

Article

A Constructed Wetland System for Rural Household Sewage Treatment in Subtropical Regions

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Abstract: A constructed wetland system, consisting of a surface-flow wetland cell connected in series with three vertical subsurface-flow wetland cells, was tested for treatment of domestic sewage from rural families in southern China. Diatomite, vermiculate, zeolite and hydrotalcite, were used, respectively, as filler adsorbents in the sequenced subsurface-flow cells for adsorption of organic, cationic and anionic pollutants. Selected trees, shrubs and annual herbs were planted to form a wetland plant community. The total treatment capacity, hydraulic loading rate and water retention time were 2 m³/d, 0.5 m/d and 48 h, respectively. Experimental data obtained from a year operation confirmed that the treatment process followed the dynamic pathway of pollutant transformation. The constructed system was effective to remove TSS, COD_{Cr} and BOD₅ and their effluent concentrations met the first grade of the discharge standards legislated in China. The removal rates of TN, NH₃-N and TP were relatively lower, and their effluent concentrations fell within the range between the first and second grade of the standards. An increase in initial pollutant loading and a decrease in temperature in winter caused apparent accumulation of TN, NH₃-N and TP in the system. Discharge of sludge at adequate intervals was shown to be effective to enhance the treatment efficiency.

Keywords: constructed wetland; multiple-cell system; pollutant transformation pathway; sewage treatment; sludge discharge; subtropical region

1. Introduction

Accounted for by its low construction and operation costs and ecologically responsible nature, constructed wetland technology (CW technology) has been rapidly developed as one of the alternative solutions for treatment of various types of wastewaters in different countries [1–7]. Examples [8,9] have shown that CW technology is very useful in rural and particularly mountainous areas where establishment of a centralized conventional treatment plant connected to long distance wastewater collection and transportation pipeline systems becomes an impossible option. With efforts to meet the demands for use of environmentally friendly and energy-saving technology to protect the rural environment, the present study was conducted to test a constructed wetland system for treatment of domestic sewage from rural families in southern China.

In order to obtain a better understanding of pollutant removal mechanisms, studies have been carried out by different workers on factors that determine the treatment efficiency of constructed wetland systems, including wetland types and combinations [5,10]; hydraulic and pollutant loads [2,11]; plant species [12–14]; substrates [15,16]; and their interactions [17,18], etc. The increasing trend in CW technology application coupled with stricter water quality standards requires development of better process design tools. Aimed at providing useful information for future design, efforts have been made in the following areas: dynamic simulation models [19–22]; treatment stages [23] and pollutant

transformation cycles [24]. A constructed wetland is a designed ecosystem. The integrated function of a wetland system is complex, involving physical processes (sedimentation, flocculation, adsorption and volatilization); chemical reactions (hydrolysis, oxidation and reduction); biochemical reactions (degradation and transformation); biological activities (uptake and assimilation) and ecological processes (matter and energy exchange among bio-species and between the system and surroundings). The rates of the reactions that occur in a wetland system are related to environmental variables including temperature, dissolved oxygen, oxidation-reduction potential, and many other physical, chemical, biological and ecological characteristics. Because of the complexity of ecological processes and the unpredictability of the changes in environmental conditions that vary in localities, the design of constructed wetland systems has so far been based mainly on experiences without commonly accepted standard procedures. Most applied systems are more or less treated as black boxes with insufficient knowledge and data concerning the exact manners, mechanisms, functions and contributions of the interrelated processes.

One of the main objectives of the present study was focused on the development of design principles using the knowledge related to the transformation of the target pollutants and pathways of pollutant accumulation and removal. The functions of the designed treatment cells corresponding to specific stages of pollutant transformation and their treatment efficiencies were examined using sewage from a teaching building at the university. The ultimate goal of this study was to provide a technical model with basic data and parameters for design and application of constructed wetland systems for rural household sewage treatment.

