



Crop Response to Low Phosphorus Bioavailability with a Focus on Tomato

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Abstract: Tomato (*Solanum lycopersicum* L.) is a high-value crop that has potential to enhance its P-use efficiency. While phosphorus (P) is an essential nutrient, supplies are finite and much of the P supply in agricultural soils is not bioavailable after application due to reactions such as soil adsorption, immobilization, or precipitation. Low-P bioavailability results in reduced growth, so plants may mobilize soil-bound P by altering root morphology, exuding root-derived compounds, or forming symbiosis with microorganisms. This review discusses (i) the significance of P in plants and agroecosystems, (ii) within-plant response to changing P bioavailabilities, and (iii) strategies to enhance P-acquisition efficiency (PAE). Phosphorus forms fluctuate in the soil and potential approaches to increase the bioavailable pool of P may focus on processes such as desorption, mineralization, or dissolving precipitated P-compounds. To enhance these processes, roots may alter their spatial arrangement, exude protons to acidify the rhizosphere, exude carboxylates to solubilize bound-P, exude phosphatase to mineralize organic P, or enhance symbiosis with native microbes. High PAE allows for use of accumulated soil P as opposed to relying on fertilizer application to meet crop demand.

Keywords: tomato; Solanum lycopersicum; phosphorus-use efficiency; phosphorus acquisition

1. Introduction

Phosphorus (P) is one of 17 essential elements that plants require to develop and function, so flexibility in plant metabolism and bioenergetics may help crops cope with low-P conditions [1,2]. Phosphorus is taken up as acid orthophosphate anions (primarily as $H_2PO_4^-$ at pH < 7.2 or as HPO_4^{2-} at pH > 7.2) and is an integral part of the chemical structures of (i) adenosine mono-, di-, tri-phosphate (AMP: $C_{10}H_{14}N_5O_7P$, ADP: $C_{10}H_{15}N_5O_{10}P_2$, ATP: $C_{10}H_{16}N_5O_{13}P_3$), (ii) nucleic acids, and (iii) phospholipids. Phosphorus is involved in the regulation of metabolic pathways such as energy transfer, protein activation, and amino acid synthesis [1,2].

Phosphorus deficiency can negatively impact crop yield and, in severe deficiency, can lead to death. However, basing nutrient management on visual deficiency symptoms is unreliable. For example, anthocyanin accumulation is typically associated with a deficiency in P, but may not be a quality indicator on its own. In their review of foliar anthocyanin, Close and Beadle [3] outline a range of contexts for accumulation of this compound; anthocyanin may accumulate in response to ultraviolet (UV) radiation, browsing herbivores, or pathogenic fungi in addition to nutrient deficiency. Instead of visual symptom reliance, soil and tissue testing may provide a reliable analysis of soil P content to establish fertility guidelines. Optimal plant P concentrations typically range from 0.1% to 0.5%, averaging approximately 0.2% [2]. The critical leaf concentration associated with a 10% reduction

in dry matter yield for perennial ryegrass (*Lolium perenne* L.) is 0.0021% [4] and for dwarf saltwort (*Salicornia bigelovii* Torr.) is 0.00078% [5]. The critical value for P may decline with age as shown by purple bush bean (*Macroptilium atropurpureum* Moc. and Sessé ex DC), where the critical concentration decreased from 0.03% at 41 days to 0.01% at 77 days [6]. The range of shoot P content tends to be higher on average for crops inoculated with mycorrhizal fungi (0.3–0.4%) compared to uninfected plants (0.1–0.3%) [7]. P-toxicity symptoms typically manifest when leaf-P concentration exceeds 1% dry weight [8], and this has been shown with tomato [9]. Fageria et al. [10] outlines P toxicity symptoms as increased red speckling in sorghum (*Sorghum bicolor* L.) and reduced yield in sorghum and cotton (*Gossypium hirsutum* L.).

Crops may cope with low-P by enhancing P-use efficiency (PUE). High PUE necessitates both elevated P acquisition by the roots and enhanced use of P in processes resulting in healthy growth. PUE may be described as the sum of P-acquisition efficiency (PAE) (ability to acquire P from the soil) and P-utilization efficiency (PUE) (ability to internally use P to result in better growth). A higher PUE could be achieved by selecting for traits that boost either PAE or PUtE. To genetically improve a crop's P-use efficiency, there needs to be an emphasis on enhancing root biology which could lead to reduced fertilization and expanding agriculture to low-P soils [11].

Tomato (*Solanum lycopersicum* L.) is a high-value perennial vegetable crop in its native habitat, but is commonly grown as an annual. Tomato is a member of the family Solanaceae, the subfamily Solanoideae, and the tribe Solaneae. The taxonomy has fluctuated between the Linnaeus identification of *Solanum lycopersicum* in 1753 and the Miller identification of *Lycopersicum esculentum* in 1768 [12]. The leaves are alternate in arrangement, simple in complexity, and pinnate in venation. Tomato plants possess a showy single flower that is bisexual, radially symmetric, and yellow. The corolla of the flower may fuse forming a corolla tube. The corolla typically contains five connate petals. The calyx has five persistent, connate sepals. The five distinct stamens that open by slits are adnate to the corolla and alternate with petals. Each flower has two carpels and a superior ovary; the placentation is axile. Tomato fruits are multi-seeded berries with small, flat seeds [13]. Many tomatoes are self-pollinating; others may cross-pollinate by anemophily (wind pollination) [14].

Tomato is a nutrient-dense and healthy vegetable. Tomato fresh tissue contains 5.0–7.5% dry matter that is mostly composed of fructose (25%) and glucose (22%) [15]. Tomato also contains four major carotenoids: alpha-carotene, beta-carotene, lutein, and lycopene [14,16,17]. Tomato health benefits include decreasing the risk of cancers (e.g., pancreatic and esophageal) and cardiovascular disease [16]. Tomato is one of the most often cultivated vegetable crops worldwide due to its numerous health benefits and high demand [18]. Worldwide tomato production was across over 5.8 million ha with a production quantity of nearly 244 million tonnes in 2018. The top five tomato producing countries in 2018 were China (approximately 60.6 million tonnes), India (approximately 19.4 million tonnes), United States (approximately 12.6 million tonnes), Turkey (approximately 12.2 million tonnes), and Egypt (approximately 6.6 million tonnes) [19].

There are few polymorphisms in tomato, but there is high morphological variation that is readily detected in self-pollinated varieties. Goals of tomato breeding programs vary widely by location and individual needs, but overarching goals have tended to be the optimization of yield in the 1970s, shelf-life in the 1980s, taste in the 1990s, and nutritional quality in the 2000s. A prominent issue for tomato breeders today is selecting for resistance to pests and pathogens [20]. While an important issue, tomato breeding goals tend to not focus as much on enhancing nutrient-use efficiency. There may be genotypic diversity in tomato regarding P-use efficiency as exemplified by Coltman et al. [21] who noticed an increase in dry weight of 77% among efficient tomato compared to inefficient tomato when growing in low-P conditions.

Crop PAE augmentation seems possible through regulating traits such as microbial symbiosis, root hair development, or organic acid exudation. An enhanced PAE could allow crops to survive in soil by mobilizing P that would otherwise be unavailable to the plant. The purpose of this review is to comprehensively analyze literature on low-P stress-induced mechanisms to better understand traits

and practices that may enhance tomato PAE. This review delineates (i) the role of P in an agroecosystem, (ii) tomato plant response mechanisms to low-P stress, and (iii) PAE enhancing strategies. Potential strategies to heighten PAE include rapid P sensing, optimization of root growth to exploit a given volume of soil, exudation of root-derived compounds, and microbial symbiosis. Understanding the structure–function relationships of morphological and physiological adaptations to low-P stress opens opportunities to increase PAE and agricultural sustainability.

