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# Effectiveness of SOx, NOx, and Primary Particulate Matter Control Strategies in the Improvement of Ambient PM Concentration in Taiwan

Jiun-Horng Tsai<sup>1,2</sup>, Ming-Ye Lee<sup>1</sup> and Hung-Lung Chiang<sup>3,\*</sup>

- <sup>1</sup> Department of Environmental Engineering, National Cheng Kung University, Tainan 701, Taiwan; jhtsai@ncku.edu.tw (J.-H.T.); cceq10502myl@gmail.com (M.-Y.L.)
- <sup>2</sup> Research Center for Climate Change and Environment Quality, National Cheng Kung University, Tainan 701, Taiwan
- <sup>3</sup> Department of Safety Health and Environmental Engineering, National Yunlin University of Science and Technology, Yunlin 64002, Taiwan
- \* Correspondence: hlchiang@yuntech.edu.tw; Tel.: +886-5-536-1489; Fax: +886-5-536-1353

Abstract: The Community Multiscale Air Quality (CMAQ) measurement was employed for evaluating the effectiveness of fine particulate matter control strategies in Taiwan. There are three scenarios as follows: (I) the 2014 baseline year emission, (II) 2020 emissions reduced via the Clean Air Act (CAA), and (III) other emissions reduced stringently via the Clean Air Act. Based on the Taiwan Emission Data System (TEDs) 8.1, established in 2014, the emission of particulate matter 2.5 (PM<sub>2.5</sub>) was 73.5 thousand tons  $y^{-1}$ , that of SOx was 121.3 thousand tons  $y^{-1}$ , and that of NOx was 404.4 thousand tons y<sup>-1</sup> in Taiwan. The CMAQ model simulation indicated that the PM<sub>2.5</sub> concentration was 21.9  $\mu$ g m<sup>-3</sup>. This could be underestimated by 24% in comparison with data from the ambient air quality monitoring stations of the Taiwan Environmental Protection Administration (TEPA). The results of the simulation of the  $PM_{2.5}$  concentration showed high  $PM_{2.5}$  concentrations in central and southwestern Taiwan, especially in Taichung and Kaohsiung. Compared to scenario I, the average annual concentrations of  $PM_{2.5}$  for scenario II and scenario III showed reductions of 20.1%and 28.8%, respectively. From the results derived from the simulation, it can be seen that control of NOx emissions may improve daily airborne PM<sub>2.5</sub> concentrations in Taiwan significantly and control of directly emitted PM<sub>2.5</sub> emissions may improve airborne PM<sub>2.5</sub> concentrations each month. Nevertheless, the results reveal that the preliminary control plan could not achieve the air quality standard. Therefore, the efficacy and effectiveness of the control measures must be considered to better reduce emissions in the future.

**Keywords:** Community Multiscale Air Quality (CMAQ); SOx; NOx; sulfate; nitrate; primary particulate matter

# 1. Introduction

Air pollutants have a significant influence on global human health, quality of life, premature deaths and mortality rate, and the occurrence of different respiratory illnesses related to ambient air quality [1–6]. Each year, approximately seven million premature deaths are recorded across the world. Specifically, one-eighth of the total deaths in the world are due to the combined effects of household and ambient air pollution [7]. Air pollution causes considerable economic impacts and also increases medical expenses, and the related loss of working days reduces economic productivity [8]. In Asia, it is estimated that about 7% of the gross domestic product (GDP) loss was attributed to local air pollution [9]. As far as the health risks associated with this particulate matter are concerned, many epidemiological studies have been published to understand associated cardiovascular diseases, respiratory diseases, cancer, and birth defects [10–17]. Moreover, a few studies have found a link between particles and inflammatory responses in sensitive



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**Copyright:** © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). people [18–20]. As discussed above, the influence of particulate matter (PM) on human health is considered a crucial factor for air quality management.

Particulate matter (PM) pollution, which shortens the lives of humans significantly, contributes to a wide range of diseases [21,22]. Many epidemiological studies have identified a positive correlation associating cardiovascular diseases, respiratory disease, and PM [10,23,24]. In addition, studies have indicated that PM is closely related to increases in the mortality rates of lung cancer and other cardiopulmonary diseases [25–27]. Fine atmospheric aerosol mass concentrations are detrimental due to their influence on the respiratory system, their impact on visibility conditions, and their role in global climate change [28,29]. In addition, to carbonaceous species, sulfate, nitrate, and ammonium are the major species present in fine particulate matter in most situations [29]. In a study by Dongarrà et al. [30], water-soluble ions consisted of a large fraction of  $PM_{10}$  and  $PM_{25}$  and ammonium, sulfate, and nitrate particles accounted for 14-29% of the PM mass concentration. In the European population, the prevalence of the secondary inorganic aerosol (SIA) could be twice as high as that of the primary fine particulate matter [31]. In a Danish cohort study, the results indicated that long-term exposure to PM<sub>2.5</sub>, black carbon/organic carbon (BC/OC), and secondary organic aerosols causes cardiovascular disease and mortality [32]. Therefore, the formation of the secondary aerosol fraction could be essential for understanding the source of fine PM contribution and set up PM control strategies. Furthermore, gas-phase reactions, such as SO<sub>2</sub> and NO<sub>2</sub>, are precursors to the production of H<sub>2</sub>SO<sub>4</sub> and HNO<sub>3</sub> via photochemical reactions, and this makes it possible for them to transfer into particulate matter [33]. Behera and Sharma [34] investigated the degradation of SO<sub>2</sub>, NO<sub>2</sub>, and NH<sub>3</sub>, and they found that these could form secondary inorganic aerosols in an environmental chamber study. Secondary inorganic particulate matter, the dominant component of fine particles, is formed by homogeneous and heterogeneous reactions among gas species [35–37].