2. Materials and Methods

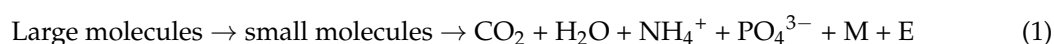
2.1. Basic Parameters Used for Treatment Process Design

Although there are numbers of factors affecting the treatment process, the hydraulic loading rate (HLR, the volumetric flow rate divided by the wetland surface area) and water retention time (HRT, the ratio of the useable wetland water volume to the average flow rate) are the two key parameters that need to be determined for the design of a constructed wetland treatment system. Given wastewater type, HLR (m/d) determines not only the hydraulic and pollutant loads but also the required land area, or the land use efficiency, which is one of the highly concerned factors for application of CW technology. When HLR is given, HRT can be adjusted by the active depth of the treatment cells. In the investigated rural regions in Hunan Province, China, the amount of sewage discharge from a farmer's house ranged between 0.5 and 1.5 t/d (approximately 180 L/d per person). Due to variation in wastewater types and localities, the HLR and HRT values applied by different workers [2,25,26] varied greatly from 0.06 to 0.49 m/d and a few hours to several days, respectively. Based on the data obtained from previous studies, the total treatment capacity, hydraulic loading rate, BOD₅ loading rate, and water retention time designed for the constructed wetland system were 2 m³/d, 0.5 m/d, 60.9 g BOD₅/m²/d and 48 h, respectively. The effluent quality was designed to meet the first grade B for the discharge standard of pollutants for a municipal wastewater treatment plant (GB 18918-2002) in China.

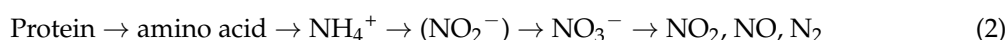
2.2. Sequence of Treatment Cells Related to the Transformation Pathway

Decomposition of organic pollutants normally takes several steps under different conditions. It is thus difficult to obtain a high pollutant removal rate from a single wetland unit without input of additional energy. In general, multiple cells connected in series are recommended to minimize short circuiting and maximize treatment efficiency [23,27]. Therefore, given HLR and HRT, it is necessary to determine the number and connection sequence of the treatment cells required to ensure treatment efficiency.

It is known that the overall decomposition reaction of organic pollutants follows the pathway,



where E denotes the energy released from the reaction and M denotes inorganic substances other than NH_4^+ and PO_4^{3-} . The sequence of the reactions related to nitrogen-containing organic compounds, for example, can be described by



There are basically 4 steps in Reaction (2) for the transformation of various types of proteins into nitrogen gases: (1) hydrolysis of protein (splitting of large protein molecules into amino acids); (2) ammonification (release of NH_3 from amino acid by oxidation, reduction or hydrolysis); (3) nitrification (oxidation of NH_4^+ to NO_3^- under aerobic conditions) and (4) denitrification (reduction of NO_3^- to NO_2 , NO or N_2 under anaerobic conditions). Apart from being either removed by plant uptake or retained by microorganisms and plant roots in the system, NH_4^+ , NO_3^- and PO_4^{3-} can be adsorbed onto filler substrate (S),



As a matter of fact, a large amount of soluble organic compounds can also be kept in precipitated or adsorbed forms. Use of substrate with high adsorption capacities can thus increase the system buffering capacity to resist pollutant loads.

The above Reactions (1)–(5) provide a theoretical background for the design of a wetland treatment process following the pollutant transformation pathway. As illustrated in Figure 1, the presented rural household system consists of four basic units with a surface-flow wetland cell A, used as the first treatment unit that allows sedimentation and precipitation of large pollutant particles, connected in series with three vertical subsurface-flow wetland cells, BI, BII and BIII. Following the reaction pathway, the major target pollutants to be removed by use of different plant species and adsorbents in units A, BI, BII and BIII, are organic (e.g., amino acids), cationic (NH_4^+) and anionic (NO_3^- , PO_4^{3-}) pollutants, respectively. The drainage outlet of each treatment cell is connected to a sludge pool (Figure 1), which allows discharge of sludge from the drainage outlets of the system when needed. In addition, the drainage system is designed to prevent clogging that will be inevitably encountered in constructed wetland systems after a long period of operation.

