

Article

Pollution Characteristics, Distribution and Ecological Risk of Potentially Toxic Elements in Soils from an Abandoned Coal Mine Area in Southwestern China

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Abstract: Acid mine drainage (AMD) from abandoned coal mines can lead to serious environmental problems due to its low pH and high concentrations of potentially toxic elements. In this study, soil pH, sulfur (S) content, and arsenic (As), cadmium (Cd), chromium (Cr), copper (Cu), lead (Pb), nickel (Ni), zinc (Zn), iron (Fe), manganese (Mn), and mercury (Hg) concentrations were measured in 27 surface soil samples from areas in which coal-mining activities ceased nine years previously in Youyu Catchment, Guizhou Province, China. The soil was acidic, with a mean pH of 5.28. Cadmium was the only element with a mean concentration higher than the national soil quality standard. As, Cd, Cu, Ni, Zn, Mn, Cr, and Fe concentrations were all higher than the background values in Guizhou Province. This was especially true for the Cd, Cu, and Fe concentrations, which were 1.69, 1.95, and 12.18 times their respective background values. The geoaccumulation index of Cd and Fe was present at unpolluted to moderately polluted and heavily polluted levels, respectively, indicating higher pollution levels than for the other elements in the study area. Spatially, significantly high Fe and S concentrations, as well as extremely low pH values, were found in the soils of the AMD sites; however, sites where tributaries merged with the Youyu River (TM) had the highest Cd pollution level. Iron originated mainly from non-point sources (e.g., AMD and coal gangues), while AMD and agricultural activity were the predominant sources of Cd. The results of an eco-risk assessment indicated that Cd levels presented a moderate potential ecological risk, while the other elements all posed a low risk. For the TM sites, the highest eco-risk was for Cd, with levels that could be harmful for aquatic organisms in the wet season, and may endanger human health via the food chain.

Keywords: acid mine drainage; abandoned coal mine; potentially toxic elements; pollution level; potential ecological risk



Citation: Pan, L.; Guan, X.; Liu, B.; Chen, Y.; Pei, Y.; Pan, J.; Zhang, Y.; Hao, Z. Pollution Characteristics, Distribution and Ecological Risk of Potentially Toxic Elements in Soils from an Abandoned Coal Mine Area in Southwestern China. *Minerals* **2021**, *11*, 330. <https://doi.org/10.3390/min11030330>

Academic Editors: Ana Romero-Freire and Hao Qiu

Received: 29 January 2021

Accepted: 16 March 2021

Published: 22 March 2021

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1. Introduction

Mining activity has important economic and social benefits; however, inadequately treated acid mine drainage (AMD) is globally recognized as one of the environmental issues, due to its acidic nature (extremely low pH of <3), and the presence of different potentially toxic elements, such as cadmium (Cd), copper (Cu), iron (Fe), manganese (Mn), lead (Pb), and zinc (Zn), toxic metalloids (e.g., arsenic (As) and selenium (Se), and sulfates) [1,2], posed high threat to soil, water, and the local inhabitants in the mine-affected areas. Acid mine drainage (AMD) occurs in sulfide-bearing mine waste due to the oxidation of iron sulfides, especially pyrite (FeS₂), pyrrhotite (FeS) and marcasite (FeS₂) leading to the creation of several soluble hydrous iron sulfates, the generation of acidity and subsequent leaching of metals [3–6]. Acid mine drainage can lead to soil acidification, which reduces soil enzyme activity, and inhibits plant respiration. Simultaneously, the low pH promotes

the growth of acidophilic microorganisms, further accelerating the generation of AMD [7]. However, the potentially toxic elements in AMD can be ingested by plants, animals, and humans through the food chain, where their enrichment can harm human health [8]. Another prominent feature of AMD is its high Fe concentration, which is of great concern because it can influence the biogeochemistry of the ambient environment [9]. A diverse range of microorganisms capable of metabolizing Fe compounds can be leached from AMD into soils, and may play an important role in Fe cycling in soils. Therefore, the Fe cycle can potentially lead to unique microbial community assemblages in AMD-impacted soils compared to pristine soils [10].

Guizhou is located in southwestern China and has some of the country's largest coal resources. The coal reserves of Guizhou, which total 100 million tons, are distributed over an area of more than 70,000 km², accounting for 40% of the total land area. It was reported that as of December 31, 2019, there were 249 coal mines in production in Guizhou Province, with a total production capacity is 133.52 million tons per year. High sulfur (S) content in the main feature of the coal in Guizhou province, which can be in excess of 3% [11]. Among the sulfide minerals, FeS₂ is the most widespread in nature and is regarded as the main cause of AMD [12,13]. Youyu Catchment in Huaxi District, locating in the upstream of Aha reservoir, which is an important water resource for the Guiyang City, has a 60-year history of coal mining. The coal resources in Huaxi District are shallow, with most of the coal seams being exposed to the ground. It has therefore been proven difficult to conduct large-scale open-pit mining, and only underground mining has been established. During mining activities undertaken in the region's small coal mines, coal bunkers, liver dumps, and sewage treatment tanks are usually established close to the main cave. Coal gangue waste piles are typically disordered, and most have not been treated or protected after mine closure; moreover, the management of coal gangue piles is even less effective. In rainy season, waste piles have been washed away by rainwater, potentially toxic elements may continue to migrate via runoff water. In early 2010, the large-scale mining activities with regard to the environmental and economic aspects had been stopped by the local authority and the government. However, Guizhou is one of the most typical areas of karst landform in China, distributing 1710 springs on the ground with an average flow of 121.23 billion m³/a. Even after abandonment, the continuously produced AMD with the presence the O₂, microorganism, and water, the AMD could still pollute the surrounding environment via springs in rainy days. Therefore, the environmental pollution and potentially ecological risks left by the decades of mining activity may last for very long time. There have been several studies focused on Fe-cycling, metals behavior, and remediation technology in the river, sediment, and soils contaminated by AMD [2,14,15]. To date, relatively few studies have reported comprehensively for soil properties and pollution level in Youyu Catchment after a long period time of closure. Even after nine years of self-restoration, Youyu Catchment's pollution is still not removed and the situation is unclear, so the aims of this study were to do the following: (1) estimate the soil acidity, soil sulfur, and levels of potentially toxic elements in the soil nine years after mine closure; (2) identify the sources of potentially toxic elements in the soil; and (3) evaluate the eco-risk posed by the distribution of potentially toxic elements in the soil.

