

*Supplementary Material for:*

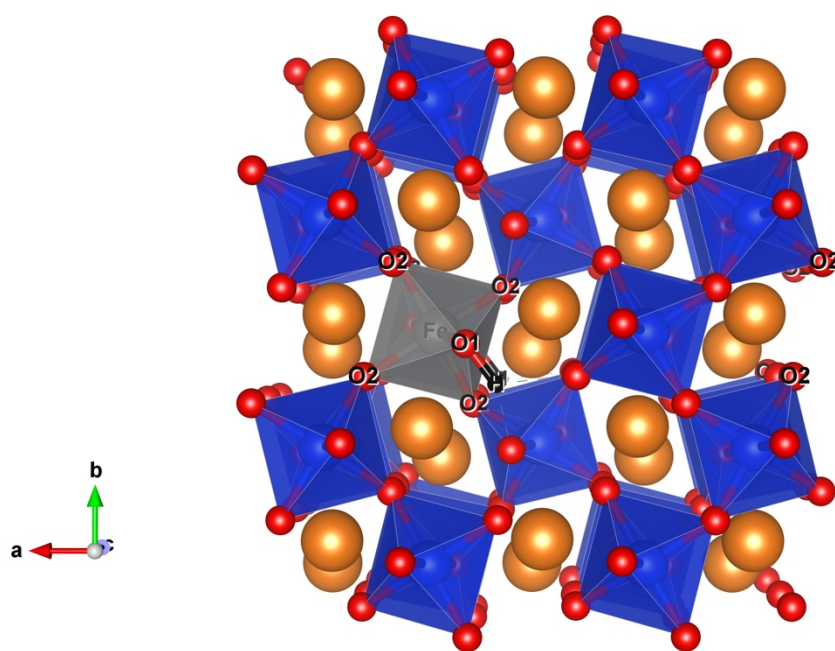
## **Vibrational and Thermodynamic Properties of Hydrous Iron-Bearing Lowermost Mantle Minerals**

**Jiajun Jiang <sup>1,2</sup>, Joshua M.R. Muir <sup>2</sup> and Feiwu Zhang <sup>2,\*</sup>**

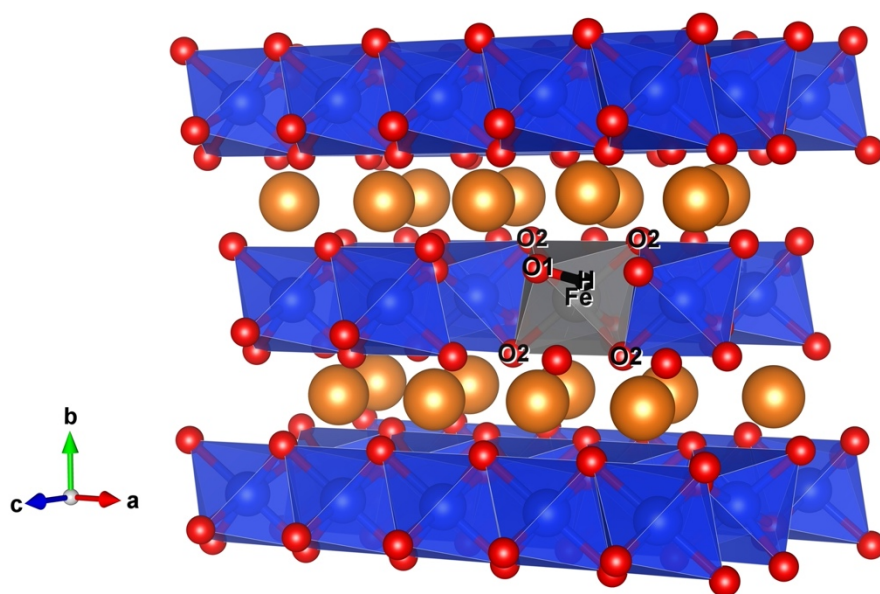
<sup>1</sup> Faculty of Land Resources Engineering, Kunming University of Science and Technology, Kunming 650093, China; jiangjiajun@kust.edu.cn

<sup>2</sup> State Key Laboratory of Ore Deposit Geochemistry, Institute of Geochemistry, Chinese Academy of Sciences, Guiyang 550081, China; j.m.r.muir@mail.gyig.ac.cn

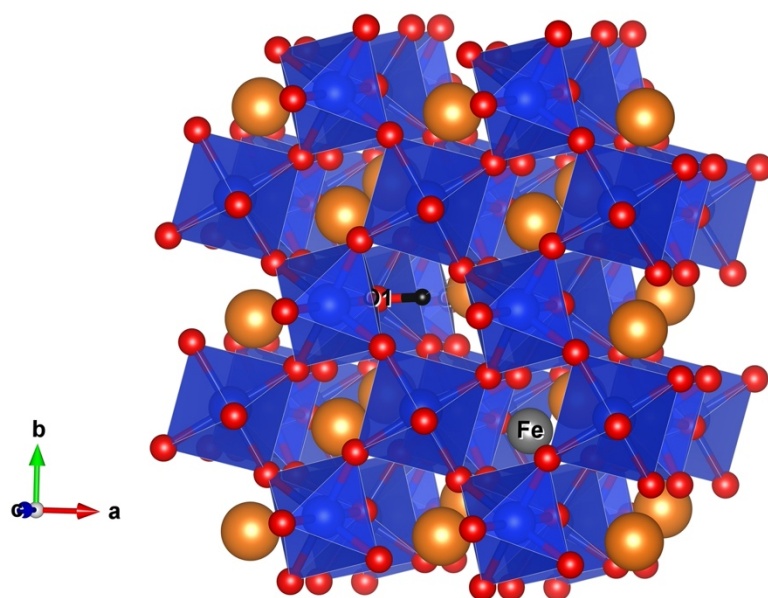
\* Correspondence: zhangfeiwu@vip.gyig.ac.cn



