

## Article

# The Potential Effectiveness of Biochar Application to Reduce Soil Cd Bioavailability and Encourage Oak Seedling Growth

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**Abstract:** Today, it is very important to protect plants in soils contaminated with metals. We investigated the behavior of cadmium during the establishment of oak seedlings (*Quercus castaneifolia* C.A. Mey.) under biochar influence. This study was conducted in pots with loamy soil. Cadmium was added to soil at 0, 10, 30, and 50 mg per kg of soil, indicated by Control, Cd10, Cd30 and Cd50. Biochar was produced at 500–550 °C from rice husk and added at 1, 3, and 5% (wt/wt) levels, indicated by B1, B3, B5, and mixed with soil at planting in three replications. Generally, increasing biochar rates had significant effects on seedling height, diameter, and biomass. This coincided with Cd immobilization in the contaminated soil which reflects a decrease in Cd concentrations in the plant bioavailability of Cd. The tolerance index increased significantly, by 40.9%, 56%, and 60.6% in B1, B3, and B5 with Cd50, respectively, compared to polluted soil. The percent of Cd removal efficiency for Cd50 was 21%, 47%, and 67% in B1, B2, and B5, respectively. Our study highlights that biochar can reduce Cd bioavailability and improve the growth of oak seedlings in contaminated soil.

Keywords: seedling biomass; heavy metals; soil amelioration

## 1. Introduction

Anthropogenic activities that lead to producing landfill leachates, vehicle emissions, and the use of sewage water irrigation or pesticides can accumulate several metals in soils [1]. Recently, the concern for metal pollution in soil has increased and is recognized as a serious environmental problem [2,3]. Heavy metals remain in the soil for a long time and are irreversible [1]. Cadmium (Cd) is a toxic element that is non-essential for plant growth [2]. Large concentrations of Cd not only reduce growth and yield but also negatively affect plant physiological activities and can even cause plant death in higher concentrations [3,4]. Cadmium is dangerous because it is highly mobile in soil–plant systems, and it is highly toxic to plants [4]. Cadmium causes reductions in photosynthesis and subsequently in plant growth, and it can ultimately result in plant death [5]. Several biosynthesis processes have been prevented by excess Cd which disrupts the photosynthetic system in the plant. Cadmium stress, furthermore, induces necrosis, alters stomatal movements, ion homeostasis, and hence limits



the availability of water and nutrients while it also affects the activities of several key enzymes and respiration in plants [6].

Applying organic amendments of plant or animal origin has the advantage of improving soil C sequestration and is an eco-friendly method for reducing the toxicity of metals [7]. Biochar is a soil modifier that is produced through pyrolysis of feedstock, such as agricultural residues, sawdust, wood chips, etc., under controlled oxygen conditions and temperature ranging from 350 to 750 °C [8]. Mohamed et al. [9] reported that biochars can retain nutrients and organic/inorganic contaminants. Biochar can trap metals and reduce their toxic effects due to the fact of its high cation exchange capacity (CEC), high pH, and large effective surface area [10–13]. Biochar is generally rich in nutrients for the plant and when added, it increases them in the soil. [8]. Cui et al. [7] examined the effect of RHB on the soil properties of an acid sulfate soil and observed an increase in soil organic carbon (SOC), soil pH, CEC, phosphorus (P), potassium (K), and calcium (Ca) and a decrease in exchangeable Al and soluble Fe.

Biochar can alter soil properties in such a way that it reduces the mobility of inorganic elements such as Cd, subsequently increasing crop production but also nutrient retention in soil [14].

*Quercus castaneifolia* (Oak) is one of the major species with industrial valuable in the northern forests of Iran with a large distribution in the Caspian Hyrcanian mixed forests (36°16′17.9″ N 57°7′32.9″ E). Oak seeds and leaves are used in pharmaceuticals, dyeing (as one of the natural dyes), and in the leather industry [15]. Unfortunately, human degradation in nature, such as mining, has caused the toxic metal contamination of Hyrcanian forest soil. Hence, protecting the Hyrcanian forest, as one of the UNESCO's World Heritage Sites, with valuable native species is necessary. Therefore, the use of biochar as a soil amendment can be a suitable approach to improve soil quality [16] and, at the same time, improve soil C sequestration.

