

Supplementary Information

Elimination of micropollutants in activated sludge reactors with a special focus on the effect of biomass concentration

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Table S1. Overview of the removal efficiency of some MPs in published studies according to some operational conditions in activated sludge.

HRT / SRT	Feed concentration (µg L ⁻¹)						Removal efficiency (%)						Sludge concentration (g _{TSS} L ⁻¹)		
	CAF	SMX	BZT	ROX	ERY	DCF	CBZ	CAF	SMX	BZT	ROX	ERY	DCF	CBZ	
>2-3 / >150d (nitrifying conditions) ^a						≤1							70 ^b		
12-17h / 1.5–5.1d ^c								44-75						8-15	
16.7h / 6d ^d	1	1				1	1	99	90				20	<10	2.4
58.4h / 54d ^d	1	1				1	1	95	45				10	<10	2.5
15h / 10-12d ^e		0.23-0.57							9		25	2			3
31h / 21-25d ^e		0.23-0.57							60						3
- / 20-40d, >40d ^f							10-40							<20	2g _{vss} L ⁻¹
- / 20-40d, >40d ^f						10-40							<20		2g _{vss} L ⁻¹
- / >40d ^f		10-40							64-70		64-70	64-70			2g _{vss} L ⁻¹
12-20h / 25d ^g		1-5	1-5			1-5	1-5		20-90 ^h				<20 ^g	<20 ^g	
12-20h / 40d ^g		1-5	1-5			1-5	1-5		20-90 ^h				<20 ^g	<20 ^g	4-5.5
12-20h / 80d ^g		1-5	1-5			1-5	1-5		20-90 ^h				<20 ^g	<20 ^g	4-5.5
12h / 3d ⁱ							0.24							--	
12.5-13.6 / 52-114d ⁱ							0.32-1.85							--	

^aVieno and Sillanpää [1]; ^bFernandez-Fontaina et al. [2] ; ^cSantos et al. [3] ; ^dMajewsky et al. [4] ; ^eGobel et al. [5] ; ^fSuárez et al. [6] ; ^gFalàs et al. [7] ^hJoss et al. [8] ⁱHai et al. [9]. / --negligeable.

Table S2. Operating schedule of SBRs.

	Operating times in each cycle								Treated wastewater (L d ⁻¹)
	SRT (d)	HRT (h)*	Duration (d)**	Filling (min)	Reaction (min)	Settling (min)	Drawing (min)	# of cycles (d ⁻¹)	
SBR-3 d	3	4	30	5	170	60	5	6	30
		8	30	5	410	60	5	3	15
		12	30	5	650	60	5	2	10
SBR-10 d	10	4	30	5	170	60	5	6	30
		8	30	5	410	60	5	3	15
		12	30	5	650	60	5	2	10
SBR-20 d	20	4	30	5	170	60	5	6	30

*. Studied HRT is equivalent to the cycle duration; **: Actual kinetics periods. These periods include transition period between phases.

Table S3. Mean values of biomass concentration ($\pm$ standard error), volatile suspended solids (VSS), dissolved oxygen (DO), pH and T ($^{\circ}$ C) in the three SBRs during seven experimental setups.

SRT (d)	3	3	3	10	10	10	20
HRT (h)	4	8	12	4	8	12	4
Biomass concentration (g L ⁻¹)	3.3 $\pm$ 0.1	3.3 $\pm$ 0.1	3.2 $\pm$ 0.1	3.0 $\pm$ 0.1	3.3 $\pm$ 0.1	3.1 $\pm$ 0.1	3.5 $\pm$ 0.1
VSS (%)	70	72	71	79	77	79	81
DO (mg L ⁻¹)	2.5	2.4	2.2	4.4	4.3	4.2	4.4
pH	8.1	8.1	8	7.5	7.8	7.9	8
T ($^{\circ}$ C)	20.1	19	19	18.1	18	18	19.5

Table S4. Composition of the synthetic wastewater and supplementary solutions I and II.

