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Effect of Austenite-to-Ferrite Phase Transformation at Grain Boundaries on PWHT Cracking Susceptibility in CGHAZ of T23 Steel

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Abstract: The post-weld heat treatment (PWHT) cracking susceptibility of a coarse grain heat-affected zone (CGHAZ) in SA213-T23 (2.25Cr-1.6W steel) that was used for boiler tubes employed in thermal power plants was investigated using a Gleeble thermal cycle simulator. The PWHT cracking susceptibility test was performed at 650 °C, 700 °C, and 750 °C, and it can be judged that the lower the reduction of the area, the more susceptible it is to PWHT cracking. The results of the test also showed higher cracking susceptibility at 650 °C and 700 °C, which mostly involved intergranular fracture, while at 750 °C, transgranular fracture was exhibited. Therefore, the PWHT cracking susceptibility is considered to be closely related to grain boundary. The microstructure of the simulated CGHAZ and PWHT at 650 °C, 700 °C, and 750 °C was observed after etching with nital and alkaline sodium picrate etchants. Alkaline sodium picrate-etched microstructures showed a white band at the grain boundary at 650 °C, 700 °C, and 750 °C, which did not appear in nital etching. An analysis of the white band using an electron probe micro-analyzer (EPMA), TEM, and nanoindentation revealed that it was intergranular ferrite depleted with C, W, and Cr as compared with that in the matrix. Based on these results, we investigated the mechanism of intergranular ferrite formation during PWHT and its effect on PWHT cracking susceptibility at 650 °C, 700 °C, and 750 °C.

Keywords: SA213-T23; Cr-Mo steel; CGHAZ; PWHT cracking; intergranular ferrite; tint etching

1. Introduction

In order to achieve a higher efficiency of thermal power plants, pipes and tubes used therein are increasingly required to sustain higher temperatures and pressures [1,2]. Therefore, SA213-T23 (2.25Cr-1.6W steel), which has better high-temperature creep strength than SA213-T22 (2.25Cr-1Mo steel), has been developed as a candidate material.

To develop SA213-T23 (2.25Cr-1.6W steel), SA213-T22 (2.25Cr-1Mo steel) was modified by the addition of tungsten (1.6%), a reduction of molybdenum (0.2%) and carbon content (0.04–0.10%), and small additions of vanadium, nitrogen, and boron [3–7].

However, in the case of T23 material, PWHT cracking (reheat cracking) has been reported in the process of post-weld heat treatment (PWHT) to mitigate the residual stress in the weld metal [8–16]. Therefore, much research has been conducted to analyze the cause of the formation of reheat cracks in 2.25Cr-1.6W steel. Nawrocki et al. explained that the cause of reheat cracking is due to the weakening



of the grain boundary by (1) the Cr-depleted and W-depleted zones formed by the M_3C and $M_{23}C_6$ carbides of Fe, Cr, and W in the grain boundaries [12,13], (2) the precipitation of an incoherent intergranular carbide M3C at the boundaries [4,5], (3) the grain boundary segregation of the Al and P elements [11,12], and (4) transgranular strengthening by the homogeneous precipitation of a fine metal carbide containing the alloying elements V and Nb [11–13].

Previous studies had been carried out from the viewpoint that the alloy components of the matrix are homogeneous. However, the research of Belde et al. recently found that there is a microsegregation in the matrix where the alloying elements are not uniform, which can affect the mechanical properties and cause cracking in the course of the heat treatment [17,18]. According to Lee et al., the $M_{23}C_6$ precipitate of the base metal in 2.25Cr-1Mo-Ti steel partially dissolves in the thermal cycle of CGHAZ, and the microsegregation of C and Cr appears as a black stripe [19,20]. It is considered that such a microsegregation can affect the mechanical properties of PWHT due to phase transformation other than the matrix.

Many researchers have studied reheat cracking, but this study aims to investigate the effect of microsegregation on the reheat cracks in the matrix during PWHT, as there has been very little research on this aspect.

2. Materials and Methods

Table 1 presents the chemical composition of the SA213-T23 (2.25Cr-1.6W) material used in this study, which was analyzed using an optical emission spectrometer (LAB LAVM 10, SPECTRO Analytical Instruments GmbH, Kleve, Germany).