In addition, the effectiveness of air pollution control strategies is assessed to improve ambient air quality, reduce exposure, and support human health. The US Environmental Protection Agency (EPA) [38] is following the Clean Air Act (CAA) to use quality models for the assessment of regulations, control strategies, and actions to not only reduce emissions but also improve ambient air quality. Air dispersion models are considered important to inform a number of different regulatory assessments, e.g., the Regulatory Impact Assessments, used to guide federal actions, and the National Ambient Air Quality Standards (NAAQS), used to assess hazardous air pollutants and other health and ecosystem measures. In Europe, the control of air pollutant emissions and related regulations were enacted with the intention to improve air quality across Europe [8]. In recent years, the Community Multiscale Air Quality (CMAQ) model has been implemented with comprehensive halogen sources and chemistry [39] to examine the overall influence of halogen species on air pollution over Europe, as the grid size can affect the predictions of the CMAQ model [40]. Some research studies have been performed in northeastern U.S., including urban health impact studies using the Community Multiscale Air Quality (CMAQ) models to determine air pollution exposure. Results indicate that a larger domain is highly recommended for summer and weekends in eastern U.S. [41].

The observations at these stations can only monitor the level of pollutants around particular locations. However, the emission, transport, and transformation of pollution over the whole region can be shown by these models. Therefore, it is essential to harmonize the criteria with capable and competent models to reproduce air quality features over a particular region in order to officially report national air pollution levels and examine compliance with regulations [6].

In China, about one million deaths per year are attributed to air pollution [42]. The Air Pollution Prevention and Control Action Plan (APPCAP) was implemented to reduce PM<sub>2.5</sub> concentration from 2013 to 2017, and the results indicated a 6.8% reduction rate in mortality attributable to PM<sub>2.5</sub> pollution in 2017 compared to that in 2013. This supports the effectiveness of the APPCAP [42]. The question of how to validate the effectiveness

of air pollution control strategies could be an important focus for future works on air quality management.

In Taiwan, some studies have indicated that the CMAQ simulation is associated with a dynamic NH<sub>3</sub> emissions approach that could improve diurnal and seasonal variations and reduce simulation bias [43]. A real-time air quality forecasting (AQF) system was developed using the Weather Research and Forecasting meteorological model and the CMAQ model for PM<sub>2.5</sub> prediction, and the AQF system was able to reduce the PM<sub>2.5</sub> forecast error and improve the root-mean-square error (RMSE) and mean bias (MB) calculations [44]. In addition, local sources could be the most important emission sources and the emissions of the power plants and iron/steel industries were under control. The management of motor vehicle emissions and construction/road dust should be taken into consideration to improve PM<sub>2.5</sub> levels in Taiwan [45].

Over the past decade,  $PM_{2.5}$  concentrations have presented a decreasing trend, as determined by the Taiwan EPA monitoring station network. The average concentration was 34 µg m<sup>-3</sup> in 2007 and 30 µg m<sup>-3</sup> in 2013. However, it still exceeds the level of 15 µg m<sup>-3</sup> (the annual average concentration) set by Taiwan's National Ambient Air Quality Standard (NAAQS). Therefore, the effectiveness of air pollution control strategies is important to enforce an emission reduction in gas precursors such as SOx and NOx and primary particulate matter for improvement in ambient air quality.

The objective of this study was improvement in ambient  $PM_{2.5}$  concentrations using various control measures under different scenarios. Emissions of primary  $PM_{2.5}$ , SOx, and NOx were estimated, and an air quality model, CMAQ, was employed in simulating the ambient concentration of  $PM_{2.5}$ . The simulated concentrations of each scenario were compared to those generated from basic cases to demonstrate the effectiveness of the  $PM_{2.5}$  control measures.

#### 2. Materials and Methods

#### 2.1. Model Simulation

The US Environmental Protection Agency's Community Multiscale Air Quality (CMAQ) modeling system version 4.7.1, associated with the fifth-generation Pennsylvania State University National Center for Atmospheric Research Mesoscale Model (MM5) version 3.7, was adopted to simulate ambient concentrations of PM<sub>2.5</sub> in Taiwan. The modeling system took into account the chemistry and physics of pollutant transport, as presented in the guidelines suggested by Byun and Schere [46] and simultaneous frameworks proposed by Wong et al. [47]. The system consists of three-level nested domains, as shown in Figure 1, where domain 1 (D1) covers east and south Asia (81 km × 81 km), domain 2 (D2) covers Mainland China and east Asia (Japan and Korea) (27 km × 27 km), and domain 3 (D3) covers southeast China and Taiwan (9 km × 9 km).

