

## SUPPORTING INFORMATION

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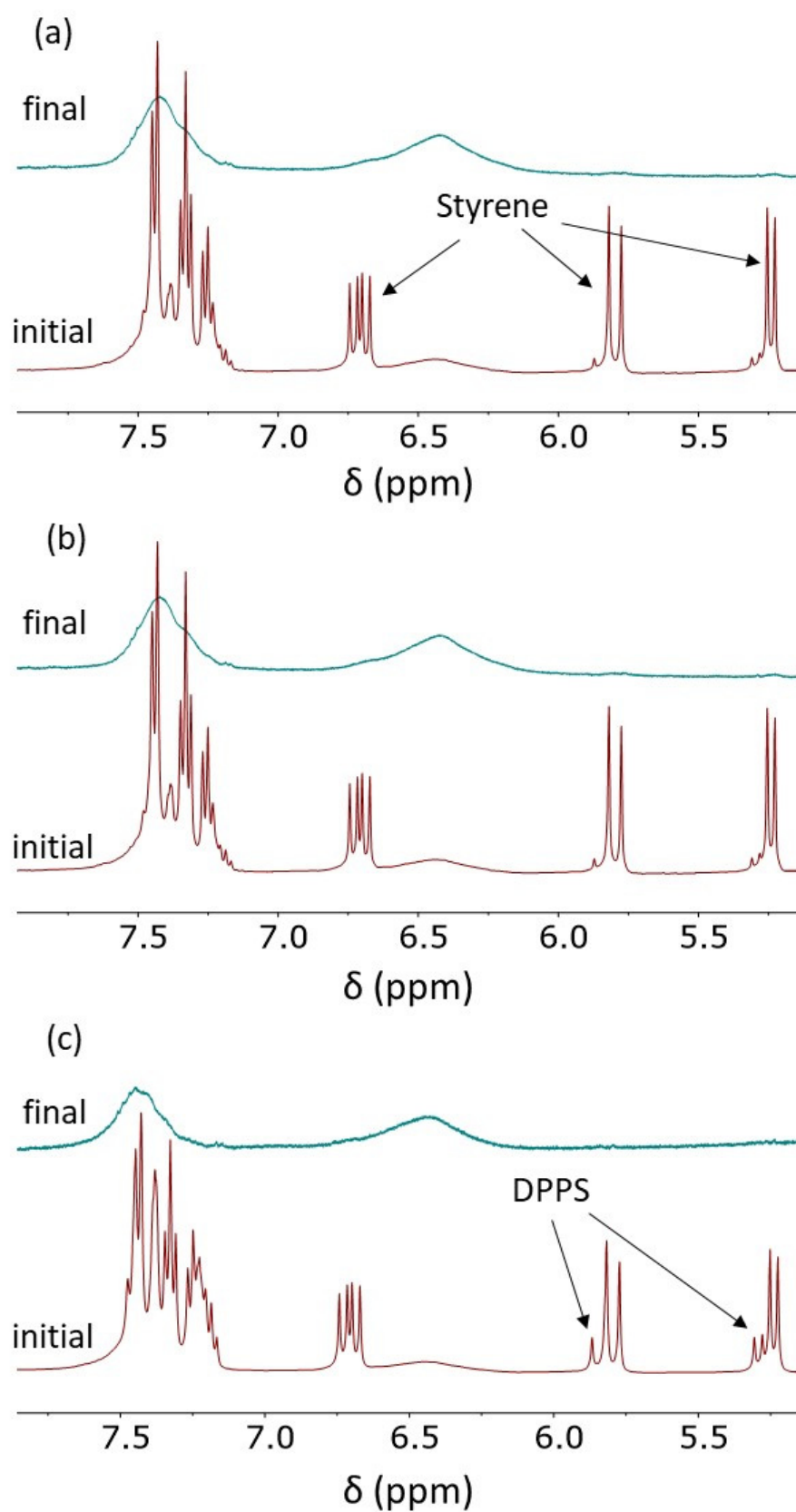
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(a) TPP-functionalized CCMs and NGs with P(SS<sup>-</sup>Na<sup>+</sup>) homopolymer blocks in the hydrophilic shell