The present study was therefore aimed at investigating the behavior of cadmium during the establishment of oak seedlings under biochar influence and performance, i.e., to examine if the use of biochar can serve as a tool for reclamation programs in mining areas.

## 2. Materials and Methods

#### 2.1. Plant and Soil Preparation

The seedlings were obtained from Savadkoh Forest Nursery, Mazandaran ( $36^{\circ}16'17.9''$  N  $57^{\circ}7'32.9''$  E). Cadmium nitrate Cd(NO<sub>3</sub>)<sub>2</sub> as Cd solution with deionized water was used to make contamination treatments [16]. Three concentrations of Cd, including 0, 10, 30, and 50 mg per kg of soil, indicated by Control, Cd10, Cd30 and Cd50, were used in the experiment [17]. Cadmium levels were combined with three levels of biochar including 1% (B1), 3% (B2), and 5% (B5) by weight. The experimental design is presented in Table 1. The pot experiment (resulting in 39 pots in total) was conducted at Sari Agricultural Science and Natural Resources University (SANRU), Mazandaran Province, Iran ( $36^{\circ}34'46.0''$  N  $53^{\circ}11'30.4''$  E). Each seedling was planted separately in 3 kg plastic pots on 20 March 2017.

Treatments	Cd10	Cd30	Cd50
Polluted soil	Cd10	Cd30	Cd50
Control	-	-	-
B1	B1 + Cd10	B1 + Cd30	B1 + Cd50
B3	B3 + Cd10	B3 + Cd30	B3 + Cd50
B5	B5 + Cd10	B5 + Cd30	B5 + Cd50

Table 1. Experimental design for treatments.

The properties of the soil used in this study are presented in Table 2.

Physicochemical Parameter	Amount
pH	$6.67\pm0.01$
EC (dS/m)	$819 \pm 1.08$
OC (%)	$6.53 \pm 0.23$
N (%)	$0.082 \pm 0.003$
$P (mg kg^{-1})$	$29.17 \pm 0.66$
K (mg kg <sup><math>-1</math></sup> )	$550.39 \pm 6.73$
CEC ( $\text{cmol}^{(+)}$ kg <sup>-1</sup> )	$7.97 \pm 0.25$
Saturation Fluid Moisture (%)	$53.65 \pm 0.48$
FC (%)	$32 \pm 0.52$
(% Sand:Silt:Clay)	50.1:31.8:18.1
Soil Texture	Loamy

**Table 2.** Physicochemical characteristics (mean values  $\pm$  SE) of the forest nursery soil before the start of the pot experiment.

EC: electrical conductivity, OC: organic matter, N: nitrogen, P: phosphorus, K: potassium, CEC: cation exchangeable capacity, FC: field capacity.

#### 2.2. Climatic Conditions and Irrigation

Climatic conditions during the experimental period are shown in Figure 1. The Fifty-year average annual rainfall sum was 768 mm, and the mean annual temperature was 19.3 °C.



**Figure 1.** Climatic conditions during the experimental period: (**a**) mean monthly temperatures; (**b**) monthly rainfall sums.

We used tap water to irrigate the seedlings over the entire experiment. Irrigation was applied during the initial establishment of the experiment and, depending on the needs of the seedlings, two to three times a week where all tree seedlings received equal amounts of water. The amount of irrigation water was based on field capacity. Irrigation water for the whole period had neutral pH and concentrations of Cd (0.005 ppm).

#### 2.3. Feedstock and Properties of Biochar

The biochar used in this study was produced from rice husk at 550 °C. [18,19]. The biochar was characterized by elemental analysis (VarioMax CHNO Analyzer). Electrical conductivity and pH were measured using a 1:20 (biochar:water) solution [20]. The detailed chemical and physical characterizations of rice husk biochar are given in Table 3. The amounts of 1%, 3%, and 5% by weight [1] of rice husk biochar were homogeneously mixed with the soil one day before planting the seedlings.