Meteorology outputs were processed with the Meteorology–Chemistry Interface Processor (MCIP) [48] (version 4.3). The purpose was to build air-quality-model-ready meteorological input files.

CMAQ default profiles were used for the 81km outermost domain's lateral boundary conditions (BCs) and initial conditions (ICs). Boundary conditions for finer domains are normally generated from coarser domains. Using ICON (Initial condition) and BCON (Boundary condition) modules, part of the preprocessing of the modules of CMAQ involved the preparation BCs and ICs. The CMAQ Chemistry-Transport Model (CCTM) with a Euler Backward Iterative (EBI) chemistry solver for photochemical mechanisms was used. The chemical mechanism used for the gaseous species was the carbon bond mechanism (CB05) [49].

There are many methods and performance indicators to validate the model simulation. Generally, the performance of a dispersion model simulation can be assessed by the mean fractional error (MFE) ( $\leq \pm 50\%$ ) and the mean fractional bias (MFB) ( $\leq \pm 30\%$ ), and extended diagnostic evaluation and sensitivity tests can improve the performance of the model simulation [50–52]. Results of the model simulation are used for a relative rather than



an absolute prediction for the selection of effective control strategies and determination of the uncertainty associated with model predictions [51].

Figure 1. The simulation domain and setting.

The forecasting of the PM concentration is important for air quality management; however, the work focused on the effectiveness of control strategies. The 9 km grid resolution could be coarse for ambient air quality forecasting; the resolution could be enough to evaluate the performance of control measures relevant to this work. More high-resolution simulations can be prepared for future work.

Some indicators were selected to evaluate the performance of the MM5. The average results indicated the MBE (Mean bias error) (0.2 °C) and MAGE (Mean absolute gross error) (1.7 °C) for temperature, MBE (2.4 m s<sup>-1</sup>) and RMSE (Root Mean Square Error) (3.7 m s<sup>-1</sup>) for wind speed, and the WNMB (Wind normalized mean bias) (-5.3%) and WNME (Wind normalized mean error) (29.7%) for wind direction. In the research statement, the wind speed was deemed an overestimation that could be one of the reasons for the underestimation of the PM<sub>2.5</sub> concentration in this work compared with that of the monitoring stations.

For the model simulation, the MFE and MFB were selected for PM<sub>2.5</sub> performance. Results indicated that the MFB and MFE were -0.48 (from -0.33 to -0.68) and 0.77 (from 0.69 to 0.88), respectively. The indices of MFE and MFB were out of the criteria of model simulation performance: MFB  $\leq \pm 0.30$  and MFE  $\leq \pm 0.50$  [51]. The emission database, complex topography, and overestimation of wind speed (results of the MM5) could be the important reasons to lead the PM<sub>2.5</sub> simulation out of criteria.

# 2.2. Emission

The Sparse Matrix Operator Kernel Emissions (SMOKE) model was developed to use matrix–vector multiplication for efficient emission processing. The SMOKE modeling system commonly used to calculate emissions for regional CMAQ applications has recently been fine-tuned to support hemispheric CMAQ applications to allow for more streamlined implementation of the various emission processing steps described above [53].

The emission data used 1 km  $\times$  1 km data from the Taiwan Emission Data System (TEDs) v8.1 by referring to the format of SMOKE. The monitoring concentrations from the Taiwan Environmental Protection Administration (TEPA) air quality monitoring station were included here to evaluate model performance.

## 2.3. Scenarios

In this study, emission scenarios consisting of three parts were evaluated: a basic case and two controlled cases. The basic case, scenario I, represented the emissions from stationary sources and mobile sources and fugitive emissions in the base year (2014) following the emission standards of TEDs 8.1 developed and proven by TEPA. It focused on permitted emissions from stationary sources, showing emission conditions in 2014. The other scenarios were based on TEDs 8.1 and reduced the air pollution emissions following the control strategies of the different scenarios in 2020. According to the contents of the control measures, the emissions of primary PM<sub>2.5</sub>, e.g., sulfur dioxide (SOx) and nitrogen dioxide (NOx), under different scenarios were estimated. Scenario II was an adapted plan (Taiwan Clean Air Act Plan) designed to include more stringent emission standards for stationary sources (especial power plants) and on-road mobile sources, eliminate highpolluting aged vehicles, and promote electric vehicles. For scenario III, it was assumed that more control measures could be conducted beyond scenario II. The measures involved included the elimination of two-stroke motorcycles, the promotion of hybrid vehicles and electric buses, and encouragement of the use of natural gas in power plants (shown in Table 1).

