

Article



Assessment of Surface Water Quality Using Water Quality Index and Discriminant Analysis Method

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Abstract: Given the complexity of water quality data sets, water resources pose a significant problem for global public order in terms of water quality protection and management. In this study, surface water quality for drinking and irrigation purposes was evaluated by calculating the Water Quality Index (WQI) and Irrigation Water Quality Index (IWQI) based on nine hydrochemical parameters. The discriminant analysis (DA) method was used to identify the variables that are most responsible for spatial differentiation. The results indicate that the surface water quality for drinking is of poor and very poor quality according to the WQI values, however, the IWQI values indicate that the water is acceptable for irrigation with restrictions for salinity sensitive plants. The discriminate analysis method identified pH, potassium, chloride, sulfate, and bicarbonate as the significant parameters that discriminate between the different stations and contribute to spatial variation of the surface water quality. The findings of this study provide valuable information for decision-makers to address the important problem of water quality management and protection.

Keywords: water quality; drinking water; irrigation water; water quality index (WQI); hydochemical parameters; discriminant analysis, Koudiat Medouar watershed, Algeria

1. Introduction

Water is essential for the survival of all living organisms and plays a vital role in the growth and development of life on Earth [1–6]. Therefore, it is imperative to take immediate action to preserve water resources through effective management and monitoring of water quality [7–11]. The usage of water for domestic purposes accounts for only 5%, with 20% being utilized by the industrial sector, leaving 75% for agricultural use [12,13]. It is crucial to implement a comprehensive program for water resource development that places emphasis on water quality management and monitoring.

It is crucial to evaluate the quality of water before it is used for human consumption [14–16]. The pollution of surface waters is a direct result of industrial development, as well as the excessive and uncontrolled use of fertilizers by man, in addition to natural processes such as the dissolution of rocks through water-rock interaction [17]. The assessment of water quality can be conducted through the use of the water quality index (WQI) model, which primarily depends on the physicochemical parameters of water, to monitor and control the quality of water intended for human consumption [18–22].

The study of the spatio-temporal variation of surface waters is a complex phenomenon, but the use of outdated methods can lead to suboptimal performance of the model and redundant results. In this study, factor analysis (FA), discriminant analysis (DA), cluster analysis (CA), and principal component analysis (PCA) methods were used to evaluate the



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). quality of surface waters and reduce the number of monitoring results and the identification of measurement stations of comparable quality, as well as to separate measurement stations where the water quality deviates from the quality of other measurement stations [23,24]. The objectives of this study are to understand the current state of water quality, determine the suitability of surface waters for human consumption, provide scientific recommendations for water resource managers and, finally, identify the most influential parameters and those responsible for the spatial differentiation of water quality using the discriminant analysis (DA) method.

2. Materials and Methods

2.1. Study Area

The Koudiat Medouar watershed is located in northeastern Algeria, in the Aurès region, and is a part of the vast watershed of the Constantine Highlands, which has a total area of 9500 km². It is between four huge basins to the north of Kebir Rhumel and Seybouse, to the east of Medjerda, to the south of Chot Hodna and Melghir, and to the west of Summam. It has a 590 km² land area. It is situated between the longitude of $35^{\circ}30'57''$ North and the latitude of $6^{\circ}30'48''$ East (Figure 1). In northeastern Algeria, the Koudiat Medouar watershed contributes significantly to the surface water mobilization and transfer system in the high plains of Constantine and the Aurès. The Oued Chemorah basin provides water for the dam, as well as sub-basins, wastewater, and industrial water from neighboring municipalities. Due to recent droughts in the region, authorities have also utilized water from the Beni Haroun dam, which has a capacity of 74 million cubic meters. This region's climate is semi-arid, with a cold and humid winter and a hot and dry summer, given the semi-arid character of the study region which results in a more or less marked drought by a very high potential evaporation rate which is estimated at 1747.34 mm per year. The average annual rainfall and temperature are around 370 mm and 15 °C, respectively [25].



Figure 1. Location of the study area and sampling points.