Material	Chemical Composition (at. %)													
	С	Si	Mn	Ni	Cr	Мо	V	Nb	Al	Ti	В	W	Ν	Fe
SA213-T23	0.07	0.24	0.50	0.11	2.09	0.15	0.22	0.028	0.018	0.042	0.0017	1.66	0.0095	Bal

Table 1. Chemical composition of the T23 steel.

In actual welding, it is impossible to carry out an evaluation of the mechanical properties, because the heat-affected zone (HAZ) is a narrow region owing to various thermal cycles. Therefore, the PWHT cracking test was performed by using the Gleeble 3500 (Dynamic Systems Inc., New York, NY, USA) thermal cycle simulator to reproduce the thermal cycle curve (1350 °C peak) of the coarse grain heat-affected zone (CGHAZ), which was calculated by using the Sysweld software 9.5 (ESI, Paris, France) for simulation of the welding analysis. The welding conditions used in the Sysweld calculations are as follows: weld method: gas tungsten arc welding (GTAW), heat input: 16 kJ/cm, thickness: 6 mm.

Figure 1 shows the schematic diagram and specimen geometry of the PWHT cracking test. After 30 min at PWHT temperatures (650 °C, 700 °C, 750 °C), the tensile test was performed at a very low strain rate (2.5×10^{-4}). As a result of the PWHT test, if the reduction in area is low, it is concluded that cracking is likely to occur because of low grain boundary ductility, and sensitivity to reheat cracking is higher. The reason for the tensile test at a very low strain rate (2.5×10^{-4}) is to prevent plastic deformation before the brittle fracture mode due to PWHT cracking.

The microstructure of the simulated CGHAZ and PWHT was observed by etching it with nital etchant (2 mL HNO₃, 50 mL methanol) and alkaline sodium picrate etchant (2 g picric acid, 25 g NaOH, 100 mL boiled water). The fracture surface was observed by using a scanning electron microscope (SEM) (SUPRA40VP, Carl Zeiss, Oberkochen, Germany). The phase analysis of the simulated CGHAZ and PWHT was performed using an electron probe micro-analyzer (EPMA) (JXA-8530F, JEOL, Tokyo, Japan) and field-emission transmission electron microscope (FE-TEM) (TALOS F200X, FEI, Hillsboro, OR, USA), and the time-temperature-transformation diagram (TTT diagram) was obtained by using JMatPro 5.1 software (Sente Software, Surrey Research Park, UK) to analyze the experimental results.



Figure 1. Schematic illustration of a post-weld heat treatment (PWHT) cracking susceptibility test at various PWHT temperatures, and the geometry of the sample.

A thin-film specimen of CGHAZ for TEM analysis was prepared by mechanical polishing, followed by jet electropolishing using an electrolyte consisting of a mixture of 95% methanol and 5% perchloric acid, maintained at a temperature of 20 °C. A thin-film specimen of 700 °C was prepared using the focused ion beam (FIB) (FEI Scios, Thermo Fisher Scientific, Hillsboro, OR, USA).

The transgranular hardness was measured using a Vickers hardness tester (HM-200, Mitutoyo Corporation, Kanagawa, Japan), and the hardness near the grain boundaries by using a nanoindenter (TI-950, HYSITON, Minneapolis, MN, USA).

3. Results and Discussion

3.1. PWHT Cracking Susceptibility Test

Figure 2 shows the results of the PWHT cracking susceptibility test of CGHAZ at various temperatures. The test was performed by holding the simulated CGHAZ at PWHT temperatures (650 °C, 700 °C, 750 °C) for 30 min, and then, the tensile test was performed under the strain rate condition of 2.5×10^{-4} .

As the PWHT temperature increases, the tensile strength tends to decrease almost linearly to 619 MPa, 466 MPa, and 278 MPa at 650 °C, 700 °C, and 750 °C, respectively. On the other hand, the reduction in area and elongation are low at 10.4% and 5.9% at 650 °C, and 9.7% and 5.2% at 700 °C, respectively. However, the reduction in area and elongation at 750 °C increased to 40.0% and 22.5%, respectively. It is found that the lower the reduction in area, the more the susceptibility to PWHT cracking [21–23]; in other words, CGHAZ is susceptible to PWHT cracking at 650 °C and 700 °C, but not at 750 °C.



