

Supplementary Material

Gold Nanoclusters Synthesized within Single-Chain Nanoparticles as Catalytic Nanoreactors in Water

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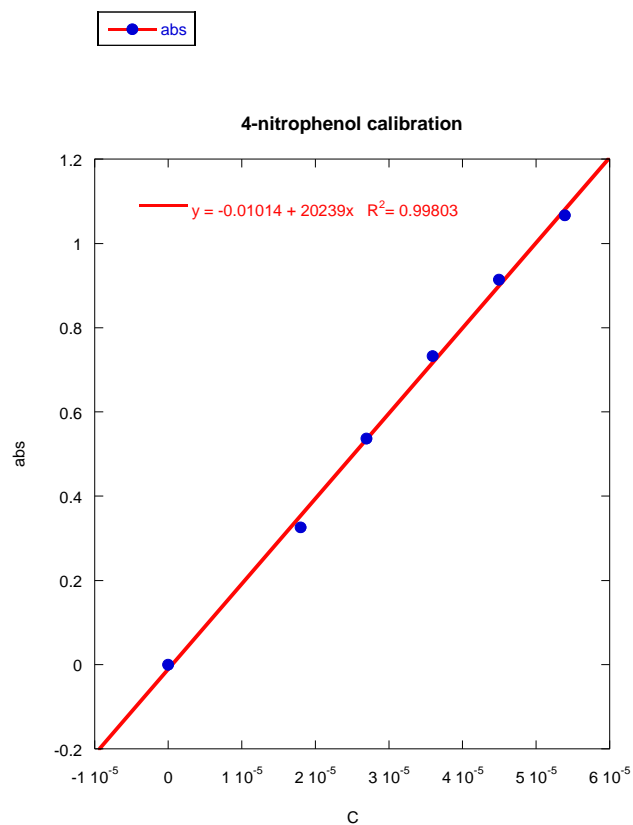


Figure S1. Calibration curve for determination of the UV-Vis molar extinction coefficient of 4-nitrophenol in water at r.t.: $\varepsilon(\lambda_{\max} = 400 \text{ nm}) \approx 20239 \text{ M}^{-1}\text{cm}^{-1}$. Literature value: $\varepsilon(\lambda_{\max} = 400 \text{ nm}) \approx 18900 \text{ M}^{-1}\text{cm}^{-1}$ (Coiffier A. *et al. J. Mater. Chem.* **2001**, 11, 2039).

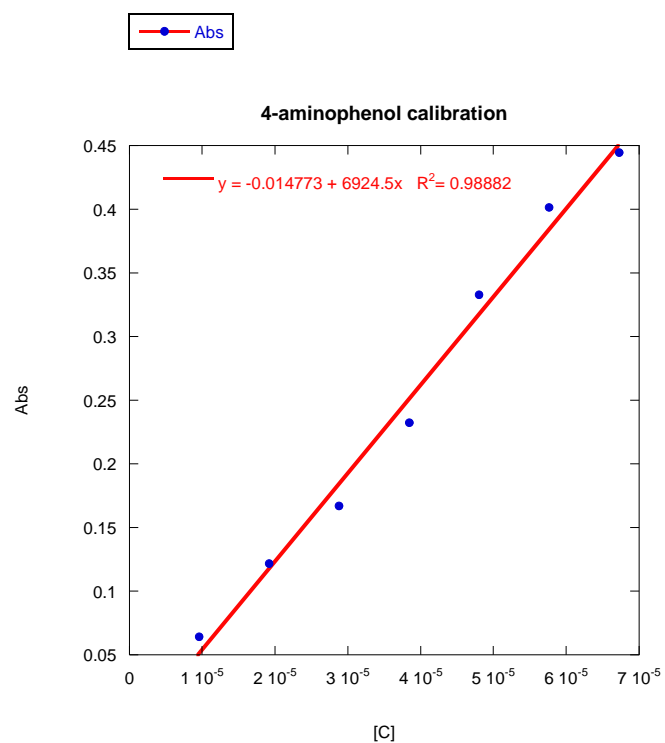


Figure S2. Calibration curve for determination of the UV-Vis molar extinction coefficient of 4-aminophenol in water at r.t.: $\varepsilon(\lambda_{\max} = 300 \text{ nm}) \approx 6925 \text{ M}^{-1}\text{cm}^{-1}$.