2.2. Sampling of Surface Water

The following sample stations were selected depending on the degree of pollution in the research area: Dam station, Oued Reboua station, Oued Timgad station, and Wastewater station. The surface water samples were collected over a two-year period (2019 to 2020), with two samples taken at the beginning and end of each month in accordance with conventional sampling protocols [26] guidelines on consumption and irrigation from the International Organization for Standardization (ISO) [27], the Bureau of International Standards (BIS) [28], and the Food and agriculture organization (FAO) [29]. The samples from the sampling stations were placed in polyethylene bottles, which were then transported in coolers to the analysis laboratory. Nitric acid (HNO₃) was added to the cationic solutions to achieve a pH of 2, but the ion balance is computed with errors of less than 10%. Table 1 summarizes the findings.

Table 1. Instrumental and titrimetric techniques used for the chemical analysis of samples of surface water.

Characteristics	Analytical Method	Range	Unit
pH	pH/EC/TDS meter	pH 4.7 and 9.2	o. /
Electrical conductivity	pH/EC/TDS meter	Potassium chloride	μS/cm
Total dissolved solids (TDS)	Calculation	EC X (0.55–0.75)	mg/L
Cotal hardness (CaCO ₃)	EDTA titrimetric	EDTA, ammonia buffer	mg/L
		and Eriochrome	
		Black-T (EBT) indicator	
Calcium (Ca)	EDTA titrimetric	EDTA, sodium hydroxide	mg/L
		and murexide	
Magnesium (Mg)	Calculation	$MgH = TH - CaH;Mg = MgH \times Eq$	mg/L
		Wt of Mg \times Normality of EDTA	
Sodium (Na)	Flame photometric	Sodium chloride (NaCl) and KCl	mg/L
Potassium (K)	Flame photometric	NaCl and KCl	mg/L
Bicarbonates (HCO ₃)	Titrimetric	Hydrosulfuric acide (H_2SO_4)	mg/L
		phenlphthalein and methyl orange	
Chloride (Cl)	Titrimetric	Silver nitrate (AgNO ₃)	mg/L
		potassium chromate	
Sulfator $(SO4)$	UV-visible	HCL athyl alcohl NaCl	ma/I
Sullates (304)	spectrophotometer	rici, euryr acom, Naci	ing/L
		barium chloride, sodium sulfate	
Nitrate (NO_{2})	UV-visible	Potassium pitrate (KNO_{2})	mg/I
(1103)	spectrophotometer	i otassium intrate (KivO3)	IIIE/ L
		Phenol disulfonic acide, ammonia	
	CharacteristicspHElectrical conductivityTotal dissolved solids (TDS)Total hardness (CaCO3)Calcium (Ca)Magnesium (Mg)Sodium (Na) Potassium (K)Bicarbonates (HCO3)Chloride (Cl)Sulfates (SO4)Nitrate (NO3)	CharacteristicsAnalytical MethodpHpH/EC/TDS meterElectrical conductivitypH/EC/TDS meterTotal dissolved solids (TDS)CalculationCalcium (Ca)EDTA titrimetricCalcium (Ca)EDTA titrimetricMagnesium (Mg)CalculationSodium (Na) Potassium (K)Flame photometric Flame photometric TitrimetricChloride (Cl)TitrimetricSulfates (SO4)UV-visible spectrophotometerNitrate (NO3)UV-visible spectrophotometer	CharacteristicsAnalytical MethodRangepHpH/EC/TDS meterpH 4.7 and 9.2Electrical conductivitypH/EC/TDS meterPotassium chlorideTotal dissolved solids (TDS)CalculationEC X (0.55–0.75)otal hardness (CaCO3)EDTA titrimetricEDTA, ammonia buffer and EriochromeCalculur (Ca)EDTA titrimetricEDTA, sodium hydroxide and murexideMagnesium (Mg)CalculationMgH = TH - CaH;Mg = MgH × Eq Wt of Mg × Normality of EDTASodium (Na)Flame photometric Flame photometricSodium chloride (NaCl) and KCl Hydrosulfuric acide (H2SO4) phenlphthalein and methyl orange Silver nitrate (AgNO3) potassium chromateSulfates (SO4)UV-visible spectrophotometerSilver nitrate (AgNO3) potassium chloride, sodium sulfateNitrate (NO3)UV-visible spectrophotometerHCI, ethyl alcohl, NaCl barium chloride, sodium sulfate Potassium nitrate (KNO3) Phenol disulfonic acide, ammonia

2.3. Water Quality Index (WQI)

In this study, two equations were used to evaluate the quality of surface water intended for consumption and irrigation. The first is the water quality index (WQI), which is classified into five different suitability classes (Table 2). The second is the Irrigation Water Quality Index (IWQI), which is classified into four different suitability classes (Table 3). Both equations abide to the parameters proposed by the World Health Organization (WHO).

