

Supporting Information

A neutral heteroleptic molybdenum cluster *trans*- [{Mo₆I₈}({py})₂I₄]

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Table S1. Selected crystallographic parameters of the single-crystal X-ray diffraction structural analysis for $(\text{pyH})_{0.2}[\{\text{Mo}_6\text{I}_8\}(\text{py})_{1.8}\text{I}_{4.2}]\cdot 1.8\text{py}$

Compound	$(\text{pyH})_{0.2}[\{\text{Mo}_6\text{I}_8\}(\text{py})_{1.8}\text{I}_{4.2}]\cdot 1.8\text{py}$
Empirical formula	$\text{C}_{19}\text{H}_{19.2}\text{I}_{12.2}\text{Mo}_6\text{N}_{3.8}$
Formula weight	2424.60
Temperature, K	150(2)
Crystal system	monoclinic
Space group	$P\ 2_1/c$
$a, \text{\AA}$	12.276(3)
$b, \text{\AA}$	14.563(3)
$c, \text{\AA}$	12.586(3)
$\beta, {}^\circ$	104.640(6)
$V, \text{\AA}^3$	2177.0(8)
Z	2
$\rho_{\text{calc}}, \text{g}/\text{cm}^3$	3.699
μ, mm^{-1}	10.345
$F(000)$	2117
Crystal size	$0.12 \times 0.08 \times 0.02$
2 Θ range for data collection, ${}^\circ$	2.180 – 26.372
	$-15 \leq h \leq 14$
Index ranges	$-18 \leq k \leq 16$
	$-15 \leq l \leq 13$
Reflections collected	12938
Independent reflections; [R _{int}]	4445; [0.0800]
Parameters refined	193
Goodness-of-fit on F ²	0.935
R ₁ / wR ₂ (I > 2 σ (I))	0.0574 / 0.1139
R ₁ / wR ₁ (all data)	0.1238 / 0.1298
$\Delta Q_{\text{max}}/\Delta Q_{\text{min}} (\text{e}\cdot\text{\AA}^{-3})$	2.360 / -1.464

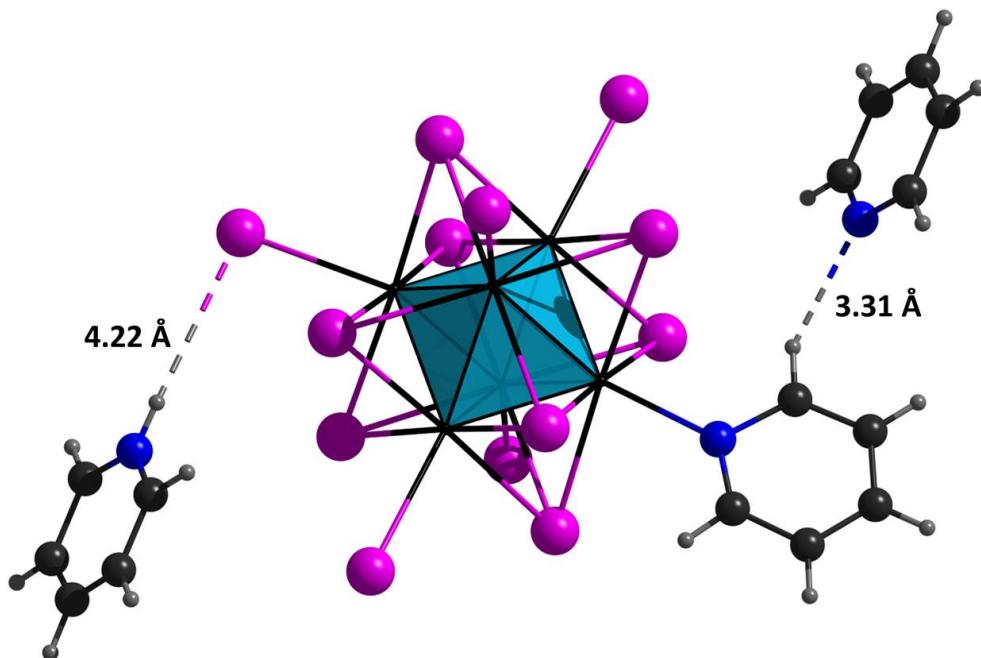


Figure S1. Illustration of $\text{py}^{\text{solv}}\cdots\text{py}^{\text{coord}}$ and $\text{pyH}\cdots\text{I}^{\text{a}}$ interactions. Color code: light blue octahedron – Mo₆, purple – iodine, blue – nitrogen, black – carbon, gray – hydrogen.

