
Vermiculite Modified with Fe³⁺ Polyhydroxy Cations Is a Low-Cost and Highly Available Adsorbent for the Removal of Phosphate Ions

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Table S1. Effect of mass of FeOH-Vt on the adsorption of Phosphate from a 15.0 mg·L⁻¹ (in P) after 200 min of contact time under stirring. Temperature = 25.0 ± 0.5°C.

| Mass (mg) | Mean Mass (mg) | Initial P | Final P | Removed P | Adsorbed P | q (mg·g ⁻¹) | Mean q (mg·g ⁻¹) | SD ^a on q | Mean % Adsorption | SD on n % |
|--------------|----------------------|-----------------------|-----------------------|-----------------------|------------|-------------------------|---------------------------------|-------------------------|----------------------|--------------|
| | | (mg·L ⁻¹) | (mg·L ⁻¹) | (mg·L ⁻¹) | (mg) | | | % | Mean % Adsorption | |
| 25.6 | 25.3 | 15.0 | 13.02 | 1.98 | 0.0198 | 0.774 | 0.836 | 0.073 | 13.21 | 14.06 |
| 25.2 | | 15.0 | 12.93 | 2.06 | 0.0206 | 0.818 | | | 13.74 | |
| 25.0 | | 15.0 | 12.71 | 2.29 | 0.0229 | 0.917 | | | 15.29 | |
| 50.7 | 50.4 | 15.0 | 9.54 | 5.46 | 0.0546 | 1.078 | 1.021 | 0.056 | 36.43 | 34.32 |
| 50.3 | | 15.0 | 10.14 | 4.86 | 0.0486 | 0.966 | | | 32.38 | |
| 50.2 | | 15.0 | 9.88 | 5.12 | 0.0512 | 1.020 | | | 34.15 | |
| 75.3 | 75.2 | 15.0 | 6.35 | 8.65 | 0.0865 | 1.149 | 1.124 | 0.045 | 57.67 | 56.36 |
| 75.4 | | 15.0 | 6.92 | 8.08 | 0.0808 | 1.072 | | | 53.86 | |
| 75.0 | | 15.0 | 6.37 | 8.63 | 0.0863 | 1.151 | | | 57.53 | |
| 100.0 | 100.2 | 15.0 | 3.68 | 11.32 | 0.1132 | 1.132 | 0.981 | 0.075 | 75.47 | 68.84 |
| 100.1 | | 15.0 | 4.65 | 10.35 | 0.1035 | 1.034 | | | 68.98 | |
| 100.4 | | 15.0 | 5.69 | 9.31 | 0.0931 | 0.928 | | | 62.08 | |

^a Standard deviation

Table S2. Adsorption kinetics data obtained in a dispersion of 2.0 g of FeOH-Vt in 200 mL of an initial $4.84 \times 10^2 \mu\text{mol L}^{-1}$ phosphate solution ($15.0 \text{ mg}\cdot\text{L}^{-1}$ in P).

| Time (min) | $q (\text{mg}\cdot\text{g}^{-1})$ | $q (\mu\text{mol}\cdot\text{g}^{-1})$ | Adsorption % |
|------------|-----------------------------------|---------------------------------------|--------------|
| 5 | 0.165 | 5.327 | 21.2 |
| 10 | 0.242 | 7.798 | 26.3 |
| 20 | 0.339 | 10.935 | 32.8 |
| 40 | 0.483 | 15.581 | 42.5 |
| 80 | 0.727 | 23.483 | 58.9 |
| 120 | 0.914 | 29.505 | 71.4 |
| 160 | 1.032 | 33.293 | 79.2 |
| 200 | 1.138 | 36.726 | 86.4 |
| 240 | 1.182 | 38.161 | 89.3 |
| 280 | 1.221 | 39.405 | 91.9 |
| 320 | 1.248 | 40.307 | 93.8 |
| 360 | 1.246 | 40.233 | 93.6 |

Table S3. Data of adsorption isotherms ($25.0 \pm 0.5^\circ\text{C}$) in dispersions of 100 mg of FeOH-Vt in 10 mL phosphate solution with initial concentrations between 1.0 and 250.0 $\text{mg}\cdot\text{L}^{-1}$ in P.

| Initial P conc. ($\text{mg}\cdot\text{L}^{-1}$) | Final P conc. ($\text{mg}\cdot\text{L}^{-1}$) | $q (\text{mg}\cdot\text{g}^{-1})$ | Adsorption % |
|--|--|-----------------------------------|--------------|
| 1.0 | <LOD | 0.097 | >99 |
| 5.0 | 0.814 | 0.406 | 83 |
| 10.0 | 3.399 | 0.659 | 69 |
| 25.0 | 12.57 | 1.168 | 49 |
| 50.0 | 31.97 | 1.762 | 35 |
| 100.0 | 75.47 | 2.488 | 24.9 |
| 200.0 | 161.0 | 3.877 | 19.5 |
| 250.0 | 210.6 | 4.313 | 15.8 |