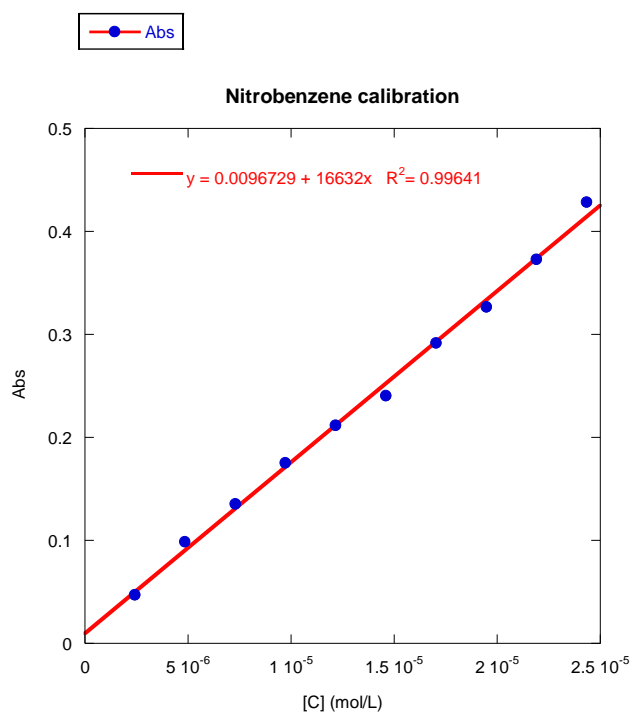


Figure S3. Calibration curve for determination of the UV-Vis molar extinction coefficient of nitrobenzene in water at r.t.: ϵ ($\lambda_{\max} = 265 \text{ nm}$) $\approx 16632 \text{ M}^{-1}\text{cm}^{-1}$.

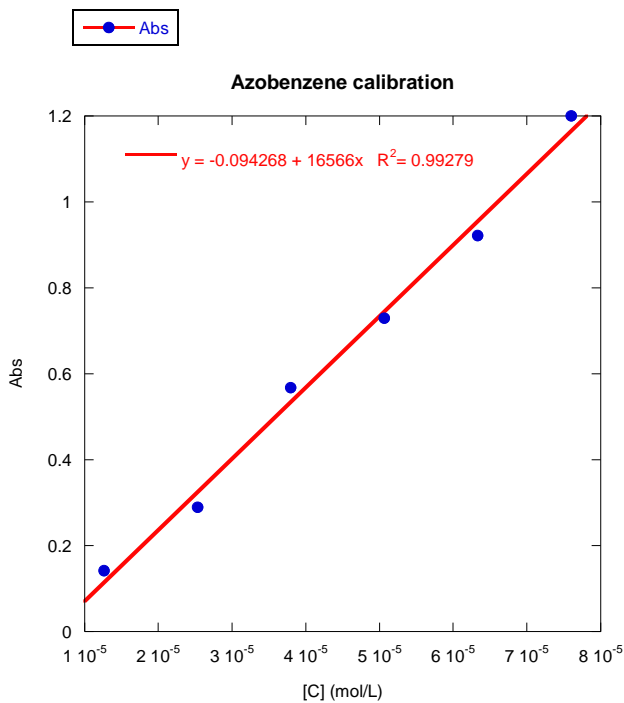


Figure S4. Calibration curve for determination of the UV-Vis molar extinction coefficient of azobenzene in water at r.t.: ϵ ($\lambda_{\max} = 230 \text{ nm}$) $\approx 16566 \text{ M}^{-1}\text{cm}^{-1}$.