Table S2. Crystal lattice parameters for $(\text{pyH})_{0.2}[\{\text{Mo}_6\text{I}_8\}(\text{py})_{1.8}\text{I}_{4.2}]\cdot 1.8\text{py}$ and $[\{\text{Mo}_6\text{X}_8\}(\text{py})_2\text{X}_4]$ ($\text{X} = \text{Cl}, \text{Br}$)

Compound	Crystal system	$a, \text{\AA}$	$b, \text{\AA}$	$c, \text{\AA}$	$\beta, {}^\circ$
$(\text{pyH})_{0.2}[\{\text{Mo}_6\text{I}_8\}(\text{py})_{1.8}\text{I}_{4.2}]\cdot 1.8\text{py}$	monoclinic	12.276(3)	14.563(3)	12.586(3)	104.640(6)
$[\{\text{Mo}_6\text{Cl}_8\}(\text{py})_2\text{Cl}_4]$		17.412(1)	20.376(2)	12.074(1)	
$[\{\text{Mo}_6\text{Cl}_8\}(\text{py})_2\text{Br}_4]$		17.780(2)	20.691(2)	12.282(1)	
$[\{\text{Mo}_6\text{Br}_8\}(\text{py})_2\text{Cl}_4]$	orthorhombic	17.761(8)	20.594(10)	12.536(6)	90
$[\{\text{Mo}_6\text{Br}_8\}(\text{py})_2\text{Br}_4]$		18.143(4)	21.102(6)	12.686(3)	

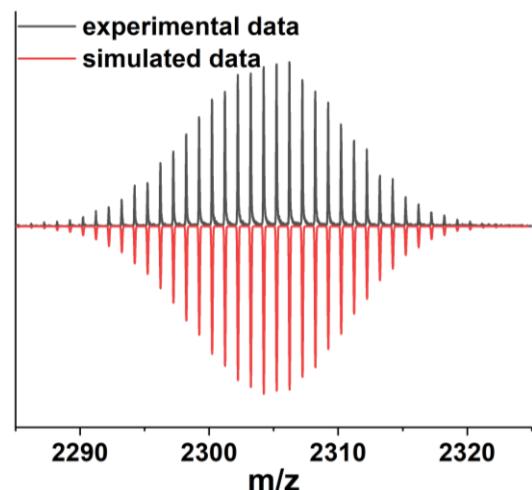


Figure S2. Fragment of ESI-MS of mother liquor illustrating the presence of $[\{\text{Mo}_6\text{I}_8\}(\text{py})\text{I}_5]^-$.