In the present research, the WQI was calculated using seven hydrochemical parameters: Ca, Mg, Na, K, Cl, SO₄, and HCO₃. There were three processes involved in calculating WQI.

First, each hydrochemical measure was allocated a weight based on its relative relevance to the overall quality of drinking and irrigation water.

The parameters (Ca and HCO₃) have significant influence on the quality of surface waters, and hence the greatest weight of 5 was allocated to them. In contrast, the parameter (K), which is believed to be non-harmful, was assigned the lowest weight of 2. (Table 4).

Step two involves calculating the relative weight (Wi) of each hydrochemical parameter using Equation (1).

$$W_i = \frac{w_i}{\sum_{i=1}^n w_i} \tag{1}$$

where W_i is the relative weight, w_i is the weight of each parameter, and n is the number of parameters.

Step three: for each hadrochemical parameter, the quality scale (qi) is computed by dividing its concentration in each water sample by its respective standard [29,30], and then multiplying the result by 100 in accordance with Equation (2).

$$q_i = \frac{C_i}{S_i} * 100 \tag{2}$$

where q_i is the quality grading, C_i is the concentration of each hydrochemical parameter in each water sample, and S_i is the concentration of standard hydrochemical concentration either for drinking water or for irrigation purposes proposed by different organizations [29,30].

Finally, Water Quality Index (WQI) is calculated by the following Equation:

$$WQI = \sum_{i=1}^{n} w_i * q_i \tag{3}$$

where WQI and IWQI are the quality indexes used to evaluate the quality of drinking water and irrigation water, respectively. W_i denotes the relative weight of each hydrochemical parameter, and q_i refers to the quality rating.

Table 2. Classification of Water Quality Index (WQI) of drinking water.

WQI Values	Water Quality for Drinking Purpose
Less than 25	Excellent
26–50	Good
51–75	Poor
76–100	Very poor
More than 100	Unsuitable

Table 3. Categorization of Water Quality Index (IWQI) of irrigation water.

IWQI Values	Class	Restriction
Less than 150	Ι	None
150-300	Π	Slight
301-450	III	Moderate
More than 450	IV	Severe

Table 4. Weight and relative weight of the hydrochemical parameters.

Parameters	Ca	Mg	Na	К	Cl	SO_4	HCO ₃
WHO (2017)	75	50	200	10	250	250	500
FAO (1985)	400	60	920	2	1065	1920	610
				Weight (w_i)			
Dam	5	3	4	2	4	3	5
Oued Rboua	5	3	4	2	3	4	5
Oued Timgad	5	3	4	2	3	4	5
Wastewater	5	3	4	2	3	4	5
			Re	elative weight (W_i)		
Dam	0.1923	0.1154	0.1538	0.0769	0.1538	0.1154	0.1923
Oued Rboua	0.1923	0.1154	0.1538	0.0769	0.1154	0.1538	0.1923
Oued Timgad	0.1923	0.1154	0.1538	0.0769	0.1154	0.1538	0.1923
Wastewater	0.1923	0.1154	0.1538	0.0769	0.1154	0.1538	0.1923

WHO: World Health Organization (2017); FAO: Food and Agriculture Organization (1985).

2.4. National Sanitation Foundation Water Quality Index

Seven parameters calcium (Ca²⁺), magnesium (Mg²⁺), potassium (K⁺), sodium (Na⁺), chloride (Cl⁻), sulfate (SO₄²⁻), and bicarbonate (CaCO₃) were used to calculate the NS-FQWI. The graphical representation shows that the water quality level varies from 0 (very poor) to 100 (excellent), the value of Q is determined by comparing the values of these parameters to this weighting curve, and the value of Q is multiplied by a weighting factor which is based on the degree its importance in the calculation (Wi) [31].