Figure 2. Results of PWHT cracking test at various temperatures.

In order to analyze the correlation between PWHT cracking susceptibility and fracture surface, the fracture surfaces of specimens that failed during the testing at 650 °C, 700 °C, and 750 °C were observed by SEM, as shown in Figure 3. The surfaces at 650 °C and 700 °C shown in Figure 3a,b largely appear as intergranular fractures, which are generally sensitive to PWHT and considered to have lower intergranular strength than transgranular strength. Figure 3d,e are the enlargements of the X and Y regions in Figure 3a,b, respectively, and show that the grain boundaries are covered by fine dimples. The surface at 750 °C shown in Figure 3c, which is not sensitive to PWHT cracking, mostly appears as a transgranular fracture. Therefore, it is considered that the intergranular strength is not lower than the transgranular strength at 750 °C. As a result of the fracture surface analysis, the reheat cracking sensitivity is considered to be correlated with intergranular and transgranular strengths.



Figure 3. Representative SEM photomicrographs of the fracture surfaces of specimens that failed during testing at (**a**) 650 °C, (**b**) 700 °C, and (**c**) 750 °C. Enlarged SEM photomicrographs (**d**–**f**) at regions denoted by X, Y, and Z.

3.2. Correlation between Microstructure and PWHT Cracking Susceptibility

Figure 4 shows the optical microscopic structure of CGHAZ at PWHT temperatures (650 °C, 700 °C, 750 °C) using nital and alkaline sodium picrate etching. In order to compare the differences between nital and alkaline sodium picrate etching, the microstructure was observed after nital etching, and it was subsequently finely polished. Afterward, alkaline sodium picrate etching was performed to observe the same position as that observed in nital etching, and the differences in the microstructures owing to etching.

The nital-etched microstructure of CGHAZ reveals the prior austenite grain boundary and lath in the matrix. The PWHT conditions at 650 °C, 700 °C, and 750 °C also showed the prior austenite grain boundary and lath in the matrix. The alkaline sodium picrate-etched microstructure of the CGHAZ also shows the prior austenite grain boundary and lath in the matrix, as in the case of nital etching, but with a black band that was not observed in the nital-etched microstructure. According to Lee's study, the $M_{23}C_6$ precipitates of the base metal are due to partial dissolution in the CGHAZ thermal cycle, and are not completely dissolved, and the fine segregation of the remaining C, W, and Cr appears as a black band in the CGHAZ [19].

White bands are not observed at the grain boundaries of CGHAZ. However, when PWHT is performed at 650 °C, 700 °C, and 750 °C, white bands appeared at the grain boundaries, which was different from the observation in CGHAZ. There is also a black band at 650 °C, 700 °C, and 750 °C, which suggests that C, W, and Cr segregations of CGHAZ are not diffused, even under PWHT. The presence of C, W, and Cr segregation in the PWHT was also examined in the EPMA analysis.



Figure 4. Optical microstructure with etchants (nital, alkaline sodium picrate) at various PWHT temperatures. (The yellow rectangle is the position observed by SEM in Figure 5).

Figure 5 shows the SEM structure of the CGHAZ at PWHT temperatures ($650 \degree C$, $700 \degree C$, $750 \degree C$) using nital and alkaline sodium picrate etching. The observed part is shown in Figure 4 as the \Box area. Nital-etched SEM microstructures show a line at the grain boundaries of the CGHAZ. However, in PWHT at 650 °C, 700 °C, and 750 °C, precipitates were observed instead of lines at the grain boundaries. Therefore, it is considered that the line of CGHAZ is transformed into precipitate during PWHT.

On the other hand, in the alkaline sodium picrate-etched microstructure, neither lines nor precipitates at grain boundaries in either CGHAZ or PWHT were observed clearly. In nital etching, the grain boundaries and precipitates are clearly visible because the matrix is reduced. However, they are not seen clearly in alkaline sodium picrate etching, because the matrix is not reduced.