2. Phosphorus in an Agroecosystem

2.1. Phosphorus Pools and Sources

As opposed to the cycles of other nutrients like nitrogen (N) [22], sulfur (S) [23], and carbon (C) [24], there is no gaseous form of P available for fixation. Therefore, P is typically supplied to plants through fertilizer application either by foliar application to the leaves and stems or to the ground, usually by banding. Although foliar fertilization is typically used for micronutrients [25], there are still benefits to using foliar sources of P. Salt-stressed tomato (60 mM NaCl applied via foliar application twice each week) recovered from P-deficiency, increased dry matter production, and increased chlorophyll concentration as a result of foliar application of supplemental P [26]. Similarly, common bean was also shown to optimize performance in saline soil when supplied with foliar P fertilizer (10 mM MAP) [27]. Additionally, Mosali et al. [28] concluded from their experiments with winter wheat (*Triticum aestivum* L.) that low rates of foliar P application may ameliorate mid-season (Feekes 10.54) P-deficiency and ultimately, lead to greater yield. Phosphorus in conventional fertilizer is categorized based on solubility: (i) water soluble P, (ii) citrate soluble P (iii) citrate insoluble P, (iv) bioavailable or phytoavailable P, and (v) total P (often expressed as P₂O₅) [2]. Fertilizers tend to resupply bioavailable pools of P through dissociating into orthophosphate and the dominant cation in the fertilizer.

Calcium phosphates (single/triple superphosphate) were the most used inorganic P fertilizers until the 1970s when ammonium phosphates (mono/di/urea ammonium phosphate, ammonium polyphosphate) became more popular. Potassium phosphates, including mono- and dipotassium phosphate, are highly water soluble and have high P and K concentrations. Phosphoric acid is another popular inorganic fertilizer source derived from reacting rock phosphate with sulfuric acid. Phosphorous acid (HPO₃^{2–}), also known as Phi, phosphite, or phosphonate, is an isotere of phosphate where an oxygen (O) is replaced by hydrogen (H) [29]. Unlike phosphoric acid, phosphorous acid has been shown to repress P limitation responses such as root hair initiation [29,30]. Plants lack an ABC-type Phi uptake system, a complex that is present in the genomes of organisms able to oxidize Phi to phosphate. Soil bacteria including Desulfotignum phosphitoxidans, D. balticum, and D. toluenicum are able to oxidize Phi (via oxidase) to phosphate [31]. Phi is typically used as a fungicide and may have preventative efficacy towards *Phytophthora* vectored diseases [32]. Beneficial effects of Phi on tomato growth have been observed, but only when Phi was supplied with phosphate at equal P concentrations of 20 mg/L [33,34]. Rock phosphate is the primary material used to manufacture P fertilizers. Rock-phosphate typically contains 25% to 36% P₂O₅ with citrate solubility ranging from 3–20% total P. Finely ground rock phosphate reacts as follows [2]:

$$Ca_{10}(PO_4)_6F_2 + 12H_2O \leftrightarrow 10Ca^{+2} + 6H_2PO_4 + 2F + 12OH_2O_4$$

While the most cost-effective current means of producing P fertilizers is mining phosphate rock, researchers such as Cordell et al. [35] have noted that the rising price and scarcity of phosphate rock coupled with eutrophication concerns necessitates increased agricultural efficiency (increasing crop yields per unit input of P). These researchers call for integrated nutrient management plans that meet crop demand through utilization of a range of inorganic and organic sources and improving timing and rates of fertilization, enhancing chemical and physical properties of soil, and stimulating microbial symbiosis in the rhizosphere [35]. We must, therefore, shift reliance from nonrenewable mineral

resources and instead to accumulated soil P supplies. Soil P concentration typically ranges from 0.003 ppm to 3 ppm, usually averaging 0.05 ppm [2]. Depending on soil test results (Mehlich 3 soil extraction method), UF IFAS recommended P fertilization as P_2O_5 for tomato production is 120–150 lb/A for low, 100 lb/A for medium, and 0 lb/A for high soil test index [25].

Organic P amendments have variable P content ranging from 0.1% (lawn clippings) to 7% (biosolids) dry matter [2]. Decomposable plant residue may be turned into stable organic humus through the activities of fungi, bacteria, protozoa, and other decomposing organisms. The range of organic P is variable, but typically represents 50% total soil P. Organic P content decreases with depth and with decreasing soil organic carbon [2].

2.2. Soil Phosphorus Fluxes

Common reactions of the P cycle include chelation, precipitation, adsorption, and mineralization. A chelating agent is a multi-dentate ligand that is able to donate pairs of electrons to metal ions and form stable metal complexes. Natural (such as citrate) or synthetic (such as Ethylenediaminetetraacetic acid (EDTA)) chelating agents interact with P-binding sites to liberate bound P [36,37]. A precipitation reaction occurs when a cation and anion interact in aqueous forms to produce a solid or a semi-crystalline structure such as aluminum (Al) or iron (Fe) phosphates [38]. Solution phosphate may precipitate with mineral nutrients such as calcium (Ca) [36]:

$$PO_4^{3-}(aq) + Ca^{2+}(aq) \rightarrow Ca_3(PO_4)_2(s)$$

While adsorption removes P from pools available for plant uptake, it also removes P from bodies of water, thereby controlling for eutrophication in water bodies. Solution phosphate may adsorb onto metal oxides and metal hydroxides. Removal efficiency depends on environmental conditions (pH, ion strength, competitive ions, dosage, phosphate concentration, temperature) and sorbent chemical (functional groups, metal content, stability) and physical (particle size, specific surface area) characteristics [39,40]. Phosphate adsorption to hydrous Al-oxide was shown to be favored at low pH values [41]. Solution pH may influence the form of bioavailable P as (i) H_3PO_4 below pH 2.12, (ii) $H_2PO_4^-$ between pH 2.12 and 7.21, (iii) HPO_4^{2-} between pH 7.21 and 12.44, and (iv) PO_4^{3-} above pH 12.44 [2]. Adsorption isotherms may help explain interactions between sorbent and adsorbate. Common isotherms include the Langmuir and Freunlich equations. The Langmuir model assumes adsorption occurs at homogenous sites in monolayer adsorption processes, whereas the Freundlich model assumes adsorption occurs at heterogenous sites. Phosphate adsorbents for metal oxides/hydroxides fitting a pseudo-first-order, a pseudo-second order, and an Elovich kinetic model include sepiolite/Al oxide hydroxide, ferric sludge/iron activated carbon fiber, and Mn-Al oxide/Al- and Fe-montmorillonite, respectively [40]. Solution phosphate may also adsorb onto Ca-minerals such as calcite. Electron probe micro-analysis revealed that coral-like growths of Ca-phosphates (predominantly dicalcium phosphate with octa-calcium phosphate present) accumulate on calcite surfaces with exchangeability of reacted P ranging from 30% to 100% depending on amount of P present [42].

Mineralization and immobilization are some of the main reactions involved with microbes turning over residue for nutrient cycling. Immobilization occurs when inorganic compounds become incorporated into organic forms. Phosphatase catalyzes mineralization of organic to inorganic P forms [2]:

$$(R - PO_4) + H_2O \xrightarrow{Phosphatase} HPO_4^{2-} + (R - OH)$$

Soil P can be taken up by roots, leached into waterways, adsorbed onto mineral surfaces, or immobilized into organic forms. Soil P reaches the roots by mass flow (less often) and diffusion (more often) [2]. Soil P can be resupplied by fertilizer application, organic amendment application, residue decomposition, or primary mineral weathering [2].

Surface adsorption and precipitation reactions are jointly referred to as P fixation or P retention reactions. Solution P may adsorb onto clay and mineral surfaces and then desorb back into solution.

Solution P can also precipitate onto secondary P minerals which can then dissolve back into solution P. Phosphate may precipitate with Al or Fe minerals at low pH and Ca or magnesium (Mg) minerals at high pH [2,43,44]. Ferric iron phosphates dominant at low pH may solubilize to ferrous iron through activity of reductants such as glucose with added C inputs [45]. Fixation reactions may be impacted by clay minerals, soil pH (general P availability being the greatest at pH 6.5), neighboring anion effects (competition for adsorption sites), soil organic matter (organic compounds tend to increase P availability), time (rapid initial reactions followed by slower reactions), temperature (an approximate doubling of mineralization rates with each 10 °C increase in temperature), flooding (bioavailable P increases after flooding), and counter cation characteristics (divalent cations on cation exchange capacity (CEC) increase P adsorption greater than monovalent cations). There is greater adsorption activity in clays saturated with Ca⁺² compared to sodium (Na⁺) because divalent cations increase availability of positively charged clay mineral edges, particularly below soil pH 6.5 [2].

