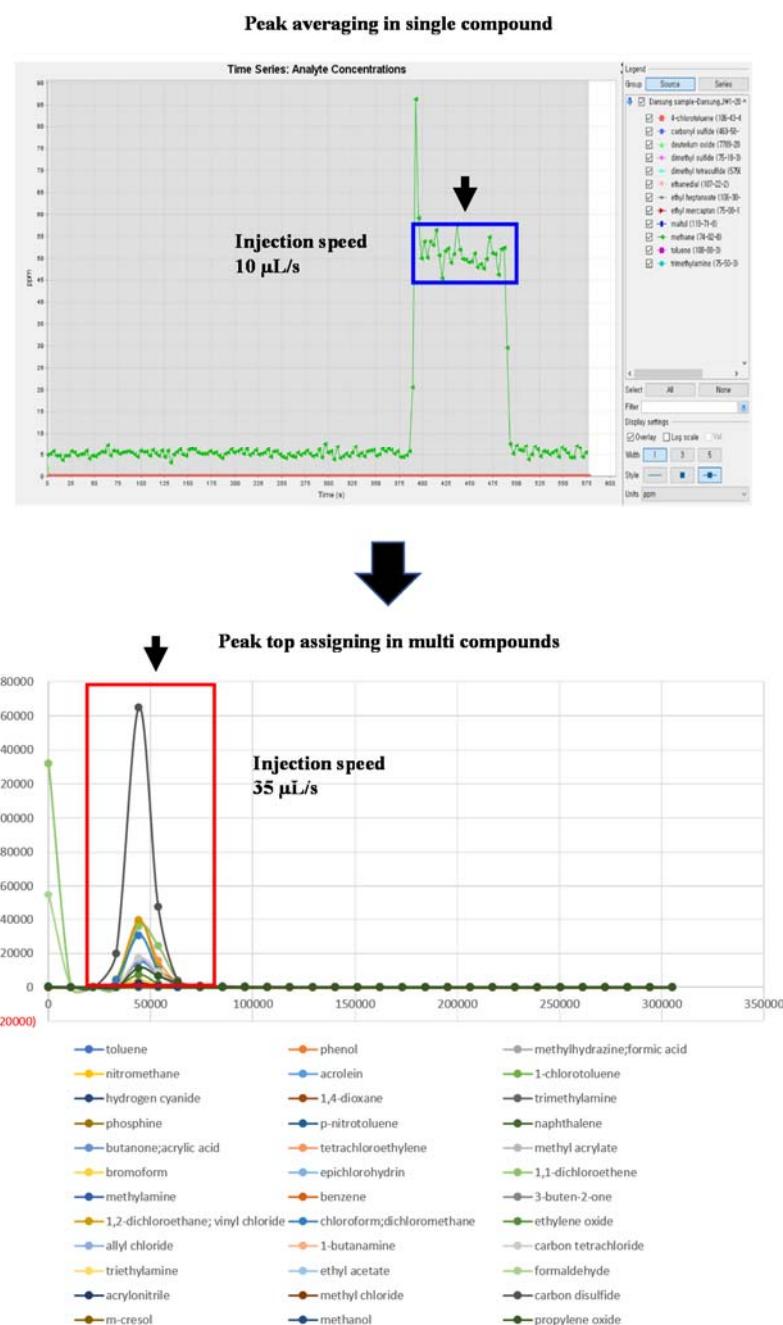


Article

# Direct mass spectrometry with online headspace sample pretreatment for continuous water quality monitoring

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**Figure S1.** Data assignment under headspace extraction and injection conditions(upper: single compound injection, lower: multi compound injection).