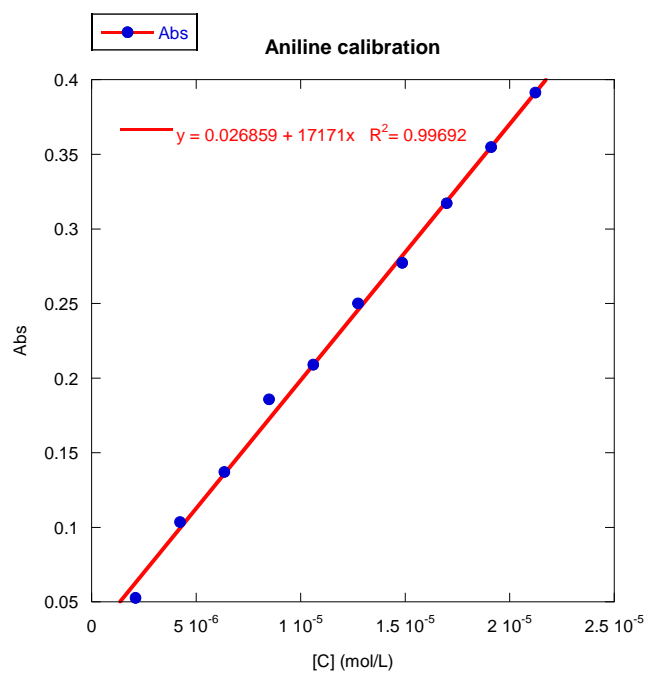


Figure S5. Calibration curve for determination of the UV-Vis molar extinction coefficient of aniline in water at r.t.: ϵ ($\lambda_{\text{max}} = 230 \text{ nm}$) $\approx 20239 \text{ M}^{-1}\text{cm}^{-1}$.

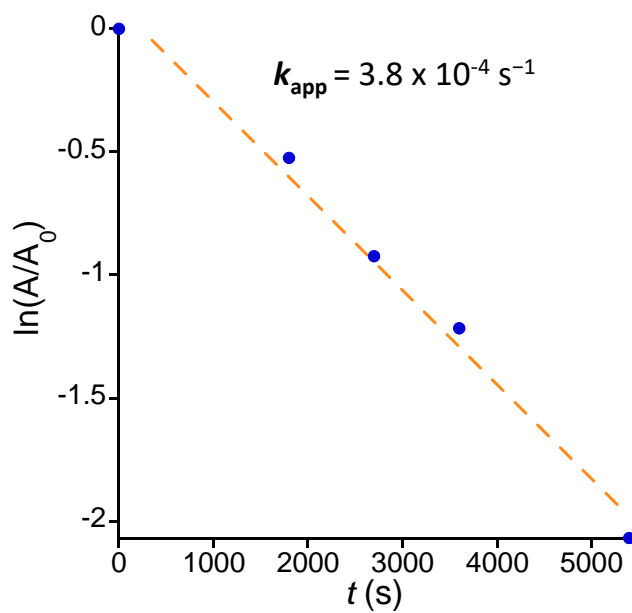


Figure S6. Apparent kinetic constant (k_{app}) of the reduction of nitrobenzene to aniline catalyzed by Au-NCs/SCNPs.