Solution P immobilization to organic P and mineralization to inorganic P is driven by soil microorganisms. Mineralization and immobilization occur simultaneously, and equilibrium is determined by the carbon-to-phosphorus ratio (C:P). When C:P is less than 200:1, between 200:1–300:1, or greater than 300:1, there is net mineralization of organic P, no gain or loss of inorganic P, and net immobilization of inorganic P, respectively [2].

Phosphorus loss from crop removal typically ranges from 11.2 to 44.8 kg P/ha (10–40 lbs P/acre) [2]. Phosphorus losses from leaching by irrigation/rainfall are relatively small because while mobile within the plant, fertilizer P is generally considered immobile in soil [46,47]. The range of the diffusion constant for $H_2PO_4^-$ is 10^{-12} to 10^{-15} m²/s. The average rate of diffusion is 0.13 mm/day. Leached dissolved reactive P and erosion of P deposits from acidic waters may result in contamination or eutrophication [48], resulting in nutrient loss from the agroecosystem and possible ecological damage to aquatic ecosystems.

2.3. Phosphorus Bioavailability Across Different Soils

Phosphorus bioavailability is related to soil characteristics such as silt content, clay content, pH, extractable Al, and extractable Fe [49]. An appropriate metric for determination of P sorption is the distribution coefficient, K_d that describes the ratio of contaminant concentration (including cations, anions, radionuclides, redox-sensitive elements) to the contaminant concentration in the surrounding aqueous solution at equilibrium [50]. A greenhouse pot experiment examining biosolid-amended acidic Greek Alfisols found that high-dose application of biosolids (300 t/ha sewage sludge) resulted in increased pH from 5.19 to 6.92 and decreased K_d from 23.3 to 12.9 L/kg [51]. Relatively fertile Alfisols commonly form over calcareous till and tend to have an accumulation of clay [2,52]. This soil type tends to occupy cool and humid regions of the Northern Hemisphere [53]. Alfisols from Southeast Sudan were found to have low available and organic P content with 40% inorganic P forms in the Fe-Al and Ca-P fractions [54].

Andisols develop from volcanic ash, and commonly occur in the Pacific Northwest in the United States [2,53]. In the native volcanic ash-derived Andisol soils of the Mexican highlands, it was found that over 19% of ³²P was recovered in bicarbonate organic P and organic sodium hydroxide forms which indicates that organic P cycling and ecological-based management systems are crucial to maintain the supplies of soil organic P and sustain agricultural production [55]. Soils with volcanic origin tend to have strong P fixation with dominant active Al in amorphous clay minerals. For instance, New Zealand andic horizons have high P retention with typically 85% added P becoming sorbed to soil colloids [52]. The Fe-rich Andisols of Hawaii tend to have predominant fixation reactions that may necessitate fertilizer of greater than 1000 k/ha [52].

Aridisol soils tend to form in arid environments such as the western United States with predominant calcium carbonate (CaCO₃) compounds [2,53]. Aridisol samples gathered from the North Kordofan state of Sudan were found to have low P content, both available and organic. Most of the inorganic forms were present in the Fe-Al fraction (>50%) as opposed to the Ca-P fraction (20%), so available P

was positively related to the Al+Fe-P content of total-P, but negatively correlated with Ca-P content of total-P [54]. Similar results were found for Vertisol samples collected from the Gezira state in Sudan in which there was low organic and available P content. However, the Vertisol samples had the most inorganic P present in the Ca-P fraction (>60%) and less in the Fe-Al fraction (40%) and, therefore, had a more alkaline pH (>8.0) [54]. Vertisols tends to be rich in clay content, which tends to result in shrinking and swelling of the soil with changes in moisture content [2].

Entisols tend to be sandy with no developed horizons and occur in places such as the African Sahara [2,53]. This soil was studied using sequential fractionation and researchers concluded that fertilization greatly impacted P availability where fertilization favored mineralization and redistributed organic and inorganic P fractions, between labile and non-labile pools and, ultimately, resulted in an accumulation of inorganic soil P [56]. While high inorganic P levels in the labile compartment may contribute to increased bioavailable pools, it also represents a risk of P-contamination to waterbodies. Esmail et al. [57] also found high inorganic P content, particularly in Ca-P fractions. These studies investigated Entisols of the Kurdistan region of Iraq and noted predominate fixation reactions from the high CaCO₃ content of the regional soils. Esmail et al. [57] also observed high CaCO₃ content in Inceptisols and Mollisols in the same region. Inceptisols are characterized by minimal horizon development commonly occurring in Mediterranean soils, while Mollisols are characterized by a prominent dark horizon with high soil organic matter content common to grasslands and the Great Plains of the United States [2,53]. Fertilization rates of 320 kg/ha P₂O₅ were found to yield both the greatest dry matter accumulation and available P concentration [57]. This study concluded by determining there to be a low P bioavailbility in Inceptisol samples and high P bioavailability in Mollisol samples, which is likely due to the high organic matter content in the Mollisol soil.

Gelisols from permafrost and Histosols from wetlands have high organic matter content, often with a surface organic layer greater than 40 cm [2,52]. Phosphorus tends to be a limiting factor to growth in these soils [58]. Histosols tend to have high water and nutrient holding capacity and form organo-P complexes [52]. Antarctic Gelisols are weakly developed soils whose chemical weathering processes are likely a result of phosphate-containing guano deposits [59].

Acidic Spodosols tend to have high soil organic matter content as well as typical Al and Fe accumulation and are commonly spread across Northern Europe [2,53]. While Spodosols of the Cook Inlet region of Alaska tend to be fertile, P-deficiency may occur from the P-fixing capacity of the allophone-imogolite minerals and Fe-hydroxides [52]. New England Spodosols are characterized by high water retention and abundant poorly crystalline structures with variable charge that are responsible for substantial P fixation [52].

Oxisols and Ultisosls are highly weathered soils that tend to have significant anion exchange capacity and high P fixation potential [52,60]. Oxisols tend to have high Al and Fe content and are the predominant soil order in Brazil [2,53]. Highly weathered Brazilian Oxisol samples were shown to have spatially dependent P availability that could be predicted by Fe-oxide content and magnetic susceptibility [61]. Ultisols occur in the humid regions of the Southeastern United States and Southeastern China, as well as the tropical regions of South America and Africa [53].

2.4. Phosphorus Uptake and Integration

Phosphorus uptake is an energy-mediated process [62]. It is comprised of multiple epidermally located transport systems driven by a proton gradient generated by H⁺-ATPase across the plasma membrane [63]. The constant K_m can help describe the kinetics of phosphate transport across a steep concentration gradient. Numerically, K_m equals the concentration of solute that yields half the maximum rate of transport. Low K_m values indicate high binding attraction of the transported compound to the transport site, whereas high K_m values indicate a lower binding attraction [64]. High-affinity phosphate transporter systems have low K_m values typically between 2.5 and 12.3 μ M and low affinity systems typically have higher K_m values between 50 to 100 μ M [65].

While high-affinity transporters are inducible in conditions of low-P, low-affinity transporters are unaffected by P status. In a study with ryegrass (*Lolium perenne* L.), the functional characterization and expression of two members of the PHT1 family were analyzed and it was determined that *LpPHT1;4* was a high-affinity transporter influenced by P status and *LpPHT1;1* was a low-affinity transporter not correlated with P status [66].

High-affinity transporters may be suppressed because overexpression of these transporters may lead to an over-accumulation of P [66]. The *PHO*-4 gene is highly stimulated by the addition of Na⁺, suggesting that *PHO*-4 is a Na⁺/P symporter of fungal origin [67]. Members of the PHT1 family utilize symport (co-transport) as means to move ions across the steep electrochemical gradient caused by coupling ATP hydrolysis with proton transport, and stoichiometry of a proton-phosphate symport system is typically 2 to 4 H⁺/H₂PO₄⁻ [68,69].

