



Article The Interaction of Seasons and Biogeochemical Properties of Water Regulate the Air–Water CO₂ Exchanges in Two Major Tropical Estuaries, Bay of Bengal (India)

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Abstract: The exchange of CO_2 between the air–water interfaces of estuaries is crucial from the perspective of the global carbon cycle and climate change feedback. In this regard, we evaluated the air-water CO2 exchanges in two major estuaries-the Mahanadi estuary (ME) and the Dhamra estuary (DE) in the northern part of the Bay of Bengal, India. Biogeochemical properties of these estuarine waters were quantified in three distinct seasons, namely, pre-monsoon (March to May), monsoon (June to October), and post-monsoon (November to February). The significant properties of water, such as the water temperature, pH, salinity, nutrients, dissolved oxygen, chlorophyll-a (chl a), and photosynthetic pigment fluorescence of phytoplankton, were estimated and correlated with CO_2 fluxes. We found that the ME acted as a source of CO_2 fluxes in the monsoon and postmonsoon, while DE acted as a sink during the monsoon. The stepwise regression model showed that the fluxes were primarily driven by water temperature, pH, and salinity, and they correlated well with the phytoplankton characteristics. The chl a content, fluorescence yield, and phycobilisomesto-photosystem II fluorescence ratios were major drivers of the fluxes. Therefore, for predicting air–water CO_2 exchanges precisely in a large area over a seasonal and annual scale in the estuaries of the Bay of Bengal, India, critical key parameters such as water temperature, pH, salinity, chl a, and fluorescence yield of phytoplankton should be taken into consideration. However, the responses of phytoplankton, both in terms of production and CO₂ capture, are critical research areas for a better understanding of air-water CO₂ exchanges in coastal ecology under climate change scenarios.

Keywords: estuary; air–water CO₂ flux; physicochemical parameters; biological parameters; chlorophyll fluorescence; Bay of Bengal

1. Introduction

A significant component of the global carbon cycle is the supersaturation of carbon dioxide (CO₂) in estuaries and the resulting efflux to the atmosphere. In biogeochemically active zones such as estuaries, dissolved and particulate materials undergo rapid changes. Estuaries are primarily net heterotrophic, contributing significantly to CO₂ emissions [1–3]. On the other hand, some estuaries occasionally exhibit net autotrophy and act as CO₂ sinks [4,5]. The physical forcing, the balance between carbon fixation and respiration, and the lateral inputs of dissolved inorganic carbon (DIC), together with heterotrophy and autotrophy, are the major factors that influence the direction and magnitude of estuarine air–water CO₂ exchanges [6,7]. The generation of CO₂ in the estuarine water column and the pattern of several biogeochemical processes depend on several variables related to each estuary's geophysical characteristics [8]. The geomorphology, structure, and function of



Citation: Pattanaik, S.; Mohapatra, P.K.; Mohapatra, D.; Swain, S.; Panda, C.R.; Dash, P.K. The Interaction of Seasons and Biogeochemical Properties of Water Regulate the Air–Water CO₂ Exchanges in Two Major Tropical Estuaries, Bay of Bengal (India). *Life* **2022**, *12*, 1536. https://doi.org/10.3390/ life12101536

Academic Editors: Anirban Akhand, Abhra Chanda and Ken Takai

Received: 25 August 2022 Accepted: 30 September 2022 Published: 2 October 2022

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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). estuaries worldwide have seen a significant transformation over the past century as a result of anthropogenic interventions as well as climate change.

Globally, estuaries account for 0.27 ± 0.23 Pg C yr⁻¹, approximately 17% and 37% of the total CO₂ uptake by the open ocean and continental shelves, respectively [9]. However, a significant uncertainty exists in flux estimation worldwide due to the exclusion of several small-scale estuaries that have been scarcely studied. Therefore, all small and big estuaries should be considered and studied on a spatial and temporal scale to provide a precise flux assessment and develop our understanding of the estuarine CO₂ budget. However, determining the spatial and temporal fluctuations of surface pCO₂ of estuarine water and CO₂ efflux at the air–water interfaces is difficult due to the rapidly changing nature of coastal environments [10]. Many biological, physical, and physicochemical factors drive the CO₂ fluxes in a particular estuary. Therefore, studying the related physicochemical and biological factors to describe the CO₂ fluxes is crucial. The biogeochemical properties such as water temperature, pH, salinity, nutrients, and dissolved oxygen, etc., play a key role in the air–water CO₂ exchanges in the estuary systems of the world [11,12]. To draw firm conclusions on the role that estuaries play in the global carbon budget, a database of small and big estuaries across the globe is necessary.

The Mahanadi estuary (ME) and the Dhamra estuary (DE) are two of the most prominent estuaries on the east coast of India. It is reported that physiochemical processes play a vital role in carbon dynamics in the Mahanadi estuary (ME) and the Dhamra estuary (DE), which regulates CO_2 fluxes [13,14]. However, the impact of biological variables and photosynthetic efficiency on CO_2 dynamics, and the inter-relationship between seasonality and biogeochemical property in the ecology, need to be studied. Biological variables could affect CO_2 exchanges significantly in estuaries [15,16]. Among the biological properties, the fluorescence parameters such as fluorescence minimum (F₀), fluorescence maximum (F_M), variable fluorescence, and photosynthetic yield, in particular, have been found to drive the CO_2 fixation rates in the different coastal and estuarine systems [17–19]. Therefore, the specific objective of this study is to assess the impact of seasons and the significant biogeochemical properties of water on air–water CO_2 exchanges, and the relative importance of the key properties on CO_2 exchanges in the ME and DE of the Bay of Bengal.

2. Materials and Methods

2.1. Study Area

The research covers two estuaries on the east coast of India: (1) the ME and (2) the DE, the confluence of rivers Brahmani and Baitarani. The ME and DE are two distinctive ecosystems because they are subjected to extreme environmental conditions and, therefore, are very susceptible to climate change and other anthropogenic interferences. The Mahanadi River is the third largest river in India and the largest river in Odisha, with an annual discharge of 66,640 mm³. The basin is marked by a tropical climate, with a mean annual rainfall of 1572 mm, of which 90% occurs during the southwest monsoon. The river starts from the Baster hills of Madhya Pradesh, flowing over different regions of the Eastern Ghats and adjacent areas, and merges into the Bay of Bengal. The main branches of Mahanadi join the Bay of Bengal at Paradip (Mahanadi estuary) and Nuagarh (Devi estuary). ME is a microtidal estuary located between $19^{\circ}40'-20^{\circ}35'$ N and $85^{\circ}40'-86^{\circ}45'$ E that receives effluents from fertilizer plants, various other industries, agricultural lands, and domestic sectors from different cities along its bank [20]. The source of pollution also includes the port activity, wastes from the Paradeep port city, and the effluents from Paradeep Phosphates Ltd. The river is a major water supply source for irrigating 13,590 km² of agricultural land in the basin [21]. Out of many ecological features, the most important is the mangrove patch that extends from the southeastern part of Mahanadi river to the estuary point of Hansua (a tributary of Brahmani) in the north and from the northeastern end of Mahanadi river up to Jamboo river in the east.

The tropical DE ($20^{\circ}35'$ to $20^{\circ}50'$ N and $86^{\circ}46'$ to $87^{\circ}05'$ E) receives an annual rainfall of about 1670 mm, mainly during August and September (monsoon season). The DE receives

water from two major rivers, viz., Brahmani and Baitarani. The pollutant source of DE is primarily contributed by coal mining activities in the Talcher area, chromite in Sukinda, and iron/manganese in Keonjhar district, along with contributions from industrial clusters of Rourkela, Dhenkanal, and Jajpur districts of Odisha, India. During the rainy season, DE receives toxic pollutants, mostly heavy metals, from mining areas and mineral-rich catchment areas. Current activities of Dhamra port, such as dredging, also impact the estuary's water and sediment quality.