Indicators	Value	Unit	Characterizations	Value	Unite
pН	8.14	-	Oxygen	0.001	%
ĒC	359	$dS m^{-1}$	Phosphorous	412	$ m mg~kg^{-1}$
H/C	0.36	Molar ratio	Sodium	76.1	mg kg <sup>-1</sup>
C/N ratio	101.7	-	Potassium	595	mg kg <sup>-1</sup>
CEC	18.28	$cmol(+) kg^{-1}$	Calcium	609	mg kg <sup>-1</sup>
Carbon	68.03	%	Magnesium	163	mg kg <sup>-1</sup>
Nitrogen	0.64	%	Iron	65	$ m mg~kg^{-1}$
Hydrogen	25.12	%	Zinc	11.5	$ m mg~kg^{-1}$

Table 3. Characteristics of rice husk biochar.

EC, electrical conductivity; CEC, cation exchangeable capacity; H/C, hydrogen/carbon ratio; C/N, carbon/nitrogen ratio.

#### 2.4. Seedling Height, Growth, and Biomass Analysis

The height and diameter of the growing seedlings were measured once monthly, and the biomass of the seedlings was recorded after harvest at the end of the experiment (October 15, 2017). After the end of the experiment, the seedlings were washed with distilled water and then cut into leaves, stems, and roots to obtain a constant weight at 70 °C. In order to measure cadmium concentration in plant organs, a laboratory mixer (IKA Labortechnik M20) was used to powder the samples.

#### 2.5. Selected Soil Properties and Heavy Metal Detection

The air-dried soil samples were analyzed for EC (by 1:5/soil:water suspension), pH (by 1:2.5/soil: water suspension), OC (by Walkley–Black method) [21], and CEC (by EDTA extraction method) [22]. The total nitrogen of the soil was measured by Kjeldahl and available phosphorus was measured by extraction with sodium bicarbonate method [23]. Measurement of soluble potassium in soil saturation extract was measured following the method of Helmke and Sparks [24]. The concentration of potassium in the extracts was measured using the Jenway PFP7 Flame Photometer.

Bioavailable cadmium was determined by the single extraction method [25]. To determine Cd concentrations in powdered plant material, 0.5 g of homogenized samples, 0.5 ml of 37% hydrochloric acid, 9.0 mL of 69% nitric acid, and 1 mL of 30% hydrogen peroxide was added into the digestion vessel in an open system. All chemical materials were purchased from Sigma-Aldrich Corporation. After the digestion program, the Cd concentration in samples was analyzed by an atomic absorption spectrophotometer (PinAAcle 900F).

The tolerance index (TI%) [26] and removal efficiency (RE%) [27] were calculated by the following formulas:

$$\mathrm{TI}(\%) = \left(1 - \frac{\overline{X}_{\mathrm{f}}}{X_{\mathrm{c}}}\right) \times 100 \tag{1}$$

where  $\overline{X}_{f}$  is the mean biomass in polluted soil, and  $X_{c}$  is the mean biomass in the control.

$$\operatorname{RE}(\%) = \left[1 - \frac{C}{C_0}\right] \times 100 \tag{2}$$

where  $C_0$  and C are the initial and final extractable Cd concentrations in the soil, respectively.

## 2.6. Data Analysis

The experiment was based on one-factorial with three replications and the Student-Newman-Keuls (SNK)test was performed to test for significant differences among treatments (p < 0.05). The data's normal distribution test was evaluated using the Kolmogorov–Smirnov test and homogeneity of variance was checked using Levene's test. SPSS 23.0 was used to analyze the data.

## 3. Results

#### 3.1. Seedling Growth

Polluted soil significantly decreased the diameter of seedlings and height compared to the control (p < 0.05) (Figure 2). There was no significant difference in the diameter (Figure 2a) and height (Figure 2b) seedlings among treatments containing biochar and the control, except for treatment B1. However, no significant difference was observed between Cd levels in biochar treatments. But, in general, the polluted soil (without biochar) had a significantly lower diameter and height of seedlings than the biochar treatments.



**Figure 2.** Effect of different levels of biochar on diameter growth and seedling height in oak seedlings at different levels of Cd contamination (means +/– standard deviation): diameter growth (**a**) and seedling height (**b**). The dashed line represents the control (no biochar, no Cd), and the term "Polluted soil" indicates pots with Cd but without biochar. Different letters represent significant differences between treatments. The star indicates a significant difference between treatments and control.