NSFWQI classes are represented in intervals that express water quality: very poor (0–25), poor (25–50), average (50–70), good (70–90), and excellent (90–100).

2.5. Discriminant Analysis (DA)

The technique of discriminant analysis is used to demonstrate the connection and linear combinations between two or more independent metric variables. Calculating the discriminant function is as follows:

$$f(G_i) = K_i + \sum_{j=1}^n W_{ij} \cdot P_{ij}$$

$$\tag{4}$$

where *i* represents the number of groups or stations (*G*). In the current study, four stations (groups) were selected: K_i represents the constant inherent to each group; *n*, is the number of variables used to categorize a set of data into a given class, W_j , is the weight coefficient, assigned by DA to a specified selected parameter P_j ; and P_j , is the analytical value of the selected variable. In this study, the independent variables were the hydrochemical parameters in the four stations.

All statistical and multivariate data analysis methods, as well as visualization, were performed using R version 4.1.3 and SPSS version 28.0.1.

3. Results and Discussion

3.1. Descriptive Statistics

The descriptive statistics of surface water hydrochemical parameters may offer an approximation of the hydrochemical characteristics of a region. Table 5 and Figure 2 provide statistics for each station's surface water hydrochemical parameters. The pH range of the surface water samples was 6.8 to 8.6, suggesting that the surface water in the research region was mainly neutral or slightly alkaline [32,33]. Electrical conductivity (EC) values for Dam, Oued Reboua, Oued Timgad, and Wastewater station were 1349, 1010, 1326, and 1173 μ S/cm, respectively. The coefficient of variation (in percent) of electrical conductivity was 22.42, 25.55, 8.81, and 15.82 for Dam, Oued Reboua, and Wastewater stations, respectively, indicating that the temporal variability of electrical conductivity was moderate at Dam, Oued Reboua, and Wastewater stations and low at Oued Timgad station. Ca > Na > Mg > K were the four locations with the highest average cation concentrations, as seen in Figure 2. In all stations, calcium was the predominant cation, followed by sodium. Dam, Oued Reboua, Oued Timgad, and Wastewater station had Ca concentrations of 94.04, 120.24, 116.93, and 107.84 mg/L, respectively. The Ca content in all surface water tests exceeded the WHO (2006) permissible limit [34]. $HCO_3 > SO_4 > Cl$ was the predominant anion in the Oued Reboua, Oued Timgad, and Wastewater sites. In the case of the Dam station, $HCO_3 > Cl > SO_4$ was the dominant anion. Bicarbonate was the most prevalent anion in all four sites, followed by sulfate at Oued Reboua, Oued Timgad, and Wastewater and chloride at the Dam station. The main concentration of HCO₃ was 173.21, 307.28, 500.51, and 758.38 mg/L in the four stations, respectively. The highest value of HCO₃ was observed in the Wastewater and Oued Timgad stations.