2. Materials and Methods

2.1. Soil Sampling

Youyu Catchment is located in Huaxi District, Guiyang City, which is in the central area of Guizhou Province, China. The catchment has a total population of approximately 14,227 and covers an area of 53.33 km². The terrain is mostly mountains and hills. The area has a humid plateau monsoon climate, characterized by a lack of severe cold in winter and of extreme heat in summer, and by abundant rainfall and high humidity. The average annual temperature and rainfall are 14.9 °C and 1178.3 mm, respectively. The altitude ranges between 1090 and 1450 m. Silico-aluminum yellow soil is the main soil type in the study area. Youyu Catchment is characterized by well-developed agricultural activities,

with the main products cultivated being tea, vegetables, and fruit. The major industries include coking, cement production, and metal foundries [16]. The soil sample collection method used herein was described by Pan et al. [17]. In June 2019, we collected soil samples surrounding eight coal gangues and six AMD effluents. Briefly, the soil samples were divided into four types: (1) soil samples surrounding (within 100 m) coal gangues (SCG); (2) samples within 100 m, where AMD emerges from underground (AMD); (3) soil samples from where tributaries merge into the Youyu River (TM); and (4) soil samples from control areas (CK). A 2 m × 2 m grid spacing was used for soil sampling, and a total of 27 surface (0–20 cm) soil samples were collected, using the five-point sampling method. At each sampling site, five subsamples were taken from the same area (approximately 4 m²) and pooled to form one composite sample. The sampling sites are illustrated in the Figure 1.

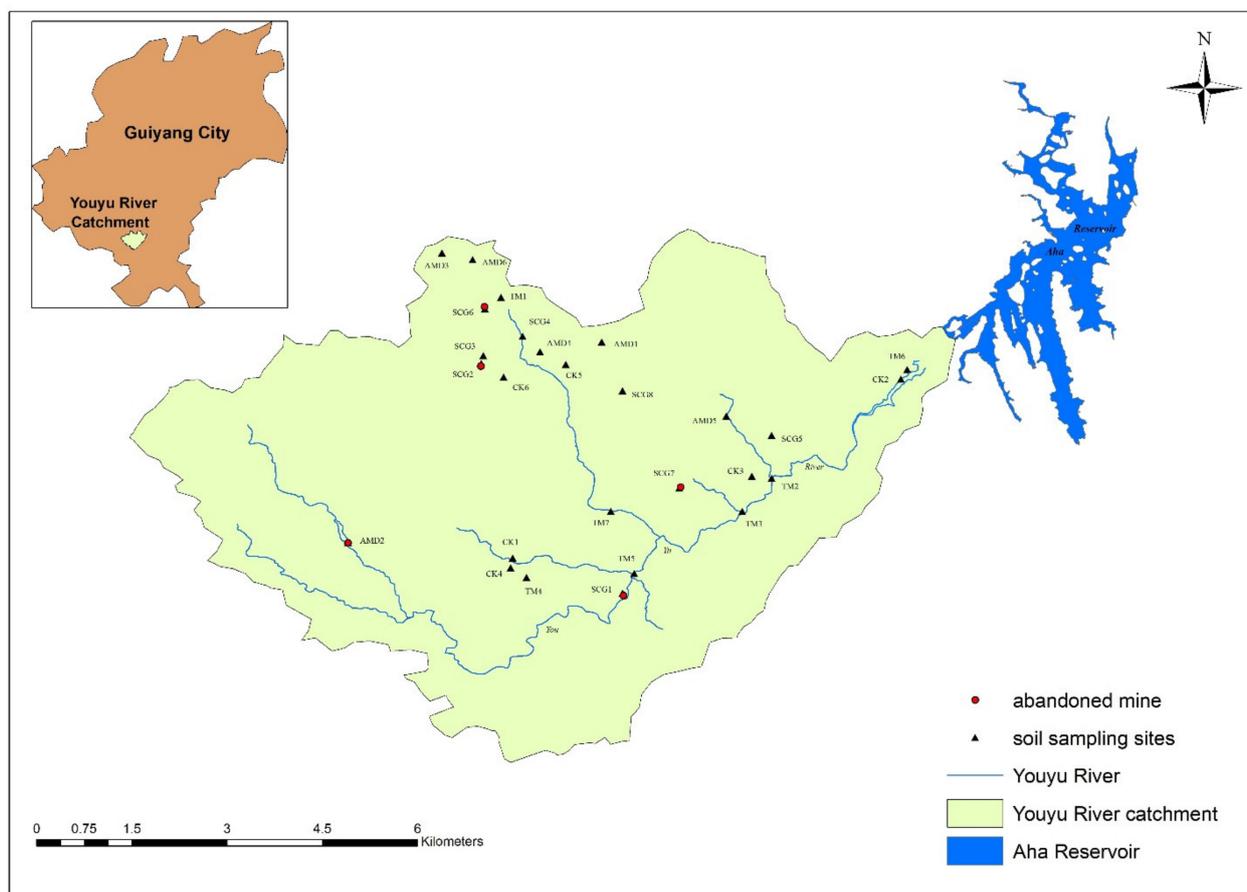


Figure 1. Locations of sampling sites in the studied area.

2.2. Sample Preparation and Analysis

All samples were air-dried for 1 week, at room temperature, in a storage room, passed through a 60-mesh sieve after removing stones, residual roots, and other unwanted materials, and then sealed in brown glass bottles and conserved in a refrigerator, at $-4\text{ }^{\circ}\text{C}$, until analysis. Samples were analyzed by following the 3050B and 6010C methods established by the United States Environmental Protection Agency (USEPA) [18,19]. In brief, approximately 10 g of sample was air-dried, gently ground with an agate mortar, and passed through a 100-mesh nylon sieve. Then, 1 g of the sample was placed into an Anton PVC (Polyvinyl chloride) digestion vessel together with 9 mL concentrated nitric acid (HNO_3) and 3 mL hydrogen peroxide (H_2O_2). The vessel was sealed and heated at $180\text{ }^{\circ}\text{C}$ for 15 min. The concentrations of eight potentially toxic elements (As, Cd, Cu, Pb, Ni, Zn, Mn, Cr, and Hg) in the digestion solution were determined by using inductively coupled

plasma–mass spectrometry (ICP–MS; POEMS3; Thermo Fisher Scientific, Waltham, MA, USA).

Iron levels in soil were determined, using the national standard method (HJ 804-2016), with 10.0 g samples (accurate to 0.01 g) placed in a 100 mL conical flask and 20.0 mL extract liquid ($c_5(\text{TEA, Triethanolamine}) = 0.1 \text{ mol /L}$, $c(\text{CaCl}_2) = 0.01 \text{ mol /L}$, $c(\text{DTPA, Diethylenetriaminepentaacetic acid}) = 0.005 \text{ mol /L}$; $\text{pH} = 7.3$) then added. The solution was shaken at $20 \pm 2 \text{ }^\circ\text{C}$ at a speed of 160~200 r/min for 2 h. The extraction solution was then slowly poured into a centrifuge tube and centrifuged for 10 min. The supernatant was isolated within 48 h of gravity filtration, using a medium-speed quantitative filter paper. Inductively coupled plasma–atomic emission spectrometry was used to detect iron levels.