Synthetic wastewater (SWW)	Concentration (mg L⁻¹)		
	SWW1 (SBR-3 d)	SWW2 (SBR-10 d)	SWW3 (SBR-20 d)
CH ₃ COONa	1200	500	100
NH ₄ HCO ₃	75	75	75
KH ₂ PO ₄	5	5	5
MgSO ₄ .7H ₂ O	50	50	50
CaCl ₂ .2H ₂ O	21	21	21
NaHCO ₃	36	36	36
Supplementary solution I			
C ₁₀ H ₆ N ₂ O ₈	50	50	5
FeSO ₄ .7H ₂ O	20	20	2
Supplementary solution II			
C ₁₀ H ₆ N ₂ O ₈	20	20	20
ZnSO ₄ .7H ₂ O	0.43	0.43	0.43
CoCl ₂ .6H ₂ O	0.24	0.24	0.24
MnCl ₂ .4H ₂ O	1	1	1
CuSO ₄ .5H ₂ O	0.25	0.25	0.25
NaMoO ₄ .2H ₂ O	0.22	0.22	0.22
NiCl ₂ .6H ₂ O	0.19	0.19	0.19
NaSeO ₄ .10H ₂ O	0.21	0.21	0.21
H ₃ BO ₄	0.14	0.14	0.14

Text S1. Calculation of Solid Retention Time (SRT).The target SRTs were maintained manually by adjustment of biomass concentration in SBRs at the end of the aeration period according to reference [10] using the Eq. (S1):

$$SRT = \frac{C_{TSS} V}{C_{TSS\ eff} Q_{eff} + C_{TSS\ w} Q_w} \quad (1)$$

where C_{TSS} : Total Suspended Solids concentration in the reactor (g TSS L⁻¹); V: reactor volume (L); $C_{TSS\ eff}$: Total Suspended Solids concentration in the effluent (g TSS L⁻¹); Q_{eff} : effluent flow rate (L d⁻¹); $C_{TSS\ w}$: Total Suspended Solids concentration of the withdrawn sludge (g TSS L⁻¹); Q_w : sludge withdrawal (L d⁻¹).

Table S5. Source parameters applied during sample measurements.

Source parameters	
Gas Temperature	200 °C
Gas Flow	8 L min ⁻¹
Nebulizer	40 psi
Sheath Gas Temperature	300 °C
Sheath Gas Flow	12 L min ⁻¹
Capillary	Positive: 4500 V Negative 3500 V
Nozzle Voltage	Positive 500 V Negative 300 V

Table S6. Mass spectrometric parameters for detection.

Compound	RT min	Polarity	Precursor Ion	Quantifier	CE	Qualifier	CE
Benzotriazole	3.3	Positive	120.1	65.0	24	92.1	20
Benzotriazole-d ₄	3.3	Positive	124.1	41.4	48	96.0	20
Caffeine	3.0	Positive	195.1	137.9	20	110.0	24
Caffeine- ¹³ C ₃	3.0	Positive	198.2	140.0	20	43.5	44
Carbamazepine	5.8	Positive	237.1	193.4	28	178.9	40
Carbamazepine-d ₈	5.8	Positive	245.2	202.1	24	200.6	40
Diclofenac	8.0	Positive	296.0	213.9	40	250.0	12
Diclofenac-d ₄	8.0	Positive	300.1	218.0	36	-	-
		Negative	298.0	-	-	254.0	8
Erythromycin	3.6	Positive	734.5	158.0	32	576.0	20
Erythromycin-d ₃	3.6	Positive	737.5	161.0	32	579.4	20
Roxithromycin	3.9	Positive	837.5	158.0	36	679.4	20
Roxithromycin-d ₇	3.9	Positive	844.6	158.0	36	686.5	20
Sulfamethoxazole	4.4	Positive	254.1	155.9	12	92.0	28
Sulfamethoxazole-d ₄	4.4	Positive	258.1	159.9	36	151.1	12