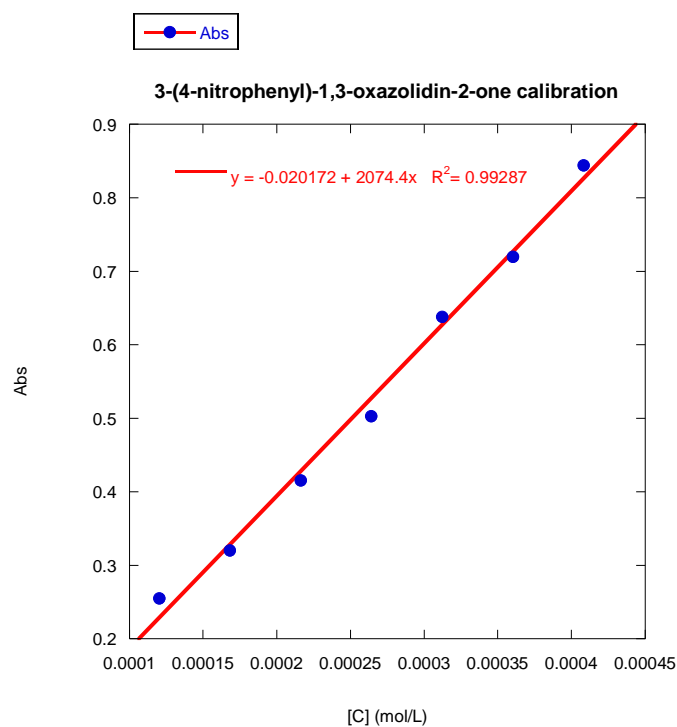


Figure S7. Calibration curve for determination of the UV-Vis molar extinction coefficient of 3-(4-nitrophenyl)-1,3-oxazolidin-2-one in water at r.t.: ϵ ($\lambda_{\max} = 320 \text{ nm}$) $\approx 2074 \text{ M}^{-1}\text{cm}^{-1}$.

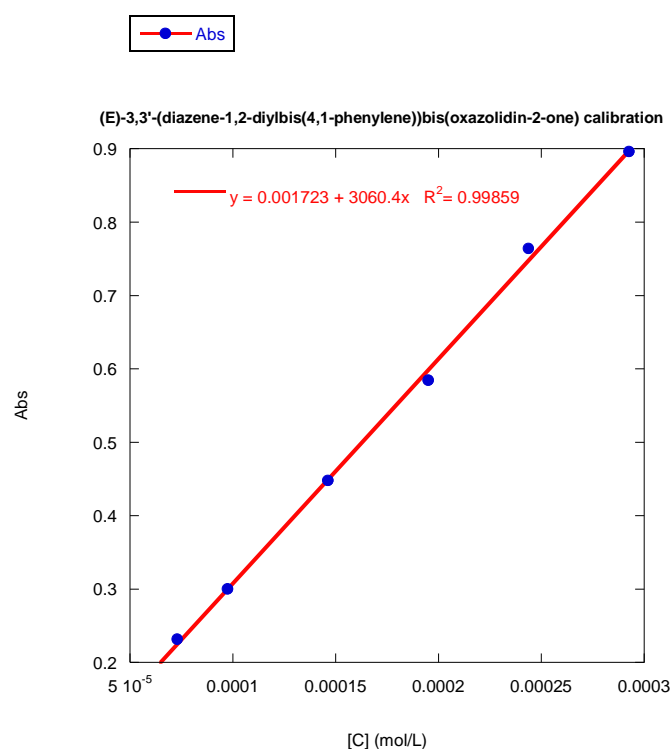


Figure S8. Calibration curve for determination of the UV-Vis molar extinction coefficient of (Z)-3,3'-(diazene-1,2-diylbis(4,1-phenylene))bis(oxazolidin-2-one) in water at r.t.: ϵ ($\lambda_{\max} = 360 \text{ nm}$) $\approx 3060 \text{ M}^{-1}\text{cm}^{-1}$.

