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Abstract: Mixed-layer illite-smectite (I-S) from a new set of Pennsylvanian-aged Illinois Basin underclays, identified as paleosols, are investigated to assess the impact of (1) regional diagenesis across the basin and (2) the extent to which ancient environments promoted illitization during episodes of soil formation. Interpretations from Reichweite Ordering and $\Delta^{\circ} 2\theta$ metrics applied to X-ray diffraction patterns suggest that most I-S in Illinois Basin paleosols are likely the product of burial diagenetic processes and not ancient soil formation processes. Acid leaching from abundant coal units and hydrothermal brines are likely diagenetic mechanisms that may have impacted I-S in Pennsylvanian paleosols. These findings also suggest that shallowly buried basins (<3 km) such as the Illinois Basin may still promote clay mineral alteration through illitization pathways if maximum burial occurred in the deep past and remained within the diagenetic window for extended periods of time. More importantly, since many pedogenic clay minerals may have been geochemically reset during illitization, sources of diagenetic alteration in the Illinois Basin should be better understood if Pennsylvanian paleosol minerals are to be utilized for paleoclimate reconstructions.

Keywords: clay mineralogy; Desmoinesian; diagenesis; Missourian; pedogenesis; Reichweite Ordering

1. Introduction

Mixed-layered, interstratified, or interlayed clay minerals possess two or more layer types or mineral components that are vertically stacked along the direction perpendicular to (001) [1,2]. Mixed-layer clay minerals likely represent intermediate products of reactions involving discrete, end-member clay minerals and may be 2:1 or 1:1 and di- or triocta-hedral [3,4]. Although there are many clay mineral transformations [5], smectite to illite (e.g., [6,7]), smectite to chlorite (e.g., [8,9]), serpentine/chlorite to chlorite (e.g., [10,11]), and smectite to glauconite (e.g., [12,13]) are attributed to occur during low-temperature (\geq 50 °C) early diagenesis. The most ubiquitous evidence of these reactions in sedimentary basins is mixed-layered illite-smectite (I-S) [2,3,6,14] and may be used to assess maximum burial and a basin's thermal evolution.

This was demonstrated in studies of pelitic sediments from the U.S. Gulf of Mexico region, which recognized patterns of diagenesis in offshore wells by observing decreasing abundances of potassium-feldspar and smectite and increasing abundances of I-S and discrete illite with increasing burial depth [6,7,15–20]. This pattern was attributed to high temperatures, ≥ 100 °C [21–23], and pressures experienced during deep burial (>3 km) diagenesis, i.e., mesogenesis [24], with available K⁺ sourced from the dissolution of K-feldspars and/or micas [25], a process defined as illitization. Smectite illitization is characterized by the collapse of the smectite 2:1 expansible interlayer into a 12-fold coordination of basal tetrahedral oxygen anions around a K⁺ interlayer cation as other, exchangeable, hydrated cations (e.g., Mg²⁺, Ca²⁺, Na⁺) are expelled from the interlayer to form illite (Figure 1) [26].



Citation: McIntosh, J.A.; Tabor, N.J.; Rosenau, N.A. Mixed-Layer Illite-Smectite in Pennsylvanian-Aged Paleosols: Assessing Sources of Illitization in the Illinois Basin. *Minerals* 2021, *11*, 108. https:// doi.org/10.3390/min11020108

Academic Editor: Javier Arostegi García Received: 20 December 2020 Accepted: 18 January 2021 Published: 22 January 2021

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Figure 1. Model of the stages of smectite illitization. (a) A cross-sectional view of an expandable 2:1 phyllosilicate at the onset of illitization. Hydrous clay minerals, i.e., smectite, readily exchange aqueous cations such as Ca²⁺, Mg²⁺, Na⁺, and K⁺, all in six-fold coordination with water molecules, in the interlayer space at this stage. Due to dehydration and replacement of Si^{4+} by Al^{3+} in the tetrahedral sheet, K⁺ becomes the preferred interlayer cation to balance the change in layer charge. Note that smectite has a layer spacing of 15–18 Å. (b) Illitization progresses during continued K⁺ fixation to the surface of tetrahedral sheet in the interlayer space while exchangeable hydrous cations, Ca²⁺, Mg²⁺, and Na⁺ are expelled from the interlayer. This intermediate stage between discrete smectite and discrete illite is called interstratified or mixed-layer illite-smectite (I-S). Note that the interlayer space begins to decrease, leading to a layer spacing for I-S of 10–15 Å. (c) As illitization is completed, there is a full conversion from smectite to the non-hydrous, 2:1 species illite. Illite is characterized by a total collapse of the interlayer space and removal of cations except for K⁺. Note that illite has a layer spacing of 10 Å. (d) A map view of the illite tetrahedral sheet, equivalent to the final stage of illitization as expressed in part C. K⁺ is the interlayer cation now in 12-fold coordination with oxygen anions that are part of the silicate tetrads. Note that the apical oxygens of the tetrads are pointing into the page, through the field of view, and the basal surface of the tetrad is represented by the 2-D viewing surface.

Although illitization was initially recognized as occurring during deep-burial diagenesis, hereafter referred to as "diagenesis/diagenetic illitization," similar sorts of mineralogical transformations also were subsequently found to occur during low-temperature, shallow-burial, or soil formation, i.e., eogenesis, hereafter referred to as "pedogenesis/pedogenic illitization" [3,7,24,27]. During pedogenesis, illitization is due to fluctuating climatic conditions, characterized by distinct wetting and drying cycles [27]. This alteration primarily occurs in seasonal climates where redox conditions fluctuate between Fe(II) and Fe(III) valence states, wherein reduction of Fe(II) dominates during wet phases and oxidation of Fe(III) dominates during dry phases (Figure 1b–d) [28–33].

X-ray diffraction (XRD) has been used to evaluate illitization, by studying I-S. These studies have found that the proportion of smectite in I-S superstructures is greater if the XRD peak at the I-S (001/002) position is closer to the discrete smectite peak, around 15 Å (5.2° 2θ Cu-K α radiation). Furthermore, as illitization progresses, the XRD I-S (001/002) peak migrates toward smaller dimensions (i.e., higher $^{\circ} 2\theta$ values; [23]). There are two primary XRD pattern analyses that have been proposed as a means of distinguishing between diagenetic and pedogenic I-S: Reichweite ordering and % illite in I-S, derived from $\Delta^{\circ} 2\theta$ measurements. Reichweite Ordering analyses of smectite illitization have determined that ordered mixed layered illite-smectite clay minerals are solely a result of deep-burial diagenesis [7,34,35]. Subsequent studies of illite-smectite occurrences suggest that randomly ordered, mixed-layered illite-smectite minerals may result primarily from pedogenic illitization, though occasionally from diagenetic illitization [27]. If a sample contains 0–60% illite in I-S, the clay is likely the product of pedogenic illitization, promoted through wetting and drying in seasonal climates [27] (see Figure 1 for illitization and I-S schematic). If a sample contains >60% illite in I-S, the clay is considered to likely result only from non-pedogenic illitization or diagenetic overprinting of pedogenic illitization during deep burial diagenesis (mesogenesis [24]) or hydrothermal alteration [1,7,27,36]. In sum, both Reichweite Ordering and % illite in I-S XRD-based proxies may provide genetic origin information for I-S identified in argillaceous rocks.

The Illinois Basin (IB) is known for its characteristic Pennsylvanian cyclothemic stratigraphy [37], which notably contains underclays, identified as fossil soils, hereafter referred to as paleosols, that formed at a tropical paleolatitude ($0^{\circ} \pm 5^{\circ}$; Figure 2 [38]). Attention has been given to IB paleosols following other Pennsylvanian-aged studies, which suggest a shift toward a drier, seasonal climate from the middle to upper Pennsylvanian, defined as the Desmoinesian-Missourian (Moscovian-Kasimovian; Figure 2) in the tropics. Evidence to support aridification in the IB at this time includes floral [39-41] and faunal shifts (e.g., [42]), increased vertic and calcic paleosol features [43,44], and increased paleotemperatures derived from pedogenic mineral proxies [45]. However, [45] (p. 391) note that some IB paleosol phyllosilicate samples show signs of diagenetic overprinting along the western margin and deep interior of the basin, because geochemical compositions of those minerals correspond to crystallization temperatures ranging from $44 - 55 \pm 3$ °C, which is abnormally high to represent typical soil temperatures (e.g., [46]). However, there are little data to provide sufficient insights into the effects of possible diagenesis on non-coal bearing Pennsylvanian-aged strata, specifically paleosols, across the entire basin. Therefore, before continuing with further geochemical, proxy-based research on Pennsylvanian paleoclimate, a rigorous assessment of the impact of diagenesis on IB paleosols should be addressed on a basin-wide scale.

This study attempts to identify mixed-layer illite-smectite within the clay-size fractions of the IB paleosol profiles to differentiate between pedogenic, or authigenic products of soil weathering, and diagenetic, or authigenic products of post-pedogenic alteration, clay minerals. This is achieved by XRD determination of Reichweite Ordering and $\Delta^{\circ} 2\theta$ illitization values and consideration of the impact of possible diagenetic mechanisms in the IB, which may inform the significance and meaning of the paleosol mineral fraction for the purposes of paleoclimatic and diagenetic reconstruction of the Pennsylvanian strata of the IB.