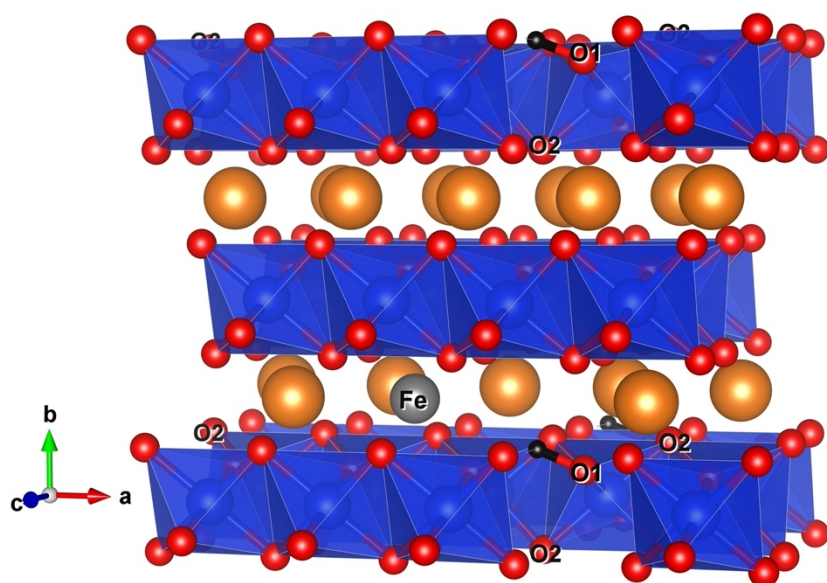
**Figure S1.** Crystal structures in details of  $[\text{Fe}^{3+}\text{-H}]_{\text{Si}}\text{-Brg}$ . In lattice structure, orange, dark blue, red, gray and black spheres represent magnesium, silicon, oxygen, iron and hydrogen, respectively.



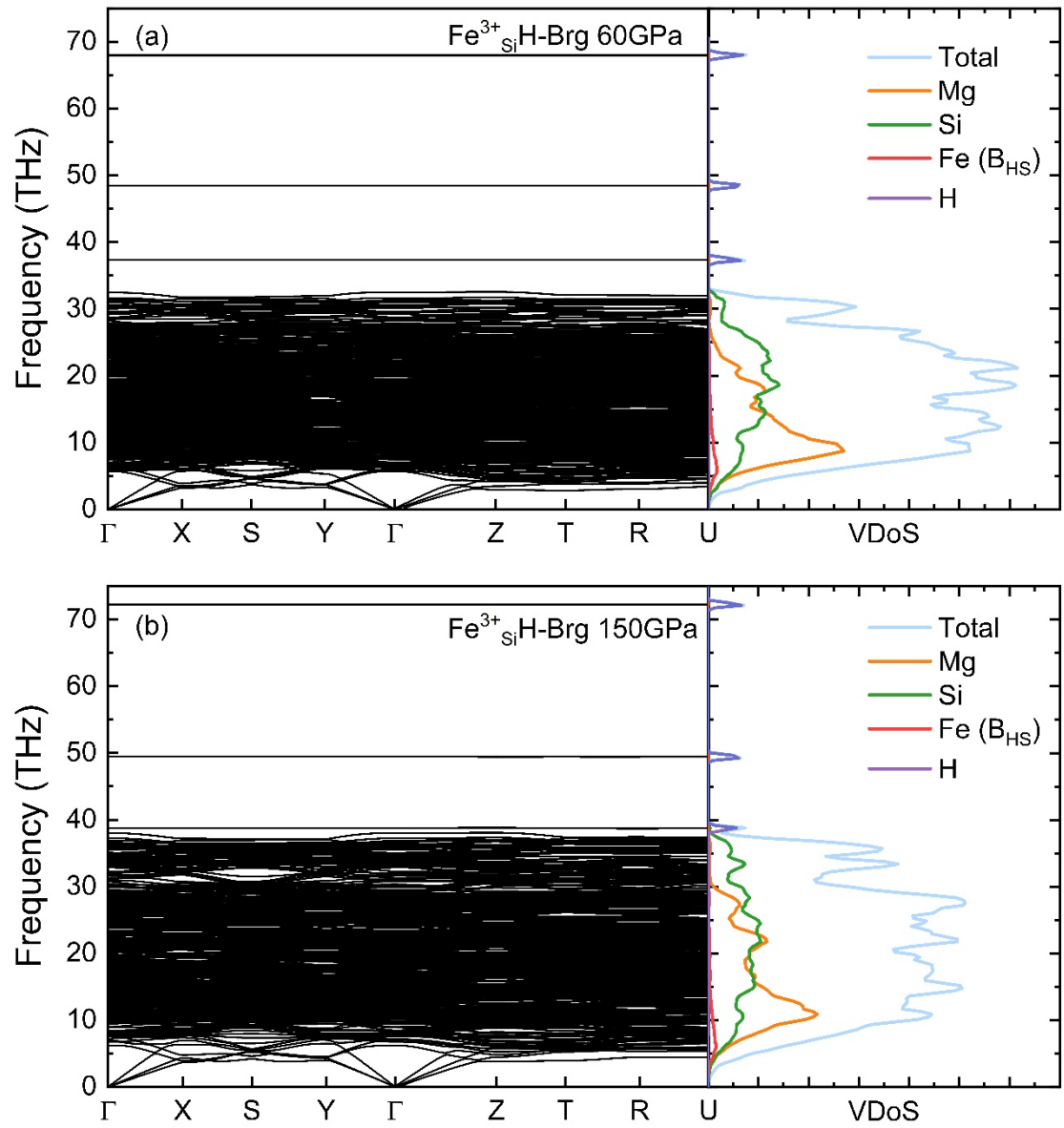
**Figure S2.** Crystal structures in details of  $[\text{Fe}^{3+}\text{-H}]_{\text{Si}}\text{-PPv}$ . In lattice structure, orange, dark blue, red, gray and black spheres represent magnesium, silicon, oxygen, iron and hydrogen, respectively.