This selected study area has great significance due to the presence of India's second largest mangrove area of Bhitarkanika wildlife sanctuary, Gahirmatha marine sanctuary (nesting ground of olive ridley turtle), and Dhamra Port [22–24]. Considering its social and ecological importance, the Bhitarkanika area has been identified as a Ramsar site [25]. The estuary area is characterized by the presence of small scattered rivulets, creeks, and channels continuously influenced by tides. Both estuaries are regularly affected by tropical cyclones in the Bay of Bengal.

Nine fixed stations from the riverside (Stations 1–3 as upper reach, 4–6 as middle reach, and 7–9 as lower reach) towards the estuary point, maintaining a distance of 1 km between two stations, were selected as sampling points for the collection of water samples, which have been mentioned as estuary stations in both the Mahanadi and Dhamra mouths (Figure 1).



Figure 1. Sketch map of two important estuaries of northeastern Bay of Bengal showing three distinct reaches of Mahanadi estuary and Dhamra estuary.

2.2. Sampling Strategy

Surface water samples (100 cm from the surface) were collected in duplicates from each station of the ME and DE with the help of a Niskin sampler covering three seasons, namely, pre-monsoon (March to May), monsoon (June to October), and post-monsoon (November to February). For water sampling, pre-cleaned glass bottles were used [for dissolved oxygen (DO)] and polythene bottles (for nutrients, chl *a*, and pigment fluorescence). Sampled water was preserved and analyzed following standard procedures [26].

2.3. Biogeochemical Properties

2.3.1. Physicochemical Properties

Water temperature (WT), pH, and salinity were measured immediately after collecting the sample on board by a multi kit (WTW Multi 340 i Set, Germany) WTW Tetracon 325 probe fitted for the analysis of salinity and WT, and WTW Sentix 41-3 probe fitted for pH. The glass electrode for pH measurements was calibrated before sampling in the NBS scale with technical buffers of pH 4.01 (Model TEP Trace at a controlled temperature of 25 °C). Dissolved oxygen content was measured by Winkler's titrimetric method (accuracy of $\pm 0.1\%$). Dissolved inorganic nutrients such as nitrate (accuracy of $\pm 2.9\%$), nitrite, ammonia, and phosphate (accuracy of $\pm 2.5\%$) were determined by spectrophotometric methods using a bio UV-visible spectrophotometer (Varian 50) [27]. Total alkalinity (TA) was measured potentiometrically using Metrohm automatic titrator inbuilt with Tiamo software (Metrohm 905 Titrando).

2.3.2. Biological Properties

Chl *a* in water samples was estimated by filtering 1000 mL of a water sample using Whatman GF/F (47 mm diameter) under vacuum pressure [28]. The filter was mixed with acetone (10 mL; Merck) and incubated for 24 h at -4 °C; after centrifuging for 15 min at 5000 rpm, the absorbance of the supernatant was measured by spectrophotometer (Varian 50 bio UV-visible). The OJIP fluorescence transient was measured after concentrating the samples and providing a saturating pulse of red light (3000 μ E/m²s) by a Handy photosynthetic efficiency analyzer (Handy PEA, Hansatech Instruments, UK). For this, 10 mL of water sample was centrifuged, and the pellet with 2 mL of supernatant was put in a 2 mL glass vial with an aluminum screw cap that was kept in the dark at room temperature for 15 min for complete relaxation of PS II reaction centers. Sample vials were put into the liquid attachment of Handy PEA. Ground level fluorescence minimum (F₀), maximum fluorescence (F_M), and the variable fluorescence (F_V = F_M - F₀) were recorded after applying a saturating light pulse (3000 μ E/m²s, duration 1 s) [29]. The photosynthetic pigment fluorescence was also measured spectrofluorimetrically using a Shimadzu spectrofluorimeter (Model RF-5301PC, Shimadzu, Japan). Ten mL of the sample was centrifuged, and the pellet with 2 mL of homogenized supernatant was poured into a quartz cuvette (3 mL). The latter was placed in the cuvette holder of the equipment, excited with a 440 nm monochromatic light beam, and the fluorescence emission spectrum was measured from 600 nm to 750 nm. The peak fluorescence at 685 nm was recorded as an emission from PS II. Similarly, the sample was excited with a 570 nm monochromatic beam, and the emission spectrum was measured from 600 to 750 nm. The peak fluorescence at 655 nm was recorded as an emission from phycobilisome (PBS). In all cases, the excitation and emission bandpasses were 10 nm each to optimize the fluorescence signal.

2.3.3. Air–Water CO₂ Flux

The data of observed salinity, temperature, pH, nutrients (phosphate and silicate), and TA were used to calculate the CO₂ fugacity of water [fCO₂ (water)] and dissolved inorganic carbon (DIC) using CO2SYS.EXE software [30]. The dissociation constants K1 and K2 were used according to Peng et al. (1987) [31]. The concentration of CO₂ in the immediate surrounding atmosphere, 10 m above the water surface, was measured by a non-dispersive infrared gas analyzer (Li-840A CO₂/H₂O gas analyzer, Li-COR Inc., Lincoln, NE, USA). Both wind speed and ambient CO₂ concentration were measured above 10 m of the water surface by mounting a portable air pump (for CO₂ concentration) and an anemometer (for wind speed) with the help of a 7 m-long pole. The measured CO₂ in ppm was converted to pCO₂ (air) using the equation of state [32].

Air–water CO₂ flux densities (μ mol/m².h), or FCO₂, were estimated by the following equation:

$$FCO_2 = k \cdot \beta \cdot \Delta fCO_2 \tag{1}$$

where 'k' is the gas transfer velocity (cm h⁻¹), ' β ' is the Ostwald dilution coefficient (mol m⁻³ atm), and ΔfCO_2 is the difference in fugacity of CO₂ between water and air, [fCO_2 (water) – fCO_2 (air)]. k was calculated according to the equation of Wanninkhof, (1992) [33] as given below:

$$k = 0.17u_{10} \left(\frac{660}{\text{Sc}}\right)^{2/3}$$
 for $u_{10} \le 3.6 \text{ m/s}$ (2)

$$k = (2.85u_{10} - 9.65) \left(\frac{660}{Sc}\right)^{0.5} \text{ for } 3.6 \le u10 \le 13 \text{ m/s}$$
(3)

where u_{10} is the wind speed (m/s), and the Schmidt number (*Sc*) for CO₂ was evaluated as per the formula:

$$Sc = A - Bt + Ct^2 - Dt^3 \tag{4}$$

where A = 1992.1, B = 121.86, C = 3.54, D = 0.04227, and t = Temperature (°C) of water [34]. Ostwald solubility coefficient was calculated by the equation:

$$\ln \beta = A_1 + A_2(100/T) + A_3 \ln(T/100) + S[B_1 + B_2(T/100) + B_3(T/100)^2]$$

where β = Ostwald solubility coefficient, ln = natural logarithm, i.e., log_e, and the T = water surface temperature (in degree Kelvin), and S = salinity in ppt.

 $A_1 = -60.2409$, $A_2 = 93.4517$, $A_3 = 23.3585$, $B_1 = 0.023517$, $B_2 = -0.023656$, $B_3 = 0.004703$. The positive magnitude of FCO₂ interprets flux from water to air, and the negative value represents the reverse flux.

2.4. Statistical Analysis

The individual, as well as cumulative, datasets of biogeochemical properties were subjected to perform correlation and stepwise regression analysis, taking CO₂ flux as the dependent variable and other associated parameters (WT, pH, salinity, chl *a*, F₀, F_M, PSII fluorescence, PBS fluorescence, PSII: PBS, yield, DO, and nutrients) as independent variables using SAAS 9.2 online version. Pearson correlation analysis of different physicochemical and biological parameters was carried out using SPSS Version 20.