Table 1 outlines the affinity and location of five phosphate transporter families. Phosphate transporter 1 family (PHT1) is involved in the initial uptake and remobilization of P [70]. Phosphate transporter 2 family (PHT2) impacts P-allocation through moderating P-starvation response genes and the translocation of P within leaves [71]. Phosphate family 3 (PHT3) includes an H⁺ symport and an OH⁻ antiport [72]. Phosphate family 4 (PHT4) has been shown to mediate P transport in yeast (*Saccharomyces cerevisiae*) with a great specificity [73]. Phosphate transporter 5 family (PHT5) functions as vacuolar-P transporters to regulate cytoplasmic homeostasis [74].

Family	Phosphate Transporter Family 1 (PHT1)	Phosphate Transporter Family 2 (PHT2)	Phosphate Transporter Family 3 (PHT3)	Phosphate Transporter Family 4 (PHT4)	Phosphate Transporter Family 5 (PHT5)
Affinity Symport agent	High and low H ⁺	High and low H ⁺	High and low H ⁺	High and low H ⁺ , Na ⁺	High and low H ⁺
Location	Plasma membrane	Inner plastid membrane of chloroplast	Mitochondrial membrane	Golgi apparatus	Vacuole
Representative homo-paralog	LePT1 to LePT5 (from tomato)	Pht2;1 (from Arabidopsis green tissue)	PtrPHT3.1a (from poplar (<i>Populus</i> trichocarpa L.))	Pht4;6-1 (from yeast)	OsSPX-MFS1 (from rice)
Notes	Acquires P in both high- and low-P soils	Moderates P translocation	Regulates P distribution	Regulates P transport between cvtosol	Also named vacuolar phosphate transporter
Sources	[75–78]	[71,75,79]	[72,75]	[73,75,80]	[74,81,82]

 Table 1. Notes on the major phosphate transporter families, PHT1, PHT2, PHT3, PHT4, and PHT5.

Tomato high-affinity phosphate transporters *LePT1* and *LePT2* consist of 12 membrane-spanning regions and have a high degree of sequence identity to other high-affinity phosphate transporters. The percentage of amino acid identity between *LePT1/LePT2* and *AtPT1* (from Arabidopsis (*Arabidopsis thaliana* L.)), *AtPT2* (from Arabidopsis), *STPT1* (from potato), *STPT2* (from potato), and *PIT1* (from *Catharanthus roseus* L.) is 78.4/77.2, 82.9/75.9, 95.5/80.0, 78.0/95.0, and 86.4/76.8, respectively [83]. *LePT1* and *LePT2* are induced by low-P conditions [84]. Increased expression of these transporters was detected when plants were supplied with 100 μ M P or less [83]. *LePT1* and *LePT2* are expressed in roots and *LePT1* is also minimally expressed in leaves, stems, and petioles of P-starved (0 μ M P) tomato plants. High transcript activities of both transporters were observed in root epidermal cells and lower levels of *LePT1* were observed in the central cylinder and *LePT1* message accumulation was detected in leaf palisade parenchyma and phloem cells [83].

Yeast has been an important model organism for identifying high-affinity phosphate transporter genes [75]. PHO84 represents a yeast high-affinity phosphate transporter homologous to the PHT1 family of P transporters [76]. As reviewed by Nussaume et al. [65], PHO84 identification allowed for further identification of homologous transporters in other species such as *Neurospora crassa*, mycorrhizal

fungus *Glomus versiforme*, and Arabidopsis. Homologs have also been identified in potato (*Solanum tuberosum* L.) [85].

Transporters from the Pht1 family have been found in barley (*Hordeum vulgare* L.). The HvPHT1;1 barley transporter is induced by P-deficiency and initiates trichoblast expression in the root epidermis [86]. HvPHT1;2 expression in the root hair zone is induced in P-starved conditions [87]. HvPHT1;6 represents a low-affinity P transporter expressed in barley root and shoot to remobilize P. Upregulation of HvPHT1;6 has been associated with greater PUtE [88]. Arabidopsis P transporter gene expression of *AtPT2* and tomato expression of *TPSI1* are inducible upon low-P and repressible upon sufficient-P. Increases in transcripts of *AtPT2* have occurred when external P concentration was below 50 µM. *TPSI1* is predominantly expressed in root tissue and other abiotic stresses have little to no impact on expression of this gene [89]. In wheat, *TaPT2* represents a high-affinity transporter gene that increases expression in the roots of P-deficient plants [90]. Phosphate transporters show a great degree of conservation among species, underscoring opportunities for identification and selection.

P-uptake is impacted by cross talks with other nutrients such as zinc (Zn) and Fe. Phosphorus deficiency results in an over-accumulation of Zn in shoots and vice versa. The genes *PHO1* and *PHO1;H1* (for P) and *HMA2* and *HMA4* (for Zn) have been identified as a set of co-expressed genes responsible for P and Zn loading into the root xylem. There is direct correlation from co-expression analysis for *PHO1* and *HMA4* and indirect correlation for *PHO1;H1* and *HMA2*, confirming existence of crosstalks between the Zn and P regulatory networks [91,92]. Furthermore, P uptake is enhanced in Fe deficiency, and conversely, P-deficiency enhances Fe availability within plants. Gene expression of *FER1*, which encodes Fe protein (ferritin) storage responds to low-P as mediated by PHR1 and Fe excess [92].

Once taken up, P may either (i) move via symplastic pathways from the roots to xylem parenchyma cells, (ii) enter the metabolic pool, (iii) become a structural component of the cell, (iv) efflux in high-P conditions, or (v) be stored in the vacuole [62,93]. The rice phosphate transporter OsSPX-MFS3 may be responsible for P efflux from the vacuole; PHT1;5 may be responsible for phloem loading [69]. In P-sufficiency, the majority of root-absorbed P is transported through the xylem to new leaves. In P-deficiency, xylem-supplied P is restricted, and P is supplemented from old leaves. Xylem P concentrations range from 1 mM (in P starved plants) to 7 mM (in plants supplied with 125 μ M P). Phosphate concentrations in the phloem range from 0.35 to 0.55 mg/mL [64].

3. Plant Physiological Responses to Low Phosphorus Stress

The standard route for P uptake, translocation, and utilization may need to be altered to allow for growth in stressful conditions of P-deficiency. The following section illustrates the processes that occur in plants when grown in P-starved media.

3.1. Internal Phosphorus Sensing

The first step in a P-deficiency-induced response is sensing when supplies are depleted. Results from a split root study with tomato indicated that transcript levels of tomato phosphate transporters (*LePT1* and *LePT2*) increase from a combination of depleted external P supply and internal P reserves; even when P was supplied only to a portion of the root, phosphate transporter expression did not increase in the portions exposed to P-deficient conditions [83]. Plants sense a within-plant deficiency when vacuolar supplies diminish with a decrease of shoot P supply, resulting in decreased photosynthesis, glycolysis, and respiration [94]. Phosphorus sensing occurs at the root tip, within leaf cells, and through the vascular system. Breeding to fine-tune low-P sensing mechanisms could enhance the overall P-deficiency response and nutrient use efficiency.

Shoot-derived microRNA (miR399) mediates PHO2 mRNA turnover in P-limitation, indicating that microRNAs may work as P-deficiency signaling agents [70]. PHR1 is responsible for upregulation of *AtSPX* to induce P-starvation coping mechanisms [95]. The PHR1 binding site has related sequences that are highly responsive to P-deficiency including *TPSI1* (from tomato), *Mt4* (from *Medicago truncatula* Gaertn), *OsPI1* (from rice), and *At4* and *AtIPS1* (from Arabidopsis) [96,97].

Signaling at P-sensing has been explored by Ham et al. [70]. These researchers found that upon sensing low external P, *LOW PHOSPHATE ROOT 1 (LPR1)* blocks symplastic communication within the stem-cell niche (SCN) in the root apical meristem (RAM), locally regulating growth in low-P conditions. The transcription factor, *SENSITIVE TO RHIZOTOXICITY (STOP1)*, controls ALUMINUM ACTIVATED MALATE TRANSPORTER 1 (ALMT1) expression, which secretes malate into the apoplast. The secreted malate results in reactive oxygen species (ROS) production and callose deposition in SCN, disrupting plasmodesmata mediate direct cell-to-cell communication and primary root growth. Shin et al. [98] reported an increase in ROS in the cortex of root cells in P-deficiency, a finding in agreement with Ham et al. [70]. ROS may act as a signaling molecule. Sensitivity to P-deficiency would be advantageous, and, therefore, investigations into ROS-responsive genotypes could be of interest.