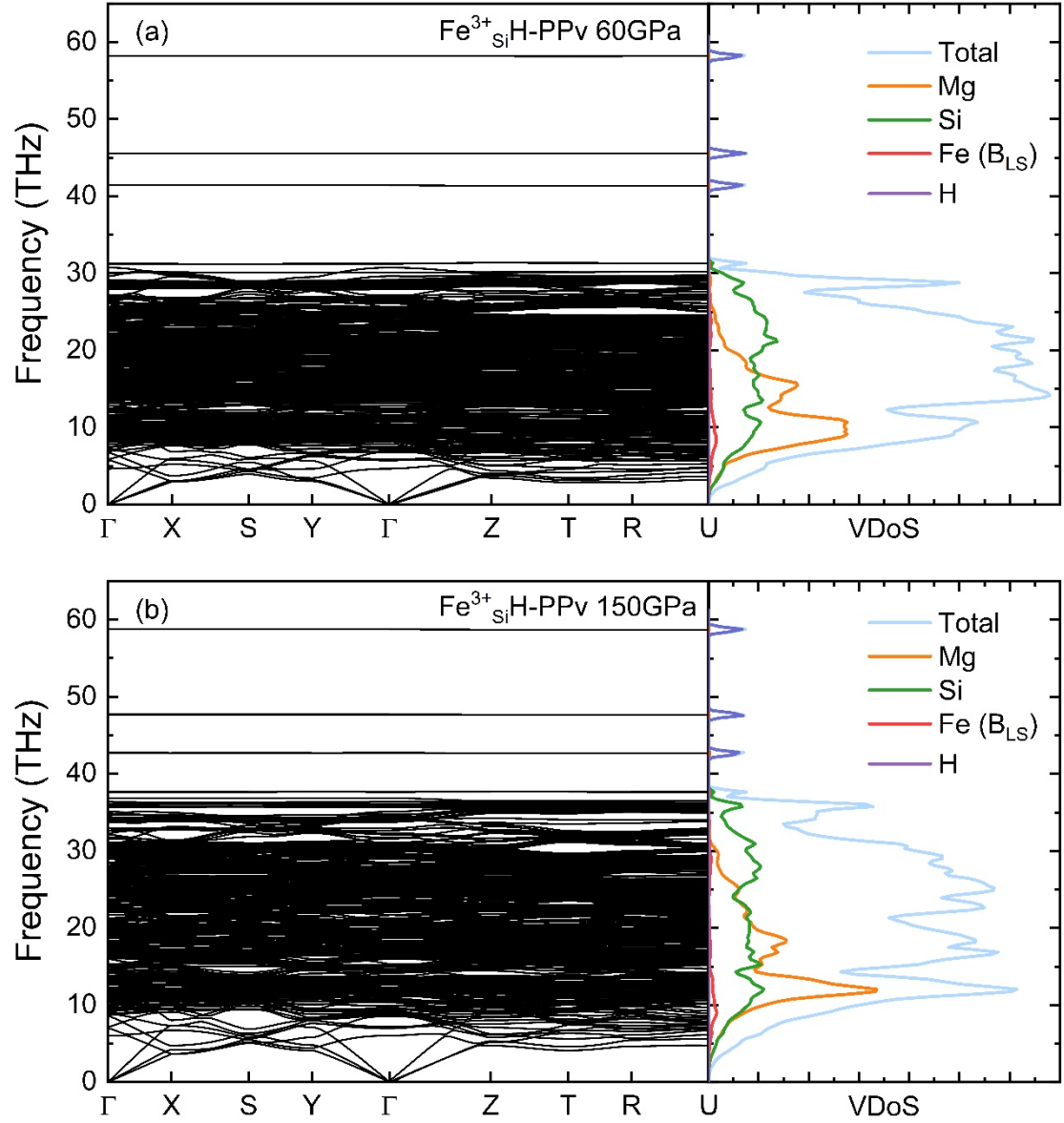
**Figure S3.** Crystal structures in details of  $[\text{Fe}^{3+}\text{-H}]_{\text{Mg-Mg-Brg}}$ . In lattice structure, orange, dark blue, red, gray and black spheres represent magnesium, silicon, oxygen, iron and hydrogen, respectively.