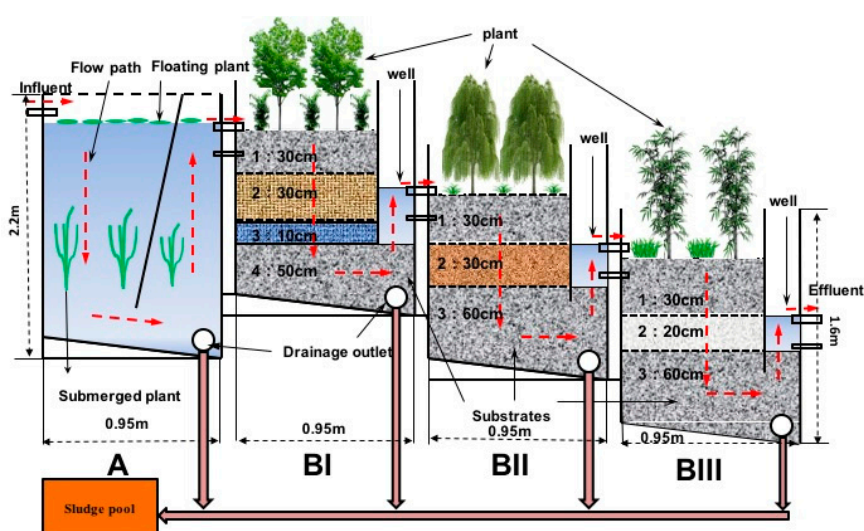


Figure 1. Constructed wetland system designed for treatment of rural household sewage (Layer numbers in BI, BII and BIII correspond to the numbers of substrate layers in Table 1).

The types, functions, introduced plant species and substrate fillers of the treatment cells are summarized in Table 1. The system used in the present test was made of steel which is firm and waterproof. For application in rural areas, other cheaper materials can be used to reduce the construction costs.

Table 1. Type, function, plant species and substrate of wetland cells.

Cell	Type	Major Target Pollutant	Main Reaction	Plant Species	Substrate
A	Surface flow	Organic pollutant Suspended solid	Sedimentation Precipitation Decomposition Uptake	<i>Hydrocotyle vulgaris</i> <i>Ipomoea aquatica</i> Forsk. <i>Oenanthe Ljavanica</i> (Blume) DC. <i>Hydrilla verticillata</i> <i>Myriophyllum verticillatum</i>	none
BI	Subsurface flow	Organic pollutant Suspended solid	Decomposition Filtration Uptake Adsorption	<i>Hibiscus mutabilis</i> Linn. <i>Metasequoia glyptostroboides</i> <i>Rosa chinensis</i> Jacq. <i>Pontederia cordata</i> <i>Brassica campestris</i> L. <i>Iris germanica</i> L.	1: Limestone (Φ1 cm–Φ2 cm) 2: Diatomite 3: Vermiculate 4: Limestone (Φ3 cm–Φ5 cm)
BII	Subsurface flow	Organic pollutant Cationic pollutant	Decomposition Uptake Adsorption	<i>Salix babylonica</i> <i>Loropetalum chinense var. rubrum</i> <i>Nerium indicum</i> Mill. <i>Acorus calamus</i> L. <i>Typha orientalis</i> Presl. <i>Oenanthe javanica</i> (Blume) DC.	1: Limestone (Φ1 cm–Φ2 cm) 2: Zeolite 3: Limestone (Φ3 cm–Φ5 cm)
BIII	Subsurface flow	Organic pollutant Anionic pollutant	Decomposition Uptake Adsorption	<i>Fargesia fungosa</i> Yi <i>Arundo donax var. versicolor</i> <i>Thalia dealbata</i> Link	1: Limestone (Φ1 cm–Φ2 cm) 2: Hydrotalcite 3: Limestone (Φ3 cm–Φ5 cm)

2.3. Plant Species and Substrates

In order to strengthen the integrated function of a wetland system, it is important to use suitable plant species and a highly cost-effective substrate. Plant richness can improve the structure, function and sustainability of wetland ecosystems to resist seasonal changes [13,28]. Use of adsorbents can enhance the buffering capacity of the treatment cells to resist pollutant loads. Based on previous studies, the plant species listed in Table 1 were used to establish a wetland plant community consisting of trees (*Hibiscus mutabilis* Linn., *Salix babylonica*), shrubs (*Metasequoia glyptostroboides*, *Rosa chinensis* Jacq., *Loropetalum chinense var. rubrum*, *Nerium indicum* Mill.) and annual herbs (*Hydrocotyle vulgaris*, *Myriophyllum verticillatum*, *Hydrilla verticillata* (L. f.) Royle, *Ipomoea aquatica* Forsk., *Eichhornia crassipes*, *Alternanthera hioxeroides* (Mart.) Griseb, *Pontederia cordata*, *Brassica campestris* L., *Acorus calamus* L., *Oenanthe javanica* (Blume) DC., *Fargesia fungosa* Yi, *Arundo donax var. versicolor*, *Thalia dealbata* Link). The selected filler adsorbents, diatomite (as molecular adsorbent), vermiculate and zeolite (both as cationic adsorbents) and hydrotalcite (as anionic adsorbent), were used in different cells for the adsorption of organic, cationic and anionic pollutants, respectively. The adsorption characteristics of the selected adsorbents were examined in earlier tests.