## 3.2. Seedling Biomass

Cadmium treatments significantly decreased the leaf dry biomass, stem dry biomass, root dry biomass, and total dry biomass of the oak seedlings (Figure 3), stronger so with increasing Cd amendment level (p < 0.05), while biochar amendments increased the biomasses of all plant organs compared to the same pollution levels without the biochar amendment. Again, an amendment of 5% of biochar always showed the best results at all contamination levels of Cd.



Figure 3. Cont.



**Figure 3.** Effect of different levels of biochar on: (a) leaf biomass; (b) stem biomass; (c) root biomass; and (d) total biomass in oak seedlings at different levels of Cd contamination (bars represent means +/– one standard deviation). The dashed line represents the control (no biochar, no Cd), and the term "Polluted soil" indicates pots with the Cd but without biochar. Different letters represent significant differences between treatments. The star indicates a significant difference between treatments and control.

## 3.3. Chemical Soil Properties

The results showed that only the main effect of biochar on soil properties was significant (p < 0.05). In the case of soil pH, the application of 3% and 5% biochar showed significant increases than control (from 6.42 to 7.17 and 7.45, respectively) (Table 4). There was no significant difference in soil EC between biochar application rates and control (Table 4). B3 and B5 caused a significant increase (p < 0.05) in the OC content. Also, B3 and B5 significantly increased CEC (p < 0.05) by 9% and 20%, respectively (Table 4). In the case of N, P, and K, all the treatments with biochar had significantly higher values than the control (p < 0.05).

Percent of Biochar	рН	EC (dS m <sup>-1</sup> )	OC (%)	CEC (cmol(+) kg <sup>-1</sup> )	N (mg kg <sup>-1</sup> )	P (mg kg <sup>-1</sup> )	K (mg kg <sup>-1</sup> )
0 (Control)	$6.4 \pm 0.02^{\text{ b}}$	$0.31 \pm 0.02^{a}$	$1.1\pm0.02$ <sup>b</sup>	8.11 ± 1.02 <sup>c</sup>	$0.15 \pm 0.01$ <sup>c</sup>	12.1 ± 1.2 <sup>c</sup>	8.3 ± 1.2 <sup>c</sup>
1% 3% 5%	$6.5 \pm 0.01 \text{ b}$ $7.17 \pm 0.03 \text{ a}$ $7.45 \pm 0.02 \text{ a}$	$\begin{array}{c} 0.27 \pm 0.05 \; ^{a} \\ 0.26 \pm 0.02 \; ^{a} \\ 0.23 \pm 0.04 \; ^{a} \end{array}$	$1.2 \pm 0.04 ^{b}$ $1.57 \pm 0.01 ^{a}$ $1.92 \pm 0.03 ^{a}$	$\begin{array}{l} 12.31 \pm 1.01 \ ^{\rm b} \\ 13.35 \pm 1.03 \ ^{\rm ab} \\ 14.68 \pm 0.44 \ ^{\rm a} \end{array}$	$0.4 \pm 0.03 ^{b}$ $0.5 \pm 0.05 ^{b}$ $0.91 \pm 0.08 ^{a}$	$21.1 \pm 2.1^{b}$ $35.5 \pm 3.3^{a}$ $39.6 \pm 3.9^{a}$	$15.4 \pm 1.1^{\text{ b}}$ 29.5 ± 1.3 <sup>a</sup> 32.7 ± 0.4 <sup>a</sup>

Table 4. Soil chemical properties (mean values  $\pm$  SE) in presence of various percent of biochar.

EC: electrical conductivity; OC, organic carbon; CEC, cation exchangeable capacity. Mean values with the same letters were not significantly different (p > 0.05).

#### 3.4. Tolerance Index and Bioavailability

Increasing the biochar rate increased the tolerance index at each Cd concentration level (Figure 4). The highest tolerance index was observed with B5 (Figure 4a). However, B1 and B3 also significantly improved the tolerance index, but 5% addition had the best effect at all levels of Cd pollution (p < 0.05). All biochar amendments decreased the bioavailability of Cd, which was most significant at biochar amendment rates of 3% and 5% and partly significant at 1% biochar application (p < 0.05) (Figure 4b). The lowest bioavailability of Cd was generally observed at a biochar application rate of 5% (Figure 4b). Generally, the bioavailability of Cd in all treatments increased with an increasing Cd level.