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6 46 Desmoinesian Carbondale gleyed calcic Vertisol (D) 97.8 1 4 Missourian Bond gleyed calcic Vertisol (D) 20.2 3 7 Missourian Bond gleyed Vertisol (D) 20.5 4 11 Missourian Patoka gleyed Vertisol (C) 73 4 9 Missourian Patoka gleyed Vertisol (C) 73.6 6 16 Missourian Patoka gleyed Protosol (B) 171.9 10 27 Desmoinesian Shelburn gleyed Protosol (B) 220.1 13 36 Desmoinesian Carbondale gleyed Protosol (B) 220.1 14 40 Desmoinesian Carbondale gleyed calcic Vertisol (D) 244.7 14 40 Desmoinesian Carbondale gleyed calcic Vertisol (D) 245.7 2 7 Virgilian Mattoon gleyed vertisol (C) 20.4 3 10 Virgilian Mattoon gleyed vertisol (C		6	49	Desmoinesian	Carbondale	gleyed calcic Vertisol (D)	97.2
1 4 Missourian Bond gleyed claic Vertisol (D) 20.2 3 7 Missourian Bond gleyed Protosol (B) 41.6 4 11 Missourian Patoka gleyed Vertisol (C) 73.6 4 9 Missourian Patoka gleyed Vertisol (C) 73.6 6 16 Missourian Patoka gleyed Protosol (B) 138.1 10 27 Desmoinesian Shelburn gleyed Protosol (B) 120.2 13 38 Desmoinesian Carbondale gleyed Protosol (B) 220.1 13 38 Desmoinesian Carbondale gleyed Protosol (B) 220.1 14 40 Desmoinesian Carbondale gleyed Protosol (B) 220.4 2 7 Virgilian Mattoon gleyed Protosol (B) 220.1 2 7 Virgilian Mattoon gleyed Protosol (B) 245.7 1 3 10 Virgilian Mattoon gleyed Protosol (B)		6	46	Desmoinesian	Carbondale	gleyed calcic Vertisol (D)	97.8
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HAM e616Missourian Desmoinesian Shelburn gleyed Protosol (B)138.1 171.91027Desmoinesian BesmoinesianShelburn gleyed Protosol (B)171.91233Desmoinesian Shelburn gleyed Protosol (B)1961338Desmoinesian Carbondale gleyed Protosol (B)220.11336Desmoinesian Carbondale gleyed Protosol (B)220.11443Desmoinesian Carbondale gleyed clcic Vertisol (D)244.71440Desmoinesian Carbondale gleyed clcic Vertisol (D)6027Virgilian WingilianMattoon Mattoon gleyed Vertisol (C)70.2310Virgilian Wissourian Mattoon gleyed Vertisol (C)80.6417Missourian Missourian Bond gleyed Protosol (B)147.8521Missourian Missourian Bond gleyed Protosol (B)171.8840Missourian Missourian Bond gleyed Protosol (B)203.1837Missourian Missourian Bond gleyed Protosol (B)214.51049Missourian Missourian Bond gleyed Vertisol (C)224.9277Missourian Missourian Bond gleyed Vertisol (C)224.1945Missourian Missourian Bond gleyed Protosol (B)214.51155Missourian Missourian Bond gleyed Vertisol (C)224.9277Missourian Missourian Bond gleyed Vertisol (D)225.6		4	9	Missourian	Patoka	gleyed Vertisol (C)	73.6
1100719Desmonnesian DesmonnesianShelburn Shelburngleyed Protosol (B)171.91233Desmoinesian StelburnShelburn gleyed Protosol (B)202.21338Desmoinesian DesmoinesianCarbondale gleyed Protosol (B)220.11336Desmoinesian DesmoinesianCarbondale gleyed Protosol (B)220.41443Desmoinesian DesmoinesianCarbondale gleyed calcic Vertisol (D)244.71440Desmoinesian CarbondaleCarbondale gleyed calcic Vertisol (D)6027Virgilian Missourian Mattoon Gleyed Vertisol (C)80.6310Virgilian Missourian 	нлме	6	16	Missourian	Patoka	gleyed Protosol (B)	138.1
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	HAM	7	19	Desmoinesian	Shelburn	gleyed Protosol (B)	171.9
		10	27	Desmoinesian	Shelburn	gleyed Protosol (D)	196
1536Destinuitesian DestinuitesianCarbondale Carbondalegleyed Protosol (B) gleyed calcic Vertisol (D)240.11443DesmoinesianCarbondale gleyed calcic Vertisol (D)244.71440DesmoinesianCarbondale gleyed calcic Vertisol (D)245.713VirgilianMattoon gleyed calcic Vertisol (D)6027VirgilianMattoon gleyed Vertisol (C)80.6310VirgilianMattoon gleyed Vertisol (C)80.6417MissourianMattoon gleyed Vertisol (D)158.7626MissourianBond gleyed vertic Calcisol (E)158.7626MissourianBond gleyed calcic Vertisol (D)203.1837MissourianBond gleyed calcic Vertisol (D)203.1837MissourianBond gleyed vertic Calcisol (E)215.3945MissourianBond gleyed Vertisol (C)224.91155MissourianPatoka gleyed Vertisol (C)225.61257MissourianPatoka gleyed Vertisol (C)225.21364DesmoinesianPatoka gleyed calcic Vertisol (D)235.61360DesmoinesianPatoka gleyed Vertisol (C)259.81473DesmoinesianPatoka gleyed Vertisol (C)259.81473DesmoinesianShelburn gleyed Vertisol (C)259.21683DesmoinesianCar		12	33 38	Desmoinesian	Carbondalo	gleyed Vertisol (C)	202.2
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		13	36	Desmoinesian	Carbondale	gleved Protosol (B)	220.1
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		13	43	Desmoinesian	Carbondale	gleved calcic Vertisol (D)	244.7
13VirgilianMattoongleyed calcic Vertisol (D)6027VirgilianMattooncalcic Vertisol (G)70.2310VirgilianMattoongleyed Vertisol (C)80.6417MissourianMattoongleyed Protosol (B)147.8521MissourianMattoongleyed vertic Calcisol (E)158.7626MissourianBondgleyed Protosol (B)171.8840MissourianBondgleyed Protosol (B)171.8837MissourianBondgleyed Protosol (B)203.1945MissourianBondgleyed Vertisol (D)204.1945MissourianBondgleyed Vertisol (C)224.91155MissourianPatokagleyed Vertisol (C)224.91151MissourianPatokagleyed Vertisol (D)258.61257MissourianPatokagleyed calcic Vertisol (D)258.61362DesmoinesianPatokagleyed Vertisol (D)259.21360DesmoinesianShelburngleyed Vertisol (C)225.61473DesmoinesianShelburngleyed Vertisol (C)259.21683DesmoinesianShelburngleyed Vertisol (C)259.21683DesmoinesianShelburngleyed Vertisol (C)295.917a98DesmoinesianShelburngleyed Vertisol (C)2		14	40	Desmoinesian	Carbondale	gleyed calcic Vertisol (D)	245.7
27VirgilianMattooncalcic Vertisol (G)70.2310VirgilianMattoongleyed Vertisol (C)80.6417MissourianMattoongleyed Vertisol (C)80.6521MissourianMattoongleyed vertic Calcisol (E)158.7626MissourianBondgleyed vertic Calcisol (D)160.6730MissourianBondgleyed calcic Vertisol (D)203.1840MissourianBondgleyed calcic Vertisol (D)203.1945MissourianBondgleyed vertic Calcisol (E)214.51049MissourianBondgleyed vertic Calcisol (E)215.31155MissourianPatokagleyed Vertisol (C)225.61257MissourianPatokagleyed calcic Vertisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed vertisol (C)259.81473DesmoinesianShelburngleyed Vertisol (C)264.41582DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.917a98DesmoinesianShelburngleyed Vertisol (C)295.917a98DesmoinesianCarbondalegleyed Vertisol (C)327.51683Desmoinesian<		1	3	Virgilian	Mattoon	gleved calcic Vertisol (D)	60
310VirgilianMattoongleyed Vertisol (C)80.6417MissourianMattoongleyed Protosol (B)147.8521MissourianMattoongleyed vertic Calcisol (E)158.7626MissourianBondgleyed calcic Vertisol (D)160.6730MissourianBondgleyed calcic Vertisol (D)203.1840MissourianBondgleyed calcic Vertisol (D)204.1945MissourianBondgleyed Protosol (B)214.51049MissourianBondgleyed vertic Calcisol (E)215.31155MissourianPatokagleyed Vertisol (C)224.92HA f1151MissourianPatokagleyed vertisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed Vertisol (D)259.21360DesmoinesianPatokagleyed Vertisol (C)259.81473DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.217a98DesmoinesianCarbondalegleyed Vertisol (C)326.117b93DesmoinesianCarbondalegleyed Vertisol (C)326.117b94Desmoines		2	7	Virgilian	Mattoon	calcic Vertisol (G)	70.2
417MissourianMattoongleyed Protosol (B)147.8521MissourianMattoongleyed vertic Calcisol (E)158.7626MissourianBondgleyed vertic Calcisol (D)160.6730MissourianBondgleyed Protosol (B)171.8840MissourianBondgleyed calcic Vertisol (D)203.1837MissourianBondgleyed calcic Vertisol (D)204.1945MissourianBondgleyed Protosol (B)214.51049MissourianBondgleyed Vertic Calcisol (E)215.31155MissourianPatokagleyed Vertisol (C)224.91151MissourianPatokagleyed calcic Vertisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed calcic Vertisol (D)259.21473DesmoinesianShelburngleyed Vertisol (C)244.41582DesmoinesianShelburngleyed Vertisol (C)254.41686DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianCarbondalegleyed Vertisol (C)327.517a98DesmoinesianCarbondalegleyed Vertisol (C)325.117b90Desmoines		3	10	Virgilian	Mattoon	gleyed Vertisol (Ć)	80.6
521MissourianMattoongleyed vertic Calcisol (E)158.7626MissourianBondgleyed calcic Vertisol (D)160.6730MissourianBondgleyed Protosol (B)171.8840MissourianBondgleyed calcic Vertisol (D)203.1837MissourianBondgleyed calcic Vertisol (D)204.1945MissourianBondgleyed vertic Calcisol (E)215.31049MissourianBondgleyed Vertisol (C)224.91155MissourianPatokagleyed Vertisol (C)224.91151MissourianPatokagleyed vertic Calcisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)258.61362DesmoinesianPatokagleyed Vertisol (D)259.21360DesmoinesianPatokagleyed Vertisol (D)259.21473DesmoinesianPatokagleyed Vertisol (D)259.21686DesmoinesianShelburngleyed Vertisol (C)246.41582DesmoinesianShelburngleyed Vertisol (D)259.21683DesmoinesianShelburngleyed Vertisol (D)326.117b93DesmoinesianShelburngleyed Vertisol (D)326.117b93DesmoinesianCarbondalegleyed Vertisol (C)327.517b90Desmoinesian		4	17	Missourian	Mattoon	gleyed Protosol (B)	147.8
626MissourianBondgleyed calcic Vertisol (D)160.6730MissourianBondgleyed Protosol (B)171.8840MissourianBondgleyed calcic Vertisol (D)203.1837MissourianBondgleyed calcic Vertisol (D)204.1945MissourianBondgleyed vertic Calcisol (E)214.51049MissourianBondgleyed vertic Calcisol (E)224.91155MissourianPatokagleyed Vertisol (C)224.91151MissourianPatokagleyed Vertisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed Protosol (B)271.51686DesmoinesianShelburngleyed Vertisol (C)264.41582DesmoinesianShelburngleyed Vertisol (C)295.91683DesmoinesianShelburngleyed Vertisol (C)295.917a98DesmoinesianCarbondalegleyed Vertisol (C)326.117b90DesmoinesianCarbondalegleyed Vertisol (C)327.517b90DesmoinesianCarbondalegleyed Vertisol (C)334.519102DesmoinesianCarbondalegleyed Vertisol (C)334.5		5	21	Missourian	Mattoon	gleyed vertic Calcisol (E)	158.7
730MissourianBondgleyed Protosol (B)1/1.8840MissourianBondgleyed calcic Vertisol (D)203.1837MissourianBondgleyed calcic Vertisol (D)204.1945MissourianBondgleyed vertic Calcisol (E)214.51049MissourianBondgleyed vertic Calcisol (E)215.31155MissourianPatokagleyed Vertisol (C)225.61257MissourianPatokagleyed calcic Vertisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed vertisol (C)259.81473DesmoinesianShelburngleyed Vertisol (C)264.41582DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.217a98DesmoinesianCarbondalegleyed Vertisol (C)327.517b90DesmoinesianCarbondalegleyed Vertisol (C)327.517b90DesmoinesianCarbondalegleyed Vertisol (C)328.418100DesmoinesianCarbondalegleyed Vertisol (C)334.519102DesmoinesianCarbondalegleyed Vertisol (D)336.7		6	26	Missourian	Bond	gleyed calcic Vertisol (D)	160.6
840MissourianBondgleyed calcic Vertisol (D)203.1837MissourianBondgleyed calcic Vertisol (D)204.1945MissourianBondgleyed Protosol (B)214.51049MissourianBondgleyed Vertic Calcisol (E)215.31155MissourianPatokagleyed Vertisol (C)224.9CHA f1151MissourianPatokagleyed Vertisol (C)225.61257MissourianPatokagleyed calcic Vertisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed calcic Vertisol (D)259.81473DesmoinesianShelburngleyed Vertisol (C)264.41582DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.917a98DesmoinesianCarbondalegleyed Vertisol (C)327.517b90DesmoinesianCarbondalegleyed Vertisol (C)327.518100DesmoinesianCarbondalegleyed Vertisol (C)334.519102DesmoinesianCarbondalegleyed Vertisol (C)336.7		7	30	Missourian	Bond	gleyed Protosol (B)	171.8
o557MissourianBondgleyed Calcic Vertisol (D)204.1945MissourianBondgleyed Protosol (B)214.51049MissourianBondgleyed vertic Calcisol (E)215.31155MissourianPatokagleyed Vertisol (C)224.9CHA f1151MissourianPatokagleyed Vertisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)258.61362DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed vertisol (C)264.41582DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.217a98DesmoinesianShelburngleyed Vertisol (C)295.917a93DesmoinesianCarbondalegleyed Vertisol (C)326.117b90DesmoinesianCarbondalegleyed Vertisol (C)328.418100DesmoinesianCarbondalegleyed Vertisol (C)334.519102DesmoinesianCarbondalegleyed Vertisol (C)336.7		8	40	Missourian	Bond	gleyed calcic Vertisol (D)	203.1
743Missourianbondgleyed retrictorsol(b)214.31049MissourianBondgleyed vertic Calcisol (E)215.31155MissourianPatokagleyed Vertisol (C)224.9CHA f1151MissourianPatokagleyed Vertisol (D)237.51257MissourianPatokagleyed calcic Vertisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)259.21362DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed vertisol (C)264.41582DesmoinesianShelburngleyed Vertisol (C)295.21686DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.917a98DesmoinesianCarbondalegleyed Vertisol (C)326.117b90DesmoinesianCarbondalegleyed Vertisol (C)328.418100DesmoinesianCarbondalegleyed Vertisol (C)334.519102DesmoinesianCarbondalegleyed Vertisol (C)336.7		0	57	Missourian	Bond	gleved Calcic Vertisol (D)	204.1
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CHA f1151MissourianPatokagleyed Vertisol (C)225.61257MissourianPatokagleyed calcic Vertisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)258.61362DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed calcic Vertisol (D)259.81473DesmoinesianShelburngleyed Vertisol (C)264.41582DesmoinesianShelburngleyed Vertisol (C)295.21686DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.917a98DesmoinesianCarbondalegleyed Vertisol (D)326.117b93DesmoinesianCarbondalegleyed Vertisol (C)327.518100DesmoinesianCarbondalegleyed Vertisol (C)328.419102DesmoinesianCarbondalegleyed Vertisol (C)336.7		10	55	Missourian	Patoka	gleved Verticol (C)	213.5
111257MissourianPatokagleyed calcic Vertisol (D)237.51364DesmoinesianPatokagleyed calcic Vertisol (D)258.61362DesmoinesianPatokagleyed calcic Vertisol (D)259.21360DesmoinesianPatokagleyed calcic Vertisol (D)259.81473DesmoinesianShelburngleyed Vertisol (C)264.41582DesmoinesianShelburngleyed Vertisol (B)271.51686DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.917a98DesmoinesianCarbondalegleyed Vertisol (D)326.117b93DesmoinesianCarbondalegleyed Vertisol (C)327.517b90DesmoinesianCarbondalegleyed Vertisol (C)328.418100DesmoinesianCarbondalegleyed Vertisol (C)334.519102DesmoinesianCarbondalegleyed calcic Vertisol (D)336.7	CHA ^f	11	51	Missourian	Patoka	gleved Vertisol (C)	225.6
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1360DesmoinesianPatokagleyed calcic Vertisol (D)259.81473DesmoinesianShelburngleyed Vertisol (C)264.41582DesmoinesianShelburngleyed Protosol (B)271.51686DesmoinesianShelburngleyed Vertisol (C)295.21683DesmoinesianShelburngleyed Vertisol (C)295.917a98DesmoinesianCarbondalegleyed Vertisol (D)326.117b93DesmoinesianCarbondalegleyed Vertisol (C)327.517b90DesmoinesianCarbondalegleyed Vertisol (C)328.418100DesmoinesianCarbondalegleyed Vertisol (C)334.519102DesmoinesianCarbondalegleyed calcic Vertisol (D)336.7		13	62	Desmoinesian	Patoka	gleyed calcic Vertisol (D)	259.2
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19 102 Desmoinesian Carbondale gleyed calcic Vertisol (D) 336.7		18	100	Desmoinesian	Carbondale	gleyed Vertisol (C)	320.4
		19	102	Desmoinesian	Carbondale	gleyed calcic Vertisol (D)	336.7