CGHAZ





650 °C

Figure 5. SEM microstructure with etchants (nital, alkaline sodium picrate) at various PWHT temperatures.

Figure 6 shows the results of the TEM analysis of the precipitates using the carbon replica method in order to identify the phases of the precipitates at the grain boundaries at PWHT temperatures. The diffraction pattern of the line at the grain boundaries in CGHAZ could not be obtained, and it was considered that it was not extracted from the carbon replica. The PWHT conditions at 650 °C, 700 °C, and 750 °C showed precipitates at the grain boundaries, and the diffraction pattern analysis of these revealed that it was $M_{23}C_6$ carbide. Observation of the microstructure shows that a line exists at the grain boundaries of CGHAZ, and white bands and $M_{23}C_6$ appear at 650 °C, 700 °C, and 750 °C when PWHT is performed.



Figure 6. TEM microstructures and diffraction patterns at the regions denoted by arrows at various PWHT temperatures.

Figure 7 shows the results of EPMA mapping for analyzing the alloy compositions of the white bands and $M_{23}C_6$ carbide at the grain boundaries, and the black band at 700 °C. The white band at the grain boundary at 700 °C can be observed to be depleted of C; W and Cr are not clearly visible, but appear to be depleted. The $M_{23}C_6$ intergranular precipitates inside the white band are concentrated in C, W, and Cr. It is considered that the depleted zone is formed around the $M_{23}C_6$ precipitate because the $M_{23}C_6$ carbide with high C, W, and Cr components precipitates at the grain boundary [24–27] when PWHT is carried out at 700 °C. According to a study by Lee et al., the black band in alkaline sodium picrate-etched CGHAZ is the segregated region of C, W, and Cr [19,20]. The black band at 700 °C is also a region where C, W, and Cr components are concentrated. It can be seen that even if PWHT is performed, C, W, and Cr are not diffused, but rather remain as segregation regions.



Figure 7. Electron probe microanalyzer (EPMA) mapping results showing the distribution of elements at 700 °C.

To summarize, at 700 °C, the white band of the grain boundaries is a C, W, and Cr-depleted zone; M23C6 precipitates in which C, W, and Cr components are concentrated are present in the white band, and the black band is an area concentrated in C, W, and Cr. At 650 °C and 750 °C, it can be considered that the white band is the region where C, W, and Cr are depleted, and the black band is a region where C, W, and Cr are concentrated.

In order to identify the phase of the white band at the grain boundary, the specimen was prepared using a FIB on the grain boundaries at 700 °C and observed using TEM. In the low-magnification TEM photograph of Figure 8, bands and precipitates are observed at the grain boundary. In Figure 8b, which is an enlarged view of the region X_1 of the grain boundary, a black dot is observed inside the band, which looks like a precipitate or dislocation. However, in the high resolution (HR) image of Figure 8c, which enlarges the X_2 region of Figure 8b, the band and black dot have the same atomic arrangement. Therefore, the black spot inside the band is judged to be a dislocation, and not a precipitate. A diffraction pattern analysis of the Y_1 region was performed to identify the phase of the bands and precipitates. As a result of the analysis, the band of the grain boundary, which was a body-centred cubic (BCC) structure, was judged to be intergranular ferrite because there was no transgranular precipitation, although there was dislocation, and the precipitate of the grain boundary was $M_{23}C_6$.



Figure 8. (a) TEM micrograph at 700 °C; (b) Enlarged micrograph of the region denoted by X_1 ; (c) HR-TEM micrograph of the region denoted by X_2 ; (d) Diffraction pattern of the region denoted by Y_1 .