**Figure 4.** Effect of different levels of biochar on: (**a**) tolerance index and (**b**) bioavailability of Cd in oak seedlings at different levels of Cd contamination (means +/– one standard error). The term "Polluted soil" indicates pots with the Cd but without biochar. Different letters represent significant differences between treatments.

## 3.5. Cadmium in Plant Tissues

Higher soil Cd concentrations significantly caused greater Cd concentrations in all plant organs (Figure 5). However, using biochar amendments decreased the concentration of Cd in plant tissues significantly (p < 0.05) compared to the control at the same Cd pollution level which was the most pronounced in the leaves (Figure 5a) and the least pronounced (although still mostly significant) in the roots (Figure 5c). Again, B5 treatment caused the lowest concentration of Cd in plant tissues in all pollution levels of Cd.



Figure 5. Cont.



**Figure 5.** Effect of different levels of biochar on the concentrations of (**a**) Cd in leaves; (**b**) Cd in the stem; and (**c**) Cd in root tissue in oak seedlings at different levels of Cd contamination (means +/– one standard error). The term "Polluted soil" indicates pots with the Cd but without biochar. Different letters represent significant differences between treatments.

## 3.6. Cadmium Removal Efficiency

Increasing the amount of biochar amendment significantly decreased the Cd removal efficiency throughout all Cd pollution levels. The maximum removal efficiency for all Cd treatments was achieved with B5, while the minimum removal efficiency was observed with B1 (Figure 6).



**Figure 6.** Effect of different levels of biochar on the removal efficiency in oak seedlings at different levels of Cd contamination (means +/– one standard error). Different letters represent significant differences between treatments.

With increasing biochar application level, Cd removal efficiency increased (Figure 7).



**Figure 7.** Cadmium removal efficiency as a function of Cd contamination level and biochar amendment level.

#### 4. Discussion

As expected, increasing levels of Cd treatments significantly reduced plant diameter, height growth, and seedlings biomass (Figures 2 and 3). Kukier and Chaney [28] reported that Cd, by disrupting nutrient absorption and photosynthesis, reduces plant growth and biomass production. Similar results were reported by other researches [29,30].

In general, biochar had a positive effect on seedling growth and establishment, alleviating Cd-induced reductions in growth and development. Compared to biochar-free Cd-contaminated soils, biochar application increased seedling height and diameter, root, shoot, and leaf growth and hence the tolerance index mainly by reducing the amount of Cd that could be taken up by the seedlings [31]. Cation exchange capacity is one of the key factors in soil conditions that provides a high level of nutritional cations for plant use [6]. The natural CEC of the soil was 7.97  $\text{cmol}^{(+)}$  kg<sup>-1</sup> before adding biochar (Table 2) and which was increased by adding B1, B3, and B5 to 12.31, 13.65, and 14.68  $\text{cmol}^{(+)}$  kg<sup>-1</sup>, respectively (Table 4). This means an increase in the provision of cations in the soil for plant roots [32]. On the other hand, the high efficiency of biochar to adsorb metals and then increase plant growth [33] comes from its high porosity, a high number of functional groups, and high pH [34,35]. Also, the mineral components in biochar, including phosphates and carbonates, can precipitate with metals and reduce their bioavailability [36,37]. The results of the meta-analysis by Chen et al. [38] claimed that increases in the pH and CEC of soils are the main mechanisms of biochar in reducing metal bioavailability. The soil pH controls the solubility and bioavailability of the chemicals in the soil, both nutrients and pollutants, and therefore influences soil quality, crop productivity, and environment pollution [39]. In this study, the soil pH was slightly acidic (6.67). The bioavailability of micronutrients was higher than that of neutral-alkaline soils (soil pH > 7), enhancing crop productivity [40], but at the same time, the solubility of heavy metals and some of these nutrients may pass the critical level of environmental toxicity in these soils. Biochars, in general, have the potential and event priority to substitute lime for improving the properties of acidic soils and, therefore, enhanced plant growth [41]. Perhaps, the increase in pH (16%) and CEC (20%) compared to the control in the present investigation were related to the same reason.

