

Supplementary

Microstructure and Phase Transition of $\text{Ag}_{50.5}\text{Cu}_{33.3}\text{Sn}_{16.2-x}\text{In}_x$ Alloys through Experimental Study and Thermodynamic Calculation

Qingsong Tong, Maohua Rong * and Jiang Wang *

School of Materials Science and Engineering & Guangxi Key Laboratory of Information Materials, Guilin University of Electronic Technology, Guilin 541004, China; 13635644432@163.com

* Correspondence: rongmh124@guet.edu.cn (M.R.); waj124@guet.edu.cn (J.W.)

Table S1 Thermodynamic parameters of the Ag-Cu-Sn-In quaternary system

Phase	Thermodynamic parameters	Reference
liquid	${}^0L_{Ag,Cu}^{liq} = 16914.949 - 14.7721T + 1.54955T\ln T$; ${}^1L_{Ag,Cu}^{liq} = -1963.300 + 0.8623T$	[28]
	${}^0L_{Ag,In}^{liq} = -26499.412 + 66.148718T - 8.6214682T\ln T$	[29]
	${}^1L_{Ag,In}^{liq} = -38058.4 + 143.60133T - 17.657918T\ln T$	[29]
	${}^2L_{Ag,In}^{liq} = -16572.628 + 121.0128T - 15.89588T\ln T$; ${}^3L_{Ag,In}^{liq} = 5791.2978$	[29]
	${}^0L_{Ag,Sn}^{liq} = -3177.49 - 10.16124T + 0.380505T\ln T$	[30]
	${}^1L_{Ag,Sn}^{liq} = -16782.28 + 2.06521T + 0.437477T\ln T$	[30]
	${}^2L_{Ag,Sn}^{liq} = 3190.34 - 107.09456T + 13.954838T\ln T$	[30]
	${}^0L_{Cu,In}^{liq} = -41564.8 + 238.616T - 29.827T\ln T$; ${}^1L_{Cu,In}^{liq} = -76057.8 + 371.306T - 44.994T\ln T$	[31]
	${}^2L_{Cu,In}^{liq} = -42076.5 + 192.395T - 23.281T\ln T$	[31]
	${}^0L_{Cu,Sn}^{liq} = -9002.8 - 5.8381T$; ${}^1L_{Cu,Sn}^{liq} = -20100.4 + 3.63666T$; ${}^2L_{Cu,Sn}^{liq} = -10528.40$	[32]
	${}^0L_{In,Sn}^{liq} = -711 - 1.6934T$; ${}^1L_{In,Sn}^{liq} = -64 - 1.3592T$	[33]
Fcc	${}^0L_{Ag,Cu,Sn}^{liq} = -80000 + 27.98T$; ${}^1L_{Ag,Cu,Sn}^{liq} = -85233 + 29.44T$; ${}^2L_{Ag,Cu,Sn}^{liq} = -40000 + 26.56T$	[12]
	${}^0L_{Ag,Cu,In}^{liq} = -50066 + 24.15T$; ${}^1L_{Ag,Cu,In}^{liq} = -134807 + 94.27T$; ${}^2L_{Ag,Cu,In}^{liq} = 31076 - 167.55T$	[34]