Figure 9 is the result of the analysis with a nanoindenter to compare the hardness of the intergranular ferrite with the matrix at 700 °C. In order to measure the hardness of intergranular ferrite with a thickness of $1.0 \sim 1.2 \,\mu$ m, the indenter size was tested to be $0.5 \,\mu$ m or less. As a result of the nanoindentation analysis, it can be seen that the hardness of intergranular ferrite is $3.92 \sim 3.94$ GPa, which is lower than that of the matrix, which is $4.43 \sim 4.51$ GPa. As a result of EPMA, nanoindentation, and TEM analysis, the white band at the grain boundary in alkaline sodium picrate etching at $650 \,^{\circ}$ C, $700 \,^{\circ}$ C, and $750 \,^{\circ}$ C is identified to be intergranular ferrite where C, W, and Cr are depleted more than in the matrix, and intergranular ferrite has a lower hardness than the matrix at $700 \,^{\circ}$ C.



Figure 9. Nanoindentation results showing hardness at 700 °C.

Therefore, it is considered that ductile intergranular ferrite is generated in the grain boundary when PWHT is carried out at 700 °C, thereby lowering the intergranular strength. This is the reason why intergranular fracture appears largely at 700 °C and is sensitive to PWHT. The fracture surfaces at 650 °C and 700 °C in Figure 3 appear as intergranular fractures mostly covered with fine dimples at the grain boundaries, which is attributed to ductile intergranular ferrite, and sensitive to reheat cracks. However, the fracture surface at 750 °C was mostly transgranular, and not sensitive to reheat cracking. It is considered that the intergranular strength is lower than transgranular strength at 650 °C and 700 °C. However, this is not the case at 750 °C.

Figure 10 shows the micro-Vickers hardness of the matrix for various PWHT temperatures. The matrix hardness at 650 °C was similar to that of the CGHAZ despite PWHT, and decreased at 700 °C and 750 °C as the temperature increased. Generally, the greater the hardness, the greater the strength [28–30]. Therefore, it is expected that as the PWHT temperature increases, the transgranular strength decreases due to the tempering effect. Therefore, the transgranular strength at 700 °C is greater than the intergranular strength, which is weakened by intergranular ferrite, but at 750 °C, it is expected to be similar to or lower than the intergranular strength.



Figure 10. Comparison of matrix hardness at various PWHT temperatures.

Figure 11 shows the results of nanoindentation analysis to compare the hardness of intergranular ferrite and the matrix at 750 °C. The analysis shows that the hardness of the matrix at 700 °C is 3.88~4.02 GPa; this is similar to that of intergranular ferrite, which is 3.94~3.98 GPa. It is considered that the transgranular strength at 750 °C is similar to the intergranular strength. Therefore, at 750 °C, the grain boundary is softened by intergranular ferrite, but the matrix is softened similarly at the grain boundary, so that intergranular fracture did not occur and CGHAZ was not sensitive to PWHT cracking.



Figure 11. Nanoindentation results showing hardness at 750 °C.

3.3. Formation Mechanism of Ferrite at Prior Austenite Grain Boundary

Figure 12 shows the results of the EPMA surface analysis for analyzing the alloy composition of the grain boundary and black band of the CGHAZ. The black band of CGHAZ is a region where C, W, and Cr components are concentrated. Lee's study showed that the $M_{23}C_6$ of the matrix in the CGHAZ thermal cycle dissolves, but is not fully diffused, revealing C, W, and Cr segregations. The line at the grain boundaries of the CGHAZ observed in the SEM microstructure of Figure 5 is the region where C is concentrated.



Figure 12. Electron probe micro-analyzer (EPMA) mapping results showing the distribution of elements in coarse grain heat-affected zone (CGHAZ).

In order to identify the phase of the line at the grain boundary of CGHAZ, a thin film was prepared and observed using TEM. In the Bright-field (BF) image of Figure 13a, the line at the grain boundary is seen. A diffraction pattern analysis of the X_1 and X_2 regions was performed to identify the phases of the line at the grain boundary and the matrix. As a result of the analysis, the diffraction patterns shows a Nishiyama-Wasserman orientation relationship between the martensite and the austenitic phase (5.26°) at boundaries. The line at the grain boundary was found to be austenite of Face Centered Cubic (FCC) structure and the matrix was of BCC structure, which was determined to be martensite considering the rapid cooling rate and hardness (340 Hv) of CGHAZ. To conclude, the result of the EPMA and TEM analysis indicated that the line at the grain boundary of CGHAZ was austenite with carbon concentration.