2.4. Operation and Analysis

The wastewater used for test of the designed system was the sewage from a septic tank installed at one of the main teaching buildings of the university. Analytical results indicated that the concentrations of the main pollutants in the tested wastewater were similar to those in the investigated rural areas in Hunan Province. The variation range of the water quality indexes of the influent, the designed

standards of the effluent and the maximum removal rate (R_M) required to reduce the highest pollutant concentration to designed standards are listed in Table 2.

Table 2. Pollutant concentration in the influent and designed standard of the effluent.

Index	TSS	COD _{Cr}	BOD ₅	TN	NH ₃ -N	TP
Influent (mg/L)	55~80	79.5~322.6	46.7~121.9	20.3~57.9	17.1~55.5	1.79~6.83
Standard * (mg/L)	20	60	20	20	8	1
R_M (%)	75	82	84	66	86	86

Note: *: First grade B, Discharge Standard of Pollutants for Municipal Wastewater Treatment [GB 18918-2002], stipulated by the Ministry of Environment Protection, China. R_M : Expected total removal rate to reduce the highest pollutant concentration to designed standards.

The operation of the system was continued through a whole year. During the operation period, water samples of 500 mL were collected in triplicate from both the inlet and outlet of each treatment cell at mostly week intervals and were prepared immediately for analysis. The standard methods stipulated in “Methods for Monitoring and Analysis of Water and Wastewater, 2000” by the Ministry of Environmental Protection, China, were used for analyses of total suspended solid (TSS), chemical oxygen demand (COD_{Cr}), biochemical oxygen demand (BOD₅), total nitrogen (TN), ammonia nitrogen (NH₃-N), and total phosphorus (TP). The relative deviations of the sample analyses were between 0.3% and 6.1%, which were within the allowed accuracy range. The average values of the measurements were presented for discussion.

3. Results

3.1. Treatment Efficiency of the System

Table 3 gives the average effluent pollutant concentration and accumulative removal rate (R_a) of the treatment cells. The accumulative removal rate is given as

$$R_a = 100 \times (C_0 - C_{\text{eff}})/C_0 \quad (6)$$

where C_0 denotes the initial pollutant concentration in the untreated influent (the influent of cell A) and C_{eff} is the pollutant concentration in the effluent of cell i. The R_a value of BIII represents the total removal rate R_T of the system.

Despite the great variation in pollutant loadings in the operation period, the designed system was very effective in the removal of TSS, COD_{Cr} and BOD₅. The observed concentrations of TSS, COD_{Cr} and BOD₅ in the effluent of the final cell (BIII) met the first grade B of the legislated discharge standards (GB 18918-2002) throughout the tested year. In comparison, however, the removal of TN, NH₃-N and TP was lower, and approximately one third of their observed values fell within the range between the first and second grade standards. The average total removal rate (the R_a value of BIII) was higher than 80% for TSS, COD_{Cr} and BOD₅ but lower than 62% for TN, NH₃-N and TP (Table 3).

Table 3. Annual average effluent concentration C_{ef} (mg/L) and accumulative removal rate R_a (%).