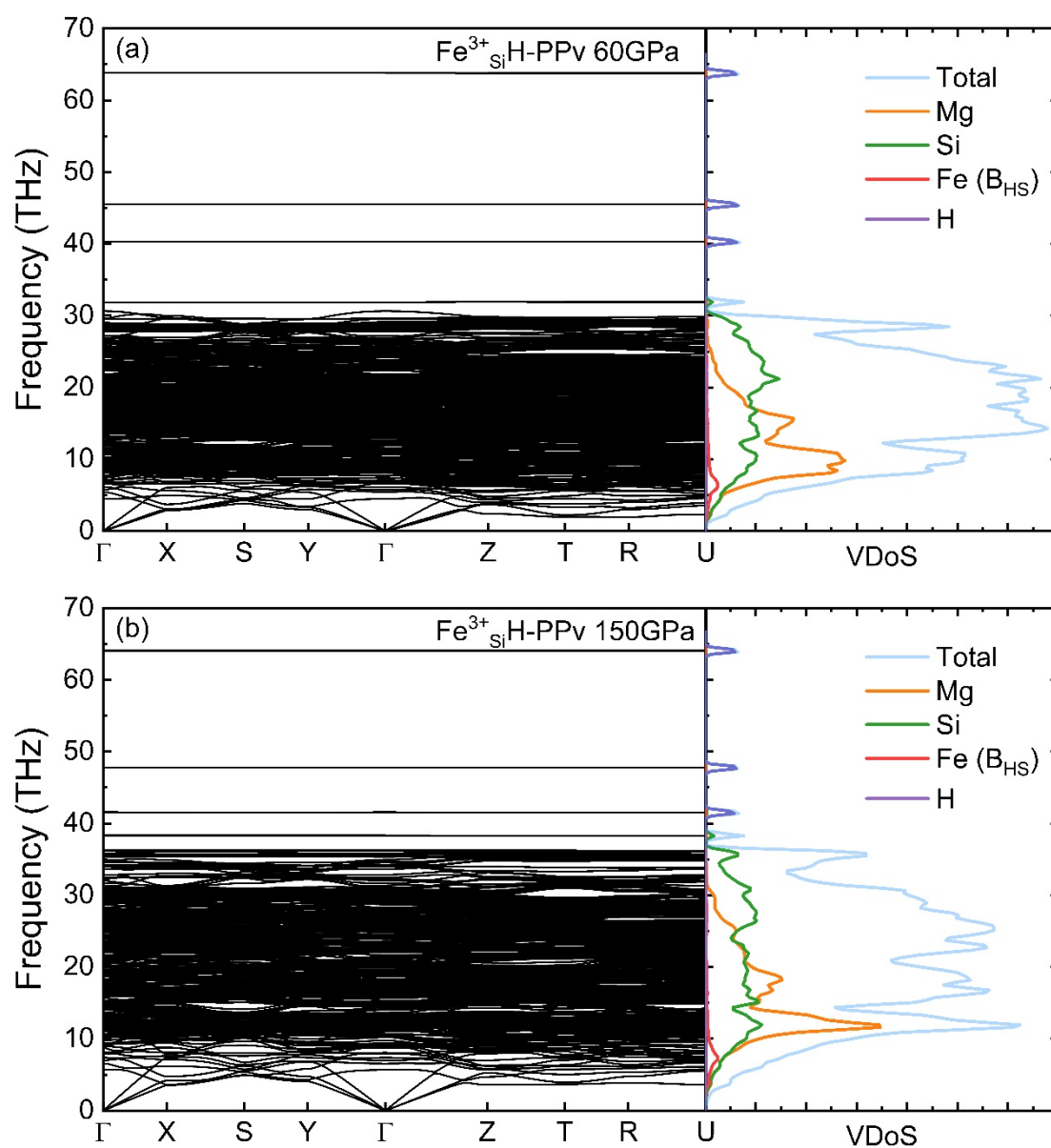
**Figure S4.** Crystal structures in details of  $[\text{Fe}^{3+}\text{-H}]\text{Mg-Mg-PPV}$ . In lattice structure, orange, dark blue, red, gray and black spheres represent magnesium, silicon, oxygen, iron and hydrogen, respectively.



**Figure S5.** Phonon dispersion curves and vibrational density of states (VDoS) of  $[\text{Fe}^{3+}\text{-H}]_{\text{Si}}\text{-Brg}$  with HS  $\text{Fe}^{3+}$  at 60 GPa (a) and 150 GPa (b).  $B_{\text{HS}}$  indicates that the high spin  $\text{Fe}^{3+}$  in B site (Si site). Total VDoS and partial VDoS of Mg, Si, Fe (HS), H are shown by light blue, orange, green, red and purple, respectively.