Figure 13. (a) TEM micrograph of CGHAZ; (b,c) Diffraction patterns at the regions denoted by X_1 and X_2 .

In order to investigate the reason for the generation of intergranular ferrite during PWHT, the TTT-curve, according to the alloy composition of the matrix, was obtained, as shown in Figure 14, by using JmatPro simulation. In the case of the PWHT of the alloy composition of the matrix for 30 min at 650 °C, 700 °C, and 750 °C, the austenite at the grain boundary of CGHAZ does not transform into ferrite, because the ferrite nose time is longer than 30 min. However, the grain boundaries are regions where the C, W, and Cr components are depleted by the $M_{23}C_6$ precipitation, due to which the ferrite nose time decreases [31]. In such a situation, austenite at the grain boundary of CGHAZ transforms into intergranular ferrite.



Figure 14. Time-temperature-transformation (TTT) diagrams for an alloy composition of matrix (Fe-0.07%C-2.09Cr-1.66W) calculated with JMatPro software. (**a**) Ferrite nose time of matrix alloy composition; (**b**) Predicted ferrite nose time when C, W, Cr is depleted.

Figure 15 is a schematic diagram for explaining microstructural evolution. (a) There are precipitates at the crystal grain boundaries of the base material. (b) When heated to 1350 °C, the matrix of the base material is transformed into austenite. The precipitates dissolve, and the grain size becomes coarse. The precipitates are not completely dissolved in the rapid heating of CGHAZ, and there are Cr, W, and C segregations. (c) When rapidly cooled to room temperature, the matrix transforms into martensite, and austenite with carbon concentration appears at grain boundaries. (d) When performing PWHT at 650 °C, 700 °C, and 750 °C, the martensite is tempered. C, W, and Cr are depleted at the grain boundary by M23C6 precipitation, and austenite at grain boundary is transformed into ferrite.





4. Conclusions

PWHT cracking susceptibility and microstructural characteristics of CGHAZ in SA213-T23 were investigated at various PWHT temperatures. The findings are summarized as follows:

- (1) As a result of the PWHT cracking susceptibility test of CGHAZ at 650 °C, 700 °C, and 750 °C, it was found that CGHAZ was sensitive to PWHT at 650 °C and 700 °C, but not at 750 °C. The surfaces at 650 °C and 700 °C contain intergranular fractures, but at 750 °C, they contain transgranular fractures.
- (2) In order to investigate the microstructural characteristics of CGHAZ at PWHT temperatures, microstructures were observed by etching with nital and alkaline sodium picrate etchants. Alkaline sodium picrate-etched microstructures at 650 °C, 700 °C, and 750 °C showed white bands at grain boundaries that did not appear in nital etching. In the EPMA and TEM analyses, the white band at the grain boundaries of 650 °C, 700 °C, and 750 °C was intergranular ferrite depleted in C, W, and Cr.
- (3) The reason why an intergranular fracture appears largely at 650 °C and 700 °C and is sensitive to PWHT cracking is that the intergranular strength is lower than the transgranular strength due to ductile intergranular ferrite. The intergranular ferrite appears due to C and Cr depletion by M23C6 precipitation during PWHT. At 750 °C, the surface fracture is mostly transgranular fracture, and is not susceptible to PWHT cracking. The reason is that the intergranular ferrite lowers the intergranular strength, but the transgranular strength is also lowered by the high tempering effect, so that the intergranular strength and the transgranular strength are similar to each other.
- (4) A line observed at the grain boundary of CGHAZ was found to be austenite with carbon concentration, on the basis of EPMA and TEM analysis.
- (5) In the case of the PWHT of the alloy composition of the matrix for 30 min at 650 °C, 700 °C, and 750 °C, the austenite at the grain boundary of CGHAZ does not transform into ferrite, because the ferrite nose time in the TTT curve is longer than 30 min. However, the grain boundary is a region where C, W, and Cr are depleted by M₂₃C₆ precipitation with high C, W, and Cr components. When C, W, and Cr are depleted, the ferrite nose time becomes shorter than 30 min, so that austenite at the grain boundary of CGHAZ is then transformed into ferrite.

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