3. Results

3.1. Biogeochemical Properties of Estuaries

3.1.1. Biological Properties of Water in Estuaries

Our studies showed that the surface chl *a* ranged between 1.7 and 8.84, and 2.4 and 10.27 mg m⁻³ in the ME and DE estuaries, respectively (Table 1). The upper reach was found to have relatively higher chl *a* than the lower reach region and was strongly associated with high nutrient concentrations. The F_0 and F_M showed site-wise, stationwise, and season-wise variations such as that of chlorophyll. F_0 and F_M values were the highest in the post-monsoon period in the ME and DE. The relationship between chl *a* and fluorescence values (F_0 and F_M) was significantly linear (r = 0.983 and r = 0.987). The yield was also proportionately high in the post-monsoon period compared to other seasons in both estuaries. However, the yield values ranged from 0.32 to 0.49, which is relatively low. Nevertheless, yield variation was also significantly correlated with chl a (r = 0.962). The PBS fluorescence measured at 570 nm excitation was the highest in the post-monsoon period in both the estuaries. A comparison of pre-monsoon data of both the estuaries showed that the PBS fluorescence of the DE water was relatively higher than the ME water during this period. While a greater seasonal variation of PBS fluorescence was seen in the ME, it was proportionately less in the DE, indicating that the DE experienced a standing population of cyanobacteria throughout the year. Still, cyanobacterial blooming peaked during the post-monsoon months. For example, compared to the pre-monsoon, the increase in PBS fluorescence during the monsoon was higher in the ME (50%) than in the DE (6%). The

			Mahanadi		Dhamra						
Parameters		Pre- Monsoon	Monsoon	Post- Monsoon	Pre- Monsoon	Monsoon	Post- Monsoon				
-3	Mean	3.41 ± 1.36	3.16 ± 0.77	6.40 ± 1.23	6.83 ± 2.13	4.23 ± 1.69	7.15 ± 2.70				
$\operatorname{Cm} u (\operatorname{mg} \operatorname{m}^{-1}) =$	Range	1.71–6.28	2.26-4.68	4.51-8.84	3.68–9.20	2.40-6.92	2.46-10.27				
Fluorescence _ minimum	Mean	44.73 ± 8.01	41.10 ± 7.36	67.86 ± 9.18	69.21 ± 15.64	50.66 ± 13.93	72.41 ± 18.95				
	Range	34.27–59.48	34.24–56.34	52.45-82.67	45.78-87.95	34.53-71.69	37.23–92.45				
Fluorescence	Mean	70.88 ± 16.5	64.86 ± 15.12	119.34 ± 20.07	125.8 ± 34.57	84.82 ± 29.91	134.08 ± 45.68				
	Range	51.78–103.86	50.35-97.44	88.67-153.75	73.95–171.83	51.68-130.79	53.24–184.62				
Viold	Mean	0.36 ± 0.03	0.36 ± 0.03	0.43 ± 0.02	0.44 ± 0.03	0.39 ± 0.05	0.44 ± 0.07				
Tiela	Range	0.32-0.43	0.32-0.42	0.40-0.46	0.38-0.49	0.33-0.45	0.30-0.51				
Phycobilisome	Mean	17.44 ± 6.93	26.42 ± 6.25	39.27 ± 7.83	35.7 ± 10.94	37.37 ± 14.28	46.07 ± 17.42				
fluorescence	Range	8.72–32.17	18.34–38.94	27.61–54.77	19.32-48.21	22.35-60.54	16.04–66.27				
PSII	Mean	3.45 ± 1.38	3.47 ± 0.79	6.60 ± 1.28	6.49 ± 1.98	3.94 ± 1.49	6.98 ± 2.52				
fluorescence	Range	1.71–6.42	2.53-4.98	4.62–9.17	3.53-8.65	2.43-6.31	2.54-9.94				
PBS/PSII -	Mean	5.07 ± 0.11	7.61 ± 0.25	5.94 ± 0.08	5.50 ± 0.10	9.51 ± 0.30	6.55 ± 0.21				
	Range	4.89–5.20	7.24–7.96	5.74-6.03	5.34-5.64	8.88–9.83	6.31–6.97				

same trend was also observed during the post-monsoon period (the increase was 50% and 26% in the ME and DE, respectively).

Table 1. Seasonal mean \pm standard deviation (SD) along with a range of essential biological parameters in the Mahanadi and Dhamra estuaries.

Unlike PBS, the PSII fluorescence in the DE during the monsoon decreased compared to the pre-monsoon season, whereas the same was more or less unchanged in the ME. During the monsoon to post-monsoon months, a significant rise in PS II fluorescence occurred in both the estuaries, though the rising rate was higher in the ME (90%) than in the DE (78%). The corresponding high value of the PBS/PSII ratio during the monsoon months, the water turbidity limited the growth of green algae. Still, cyanobacterial growth was favored over green algae due to their low light adaptation. Irrespective of the season, the PBS/PS II ratio was higher in the DE than in the ME.

3.1.2. Physicochemical Properties of Water in Estuaries

The variation of WT was not significant between the estuaries. However, seasonal variation at both the estuaries was prominent. Low temperatures were observed during the post-monsoon at both the estuaries (Figure 2). However, the DE showed more decline in WT (23.1 \pm 0.1 °C) in the post-monsoon period as compared to the ME (28.8 \pm 0.01 °C). Surface water pH attained the highest average value in the pre-monsoon in both the ME (8.197 ± 0.015) and DE (8.059 ± 0.038) . The salinity of Mahanadi and Dhamra was primarily regulated by river discharges and was lowest during the monsoon at both the estuaries (4.2 \pm 0.6 PSU in ME and 10.8 \pm 3.5 PSU in DE). The nitrate concentrations were between 1.977 and 15.653 and 6.454 and 19.134 μ mol L⁻¹ in the ME and DE, respectively, whereas phosphate varied from 0.324 to 8.392 and 0.284 to 1.889 μ mol L⁻¹, respectively (Table 2). On the other hand, nitrite concentrations ranged from 0.048 to 2.678 in the ME and below the detection limit to 1.650 μ mol L⁻¹ in the DE. Similarly, the variations in DO concentration were observed in a range between 4.0 and 6.9 mg L^{-1} in the ME and 4.0 to 7.0 mg L⁻¹ in the DE and were more or less consistent with chl *a* (Tables 3–5; $r \ge 0.4$; p = 0.05), suggesting biological processes have significant control over water biogeochemistry. The fCO_2 (water) ranged between 240.5 and 1268.0 µatm in the ME and 208.8 and 485.6 µatm in

the DE, whereas the variation of fCO_2 (air) was 378.2–432.7 in the ME and 378.8–408.7 µatm in the DE. In the ME, the highest TA and DIC were observed in the pre-monsoon (1873.4 ± 213.1 µmol kg⁻¹ and 1415.9 ± 49.0 µmol kg⁻¹, respectively). In the DE, the highest TA was observed in the pre-monsoon (1873.4 ± 213.1 µmol kg⁻¹), and the highest DIC was observed in the post-monsoon (1683.5 ± 28.8 µmol kg⁻¹).



Figure 2. Seasonal variability of temperature (*p* < 0.001) in Mahanadi and Dhamra estuaries.

Table 2. Seasonal mean \pm SD, along with a range of essential physicochemical parameters in the Mahanadi and Dhamra estuaries.