Index	TSS	□	COD _{Cr}	□	BOD ₅	□	TN	□	NH ₃ -N	□	TP	□
Cell	C_{ef}	R_a	C_{ef}	R_a	C_{ef}	R_a	C_{ef}	R_a	C_{ef}	R_a	C_{ef}	R_a
O	68.0 ± 4.0	0.0	172.5 ± 9.8	0.0	88.8 ± 5.4	0.0	38.3 ± 1.9	0.0	36.4 ± 2.0	0.0	3.92 ± 0.24	0.0
A	58.0 ± 2.0	14.7 ± 0.5	123.9 ± 5.7	28.2 ± 1.2	73.8 ± 4.1	16.9 ± 1.0	35.3 ± 2.1	7.9 ± 0.4	32.6 ± 1.9	10.4 ± 0.6	3.32 ± 0.20	15.2 ± 0.9
BI	20.0 ± 1.0	70.6 ± 2.7	85.6 ± 3.9	50.4 ± 2.6	40.5 ± 2.3	54.4 ± 3.1	31.2 ± 1.9	18.5 ± 1.1	27.7 ± 1.6	23.9 ± 1.4	2.84 ± 0.17	27.5 ± 1.6
BII	14.0 ± 0.5	79.4 ± 3.0	56.0 ± 2.4	67.5 ± 3.2	30.1 ± 1.5	66.1 ± 3.5	25.1 ± 1.5	34.6 ± 1.9	22.0 ± 1.3	39.6 ± 2.1	2.23 ± 0.13	43.1 ± 2.4
BIII	10.0 ± 0.3	85.3 ± 3.4	30.4 ± 1.6	82.4 ± 4.0	17.7 ± 0.9	80.1 ± 4.2	20.1 ± 1.2	47.5 ± 2.8	17.8 ± 1.0	51.0 ± 2.7	1.50 ± 0.08	61.8 ± 3.6
GB *	1st	□	1st	□	1st	□	2nd	□	2nd	□	2nd	□

Note: *: Discharge standard (GB 18918-2002); 1st: First grade B; 2nd: Second grade.

3.2. Effect of Pollutant Loading, Temperature and Sludge Discharge

Given the hydraulic loading rate HLR, the pollutant concentration of the initial influent (influent of cell A) determines the pollutant loading rate of the treatment system. Table 4 gives the coefficient values obtained from the correlation analyses on pollutant concentrations between effluent and initial influent. All correlation coefficient values are positive, showing the impact of the pollutant loading on treatment efficiency. A clear trend can be noticed in Table 4: the coefficient values for all analyzed pollutants decrease along the treatment process from A to BIII. Since in the designed system, the effluent of a cell is the influent of its successor, it is expected that the initial pollutant loading rate should have less influence on subsequent cells. This appears to be the case for TSS and TP. The correlation coefficient values for TSS and TP are very low at BII and BIII, suggesting that the initial pollutant loading rate is not the unique factor determining the effluent concentration of TSS and TP. The low coefficient values for TSS observed at cells BII and BIII can be explained by the high removal capacity of BI. As noticed in Table 3, the TSS level in the effluent of BI was low, which already met the designed standard and thus reduced the dependence of BII and BIII on initial pollutant loading. The low coefficient values for TP at BII and BIII, however, were mainly attributed to the changes in temperature. The low temperature in the winter period caused an apparent reduction of TP removal in all subsurface-flow wetland cells.

Table 4. Correlation coefficient between effluent and initial influent concentrations.

Cell	TSS	COD _{Cr}	TN	NH ₃ -N	TP
A	0.88	0.90	0.96	0.97	0.88
BI	0.72	0.79	0.94	0.91	0.58
BII	0.38	0.70	0.88	0.85	0.46
BIII	0.21	0.69	0.77	0.79	0.38

The effect of an increase in initial influent concentration and a decrease in temperature in winter on the effluent concentration of BIII is illustrated in Figure 2. In order to give a clear picture to show the change in system treatment efficiency with operation time, the average weekly measurement values for cells A, BI and BII were not presented. In general, the trends of the changes in week observations for cells A, BI and BII are similar to that for BIII shown in Figure 2. The operation time is separated into warm and cold periods by two dotted lines (Figure 2). The interval between the two dotted lines denotes the cold period phase (late November to early March) with a temperature range between approximately -2°C and 15°C . The remaining sections are the warm periods with a temperature range between 16°C and 39°C . The two dotted arrow lines in Figure 2 indicate the day on which sludge drainage occurred.

The high difference in temperature between the cold and warm periods did not cause significant changes in the effluent concentration of COD_{Cr}. Despite the sharp increase in initial pollutant loadings in the cold period, the concentration of COD_{Cr} (as well as BOD₅ and TSS) in the effluent of BIII remained below the GB1 (the stricter first B discharge standard) line throughout the operation period (Figure 2a). In comparison, the negative impact of the pollutant loadings on the removal of TN (Figure 2b), NH₃-N (Figure 2c) and TP (Figure 2d) was apparently exaggerated by the low temperature in the cold period, which raised the effluent concentrations of these three pollutants to significantly higher levels than their discharge standards. There were clear trends of increasing pollutant concentrations in the effluent of BIII under increasing operation time before the day when the first sludge drainage was performed (Figure 2). The increasing trends of the effluent concentrations were indications of pollutant accumulation in the solid phase of the system. As shown in Figure 2, the discharge of the sludge using the drainage pipelines (Figure 1) to empty the stored wastewater from the whole system for 24 h was carried out twice (week 34, the end of the cold period, and week 45, the end of the spring) in the later operation period. The effect of sludge discharge with the purpose of removing the accumulated pollutants was effective as it resulted in a significant reduction of the effluent concentration for all