For measuring pH, 10 g of soil was mixed with 25 mL of deionized water (1:2.5 (m/v) soil-to-water ratio), and the supernatant was measured by using a pH meter (HQ30d; Hach, Loveland, CO, USA), according to the methods of the People’s Republic of China National Environmental Protection Standards (HJ962–2018) [20]. The S content in soil was determined by using an elemental analyzer (PE 2400-II; Perkin Elmer, Waltham, MA, USA).

2.3. Quality Assurance and Quality Control

Soil standard reference materials (GBW07401 and GSS-1), obtained from the Center of National Standard Reference Material of China, was used for quality assurance and quality control (QA/QC). The accepted recoveries ranged from 81.0% to 109%. Analysis methods were evaluated by using blank ($n = 7$) and duplicate samples ($n = 13$) for each set of samples. The relative deviation of the duplicate samples was $<7\%$ for all batch treatments.

2.4. Data Analysis

The data analysis was performed, using SPSS software (version 18.0; SPSS Inc., Chicago, IL, USA). Descriptive statistics generated for analysis of the potentially toxic elements in the soil samples included the maximum, minimum, median, and mean values, and the standard deviation (SD). Relationships between concentrations of potentially toxic elements were investigated, using Spearman correlation coefficients. The spatial distribution of the concentrations and eco-risks of potentially toxic elements in Huaxi District was visualized, using ArcGIS software (version 10.3; ESRI, Redlands, CA, USA).

2.5. The Geoaccumulation Index (Igeo)

The Igeo is a geochemical criterion introduced by Müller [21]. It can be used to evaluate soil contamination by comparing the differences between current and preindustrial concentrations of potential contaminants. Unlike other methods of pollution assessment, the Igeo takes the natural diagenesis process into account, making the assessments more practical. The Igeo was calculated by using the following equation:

$$I_{geo} = \log_2 (C_n / 1.5 \times B_n) \quad (1)$$

where C_n is the measured concentration of the elements in soil (mg kg^{-1}), and B_n is the geochemical background value (mg kg^{-1}). A coefficient of 1.5 was applied due to potential variations in the baseline data [22]. The relationships between the Igeo and the pollution levels are given in Supplementary Materials Table S1.

2.6. Potential Ecological Risk Index (RI)

The RI, proposed by Hakanson [23], is a relatively rapid, simple, and standard method for assessing the ecological risk posed by potentially toxic elements in soils or sediments [24]. The RI can reflect the potential ecological risk according to the overall pollution level. The formula is as follows:

$$RI = \sum E_r^i = \sum T_r^i (C_s^i / C_B^i) \quad (2)$$

The RI is the sum of ecological risk indices of various potentially toxic elements; E_r^i is the ecological risk factor for individual elements; C_s^i (mg/kg) is the concentration of a target element in a sample soil; C_B^i (mg/kg) is the background value of the element in Guizhou Province (Table 1); and T_r^i is the toxicity response factor of each metal. Liu et al. [25] calculated the toxicity response coefficient of TI in a potential ecological risk assessment, with the value set to 10. The toxic response coefficients of As, Cd, Cr, Cu, Pb, and Zn were 10, 30, 2, 5, 5, and 1, respectively. The classification standard for the RI of the potentially toxic elements is also shown in Supplementary Materials Table S1.

3. Results and Discussion

3.1. pH

Soil pH is usually classified into the following five categories: strongly acidic (pH < 4.5), acidic (4.5 ≤ pH < 5.5), slightly acidic (5.5 ≤ pH < 6.5), neutral (6.5 ≤ pH < 7.5), alkaline (7.0 ≤ pH < 8.5), and strongly alkaline (pH ≥ 8.5) [26]. As shown in Table 1, the range of soil pH values in the study area was 0.94–7.77 (mean, 5.44 ± 1.46, i.e., acidic). In contrast to the soil background pH value (6.2) in Guizhou Province, 60% of the samples were classified as strongly acidic. The sample site with the most acidic soil conditions was AMD4 (0.94), which was located in an area affected by the largest flows of AMD effluents. Most of the other sites with strongly acidic soil conditions (pH range of 3.27–5.89) were located on or surrounding coal gangues, e.g., AMD1, SCG3, and SCG5. In contrast, the soil pH in sites relatively remote from the coal gangues (and thus from the impact of AMD effluents) was neutral to alkaline. Similar trends were also observed in other studies conducted in Youyu Catchment. For example, the pH was found to be 4.48 ± 0.93 in farmland soil from Youyu Catchment, and irrigation by AMD was considered to be the main reason for soil acidification [27]. The farmland and paddy soil pH values were reported to be 4.52–6.40 and 4.71–5.55, respectively, in a study conducted in the same area ago, and the soil pH within 100 m of coal gangues was lower than that at sites 300 m away [28]. Studies in other coal-mining areas in Guizhou Province have also revealed a strongly acidic soil pH, e.g., Zhijin mine in Bijie City (4.78–5.38 for paddy soil and 4.995–5.77 for farmland soil), with AMD resulting from high-S ores and gangues identified as the dominant source. The study area was located in the karst landform region of Guizhou Province, which is dominated by carbonate rocks, and the background soil pH is neutral to slightly alkaline. Since the early 1980s, coal resources have been exploited on a large scale in Youyu Catchment, with numerous small coal mines and several larger private mines forming large-scale goafs. Communication between coal bearing strata and the overlying karst strata has led to karst water flowing into pits and interacting with coal seams, which has generated acidic wastewater containing rusty, yellow-colored granular suspended matter. Because most coal mines are located along the Youyu River, the acid waste water from mining has typically been discharged into the river along the wellhead.

3.2. Concentrations of Potentially Toxic Elements in Soils

Descriptive statistics for As, Cd, Cu, Pb, Ni, Zn, Mn, Cr, Hg, Fe, and S in the 27 soil samples from Youyu Catchment are presented in Table 1 and Figure 2. The background pH and concentrations of potentially toxic elements in Guizhou Province and the Chinese Environmental Quality Standards for Soils [29] are listed together for reference. Potentially toxic elements were found to be widespread in the study area, with concentration ranges of 4.85–12.94, 0.10–2.85, 4.04–167.92, 5.15–43.11, 3.39–77.73, 10.60–264.58, 6.54–3524.02, 56.37–118.25, 0.006–0.015, 3764.24–185,519.41 mg/kg for As, Cd, Cu, Pb, Ni, Zn, Mn, Cr, Hg, and Fe, and 0.003–2.91% for S; the respective mean values were 8.74, 1.12, 62.68, 18.40, 25.95, 72.52, 593.60, 86.06, 0.010, and 35,575.95 mg/kg, and 0.44%. In general, except for Cd, the concentrations of the potentially toxic elements (As, Cu, Pb, Ni, Zn, Cr, and Hg) were all below the Chinese environmental quality standards for soils. For Cd, the mean value was 1.12 mg/kg, which was 2.73 times higher than the Chinese standard (0.30 mg/kg). The As, Cd, Cu, Ni, Zn, Mn, Cr, and Fe concentrations were higher than the soil background

values in Guizhou Province [30], with the Cd, Cu, and Fe concentrations being particularly high at 1.12, 62.68, and 35,575.95 mg/kg, which were 1.69, 1.95, and 12.18 times higher than their respective background values. There are no published records of soil background values for S in Guizhou Province, but when compared to a mountain soil in the adjacent area (0.11%) [31], the mean S content in the study area was 3.4 times higher (0.44%).