	${}^0L_{Ag,In,Sn}^{liq} = 64697 - 8.82T$; ${}^1L_{Ag,In,Sn}^{liq} = 23474 - 22.792T$; ${}^2L_{Ag,In,Sn}^{liq} = 13374 - 27.171T$	[34]
	${}^0L_{Cu,In,Sn}^{liq} = 92284.1485 - 57.897T$; ${}^1L_{Cu,In,Sn}^{liq} = -17306.1096 + 20.828T$	[34]
	${}^2L_{Cu,In,Sn}^{liq} = -1332.422 - 26.208T$	[34]
	${}^0L_{Ag,Cu}^{Fcc} = 32580.365 - 7.4547T$; ${}^1L_{Ag,Cu}^{Fcc} = -10144.596 + 5.5617T$	[28]
	${}^0L_{Ag,In}^{Fcc} = -17978.868 + 10.39293T$; ${}^1L_{Ag,In}^{Fcc} = -32794.133$	[29]
	${}^0L_{Ag,Sn}^{Fcc} = 745.45 + 11.498027T$; ${}^1L_{Ag,Sn}^{Fcc} = -36541.5$	[30]
	${}^0L_{Cu,In}^{Fcc} = -6475.9 + 21.830T$; ${}^1L_{Cu,In}^{Fcc} = -29935.2 - 5.672T$; ${}^2L_{Cu,In}^{Fcc} = 47350.2 - 40.210T$	[31]
	${}^0L_{Cu,Sn}^{Fcc} = -10672 - 1.4837T$; ${}^1L_{Cu,Sn}^{Fcc} = -15331.3 + 6.9539T$	[32]
	${}^0L_{In,Sn}^{Fcc} = 25000$	[33]
Bcc	${}^0L_{Ag,Cu,In}^{Fcc} = -273196 + 190T$; ${}^1L_{Ag,Cu,In}^{Fcc} = 85251.5$	[34]
	${}^0L_{Ag,In,Sn}^{Fcc} = 105437 - 27.156T$; ${}^1L_{Ag,In,Sn}^{Fcc} = 173115$; ${}^2L_{Ag,In,Sn}^{Fcc} = 121030$	[34]
	${}^0L_{Cu,In,Sn}^{Fcc} = -55000$; ${}^1L_{Cu,In,Sn}^{Fcc} = -55000$; ${}^2L_{Cu,In,Sn}^{Fcc} = -55000$	[34]
	${}^0L_{Ag,In}^{Bcc} = 2858.3 - 14.059T$; ${}^1L_{Ag,In}^{Bcc} = -80876.7 + 17.445T$; ${}^2L_{Ag,In}^{Bcc} = 31125.1$	[29]
	${}^0L_{Cu,In}^{Bcc} = -20532.763 + 12.724T$; ${}^1L_{Cu,In}^{Bcc} = -13379.27 - 12.358T$	[31]
	${}^0L_{Cu,Sn}^{Bcc} = -32656.8 + 25.0158T$; ${}^1L_{Cu,Sn}^{Bcc} = -13862.5 - 32.0218T$	[32]
	${}^2L_{Cu,Sn}^{Bcc} = -4175.47 + 5.0083T$	[32]
	${}^0L_{Ag,Cu}^{Bcc} = -12440.6 + 24.382T$	[34]
	${}^0L_{In,Sn}^{Bcc} = 25000$	[34]
	${}^0L_{Ag,Cu,In}^{Bcc} = 313800 - 266.2T$; ${}^1L_{Ag,Cu,In}^{Bcc} = 210500 - 276.05T$; ${}^2L_{Ag,Cu,In}^{Bcc} = 1200$	[34]