Table 1. Sample details from Illinois Basin paleosols.

^a For more details on paleosol identification see [50] (p.79). For more details on Pennsylvanian-aged Illinois Basin pedotype scheme see [43] (p. 610–622). ^b ADM = Illinois State Geological Survey (ISGS)-Archer-Daniels-Midland, Borehole MMV-04B core. ^c VERM = the Materials Service Corps #F-72-8 core. ^d FC = flint clay or a kaolinite rich claystone. ^e HAM = the American Coal Company Borehole 7510-20 core. ^f CHA = ISGS #1 City of Charleston core.



Figure 2. (a) Stratigraphic position of cores and paleosols identified in the Illinois Basin (IB) for this study. The cores include the Illinois State Geological Survey (ISGS) #1 City of Charleston (CHA), ISGS-Archer-Daniels-Midland, Borehole MMV-04B (ADM), the Materials Service Corps #F-72-8 (VERM), and the American Coal Company Borehole 7510-20 (HAM). P(#) notation indicates the paleosol identification number collected from the corresponding core, as well as its relative stratigraphic position (see Table 1). P(#)s denoted in black are depicted in Figure 2c herein. TD = total depth of the core from the surface, in meters. The Chapel Coal (dashed horizontal line) is delineated for reference. IUGS = nomenclature for geologic time terms reported by the International Union of Geological Sciences. North Am. = nomenclature for geologic time terms utilized in North America. (b) Map of the IB showing geographic locations of cores sampled for this study. Core locations represented by colored circles are the focus of methods, results, and discussion for this work. Core locations represented by colored triangles are sites studied in [43] (p. 607) that have complementary $\Delta^{\circ} 2\theta$ data incorporated into the discussion of this work (data on Reichweite Ordering from the Lone Star Cement Company #TH-1 (LSC) and Monterey Coal

Company Mac 1 #CBM4 (MAC) cores and are not included in this paper). Map modified after [43] (p. 607). Paleolatitudinal information from [38] (p. 33–39). Mineral districts and Reelfoot-Rift-Rough Creek Graben depiction modelled after [47] (p. 258). (c) Core photos with horizon identifiers based on modern soil classification of soil horizons applied to paleosols [48,49]. All core boxes are 2 feet or ~0.6 m long. Paleosols are named after the IB pedotype scheme developed by [43] (p. 610–622). (i) Paleosol 4 from the VERM core classified as a gleyed calcic Vertisol due to the presence of wedge shaped peds, slickensides, carbonate nodules, mottling, and Fe- and/or Mn- oxide minerals (in order of descending importance). (ii) Paleosol 5 from the CHA core, classified as a gleyed vertic Calcisol due to the presence of pedogenic carbonates, wedge shaped peds, slickensides, and drab matrix colors. Paleosol characteristics are not discussed further in this paper, see [50] (p.79) for corresponding classifications for this sample set.