Scenario	Strategies		
<b>Scenario I</b> Year–2014	Baseline year–2014		
<b>Scenario II</b> Taiwan Clean Air Act (TCAA) 2020	Allow emission growth by demand. Power plant: replace the generation sector. Conduct the state implement plan. Make stringent the emission standards for power plants. Identify mobile sources. Replace two-stroke motorcycles. Replace old buses. Fugitive source: control river dust.		
Scenario III TCAA and implement more control strategies Year–2020	<ul> <li>Follow the strategies of scenario II.</li> <li>Implement more control strategies.</li> <li>Make stringent the emission standards of existing sources (power plants; basic iron and steel manufacturing; petroleum and coal product manufacturing; chemical material manufacturing; petrochemical manufacturing; pulp, paper, and paper product manufacturing; and others).</li> <li>Phase out two-stroke motorcycles.</li> <li>Promote clean fuels.</li> <li>Promote liquefied natural gas (LNG) for power plants.</li> <li>Promote use of hybrid oil-electricity passenger cars.</li> </ul>		

Table 1. The control strategies of the different scenarios.

# 3. Results and Discussion

#### 3.1. Baseline Conditions: Scenario I

The emissions of SOx, NOx, and  $PM_{2.5}$  were 121.3, 404.5, and 73.5 thousand tons per year, respectively, for scenario I. The emission distributions of  $PM_{2.5}$ , SOx, and NOx are illustrated in Figure 2. SOx were mainly emitted from stationary sources (a fraction of 90% from stationary sources, 36% from power plants, 14% from the iron and steel industry, and

9% from the manufacture of chemical materials) (shown as Figure 3). A total of 90% of NOx was attributed to stationary sources and mobile sources; heavy-duty diesel vehicles (20%), power plants (18%), gasoline passenger cars (8%), diesel buses (6%), and the manufacture of chemical materials (5%) were considered dominant sources (shown as Figure 4). The most abundant  $PM_{2.5}$  emissions were from fugitive emissions (such as cooking fumes from restaurants (10%), construction engineering (6%), and agricultural burning (6%)), mobile source emissions (11% from motor vehicles, 6% from gasoline passenger cars, and 6% from heavy-duty diesel vehicles), and stationary sources (the iron and steel industry and power plants), as shown in Figure 5.



Figure 2. The emission distribution (ton  $y^{-1}$ ) of PM<sub>2.5</sub> (a), SOx (b), and NOx (c) in the baseline year (scenario I).





**Figure 3.** SO<sub>2</sub> emission sources for different scenarios (ton  $y^{-1}$ ).



Other\*: 4-stroke Motorcycle, Petrochemicals Manufacturing, Bus, Petroleum and Coal Products Manufacturing, Light Duty Gasoline Vehicle, Other Area Sources, Agriculture Burning, Business Passenger Car-gasoline, Light-duty Diesel Vehicle, 2-stroke Motorcycle, Commercial Combustion-heavy Oil, Aircraft Combustion Emission, Ship and Boat Combustion-diesel, Passenger Car-diesel, Passenger Car-LPG, Commercial Combustion-diesel

scenarios III

scenarios II

scenarios I





Other\* : Chemical Material Manufacturing, Non-metallic Mineral Products Manufacturing, Bus-diesel, River Fugitive Dust, Farming, Light Duty Gasoline Vehicle, Metallic Mineral Products Manufacturing, Petrochemicals Manufacturing, Light-duty Diesel Vehicle, Business Passenger Car-gasoline, Bus, Pulp, Paper and Paper Products Manufacturing, Motor Vehicle (Non-paved road), Mining, Petroleum and Coal Products, Manufacturing, Passenger Car-diesel, Ship-boat-diesel Power Generator, Ship-boat-diesel, Passenger Car-LPG, 2-stroke motorcycle

**Figure 5.**  $PM_{2.5}$  emission sources for different scenarios (ton  $y^{-1}$ ).

SOx emission distributions were mainly concentrated in urban and industrial complex areas and harbors. NOx emissions were in the urban areas of western Taiwan, and transportation activities involving the use of freeways and highways were the main sources of NOx emissions.  $PM_{2.5}$  emissions were abundant in the urban areas of western Taiwan, where activities of transportation take place, as given in Figure 2.

The annual average concentration of  $PM_{2.5}$  was 21.9 µg m<sup>-3</sup> after the CMAQ simulation; there was a 24% underestimation in comparison with the average monitoring data derived from the Taiwan EPA air quality monitoring stations in 2014, as depicted in Figure 6. The MFB and MFE did not meet the criteria, which could be important reasons for the underestimation of the PM simulation. The model simulation did not clearly determine high-PM episodes, especial in January, which was a limitation of the application of the model to predict the ambient air quality. PM concentration forecasting is important for air quality management; however, the study evaluated the effectiveness of control strategies under different scenarios.



Figure 6. The modeling simulation of  $PM_{2.5}$  concentrations ( $\mu g m^{-3}$ ) of the ambient air quality monitoring stations.

In addition, this work discussed SOx and NOx reactions and their transfer into particulate matter. However, we do not believe that NH<sub>3</sub> and the secondary organic aerosol come from photochemical reactions. According to the literature, the total mass of sulfate, nitrate, and ammonium constituted 39.9-44.3% of the PM<sub>2.5</sub> mass [54] in southern Taiwan, and the mass fractions of secondary organic matter in PM<sub>2.5</sub> were estimated to be 14–22% [55]. Over 50% of the mass of fine particulate matter came from the chemical reactions required to form secondary PM<sub>2.5</sub>. The overestimation of wind speed and chemical reactions did not include NH<sub>3</sub> and the secondary organic aerosol, which could explain the underestimation of ambient PM<sub>2.5</sub> concentrations.