3.2. Phosphorus Reprioritization

Phosphorus deficiency is typically accompanied by within-plant remobilization of P from senescent to developing tissue [71,88]. Robinson et al. [99] furthers this idea through studying delayed leaf senescence in Arabidopsis and ultimately, claimed that P-efficiency can be enhanced through improved remobilization. Within-plant P is mobile and P recycling is influenced by source and sink relationships, with sinks being young leaves and sources being older leaves [68].

Li et al. [100] investigated source to sink relationships in P-deficiency and found that a rice phosphate transporter (OsPht1;8) redistributed P from old to young leaves and from endosperm to embryo in seeds. Phosphorus-efficient *Banksia* species (Proteaceae) remobilized P resulting in a leaf-P concentration of 0.027 to 0.196 mg/g P dry matter (DM) after an initial concentration of 0.14 to 0.32 mg/g P DM [101]. Remobilization of internal P is an effective way to temporarily ensure maximum use of P reserves. This strategy may result in reduced growth rates, decreased vacuolar P content, and reduced P nucleic acid pools, indicating that it is not sustainable long-term [62]. Proteaceae members are non-mycorrhizal plants that generally have high P-efficiency partly due to their high P-remobilization efficiency [102]. Besford [103] noted P-deficiency induced nutrient remobilization in tomato. This study found mobilization and a rapid net export of P from the leaves of tomato plants transferred from a medium with 2340 ppm (2.34 kg/m³) of superphosphate (Ca(H₂PO₄) to a medium with no superphosphate. Similarly, Irshad et al. [104] tested cotton (*Gossypium* sp.) and found that P remobilization enabled efficient cultivars to establish a better rooting system when grown without added P for 30 days after an initial period of optimal P nutrition.

Intracellular acid phosphatase (IAP) is ubiquitous in vascular plants and activity is a helpful marker of P-deficiency. IAP functions to recycle P from expendable intracellular organophosphate pools. A study with tomato suspension cells revealed expression of a low-P induced IAP composed of a 1:1 ratio of 63 and 57 kDa subunits [105]. Because of the IAP-catalyzed P remobilization, tomato seedlings utilized stores of phytic acid (IAP upregulation corresponded to 20 fold reduction in intracellular free phosphate levels) and avoided morphological and biochemical symptoms of P-deficiency during the first 10 days of growth [105].

Membrane lipid composition may change in a P deficit because phospholipids could be used as P reserves [106,107]. Phosphatase drives remobilization of organic P sources by catalyzing the hydrolysis of orthophosphate-monoesters and anhydrides [94]. The PHOSPHATE2 (PHO2) and RNA isomers are principle regulators of within-plant P remobilization; microRNA399 (miR399) targets *pho2* that negatively regulates P uptake resulting in increased plant P uptake and microRNA827 (miR827) interacts with the *SPX-MSF* genes to moderate P sensing and homeostasis [108].

3.3. Cellular Phosphorus Homeostasis

The primary cellular component for P storage is the vacuole [62,109]. Excess P gets stored in organic compounds such as phytic acid in the vacuoles of leaf cells [110]. The vacuole typically stores 85–95% total P [111]. A protoplast study found that P-sufficient vacuoles contained 87–94% total cell-P [112]. Phosphate influx transporters across the tonoplast (vacuolar membrane) are active

in conditions of sufficient P, and in low P, vacuolar P efflux is active [69]. Passage of P across the tonoplast regulates cytoplasmic-P levels and buffers against fluctuations of external P and metabolic activities [74]. Glycine betaine (GB)-regulated phosphate homeostasis in tomato transformed with a choline oxidase gene (*codA* from *Arthrobacter globiformis*) resulted in a more resistant tomato to low-P stress compared to the wild type in part due to the differential expression of the 'PHO regulon' genes to maintained intracellular phosphate homeostasis [113].

Low cytoplasmic-P concentrations cause an inhibition of ATP synthesis, deactivation of RuBisCO (Ribulose-1,5-bisphosphate carboxylase/oxygenase), and accumulation of RuBP (Ribulose-1,5-bisphosphate) [114]. The metabolically active cytoplasmic P pool tends to range between 10 and 15 mM [71].

4. Strategies to Enhance PAE

There is potential to select for high PAE because many genes alter expression in a P deficit. For example, approximately 29% of Arabidopsis genes microarrayed were up- or down-regulated by a factor or two or more during a P deficit [115]. In yeast, 22 genes from a whole genome microarray were shown to be regulated by the PHO pathway [116]. There may be opportunities for screening genotypes effective in strategies that acquire sparingly available P including root morphological responses to low soil-P, exudation of root derived compounds, and microbial symbiosis (Figure 1).



Figure 1. A tomato plant efficient in phosphorus (P) acquisition. A hypothetical tomato adapted for P-acquisition efficiency (PAE) (left) can be compared to a hypothetical tomato inefficient in P acquisition (right). Root morphological strategies a tomato could employ to cope with a P deficit include adventitious rooting (A), topsoil foraging (B), or root hair growth (C) to better exploit a given volume of soil. Root proliferation in nutrient dense patches (D) exemplifies plasticity of an efficient root system to optimize a heterogeneous supply of soil P. Roots may exude enzymes such as phosphatase (E) to mineralize organic P sources. Roots may also exude organic acids such as citrate, malate, fumarate, or oxalate (F) to chelate bound-P. Reductants such as glucose (G) may reduce ferric iron to mobilize P. Exudation of hydrogen ions (H) acidify the rhizosphere to solubilize calcium phosphates. Rhizosphere acidification can also be achieved through excess cation uptake (I). Symbiosis with mycorrhizal fungi (J) allows for uptake of P transferred from sources unavailable to the root. There are no mechanisms illustrated for the P-inefficient representative. However, any listed strategy could be present in the root system of a P-inefficient tomato, but likely expressed to a lesser degree than a P-efficient representative.

4.1. Root Morphological Responses

Root morphological responses reflect changes in root architecture (spatial configuration), topology (connection and branching), or distribution (position in soil). Roots exhibit great architectural variation among species and genotypes, as demonstrated by soybean with variations in yield, P accumulation, and root morphology and architecture [117]. Lynch and Brown [118] suggested that root architectural traits can be mapped through quantitative trait locus (QTL) tagging (data unpublished) that could allow breeders to select for a more P efficient crop.

Root plasticity. Root plasticity is the ability of a root system to alter its typical structure in response to changing environmental conditions in order to acquire P at minimal metabolic costs. Because there is no single universal least-cost mode of P acquisition, plasticity is important for a root system to adapt to the changing costs of adaptive strategies when external P concentrations fluctuate [119]. An example of root plasticity is topsoil foraging; topsoil foraging occurs when plants concentrate root dispersal to the shallow soil horizons [120]. Computer models have shown that topsoil foraging root systems are more P efficient than a typical root system with an equivalent size. Topsoil foraging traits have been genetically mapped and can be tagged with QTLs in plant breeding programs [118]. Adventitious rooting (roots developed from non-root organs) tends to occur in the topsoil portion of the soil. Adventitious roots require minimal construction cost compared with other root types [121]. A study of common bean found that P-deficiency enhanced adventitious rooting in P-efficient genotypes (G2333 and G19839) but not in P-inefficient genotypes ('Carioca' accession G4017 and DOR364) [121]. These fast-growing, small roots help explore the soil. Importantly, Walk et al. [122] noted that an adventitious rooting response is beneficial only when specific respiration rates are similar to those of the basal roots to avoid excess carbohydrate reallocation.

Lynch and Brown [118] concluded that the highly plastic roots directly and positively impacted the P-efficiency of common bean. These plasticity responses included increased presence of adventitious roots and a gravitropic trajectory of 75° – 90° . Low levels of ethylene (0.0005–0.0010 µL/L) have been shown to restore a full gravitropic response in tomato [123].