2. Materials and Methods

2.1. Samples

Bulk paleosol matrix samples (n = 140) were collected from four cores in the IB (Figure 2). Cores include Illinois State Geological Survey (ISGS) #1 City of Charleston (CHA), ISGS-Archer-Daniels-Midland, Borehole MMV-04B (ADM), the Materials Service Corps #F-72-8 (VERM), and the American Coal Company Borehole 7510-20 (HAM). All cores were previously logged using sedimentological and stratigraphic frameworks employed by ISGS geologists that provides basin-wide correlations with other cores. Once paleosol profiles were identified and divided into horizons (sensu [48]), matrix samples were taken in 10-20 cm intervals. Paleosols were classified based on the [49] (p. 130–133) paleosol classification scheme. Note paleosol identifications are listed for reference in Table 1 but will not be discussed in detail here (Figure 2c; see [50] (p. 79) for more information on paleosol classification associated with this samples set). It should be noted that coals, or Histosols [49] (p. 131), were not sampled for this study, but their stratigraphic position was noted. Furthermore, the Shelburn Formation (Fm.) in the VERM core was removed from the ISGS Core Repository and therefore could not be sampled for this study. Once samples were identified, cores were correlated using laterally extensive overlying coal and limestone units, or members, to evaluate lateral differences in penecontemporaneous rock units near the Desmoinesian-Missourian Boundary (Figure 2).

2.2. X-ray Diffraction Analyses

Sixty-four bulk matrix samples were individually suspended in deionized water and disaggregated by aqueous ultrasonic agitation. Samples were isolated to the <2.0 μ m equivalent spherical diameter size fraction by centrifugation [51]. Sixteen of these 64 samples were subject to further centrifugation to isolate the <0.2 μ m equivalent spherical diameter clay fraction, which may contain primarily authigenic clay minerals (e.g., [45,52]). The <2.0 and <0.2 μ m fractions were suspended in DI water and prepared as oriented aggregates on filter membrane peels and transferred to cleaned glass slides [53]. Samples were subject to ethylene glycol (EG) solvation at 60 °C for 8 h, enhancing peak intensity of mixed-layer clay minerals that otherwise could be indistinguishable from background or interference with neighboring peaks [1,54,55]. Step-scan analyses of the samples were performed using a Rigaku Ultima III X-ray Diffractometer at Southern Methodist University with Cu-K α radiation over a range of 2 to 30° 2 θ , and a step scan size of 0.04° 2 θ per 1 s.

2.3. Reichweite Ordering

Reichweite Ordering was developed to identify ordered versus randomly ordered clay minerals [56]. Ordering in this particular mineralogical sense is dependent on the number of interactions between neighboring layers in a clay mineral and whether they show a pattern that permits X-ray data to predict a repetitive relationship(s), or interaction, of crystallographic form across the d_{001} axis [2]. Randomly ordered interstratifications (R0) show no interaction between layers. Based upon observation, ordered interstratifications are ordered by nearest-neighbor (R1) interlayers and thrice nearest neighbors (R3). The identifiable difference between R1 and R3 ordering is the amount of illite in the sample,

which tends to increase with increasing ordering value (i.e., R3 ordered I-S has more % illite in mixed-layer illite-smectite than R1 ordered I-S or R0 randomly ordered I-S [1,25,36]). Ordering can be identified by examining X-ray diffraction patterns, generated by Cu-Ka radiation, between 5 and 9° 2 θ . If there is an X-ray diffraction pattern from 2:1 phyllosilicate mineral with a peak closer to 5° 2 θ , the mineral in the sample likely is randomly interstratified I-S. If a similar sample exhibits an XRD pattern with a peak closer to 9° 2 θ , the sample is considered to be ordered and there is a predictable, repeating, long-distance pattern of layering of illite-type and smectite-type layers in the sample. The intensity of a peak is higher for R3 ordered minerals due to overall greater crystallinity and long-range order, than R1 ordered minerals. It should be noted that when identifying peaks from an XRD pattern, an error of $\pm 0.02-0.08^{\circ}$ 2 θ should be expected [57].

2.4. Δ° 2θ

X-ray diffraction $\Delta^{\circ} 2\theta$ measurement of phyllosilicate minerals is used to identify interstratified clay mineral thicknesses of samples solvated by EG [57]. The following equation is used to estimate the $\Delta^{\circ} 2\theta$ value of EG pretreated, mixed-layer illite-smectite samples:

$$\Delta^{\circ} 2\theta = [\text{illite (002)/smectite (003)}] - [\text{illite (001)/smectite (002)}]$$
(1)

where illite (002)/smectite (003) represents the XRD pattern peak position of the composite interstratified I-S (002) phase of illite and (003) phase of smectite, respectively; both of which occur near 16° 2 θ (using Cu-K α radiation [57]). Illite (001)-smectite (002) represents the XRDpattern peak position of the composite interstratified I-S (001) phase of illite and the (002) phase of smectite, respectively; both of which occur near 9° 2 θ (using Cu-K α radiation [57]). As mentioned, peak positions in an XRD pattern have an error of $\pm 0.02-0.08^{\circ}$ 2 θ , which contributes to a variable uncertainty of the reported Δ° 2 θ value [57]. This Δ° 2 θ EG solvation method allows for the requisite sensitivity of XRD peak positions that is necessary in order to assess compositional changes between illite-smectite superstructure minerals. This reflects the fact that the Δ° 2 θ EG solvation methods cause all X-ray diffraction peaks to be displaced in the same direction, thus not impacting the measured Δ° 2 θ value [1]. Ultimately, the calculated Δ° 2 θ data can be used to estimate the percent illite in I-S [1].

3. Results

Sixty-four samples were chosen for isolation of the <2.0 μ m fraction prior to XRD analysis from among the 140 samples collected from VERM, ADM, CHA, and HAM (Figure 2) paleosol profiles in order to better understand trends in I-S interlayering across the IB. Of those 64 samples selected, eight samples were removed from further analysis because they lacked a sufficient clay-sized fraction of materials for analysis, leaving 56 total samples (Table 2). Of those 56, 3 samples exhibited no XRD detectable I-S, leaving a total of 53 samples for an assessment of I-S of the <2.0 μ m fraction from amongst the four cores from the IB (Table 2). Sixteen samples were chosen for isolation of the <0.2 μ m fraction prior to XRD analysis. Of those 16, three samples exhibited no XRD detectable I-S, leaving a total of 13 samples for assessment of the <0.2 μ m fraction from amongst the four cores in the IB (Table 3).

3.1. Clay Mineralogy

XRD peaks of interest for this work are located near 15 Å (5° 2 θ Cu-K α) for discrete smectite (001), 10 Å (9° 2 θ) for discrete illite (001) or mica, 17.5 Å (9.8° 2 θ) for I-S (001/002), 5.2 Å (17.5°2 θ) for I-S (002/003), and 5 Å (17.8° 2 θ) for discrete illite (002) or mica (Figures 3–7) [1,58]. I-S may replace traditional discrete smectite peaks in low-angle reflections, i.e., ~5–8° 2 θ (Figures 3–7). Peaks associated with chlorite (14.2, 7.1, 4.74, and 3.53 Å), kaolinite (7.16 and 3.18 Å), and quartz (4.26 and 3.35 Å) are also distinguishable in XRD patterns of the <2.0 µm fractions (Figures 3–6) and the <0.2 µm fractions though to a lesser extent (Figure 7).

Core	Paleosol (P)#	Sample ID	I-S (002/003) (° 2θ) ^a	I-S (001/002) (° 2θ) ^a	$\Delta^{\circ} 2\theta^{b}$	% Illite in I-S (\pm 3%)	Reichweite Order
	1	6	16.72	9.56	7.16	63%	1
	2	18	16.32	9.64	6.68	53%	0
	3	25	16.04	9.84	6.2	42%	0
	3	22	16.6	9.48	7.12	63%	1
ADM	3	21	16.84	9.52	7.32	68%	1
	4	32	16.9	9.3	7.6	74%	1
	4	29	16.68	9.36	7.32	68%	1
	4	27	16.84	9.56	7.28	67% 50%	1
	5	22	16.04	10.04	0.0	5270 629/	1
	1		10.92	9.0	7.12	710/	1
		2	16.76	9.32	7.44 NJ (A	/1%	l N () d
	FC	0	IN/A 16.6	N/A	$\frac{N}{A}$		N/A ^u
	2	13	16.8	9.4	7.2	67%	1
	2	7	17.04	9.44	7.6	74%	1
	3	22	16.72	9 56	7.16	64%	1
	4	31	N/A	N/A	N/A	N/A	N/A
VERM	4	26	17.04	9.56	7.48	72%	1
	4	23	16.92	9.52	7.4	70%	1
	5	41	17.04	9.48	7.56	74%	1
	5	35	17.08	9.12	7.96	80%	1
	5	33	17	9.44	7.56	74%	1
	6	51	16.8	9.4	7.4	70%	1
	6	49	16.72	9.56	7.16	64%	1
	6	46	17.12	9.56	7.56	74%	1
	1	4	16.88	9.48	7.4	70%	1
	1	2	16.8 NI (A	9.32 NI (A	7.48 NI/A	/2% NI/A	
	3	/	IN/A	IN/A NI/A	N/A	IN/A N/A	N/A N/A
	4	11	IN/A 16.76	N/A 0.26	$\frac{1N}{A}$	IN / A 70%	N/A 1
	4	16	17.04	9.30	7.4	70%	1
HAM	7	10	N/A	N/A	N/A	N/A	N/A
	10	27	N/A	N/A	N/A	N/A	N/A
	12	33	17.12	9.28	7.84	79%	1
	13	38	N/A	N/A	N/A	N/A	N/A
	13	36	17.08	9.45	7.63	75%	1
	14	43	17	9.2	7.8	78%	1
	14	40	17.16	9.16	8	82%	1
	1	3	16.8	9.62	7.18	65%	1
	2	7	16.96	9.64	7.32	68%	1
	3	10	16.64	9.62	7.02	60%	1
	4	17	16.98	9.64	7.34	69%	1
	5	21	16.56	9.64	6.92	58%	0
	6 7	20	16.68	9.4	7.28	67 % 50%	0
	8	30	10.36	9.04	7.64	75%	0
	8	37	16.96	9.52	7.04	73%	1
	9	45	16.72	9.28	7.44	71%	1
	10	49	17	9.4	7.6	74%	1
	11	55	16.72	9.52	7.2	65%	0
CHA	11	51	N/A	N/A	N/A	N/A	N/A
	12	57	17.12	9.48	7.64	75%	1
	13	64	16.88	9.44	7.44	71%	1
	13	62	16.68	9.6	7.08	62%	1
	13	60	16.92	9.52	7.4	70%	1
	14	73	17	9.6	7.4	70%	U NT (1
	15	82	N/A	N/A	N/A	N/A	N/A
	10 14	80 82	10.84	9.56 0.24	7.28 7.74	0/% 78%	1
	10 175	00	1/ 16.8/	7.24 9.52	7.70	70% 68%	1
	17a 17b	93	16.04	9.52	7.32	72%	1 1
	17b 17b	90	N/A	N/A	N/A	N/A	N/A
	18	100	N/A	N/A	N/A	N/A	N/A
	19	102	16.92	9.28	7.64	75%	1

Table 2. Results of XRD analysis the <2.0 μ m clay-sized fraction of Illinois Basin paleosol matrices.