	${}^0L_{Cu,In,Sn}^{Bcc} = 18052.7; {}^1L_{Cu,In,Sn}^{Bcc} = -9000; {}^2L_{Cu,In,Sn}^{Bcc} = 25000$	[34]
	${}^0L_{Ag,In}^{Hcp} = -15502.468 + 9.129T; {}^1L_{Ag,In}^{Hcp} = -60975.391 + 4.723T; {}^2L_{Ag,In}^{Hcp} = 31769.287$	[29]
	${}^0L_{Ag,Sn}^{Hcp} = 1046.1 + 10.23693T; {}^1L_{Ag,Sn}^{Hcp} = -40505.5$	[30]
Hcp	${}^0L_{Ag,Cu}^{Hcp} = 20573.1$	[34]
	${}^0L_{In,Sn}^{Hcp} = 25700$	[34]
	${}^0L_{Ag,Cu,In}^{Hcp} = -308146.1 + 334.359T$	[34]
	${}^0L_{Ag,In,Sn}^{Hcp} = 9456 - 10.641T; {}^1L_{Ag,In,Sn}^{Hcp} = 77280 + 1.308T; {}^2L_{Ag,In,Sn}^{Hcp} = 76616 - 29.97T$	[34]
β -InSn	${}^0L_{In,Sn}^{\beta-InSn} = 35.7 - 4.561T$	[33]
γ -InSn	${}^0L_{In,Sn}^{\gamma-InSn} = -15715.5 + 19.3402T$	[33]
Bct	${}^0L_{In,Sn}^{Bct} = -239 + 2.8509T$	[33]
Tetragonal	${}^0L_{In,Sn}^{Tetragonal} = 718 - 1.495T; {}^1L_{In,Sn}^{Tetragonal} = -2217$	[33]
	${}^0G_{Ag:In}^{AgIn2} = -8059.1667 + 8.583T + 0.333{}^0G_{Ag}^{Fcc} + 0.667{}^0G_{In}^{Tetragonal}$	[29]
AgIn ₂	${}^0G_{Cu:In}^{AgIn2} = 0.333{}^0G_{Cu}^{Fcc} + 0.667{}^0G_{In}^{Tetragonal}$	[29]
	${}^1G_{Ag,Cu:In}^{AgIn2} = -9466.667$	[29]
	${}^0G_{Ag:Ag:In:In}^{Ag9In4} = -8200.208 - 0.48T + 0.692{}^0G_{Ag}^{Fcc} + 0.308{}^0G_{In}^{Tetragonal}$	[29]
Ag ₉ In ₄	${}^0G_{Ag:In:In:In}^{Ag9In4} = 0.462{}^0G_{Ag}^{Fcc} + 0.538{}^0G_{In}^{Tetragonal}$	[29]
	${}^0G_{Ag:Cu:In:In}^{Ag9In4} = -3150.854 + 3.578T + 0.462{}^0G_{Ag}^{Fcc} + 0.231{}^0G_{Cu}^{Fcc} + 0.307{}^0G_{In}^{Tetragonal}$	[29]
	${}^0G_{Ag:Ag,In:In:In}^{Ag9In4} = -5300; {}^0G_{Ag:Ag,Cu:In:In}^{Ag9In4} = -8174.992$	[29]
Ag ₃ Sn	${}^0G_{Ag,Ag}^{Ag3Sn} = 4750 - 0.5T + {}^0G_{Ag}^{Fcc}$	[30]
	${}^0G_{Ag,Sn}^{Ag3Sn} = -11085.3 + 110.01471T - 23.18TlnT - 0.00359T^{-2} + 4389.5T^{-1}$	[30]
	${}^0G_{Cu,Cu:In}^{\gamma-CuIn} = -2201.8 - 3.443T + 0.769{}^0G_{Cu}^{Fcc} + 0.231{}^0G_{In}^{Tetragonal}$	[31]
	${}^0G_{Cu:In:In}^{\gamma-CuIn} = -7131.6 + 0.112T + 0.654{}^0G_{Cu}^{Fcc} + 0.346{}^0G_{In}^{Tetragonal}$	[31]
γ -CuIn	${}^0G_{Ag:Ag:In}^{\gamma-CuIn} = -1620.0692 - 6.5712T + 0.769{}^0G_{Ag}^{Fcc} + 0.231{}^0G_{In}^{Tetragonal}$	[34]
	${}^0G_{Ag:In:In}^{\gamma-CuIn} = 200 + 0.654{}^0G_{Ag}^{Fcc} + 0.346{}^0G_{In}^{Tetragonal}; {}^0G_{Ag:Ag,In:In}^{\gamma-CuIn} = -8456.9907 + 9.1312T$	[34]
	${}^0G_{Ag,Cu,Cu:In}^{\gamma-CuIn} = -24700 - 25T$	[34]
δ -Cu ₇ In ₃	${}^0G_{Cu:In}^{\delta-Cu_7In_3} = -7991.3 + 1.170T + 0.7{}^0G_{Cu}^{Fcc} + 0.3{}^0G_{In}^{Tetragonal}$	[31]
	${}^0G_{Ag:In}^{\delta-Cu_7In_3} = 0.7{}^0G_{Cu}^{Fcc} + 0.3{}^0G_{In}^{Tetragonal}$	[29]
	${}^1G_{Ag,Cu:In}^{\delta-Cu_7In_3} = 53045.9 - 93.729T$	[29]
η -LT	${}^0G_{Cu,In}^{\eta-LT} = -8173.8 + 1.380T + 0.64{}^0G_{Cu}^{Fcc} + 0.36{}^0G_{In}^{Tetragonal}$	[31]