There have been several other studies reporting the levels of potentially toxic elements in the soils of the studied area. According to a study conducted in Youyu Catchment in 2014, the Cr, Cu, Zn, As, Cd, and Pb concentrations in the soil surrounding the coal mine were 79.09 ± 16.48 , 87.07 ± 13.28 , 127.5 ± 36.23 , 30.53 ± 10.82 , 0.36 ± 0.15 , and 29.05 ± 9.04 mg/kg, respectively [27]. In a study in Huaxi District conducted in 2010, the Hg, As, Cr, Pb, Cd, Fe, and Mn concentrations were 0.15, 29.60, 62.42, 26.56, 0.61, 31,300, and 437.52 mg/kg, respectively [32]. Gao et al. [15] determined the total Fe concentration in an abandoned coal mine in Youyu Catchment and reported an AMD volume of >7000 m³, and a mean Fe concentration of 20,929.65 mg/kg, which was lower than the value reported in this study. Since 2010, the As, Hg, and Pb concentrations in soils in Youyu Catchment have displayed a downward trend, while the Cd, Fe, Mn, and Cr concentrations have displayed upward trends.

There have been several studies of other high-S coal mines in Guizhou Province [33] determined the As, Cd, Cr, Cu, Pb, and Zn concentrations in the soils of Xingren County, which is located in southwest Guizhou where numerous high-S coal mines are located. The As, Cd, Cr, Cu, Pb, and Zn concentrations were 28.82, 133.75, 1.14, 52.11, 79.63, 44.39, and 118.23 mg/kg, respectively. According to Li [34], at the Zhijin coal mine in western Guizhou Province, the Hg, As, Cu, Zn, Cd, Pb, Cr, Fe, and Mn concentrations in the soil surrounding coal gangues were 0.128, 6.94, 126.49, 85.92, 0.56, 17.62, 157.69, 61,689, and 435.1 mg/kg, respectively. In comparison to these coal-mining areas in Guizhou Province, the concentrations of potentially toxic elements in this study were relatively low. This disparity reflects the variation in potentially toxic element concentrations among different regions in coal-mining areas, as well as the enrichment of potentially toxic elements in soils by anthropogenic activities.

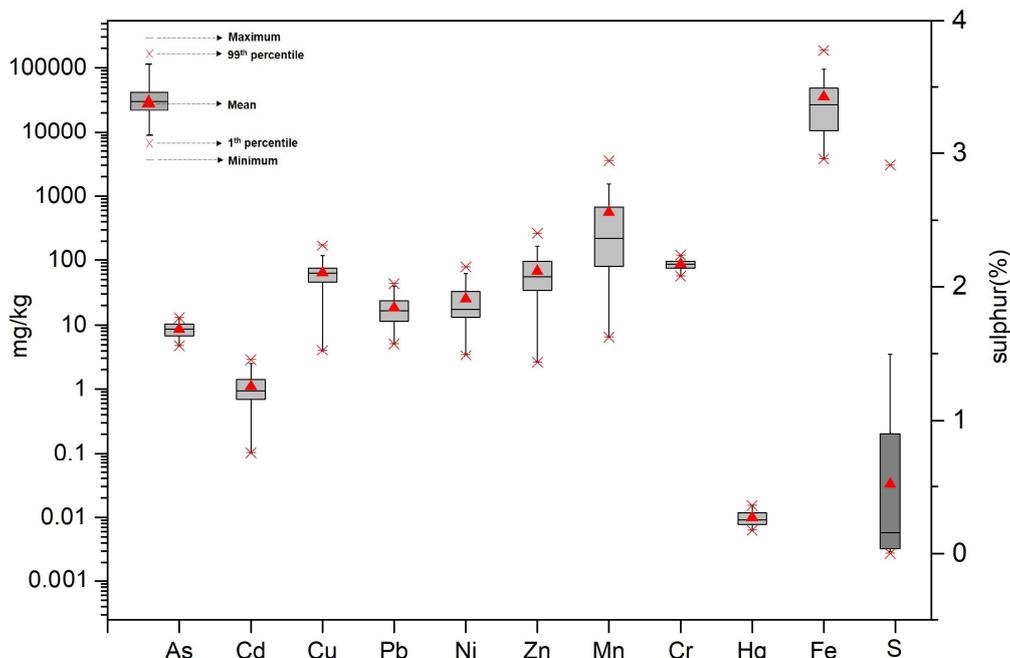


Figure 2. The statistical characteristics (maximum, 99th percentile, median, 1st percentile, and minimum values) of potentially toxic elements in the soil from Youyu Catchment. The concentrations of As, Cd, Cu, Pb, Ni, Zn, Mn, Cr, Hg, Fe refer to the left axis; the percentage of sulfur refers to right axis.

Table 1. The concentrations, coefficient of variation, and geoaccumulation index (Igeo) values of potentially toxic elements in soils of Youyu Catchment.

-	pH	As (mg/kg)	Cd (mg/kg)	Cu (mg/kg)	Pb (mg/kg)	Ni (mg/kg)	Zn (mg/kg)	Mn (mg/kg)	Cr (mg/kg)	Hg (mg/kg)	Fe (mg/kg)	S (%)
Mean	5.44	8.74	1.12	62.68	18.40	25.95	72.52	593.60	86.06	0.010	35,575.95	0.44
SD	2.06	0.70	27.52	9.05	18.88	49.12	775.69	14.30	0.00	33,710.01	0.59	SD
CV	23.59	62.17	43.90	49.19	72.76	72.65	130.68	16.62	24.76	94.76	132.53	CV
Igeo	-	-0.13	0.22	-1.26	-0.79	-1.05	-1.51	-0.11	-2.65	2.47	-	Igeo
CEQS	-	25.00	0.30	200.00	350.00	50.00	300.00	-	250.00	1.00	-	-
Background values in Guizhou province	6.2	8.6	0.21	32.0	26.2	23.4	67.0	482	61.0	0.040	2920	-

CEQS: Chinese Environmental Quality Standards for Soils (CEPA 1995); Background Values in Guizhou Province: CNEMC, 1990.