Besides that, the use of biochar had positive effects on soil macronutrients (Table 4). Biochars, generally, are rich in plant nutrients (except for N when woody) and when added to soil, they usually increase soil fertility [30]. Masulili et al. [42] examined the effect of rice husk biochar on the soil properties of an acid sulfate soil and observed an increase in SOC, soil pH, CEC, N, P, K, and Ca and a decrease in exchangeable Al and soluble Fe. However, the amount of Mg and Na remained unaffected by biochar. Singh et al. [43] reported a significant increase in the water holding capacity, total C, N, and P, and soil moisture content.

According to the results, the addition of biochar significantly reduced the bioavailability of Cd in the soil (Figure 4), and the Cd concentrations in soil decreased with increasing biochar application levels (although not linearly). A reduced bioavailability of metals with biochar has been reported previously for Cd contamination in sandy soil to maize [44] and Cu toxicity in sandy soil to quinoa seedlings [45]. Several researchers [38,46,47] showed that biochar application reduced the concentration of metals and improved the condition of the plant and soil in terms of contamination because of its high surface area and CEC, as well as the presence of carboxyl, phenolic, hydroxyl, and other functional groups that contain superficial oxygen. On the other hand, the application of biochar often increases the pH value significantly, as shown by Lucchini et al. [48], and could hence decrease the mobility of the Cd.

The mechanism for this is that biochar increases the adsorption capacity of cations by increasing the negative charge on the soil surface [49–51]. Also, biochar is a porous substrate with a high surface area for the absorption of heavy metals and the formation of a complex with soluble organic carbon which immobilizes heavy metals [52]. According to Table 4, soil containing treatment B5 had the highest amount of organic carbon (1.92%). Similar results were reported by other studies [53,54].

Our findings show that the addition of biochar significantly reduced the concentration of Cd in the tissue of plants (Figure 5) as reported previously for maize by Liu et al. [3]. The low concentration of metals in the plant was directly related to the loss of metal bioavailability in the soil [38]. The pH of soil with the B5 treatment increased by 1.03 in the soil over the control. This increase may be related to pH = 8.14 in biochar. Li et al. [54] stated that the high pH of biochar is due to the dominance of carbonates and organic anions over ash. Generally, data on Cd accumulation in roots, stems, and leaves shows that Cd accumulates in the roots more than other plant organs (Figure 5), a strategy that was also employed by quinoa seedlings regarding Cu toxicity [45]. Due to the toxicity of Cd to cytosol in its free form, plant cells likely try to fix it at the root by ways such as attaching it to the cell wall [55], storing it in vacuoles [56], and chelating it with phytochelatin [55], thus reducing its toxicity. The result will be a low transfer of Cd to the upper plant organs.

As shown in Figure 6, increasing the level of biochar application decreased the cadmium concentration at all levels. Adding biochar to the soil will have a series of incremental consecutive consequences, including porosity, surface reactive sites, contact between Cd and biochar, the redox potential, and the removal efficiency of Cd [57] (Figure 7). The high CEC of biochar is directly related to its specific surface area and high porosity and therefore the CEC of soil can be increased (Table 4) [53]. The increase in negative charge due to the biochar oxidation is also a reason for the high biochar CEC [58].

#### 5. Conclusions

In this study, biochar amendments of rice husk biochar clearly reduced the bioavailability of cadmium in increasingly contaminated soil, alleviating the negative impact of the Cd pollution on seedling physiology and growth. The plant tolerance index for the highest Cd rate of 50 mg kg<sup>-1</sup> increased significantly by 40.9%, 56%, and 60.6% with rice husk biochar addition rates of 1%, 3%, and 5%, respectively. Therefore, biochar application in Cd-contaminated oak forests can be a viable solution for oak seedling establishment. It remains to be tested if contaminations with other cationic heavy metals will be likewise remediated by biochar use, but the encouraging results obtained in this study, combined with that of other studies [44,45], call for the real-world first field testing in contaminated mining soils in Iran.

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