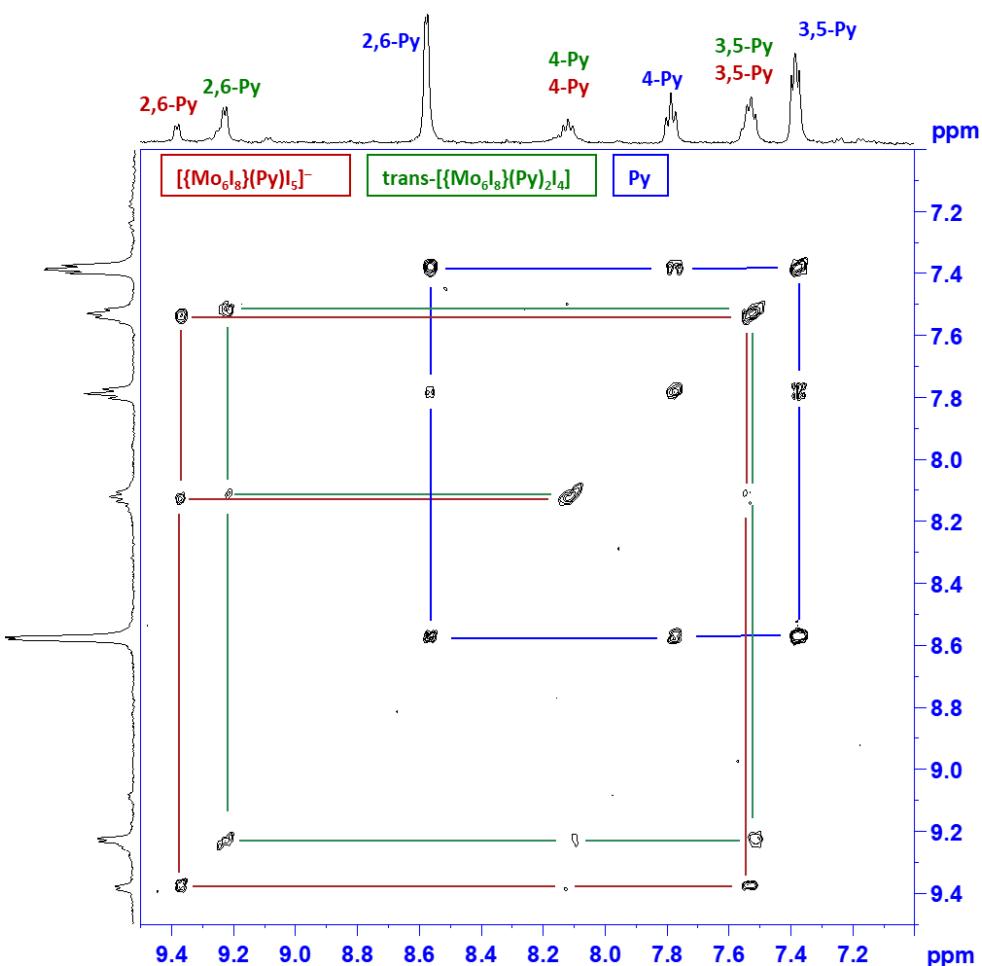


Figure S3. ^1H -COSY NMR spectrum of $(\text{pyH})_{0.2}[\{\text{Mo}_6\text{I}_8\}(\text{py})_{1.8}\text{I}_{4.2}]\cdot 1.8\text{py}$.

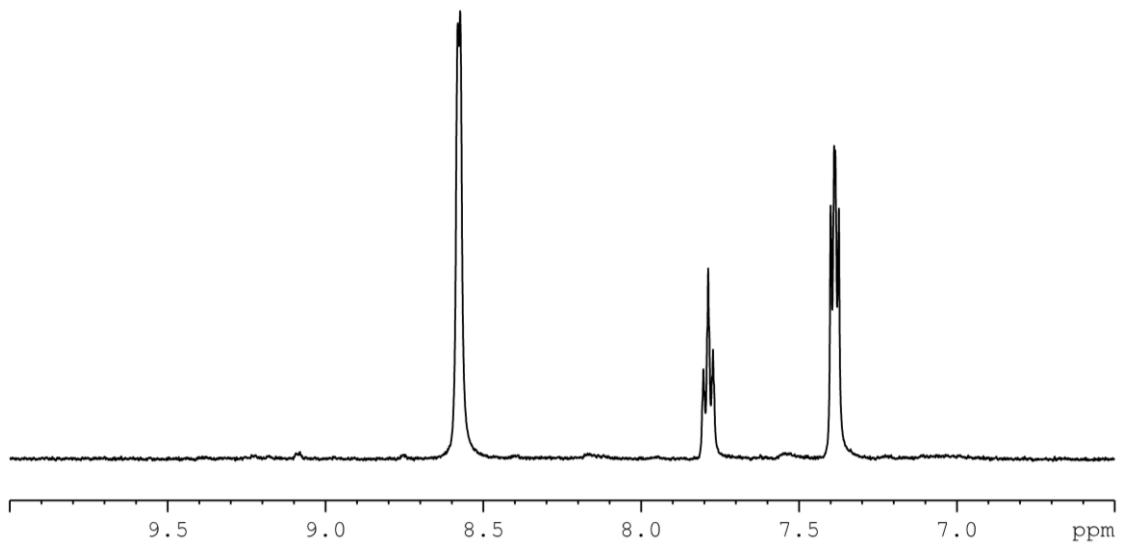


Figure S4. ^1H -NMR spectrum of $(\text{pyH})_{0.2}[\{\text{Mo}_6\text{I}_8\}(\text{py})_{1.8}\text{I}_{4.2}]\cdot 1.8\text{py}$ after heating in DMSO-d_6 .