3.3. Spatial Distribution of Potentially Toxic Elements

In general, the sample types were divided into four classes, to represent the abandoned coal area: (1) soil samples surrounding (within 100 m) the abandoned coal mines (SCG), (2) samples from where AMD emerges from underground (AMD), (3) soil samples from where tributaries merge into the Youyu River (TM), and (4) soil samples from control areas (CK). As shown in Figures 3 and 4, different soil sample sites displayed varying degrees of pollution. Generally, the soil sampled at TM sites had the highest Cd, Cu, Ni, Zn, and Mn concentrations, while the As, Pb, Cr, and Fe concentrations were highest in the soils at AMD sites; the Hg concentrations were similar across the five sampling sites. The soil at SCG sites had the highest S content, and conversely had the lowest levels of all potentially toxic elements, except for Fe. Therefore, Cd and Fe were considered to be the elements of most concern; however, these two elements had different pollution characteristics. The Cd concentrations followed the order of TM > AMD > SCG > CK, while the Fe concentrations followed the order of AMD > SCG > CK > TM.

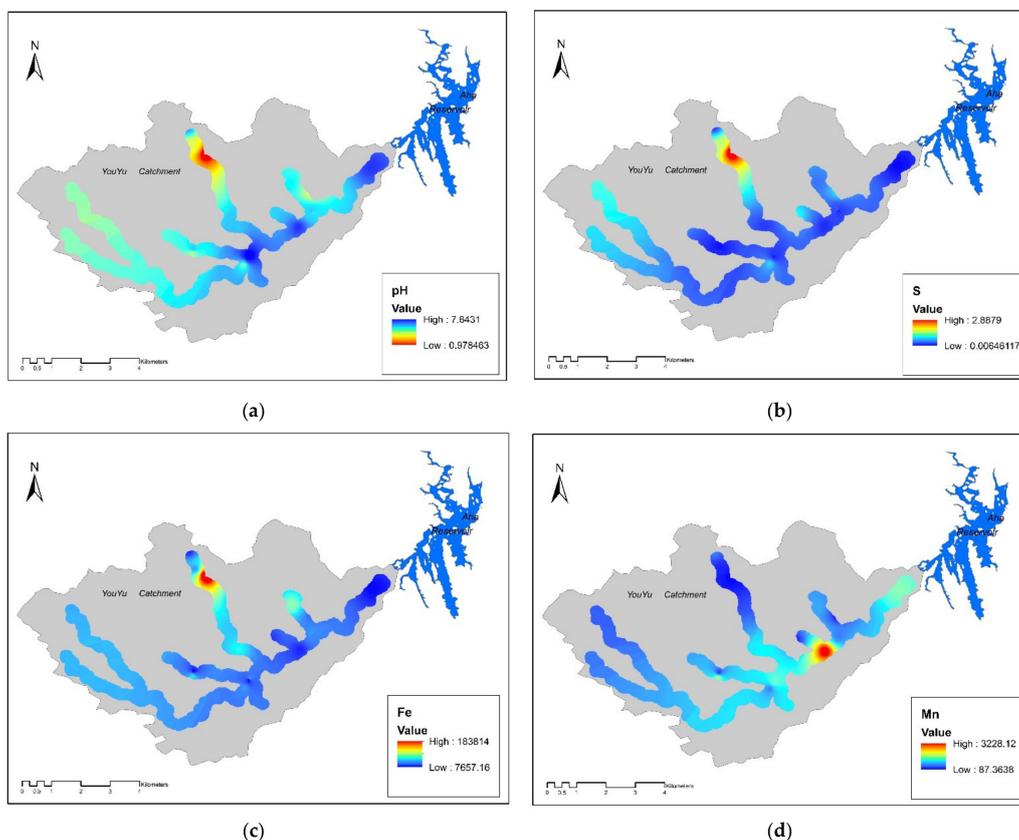


Figure 3. Cont.

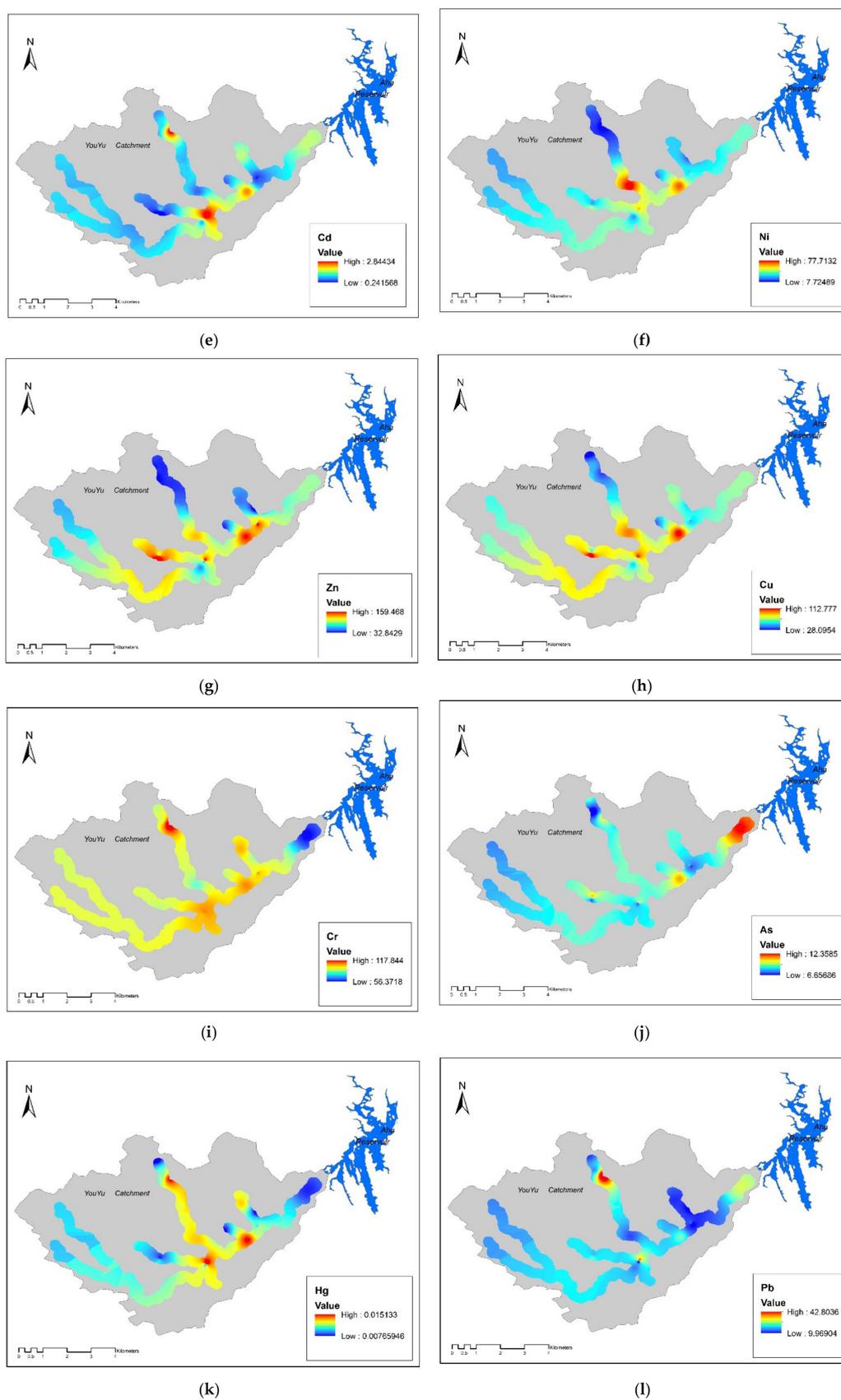


Figure 3. Spatial distribution of the concentrations of potential toxic elements in Youyu catchment. (a) pH value; (b) S value; (c) Fe value; (d) Mn value; (e) Cd value; (f) Ni value; (g) Zn value; (h) Cu value; (i) Cr value; (j) As value; (k) Hg value; (l) Pb value.