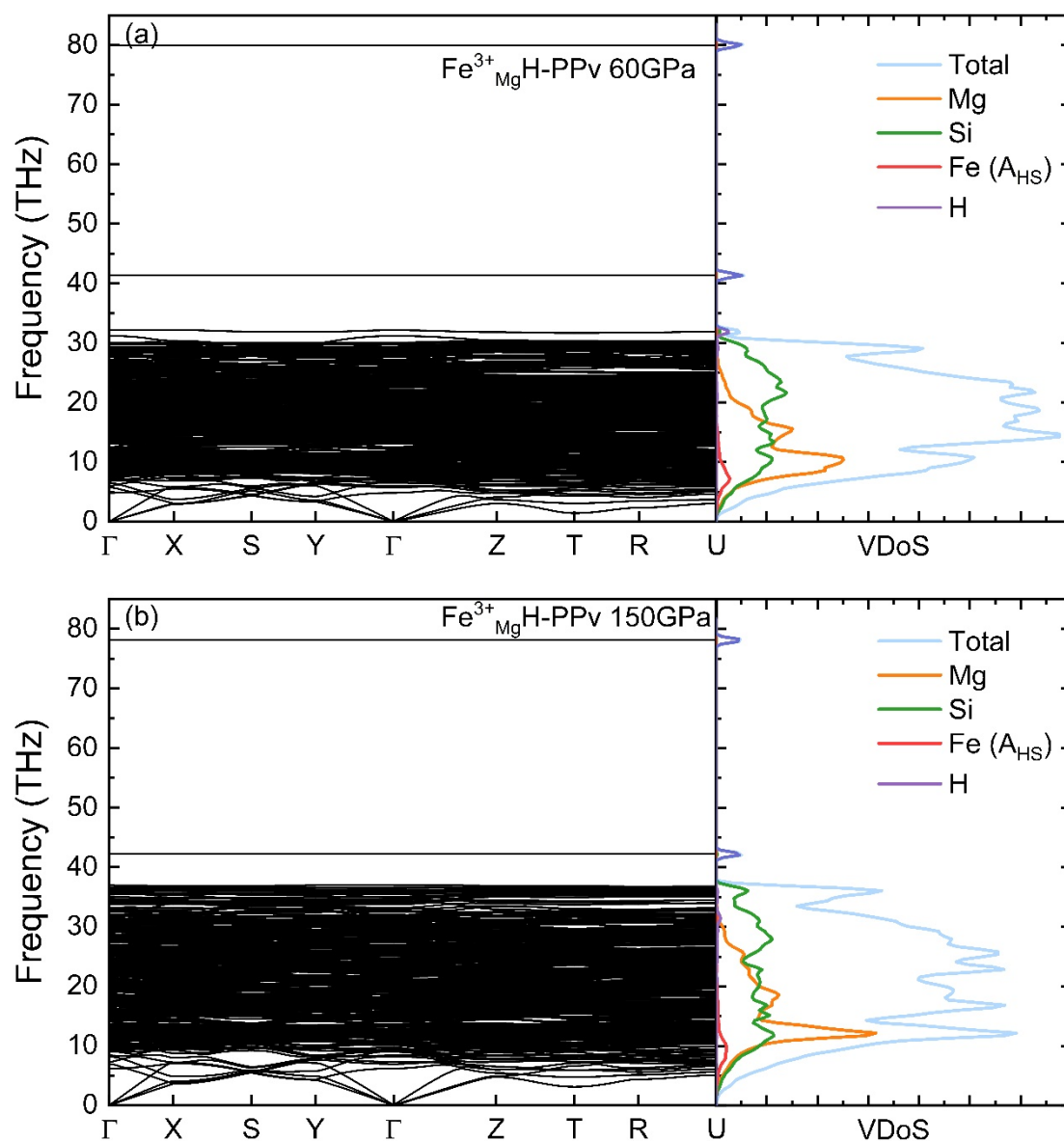


**Figure S6.** Phonon dispersion curves and vibrational density of states (VDoS) of  $[\text{Fe}^{3+}\text{-H}]_{\text{Si}}\text{-PPv}$  with LS  $\text{Fe}^{3+}$  at 60 GPa (a) and 150 GPa (b).  $\text{B}_{\text{LS}}$  indicates that the low spin  $\text{Fe}^{3+}$  in B site (Si site). Total VDoS and partial VDoS of Mg, Si, Fe (LS), H are shown by light blue, orange, green, red and purple, respectively.



**Figure S7.** Phonon dispersion curves and vibrational density of states (VDoS) of  $[\text{Fe}^{3+}\text{-H}]_{\text{Si}}\text{-PPv}$  with HS  $\text{Fe}^{3+}$  at 60 GPa (a) and 150 GPa (b).  $B_{\text{HS}}$  indicates that the high spin  $\text{Fe}^{3+}$  in B site (Si site). Total VDoS and partial VDoS of Mg, Si, Fe (HS), H are shown by light blue, orange, green, red and purple, respectively.





**Figure S8.** Phonon dispersion curves and vibrational density of states (VDoS) of  $[\text{Fe}^{3+}\text{-H}]_{\text{Mg-Mg-PPv}}$  at 60 GPa (a) and 150 GPa (b).  $A_{\text{HS}}$  indicates that the high spin  $\text{Fe}^{3+}$  in A site (Mg site). Total VDoS and partial VDoS of Mg, Si, Fe (HS), H are shown by light blue, orange, green, red and purple, respectively.