^a \circ 2 θ values reported using Cu-K α radiation. ^b Calculated Δ° 2 θ parameter using Equation (1). ^c FC = flint clay or a kaolinite rich claystone of detrital or volcanic origin. ^d Samples classified as N/A did not have a sufficient quantity of the <2.0 µm fraction of the paleosol matrix or any identifiable XRD peaks associated with interstratified illite-smectite. Therefore, these samples could not provide any illitization information.

Core	Paleosol (P)#	Sample ID	I-S (002/003) (° 2θ) ^a	I-S (001/002) (° 2θ) ^a	$\Delta^{\circ} 2\theta^{b}$	% Illite in I-S (\pm 3%)	Reichweite Order
	3	22	16.68	9.32	7.36	69%	1
ADM	4	27	N/A ^c	N/A	N/A	N/A	N/A
VERM	2	12	16.86	9.62	7.24	66%	1
VEDM	4	23	17.06	9.22	7.84	79%	1
VEINVI	6	49	17	9.44	7.56	74%	1
	6	46	17.06	9.4	7.66	75%	1
	4	9	17.04	9.24	7.8	78%	1
LIAM	6	16	17.16	9.28	7.88	80%	1
ПАМ	13	36	17.04	9.56	7.48	72%	1
	14	43	17	9.24	7.76	78%	1
СНА	9	45	16.76	9.44	7.32	68%	1
	10	49	N/A	N/A	N/A	N/A	N/A
	12	57	N/A	N/A	N/A	N/A	N/A
	13	62	16.8	9.64	7.16	64%	1
	16	83	17.04	9.48	7.56	74%	1
	17b	93	16.96	9.34	7.62	74%	1

Table 3. Results of XRD analysis of the <0.2 μm clay-sized fraction of Illinois Basin paleosol matrices.

^a $^{\circ}$ 2 θ values reported using Cu-K α radiation. ^b Calculated Δ° 2 θ parameter using Equation (1). ^c Samples classified as N/A did not have any identifiable XRD peaks associated with interstratified illite-smectite. Therefore, these samples could not provide any illitization information.

In general, XRD patterns display similar mineral compositions for all <2.0 and <0.2 μ m fractions of paleosol samples in each of the four cores, though peak intensity is variable throughout the dataset (Figures 3–7). The general mineralogy of paleosol matrix samples does not have any strong dependence on paleosol type or sampling location (Figures 2–6). Furthermore, there are not significant changes in mineralogy across the Desmoinesian–Missourian boundary in the <2.0 μ m fraction (Figures 3–6). XRD peakedness is strong at 7.15 and 3.58 Å, associated with kaolinite (001) and (002), respectively (Figures 3–6). Peakedness is poor around most other peaks, possibly indicating a prevalence of small crystal size, i.e., clay minerals, and/or poor crystallinity. For example, the peak around 3.35 Å can be associated with quartz (001), illite (003), mica (003), and, to a lesser extent, smectite (004), but due to the high intensity, broadness of the peak at half height, and the presence of multiple equidistant peaks in the pattern, it possibly represents both clay and nonclay minerals.

The Bond Fm. contains the most prominent peaks of I-S in the (001/002) and (002/003) positions in the ADM, VERM, and CHA cores compared to other formations (Figures 3–6). The ADM core contains some of the broadest I-S peaks at low angles, i.e., $5-8^{\circ} 2\theta$, indicating greatest expansibility or amount of smectite in I-S (e.g., ADM-25; Figure 3). The HAM core contains the weakest intensity peaks of I-S at all reflection angles (Figure 6); though peaks for chlorite have a higher intensity in the HAM core compared to the other three cores (Figures 3–6).

3.2. Reichweite Ordering

Reichweite Ordering results from XRD patterns of the <2.0 μ m fraction of IB paleosol matrices indicate that I-S possess random and ordered interstratification (Table 2; Figure 8), though ordered I-S exists only as R1 and not long-range order (R3). Samples without observable I-S, and thus no Reichweite Order, are noted relative to depth for reference (Figure 8). The VERM and HAM cores contain I-S with only R1 ordering, whereas the ADM and CHA cores contain both R0 and R1 ordering. The Bond Fm. contains the greatest number of I-S samples with random interstratification (n = 4) in the CHA and ADM cores compared to the Carbondale (n = 0), Shelburn (n = 1), Patoka (n = 1), and Mattoon Fms. (n = 1; Tables 1 and 2). There is only one sample from the Desmoinesian Shelburn Fm. that contains R0 ordering (CHA-73), and the remaining samples with R0 ordering are from Missourian strata (Tables 1 and 2; Figures 6 and 8).



Figure 3. XRD patterns of ethylene glycol saturated oriented aggregates of the <2.0 μ m equivalent spherical diameter claysized fraction from paleosols in the Illinois State Geological Survey (ISGS) -Archer-Daniels-Midland, Borehole MMV-04B (ADM) core (*n* = 10). Diffraction patterns are organized with respect to their relative stratigraphic position, i.e., ADM-33 and ADM-6 are stratigraphically the lowest and highest samples, respectively. D-spacing information for minerals discussed herein are noted vertically and in angstroms (Å). Diffraction patterns are colored to correspond with paleosol type, i.e., dark green = gleyed Vertisol (Pedotype C [43,44]), light green = gleyed calcic Vertisol (Pedotype D), ochre = gleyed vertic Calcisol (Pedotype E), and red = calcic Vertisol (Pedotype G). D-M = Desmoinesian–Missourian boundary. Core location noted in Figure 2. See Tables 1 and 2 for more details.



Figure 4. XRD patterns of ethylene glycol saturated oriented aggregates of the <2.0 μ m equivalent spherical diameter claysized fraction from paleosols in the Materials Service Corps #F-72-8 (VERM) core (*n* = 13). Diffraction patterns are organized with respect to their relative stratigraphic position, i.e., VERM-46 and VERM-2 are stratigraphically the lowest and highest samples, respectively. D-spacing information for minerals discussed herein are noted vertically and in angstroms (Å). Diffraction patterns are colored to correspond with paleosol type, i.e., light green = gleyed calcic Vertisol (Pedotype D [43,44]) and red = calcic Vertisol (Pedotype G). D-M = Desmoinesian–Missourian boundary. Core location noted in Figure 2. See Tables 1 and 2 for more details.



Figure 5. XRD patterns of ethylene glycol saturated oriented aggregates of the <2.0 μ m equivalent spherical diameter clay-sized fraction from paleosols in the Illinois State Geological Survey (ISGS) #1 City of Charleston (CHA) core (*n* = 23). Diffraction patterns are organized with respect to their relative stratigraphic position, i.e., CHA-102 and CHA-3 are stratigraphically the lowest and highest samples, respectively. D-spacing information for minerals discussed herein are noted vertically and in angstroms (Å). Diffraction patterns are colored to correspond with paleosol type, i.e., grey = gleyed Protosol (Pe-dotype B [43,44]), dark green = gleyed Vertisol (Pedotype C), light green = gleyed calcic Vertisol (Pedotype D), ochre = gleyed vertic Calcisol (Pedotype E), and red = calcic Vertisol (Pedotype G). D-M = Desmoinesian–Missourian boundary. Core location noted in Figure 2. See Tables 1 and 2 for more details.



Figure 6. XRD patterns of ethylene glycol saturated oriented aggregates of the <2.0 μ m equivalent spherical diameter claysized fraction from paleosols in the American Coal Company Borehole 7510-20 (HAM) core (*n* = 10). Diffraction patterns are organized with respect to their relative stratigraphic position, i.e., HAM-40 and HAM-4 are stratigraphically the lowest and highest samples, respectively. D-spacing information for minerals discussed herein are noted vertically and in angstroms (Å). Diffraction patterns are colored to correspond with paleosol type, i.e., grey = gleyed Protosol (Pedotype B [43,44]), dark green = gleyed Vertisol (Pedotype C), and light green = gleyed calcic Vertisol (Pedotype D). Desmoinesian–Missourian boundary. Core location noted in Figure 2. See Tables 1 and 2 for more details.

Relative Intensity





Figure 7. XRD patterns of ethylene glycol saturated oriented aggregates of the < 0.2μ m equivalent spherical diameter clay-sized fraction from select paleosols in the ADM, VERM, CHA, and HAM cores. Diffraction patterns are organized with respect to their relative stratigraphic position, i.e., HAM-40 and HAM-4 are stratigraphically the lowest and highest samples, respectively, from the HAM core. D-spacing information for minerals discussed herein are noted vertically and in angstroms (Å). Diffraction patterns are colored to correspond to core, i.e., red = ADM, green = VERM, orange = CHA, yellow = HAM. Core location noted in Figure 2. See Tables 1 and 3 for more details.



Figure 8. Reichweite Ordering values relative to stratigraphic position for each <2.0 μ m (circles) and <0.2 μ m (diamonds) equivalent spherical diameter clay-sized fraction of paleosol samples from the the Materials Service Corps #F-72-8 (VERM; green; <2.0 μ m, *n* = 13; <0.2 μ m. *n* = 4), Illinois State Geological Survey (ISGS)-Archer-Daniels-Midland, Borehole MMV-04B (ADM; red; <2.0 μ m, *n* = 10; <0.2 μ m, *n* = 2), ISGS #1 City of Charleston (CHA; orange; <2.0 μ m, *n* = 20; <0.2 μ m, *n* = 6), and the American Coal Company Borehole 7510-20 (HAM; yellow; <2.0 μ m, *n* = 10; <0.2 μ m, *n* = 4) cores associated with colored circles in of the IB map (Figure 2). The northernmost core is on the far left and the southernmost core is on the far right. The Chapel Coal (black dashed lines) is delineated for reference in each core to demonstrate stratigraphic control. Samples classified as N/A did not have any noticeable interstratification of illite-smectite visible in XRD patterns and therefore could not provide any ordering or illitization information (see Tables 2 and 3 for tabulated Reichweite Ordering values).