analyzed pollutants in all treatment cells. The observed differences between the month before and after the sludge drainage showed that the sludge discharge enhanced the average monthly removal rates by 16.8%, 20.1% and 34.6% at week 34, and 8.2%, 11.5% and 16.2% at week 45 for TN, NH₃-N and TP, respectively. The sludge discharge measure was shown to be effective to reduce the effluent concentrations of NH₃-N and particularly TP to meet the designed discharge standards.

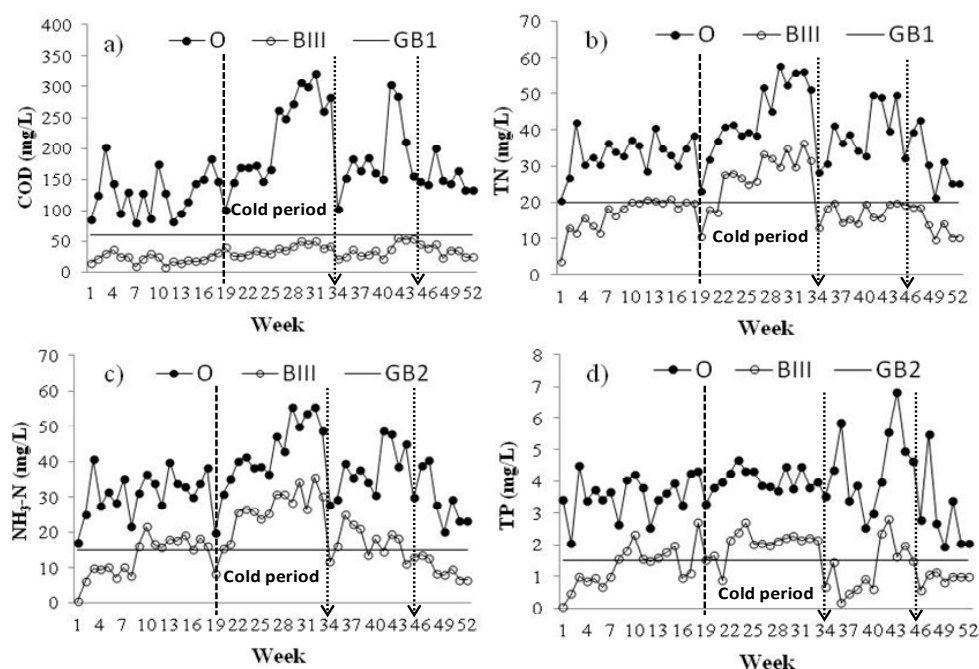


Figure 2. Change in pollutant concentration with operation time (O: influent; BIII: effluent; GB1: stricter first B discharge standard; GB2: old first B discharge standard; Interval between two dotted lines: cold period; Dotted arrow lines: point of sludge drainage. The relative deviations of the average values were within the range between 0.3% and 6.1%).

3.3. Function and Contribution of Treatment Cells

For design of a multiple-cell system, it is necessary to determine the function and contribution of each treatment cell. The function of an individual wetland cell can be evaluated by its relative removal rate R_r given as

$$R_r = 100 \times (C_{in} - C_{ef}) / C_{in} \quad (7)$$

where C_{in} and C_{ef} denote, respectively, the influent and effluent concentrations of the treatment cell. The applied HRT was 48 h, which was long enough for pollutant transfer to reach equilibrium among liquid, gas and solid phases in the system. The R_r values given by Equation (7) and listed in Table 5 can thus be regarded as equilibrium ratios.

The R_r values of COD_{Cr} were significantly higher than those of TN, NH₃-N and TP in all treatment cells. The highest R_r values obtained for NH₃-N (as well as TN) at BII and TP at BIII confirmed the functions of BII and BIII designed, respectively, to remove cationic and anionic pollutants.

Table 5. Average relative pollutant removal rate R_r (%) of treatment cells.