The monthly average  $PM_{2.5}$  simulation concentrations were in the order 29.7 µg m<sup>-3</sup> (April) > 28.0 µg m<sup>-3</sup> (January) > 19.7 µg m<sup>-3</sup> (October) > 10.3 µg m<sup>-3</sup> (July). Relatively high  $PM_{2.5}$  concentrations were observed in central and southern Taiwan, as presented in Figure 7. Moreover, relatively high PM concentrations were determined in April, and the same data were also collected from TEPAair monitoring stations. Therefore, the PM concentration was observed to be high in winter and spring. In contrast, the PM concentration was low in summer. That was because the rainy season is from May to September and sometimes can last until October or November.

Primary PM<sub>2.5</sub>, known as directly emitted PM, is the dominant fraction of the total PM<sub>2.5</sub>. The concentrations of nitrate were higher than those of sulfate in central and southern Taiwan in January, April, and October, as described in Figure S1. Based on the model simulation and statistical analysis, results of the correlation between PM<sub>2.5</sub> and the composition of airborne PM<sub>2.5</sub> showed that the PM<sub>2.5</sub> concentrations were closely related to nitrate (r = 0.93), primary PM<sub>2.5</sub>, (r = 0.80), and sulfate (r = 0.70), with a strong correlation (r > 0.7).

According to the emissions of the baseline year, high sulfate concentrations were located in the southwest region and the highest abundant concentrations were observed in April, as can be seen in Figure S2. Arather high nitrate concentration was found in central and southern Taiwan. In addition, in January, the highest average nitrate concentration was found around the Kaoping air basin, located in southwestern Taiwan (Figure S3). Primary PM<sub>2.5</sub> was high in central and southwestern Taiwan. The highest concentration period was observed in January, especially in the southwestern area of Taiwan.



**Figure 7.**  $PM_{2.5}$  concentration (µg m<sup>-3</sup>) simulations for January (**a**), April (**b**), July (**c**), and October (**d**) under baseline year conditions.

According to the model simulation, primary  $PM_{2.5}$  was found to contribute 20–40% of the mass of the total  $PM_{2.5}$ . High primary  $PM_{2.5}$  contributions were found to be in the northern and central parts of western Taiwan. The nitrate mass fraction tended to increase in the seasons with high levels of PM mass, especially in January and April, in central and southwestern Taiwan.

# 3.2. Scenario II: The Taiwan Clean Air Act (TCAA) Plan Adopted

The reduction in SOx was 17.9 thousand tons per year, and the reduction in NOx was 158 thousand tons per year. In addition, the reduction in  $PM_{2.5}$  emissions was 13.9 thousand tons per year. For scenario II, the emissions of SOx, NOx, and  $PM_{2.5}$  were 103.4, 246.3, and 59.7 thousand tons per year, respectively, as shown in Figure 8.



**Figure 8.** The emission distribution (ton  $y^{-1}$ ) of PM<sub>2.5</sub> (**a**), Sox (**b**), and NOx (**c**) for scenario II.

The SOx reductions were mainly from changes in power plants (the installation of new electricity-generating units and change of fuels from coal or heavy oil to liquefied natural gases). The emission of 12thousand tons was attributed to changes in basic iron

and steel production, 1.1 thousand tons to the manufacture of chemical materials, and 0.7 thousand tons to other sources, as depicted in Figure 3 for scenario II.

NOx reduction via power plants was 73 thousand tons per year, via heavy-duty diesel vehicles was 43 thousand tons per year, via gasoline passenger cars was 14 thousand tons per year, and via buses powered by diesel was 13 thousand tons per year (shown as Figure 4, scenario II).

Considering different emission sources, the PM<sub>2.5</sub> concentration could be reduced. The major reduction came from the following human activities and motor vehicles: (1) heavy-duty diesel vehicles, 2.1 thousand tons per year; (2) motor vehicle driving, 1.7 thousand tons per year; (3) agricultural burning, 1.1 thousand tons per year; (4) passenger cargasoline, 1.1 thousand tons per year; and (5) two-stroke motorcycles, 1.0 thousand tons per year, as illustrated in Figure 5, scenario II.

The monthly average  $PM_{2.5}$  simulation concentrations of scenario II were shown as Figure 9.



Figure 9. PM<sub>2.5</sub> concentration (µg m<sup>-3</sup>) simulations for January (a), April (b), July (c), and October (d) under scenario II.

Primary PM<sub>2.5</sub> concentrations increased from northwestern (2.8–5.2  $\mu$ g m<sup>-3</sup>) to southwestern (3.4–9.6  $\mu$ g m<sup>-3</sup>) Taiwan. The highest concentration was determined in southwestern Taiwan in January. In east Taiwan, the primary PM concentration was 0.8–2.6  $\mu$ g m<sup>-3</sup>, as described in Figure S4.