**Table S1.** Physico/chemical properties of target compounds for chemical accidents and leakages.

Compounds	CAS no	Elemental composition	Molecular weight	Vapor pressure (mmHg) @ 25°C	Solubility (mg/L) @ 25°C
Toluene	108-88-3	C7H8	92.14	28.4	526
Phenol	108-95-2	C6H6O	94.11	0.35	82 800
Formic acid	64-18-6	CH2O2	46.025	42.59 (extrapolated)	1 000 000
Nitromethane	75-52-5	CH3NO2	61.04	35.8	111 000
Acrolein	107-02-8	C3H4O	56.06	274	212 000
Benzyl chloride	100-44-7	C7H7Cl	126.58	1.23	525
Acrylamide	79-06-1	C3H5NO	71.08	0.007	2 040 000
1,4-dioxane	123-91-1	C4H8O2	88.11	38.1	800 000
Trichloroethylene	79-01-6	C2HCl3	131.38	69	1 280
p-nitrotoluene	99-99-0	C7H7NO2	137.14	0.0157 (extrapolated)	361
Naphthalene	91-20-3	C10H8	128.17	0.085	31
Methyl ethyl ketone	78-93-3	C4H8O	72.11	90.6	223 000
Tetrachloroethene	127-18-4	C2Cl4	165.8	18.5	206
Methyl acrylate	96-33-3	C4H6O2	86.09	86.6	49 400
Bromoform	75-25-2	CHBr3	252.73	5.4 (estimated)	3 100
1,1-dichloroethylene	75-35-4	C2H2Cl2	96.94	600	2 420
Benzene	71-43-2	C6H6	78.11	94.8	17 900
Methyl vinyl ketone	78-94-4	C4H6O	70.09	152	100 000
Vinyl chloride	75-01-4	C2H3Cl	62.5	2980	8 800
Dichloromethane	75-09-2	CH2Cl2	84.93	435	13 200
Ethylene oxide	75-21-8	C2H4O	44.05	1310	237
Allyl chloride	107-05-1	C3H5Cl	76.52	368	3.37
Carbon tetrachloride	56-23-5	CCl4	153.82	115	793
Ethyl acetate	141-78-6	C4H8O2	88.11	93.2	80 000
Formaldehyde	50-00-0	CH2O	30.03	10.5 @ 20°C	400 000

Acrylonitrile	107-13-1	C3H3N	53.06	109	74 500
Methyl chloride	74-87-3	CH3Cl	50.49	4300	5 040
Carbon disulfide	75-15-0	CS2	76.14	359	2 160
m-cresol	108-39-4	C7H8O	108.14	0.11	22 700
Methanol	67-56-1	CH3OH	32.04	127	1 000 000
Propylene oxide	75-56-9	C3H6O	58.08	538	590 000