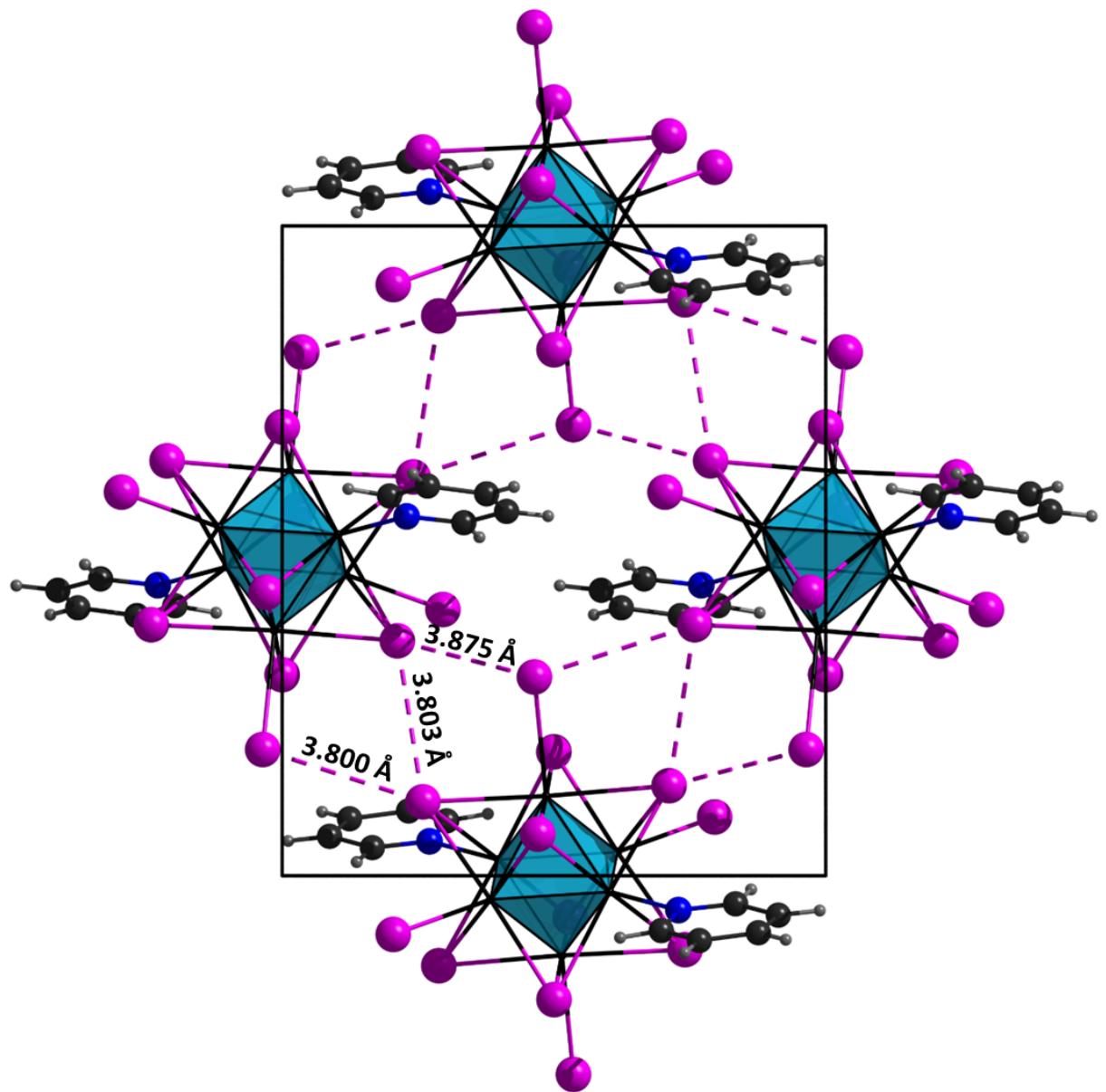


Figure S5. $\text{I}^{\text{i}}\text{-}\text{I}^{\text{a}}$ and $\text{I}^{\text{i}}\text{-}\text{I}^{\text{i}}$ interactions in $(\text{pyH})_{0.2}[\{\text{Mo}_6\text{I}_8\}(\text{py})_{1.8}\text{I}_{4.2}]\cdot 1.8\text{py}$.