Regarding the sulfate concentration distribution, high sulfate concentrations were determined in southern Taiwan and their concentration could be over 2.5  $\mu$ g m<sup>-3</sup> in January. The sulfate concentration was also high in central Taiwan (2.1  $\mu$ g m<sup>-3</sup>). In April, the sulfate concentration inland was higher than that near the seaside areas in central Taiwan. The sulfate concentrations were 2.1–2.8  $\mu$ g m<sup>-3</sup> in western areas and 1.4–1.5  $\mu$ g m<sup>-3</sup> in eastern areas. In July, the sulfate concentration was reduced to half of the concentration observed in January and April. After the rainy season, the concentration increased to 1.1–2.1 $\mu$ g m<sup>-3</sup> in the west and 0.9–1.0  $\mu$ g m<sup>-3</sup> in western Taiwan in October, as shown in Figure S5.

Nitrate concentrations were high (January:  $1.4-7.1 \ \mu g \ m^{-3}$ ; April:  $1.5-4.7 \ \mu g \ m^{-3}$ ; July:  $0.2-0.7 \ \mu g \ m^{-3}$ ; October:  $0.2-3.6 \ \mu g \ m^{-3}$ ) in western Taiwan, especially southwestern Taiwan ( $0.4-7.1 \ \mu g \ m^{-3}$ ). Lower concentrations ( $0.00.7 \ \mu g \ m^{-3}$ ) were found in July due to the rainy season in this period. In east Taiwan, the nitrate concentrations were within the range of  $0.0-0.4 \ \mu g \ m^{-3}$ , as presented in Figure S6.

State and local governments have planned to eliminate older two-stroke motorcycles and older buses and control the fugitive dust emissions from the river. The annual average concentration of  $PM_{2.5}$  in scenario II was 17.5 µg m<sup>-3</sup>. There was a 20.1% concentration reduction in ambient  $PM_{2.5}$  between scenarios I and II. The monthly average reductions in  $PM_{2.5}$  concentrations were 18.3% in October, 17.2% in April, 14.4% in January, and 14.3% in July. High  $PM_{2.5}$  concentration reduction was observed in central and southern Taiwan, as shown in Figure S2a. Primary  $PM_{2.5}$  is still the dominant component. The

results of the correlation between  $PM_{2.5}$  and the composition of airborne  $PM_{2.5}$  proved the  $PM_{2.5}$  concentration to be related to nitrate (r = 0.92), primary  $PM_{2.5}$  (r = 0.82), and sulfate (r = 0.76).

In January, the primary  $PM_{2.5}$  mass fractions were 28–38% for different air basins. The sulfate mass in  $PM_{2.5}$  fractions (9.1–24%) was higher than that observed in both northern and eastern Taiwan. High nitrate mass fractions were determined in central (16%) and southern (18%) Taiwan. In April, primary  $PM_{2.5}$  mass fractions were 22–32%, the sulfate fraction was 12–24%, and the nitrate fraction was 1.2–12%. A low PM mass was observed in July. The primary PM and sulfate mass fractions were found to be increasing, and a decrease in the nitrate fraction could be attributed to the rainy season in July, as nitrate compounds easily dissolve and react under high-moisture conditions. In October, the PM and sulfate fraction increased instead.

# 3.3. Scenario III: Conducted TCCA and More Control Strategies

The reduction in SOx was 59.5 thousand tons per year, NOx was reduced to 200 thousand tons per year, and  $PM_{2.5}$  reduction was 18.8 thousand tons per year when compared with the emissions of scenario I. The emissions of SOx, NOx, and  $PM_{2.5}$  were 61.8, 204.7, and 54.8 thousand tons per year, respectively, for scenario III, as shown in Figure 10. The differences of emissions between scenarios II and III are shown in Figure S7. The reductions were mainly because of phasing out two-stroke motorcycles, promoting electric motor vehicles and buses in downtown, and increasing liquefied natural gas (LNG) consumption at power plants.



Figure 10. The emission distribution (ton  $y^{-1}$ ) of PM<sub>2.5</sub>, Sox, and NOx for scenario III.

For scenario III, the control strategies followed those adopted in scenario II and more strategies and control measures for SOx and NOx such as the phasing out of two-stroke motorcycles, the clean fuel program (liquefied natural gas (LNG) is used in power plants), and the electric bus program were included, as given in Figures 3 and 4, scenario III).

SOx reductions were mainly from LNG application in power plants (a reduction of 31.7 thousand tons per year), and reductions from other categories were less than 1.5 thousand tons per year.

NOx reductions were mainly from power plants and electric buses. The former contributed 24 thousand tons per year. The latter contributed 4.0 thousand tons per year. Chemical material manufacturing also contributed 2.2 thousand tons per year, and the cement industry contributed 3.3 thousand tons per year.

The monthly average  $PM_{2.5}$  simulation concentrations of scenario III were shown as Figure 11.



Figure 11. PM<sub>2.5</sub> concentration (µg m<sup>-3</sup>) simulations for January (a), April (b), July (c), and October (d) under scenario III.