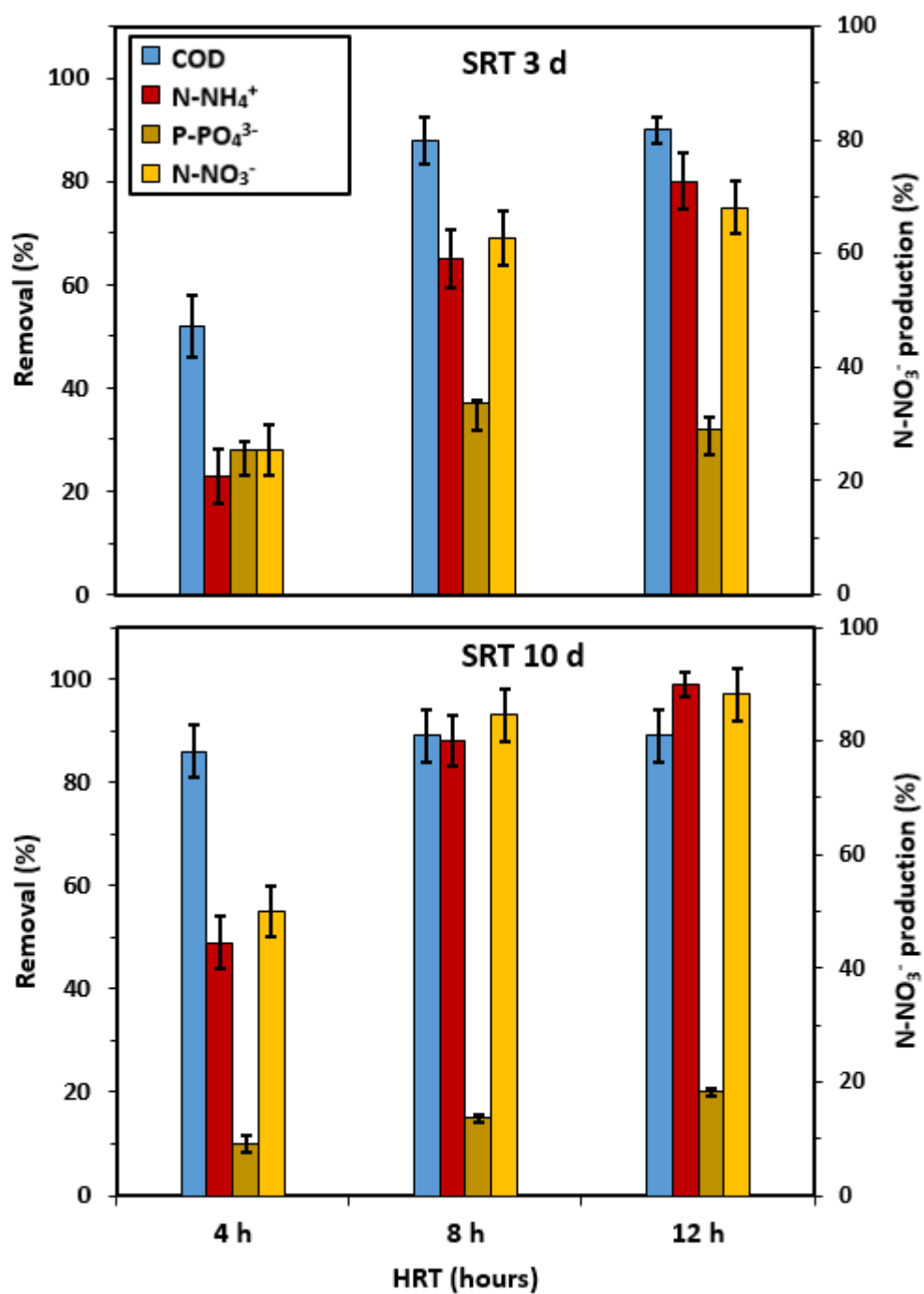


Figure S1. Removal efficiencies of wastewater parameters (COD, N-NH₄⁺ and P-PO₄³⁻) and nitrate-nitrogen production over 4 h, 8 h and 12 h HRT for SRT - 3 d and SRT - 10 d (error bars present standard error).