**Table S1.** Calculated thermodynamic parameters of hydrous iron-bearing and pure structure for bridgmanite\*

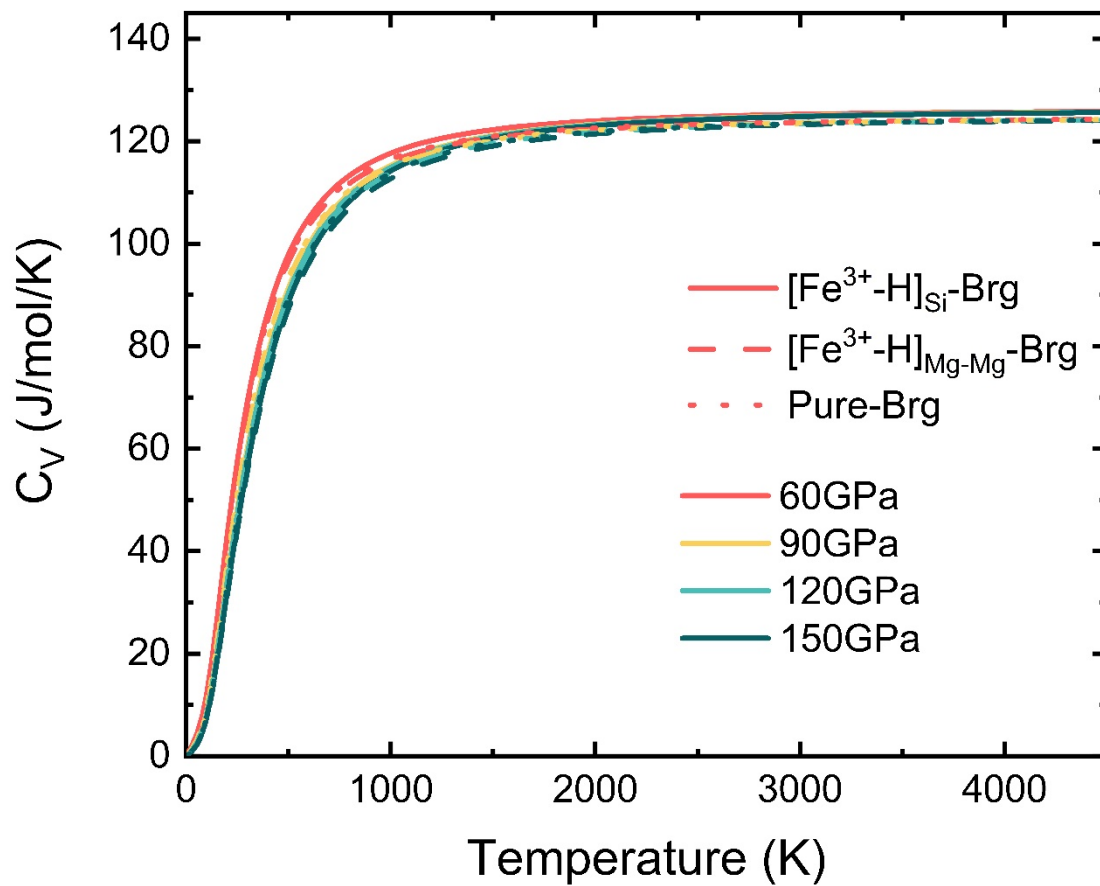
	90 GPa				120 GPa				150 GPa			
	300K	1000K	2000K	3000K	300K	1000K	2000K	3000K	300K	1000K	2000K	3000K
$K_T$ (GPa)	534.1	518.5	493.9	467.5	628.2	614.6	593.1	570.2	714.8	703.3	685.2	665.7
	536.9	520.6	495.1	468.4	632.1	617.8	595.1	571.2	719.8	707.7	688.1	667.3
	540.1	525.8	502.4	476.9	630.2	619.6	601.3	580.9	709.5	702.9	690.4	675.5
$\alpha$ ( $10^{-5}$ K $^{-1}$ )	0.74	1.38	1.53	1.63	0.60	1.17	1.29	1.35	0.51	1.04	1.14	1.18
	0.71	1.36	1.51	1.61	0.56	1.14	1.26	1.32	0.47	1.00	1.10	1.14
	0.69	1.34	1.48	1.57	0.58	1.16	1.26	1.31	0.53	1.07	1.15	1.17
$C_P$ (J mol $^{-1}$ K $^{-1}$ )	64.4	118.9	128.7	133.1	60.2	117.4	127.4	131.3	56.7	116.1	126.5	130.2
	63.1	117.3	126.9	131.2	59.0	115.7	125.6	129.4	55.6	114.4	124.7	128.2
	63.3	117.7	127.0	131.1	59.1	116.2	125.9	129.5	55.6	115.0	125.2	128.7
$\gamma$	1.23	1.25	1.26	1.28	1.18	1.20	1.21	1.22	1.15	1.18	1.18	1.18
	1.20	1.23	1.25	1.26	1.13	1.17	1.18	1.19	1.10	1.14	1.14	1.15
	1.17	1.23	1.25	1.26	1.17	1.20	1.21	1.21	1.23	1.21	1.20	1.20
$S_{\text{vib}}$ (J mol $^{-1}$ K $^{-1}$ )	38.6	153.0	236.2	286.7	35.5	147.4	230.6	281.0	32.4	141.4	224.2	274.5
	38.2	151.6	234.2	284.0	34.5	144.8	226.9	276.7	31.7	139.1	220.9	270.5
	37.5	151.4	234.2	284.1	33.8	144.5	226.8	276.6	31.3	141.7	223.7	273.4

\*Values in the first, second and third lines in each thermodynamic parameter are for [Fe $^{3+}$ -H]<sub>Si</sub>-Brg, [Fe $^{3+}$ -H]<sub>Mg-Mg</sub>-Brg and pure-Brg, respectively.

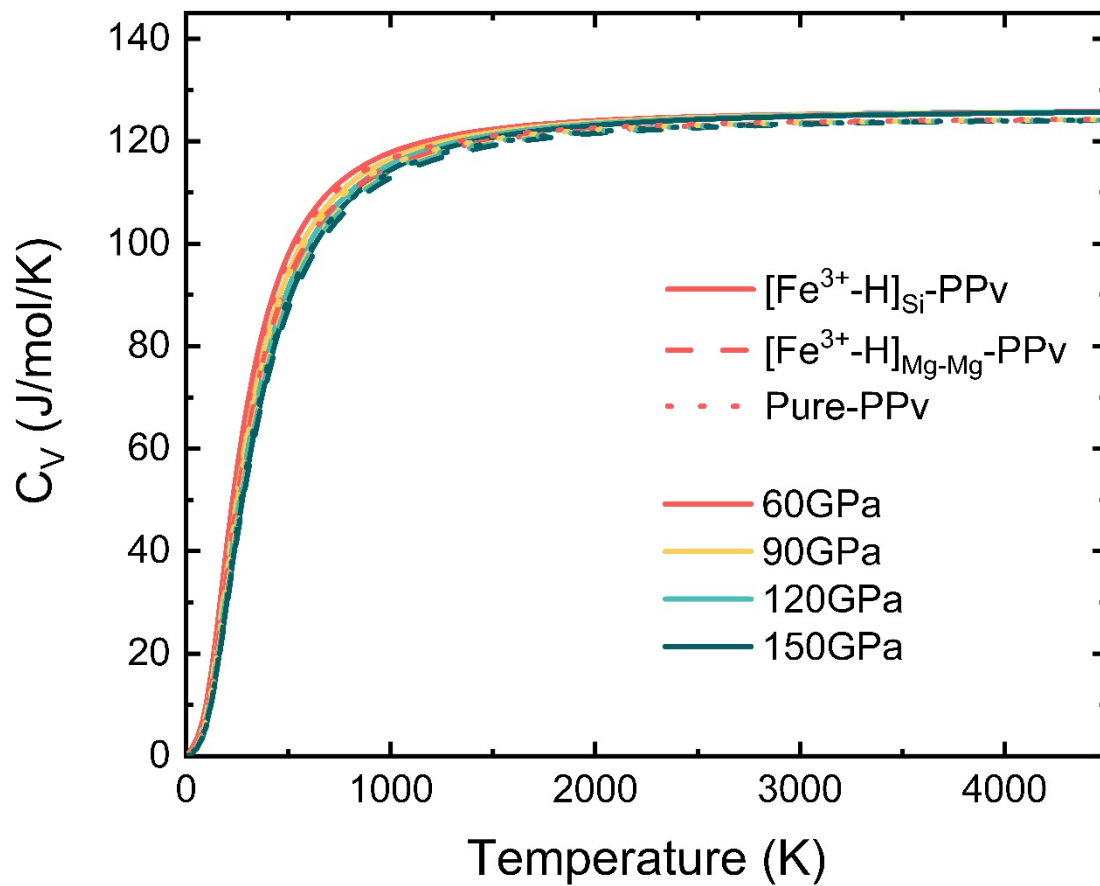
**Table S2.** Calculated thermodynamic parameters of hydrous iron-bearing and pure structure for post-perovskite\*