	${}^0G_{Cu:In}^{Cu_{11}In_9} = -7525.6 + 1.703T + 0.55{}^0G_{Ag}^{Fcc} + 0.45{}^0G_{In}^{Tetragonal}$	[31]
Cu ₁₁ In ₉	${}^0G_{Ag:In}^{Cu_{11}In_9} = 0.55{}^0G_{Ag}^{Fcc} + 0.45{}^0G_{In}^{Tetragonal}$	[29]
	${}^0G_{Ag,Cu:In}^{Cu_{11}In_9} = 10800 - 12T$	[34]
	${}^0G_{Cu:Sn}^{Cu_3Sn} = -8194.2 - 0.2043T + 0.75{}^0G_{Cu}^{Fcc} + 0.25{}^0G_{Sn}^{Bct}$	[32]
Cu ₃ Sn	${}^0G_{Cu:In}^{Cu_3Sn} = -7200 - 4T + 0.75{}^0G_{Cu}^{Fcc} + 0.25{}^0G_{In}^{Tetragonal}$	[34]
	${}^0G_{Cu:In,Sn}^{Cu_3Sn} = -5800 + 3T$	[34]
Cu ₁₀ Sn ₃	${}^0G_{Cu:Sn}^{Cu_{10}Sn_3} = -6655.1 - 1.485T + 0.769{}^0G_{Cu}^{Fcc} + 0.231{}^0G_{Sn}^{Bct}$	[32]
	${}^0G_{Cu:Sn}^{Cu_{41}Sn_{11}} = -6323.5 - 1.2808T + 0.788{}^0G_{Cu}^{Fcc} + 0.212{}^0G_{Sn}^{Bct}$	[32]
Cu ₄₁ Sn ₁₁	${}^0G_{Cu:In}^{Cu_{41}Sn_{11}} = -5350 + 1.5T + 0.788{}^0G_{Cu}^{Fcc} + 0.212{}^0G_{In}^{Tetragonal}$	[34]
	${}^0G_{Cu:In,Sn}^{Cu_{41}Sn_{11}} = -13000 + 14.77T$	[34]
Cu ₆ Sn ₅ -l	${}^0G_{Cu,Sn}^{Cu_6Sn_5-l} = -7129.7 + 0.4059T + 0.545{}^0G_{Cu}^{Fcc} + 0.455{}^0G_{Sn}^{Bct}$	[32]
	${}^0G_{Cu:Cu:In}^{Cu_6Sn_5-h} = -6301.5 - 0.940T + 0.667{}^0G_{Cu}^{Fcc} + 0.333{}^0G_{In}^{Tetragonal}$	[31]
	${}^0G_{Cu,Cu:Sn}^{Cu_6Sn_5-h} = 3200 + 2T + 0.667{}^0G_{Cu}^{Fcc} + 0.333{}^0G_{Sn}^{Bct}$	[32]
	${}^0G_{Cu:In:In}^{Cu_6Sn_5-h} = -156.7 - 7.030T + 0.545{}^0G_{Cu}^{Fcc} + 0.455{}^0G_{In}^{Tetragonal}$	[32]
	${}^0G_{Cu:In,Sn}^{Cu_6Sn_5-h} = -2492 + 4T + 0.545{}^0G_{Cu}^{Fcc} + 0.122{}^0G_{In}^{Tetragonal} + 0.333{}^0G_{Sn}^{Bct}$	[34]