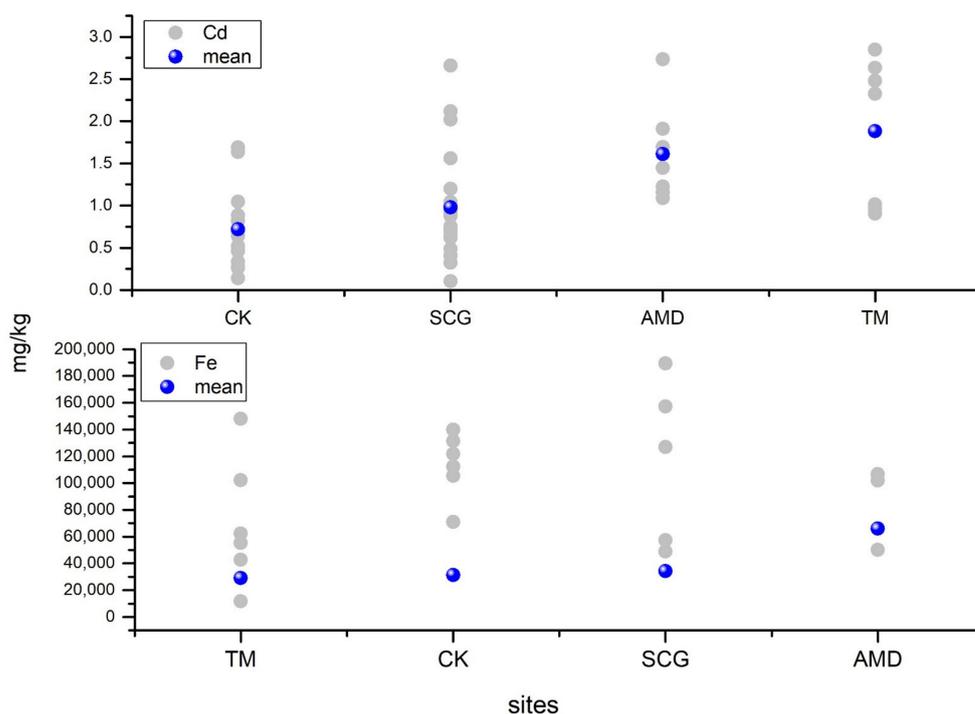


Figure 4. The concentrations of Cd and Fe in the soil from different sites in Youyu Catchment.

3.4. Ecological Risk of Potentially Toxic Elements

The method of Hakanson [23] was adopted to assess the ecological risk posed by heavy metal pollution in the topsoil of the study area. Hakanson developed a quantitative approach that has been used in many other studies, to assess the ecological risk posed by heavy metals in soil [35,36]. Supplementary Materials Table S2 shows the potential ecological risk values, $E(i)$, of eight potentially toxic elements in the topsoil in Youyu Catchment. The mean $E(i)$ values of As, Cr, Cu, Ni, Pb, Zn, and Hg were all less than 40, indicating that these metals posed a low ecological risk. Cadmium had a higher mean $E(i)$ value (51.16) than the other seven heavy metals, thus presenting a moderate potential ecological risk. Overall, 61.8% of all samples had an $E(\text{Cd})$ value in the range of 40 to 80, while 14.5% were in the range of 80 to 160, representing moderate and considerable potential ecological risk, respectively. The RIs were calculated to assess the risk of multiple potential toxic elements in soils. The RI values for 91.0% of all samples were <160 , indicating a low potential ecological risk, while the other 9.0% of samples posed a moderate ecological risk. As shown in Figures 5 and 6, the spatial distribution of $E(\text{Cd})$ is consistent with that of RI, indicating that the potentially ecological risks in the studied area was dominated by Cd.

The oxidation of sulfides releases H^+ , Fe^{2+} , and SO_4^{2-} , which is the first step in the environmental accumulation of potentially toxic elements in the environment. However, the subsequent migration of elements is also controlled by a series of complex precipitation, hydrolysis, adsorption and desorption, co-precipitation, and ion exchange reactions. In general, the highest concentrations of Fe, S and the lowest pH values were all observed in the AMD sites. At the AMD sites, soil Fe and S concentrations had significant negative correlations with soil pH ($P > 0.01$, $\text{Fe}:\text{R} = -0.83$, $\text{S}:\text{R} = -0.93$), suggesting that the Fe and S concentrations were higher at lower pH values. This was the karst landscape facilitated AMD outflows, resulting in massive amounts of Fe^{3+} , H^+ , and SO_4^{2-} being leached into the soil, thus causing significantly high Fe and S concentrations and extremely low pH values in the soil at AMD sites. It was also apparent that the Fe concentration displayed a decreasing trend with distance from AMD and SCG sites, while pH displayed the opposite trend.