			Mahanadi		Dhamra					
Parameters		Pre-Monsoon	Monsoon	Post- Monsoon	Pre-Monsoon	Monsoon	Post- Monsoon			
CO ₂ flux	Mean	-6.5 ± 1.6	109.5 ± 31.5	28 ± 20.1	-4.7 ± 3.7	-14.2 ± 11.1	1.8 ± 2.2			
$(\mu mol m^{-2} h^{-1})$	Range	-9.23.5	63.8–154.1	-0.7 - 57.1	-9.2-0.3	-25.8-12.2	-0.7 - 5.6			
fCO ₂ (water) (µatm)	Mean	270.3 ± 17.4	759.4 ± 109.9	840.0 ± 294.2	300.2 ± 31.4	258.8 ± 38.5	432.2 ± 31.2			
	Range	240.5–289.6	600.3–910.9	417.6–1268.0	270.5–365.2	208.8-327.3	395.4-485.6			
fCO ₂ (air)	Mean	381.7 ± 2.724	379.7 ± 2.681	421.7 ± 7.397	382.2 ± 2.118	386.6 ± 3.9	406.3 ± 0.993			
(µatm)	Range	379.4–388.2	378.2–386.8	414.1-432.7	378.8–384.9	382.2–394.90	405.4-408.7			
TA	Mean	1564.0 ± 54.63	1240.8 ± 105.8	1284.4 ± 250.2	1873.4 ± 213.1	1022.0 ± 52.3	1831.9 ± 29.1			
$(\mu mol \ kg^{-1})$	Range	1476.9–1669.5	1123.8–1457.9	1012.4–1669.3	1513.0–2151	919.8–1079.7	1769.2–1876.2			
DIC	Mean	1415.9 ± 49.0	1234.6 ± 105.6	1270.1 ± 249.5	1606.4 ± 140.9	934.2 ± 46.4	1683.5 ± 28.8			
$(\mu mol \ kg^{-1})$	Range	1346.8–1502.2	1112.5–1448.9	998.1–1648.7	1329.3–1774.6	853.5–998.0	1622.3–1731.1			
pН	Mean	8.197 ± 0.015	7.847 ± 0.048	7.716 ± 0.153	8.059 ± 0.038	8.02 ± 0.088	7.974 ± 0.021			
	Range	8.180-8.230	7.750–7.911	7.541–7.962	8.011-8.101	7.840-8.101	7.940-8.001			

			Mahanadi		Dhamra					
Parameters		Pre-Monsoon	Monsoon	Post- Monsoon	Pre-Monsoon	Monsoon	Post- Monsoon			
Water	Mean	31.4 ± 0.21	30.9 ± 0.32	28.8 ± 0.01	30.5 ± 0.71	$\begin{array}{c} 30.8 \pm 0.62 \\ 30.3 31.7 \end{array}$	23.1 ± 0.13			
temperature (°C)	Range	30.9–31.6	30.5–31.4	28.8-28.9	29.6–31.0		22.9–23.3			
Salinity (psu)	Mean	8.8 ± 1.8	4.2 ± 0.6	6 ± 2.3	22.7 ± 2.1	10.8 ± 3.5	22.6 ± 0.5			
	Range	5.8–10.8	3.1–5.1	1.8–9.6	19.5–25.5	4.7 - 18.4	21.9–23.3			
D.O. (mg L ⁻¹)	Mean	5.9 ± 0.4	6.1 ± 0.1	5.2 ± 0.5	5.5 ± 0.2	4.6 ± 0.4	6.5 ± 0.2			
	Range	5.5–6.9	6.0–6.3	4.0–5.7	5.3–6.1	4.0–5.2	6.2–7.0			
Nitrite	Mean	1.562 ± 0.661	0.262 ± 0.035	0.100 ± 0.038	0.737 ± 0.572	0.309 ± 0.088	0.247 ± 0.25			
$(\mu mol L^{-1})$	Range	0.52–2.678	0.209–0.313	0.048-0.152	0.052-1.650	0.174-0.483	BDL-0.774			
Nitrate	Mean	7.839 ± 4.771	14.731 ± 0.752	4.259 ± 2.579	9.028 ± 1.387	8.114 ± 1.81	16.391 ± 1.621			
$(\mu mol L^{-1})$	Range	2.639–15.038	13.542–15.653	1.977-10.728	7.358–10.915	6.454-12.202	14.235–19.134			
Ammonia	Mean	0.055 ± 0.108	1.217 ± 0.36	1.351 ± 1.658	7.976 ± 11.246	1.127 ± 0.973	5.369 ± 6.809			
$(\mu mol L^{-1})$	Range	BDL-0.338	0.832-1.756	BDL-5.273	BDL-34.315	0.211-3.464	0.658–20.516			
Phosphate	Mean	2.554 ± 1.198	5.819 ± 1.423	0.967 ± 0.504	0.467 ± 0.124	1.485 ± 0.194	0.363 ± 0.203			
$(\mu mol L^{-1})$	Range	1.348-5.037	4.250-8.392	0.324-1.758	0.284-0.659	1.289–1.889	0.137-0.811			

Table 2. Cont.

3.2. Air–Water CO₂ Fluxes in Estuaries and Their Drivers

The regression models built from the whole dataset showed that the overall air–water CO_2 fluxes (FCO₂) were not significantly dependent on any biological parameter. However, the CO₂ flux of the ME during the monsoon was significantly dependent on pH (Table 6). In contrast, in the DE, the CO₂ flux depended on pH and salinity. The FCO₂ in the ME, irrespective of the seasons, was significantly driven by biological parameters such as chl *a*, yield, and PBS: PSII, whereas only PBS: PSII significantly affects FCO₂ in the DE among the biological parameters. The estuarine effect on biological parameters is more prominent, but the physicochemical parameters such as WT, pH, and salinity were the governing parameters in both the estuaries.

3.3. Correlation among Physicochemical and Biological Parameters of Estuaries

The PS II fluorescence intensity was directly correlated with the chl *a* content of the water sample ($r \ge 0.99$). However, in the monsoon season, it was found less due to the water turbidity and greater PBS fluorescence. There was an inter-seasonal variation of PBS fluorescence at both the study sites, and such variations were found to be significant. While in the ME, no significant improvement in the PSII fluorescence was observed in the monsoon compared to the pre-monsoon, the corresponding PBS fluorescence increment was relatively high. However, increased phytoplankton density in the post-monsoon, presumably driven by nutrient availability, led to higher PBS and PSII fluorescence compared to the monsoon season. On the other hand, there was a significant reduction of the PSII fluorescence, and no significant change in the PBS fluorescence was noted in the monsoon than in the pre-monsoon of the DE. However, the trend from monsoon to post-monsoon in the DE was similar to the ME. The fluorescence parameters were found to be significantly correlated with water temperature, thus showing that the phytoplankton density/diversity of the estuaries was well-influenced by water temperature (r = -0.40 to -0.46) but independent of the pH changes of the surface water (r = -0.002 to -0.114).