	pН	EC	Ca	Mg	Na	К	Cl	SO_4	HCO ₃	WQI	IWQI
						Dam					
Min	7.4	1040	88.18	39.36	80.00	0.812	75.25	78.00	134.20	55.36	23.50
Max	7.8	1800	99.70	45.49	103.30	1.289	124.25	92.00	201.30	62.18	27.80
Mean	7.5	1349	94.04	42.72	93.92	1.007	90.32	85.77	173.21	58.15	25.46
SD	0.14	302.44	4.14	2.07	6.22	0.14	13.74	4.97	19.87	1.98	1.45
CV	1.90	22.42	4.40	4.85	6.62	14.40	15.21	5.79	11.47	3.41	5.69
_					(Oued Rboua	a				
Min	7.5	680	88.18	21.48	93.70	0.908	17.75	75.00	231.80	50.88	22.39
Max	8.4	1230	136.27	45.44	120.70	1.638	53.25	98.00	372.10	78.22	36.22
Mean	7.75	1010	120.24	37.67	109.35	1.371	35.50	91.00	307.28	68.05	30.93
SD	0.44	258.07	21.95	10.97	12.38	0.35	14.49	10.74	72.25	11.94	6.21
CV	5.62	25.55	18.26	29.11	11.32	25.19	40.82	11.80	23.51	17.55	20.09
					C	Dued Timga	d				
Min	7.1	1200	108.20	30.60	80.00	0.812	35.50	86.00	402.60	69.45	32.01
Max	8.6	1500	132.26	47.84	130.50	1.835	71.00	115.00	591.70	84.75	41.26
Mean	7.9	1326	116.93	37.49	105.64	1.257	45.18	101.32	500.51	75.30	36.51
SD	0.49	116.90	8.39	5.14	15.02	0.35	14.56	10.96	58.76	5.39	3.34
CV	6.30	8.81	7.17	13.72	14.22	28.09	32.22	10.82	11.74	7.16	9.15
						Wastewater					
Min	6.8	1010	92.18	27.84	82.00	0.830	51.57	88.00	597.80	72.47	38.53
Max	7.6	1660	128.30	38.28	133.50	1.815	106.50	98.00	1000.40	95.26	54.44
Mean	7.1	1173	107.84	33.48	101.53	1.30	65.87	92.27	758.38	82.07	43.68
SD	0.28	185.53	11.50	4.51	19.74	0.36	21.77	3.53	116.18	6.11	4.59
CV	3.96	15.82	10.67	13.48	19.44	27.71	33.06	3.83	15.32	7.44	10.52

Table 5. Summary statistics of hydrochemical parameters of the surface water samples.

All parameters are expressed in mg/L while pH on scale and EC in μ S/cm; WQI: water quality index for drinking purposes. IWQI: water quality index for irrigation purpose; Min: minimum; Max: maximumSD: standard deviation; CV: coefficient of variation (in %).



Figure 2. Boxplot of the hydrochemical parameters of the surface water.

3.2. Evaluation of Surface Water Quality

This research used the water quality index for drinking (WQI) and irrigation (IWQI) water purposes for assessing surface water quality in terms of analyzed parameters. Table 5 displays the summary statistics for each station's water quality index. Boxplots of WQI and IWQI for each station are shown in Figure 3.

The calculated WQI values were divided into five categories: excellent, good, poor, very poor, and unsuitable for human consumption. Dam, Oued Reboua, Oued Timgad, and Wastewater station had WQI values of 58.15, 68.05, 75.30, and 82.07, respectively. The Wastewater station's water samples had the highest WQI values. Estimated WQI values varied from 50.88 to 95.26, indicating that none of the surface water samples contain 'Excellent' or 'Good' drinking water (WQI 50). As demonstrated in Figure 3, all surface water samples collected at the Dam station were of poor quality. A total of 75% of the water samples collected at the Oued Reboua station were judged to be of poor quality. However, almost all water samples from the Oued Timgad and Wastewater sites were determined to be of very poor quality.

The Water Quality Indices for irrigation (IWQI) reveal the level of pollution based on the restriction standards established for each class. According to Table 4, zone one (I) is the unrestricted class, zone two (II) expresses the class requiring restriction, zone three (III) indicates the moderate restriction class, and zone four (IV) expresses the severe restriction class [35].

Figure 3 displays that the highest surface water quality index values for irrigation in the research region were 27.80, 36.22, 41.26, and 54.44, respectively. However, these values indicate that they are classed in zone one, which indicates class (I) of no restrictions on soils and crops.



Figure 3. Boxplot of water quality index: WQI for drinking purposes (**left**) and IWQI for irrigation purposes (**right**).

The NSFWQI values presented in Table 6 show that all the samples were classified as medium (NSFWQI = 50–70), except for the samples from the dam station which were classified as good (NSFWQI = 72.82); this increase is due to the decrease in potassium (K⁺) and chloride (Cl⁻) concentrations [31].

Variables	Ca	Mg	Na	К	Cl	SO ₄	HCO ₃
			Weightin	ng factor			
Dam	0.19	0.11	0.15	0.07	0.11	0.15	0.19
Oued Reboua	0.19	0.11	0.15	0.07	0.11	0.15	0.19
Oued Timgad	0.19	0.11	0.15	0.07	0.11	0.15	0.19
Wastewater	0.19	0.11	0.15	0.07	0.11	0.15	0.19

Table 6. Results of NSFWQI for water quality of stations.