As to the primary PM<sub>2.5</sub> distribution (0.8–9.2  $\mu$ g m<sup>-3</sup>), the results indicated that high PM<sub>2.5</sub> concentrations were observed in southern (3.2–9.2  $\mu$ g m<sup>-3</sup>) and central (3.2–7.5  $\mu$ g m<sup>-3</sup>) Taiwan. The highest primary PM<sub>2.5</sub> concentration was about 10  $\mu$ g m<sup>-3</sup> in southern Taiwan in January. Low PM<sub>2.5</sub> concentrations (0.8–3.2  $\mu$ g m<sup>-3</sup>) were determined in July. (shown as Figure S8).

 $PM_{2.5}$  emissions were reduced via power plants (a reduction of 2.8 thousand tons per year), and the sulfate concentration distribution was 0.7–2.3 µg m<sup>-3</sup>. The results indicated the highest sulfate concentration (1.1–2.3 µg m<sup>-3</sup>) in southern Taiwan. There was a relatively high sulfate concentration (1.3–2.1 µg m<sup>-3</sup>) in central Taiwan in comparison with that in north Taiwan (1.1–1.9 µg m<sup>-3</sup>) and that in east Taiwan (0.7–1.4 µg m<sup>-3</sup>). The months of January (1.4–2.2 µg m<sup>-3</sup>), October (0.9–1.8 µg m<sup>-3</sup>), and April (1.4–2.3 µg m<sup>-3</sup>) presented relatively high sulfate levels, with relatively low ones in July (0.7–1.3 µg m<sup>-3</sup>) due to the rainy season from May to September in Taiwan, as illustrated in Figure S9.

In terms of the nitrate concentration distribution  $(0-6.2 \ \mu g \ m^{-3})$ , high concentrations  $(0.3-6.2 \ \mu g \ m^{-3})$  were observed in southern Taiwan. High nitrate concentrations were found in central Taiwan in April. In July, the nitrate concentrations were less than  $0.5 \ \mu g \ m^{-3}$  because of the rainy season in Taiwan (shown as Figure S10).

The annual average concentration of  $PM_{2.5}$  for scenario III was 17.5 µg m<sup>-3</sup>. Compared with the basic case in 2014, the simulated concentration was reduced by 28.8%. The monthly average  $PM_{2.5}$  concentration reductions were 10.5% (April), 10.0% (October), 8.6% (July), and 5.5% (January), sequentially. A significant reduction in the  $PM_{2.5}$  concentration was also observed in central and southern Taiwan, as described in Figure 11.

In January, sulfate concentrations were within the range of 1.7–2.2  $\mu$ g m<sup>-3</sup> in western Taiwan and a high-abundance area was located in southwestern Taiwan. High nitrate concentrations were located in central and southwestern Taiwan, and the concentration range was 4.1–6.2  $\mu$ g m<sup>-3</sup> (nitrate fraction of PM<sub>2.5</sub> was 15–17%). The primary PM<sub>2.5</sub> concentration ranged from 7.5 to 9.2  $\mu$ g m<sup>-3</sup> (the fraction was 31.3–38.6%), and the corresponding areas were from the central part of Taiwan to the southwestern part of Taiwan. In the western part of Taiwan, the PM concentration increased from the northern area to the southern area.

In April, sulfate concentrations were  $1.9-2.3 \ \mu g \ m^{-3}$ , and these were slightly increased in comparison with those observed in January in the same area. For nitrate, the concentrations were within the range of  $3.3-3.8 \ \mu g \ m^{-3}$  (the nitrate fraction of PM<sub>2.5</sub> was 10–11%) in western Taiwan. High primary PM<sub>2.5</sub> levels were located at central (concentration:  $5.6 \ \mu g \ m^{-3}$ ) and southern (concentration:  $5.7 \ \mu g/m^3$ ) Taiwan. In July, a high sulfate concentration was in the central air basin, and its concentration was  $1.3 \ \mu g \ m^{-3}$ . In addition, the concentration of nitrate was high ( $1.5 \ \mu g \ m^{-3}$ ) as well in the air basin. The primary PM concentration was about  $3.2 \ \mu g \ m^{-3}$  in the central and southern air basins. Sulfate ( $1.5-1.8 \ \mu g \ m^{-3}$ ), nitrate ( $2.0-2.7 \ \mu g \ m^{-3}$ , a fraction representing 7.9–11% of the PM

mass), and primary PM concentrations increased in October compared with those observed in July.

# 3.4. Effectiveness of Pollutant Reductions

Three major emission types consisting of NOx, SOx, and directly emitted  $PM_{2.5}$  were determined to contribute to the ambient  $PM_{2.5}$  aerosol mass. The implemented control strategies of the air quality management program (AQMP) help reduce SOx, NOx, and direct  $PM_{2.5}$  emissions substantially. As presented earlier, the trends in SOx, NOx, and  $PM_{2.5}$  emissions suggest a direct relationship between lower emissions and improvements in air quality.

In scenario II, it was shown that replacement of older power plants could be an effective strategy for reducing directly emitted PM, SOx, and NOx. In addition, the replacement of old heavy-duty diesel vehicles could reduce NOx and PM emissions, as shown in Figures 3–5.