	FCO ₂ (Flux Density)	РН	Water Tem- perature	Salinity	Chl a	F ₀	F _m	Yield	PBS	PS II	PBS/PSII	DO	NO_2^-	NO ₃ -	NH ₃	PO ₄ ³⁻
FCO ₂ (Flux density)	1															
PH	-0.597 **	1														
Water tem- perature	0.152	0.186	1													
salinity	-0.504 **	0.313 *	-0.535 **	1												
Chl a	-0.327*	-0.016	-0.448 **	0.438 **	1											
F ₀	-0.359 **	-0.017	-0.459 **	0.441 **	0.990 **	1										
F _m	-0.351 **	-0.002	-0.468 **	0.454 **	0.994 **	0.996 **	1									
Yield	-0.331 *	-0.024	-0.401 **	0.432 **	0.962 **	0.9 57**	0.965 **	1								
PBS	-0.222	-0.114	-0.456 **	0.289 *	0.902 **	0.896 **	0.906 **	0.890 **	1							
PS II	-0.262	-0.076	-0.457 **	0.381 **	0.994 **	0.984 **	0.985 **	0.949 **	0.892 **	1						
PBS/PSII	0.133	-0.160	0.077	-0.228	-0.206	-0.216	-0.198	-0.159	0.196	-0.238	1					
DO	0.265	0.056	-0.452 **	0.271 *	0.043	0.029	0.049	0.013	-0.077	0.072	-0.394 **	1				
NO ₂ -	-0.136	0.217	0.114	0.207	-0.087	-0.109	-0.096	0.008	-0.196	-0.105	-0.229	-0.017	1			
NO ₃ ⁻	0.339 *	0.008	-0.443 **	0.292 *	-0.016	-0.057	-0.022	-0.037	0.074	-0.011	0.156	0.628 **	-0.100	1		
NH ₃	-0.123	0.086	-0.109	0.395 **	0.360 **	0.361 **	0.365 **	0.334 *	0.244	0.328 *	-0.122	0.070	-0.102	0.036	1	
PO4 ³⁻	0.718 **	-0.104	0.432 **	-0.612 **	-0.571 **	-0.587 **	-0.570 **	-0.566 **	-0.428 **	-0.534 **	0.219	0.157	-0.096	0.257	-0.218	1

Table 3. Pearson correlation analysis of different physicochemical and biological parameters using the cumulative dataset of Mahanadi and Dhamra estuaries.

* Correlation is significant at the 0.05 level. ** Correlation is significant at the 0.01 level.

	FCO ₂ (Flux Density)	РН	Water Tem- perature	Salinity	Chl a	F ₀	F _m	Yield	PBS	PS II	PBS/PSII	DO	NO ₂ -	NO ₃ -	NH ₃	PO4 ³⁻
FCO ₂ (Flux density)	1															
PH	-0.716 **	1														
Water temperature	-0.601 **	0.479 *	1													
salinity	0.532 **	-0.001	-0.485 *	1												
Chl a	0.105	0.283	-0.288	0.435 *	1											
F ₀	0.089	0.286	-0.302	0.436 *	0.993 **	1										
F _m	0.098	0.273	-0.309	0.427 *	0.996 **	0.997 **	1									
Yield	0.070	0.297	-0.223	0.407 *	0.954 **	0.951 **	0.960 **	1								
PBS	-0.149	0.223	-0.283	0.063	0.867 **	0.871 **	0.881 **	0.855 **	1							
PS II	0.145	0.247	-0.334	0.473 *	0.997 **	0.991 **	0.994 **	0.945 **	0.853 **	1						
PBS/PSII	-0.544 **	-0.099	0.288	-0.918 **	-0.462 *	-0.457*	-0.442*	-0.411 *	0.000	-0.492 **	1					
DO	0.608 **	-0.250	-0.829 **	0.781 **	0.460*	0.464 *	0.464 *	0.378	0.240	0.509 **	-0.644 **	1				
NO ₂ -	0.117	0.027	0.116	0.221	-0.091	-0.121	-0.106	0.036	-0.210	-0.096	-0.226	-0.056	1			
NO ₃ ⁻	0.625 **	-0.541 **	-0.903 **	0.502 **	0.299	0.319	0.315	0.213	0.251	0.351	-0.343	0.841 **	-0.193	1		
NH ₃	0.077	0.245	0.032	0.298	0.324	0.331	0.320	0.284	0.127	0.308	-0.329	0.148	-0.121	-0.073	1	
PO4 ³⁻	-0.589 **	-0.005	0.580 **	-0.886 **	-0.505 **	-0.499 **	-0.498 **	-0.450 *	-0.156	-0.536 **	0.884 **	-0.796 **	-0.198	-0.512 **	-0.251	l 1

 Table 4. Pearson correlation analysis of different physicochemical and biological parameters using the dataset of Mahanadi estuary.

* Correlation is significant at the 0.05 level. ** Correlation is significant at the 0.01 level.

	FCO ₂ (Flux Density)	РН	Water Tem- perature	Salinity	Chl a	F ₀	F _m	Yield	PBS	PS II	PBS/PSII	DO	NO_2^-	NO ₃ -	NH ₃	PO4 ³⁻
FCO ₂ (Flux density)	1															
PH	-0.543 **	1														
Water temperature	0.066	0.734 **	1													
salinity	-0.577 **	0.390*	0.197	1												
Chl a	-0.294	-0.353	-0.740 **	-0.318	1											
F ₀	-0.354	-0.338	-0.760 **	-0.232	0.983 **	1										
Fm	-0.343	-0.340	-0.762 **	-0.263	0.987 **	0.998 **	1									
Yield	-0.285	-0.373	-0.702 **	-0.300	0.957 **	0.950 **	0.958 **	1								
PBS	0.053	-0.545 **	-0.733 **	-0.587 **	0.919 **	0.891 **	0.903 **	0.898 **	1							
PS II	-0.241	-0.384^{*}	-0.742 **	-0.366	0.998 **	0.976 **	0.981 **	0.952 **	0.940 **	1						
PBS/PSII	0.907 **	-0.506 **	0.005	-0.721 **	-0.183	-0.235	-0.211	-0.140	0.208	-0.128	1					
DO	0.290	0.280	0.683 **	0.021	-0.650 **	-0.669 **	-0.666 **	-0.569 **	-0.563 **	-0.644 **	0.250	1				
NO ₂ -	-0.526 **	0.801 **	0.624 **	0.498 **	-0.344	-0.349	-0.351	-0.321	-0.582 **	-0.378	-0.611 **	0.255	1			
NO ₃ ⁻	0.646 **	0.036	0.528 **	-0.326	-0.545 **	-0.632 **	-0.606 **	-0.502 **	-0.302	-0.510 **	0.625 **	0.574 **	-0.040	1		
NH ₃	0.328	-0.523 **	-0.375	-0.319	0.219	0.241	0.250	0.248	0.361	0.244	0.366	-0.135	0.470 *	0.038	1	
PO4 ³⁻	0.637 **	0.086	0.505 **	-0.317	-0.651 **	-0.673 **	-0.652 **	-0.643 **	-0.369	-0.614 **	0.694 **	0.490 **	-0.085	0.727 **	0.058	1

Table 5. Pearson correlation analysis of different physicochemical and biological parameters using the dataset of Dhamra estuary.

* Correlation is significant at the 0.05 level. ** Correlation is significant at the 0.01 level.

	Regression Equation	$ m r^2$ ($p \le$ 0.05)
	Mahanadi	
Pre-Monsoon	$FCO_2 = -235.2 + 30.5 \text{ pH} - 1.4 \text{ salinity} - 0.12 \text{ Fm} + 5.3 \text{NH}_4$	$r^2 = 0.94$
Monsoon	FCO ₂ = 4436.2 - 552.4 pH	$r^2 = 0.69$
Post-Monsoon	FCO ₂ = 9397.4 - 171.1 pH - 279.8 Water temperature +179.1 NO ₂	$r^2 = 0.92$
	Dhamra	
Pre-Monsoon	$FCO_2 = 160.3 - 5.41$ temperature	$r^2 = 0.95$
Monsoon	FCO ₂ = 867.7 – 108.7 pH – 0.92 salinity	$r^2 = 0.93$
Post-Monsoon	$FCO_2 = 363.7 - 58.6 \text{ pH} + 5.4 \text{ Water temperature } 7 - 0.91 \text{ salinity}$	$r^2 = 0.98$
Mahanadi Annual	FCO ₂ = 620.4 – 160.5 pH +24.9 temperature +17.5 Chl <i>a</i> – 836.8 yield + 29.3 PBS:PSII	$r^2 = 0.94$
Dhamra Annual	$FCO_2 = 965.9 - 116.1 \text{ pH} - 0.39 \text{ salinity} - 4.6 \text{ PBS:PSII}$	$r^2 = 0.90$
Mahanadi and Dhamra Annual	FCO ₂ = 1079.8 – 162.6 pH + 4.0 temperature + 13.8 DO +1.9 nitrate +10.1 phosphate	$r^2 = 0.86$

Table 6. Regression models of FCO₂ with related physicochemical and biological parameters.