**Table S2.** Reaction ion, mass, theoretical and measured reaction ratio of target compounds.

Compounds	Reaction Ion	m/z	Reactio n ratio	Compounds	Reaction Ion	m/z	Reactio n ratio	Compounds	Reactio n Ion	m/z	Reactio n ratio	
Toluene	H <sub>3</sub> O <sup>+</sup>	93*	1	Naphthalene	H <sub>3</sub> O <sup>+</sup>	129*	1	Ethylene oxide	H <sub>3</sub> O <sup>+</sup>	45*	1	
	NO <sup>+</sup>	92	1		NO <sup>+</sup>	128*	1		O <sub>2</sub> <sup>+</sup>	44*	1	
	O <sub>2</sub> <sup>+</sup>	92*	1		O <sub>2</sub> <sup>+</sup>	128	1		Allyl chloride	41*	0.87	
Phenol	H <sub>3</sub> O <sup>+</sup>	95*	1	Butanone	H <sub>3</sub> O <sup>+</sup>	73	1	Allyl chloride	H <sub>3</sub> O <sup>+</sup>	59*	0.13	
	NO <sup>+</sup>	94	1		NO <sup>+</sup>	102*	1		O <sub>2</sub> <sup>+</sup>	41*	0.65	
	O <sub>2</sub> <sup>+</sup>	94	1		O <sub>2</sub> <sup>+</sup>	72*	0.3			76*	0.3	
Formic acid	H <sub>3</sub> O <sup>+</sup>	47	0.9	Tetrachloroethylene	O <sub>2</sub> <sup>+</sup>	166*	0.42	Tetrachloromethane	H <sub>3</sub> O <sup>+</sup>	119*	0.48	
	NO <sup>+</sup>	46	0.35			164*	0.32			117*	0.52	
	O <sub>2</sub> <sup>+</sup>	45*	0.4			168*	0.21		O <sub>2</sub> <sup>+</sup>	117	0.52	
Nitromethane	H <sub>3</sub> O <sup>+</sup>	62*	1	Methyl acrylate	H <sub>3</sub> O <sup>+</sup>	87*	1	Tetrachloromethane		119	0.48	
	NO <sup>+</sup>	61*	0.9		NO <sup>+</sup>	116*	0.9		Ethyl acetate	H <sub>3</sub> O <sup>+</sup>	89*	1
Acrolein	H <sub>3</sub> O <sup>+</sup>	57*	1		O <sub>2</sub> <sup>+</sup>	85*	0.4			O <sub>2</sub> <sup>+</sup>	88*	0.2
	NO <sup>+</sup>	55*	0.6	Bromoform	H <sub>3</sub> O <sup>+</sup>	173*	0.5	Ethyl acetate	Formaldehyde	H <sub>3</sub> O <sup>+</sup>	31*	1
		86*	0.4		O <sub>2</sub> <sup>+</sup>	175*	0.25			O <sub>2</sub> <sup>+</sup>	31*	1
Benzyl chloride	NO <sup>+</sup>	91*	1			173*	0.5	Formaldehyde	H <sub>3</sub> O <sup>+</sup>	54	1	
	O <sub>2</sub> <sup>+</sup>	91	0.6			171*	0.25		O <sub>2</sub> <sup>+</sup>	83*	1	
		126	0.3	1,1-dichloroethylene	H <sub>3</sub> O <sup>+</sup>	97*	0.57	Acrylonitrile	H <sub>3</sub> O <sup>+</sup>	49*	0.76	
Acrylamide	H <sub>3</sub> O <sup>+</sup>	72*	1			99*	0.37		O <sub>2</sub> <sup>+</sup>	51*	0.24	

	NO <sup>+</sup>	101*	1		O <sub>2</sub> <sup>+</sup>	96	0.57	Carbon disulfide	H <sub>3</sub> O <sup>+</sup>	77	1
	O <sub>2</sub> <sup>+</sup>	71	0.55			98	0.37		O <sub>2</sub> <sup>+</sup>	76*	1
1,4-Dioxane	H <sub>3</sub> O <sup>+</sup>	89*	1	Benzene	H <sub>3</sub> O <sup>+</sup>	79*	1	m-cresol	H <sub>3</sub> O <sup>+</sup>	109	1
	NO <sup>+</sup>	88*	0.45		NO <sup>+</sup>	108*	0.24		NO <sup>+</sup>	108*	1
	O <sub>2</sub> <sup>+</sup>	88	0.65		O <sub>2</sub> <sup>+</sup>	78*	1		O <sub>2</sub> <sup>+</sup>	108	1
Trichloroethylene	H <sub>3</sub> O <sup>+</sup>	131*	0.42	Methyl vinyl ketone	H <sub>3</sub> O <sup>+</sup>	71*	1	Methanol	H <sub>3</sub> O <sup>+</sup>	33*	1
		133	0.42		NO <sup>+</sup>	100*	1		NO <sup>+</sup>	62	1
	O <sub>2</sub> <sup>+</sup>	130	0.42		O <sub>2</sub> <sup>+</sup>	70*	0.5		O <sub>2</sub> <sup>+</sup>	31	0.5
		132	0.42		Vinyl chloride	O <sub>2</sub> <sup>+</sup>	62*	0.75		32	0.5
p-nitrotoluene	H <sub>3</sub> O <sup>+</sup>	138*	1			64*	0.25	Propylene oxide	H <sub>3</sub> O <sup>+</sup>	59*	1
	NO <sup>+</sup>	167	1	Dichloromethane	O <sub>2</sub> <sup>+</sup>	83*	0.56		NO <sup>+</sup>	57*	0.35
	O <sub>2</sub> <sup>+</sup>	137	1			85*	0.38			88	0.65

\* : mass-to-charge ratio (m/z) chosen for quantification and qualification of each compound.