Reichweite Ordering results from the XRD patterns of the <0.2 μ m fraction of IB paleosol matrices (Figure 7) indicate that I-S possesses only R1 ordered interstratification (Table 3; Figure 8). Generally, the Reichweite Order is the same for a given sample regardless of the size fraction (Tables 2 and 3; Figures 3–8). The exceptions are two samples from the CHA core and one sample from the ADM core, where I-S is not detectable. In sum, R1 ordering is the most common Reichweite Order amongst all cores and size fractions sampled for this work (Tables 2 and 3; Figure 8). Since R1 is usually the product of diagenetic illitization [7,34,35], the Reichweite Ordering results presented herein suggest that most of the I-S in the sample set has been impacted by deep burial diagenesis.

3.3. Δ° 2 θ Analyses

 $\Delta^{\circ} 2\theta$ data derived from the <2.0 and <0.2 µm fractions of IB paleosol matrices from the ADM, VERM, CHA, and HAM cores have a range of 6.2 to 8° 2 θ . When converted to % illite layers in I-S, the range is from 42% to 82% (Tables 2 and 3; Figure 9). The ADM core possess most of the samples that contain <60% illite in I-S, and these were collected from <140 m core depth (Tables 1–3; Figure 9a). The HAM and CHA cores contain the deepest samples, many recovered from depths exceeding 150 m core depth (Table 1; Figure 9a). Twenty-nine samples from >150m core depth in the HAM and CHA cores contain >60% illite in I-S, as opposed to two samples that are <60% illite in I-S (Tables 1–3; Figure 9a). Stratigraphically, the % illite in I-S from these samples have varying percentages (Tables 1 and 2; Figure 9b). For example, samples collected within ±10 m from the Chapel Coal from the VERM (*n* = 3), HAM (*n* = 1), CHA (*n* = 4), and ADM (*n* = 2) cores have a range from 52% to 80% illite in I-S in the <2.0 µm (Tables 1 and 2; Figure 9b) whereas the <0.2 µm fraction has a range of 64–80% illite in I-S from the CHA (*n* = 1), HAM (*n* = 1), and VERM (*n* = 1) cores (Tables 1 and 3; Figure 9b). Moreover, the lowest % illite in I-S values are from the Bond (42%) and Patoka Fms. (52%), while the highest % illite in I-S value

is from the Carbondale Fm. (82%; Tables 1 and 2; Figure 9b). In sum, since % illite in I-S superstructures that is >60% is usually the product of diagenetic illitization [1,7,27,36], the $\Delta^{\circ} 2\theta$ data % illite in I-S data presented suggest that most of the I-S in the sample set herein are products of deep burial diagenesis.



Figure 9. $\Delta^{\circ}2\theta$ values were calculated from Equation (1) and converted to % illite in illite-smectite superstructures [1] for each <2.0 μ m (circles, this study; triangles from [43]) and <0.2 μ m (diamonds, this study) clay-sized fraction of paleosol samples taken from the Materials Service Corps #F-72-8 (VERM; green; <2.0 μ m, n = 13; <0.2 μ m, n = 4), Illinois State Geological Survey (ISGS)-Archer-Daniels-Midland, Borehole MMV-04B (ADM; red; <2.0 µm n = 10; <0.2 µm, n = 1), ISGS #1 City of Charleston (CHA; orange; <2.0 μ m, n = 22; <0.2 μ m, n = 4), the American Coal Company Borehole 7510-20 (HAM; yellow; <2.0 μm, *n* = 8; <0.2 μm, *n* = 4), the Lone Star Cement Company #TH-1 (LSC; blue; <2.0 μm, *n* = 32 [43]), and the Monterey Coal Company, Mac 1 #CBM4 (MAC; grey; <2.0 μ m, n = 31 [43]) cores. Samples that have <60% illite in I-S are interpreted to result from pedogenic illitization [27], whereas samples >60% are interpreted to result from diagenetic illitization [7]. See Tables 2 and 3 for tabulated data. (a) % illite in I-S relative to depth form the surface, where 0 m represents the top of each core. Each core is color coded to the map in Figure 2b. (b) % illite in I-S is plotted relative to stratigraphic position in the basin, where the 0 m datum is equivalent to the location of the Chapel Coal in each core. The Chapel Coal, of the lowermost Patoka Fm (Figure 2a) is shown as stratigraphic data to the depth measurements. Each data point color follows the legend in part a. (c) This is the same plot as part b, except each sample point is color coded towards the formation from which it was sampled, including: the Mattoon (red; <2.0 μ m, n = 5), Bond (lilac; <2.0 μ m, n = 26; <0.2 μ m, *n* = 3), Patoka (navy; <2.0 μm, *n* = 35; <0.2 μm, *n* = 4), Shelburn (turquoise; <2.0 μm, *n* = 25; <0.2 μm, *n* = 1), Carbondale (ochre; <2.0 μ m, n = 17; <0.2 μ m, n = 5), and Tradewater Formations (green; <2.0 μ m, n = 8).

4. Discussion

4.1. Provenance and Detrital Components of Illinois Basin Paleosols

Provenance studies of the IB indicate that Pennsylvanian strata are composed of intrabasinal derived, reworked Mississippian strata and extrabasinal sediments [59]. Extrabasinal sediments contain detrital zircon signatures associated with the Neoproterozoic, Grenville, Granite-Rhyolite, and Yavapai-Mazatzal basement terranes [59–62]. Pennsylvanian-aged back-fill of fluvially incised paleo-valleys indicate intrabasinal sediment sources while paleoflow indicators in crossbedding, indicate a south-southwest trend during fluvial deposition [63,64]. In sum, it is likely that extrabasinal IB sediments were likely derived from eastern Laurentia, possibly as the Appalachian Mountains began to uplift. Moreover, using these data, parent material is characterized as felsic in order to evaluate findings on Pennsylvanian detrital rocks and weathering products within the IB.

Sedimentary petrology of Pennsylvanian units [65,66] and underclays or paleosols [43–45,67–71] indicate the presence of quartz, kaolinite, illite and/or mica, chlorite, and I-S. Feldspar, pyrite, and muscovite are common accessory minerals. Many of the major minerals identified in previous studies are corroborated in XRD patterns presented herein of both the <2.0 and <0.2 μ m clay-sized fractions (Figures 3–7).

The <0.2 µm fraction likely contains only kaolinite, I-S, and illite (Figure 7). From sedimentary petrology (e.g., [72]), stable isotope (δ D and δ ¹⁸O; e.g., [73]), and radio isotope studies (K-Ar; e.g., [74]) of clay minerals, it is likely that finer size fractions (<0.1 or <0.2 µm) of sedimentary rocks, including paleosols, are composed dominantly of authigenic clay minerals or crystallites. The lack of chlorite in the <0.2 µm clay-sized fraction diffraction patterns suggests that its presence in coarser-sized fractions may be a detrital component of IB paleosols. Due to the broadness of the peak and the decrease and intensity around 3.35 Å when comparing samples from the <2 µm and <0.2 µm (Figures 3–7), it is likely that quartz is also a detrital mineral-grain component. Due to the scope of this study on I-S, only mica and illite type phyllosilicates will be discussed further.

Differentiating between mica, specifically muscovite, and illite using XRD is challenging because they have nearly identical sharp, peak patterns at 10, 5, and 3.35 Å associated with crystallographic d(hkl) planes (001), (002), and (003), respectively (Figures 3–6). Genetically, micas are frequently detrital components of soils and paleosols whereas illite can be detrital, pedogenic, and/or diagenetic components [72,75–78]. However, pedogenic mica has been documented in some soil environments where there is K⁺ or NH₄⁺ to promote transformation from vermiculite to mica [79,80], though micas frequently weather to other clay minerals such as smectite or vermiculite as soil formation continues [81]. It is also possible that micas may be products of late-stage diagenesis in the epizone (e.g., [82]), at higher temperatures than what is expected during diagenesis (around 300 °C, [83,84] in the metamorphic window). Therefore, although pedogenic and diagenetic micas may be part of these paleosol clay-sized fractions, illite is a more likely candidate to be associated with the XRD peaks around 10 and 5Å observed in IB paleosols (Figures 3–7).

Regardless of its exact origin, since the <0.2 μ m clay-sized fraction is suspected to be dominated by authigenic clay minerals, any mica/illite peaks from XRD patterns of the <0.2 μ m fraction of paleosol matrices reported herein (Figure 7) are likely authigenic illite (and/or mica). Moreover, when considering the XRD results of <2.0 (Figures 3–6) and <0.2 μ m (Figure 7) clay-sized fractions, including results from Reichweite Ordering and % illite in I-S for <2.0 and <0.2 μ m clay-sized fractions (Tables 2 and 3; Figures 8 and 9), the XRD pattern analyses produce similar results. Therefore, since the <0.2 μ m fraction is generally considered to be authigenic, it is possible that much of the I-S and illite (and/or mica) in the analogous <2.0 μ m fraction is also authigenic.

Thus, for the remainder of the discussion, I-S analyzed herein is considered to be only authigenic, so that the probability of I-S genesis in the ancient soil formation versus the deep burial diagenetic environments in the IB may be assessed. The abundances of detrital and diagenetic illite and I-S in IB paleosol fractions will be explored in future research.

4.2. Origins of Authigenic I-S From Illinois Basin Strata

4.2.1. Pennsylvanian Paleosols

By comparing Reichweite Ordering and % illite in I-S results, we can verify whether both methods conclusively identify deep-burial diagenesis as the primary process responsible for illitization and I-S in IB paleosols. Although 17% of samples studied herein display no evidence for I-S, both Reichweite Ordering and % illite in I-S proxies agree that a majority of samples have I-S, though the percentages for genetic origins are unequal (Figure 8, Figure 9 and Figure S1a, Supplementary Materials). When considering the results from each method for each sample, 81% of samples agree that I-S in that sample is the results of deep-burial diagenesis while only 6% of samples agree that I-S in that sample is the results of pedogenesis (Figure S1c). However, 13% of samples produce conflicting results such that Reichweite Ordering or % illite in I-S yield a pedogenic origin while the other method yields a diagenetic origin (Figure S1b,c). Since there is an expected error of $\pm 0.02-0.08^{\circ}$ 2θ while identifying peaks [57], any % illite in I-S value within 60% \pm 5% should also be suspect (Table 2; Figure S1b). This result suggests that although these proxies are generally corroborative (87%; Figure S1c), they should not be used independently to determine genetic origins of I-S or there may be some conflicting results.