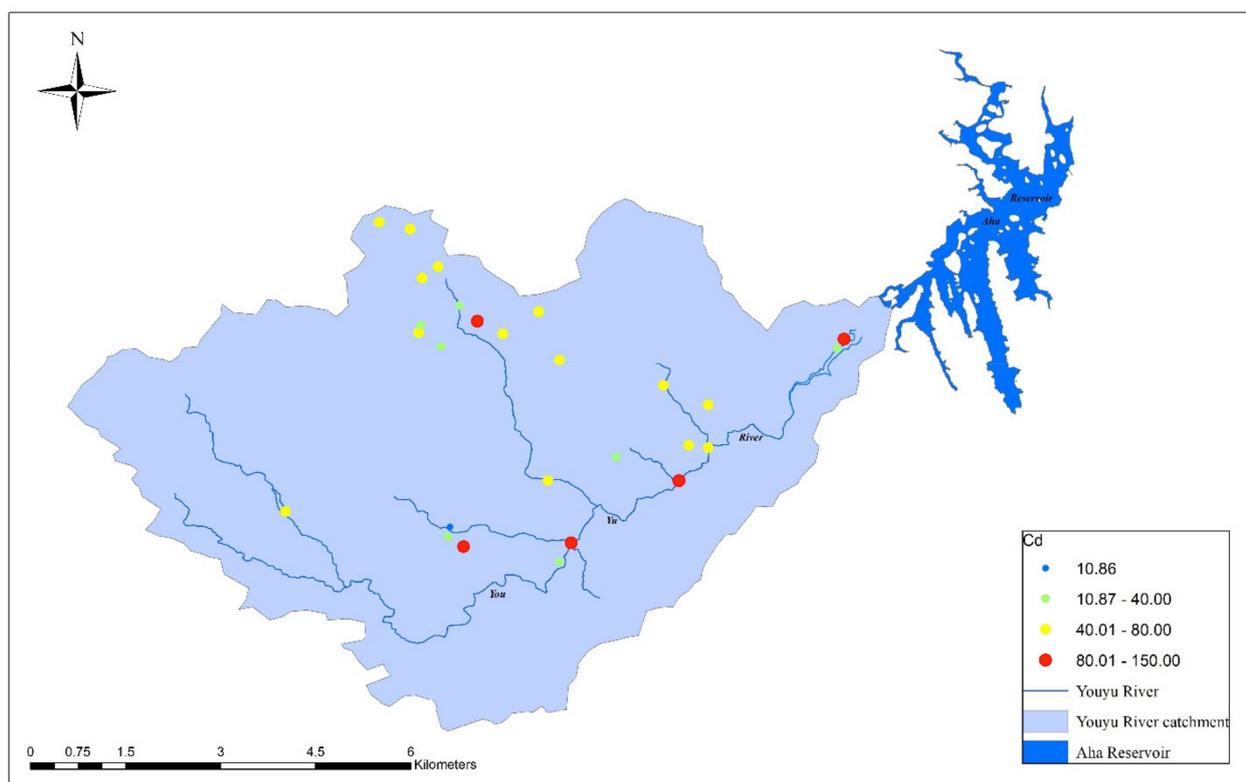


Figure 5. Spatial distribution of the potential ecological risks (E(i)) for Cd in the soil of the studied area.

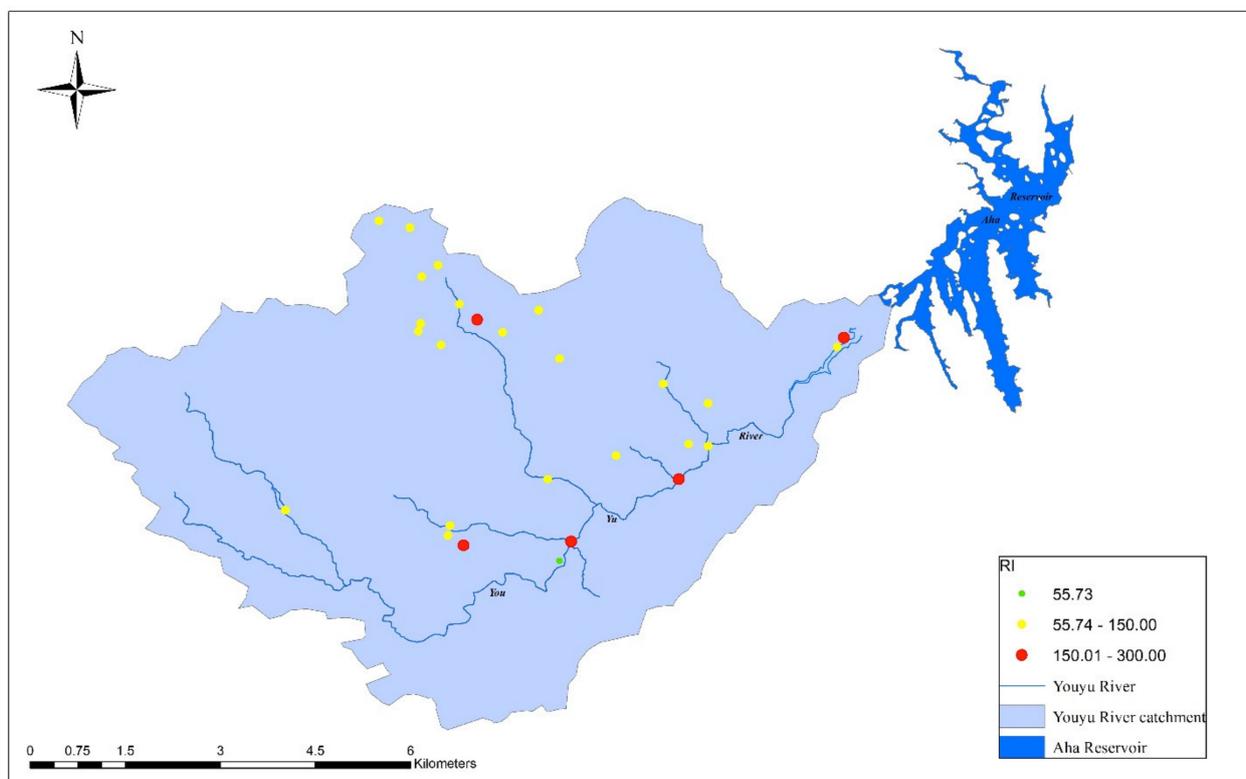


Figure 6. Spatial distribution of the potential ecological risk indices (RIs) in the soil of the studied area.

However, compared to Fe, the Cd concentrations displayed the opposite trend, being highest in the TM sites and thus implying other pollution sources. Based on the field investigation, coal mining at the AMD sites and abandoned coal mines was responsible for the direct introduction of Cd into the surface environment. However, there were also numerous exposed coal seams upstream of TM sites. The local karst area was dominated by a lithologic soil that has been substantially affected by the carbonate rocks. The soil was shallow and neutral to slightly alkaline due to the lack of a transition layer, and the affinity and adhesion between the underlying rock and soil was poor. For a long period of time, the local area has had a tropical and subtropical climate, and has therefore experienced humid and hot conditions with high precipitation, resulting in strong chemical leaching activities. Leaching or weathering of these residues may transfer Cd and other elements downhill to the TM soils, which can then migrate onto farmland and into surface waters. It has been reported that soil Cd is more likely to migrate under acidic conditions ($\text{pH} < 6$) [14,37], but is more likely to deposit as $\text{Cd}(\text{OH})_2$ or other precipitates in an alkaline environment ($\text{pH} > 6$). In contrast, Fe is more inclined to precipitate under acidic conditions [38], which was the case in this study. In the TM sites, the Cd concentration showed a significantly negative correlation ($p < 0.01$, $R = 0.97$) with pH (mean value, 6.76), while in the SCG and AMD sites, which had the highest soil Fe concentrations, there was a significant positive correlation ($p < 0.01$, $R = 0.96$) between the Fe concentration and pH (mean value, 4.35). The above results indicated that with distance from AMD and SCG sites, the pH tended to be higher and soil Cd tended to precipitate, while Fe tended to precipitate under a low soil pH. However, it should be recognized that the migration of various elements in the mining area varied with the chemical properties of solid minerals and solutions. There are likely to be multiple migration mechanisms, which should be studied further in combination with an analysis of the various chemical forms of soil elements. The TM sites showed higher Cd concentrations compared to other sites in the upper reach; as mentioned above, Cd is more likely migrate under acid condition and absorbed into sediment under opposite condition. Yuan et al. [39] reported that, under flood conditions, sediments are easily mixed with water, and potentially toxic elements-rich sediments are enriched in the soil around the river and in the lower-lying terrain. It can be seen that river sediments are not only the enrichment area of potentially toxic elements but also the secondary pollution sources with potential hazards to water quality and the surrounding soil. In addition, coal wastes are often used by local residents to increase the height of riverbanks, and can also be enriched in river sediment. Under flood conditions, the sediments will mix with river water, and the soil around the river and surrounding low-lying terrain can become enriched with sediment [39]. Runoff from tributaries of the Youyu River has decreased annually, especially in the dry season, and the bare riverbed has formed a soil over a long period. The levels of Cd and other potentially toxic elements in soil may also be influenced by agricultural activities, such as irrigation with contaminated water and fertilization. In addition, due to the prohibition of mining activities, tea plantations have been established by local famers in recent years. Phosphate fertilizer is used by local farmers, to improve soil fertility, and has had an important influence on soil quality.