4. Discussion

4.1. Biological Bloom and Behavior in Estuaries

To the best of our knowledge, this study reports for the first time the efficiency of Chl a fluorescence as a marker for predicting the air–water CO_2 flux in estuaries. The photosynthetic activity of phytoplankton in the coastal ocean plays a critical role in maintaining the gradient of CO_2 concentration between the surface layer of the ocean and its overlying atmosphere, mainly regulating the air–water CO₂ exchange and its spatial distribution. Photosynthesis in the coastal ocean reduces the CO₂ concentration by converting dissolved inorganic carbon into an organic form of carbon that decomposes and remineralizes back to dissolved organic carbon at ocean depth. This biological cycle is responsible for the net movement of carbon from the surface layer to the subsurface and subsequently to the ocean depth, referred to as the "biological pump". This study found that the biological pump was very active during the post-monsoon in both the estuaries. The chl *a* and the associated photosynthetic performance parameters were high in the post-monsoon compared to other seasons. Irrespective of seasons, the biological parameters were higher in magnitude in the DE compared to the ME. While PS II fluorescence was not much influenced by the change of season from the pre-monsoon to monsoon in the ME, a significant decrease of fluorescence intensity was observed in the DE during the period indicating that monsooninduced increase in turbidity of DE water limited the growth of green phytoplankton and, consequently, directed the resources for an increase in density of cyanobacteria as the latter are adapted to low-light intensities. The corresponding rise in PBS fluorescence corroborated this. Throughout the year, the photosynthetic yield of the DE was high compared to that of the ME, indicating a high planktonic producer load in the DE. No significant seasonal change in yield was observed in the DE, but in the ME, the yield significantly increased in the post-monsoon season as compared to the monsoon months, and it was attributed to the high growth of green phytoplankton. When the biological data of both the estuaries were pooled together, the correlation analysis showed that CO_2 flux was negatively correlated with the fluorescence parameters. Thus, it can be inferred that the growth and photosynthetic performance of phytoplankton, irrespective of the qualitative variation, have a significant impact on the CO₂ flux of the estuaries.

It has also been reported that more than 1 million tons of carbon are fixed into organic matter each day by ocean biological fixation [35]. Due to the biological control of the dissolved inorganic carbon in the ocean, the current average atmospheric CO_2 concentration of earth is ~410 ppm, which would be 150–200 µatm greater than the current value in the absence of ocean biological fixation. From various research, it is also clear that the coastal

bloom takes up a large portion of anthropogenic CO_2 , adding organic carbon to the coastal ocean and contributing to the inner ocean's primary production [16,36,37].

Chl *a* fluorescence has become an essential analytical parameter in oceanography and limnology to predict phytoplankton productivity and examine the diversity of producers in the aquatic systems [38,39]. Active chl *a* fluorescence measurement evaluates the efficiency by which absorbed light is utilized by photosynthesis. The fluorescence ratios have also become useful to determine the dominance of one group of algae or the other in a system. Because of the efficient transfer of energy to reaction center chlorophylls, the fluorescence emission from chl *a* usually acts as an indicator of the phytoplankton density vis-a-vis primary production [40]. Fluorescence spectra are useful for determining the phytoplankton composition and for measuring the dynamics of change in phytoplankton dominance. However, samples' excitation and emission characteristics vary with the qualitative and quantitative changes in phytoplankton composition [41]. Nevertheless, PS II and PBS's intensity are beneficial for predicting phytoplankton dominance and production potential [42]. The relationships between algal growth and environmental factors have been previously discussed in many Indian estuaries, but the use of the physiological parameters of algae about CO_2 sequestration is rarely discussed [43].

4.2. Temporal and Spatial Relationship of Physicochemical Properties of Estuaries

The Bay of Bengal receives a good amount of freshwater discharge during the monsoon season; therefore, a weakened coastal upwelling was observed. During the period of our observations, the DE was found to be more saline than ME (Figures 3 and 4) because the Mahanadi river has a greater number of tributaries, and it receives more fresh water in comparison to Dhamra (confluence of Brahmani and Baitarani) [44]. Seasonal variations in chl a and nutrients were primarily due to river discharge and other associated biogeochemical processes. The source of nutrients is usually associated with low salinity, and temperature signifies that the nutrient sources are river-driven. During monsoon, the chl a concentration was low despite the high amount of nutrients. This led to the conclusion that even though there were sufficient nutrients by land contribution during the monsoon season, the growth of chlorophyll-dominated phytoplankton was limited by suspended particulate matters, and cyanobacteria well exploited the available resources because of their low light adaptability. In the Arabian Sea, the monsoon-driven changes in water chemistry cause a shift in the dominance of phytoplankton, the cyanobacteria being the dominant group during monsoon months [45]. The strong seasonal variation of PBS/PS II ratio and its high values in the monsoon seasons in this study supports the conclusion that the cyanobacteria are abundant during the monsoon relative to green algae and diatoms in post-monsoon conditions. It is essential to mention that the increase in chl a and associated changes in nutrients, WT, and salinity can be useful to evaluate possible mechanisms regulating the CO_2 flux.

4.3. Drivers of Variations of Air–Water CO₂ Exchanges and Their Prediction

The detailed estimation of air–water CO_2 fluxes in our study revealed sharp seasonal changes in the Mahanadi and Dhamra estuaries (Figure 5). The biological parameters had significantly driven the fluxes in the ME. However, in the DE, the physicochemical variables such as WT, pH, and salinity regulated the flux behavior rather than the biological parameters. Chl *a* reveal a negative correlation with CO_2 flux, suggesting that phytoplankton growth has a vital role in controlling the capacity of the estuary to atmospheric uptake CO_2 . The stepwise regression model revealed that FCO_2 could be predicted precisely with the help of specific physicochemical and biological parameters. This could be a valuable tool for understanding the regulation mechanism of these parameters over CO_2 flux and vice-versa.



Figure 3. Seasonal variability of pH, salinity, chl *a*, and dissolved oxygen in Mahanadi estuary (p < 0.001).



Figure 4. Seasonal variability of pH (p < 0.05), salinity (p < 0.001), chl *a* (p < 0.05), and dissolved oxygen (p < 0.001) in Dhamra estuary.



Figure 5. Seasonal variability of CO_2 fluxes (p < 0.001) in Mahanadi and Dhamra estuaries.

5. Conclusions

The Mahanadi estuary, Bay of Bengal, India, acted as a source of CO_2 fluxes in the monsoon and post-monsoon. On the contrary, the Dhamra estuary acted as a sink of CO_2 fluxes during the monsoon and a weak source in the post-monsoon. There is not much variation in the sink strength in the pre-monsoon between the ME and DE. The upper reaches (near the sea) of the estuaries were characterized by more saline water, with more pH and sink strength. Air-water CO₂ flux and surface chl a concentration were negatively correlated, suggesting that photosynthesis is one of the primary processes responsible for the substantial CO_2 sequestration in estuaries and other coastal oceans. The biological CO_2 drawdown is site-specific and significant on a large area over a long temporal scale. Still, the seasonal cycle of CO_2 flux is regulated less by biological and more by physicochemical parameters such as WT, pH, and salinity. The algal composition and assemblage showed distinct characteristics due to variable nutrient concentration and water transparency. Our stepwise regression model suggested that biological parameters are more site-specific and sensitive to WT, pH, and salinity. Long-term analysis of biological parameters with realtime air–water CO₂ exchanges should be monitored over a spatial scale to judge the climate change feedback behavior of estuaries in the Bay of Bengal. However, the collected data in this study and other studies may help understand the future climate trend. Further, it may help the global community understand the flux behavior and associated dynamics when seasonal data from such marginally studied estuaries are considered for scale-up processes.