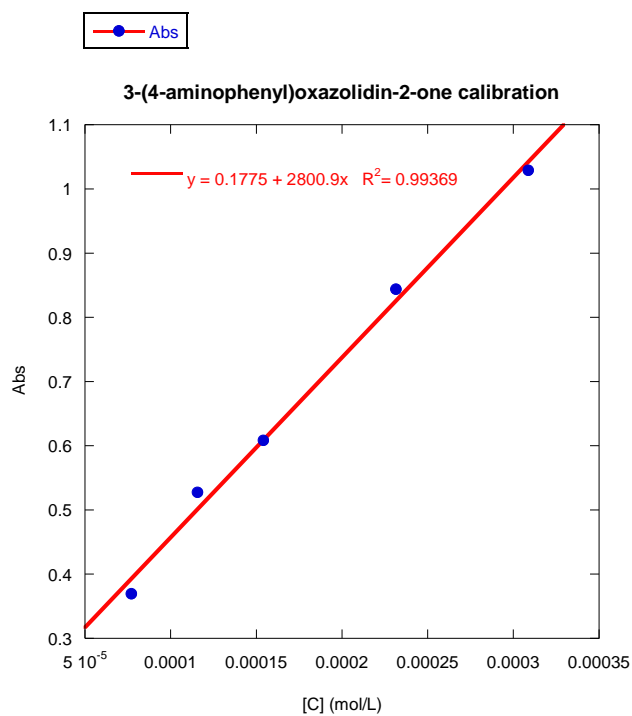


Figure S9. Calibration curve for determination of the UV-Vis molar extinction coefficient of 3-(4-aminophenyl)-1,3-oxazolidin-2-one in water at r.t.: $\epsilon(\lambda_{\text{max}} = 250 \text{ nm}) \approx 2801 \text{ M}^{-1}\text{cm}^{-1}$.

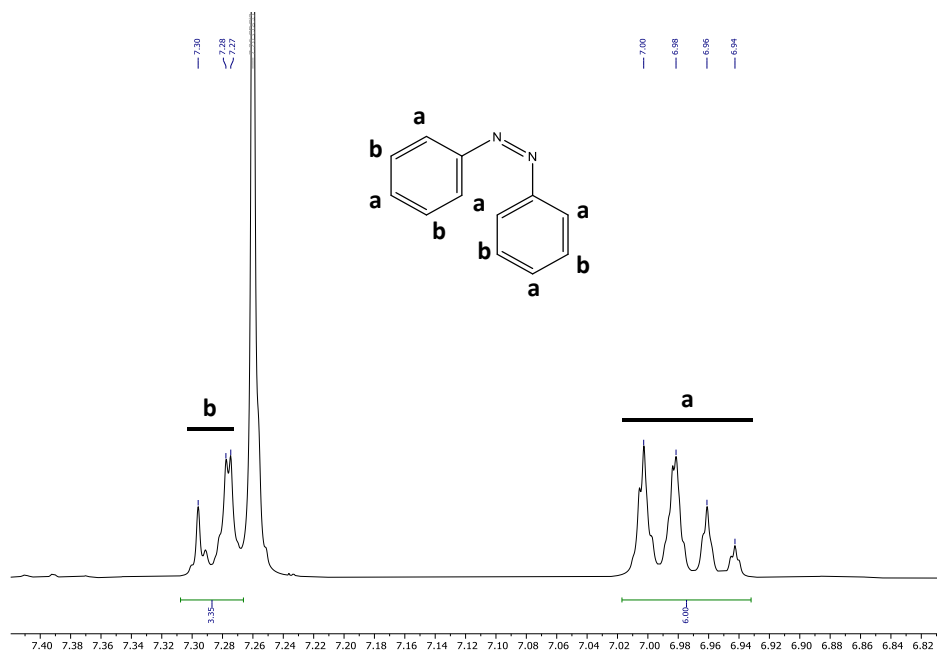


Figure S10. ^1H MNR spectrum after isolation *via* preparative TCL of the *cis*-azobenzene intermediate species.

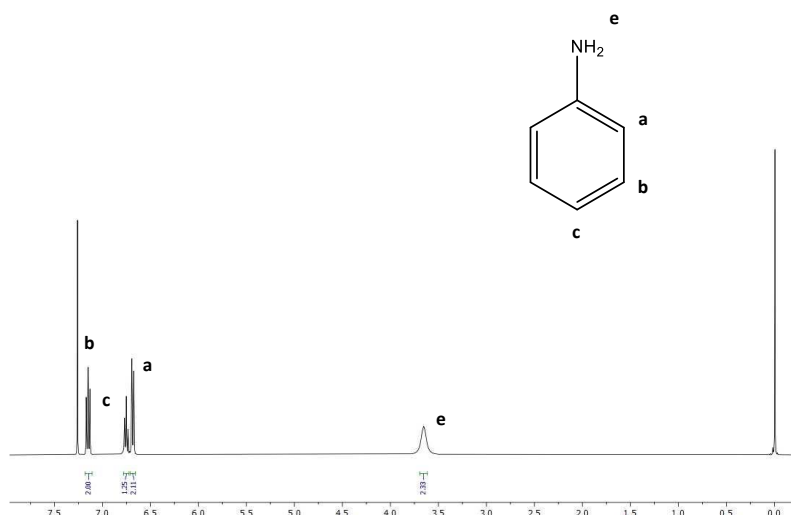


Figure S11. ¹H MNR spectrum after isolation *via* preparative TCL of the aniline product.