Root plasticity is advantageous because many soils have an unevenly distributed P supply. Therefore, soils such as Oxisols [61] may enhance P bioavailability with plastic root systems (Table 2). Researchers found that biomass and P content were greater for P-efficient wheat and white lupin (*Lupinus albus* L.) grown in a heterogeneous and localized P supply compared with a uniform P supply [124]. Jing et al. [125] concluded that localized nutrition led to local root proliferation in maize. Plants with high PAE express burgeoning root growth upon sensing nutrient-dense volumes to optimize nutrient influx. Kumar et al. [126] described the advantage of plasticity between root traits as minimizing trade-offs between the costs of maintaining root functional traits and increased nutrient acquisition.

Soil Order	Characteristic Impacting Phosphorus Bioavailability	Potential Strategy to Cope with Phosphorus Availability	Relevant Source
Alfisol	Common fixation to Ca-phosphates	Microbial symbiosis	[54]
Andisol	High P sorption Al-compounds	Dense root hair growth on lateral roots	[52]
Aridisol	High calcium carbonate concentration	Rhizosphere acidification from proton exudation	[54]
Entisol	Predominate Ca-phosphate compounds	Rhizosphere acidification from proton exudation	[57]
Gelisol	Inorganic phosphate immobilization	Phosphatase exudation	[58]
Histisol	Inorganic phosphate immobilization	Phosphatase exudation	[52]
Inceptisol	High calcium carbonate concentration	Rhizosphere acidification from proton exudation	[57]
Mollisol	Inorganic phosphate immobilization	Microbial symbiosis	[57]
Oxisol	Spatially dependent high Fe-oxide content	Root plasticity with fine root proliferation	[61]
Spodosol	High Al and Fe content with high P fixation	Orgnic acid exudation	[52]
Ultisol	Common P retention with Fe-oxides	Use of native bacteria as biofertilizer to release sorbed P	[60]
Vertisol	Predominant Ca-phosphate compounds	Rhizosphere acidification from proton exudation	[54]

Table 2. Strategies to enhance phosphorus acquisition across different soil orders.

Lateral root growth to enhance surface area. The lateral root response to P-deficiency varies among crops. While many crops may benefit from enhanced lateral rooting, Lynch and Brown [118] suggested that minimal lateral rooting benefited the P-efficiency of common bean. This response has also not been reported in tomato where high P levels (50 mg/kg) have been shown to increase lateral root number [127]. Tomato root morphology has instead been shown to respond to P-deficiency by increasing root surface area and decreasing total root weight and average root diameter [128]. Although increasing lateral rooting has not yet been reported in tomato, many other crops have been studied that employ this strategy.

The number and length of lateral roots was shown to be greater among higher-yielding and P-efficient accessions of Brassica oleracea L. than lower-yielding accessions [129]. Similarly, greater lateral root branching density with lower average root diameter has been shown to increase P uptake while not substantially increasing root competition in maize [130]. Lateral rooting has been shown to vary greatly in Arabidopsis. Chevalier et al. [131] observed a range of Arabidopsis responses to P-deficiency: 50% reduced length of primary and number of lateral roots, 25% were not sensitive, 16% reduced length of primary root only, and 9% reduced number of lateral roots only (of 73 accessions). Williamson et al. [120] and Reymond et al. [132] both observed an inhibition of primary root elongation and enhancement of lateral rooting in Arabidopsis when grown in low P. Zhu et al. [133] mapped the QTLs for lateral rooting in maize in P-deficiency and found that in low-fertility treatments, six QTLs (flanked by phi001/csu3, scu164a/phi055, nc003/umc36b, bn16.16/umc17, phi070/umc62, bn17.08a/phi121 on chromosomes 1, 1, 2, 3, 6, 8, respectively) were associated with lateral root length response and one QTL (flanked by umc131/nc003 on chromosome 2) was associated with lateral root number. The researchers found that three QTLs are associated with the reduction of primary root length (LPR1 on chromosome 1, LPR2 on chromosome 3, LPR3 on chromosome 4). Ultimately, lateral rooting may be conditionally advantageous to acquire P if root diameter is small. These listed P-efficient crops were able to increase the surface area of their root system, while other crops such as tomato and bean may rely on other strategies.

Root hair production. Root hairs are extensions occupying up to 90% of a root's surface and they facilitate water and nutrient acquisition [134]. Root hairs have optimal geometry for P capture. Their small radius helps to reduce carbon cost to the plant while also extending soil exploration [135]. Root hairs are fine, which helps with P acquisition. A plant growth model based on P-deficient rice revealed that increasing root fineness by 22% increased P uptake 3-fold [136]. Genetic control of root hair growth has been explored in rice by Giri et al. [137] who found that the auxin influx carrier *OsAUX1* controls primary and crown root gravitropic responses and promotes root hair growth at low-P levels (3 μ M P). Homologous genes of TRIPTYCHON (TRY) and GLABRA3 (GL3) in Arabidopsis have been identified in tomato by Tominaga-Wada et al. [138] and, respectively, named *SITRY* and *SIGL3*. These genes are dominantly expressed in developing tomato shoots and have been determined to be functionally similar to the CAPRICE (CPC) like MYB transcription factors of Arabidopsis that regulate trichome and root hair development. Furthermore, tomato root hair length was shown to increase from 0.1 mm to 0.2–0.7 mm when phosphate concentration decreased from 100 to 2 μ M [139].

Root hairs have high plasticity that enables navigation of soils exposed to environmental changes. Their presence is advantageous; a crop with more densely packed root hairs will be better adapted for an environment with sparingly available P. Bayuelo-Jimenez et al. [140] grew 242 maize accessions in low (23 kg/ha P_2O_5) and high (97 kg/ha P_2O_5) P treatments and determined that the dense root hairs on the main root and first order laterals were responsible for the enhanced P-efficiency of high yielding varieties gown on the Mexican highlands with predominant Andisol soils (Table 2). Root hairs develop from epidermal trichoblasts overlaid onto the connection of two cortical cells [134]. More numerous and smaller cortical cells form in P-limitation, leading to a greater density of trichoblasts and, therefore, root hairs [127] and this trend has been seen in maize [141]. Root hair initiation sites are actively and progressively specified as opposed to being determined during trichoblast development. The adaptability and function of root hairs substantiate claims of call-autonomous external-P sensing.

was shown to be unaffected by plant Zn status [143]. This trend may be due to a greater reliance of H⁺-ATPase for P uptake compared with Zn uptake. Plants with root hairs tend to reach maximum biomass at a lower P concentration than root-hairless mutant plants due to their high concentrations H⁺-ATPase activity [134,144]. Root hair elongation may be driven by a shoot-originating signal, possibly auxin-derived, translocated to the roots upon sensing low-P [145].

The P-inefficient root hairless barley mutant (brb) has been extensively studied for root-hair traits. This inefficiency is likely due to its smaller root surface area compared with the root hair covered wild type [88]. Wild-type barley root hair length increased in P-deficiency [146]. Similar results were found in investigations with Arabidopsis. Bates and Lynch [147] compared growth and P accumulation of wild-type Arabidopsis with root hairless mutants (*rhd6* and *rhd2*). Root hairless mutants were shown to acquire less P than plants with a root hair response. Phosphorus-efficient Arabidopsis have longer root hairs at greater density, resulting in higher rates of nutrient uptake per unit root length [148]. Arabidopsis root-hair density is stimulated by low-P and suppressed logarithmically in response to an increase in P supply [144,149]. The presence or absence of root hairs did not affect growth in nutrient-sufficient conditions [147,150].

Denser root hair patches may lead to overlapping zones of depletion among root hairs [149]. Increasing the length of root hairs could alleviate this issue, and in P-starvation (<1 mmol/m³), root-hair length can exceed 1 mm [144]. However, longer root hairs come at a metabolic cost to the plant [146]. This cost is small because of the size of root hairs, their ephemeral nature, and the comparative benefit of growing root hairs. Root hairs can extend the total zone of depletion and explore a greater volume of soil. This enhanced exploration of the soil from higher root length density is important for nutrients such as P that diffuse slowly through the soil [119]. The Arabidopsis root hairless mutant *rhd6* has been used to study hormonal effects on root hair development. Application of indole-3-acetic acid (IAA) at 30 nM suppressed mutant defects and increased root hair elongation [151]. Auxin was also able to restore root hair growth in the ethylene insensitive Arabidopsis mutant (*ein2-1*) [152]. Application of exogenous auxin may be beneficial to P acquisition because of its ability to restore root hair production.