Considering the results of Reichweite Ordering and % illite in I-S superstructure measurements presented herein, samples from the ADM, CHA, VERM, and HAM cores have been impacted by deep burial diagenetic processes. This implication can be tested by comparing this dataset to comparable data collected from penecontemporaneous IB paleosols, originally reported by [43] (p. 625–631), from the Lone Star Cement Company #TH-1 (LSC) and Monterey Coal Company Mac 1 #CBM4 (MAC) cores (Figure 2b). Note, there are no comparable Reichweite Ordering data from the samples collected from the LSC or MAC cores therefore only % illite in I-S data will be discussed, the aforementioned provision that using only data to assess illitization processes notwithstanding for this particular comparison.

The % illite in I-S data collected from the LSC and MAC cores range from 3% to 89% (Figure 9a,b) [43], which covers a larger range than the dataset presented herein (42–82%; Table 2). Moreover, the samples were recovered from shallower core depths, 3.05–142.64 m [43], compared to 20.2–336.7 m for samples collected from the ADM, CHA, VERM, and HAM cores (Figure 9a; Table 2). This follows because the depth of the Pennsylvanian strata in the IB is greater in the south and the interior where the CHA and HAM cores were drilled, relative to their occurrence at shallower depths in the northern (LSC) or western basin margin (MAC; Figure 2 [85,86]). Regardless of these differences, the data from all six cores indicate that % illite in I-S increases with depth, but the % illite in I-S is more variable between 0 and 150 m (3–89%) compared to the data >150 m which are between 58–82% (Figure 9a). Stratigraphically, since all the formations contain variable % of illite in I-S in each core, there is no clear, basin-wide pattern to suggest homogenous preservation of exclusively low-temperature, pedogenetic illitization in these data (Figure 9b,c). Rather, the "rule" across the IB for I-S crystal properties is indicative of deep-burial diagenesis.

As stated in the introduction, illitization in the pedogenic environment may be promoted by fluctuating redox conditions and concomitant wet-dry cycles [27–30,32,33]. Since seasonality and thus changing redox states were likely promoted during the transition from the middle to upper Pennsylvanian in this region [43–45,87], we would expect to see increasing % illite in I-S from the lower formations, i.e., Tradewater, Carbondale, and Shelburn, to the upper formations, i.e., Patoka, Bonda, and Mattoon (Figure 2a), and % illite in I-S should be <60% [27] if illitization was primarily promoted during pedogenesis. However, the composite % illite in the I-S dataset of the ADM, VERM, CHA, HAM, MAC, and LSC cores rejects this hypothesis (Figure 9a–c).

The variability in % illite in I-S could be attributed to heterogenous temperatures during diagenetic illitization in the IB. This is a common conclusion from diagenetic clay mineral studies because sedimentary basins have spatially and temporally variable subsidence, uplift, erosion, and heat flow (e.g., [88]). Since illitization is likely dependent on kinetic conditions, its occurrence is in part temperature dependent, and thus should increase with increasing burial depth [3,23,89–91]. Still, according to vitrinite reflectance data collected from Pennsylvanian coals, the IB likely experienced a shallow burial, with a maximum depth of ~1–3 km [92–95], where the northern IB was buried to a lesser extent than the southern IB. Still, these trends of high % illite in I/S and R1 ordering are seen across all cores in the basin.

4.2.2. Devonian-Pennsylvanian Siliciclastic Rocks

To compare the potential impact of diagenesis on older rock units in the IB we can consider also previous studies of I-S in other clastic rocks from different lithologies (i.e., not paleosol profiles) from the IB. In siliciclastics however, there is not the issue of differentiating between I-S developed in the pedogenic or diagenetic environments. For instance, [66] (p. 358–361) studied I-S in the Devonian-Mississippian-aged New Albany shale of the IB, finding that that % illite in I-S is between 80–95% and increased with depth to 1 km. However, % illite in I-S was unchanged between 1 and 2.5 km, and total concentration of stochiometric illite became more abundant within this same region of deeper burial [66]. This is significant because the New Albany Shale % illite in I-S values are similar to values found herein, even though the New Albany shale is older (and deeper) than Pennsylvanian paleosols investigated in this study.

Similarly, studies of the Pennsylvanian-aged Browning Sandstone and Purington Shale (Tradewater–Carbondale Fms.) also possess I-S with <10% smectite [96,97], consistent with findings presented herein and [43,66] (p. 358–361). In both Pennsylvanian clastic studies, the lack of substantial burial of the IB coupled with advanced stages of diagenetic maturity may indicate low-temperature, time-dependent, or protracted, diagenesis, which drives illitization processes [96,97]. Although, the importance of temperature and time to the illitization reaction is debatable [90,91].

Considering the consistent signatures of deep-burial diagenesis recorded in I-S clay minerals from various stratigraphic intervals in the IB, I-S may record influence(s) of variable heat flow sources associated with tectonic and burial changes in the IB after Pennsylvanian deposition, pedogenesis, and burial.

4.3. Potential Sources of Alteration in Pennsylvanian Phyllosilicates

The results from this work suggest that I-S in IB paleosols has primarily formed following pedogenesis, during shallow to deep burial (Figures 8 and 9). We can now consider how non-pedogenic signatures in Pennsylvanian-aged paleosols were acquired in the IB, acknowledging that the stability of most pedogenic phyllosilicates wanes around \sim 50–200 °C [7,20,98] and undergoes mineralogical, diagenetic changes. Both temperature change and water/rock interactions will be discussed here to assess the probability of diagenetic alteration of original paleosol phyllosilicates.

4.3.1. Coals and Acid Leaching

Some of the Pennsylvanian paleosols studied herein are underclays, or argillaceous rocks that formed before, and stratigraphically beneath, a coal bed [37,67,68]. Although not the focus of this work, kaolinite was identified in the XRD patterns (Figures 3–7) of these paleosol underclays. If kaolinite formed during pedogenic weathering in the Pennsylvanian period, we would expect it to form under humid conditions in well-drained soils [78]. However, it is also possible that acidic fluids leaching from overlying peats could have spurred hydrolytic reactions to form kaolinite in underlying paleosols during shallow burial, a suggestion by previous underclay mineralogy studies [43,69,71,99].

Similarly, it has been suggested that organic acids may impact illitization by increasing the rate of smectite, K-feldspar, and muscovite dissolution, promoting the availability of K and Al cations into the illite layers [27,100,101]. [71] (p. 425) studied the underclay paleosols of the Herrin Coal from the Carbondale Fm. from IB cores in the southwestern

part of the basin and found that I-S expandability decreased and ordered I-S increased relative to randomly interstratified I-S as depth below coal increased, demonstrating the relation between in situ acid leaching and illitization. Similarly, there are 26 samples from the <2.0 μ m and 9 samples from the <0.2 μ m clay-sized fractions analyzed herein that are from paleosols directly underlying coal seams (Figure 10 and Table 4). When comparing the depth of these samples below the coal seam, % illite in I-S increases with increasing depth, though there is some scatter (Figure 10). These data may support the conclusions of [71] (p. 429) such that proximity to coal seams and thus acid leaching may instigate some illitization possibly through fixation of not only K⁺, but also NH₄⁺. This may have occurred shortly after soil formation, concurrent with coal forming swamps, or after burial of the soil, now paleosol. However, the other 27 samples of the <2.0 μ m fraction of paleosols in this study display >60% illite in I-S (Figure 9 and Table 2) even though they are not sampled from paleosols beneath coals. Therefore, there may be multiple mechanisms influencing post-pedogenic illitization processes within the IB paleosol profiles.



Figure 10. % illite in I-S of the <2.0 and <0.2 μ m clay-sized fraction of paleosol matrices relative to depth below coal seam for paleosols directly underlying coals (*ntot* = 35). See Table 4 for tabulated data and information about specific IB coals.

Core	Paleosol (P)#	Sample ID	Overlying Coal	Coal Depth in Core (m)	Paleosol Sampling Depth below Coal (cm)
	2	18	Flat Creek	79	70
ADM	3	25	Fairbanks	94.94	26
	3	22	Fairbanks	94.94	146
	3	21	Fairbanks	94.94	176
	5	35	Chapel	129.63	17
	5	33	Chapel	129.63	57
VFRM	4	31	Womac	48.31	9
VEKIVI	4	26	Womac	48.31	189
	3	7	Fairbanks	41.39	21
	4	11	New Haven	72.67	33
	4	9	New Haven	72.67	93
	6	16	Chapel	137.67	43
	7	19	Athensville	171.75	15
HAM	10	27	Danville	195.85	15
	12	33	Baker	201.158	104.2
	13	38	Herrin	219.03	107
	13	36	Herrin	219.03	137
	14	43	Springfield	244.224	47.6
	14	40	Springfield	244.224	147.6
	5	21	McCleary's Bluff	157.76	94
	8	40	Flat Creek	201.98	112
	9	45	Fairbanks	213.72	78
	10	49	Fairbanks	214.87	43
	12	57	Womac	236.79	71
	13	64	Chapel	258.34	26
СНА	13	62	Chapel	258.34	86
CIIA	13	60	Chapel	258.34	146
	14	73	Rock Branch	262.72	168
	16	86	Danville	294.24	96
	16	83	Danville	294.24	166
	17a	98	Herrin	325.85	25
	17b	93	Herrin	325.85	165
	18	100	Springfield	334.04	46

Table 4. Paleosols samples and their overlying coal seams.

4.3.2. The Geothermal Gradient

Another common trigger of phyllosilicate alteration in any sedimentary basin is during shallow to deep burial, as mentioned in the previous section. In sedimentary basins, the average geothermal gradient is about 30 °C/km [102]. Since IB vitrinite reflectance data from Pennsylvanian coals supports a maximum burial depth of around 1–3 km, there was likely a regional geothermal gradient between 20 and 23.5 °C/km in the IB [92–95]. A more recent study modelled vitrinite reflectance and basin analysis to suggest that the geothermal gradient may have been 30–50 °C/km in the Pennsylvanian strata during maximum burial, though this might not be solely due to burial but also hydrothermal fluid flow [103]. Therefore, it is possible that temperatures during maximum burial were 70–150 °C, which may have facilitated diagenetic alteration of Pennsylvanian phyllosilicates [66].