Variables	Ca	Mg	Na	К	Cl	SO ₄	HCO ₃
			W _i I _i Y	Values			
Dam	13.31	8.24	12.73	4.52	8.24	12.73	13.05
Oued Reboua	13.06	8.53	12.29	2.89	7.61	11.15	13.02
Oued Timgad	13.58	8.95	11.61	1.98	6.81	10.32	12.66
Wastewater	10.08	6.32	7.43	1.03	5.35	9.28	11.09
			NSF	WQI			
Dam	72.82						
Oued Reboua	68.55						
Oued Timgad	65.91						
Wastewater	50.58						

Table 6. Cont.

3.3. Correlation Analysis

In addition to identifying the origins of the hydrochemical formation process and establishing the degree of relevance between elements [36], the correlation research was used to demonstrate the numerous types of correlations between the physico-chemical parameters evaluated in the study region.

Using the corrplot package in R [37], the correlation matrix between the hydrochemical parameters, WQI, and IWQI for each station is presented in Figure 4 in this section. The correlation matrix was created using Pearson's correlation analysis.



Figure 4. Correlation analysis of hydrochemical parameters for each station (*p*-value < 0.01).

3.3.1. Dam Station

Figure 4 demonstrates that EC had a negative correlation with Mg, HCO₃, SO₄, and IWQI, indicating that ions are prevalent in the research region. HCO₃, Cl, and IWQI all correlate positively with WQI. IWQI demonstrated a statistically significant positive correlation with Mg, HCO₃, Cl, and SO₄. In addition, a substantial positive correlation was identified between Mg-HCO₃ and SO₄-K.

3.3.2. Oued Reboua Station

Ca had a substantial positive correlation with SO_4 according to the data, suggesting that both parameters may have the same source. In addition, a substantial association was found between IWQI and K. (0.99).

3.3.3. Oued Timgad Station

Figure 4 shows that IWQI, Cl, SO₄, Mg, Na, and K exhibited a substantial positive correlation with WQI, suggesting that these hydrochemical parameters have a considerable impact on water quality. IWQI was positively correlated with Cl, SO₄, Mg, and NA. There was a substantial correlation between potassium (K) and pH, Cl, SO₄, and Na.

3.3.4. Wastewater Station

Figure 4 reveals that the bicarbonate parameter had a strong correlation with WQI and IWQI, showing that HCO₃ plays a significant role in influencing the surface water quality in the research region. The sodium (Na) parameter correlated positively with EC, Cl, Mg, and K, but negatively with pH and Ca. In addition, a high positive correlation between Ca and Mg suggests that these parameters may have the same source.

Significant correlations were explained by the dissolution of the rock's mineral composition and the exchange of ions, which generates precipitation, human activity, and the addition of chemical fertilizers that pollute surface waters via runoff and leaching from agricultural land [32,33].

3.4. Spatial Variation in Surface Water Quality

The spatial variation in surface water quality was analyzed using discriminant analysis (DA) conducted on the raw dataset consisting of 10 variables grouped into four stations (pH, EC, Ca, Mg, Na, K, Cl, SO₄, HCO₃, WQI, and IWQI). The stepwise mode of DA was employed, in which at each step, one variable that minimized the overall Wilks' Lambda statistic was added or removed. The aim was to identify the most crucial variables that contributed to the differences among the various stations. The independent variables were the main hydrochemical parameters that were measured, and the dependent variable consisted of four stations.

Since the *p*-values are less than the 0.01 level of significance, the Wilks' Lambda statistic for group mean equality is significant. Test F for Wilks' Lambda was used to identify the factors that substantially contributed to station differentiation. Table 7 displays the ANOVA findings. From the results, we can see that the Test F is significant for eight variables (pH, Ca, Mg, Cl, SO₄, HCO₃, WQI, and IWQI) out of eleven variables at *p*-level < 0.05. The three variables of electrical conductivity (EC), sodium (Na), and potassium (K), for which the significance (Sig.) is higher than 0.01, should be eliminated from the model.