	90 GPa				120 GPa				150 GPa			
	300K	1000K	2000K	3000K	300K	1000K	2000K	3000K	300K	1000K	2000K	3000K
$K_T$ (GPa)	529.5	512.2	485.7	458.2	630.1	614.7	590.7	565.7	723.4	709.9	688.6	666.3
	527.9	511.0	485.1	458.5	632.8	617.4	593.4	568.7	731.9	718.0	696.3	673.6
	532.9	517.1	492.3	466.1	629.0	616.0	595.1	572.7	714.3	704.3	687.8	669.6
$\alpha$ ( $10^{-5}$ K $^{-1}$ )	0.76	1.42	1.58	1.68	0.59	1.18	1.31	1.38	0.49	1.03	1.14	1.18
	0.74	1.39	1.54	1.64	0.57	1.15	1.27	1.34	0.47	1.00	1.10	1.15
	0.72	1.38	1.53	1.62	0.58	1.17	1.28	1.34	0.51	1.05	1.15	1.18
$C_P$ (J mol $^{-1}$ K $^{-1}$ )	63.7	119.2	128.7	133.4	59.4	117.6	127.5	131.4	55.8	116.3	126.6	130.2
	62.3	117.4	127.0	131.2	58.1	115.9	125.7	129.4	54.7	114.6	124.8	128.3
	62.6	117.9	127.1	131.3	58.2	116.4	125.9	129.6	54.7	115.1	125.2	128.7
$\gamma$	1.24	1.25	1.26	1.28	1.17	1.19	1.20	1.21	1.13	1.16	1.17	1.17
	1.21	1.22	1.23	1.24	1.15	1.17	1.18	1.19	1.12	1.14	1.15	1.16
	1.19	1.23	1.24	1.26	1.15	1.19	1.20	1.20	1.17	1.18	1.18	1.18
$S_{\text{vib}}$ (J mol $^{-1}$ K $^{-1}$ )	37.1	152.3	236.1	286.7	33.3	145.2	228.5	279.0	30.4	139.3	222.3	272.6
	36.2	149.5	232.2	282.0	32.5	142.6	224.9	274.6	29.7	137.0	218.8	268.5
	35.1	149.1	232.0	281.9	31.5	142.2	224.6	274.4	29.1	136.9	222.1	271.8

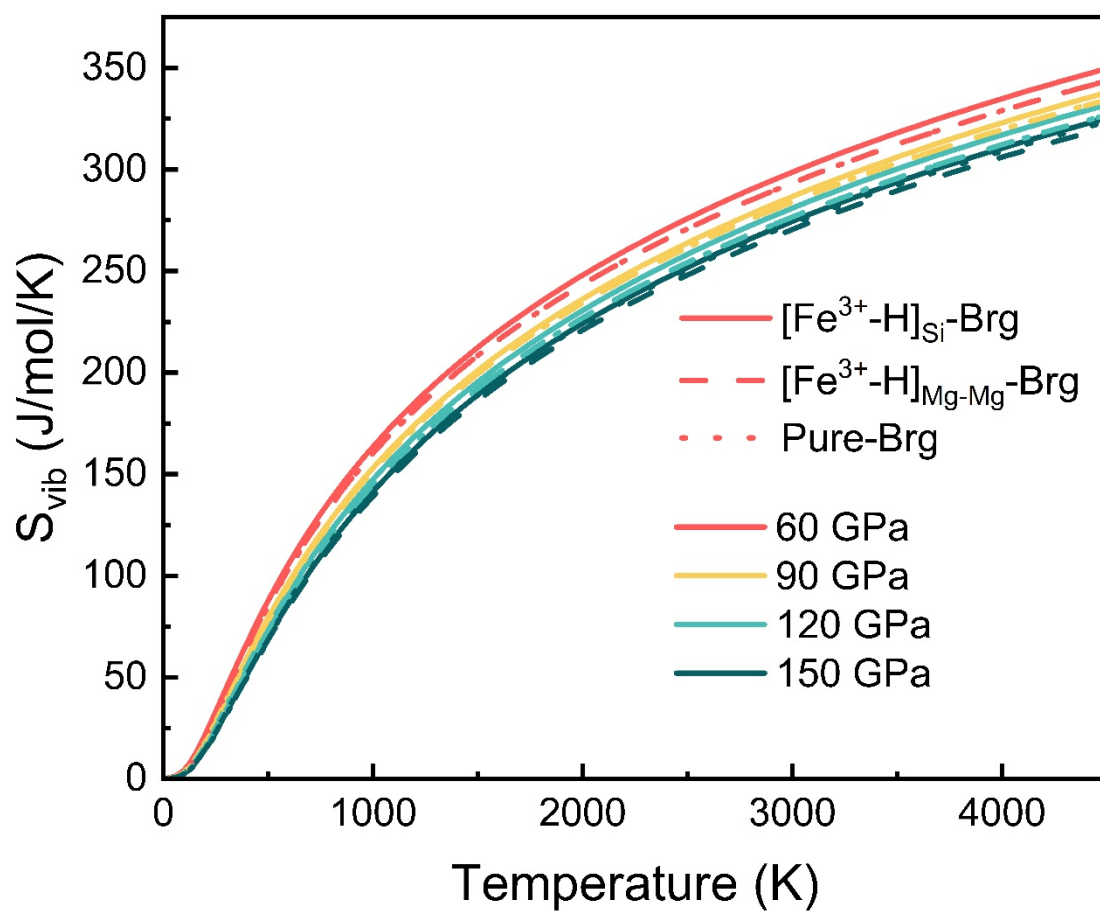
\*Values in the first, second and third lines in each thermodynamic parameter are for [Fe<sup>3+</sup>-H]<sub>Si</sub>-PPv, [Fe<sup>3+</sup>-H]<sub>Mg-Mg</sub>-PPv and pure-PPv, respectively.