Author Contributions: Conceptualization, S.P.; methodology, S.P., S.S. and D.M.; software, P.K.D.; formal analysis, S.P., S.S. and D.M.; resources, P.K.M. and C.R.P.; writing—original draft preparation, S.P.; writing—review and editing, S.P. and P.K.M.; visualization, S.P. and S.S.; supervision, P.K.M. and C.R.P.; project administration, C.R.P.; funding acquisition, P.K.M. and C.R.P. All authors have read and agreed to the published version of the manuscript.

Funding: This research was funded by [ISRO-NRSC] grant number [GAP-242], [CSIR-HRDG] grant number [31/9(127)/2017-EMR-I], [DST-FIST] grant number [SR/FST/LS II/2018/219].

Institutional Review Board Statement: Not Applicable.

Informed Consent Statement: Not Applicable.

Data Availability Statement: The authors themselves have measured the primary data used in this study and can be provided by the corresponding author on reasonable request and the secondary data used in this study is duly cited all through the manuscript.

Acknowledgments: The authors are thankful to DST-FIST, Government of India, and ISRO-National Remote Sensing Centre, Government of India, for funding the research work. S.P. is grateful to the Council of Scientific and Industrial Research (CSIR) for awarding the CSIR-SRF fellowship.

Conflicts of Interest: The authors declare no conflict of interest.

References

- 1. Borges, A.V.; Delille, B.; Frankignoulle, M. Budgeting Sinks and Sources of CO₂ in the Coastal Ocean: Diversity of Ecosystems Counts. *Geophys. Res. Lett.* **2005**, *32*, 14. [CrossRef]
- Testa, J.M.; Kemp, W.M.; Hopkinson, C.S.; Smith, S.V. Ecosystem Metabolism. In *Estuarine Ecology*; John Wiley & Sons, Inc.: Hoboken, NJ, USA, 2012; pp. 381–416.
- Laruelle, G.G.; Dürr, H.H.; Lauerwald, R.; Hartmann, J.; Slomp, C.P.; Goossens, N.; Regnier, P.A.G. Global Multi-Scale Segmentation of Continental and Coastal Waters from the Watersheds to the Continental Margins. *Hydrol. Earth Syst. Sci.* 2013, 17, 2029–2051. [CrossRef]
- Eyre, B.D.; Ferguson, A.J.P.; Webb, A.; Maher, D.; Oakes, J.M. Metabolism of Different Benthic Habitats and Their Contribution to the Carbon Budget of a Shallow Oligotrophic Sub-Tropical Coastal System (Southern Moreton Bay, Australia). *Biogeochemistry* 2011, 102, 87–110. [CrossRef]
- 5. Maher, D.T.; Eyre, B.D. Carbon Budgets for Three Autotrophic Australian Estuaries: Implications for Global Estimates of the Coastal Air-Water CO₂ Flux. *Glob. Biogeochem. Cycles* **2012**, *26*, 1. [CrossRef]
- Cai, W.-J. Estuarine and Coastal Ocean Carbon Paradox: CO₂ Sinks or Sites of Terrestrial Carbon Incineration? *Ann. Rev. Mar. Sci.* 2011, 3, 123–145. [CrossRef]
- 7. Van Dam, B.R.; Crosswell, J.R.; Anderson, I.C.; Paerl, H.W. Watershed-Scale Drivers of Air-Water CO₂ Exchanges in Two Lagoonal North Carolina (USA) Estuaries. *J. Geophys. Res. Biogeosci.* **2018**, 123, 271–287. [CrossRef]
- Jiang, Z.-P.; Cai, W.-J.; Lehrter, J.; Chen, B.; Ouyang, Z.; Le, C.; Roberts, B.J.; Hussain, N.; Scaboo, M.K.; Zhang, J.; et al. Spring net community production and its coupling with the CO2 dynamics in the surface water of the northern Gulf of Mexico. *Biogeosciences* 2019, 16, 3507–3525. [CrossRef]
- 9. Laruelle, G.G.; Dürr, H.H.; Slomp, C.P.; Borges, A.V. Evaluation of Sinks and Sources of CO₂ in the Global Coastal Ocean Using a Spatially-Explicit Typology of Estuaries and Continental Shelves. *Geophys. Res. Lett.* **2010**, *37*, 15. [CrossRef]
- Zhai, W.; Dai, M. On the Seasonal Variation of Air—Sea CO2 Fluxes in the Outer Changjiang (Yangtze River) Estuary, East China Sea. Mar. Chem. 2009, 117, 2–10. [CrossRef]
- 11. Sarma, V.V.S.S.; Viswanadham, R.; Rao, G.D.; Prasad, V.R.; Kumar, B.S.K.; Naidu, S.A.; Kumar, N.A.; Rao, D.B.; Sridevi, T.; Krishna, M.S.; et al. Carbon Dioxide Emissions from Indian Monsoonal Estuaries. *Geophys. Res. Lett.* **2012**, *39*, 3. [CrossRef]
- Akhand, A.; Chanda, A.; Manna, S.; Das, S.; Hazra, S.; Roy, R.; Choudhury, S.B.; Rao, K.H.; Dadhwal, V.K.; Chakraborty, K.; et al. A Comparison of CO2 Dynamics and Air-Water Fluxes in a River-Dominated Estuary and a Mangrove-Dominated Marine Estuary. *Geophys. Res. Lett.* 2016, 43, 11726–11735. [CrossRef]
- Pattanaik, S.; Acharya, D.; Sahoo, R.K.; Satapathy, D.R.; Panda, C.R.; Choudhury, S.B.; Nagamani, P.V.; Roy, R. Short-Term Variability of Physico-Chemical Properties and PCO2 Fluxes off Dhamra Estuary from North-Eastern India. *J. Indian Soc. Remote Sens.* 2019, 47, 1197–1208. [CrossRef]
- Pattanaik, S.; Chanda, A.; Sahoo, R.K.; Swain, S.; Satapathy, D.R.; Panda, C.R.; Choudhury, S.B.; Mohapatra, P.K. Contrasting Intra-Annual Inorganic Carbon Dynamics and Air–Water CO2 Exchange in Dhamra and Mahanadi Estuaries of Northern Bay of Bengal, India. *Limnology* 2020, 21, 129–138. [CrossRef]
- Migné, A.; Gévaert, F.; Créach, A.; Spilmont, N.; Chevalier, E.; Davoult, D. Photosynthetic Activity of Intertidal Microphytobenthic Communities during Emersion: In Situ Measurements of Chlorophyll Fluorescence (PAM) and CO₂ Flux (IRGA)¹. *J. Phycol.* 2007, 43, 864–873. [CrossRef]
- 16. Bennington, V.; McKinley, G.A.; Dutkiewicz, S.; Ullman, D. What Does Chlorophyll Variability Tell Us about Export and Air-Sea CO₂ Flux Variability in the North Atlantic? *Glob. Biogeochem. Cycles* **2009**, *23*, 3. [CrossRef]
- 17. Serôdio, J.; Cruz, S.; Vieira, S.; Brotas, V. Non-Photochemical Quenching of Chlorophyll Fluorescence and Operation of the Xanthophyll Cycle in Estuarine Microphytobenthos. *J. Exp. Mar. Biol. Ecol.* **2005**, *326*, 157–169. [CrossRef]
- Kromkamp, J.C.; Morris, E.P.; Forster, R.M.; Honeywill, C.; Hagerthey, S.; Paterson, D.M. Relationship of Intertidal Surface Sediment Chlorophyll Concentration to Hyperspectral Reflectance and Chlorophyll Fluorescence. *Estuaries Coasts* 2006, 29, 183–196. [CrossRef]
- 19. Maricle, B.R.; Lee, R.W.; Hellquist, C.E.; Kiirats, O.; Edwards, G.E. Effects of salinity on chlorophyll fluorescence and CO2 fixation in C4 estuarine grasses. *Photosynthetica* 2007, *45*, 433–440. [CrossRef]