Table 8 provides an overview of canonical discriminant functions. Four stations were analyzed using three discriminant functions (DFs), and therefore, three eigenvalues in a discriminant analysis.

The first discriminant function has the greatest eigenvalue, suggesting that it has the strongest discrimination power of the three. From Table 8, we can see that the first function explained 77.2% of the total variance between stations, compared to the other two functions that unite the studied parameters, indicating a dispersion of less than 20%. However, the canonical correlation coefficient, which measures the ratio between the coordinates of the

discriminating factor and the grouping variable, explained only 94.3%, or (0.971)2, of the total variance, indicating a difference.

Variables	Wilks' Lambda	F	df1	df2	p-Level (Sig.)
pН	0.555	8.808	3	33	0.000
ĒC	0.773	3.226	3	33	0.035
Ca	0.502	10.924	3	33	0.000
Mg	0.649	5.961	3	33	0.002
Na	0.868	1.674	3	33	0.192
Κ	0.823	2.361	3	33	0.089
Cl	0.388	17.358	3	33	0.000
SO_4	0.586	7.777	3	33	0.000
HCO ₃	0.087	116.063	3	33	0.000
WQI	0.250	33.031	3	33	0.000
IWQI	0.193	45.914	3	33	0.000

Table 7. Tests of group mean quality for discriminant analysis.

Table 8. Eigenvalues for the discriminant analysis.

Function	Eigenvalue	% of Variance	Cumulative %	Canonical Correlation
1	16.481 ^a	77.2	77.2	0.971
2	4.324 ^a	20.2	97.4	0.901
3	0.553 ^a	2.6	100	0.597
3 11 1 0	1 1	1.	.1 1 .	

^a First 3 canonical discriminant functions were used in the analysis.

Table 9 displays the Wilks' Lambda and Chi-square statistics for each discriminant function (DF). According to Table 8, the Wilks' Lambda and Chi-square statistics for each discriminant function (DF) were between 0.007 and 0.64 and 13.859 and 156.66, respectively. Wilk's Lambda statistic indicated that all three functions were statistically significant since their values are were less than 0.01, showing that the spatial discriminant analysis was effective and significant.

Table 9. Wilks' Lamda and Chi-square test for the discriminant analysis.

Test of Function(s)	Wilks' Lambda	Chi-Square	df	<i>p</i> -Level (Sig.)
1 through 3	0.007	156.660	15	0.000
2 through 3	0.121	66.535	8	0.000
3	0.644	13.859	3	0.003

Considered discriminating variables are expressed in various units of measurement [38]. The coefficients of the discriminant function are used to determine the discriminant score for each case. Given that the first function has the largest discriminating power, it is clear to focus on analyzing its output. The findings of the first discriminant function demonstrate that bicarbonate (HCO₃) was an important hydrochemical parameter for discriminating the spatial variation in surface water quality across four stations. Table 10 presents the normalized coefficients of the discriminant function.

 Table 10. Standardized canonical discriminant function coefficients.

x7 · 11		Function	
Variables	1	2	3
pН	0.432	0.592	0.308
K	-0.403	0.668	-1.023
Cl	0.723	-0.937	0.536
SO ₄	0.305	0.650	0.995
HCO ₃	-1.057	-0.272	0.136

The structure matrix coefficient represents the relationship between every predictor variable and the discriminant function. Although the correlation coefficient of bicarbonate, which admits a high value, is strongly correlated with the first discriminant function, the chloride and pH are strongly correlated with the second discriminant function, whereas the coefficients of sulfate and potassium are well correlated with the third discriminant function. The values of the calculated structural coefficients are shown in Table 11.

** • • •		Function	
Variables	1	2	3
HCO ₃	-0.791 *	-0.157	0.497
IWQI ^b	-0.529 *	-0.061	0.346
Cl	0.156	-0.515 *	0.244
рН	0.106	0.366 *	0.252
EC ^b	-0.039	-0.110 *	-0.105
SO ₄	-0.106	0.291	0.529 *
WQI ^b	-0.370	-0.042	0.472 *
Ca ^b	0.072	0.172	0.363 *
Na ^b	-0.102	-0.012	-0.199 *
Κ	-0.095	0.103	-0.193 *
Mg ^b	-0.013	0.027	0.117 *

Table 11. The structure matrix for the discriminant analysis.