Unit	TSS	COD _{Cr}	TN	NH ₄ ⁺ -N	TP
A	14.7 ± 0.8	28.2 ± 1.6	7.9 ± 0.4	10.4 ± 0.5	15.2 ± 0.6
BI	65.5 ± 3.7	30.9 ± 1.5	11.5 ± 0.5	15.1 ± 0.8	14.5 ± 0.5
BII	30.0 ± 1.8	34.5 ± 1.6	19.8 ± 1.0	20.6 ± 0.9	21.5 ± 0.9
BIII	28.6 ± 1.6	45.7 ± 2.2	19.7 ± 1.1	18.9 ± 0.7	32.8 ± 1.3

The relative removal rate cannot fully reflect the contribution of the cell to the system treatment efficiency as it is a function of the cell's influent concentration which varies in cells connected in series. Replacing the denominator C_{in} (the influent concentration of cell i) in Equation (7) by C_0 (the initial influent concentration of cell A) gives the rate of contribution of cell i to the system treatment efficiency

$$R_c = 100 \times (C_{in} - C_{ef})/C_0 \quad (8)$$

The difference in average R_c values for pollutants in different cells is shown in Figure 3.

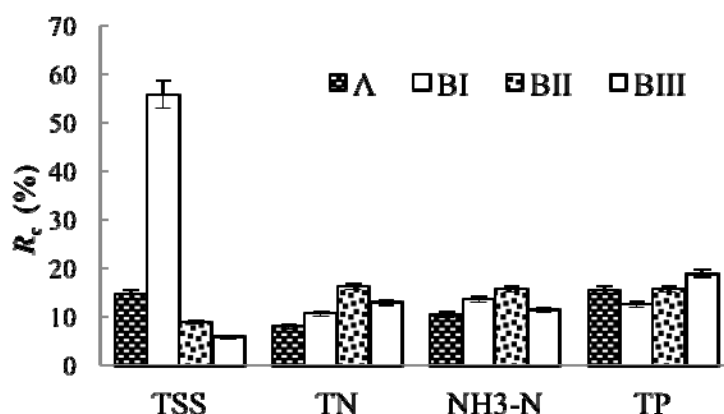


Figure 3. Contribution rate (R_c) of treatment cells.

The highest R_c value of TSS obtained at BI indicated that the subsurface-flow wetland cell was an excellent sedimentation and filtration system for removal of suspended solids. Unlike the R_r value of COD_{Cr} (Table 5), which increased from A to BIII due to a decrease in denominator C_{in} , the R_c value of COD_{Cr} decreased along the treatment pathway (Figure 3). The highest contribution rate for COD_{Cr} was obtained in cell A because, as the first unit, it received the highest pollutant loads. As expected, BII (designed for removal of cationic pollutants) had the highest contribution for the removal of TN and NH_3 -N while BIII (designed for removal of anionic pollutants) had the highest contribution for the removal of TP. The difference in contribution rates between cells was significant ($p < 0.001$) for all analyzed pollutants.

The released ionic species (NH_4^+ , NO_3^- and PO_4^{3-} , etc.) from decomposition of organic pollutants can be removed by plant uptake. Therefore, as an essential component, wetland plants play an important role in both the constitution of a wetland system and enhancement of its treatment efficiency [12,14]. However, the amounts of nitrogen and phosphorus removed directly by plant uptake are relatively low, which, according to Dzakpasu et al. [29], contributes less than 16% of the total removal rates. Thus, in the treatment process, large amounts of nitrogen and phosphorus will remain in the substrate bed in either organic-bound or mineralized forms. As a critical parameter for design of a treatment cell, the relative removal rate is only partially related to the buffering capacity of the cell. Use of fillers with higher ion adsorption capacity for construction of a wetland cell bed, for an example, can increase the buffering capacity of the cell to resist ionic pollutant loads for a longer operation period [16]. However, regardless of how high the adsorption capacity of the cell is, the concentration of the adsorbed pollutants in the solid phase will increase and finally reach saturation if the adsorption capacity of the cell bed cannot be simultaneously renewed by other means. This demands an effective pathway for regeneration of the system adsorption capacity.