4.2. Exudation of Root Derived Compounds

The rhizosphere is a biologically and chemically active area surrounding the root from which plants take up nutrients. Roots exude compounds into the rhizosphere including protons, hydroxide, organic anions, enzymes (such as phosphatase), sugars, vitamins, amino acids, purines, gaseous molecules (such as H_2), root border cells, and phytosiderophores [153], which help to produce bioavailable P.

Proton exudation and rhizosphere acidification. The advantage of rhizosphere acidification may be dependent on initial soil pH and buffering capacity. Phosphorus bioavailability in high-pH soils such as Entisols or Inceptisols buffered with CaCO₃ benefits from high Ca uptake and proton exudation to maintain electro-neutrality [154] (Table 2). However, in initially acidic soil, exuded protons could interact with the dominant Al-phosphates and may result in Al toxicity in plants [155]. Studies have shown that decreasing soil-pH resulted in increased exchangeable Al and decreased exchangeable Ca [156]. Calcium phosphates (including Ca₃(PO₄)₂; CaHPO₄⁻•2H₂O) are dominant in the soil with increasing pH [157], but have negligible precipitation in solutions of pH 4.4 [158]. Therefore, in non-acidic soils dominated by Ca-phosphates, proton secretion is a significant process that enhances P-bioavailability.

H⁺-ATPase is the driving force behind rhizosphere acidification. In the plasma membrane, this enzyme couples ATP hydrolysis with proton transport, establishing electrochemical gradients across the plasma membrane [63,159]. Upregulation of plasma membrane H⁺-ATPase is largely beneficial to P uptake. A study with soybean (*Glycine max* L.) showed that plants treated with fusicoccin (plasma

membrane H⁺-ATPase activator) increased P uptake by 35%, but plants treated with vanadate (plasma membrane H⁺-ATPase inhibitor) suppressed P uptake [160].

Hydrogen ion efflux is genetically controlled. In tomato, the gene TFT7 (a member of the 14-3-3 gene family) was found to activate H⁺-ATPase and subsequent H⁺ release in low-P conditions (2 μ M P). After one day, TFT7 expression increased 2.5 times compared with P-sufficient plants [161]. The activity of 14-3-3 proteins and auxin regulatory pathways were shown to modulate H⁺ efflux by affecting the AHA2 or AHA7 genes in Arabidopsis; AHA7 moderates H⁺ exudation in the root hair zone and AHA2 regulates primary root elongation and mediates H⁺ exudation in the root elongation zone [162]. Wild-type tomato was shown to exude H⁺ at approximately $0.3 \,\mu$ M/hr/10g FW at high P (200 μ M P) and approximately 0.8 µM/hr/10g FW at low P (10 µM P) [84]. This study also found that transgenic tomato treated with General Regulatory Factor 9 (GRF9) (an expression vector with Arabidopsis 14-3-3 protein cDNA) exuded more H⁺ (approximately 1.6 μ M/hr/10g FW at low P), and its high higher H⁺-ATPase activity helped these transgenic lines accumulate more shoot-P. Hydrogen-ion exudation as a coping mechanism for a P-deficit has been observed in numerous species such as bean [163], white lupin [164], and tea (Camellia sinensis L.) [165]. The tea genotypes, TRI 2023, TRI 2025 and S 106 accumulated approximately 3.3, 2.6 and 1.1 mg P/plant and decreased rhizosphere pH by 0.30, 0.19 and 0.17 units, respectively. Compared with P-sufficient white lupin, P-deficient samples increased (i) ATPase activity, (ii) plasma membrane H^+ -ATPase concentration, (iii) H^+ pumping activity, and (iv) H⁺ plasma membrane permeability.

Nitrogen exists in the soil primarily as nitrate (NO_3^-) and ammonium (NH_4^+) [155]. Rhizosphere pH increases during NO_3^- nutrition and decreases during NH_4^+ nutrition [2]. In a study with maize, Jing et al. [125] found that localized application of P with ammonium decreased rhizosphere pH by 3 units and subsequently increased leaf expansion by 20–50%, root length 23–30%, and plant growth rate 18–77%. Understanding how different N sources impact P-bioavailability can help manage P-deficient crops. Other management strategies to enhance plasma membrane H⁺-ATPase may include attracting earthworms (*Eisenia foetida*) or applying humic acid to the soil. Earthworms produce humic substances, and isolated humic acid from earthworm compost has been shown to enhance root growth of maize and stimulate expression of plasma membrane H⁺-ATPase [166]. Additionally, auxin may impact H⁺-ATPase activity. Application of IAA to endogenous auxin-depleted Arabidopsis increased activity of the H⁺-ATPase enzyme through phosphorylation of threonine [167].

Organic acid exudation. Organic acids/anions (OAs) (including malic, fumaric, oxalic, oxalo-acetic, succinic, α -cetoglutaric, isocitric, citric, aconitic, formic, piscidic, shikimic) are important metabolites with typically one or more carboxylic group that dissociates in the cytosol of root cells. The high exudation of carboxylates from ephemeral cluster roots is largely why researchers consider members of Proteaceae to be highly P-use efficient [102]. OAs complex with metal cations and displace anions like phosphate from the soil matrix [168,169]. OA exudation significantly helps plants cope with Al toxicity and P-deficiency in acid conditions, which could benefit soils such as Spodosols (Table 2). Exudation is regulated by membrane-localized transporters including Aluminum Activated Malate Transporter (ALMT), Multidrug and Toxic Compound Extrusion (MATE), and plasma membrane H⁺-ATPase [170]. In P-deficient proteoid roots of white lupin, there was an increase of ATPase activity, V_{max} and K_m, H⁺-ATPase concentration in the plasma membrane, H⁺ pumping activity, pH gradient, and passive H⁺ permeability in the plasma membrane [164]. Lupin and other members of the *Lupinus* genus exude high concentrations of carboxylates, which accounts for their great efficiency in P-use and may also account for their role as an aggressive invasive species [171]. Exudation of OAs may also increase P bioavailability by enhancing soil microbial activity. The C compounds in OAs stimulate microflora which may help plants acquire P through symbiosis [155].

The dominant OA exuded may be species-specific. Citric acid has been reported to be dominantly exuded in tomato. There has been an accumulation of citric acid observed in P-deficient roots of tomato and an increase in PEP carboxylase needed for citrate biosynthesis [172]. Luo et al. [173] found that citric acid exuded by tomato roots increased phosphate solubility in the rhizosphere. Tomato roots

were shown to predominantly exude fumarate, citrate, and succinate at 0 M P and predominantly exude succinate and citrate at 0.5 M P [174].

Citric and malic acid may be the most frequently involved OA in responding to low-P conditions [155]. Citric and malic acid exuded from sweet potato (*Ipomoea batatas* L.) benefited root growth in low-P [169]. Citric and oxalic acid exudation increased in mungo bean (*Vigna mungo* L.) [168] and maize [169] in response to low-P stress. Similarly, P-sorption decreased with addition of citric acid [175]. However, citric acid exudation may not be beneficial in soils with high a concentration of adsorbed Ca and 2:1 clay minerals. Citric acid addition at 10 µmol/kg decreased P availability in chromic Cambisol and Luvisol samples, but increased availability in Ferralsol samples [176].

Selecting to enhance organic acid exudation could prove beneficial in searching for a more P-efficient tomato as long as considerations have been made to account for soil type and OA species specificity. Managing soils to see the advantages of OA exudation on P bioavailability would include adding chelates to the soil. Chelating agents are intricate organic acids, so adding chelating agents may help solubilize bound-P supplies. Repeated addition of synthetic ethylenediamine di(o-hydroxy-phenylacetic) acid (EDDHA), a strong Fe-chelating agent, increased P uptake and growth of big bluestem (*Andropogon gerardii* Vitman) in both laboratory conditions and in a prairie soil water slurry [177]. Furthermore, fertilization practices may impact OA effects. Fertilizing maize with monoammonium phosphate (MAP) coated with peat humic organic acid increased agronomic efficiency 11% and apparent P recovery 41% compared to conventional MAP fertilization [178]. The greater P content may be a result of the slower release from coating the fertilizer and from the solubilization effects from the organic acid coating. It was also found that fertilization with greater rates of NO₃⁻ compared to NH₄⁺ resulted in greater exudation of citrate, malate, and fumarate in tomato plants [179].