In scenario III, SO<sub>2</sub> emission reductions were mainly from the use of clean fuel (LNG) in power plants, basic iron and steel manufacturing, chemical material manufacturing, and petroleum and coal product manufacturing. NOx reductions were from the use of LNG in power plants, the phasing out of heavy-duty diesel vehicles, and the replacement of diesel buses with electric buses. LNG power plants, vehicle phase-outs, exhaust control devices, and the retrofitting of old heavy-duty diesel vehicles could be relatively effective control strategies for reducing PM emissions in scenario III.

It is useful to evaluate the relative value of the ambient microgram per cubic meter improvements in ambient  $PM_{2.5}$  under the per ton precursor emission reductions. Table 2 presents the relative contributions of precursor emission reductions to ambient  $PM_{2.5}$  concentrations in scenario II and scenario III. The analysis determines that SOx emission reductions had the lowest return in terms of micrograms per cubic meter on the per ton emissions of  $PM_{2.5}$  reductions, about half that of NOx reductions. NOx emission reductions were about twice as effective as SOx emission reductions for ambient  $PM_{2.5}$  concentration reduction. However, the reductions in directly emitted  $PM_{2.5}$  emissions were approximately 14–19 times more effective than SOx emission reductions. The results of scenarios II and III indicated that the sequence of the effectiveness of  $PM_{2.5}$  concentration reduction was primary  $PM_{2.5} > NOx > SOx$ .

Precursor	PM Fraction	Scenario II	Scenario III
SOx	Sulfate	1	1
NOx	Nitrate	2.1	1.7
PM <sub>2.5</sub>	Primary PM <sub>2.5</sub>	19	14

Table 2. Relative contributions of precursor emission reductions of the simulated PM<sub>2.5</sub> concentration.

## 4. Conclusions

A simulated high PM<sub>2.5</sub> concentration of the baseline scenario was observed in central and southern Taiwan. The period with the higher PM level was found to range from October to April of the following year. This was because the rainy season is from May to September in Taiwan. Directly emitted  $PM_{2.5}$  was found to be the major component of total  $PM_{2.5}$ . Nitrate was considered a higher fraction of the  $PM_{2.5}$  mass than sulfate in high- $PM_{2.5}$ -concentration areas and periods. There was also a relatively high correlation between simulated  $PM_{2.5}$  concentration and nitrate during high-PM-concentration periods. The emission reduction rates of scenario II and scenario III were 20.1% and 28.8%, respectively, in comparison with the rate of scenario I. The replacement of antiquated power plants and the use of clean LNG fuel, the replacement of old motor vehicles, and stringent emission standards for stationary sources could be effective strategies for not only reducing emissions but also improving ambient air quality. However, the reduction in emissions by enactment of the Taiwan Clean Air Act Plan and enhanced control measures is still unable to achieve the annual average  $PM_{2.5}$  standard of 15 µg m<sup>-3</sup>. SO<sub>2</sub> emission reduction might help improve ambient air  $PM_{2.5}$ . In addition, the relative contribution of precursor emission reductions to  $PM_{2.5}$  suggests that NOx emission control could lessen the problem of high  $PM_{2.5}$  concentrations and primary  $PM_{2.5}$  emission control might improve ambient  $PM_{2.5}$  concentrations in Taiwan, regardless of time and space. For the simulation in this work, the metrics of the MFE and MFB could not meet the criteria of model performance. The modeling system can be updated to more recent versions, which could be one of the routes to improving its simulation performance benchmarks.

**Supplementary Materials:** The following are available online at https://www.mdpi.com/article/10 .3390/atmos12040460/s1, Figure S1: The stimulation of primary PM<sub>2.5</sub> concentrations for January (a), April (b), July (c) and October (d) under the baseline year condition, Figure S2: The stimulation of sulfate concentrations for January (a), April (b), July (c) and October (d) under the baseline year condition, Figure S3: The stimulation of nitrate concentrations for January (a), April (b), July (c) and October (d) under the baseline year condition, Figure S4: The stimulation of primary PM2.5 concentrations for January (a), April (b), July (c) and October (d) under Scenario II, Figure S5: The stimulation of sulfate concentrations for January (a), April (b), July (c) and October (d) under the scenario II, Figure S6: The stimulation of nitrate concentrations for January (a), April (b), July (c) and October (d) under Scenario II, Figure S7: The stimulation of Primary PM2.5 concentrations for January (a), April (b), July (c) and October (d) under the scenario II, Figure S8: The stimulation of sulfate concentrations for January (a), April (b), July (c) and October (d) under the scenario II, Figure S6: The stimulation of nitrate concentrations for January (a), April (b), July (c) and October (d) under Scenario II, Figure S7: The stimulation of Primary PM2.5 concentrations for January (a), April (b), July (c) and October (d) under the scenario III, Figure S8: The stimulation of sulfate concentrations for January (a), April (b), July (c) and October (d) under the scenario III, Figure S9: The stimulation of Nitrate concentrations for January (a), April (b), July (c) and October (d) under the scenario III, Figure S10: Comparison the differences of Scenario II (Figure 5) and III (Figure 6). (a) PM2.5; (b) Sox; (c) NOx.

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