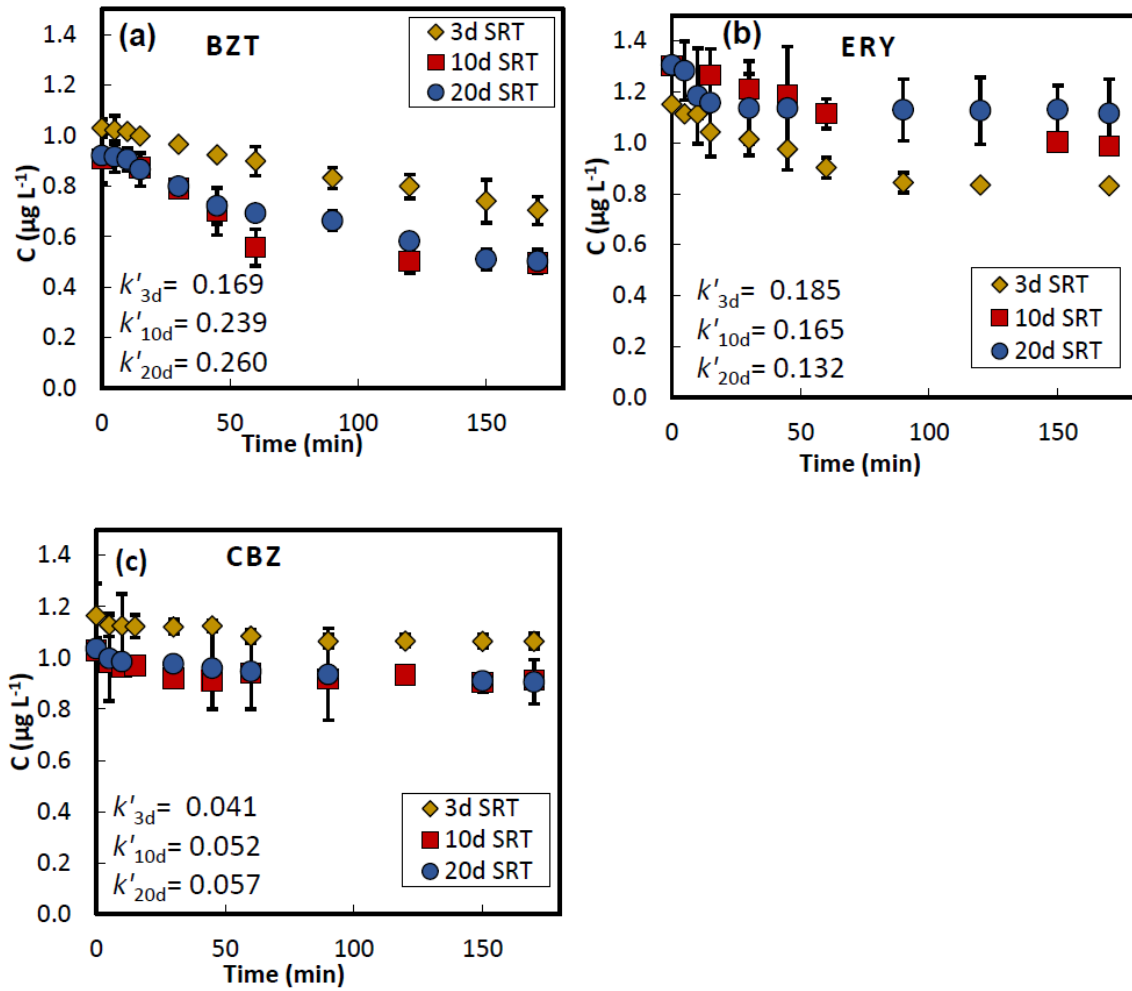


Figure S2. Removal and k' : apparent-first-order fits (h^{-1}) of other investigated MPs; number of replicates = 3; error bars indicate one standard error.