Phosphatase exudation. Phosphatase enzymes enhance soil-P bioavailability through catalyzing hydrolysis of phytic acid (inositol hexaphosphate (IP6)) thereby converting stores on unavailable organic P to available inorganic P, which may help plants growing in soils with high organic P content such as Gelisols and Histosols (Table 2). Having the ability to mineralize organic P is invaluable; 30% to 80% of total P in an agroecosystem is in organic forms, unable to be used by the plant before mineralization [2]. Both acid and alkaline phosphatase are active over a range of orthophosphoric acid monoesters [180]. Warming soils may benefit phosphatase activity. In a study with *Erica multiflora*, increasing soil and air temperature by an average of 1 °C resulted in a 68% increase in soil acid phosphatase and 22% increase in alkaline phosphatase activity [181].

Acid phosphatase is a type of phosphatase present primarily in the mitochondria and is exuded at low pH to make organic P bioavailable [180]. Two monomeric secreted acid phosphatase (SAP) isozymes have been identified in tomato (84 kDa SAP1 and 57 kDa SAP2) that mobilize external organophosphates [105]. Chickpea roots supplied with phytate (organic P) exuded acid phosphatase to a greater degree than those supplied with KH_2PO_4 or $Ca(H_2PO_4)_2$ (inorganic P) and were able to utilize the organic P source in both hydroponic and soil culture [182]. White lupin has been extensively studied for its high P-efficiency in part because white lupin secretes acid phosphatase when there are minimal surrounding stores of bioavailable P [183]. Playsted et al. [184] concluded that the ecophysiological advantage of a rhizomatous sedge (*Caustis blakei* Kuk.) can be attributed to the high concentration of carboxylates (citrate released at 0.12 nm/g FW/s) and phosphatases (acid phosphatase released at 150 µmol/g FW/s) exuded from their dauciform roots that grew in organic soil horizons. Dauciform roots as well as cluster roots rigorously scavenge pools of local external P [119].

Reactions of alkaline phosphatase can proceed in both directions, catalyzing the synthesis and the hydrolysis of phosphate esters [185]. This enzyme may be responsible for both hydrolyzing and transporting phosphate at high pH [180,185]. Alkaline phosphatase is primarily present in the microsomes of cells [180]. While many higher plants express high acid phosphatase activity, there is significantly less activity for alkaline phosphatase [185]. However, numerous species of fungi synthesize alkaline phosphatase including *Neurospora crassa* and *Saccharomyces cerevisiae* [185]. Treatments with higher alkaline phosphatase activity were observed in soils inoculated with *Glomus etunicatum* (vesicular

arbuscular mycorrhizal (VAM) fungus) and *Enterobacter agglomerans* (bacterium) and resulted in higher concentrations of soluble P, suggesting a synergistic interaction between the microbes and alkaline phosphatase activity to increase pools of bioavailable P [186]. Fungal production of alkaline phosphatase is dependent on external P supply; medium with high inorganic P content hinders alkaline phosphatase production [185].

4.3. Microbial Symbiosis

Because microbes can produce beneficial exudates and facilitate soil P transfer, soils that favor bacterial growth (as in soils with high soil organic matter such as Alfisols and Mollisols (Table 2) may enhance phosphatase activity as opposed to soils that reduce bacterial growth such as fumigated soils). Numerous organisms including bacteria, fungi, actinomycetes, and algae are able to solubilize P through mineralization [11,187]. Highly weathered soils may benefit from microbial inoculation. In a slightly acidic savanna Ultisol, the dominant native bacteria species *Burkholderia cepacia* was shown to improve phosphate availability when used as a biofertilizer [60] (Table 2).

Plant roots colonized by arbuscular mycorrhizal (AM) fungi are capable of phosphate uptake through typical direct root epidermal uptake and through mycorrhizal uptake. Mycorrhizal P uptake in tomato is at least partially regulated by plant P status. At high P (3.5 mg P/g dry weight), the mycorrhizal uptake pathway in tomato was repressed almost completely (10% P taken up via mycorrhizal uptake pathway), whereas at low P (1.5 mg P/g dry weight), the mycorrhizal uptake pathway in tomato was dominant (75% P taken up via mycorrhizal uptake pathway) in mycorrhiza colonized conditions [188]. Three phosphate transporters that are mycorrhiza-inducible have been identified in tomato: *LePT3*, *LepPT4*, and *LePT5* (within Pht1 family). *LePT4* has a great degree of sequence identity to *StPT4* from potato, another solanaceous crop [189]. The transcripts of these transporters, *LePT3*, *LePT4*, and *LePT5*, were exclusive to arbuscule-containing cells [190]. *LePT3* and *LePT4* transcripts are reliable markers of a functional mycorrhiza uptake pathway in tomato as shown by their expression in *Glomus intraradices* colonized roots with symbiotic phosphate transfer [191].

Calcium phosphates may solubilize when interacting with OAs exuded by microorganisms [192]. When compared with symbiotic bacteria, mycorrhizal fungi are able to secrete more OAs and diffuse greater distances through their hyphae thereby further enhancing P availability [187,193]. Mycorrhizal fungi acquire P more rapidly from a lower concentration than their plant host because they possess an accumulation mechanism with a higher affinity than that of plant roots [32]. Inoculation of mycorrhizal fungi benefits P-efficiency by reducing P loss as shown by studies inoculating *Glomus mosseae* into rice paddies [194]. These associations are so helpful in P acquisition that other P adaptations become less pronounced in plants with mycorrhizal symbiosis [195].

Root exudates may stimulate microbial growth and favor an environment with symbiotic relationships. Hyphal growth of arbuscular mycorrhizal (AM) fungi is stimulated by flavonoids that can be exuded from roots [187]. Roots may secrete a branching factor (BF) to stimulates hyphae branching during AM fungi spore germination. In lotus (*Nelumbo* spp.) and sorghum, the active BF was a strigolactone [196]. Exploration of the soil from microbial root extensions scavenges the soil widely, converse to the exploitation from cluster or dauciform root exudates [119].

Symbiosis occurs at a C cost to the host (typically below 10% photosynthetically fixed C) [197]. Therefore, overly stressed plants may become more stressed with symbiosis, so ensuring that proper management techniques are in place (including the 4-Rs of nutrient stewardship and integrated pest management) would allow for an environment where breeding for enhanced microbial symbiosis and inoculation with beneficial microbes would enhance PAE.

5. Conclusions

Phosphorus is a constituent of ATP/ADP, phospholipids, and nucleic acids, enabling growth and development in plants. However, P tends to be scarcely available in agroecosystems due to processes such as precipitation, adsorption, or immobilization. Roots may alter characteristics of the rhizosphere

to facilitate P-bioavailability-enhancing reactions such as mineralization, desorption, and chelation. These strategies may include alterations to root structure such as plasticity in low-P stress responses and root spatial arrangement. Enhanced adventitious root growth, expansion of fine roots, and longer and denser root hairs benefit root P acquisition by increasing root surface area. Physiological adaptations such as increased exudation of protons, organic acids, reductants, and phosphatases help to solubilize bound-P. Enhancement of microbial symbiosis also greatly enhances P acquisition from the microbes exploiting greater volumes of soil and taking up phosphate at low concentrations. Plants that employ these strategies are capable of healthy growth in conditions of low bioavailable P. Understanding the morphological and physiological responses to low-P stress can help provide the tools necessary to understand and screen P-efficient crops and enhance agricultural sustainability. It is important to select for crops with high PAE to optimally utilize the soil-P supply. It seems plausible to select for a tomato with greater PAE because of the numerous genetically regulated responses. Management strategies to enhance P cycling may also increase pools of bioavailable P. Enhancing P acquisition may reduce the necessity for off-farm inputs and extend the longevity of indispensable phosphate rock reserves.

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