	${}^0G_{Cu:Sn:In}^{Cu_6Sn_5-h} = 16000 + 0.545{}^0G_{Cu}^{Fcc} + 0.122{}^0G_{Sn}^{Bct} + 0.333{}^0G_{In}^{Tetragonal}$	[34]
	${}^0G_{Cu:Sn:Sn}^{Cu_6Sn_5-h} = -6869.5 - 0.1589T + 0.545{}^0G_{Cu}^{Fcc} + 0.455{}^0G_{Sn}^{Bct}$	[32]
	${}^0G_{Cu:Cu,In:In}^{Cu_6Sn_5-h} = -14526.5 + 18.020T$	[31]
Cu ₆ Sn ₅ -h	${}^0G_{Cu,Cu,In,Sn}^{Cu_6Sn_5-h} = -14526.546 + 18.020T; {}^0G_{Cu:Cu,Sn:In}^{Cu_6Sn_5-h} = -37093.16 + 18.520T$	[34]
	${}^0G_{Cu:Cu,Sn:Sn}^{Cu_6Sn_5-h} = -8300; {}^0G_{Cu:In,Sn:In}^{Cu_6Sn_5-h} = -8300; {}^0G_{Cu:In,Sn:Sn}^{Cu_6Sn_5-h} = -8300$	[34]
	${}^0G_{Cu:Cu,In,Sn}^{Cu_6Sn_5-h} = -19650.8 - 0.4T; {}^0G_{Cu:In,In,Sn}^{Cu_6Sn_5-h} = -44570.8 + 39.6T$	[34]
	${}^0G_{Cu:Sn,In,Sn}^{Cu_6Sn_5-h} = -30000 + T$	[34]
	${}^0G_{Ag:Cu:In}^{Cu_6Sn_5-h} = 200 + 0.545{}^0G_{Ag}^{Fcc} + 0.122{}^0G_{Cu}^{Fcc} + 0.333{}^0G_{In}^{Tetragonal}$	[34]
	${}^0G_{Ag:In:In}^{Cu_6Sn_5-h} = 200 + 0.545{}^0G_{Ag}^{Fcc} + 0.455{}^0G_{In}^{Tetragonal}$	[34]
	${}^0G_{Cu,Cu,In:In}^{Cu_6Sn_5-h} = -14526.5 + 18.020T$	[31]
	${}^0G_{Cu:Ag,Cu:In}^{Cu_6Sn_5-h} = -1000; {}^0G_{Ag,Cu,Cu:In}^{Cu_6Sn_5-h} = -2000; {}^0G_{Ag,Cu:In:In}^{Cu_6Sn_5-h} = 2000$	[34]
	${}^0G_{Cu:In}^{Cu_{11}In_2Sn} = -5250 + 0.77{}^0G_{Cu}^{Fcc} + 0.23{}^0G_{In}^{Tetragonal}$	[34]
τ_1 -Cu ₁₁ In ₂ Sn	${}^0G_{Cu:Sn}^{Cu_{11}In_2Sn} = -7000 + T + 0.77{}^0G_{Cu}^{Fcc} + 0.23{}^0G_{Sn}^{Bct}$	[34]
	${}^0G_{Cu:In,Sn}^{Cu_{11}In_2Sn} = -4800$	[34]
τ_2 -Cu ₂ In ₃ Sn	${}^0G_{Cu:In,Sn}^{Cu_2In_3Sn} = -9235 + 5T + 0.333{}^0G_{Cu}^{Fcc} + 0.5{}^0G_{In}^{Tetragonal} + 0.167{}^0G_{Sn}^{Bct}$	[34]