- Sundaray, S.K.; Nayak, B.B.; Lin, S.; Bhatta, D. Geochemical Speciation and Risk Assessment of Heavy Metals in the River Estuarine Sediments—A Case Study: Mahanadi Basin, India. J. Hazard Mater. 2011, 186, 1837–1846. [CrossRef]
- 21. CPCB. Basin Sub-Basin Inventory of Water Pollution: The Mahanadi Basin, ADSORBS/23/1999–2000; Central Pollution Control Board: New Delhi, India, 2000.
- 22. Woodroffe, C. Mangrove Sediments and Geomorphology. In *Tropical Mangrove Ecosystems*; Robertson, A.I., Alongi, D.M., Eds.; American Geophysical Union: Washington, DC, USA, 1992; Volume 41, pp. 7–41.
- Kumar, M.; Mohapatra, S.; Karim, A.A.; Dhal, N.K. Heavy Metal Fractions in Rhizosphere Sediment Vis-à-Vis Accumulation in *Phoenix Paludosa (Roxb.)* Mangrove Plants at Dhamra Estuary of India: Assessing Phytoremediation Potential. *Chem. Ecol.* 2021, 37, 1–14. [CrossRef]
- Friess, D.A.; Yando, E.S.; Alemu, J.B.; Wong, L.W.; Soto, S.D.; Bhatia, N. Ecosystem Services and Disservices of Mangrove Forests and Salt Marshes. In *Oceanography and Marine Biology: An Annual Review*; Hawkins, S.J., Allcock, A.L., Bates, A.E., Evans, A.J., Firth, L.B., McQuaid, C.D., Russell, B.D., Smith, I.P., Swearer, S.E., Todd, P.A., Eds.; Taylor & Francis: Baco Raton, FL, USA; London, UK; New York, NY, USA, 2020; Volume 58, pp. 107–141.
- 25. Patnaik, M.R.; Purohit, K.L.; Patra, A.K. Mangrove Swamps of Bhitarkanika, Orissa, India—A Great Eco Habitat for Wildlife. *Cheetal* **1995**, *34*, 1–9.
- 26. APHA; AWWA. Standard Methods for Examination of Water and Wastewater, 20th ed.; American Public Health Association: Washington, DC, USA, 1998.
- 27. Grasshoff, K.; Kremling, K.; Ehrhardt, M. (Eds.) Methods of Seawater Analysis, 3rd ed.; John Wiley & Sons: Hoboken, NJ, USA, 1999.
- 28. Jeffrey, S.W.; Humphrey, G.F. New Spectrophotometric Equations for Determining Chlorophylls a, b, C1 and C2 in Higher Plants, Algae and Natural Phytoplankton. *Biochem. Und Physiol. Pflanz.* **1975**, *167*, 191–194. [CrossRef]
- Strasserf, R.J.; Srivastava, A. Polyphasic chlorophyll a fluorescence transient in plants and cyanobacteria. *Photochem. Photobiol.* 1995, 61, 32–42. [CrossRef]
- Lewis, E.; Wallace DW, R.; Allison, L.J. Program Developed for CO₂ System Calculations; Oak Ridge National Laboratory: Oak Ridge, TN, USA, 1998.
- 31. Peng, T.-H.; Takahashi, T.; Broecker, W.S.; Olafsson, J. Seasonal Variability of Carbon Dioxide, Nutrients and Oxygen in the Northern North Atlantic Surface Water: Observations and a Model. *Tellus B* **1987**, *39 B*, 439–458. [CrossRef]
- 32. Weiss, R.F. Carbon Dioxide in Water and Seawater: The Solubility of a Non-Ideal Gas. Mar. Chem. 1974, 2, 203–215. [CrossRef]
- 33. Wanninkhof, R. Relationship between Wind Speed and Gas Exchange over the Ocean. J. Geophys. Res. 1992, 97, 7373. [CrossRef]
- 34. Liss, P.S.; Merlivat, L. Air Sea Gas Exchange Rates: Introduction and Synthesis. In *The Role of Air Sea Exchange in Geochemical Cycling*; Buat-Menard, P., Ed.; Springer: Dordrecht, The Netherlands, 1986; Volume 1986, pp. 113–119.
- 35. Behrenfeld, M.J.; O'Malley, R.T.; Siegel, D.A.; McClain, C.R.; Sarmiento, J.L.; Feldman, G.C.; Milligan, A.J.; Falkowski, P.G.; Letelier, R.M.; Boss, E.S. Climate-Driven Trends in Contemporary Ocean Productivity. *Nature* **2006**, 444, 752–755. [CrossRef]
- 36. Schuster, U.; Watson, A.J. A Variable and Decreasing Sink for Atmospheric CO₂ in the North Atlantic. *J. Geophys. Res.* 2007, 112, C11006. [CrossRef]
- Sabine, C.L.; Feely, R.A.; Gruber, N.; Key, R.M.; Lee, K.; Bullister, J.L.; Wanninkhof, R.; Wong, C.S.; Wallace, D.W.R.; Tilbrook, B.; et al. The Oceanic Sink for Anthropogenic CO₂. *Science* 2004, *305*, 367–371. [CrossRef]
- Strasser, R.J.; Srivastava, A.; Tsimilli-Michael, M. The Fluorescence Transient as a Tool to Characterize and Screen Photosynthetic Samples. In Probing Photosynthesis: Mechanism, Regulation & Adaptation; CRC Press: Boca Raton, FL, USA, 2014; p. 445.
- Prasil, O.; Suggett, D.J.; Cullen, J.J.; Babin, M. Aquafluo 2007: Chlorophyll Fluorescence in Aquatic Sciences, an International Conference Held in Nové Hrady. *Photosynth. Res.* 2007, 95, 111–115. [CrossRef]
- 40. Eggert, A.; Häubner, N.; Klausch, S.; Karsten, U.; Schumann, R. Quantification of Algal Biofilms Colonising Building Materials: Chlorophyll *a* Measured by PAM-Fluorometry as a Biomass Parameter. *Biofouling* **2006**, *22*, 79–90. [CrossRef] [PubMed]
- 41. Falkowski, P.G.; Raven, J.A. Aquatic Photosynthesis, 2nd ed.; Princeton University Press: Princeton, NJ, USA, 2007.
- 42. Mohapatra, P.K.; Schiewer, U. Influence of Dimethoate on the Structure and Function of the Natural Phytoplankton Assemblage of the Darss-Zingstbodden Chain Reared in a Laboratory. *Pol. J. Environ. Stud.* **1996**, *2*, 31–36.
- Wang, C.; Wang, Z.; Wang, P.; Zhang, S. Multiple Effects of Environmental Factors on Algal Growth and Nutrient Thresholds for Harmful Algal Blooms: Application of Response Surface Methodology. *Environ. Modeling Assess.* 2016, 21, 247–259. [CrossRef]
- 44. Chakrapani, G.J.; Subramanian, V. Preliminary Studies on the Geochemistry of the Mahanadi River Basin, India. *Chem. Geol.* **1990**, *81*, 241–253. [CrossRef]
- Giannini, F.; Mendes, C.R.B.; Garcia, C.A.E.; Carvalho, A.C.O.; Ciotti, A.M. Phytoplankton Community and the Fluorescence-Derived Photo-Physiological Parameters in the South Atlantic Ocean. J. Mar. Syst. 2021, 218, 103538. [CrossRef]