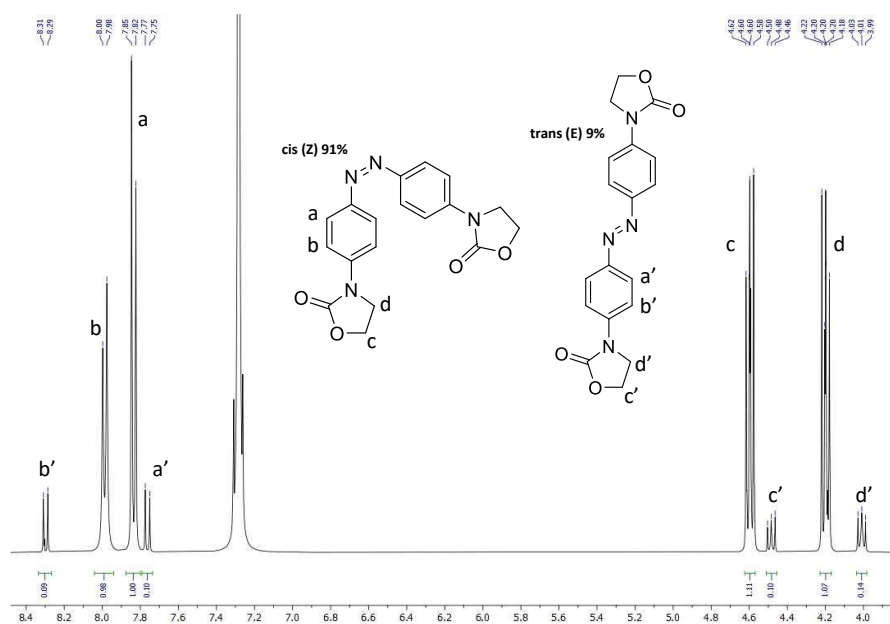


Figure S12. ¹H MNR spectrum after isolation *via* preparative TCL of the (Z)-3,3'-(diazene-1,2-diylbis(4,1-phenylene))bis(oxazolidin-2-one) intermediate species.

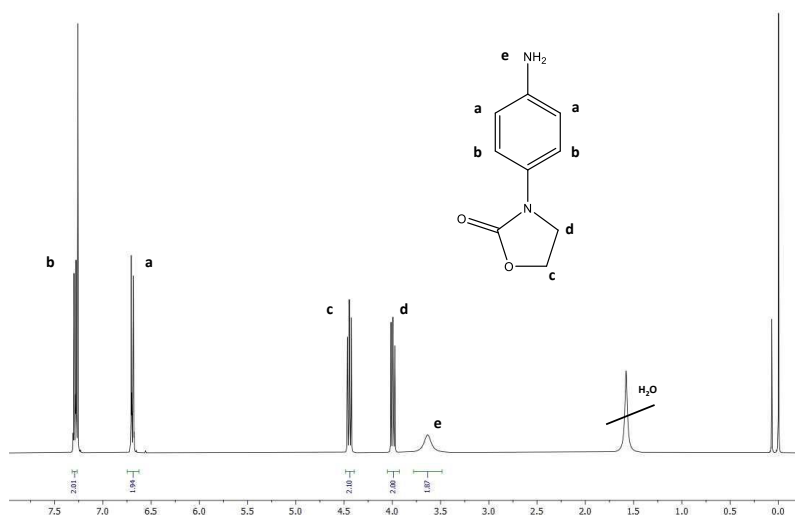


Figure S13. ^1H MNR spectrum after isolation *via* preparative TLC of the 3-(4-aminophenyl)-1,3-oxazolidin-2-one product.