Pooled within-groups correlations between discriminating variables and standardized canonical discriminant functions; variables ordered by absolute size of correlation within function. * Largest absolute correlation between each variable and any discriminant function; ^b this variable not used in the analysis.

The first discriminant function clearly illustrates the spatial variation between the four sites, as seen in Table 12.

Tables 13 and 14 show the resulting classification matrices (CMs) and classification functions (CFs) from the discriminant analysis. From the findings, we can observe that the DA properly categorized 97.3% of the cases that were originally grouped. According to Tables 13 and 14, the DA effectively categorized 94.6% of the initially grouped cases in the classification matrix. Based on the findings of discriminant analysis, the most significant discriminant variables across the four geographical stations (Table 13) were pH, potassium, chloride, sulfate, and bicarbonate, causing the majority of expected surface water quality variations.

 Table 12. The canonical discriminant function coefficients.

** • • •	Function				
Variables	1	2	3		
pН	1.237	1.697	0.881		
ĸ	-1.312	2.177	-3.334		
Cl	0.043	-0.056	0.032		
SO ₄	0.040	0.085	0.130		
HCO ₃	-0.014	-0.004	0.002		
(Constant)	-7.777	-18.124	-17.580		

Unstandardized coefficients.

Table 13. The classification function coefficients.

Variables -	Station				
	Dam	Oued Rboua	Oued Timgad	Wastewater	
pН	114.003	114.213	113.702	101.158	
K	15.780	36.539	29.461	28.856	
Cl	0.757	0.285	0.319	0.342	
SO_4	2.816	2.761	2.983	2.370	
HCO ₃	-0.158	-0.117	-0.089	-0.019	
(Constant)	-580.199	-581.729	-602.609	-493.412	

Fisher's linear discriminant functions.

		Station –	Predicted Group Membership				
			Dam	Oued Rboua	Oued Timgad	Wastewater	Iotal
Original		Dam	11	0	0	0	11
		Oued Rboua	0	4	0	0	4
	Count	Oued Timgad	0	1	10	0	11
		Wastewater	0	0	0	11	11
		Dam	100	0	0	0	100
	0/	Oued Rboua	0	100	0	0	100
	70	Oued Timgad	0	9.1	90.9	0	100
		Wastewater	0	0	0	100	100
Cross- validated b		Dam	11	0	0	0	11
		Oued Rboua	0	3	1	0	4
	Count	Oued Timgad	0	1	10	0	11
		Wastewater	0	0	0	11	11
	0/	Dam	100	0	0	0	100
		Oued Rboua	0	75	25	0	100
	70	Oued Timgad	0	9.1	90.9	0	100
		Wastewater	0	0	0	100	100

Table 14. Classification results ^{a,c}.

^a 97.3% of original grouped cases correctly classified. ^b Cross validation is performed only for those cases in the analysis. In cross validation, each case is classified by the functions derived from all cases other than that case. ^c A total of 94.6% of cross-validated grouped cases correctly classified.

4. Conclusions

In the context of this work, physico-chemical parameters (pH, EC, Ca, Mg, Na, K, Cl, SO₄, and HCO₃) were used to calculate the surface water quality indexes for potability (WQI) and irrigation (IWQI). All samples were collected monthly at four sites throughout 2019 and 2020.

Discriminant analysis (DA) is a crucial technique for producing a thorough interpretation of the acquired data, allowing managers to use them for improved surface water pollution management of water resources.

Based on the WQI, the results show that the water samples from the four stations are of poor and very poor quality for potability. According to the IWQI, the water quality of the research area is adequate for irrigation, with restrictions for salinity-sensitive plants, while the NSFWQI reflects the same water quality as the IWQI or samples are classified as fair water, except for samples from dam stations which classified as good quality water.

Using discriminant analysis, the spatial variation in surface water quality across the four sites was examined (DA). The discriminant analysis revealed the presence of pH, potassium, chloride, sulfate, and bicarbonate. These discriminating parameters are the most relevant at the four sites and contribute in the majority of cases to the predicted changes in surface water quality.

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