Depth of burial is perhaps the most typical explanation for I-S in sedimentary basins, which was proposed in the initial studies of Eocene-Pleistocene-aged sediment cores from the U.S. Gulf Coast, where illitization was found to have occurred as depth of burial increased [6,7,15–20]. However, the work in the Gulf Coast, suggested that illitization promoted during burial occurs at depths >3 km and temperatures >100 °C, yet the data

presented herein suggest that signals of illitization and burial diagenesis can be detected at maximum burial depths <3 km and potentially temperatures well below <100 °C. This is an important comparison between the IB and the U.S. Gulf Coast, which are relatively old and young basins, respectively, highlighting the significance of time as a process during burial diagenesis. Therefore, this work may indicate that signatures of illitization may not be dependent merely on depth of burial but the timing and duration of geologic history in which these processes have occurred.

It should also be noted that there must be sufficient abundances of K^+ in the system, likely at the expense of K-bearing minerals such as K-feldspar and mica, to promote potassium fixation during diagenetic illitization [7]. Typical silicate weathering reactions include hydrogen ions and water as reactants to form clay minerals, such as kaolinite, silica while releasing K^+ [104]. The problem with applying this weathering scheme to the IB is the minimal K-feldspar in IB siliciclastic rocks compared to quartz, though some sandstones are micaceous [65,105]. Perhaps more importantly, a closed system model of burial without a fluid interaction may not trigger this weathering reaction. Therefore, due to variable quantities of K-bearing minerals and no fluid flow, the impact of the depth of burial mechanism to prompt illitization in Pennsylvanian paleosols is unclear.

4.3.3. Intrusive Activity

It is possible that igneous intrusions may have altered the sedimentary strata in the IB. During the Ouachita orogeny, reactivation of faults in the Reelfoot Rift-Rough Creek Graben area led to emplacement of magma in the southern portion of the basin in the Permian (272–270 Ma [106,107]). It is likely that heat was conductively transferred from these magmatic intrusions into IB sedimentary strata. Other studies on the effects of igneous intrusions on clay minerals in sedimentary basins find similar % illite in I-S values to those reported herein. For example, the North Sea has values between 65–95% illite in I-S as stratigraphic position relative to the granitic basement increases [108]. In fact, in the IB, thermal maturity calculations of kerogen from coals, and bounding units, in the southern portion of the basin indicate regional diagenesis and contact metamorphism of IB strata that was concomitant with Permian-age intrusive magmatism, supporting this hypothesis [109–113]. It should be noted that this intrusive activity may have led to liberation of volatiles into permeable coals [111]. However, intrusive magmatism and subsequent conductive, and to some extent advective, heat flow is expected to have a regional, rather than basin wide, impact [111]. Also, similarly to the argument for the depth of burial, the lack of a definite K⁺ source and limited fluid availability may weaken the viability of this diagenetic illitization mechanism in Pennsylvanian paleosols across the IB.

4.3.4. Hydrothermal Brines

There is also evidence for topographically derived long-range brine migration in the IB strata, originating south of the IB near the Ouachita orogeny, and propagating north [114–117]. This brine migration is also suspected to have triggered the mineralization of the Upper Mississippi Valley and Fluorspar mineral districts during the Permian [115–120]. Fluid inclusion measurements from these mineral district yield temperatures between 50–170 °C [121–123]. Moreover, these briny fluids are hypothesized to be potassic due to evidence of authigenic K-feldspar overgrowths, K-leaching, and authigenic illite and I-S in sandstone aquifers [119,124–128]. Therefore, due to elevated temperatures and potassium availability, it is possible that illitization could be promoted in the IB due to these long-range, hydrothermal brines. However, most evidence for advective heat flow by brine migration through the IB is focused on the Cambrian and Ordovician siliciclastic strata [120,125,129–133] and Precambrian basement [124,127]. For instance, [133] (p. 1096–1102) found that ore-bearing fluids may have formed some or all of illite or I-S in Ordovician Maquoketa group of the IB. Therefore, it is necessary to constrain possible pathways to Pennsylvanian strata to assess viability of this illitization mechanism for paleosols.

Studies of hydrothermal advective activity in the Pennsylvanian strata is limited to the effects on coals. This includes fluid inclusions in coal minerals, including sphalerite and pyrite [134] and the relation between vitrinite reflectance and hydrothermal fluid flow modelling [47,103,135]. In fact, the model from [103] (p. 1810–1821) notes that the strata at or just below the Absaroka unconformity, near the Mississippian–Pennsylvanian boundary, may have acted as an aquifer for fluid flow during the hypothesized Permian long range brine migration. It is also possible that faults and fractures associated with the Reelfoot Rift-Rough Creek Graben area, the La Salle Anticline, the Wabash Valley Fault Zone, the Sandwich Fault Zones, or Plum River Fault Zone [136] may provide conduits to the Pennsylvanian stratigraphy. Since the possible impact of advective heat flow on the Pennsylvanian strata is limited to regional studies, the effects of hydrothermally motivated diagenesis across the basin remains unclear.

4.4. Implications for Pennsylvanian Paleoclimate Reconstructions from the Illinois Basin

Although the I-S XRD data and subsequent analyses presented herein may not precisely ascertain which diagenetic mechanism(s) discussed here were specifically responsible for mineralogical alteration of Pennsylvanian paleosols, these varying and welldocumented mechanisms highlight the many ways by which diagenetic alteration may have occurred in the IB. More importantly, since the Pennsylvanian strata has likely been impacted by post-pedogenic diagenesis, any geochemical-based paleoclimate study derived from the Pennsylvanian rock units may need to be re-evaluated. This is because paleoclimate studies frequently rely on the elemental and stable isotopic composition of non-altered paleosol clay minerals (e.g., [45]) or pedogenic carbonates (e.g., [137]). However, temperature stability of many subaerially or pedogenically formed clay minerals and carbonates wanes at temperatures ~50–200 °C [7,20,98]. This work has highlighted that this temperature range was likely sustained during burial, hydrothermal fluid flow, and/or conductive heat flow from intrusive magmatism in the IB. If pedogenic minerals geochemically exchange under either solid-state or dissolution/crystallization conditions [26,138–140], some or all of the original Pennsylvanian geochemical signature of soil formation preserved in the paleosol may be compromised. More specifically, this means that results of geochemical studies of altered clay minerals would provide insights on conditions involved during mineral transformation during diagenesis rather than Pennsylvanian soil formation. Therefore, recognition of I-S in paleosols is not inherently indicative of ancient soil formation processes, especially in ancient and shallow sedimentary basins, and should be evaluated more carefully before geochemical paleoclimate reconstruction studies are initiated.

5. Conclusions

By assessing the Reichweite ordering and % illite in I-S superstructures of illitesmectite mineralogical intergrades from IB paleosols, this study attempts to understand the post-Pennsylvanian history and diagenetic impact on clay minerals originally formed during pedogenesis. This study finds that deep-burial diagenesis is a more probable source of illitization in IB paleosols than a pedogenic origin for these mineralogical phases. The conclusions arrived at herein are further supported by previous studies of diagenetic I-S in non-paleosol rocks in the IB and a comprehensive discussion of the many possible mechanisms that may have promoted circumstances conducive to illitization during post-pedogenic diagenesis. In fact, it is likely that acid leaching from coal seams and hydrothermal fluid flow models provide enough evidence to support diagenetic illitization in observed in IB Pennsylvanian paleosols. Though the geothermal gradient determined by the depth of maximum basin burial and intrusive igneous activity may also have an impact on illitization if there is enough potassium available to promote fixation. It is possible that potassium is sourced from the weathering of mica group minerals since there are low abundances of K-feldspar compared to quartz in the basin. This study also highlights the importance of understanding time in diagenetic processes, specifically when in the geologic past the basin formed and was subject to diagenesis and the duration of diagenesis. These

results are significant for future studies seeking to reconstruct paleoenvironments using geochemical proxies from shallowly buried sedimentary basins due to the many possible diagenetic mechanisms that may impact the preservation of a paleo-proxy in an ancient basin's history.

Future work will attempt to identify illite polytypes and trace the K-Ar composition to differentiate between detrital and diagenetic I-S in IB paleosols. Future work also seeks to measure crystallization temperatures of phyllosilicates using δD and $\delta^{18}O$ stable isotope proxies on a basin-wide scale to further assess possible illitization mechanisms and the impacts of diagenesis on Pennsylvanian-aged paleosols in the IB.

Supplementary Materials: The following are available online at https://www.mdpi.com/2075-163 X/11/2/108/s1, Figure S1: Comparing data from Reichweite Ordering and % illite in I-S derived from $\Delta^{\circ} 2\theta$ for samples from the <2.0 µm clay-sized fraction of the ADM, VERM, HAM, and CHA cores. Recall, Reichweite Ordering data is not available for LSC and MAC cores.

Author Contributions: Contributions to the manuscript from first author J.A.M. include: Project design, identification of the main science question, funding acquisition, fieldwork, sample preparation, laboratory work, interpretation of results, and manuscript writing. Contributions from second author N.J.T. include: Analytical design, interpretation of results, contributions to text, and overall text editing. Contributions from third author N.A.R. include: Assistance during fieldwork, interpretation of results, consultation during manuscript writing, and overall text editing. All authors have read and agreed to the published version of the manuscript.

Funding: This study was funded by the Institute for the Study of Earth and Man at Southern Methodist University.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Data is contained within the article or Supplementary Material.

Acknowledgments: This work would not have been possible without the collaboration of the Illinois State Geological Survey. Specifically, thank you to Scott Elrick, W. John Nelson, and Bob Mumm. J.A.M. would like to thank T. Scott Myers and Roy Beavers for assistance in the laboratory. The authors would like to thank Crayton Yapp for thoughtful and constructive discussions throughout the stages of this work. This manuscript benefited from the commentary of three anonymous reviewers.

Conflicts of Interest: The authors declare no conflict of interest. The funders had no role in the design of the study; in the collection, analyses, or interpretation of data; in the writing of the manuscript, or in the decision to publish the results. The views expressed in this article are those of the authors and do not necessarily represent the views or policies of the U.S. Environmental Protection Agency.

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