4. Discussion

The observed changes in effluent concentrations (Table 3, Figure 2) and correlation coefficients (Table 4) and variations in relative removal rates (Table 5) and contribution rates (Figure 3) confirm

that the designed treatment process follows the pathway described by Reactions (1)–(5). The observed high and stable COD_{Cr} and BOD_5 removal rates are expected results from the designed multiple-cell process because a reduction in COD_{Cr} can be attained in all cells under both aerobic and anaerobic conditions. The highest contribution rate of COD_{Cr} obtained in cell A (Figure 3) shows the importance of the pretreatment unit. In the multiple-cell system, cell A functions not only as a sedimentation pool to remove large solid particles but also as a facultative reactor for decomposition of soluble organic compounds and release of volatile organics. Removal of nitrogen and phosphorus present in inorganic forms occurs at later stages of the treatment process, and this is one of the major reasons for their lower removal rates. Apart from plant uptake, removal of $\text{NH}_3\text{-N}$ as NH_3 and N_2 gases from the system demands optimal pH, DO and temperature conditions in the sequenced ammonification, nitrification and denitrification processes [23]. Development of cost-effective measures for control of microbial activities to enhance nitrogen removal in wetland systems may still remain important topics in future studies [17]. In a natural process, the unique route to remove phosphorus completely from the wetland system is plant uptake. This results in difficulties in phosphorus removal particularly in winter when plant growth is lower. Reduction of pollutant loadings by increasing land use area or improvement of the wetland structure and function by using high adsorption capacity fillers, etc., can be effective solutions to obtain better performance. Applications of such measures, however, will inevitably lead to increases in construction costs.

In general, maintaining a high and stable treatment efficiency of the designed wetland system demands simultaneous removal of the pollutants, not only from the liquid phase, but also from the solid phase when influent is continuously inputted into the system. It is necessary to mention that sedimentation, precipitation and adsorption do not remove the pollutants from the wetland cells. To renew the pollutant adsorption capacity in the solid phase, for example, the pollutants adsorbed from the influent onto the solid surfaces need to be removed by continuous bio-transformation and translocation. At deeper layers of treatment cells, however, both aerobic reaction and bio-translocation rates are relatively low due to low DO (dissolved oxygen) levels and limited root and microbial activities [27]. This can consequently lead to accumulation of the pollutants at the bottom layer of the cells and in turn cause an increase in pollutant concentration in the whole system. Therefore, discharge of the sludge from the bottom layer of the wetland cells may become a necessary measure to maintain a high treatment efficiency of the system. The results shown in Figure 2 confirmed that the discharge of sludge at certain operation intervals was an effective solution. As a matter of fact, pollutant accumulation in the wetland bed not only resulted from the influent input. Constructed wetlands are primary biomass production ecosystems. Along with the degradation of the organic pollutants contained in the influent in the operation period, the released inorganic carbon, nitrogen and phosphorus are simultaneously synthesized by both microorganisms and plants to form biomass [27]. Thus, in the treatment process, a decrease in organic and inorganic pollutants in the effluent is simultaneously associated with an increase in biomass in the wetland ecosystem. This can be crucial in a subtropical region when fast-growing plant species are introduced into wetland cells. In addition to an increase in plant uptake in the process of vegetation growth, plant roots can also contribute a certain amount of oxygen to the granular bed and can provide an additional surface for bio-film attachment, which facilitates the decomposition of organic matters. As root penetration in the bed is only partial, and there is a significant amount of flow under the root zone, removal of the pollutants from the deep layer of the system will always be limited to relatively low levels. Because of this, it is necessary to design a pathway for the discharge of sludge from the bottom layers of wetland beds. In addition, harvest of mature plants and removal of plant litter as well as reuse of discharged sludge are also important procedures to avoid secondary pollution.

5. Conclusions

(1) The tested constructed wetland system was effective to remove TSS, COD_{Cr} and BOD_5 , and the effluent concentrations of these three pollutants met the legislated discharge standards throughout

the operation period. The average total removal rates determined in the tested period for TSS, COD_{Cr}, BOD₅, TN, NH₃-N and TP were 85.3%, 82.4%, 80.1%, 47.5%, 51.0% and 61.8%, respectively. Discharge of sludge at adequate intervals could further enhance the treatment efficiency.

(2) The obtained experimental data confirmed that the removal of the organic, cationic and anionic pollutants in the designed treatment process followed the dynamic pathway of organic pollutant transformation. Relative removal and contribution rates were found to be useful indexes for evaluating the functions and contributions of the sequenced treatment cells.

(3) The results obtained from the present study provided a technical system with basic data and parameters for the design and application of constructed wetland systems for rural household sewage treatment in subtropical regions of China.

6. Patents

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