After 2010, the mines in this area were successively closed, and the pits were all sealed. However, AMD formation became more extensive after mine closure, because the pumps used to keep the water table near open-pit and/or underground mines artificially low, to facilitate mining activities, were turned off. This has resulted in a rise in the groundwater level, which then comes into contact with the exposed sulfide minerals on wall rocks, causing AMD to be continuously generated. In addition, even if all the pits were sealed, groundwater would flow in the form of karst gaps, pores, karst caves, pipes, and underground rivers, due to the karst landforms. Rock cutting and fragmentation create favorable conditions for groundwater storage, migration, and dissolution. Surface water and groundwater are exchanged frequently, and karstification is strong. Water usually emerges from the ground in the form of springs, seriously contaminating the surrounding aquatic and soil environment. At the same time, coal gangues from the coal mines have

been randomly deposited in the area, and harmful components are leached out by rainfall leaching, which then pollutes surface water, groundwater, and soil. According to the spatial distribution of potentially ecological risk posed by Cd (Figure 5 and Table 2), the sites where tributaries merged with the main stream of the river had the highest pollution levels. Soil pollutants can enter rivers through leaching, where they may harm aquatic organisms in the wet season, and further endanger human health via the food chain.

Table 2. The potential ecological risks of potentially toxic elements in soils from different sites of Youyu Catchment.

Sampling Sites	As	Cd	Cu	Pb	Ni	Zn	Cr	Hg	RI
TM	10.39	85.60	11.66	3.87	10.71	1.66	3.08	14.50	141.47
AMD	11.27	73.17	8.93	3.61	5.11	0.91	3.26	12.30	118.56
SCG	9.63	44.78	8.95	3.11	3.69	0.69	2.75	9.45	83.06
CK	10.78	32.84	11.29	3.62	5.96	1.35	2.66	9.00	77.50
Mean	10.17	51.16	9.79	3.51	5.55	1.01	2.82	9.87	93.87
Median	10.46	43.84	9.82	3.14	3.91	0.82	2.82	9.12	82.15
Maximum	15.04	129.70	26.24	8.23	16.61	3.95	3.88	15.24	199.51
Minimum	5.64	4.69	0.63	0.98	0.72	0.04	1.85	6.32	38.96
SD	2.40	31.81	4.30	1.73	4.04	0.73	0.47	2.44	36.25

TM: Youyu River; AMD: acid mine drainage; SCG: surrounding (within 100 m) coal gangues; CK: control areas.

Cadmium is considered to be one of the most toxic trace elements in the environment, and can cause serious health problems in animals and humans [40]. Cadmium accumulation in grains and vegetables has received much attention [41]. Cadmium pollution of the edible parts of grains and vegetables is particularly serious in China, due to rapid industrialization and the wide application of agrochemicals in agricultural activities. It has been reported that Cd is the main element responsible for soil heavy-metal contamination in China, and poses the highest potential ecological risk in both rural and urban areas. The highest soil Cd concentrations have been reported in south China, e.g., the Yunnan–Guizhou Plateau. However, there are many other hotspots in China, due to mining and smelting activities, with the soil in mining and smelting areas having much higher Cd concentrations than in areas where other land uses predominate, such as irrigation, urban, and remote areas [42–44].

4. Conclusions

Obvious acidification and enrichment of potentially toxic elements was observed in the soils of Youyu Catchment, especially Cd, S, and Fe. Among all elements in the study area, the highest concentrations were observed for Cd and Fe, with Cd and Fe being present at unpolluted to moderately polluted and heavily polluted levels, respectively. Significantly high Fe and S concentrations and extremely low pH values were found in the soils of AMD sites, and the sites where tributaries merged with the TM had the highest Cd concentrations. Iron contamination originated mainly from the non-point sources of AMD and coal gangues, while the adsorption and desorption of AMD and agricultural activity resulted in high Cd levels in the estuary. Cadmium levels posed a moderate ecological risk, while the other elements, such as As, Cr, Pb, Zn, Cu, Ni, and Hg, all posed a low eco-risk. Spatially, the TM sites had the highest Cd eco-risk; moreover, they tended to pose a significant risk of harm to aquatic ecosystems, particularly in the wet season, and endangered human health via the food chain.

Supplementary Materials: The following are available online at <https://www.mdpi.com/2075-163X/11/3/330/s1>, Table S1: Contamination categories on the basis of Igeo values and ecological risk indices; Table S2: Potentially ecological risks values of the elements of As, Cd, Cu, Pb, Ni, Zn, Cr, and Hg.

Author Contributions: Conceptualization, L.P. and X.G.; methodology, L.P.; software, L.P. and B.L.; validation, L.P. and Y.C.; formal analysis, L.P.; investigation, X.G., B.L., Y.C., Z.H., J.P., Y.P., Y.Z.; resources, L.P.; data curation, L.P.; writing—original draft preparation, L.P. and B.L.; writing—review

and editing, L.P.; visualization, L.P.; supervision, X.G. and L.P.; project administration, Chinese Research Academy of Environmental Sciences; funding acquisition, X.G. All authors have read and agreed to the published version of the manuscript.

Funding: This study was supported by the National Key R&D Plan (2018YFC1801302) and Chinese Research Academy of Environmental Sciences, State Key Laboratory of Environmental Criteria and Risk Assessment (201520 and 201542).

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Acknowledgments: We would like to thank all those who helped with the sampling.

Conflicts of Interest: The authors declare no conflict of interest.

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