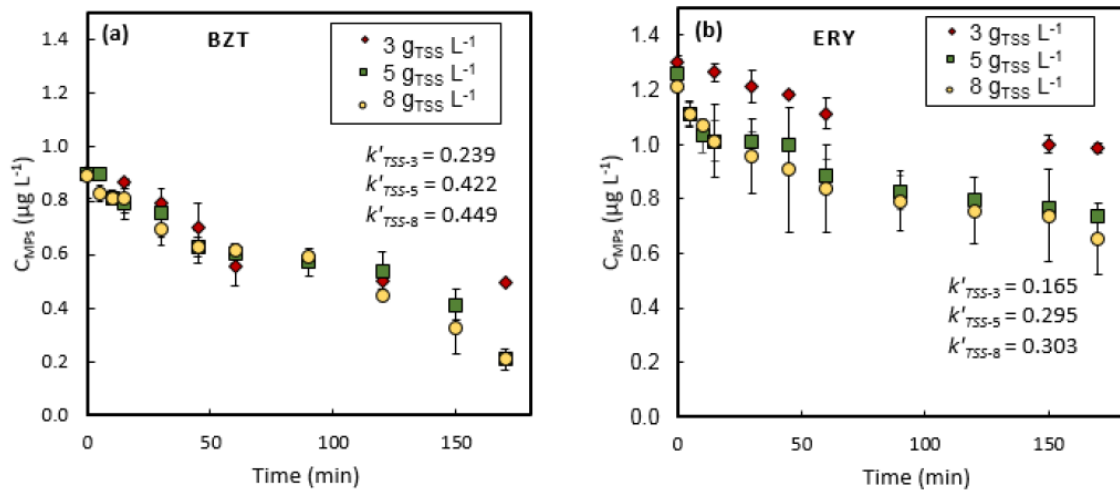


Figure S3. Change of concentration of (a) BZT and (b) ERY during biological treatment in reactors inoculated with biomass at concentrations of 3, 5 and 8 g L^{-1} ; k' : apparent-first-order removal rate constant (h^{-1}) at SRT - 10 d and HRT - 4 h.

References

1. Vieno, N.; Sillanpää, M. 2014. Fate of diclofenac in municipal wastewater treatment plant — A review. *Environ.Int*, 69, 28–39.
2. Fernandez-Fontaina, E.; Omil, F.; Lema, J.M.; Carballa, M. 2012. Influence of nitrifying conditions on the biodegradation and sorption of emerging micropollutants. *Water Research*, 46, 5434–5444.
3. Santos, J.L., Aparicio, I., Callejón, M., Alonso, E. 2009. Occurrence of pharmaceutically active compounds during 1-year period in wastewaters from four wastewater treatment plants in Seville (Spain). *Journal Hazardous Materials*, 164, 1509-16.
4. Majewsky, M.; Gallé, T.; Yargeau, V.; Fischer, K. 2011. Active heterotrophic biomass and sludge retention time (SRT) as determining factors for biodegradation kinetics of pharmaceuticals in activated sludge. *Bioresource Technology*, 102, 7415–7421.
5. Gobel, A.; Mcardell, C.; Joss, A.; Siegrist, H.; Giger, W. 2007. Fate of sulfonamides, macrolides, and trimethoprim in different wastewater treatment technologies. *Science of The Total Environment*, 372, 361–371.
6. Suárez, S.; Reif, R.; Lema, J.M.; Omil, F. 2012. Mass balance of pharmaceutical and personal care products in a pilot-scale single-sludge system: Influence of T, SRT and recirculation ratio, *Chemosphere*, 89, 164-171.
7. Falås. P.; Wick, A.; Castronovo, S.; Habermacher, J.; Ternes, T.A.; Joss, A. 2016. Tracing the limits of organic micropollutant removal in biological wastewater treatment. *Water Resource*, 95, 240-249.
8. Joss, A.; Zabczynski, S.; Göbel, A.; Hoffmann, B.; Löffler, D.; McArdell, C.S.; Ternes, T.; Thomsen, A.; Siegrist, H. 2006. Biological degradation of

pharmaceuticals in municipal wastewater treatment: proposing a classification scheme. *Water Resource*, 40, 1686-1696.

9. Hai, F.; Shufan, Y.; Muhammad, A.; Vitor, S.; Samia, S.; Martina, S.S.; Jody, G.; Zhi-Qiang, X.; Kazuo, Y. 2018. Carbamazepine as a possible anthropogenic marker in water: occurrences, toxicological effects, regulations and removal by wastewater treatment technologies. *Water* 10, 107.
10. Henze, M.; Van Loosdrecht, M.C.M.; Ekama, G.; Brdjanovic, D. 2008. *Biological wastewater treatment: principles, modelling and design*. London, IWA Publishing.