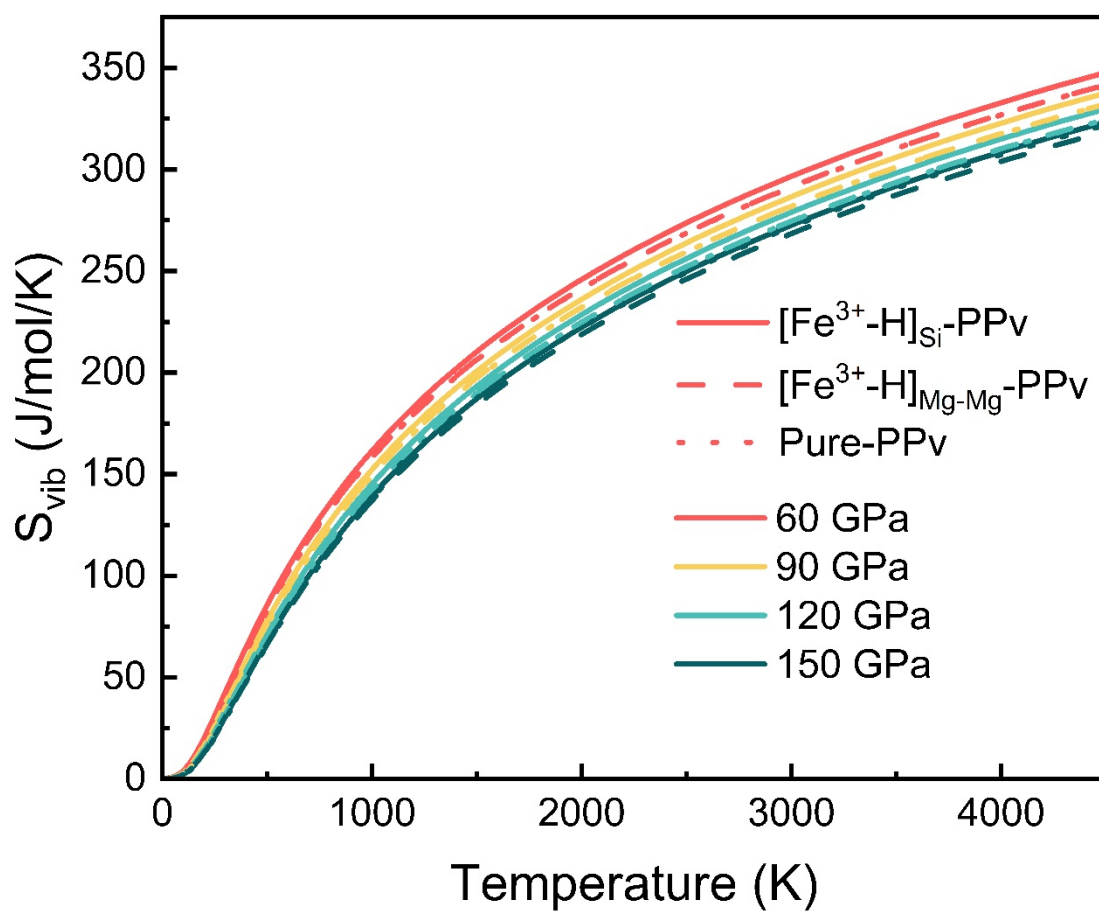
**Figure S9.** The heat capacity at constant volume as a function of temperature at different pressures. Solid, dashed and dotted lines represent the  $[\text{Fe}^{3+}\text{-H}]_{\text{Si-Brg}}$ ,  $[\text{Fe}^{3+}\text{-H}]_{\text{Mg-Mg-Brg}}$  and Pure-Brg, respectively.



**Figure S10.** The heat capacity at constant volume as a function of temperature at different pressures. Solid, dashed and dotted lines represent the  $[\text{Fe}^{3+}\text{-H}]_{\text{Si}}\text{-PPv}$ ,  $[\text{Fe}^{3+}\text{-H}]_{\text{Mg-Mg}}\text{-PPv}$  and Pure-PPv, respectively.



**Figure S11.** Temperature dependence of vibrational entropy at different pressures. Solid, dashed and dotted lines represent the  $[\text{Fe}^{3+}\text{-H}]_{\text{Si-Brg}}$ ,  $[\text{Fe}^{3+}\text{-H}]_{\text{Mg-Mg-Brg}}$  and Pure-Brg, respectively.



**Figure S12.** Temperature dependence of vibrational entropy at different pressures. Solid, dashed and dotted lines represent the  $[\text{Fe}^{3+}\text{-H}]_{\text{Si}}\text{-PPv}$ ,  $[\text{Fe}^{3+}\text{-H}]_{\text{Mg-Mg}}\text{-PPv}$  and Pure-PPv, respectively.