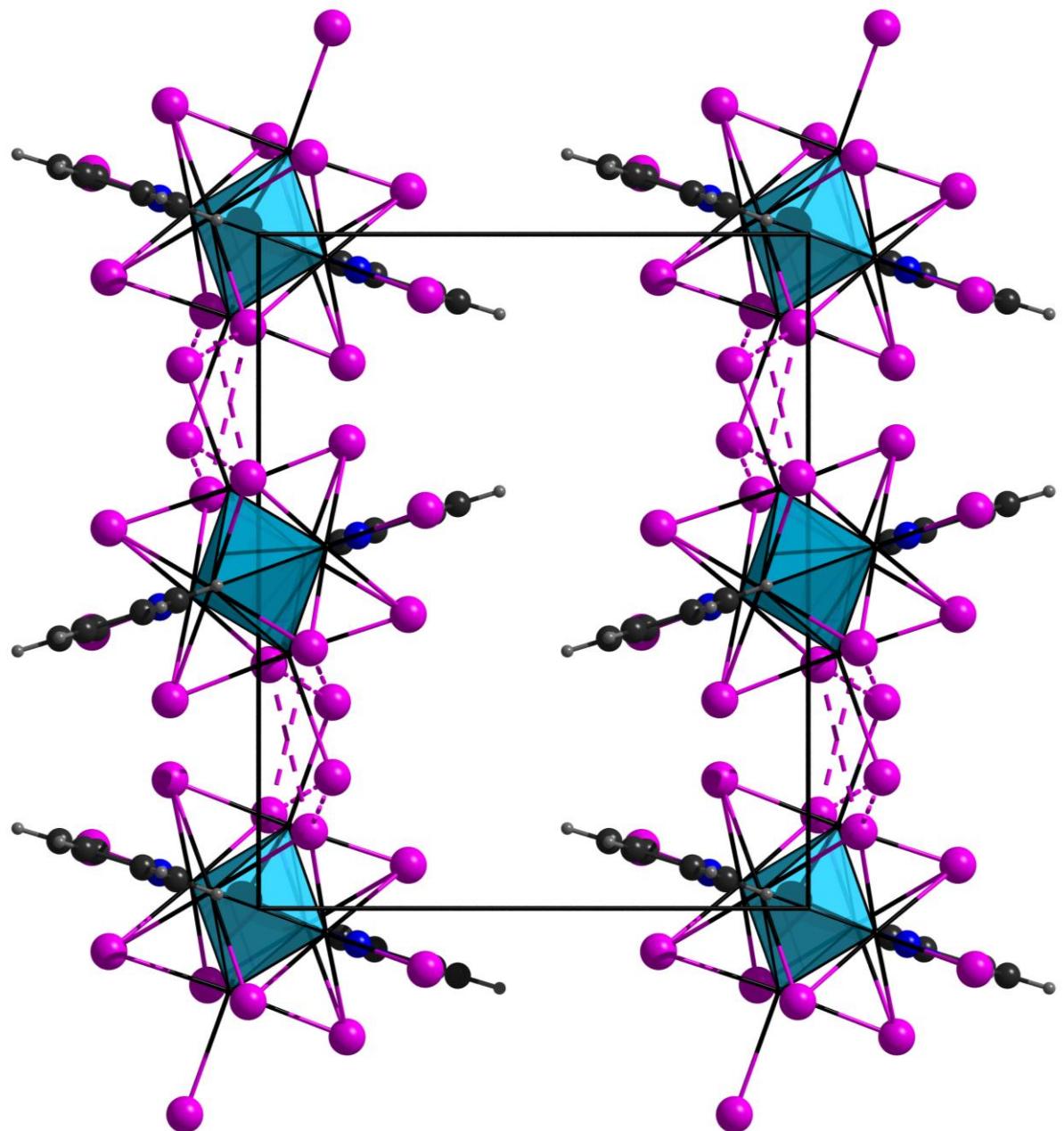


Figure S6. Layers in $(\text{pyH})_{0.2}[\{\text{Mo}_6\text{I}_8\}(\text{py})_{1.8}\text{I}_{4.2}]\cdot 1.8\text{py}$.