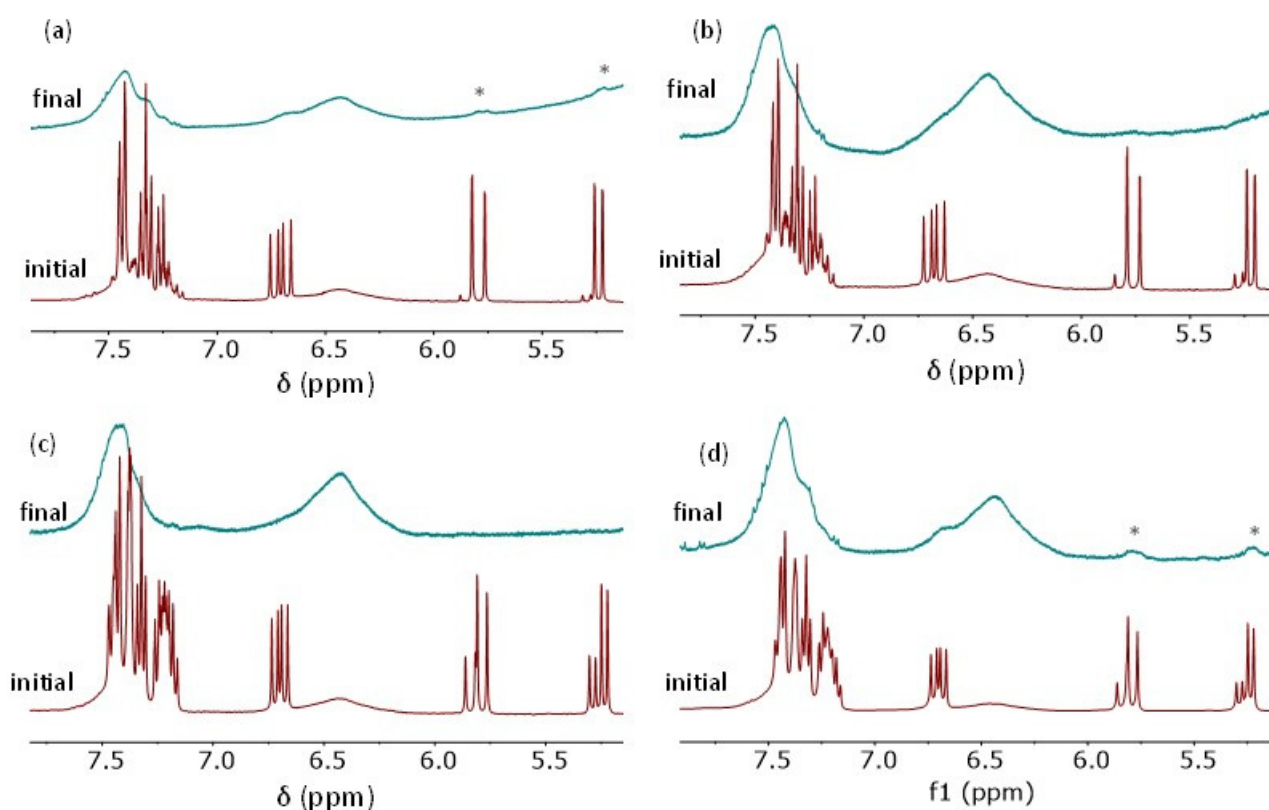
**Table S 1.** List of all polymers with a P(SS<sup>-</sup>Na<sup>+</sup>) shell synthesized in this study and reference to their characterization.

Formula <sup>a</sup>	SEC	<sup>1</sup> H NMR	<sup>31</sup> P NMR	DLS	TEM	ζ (mV)
<b>1. Diblock PSS<sup>-</sup>Na<sup>+</sup>-co-P(St-co-DPPS) micelles</b>						
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-St <sub>50</sub> -b-(St <sub>0.95</sub> -co-DPPS <sub>0.05</sub> ) <sub>300</sub> -SC(S)SnPr	/	Figure S 1		/	/	/
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-St <sub>50</sub> -b-(St <sub>0.9</sub> -co-DPPS <sub>0.1</sub> ) <sub>300</sub> -SC(S)SnPr	/	Figure S 1		/	/	/
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-St <sub>50</sub> -b-(St <sub>0.8</sub> -co-DPPS <sub>0.2</sub> ) <sub>300</sub> -SC(S)SnPr	/	Figure S 1 Figure S 4	Figure S 4	Figure S 3	Figure S 3	/
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-(St <sub>0.95</sub> -co-DPPS <sub>0.05</sub> ) <sub>300</sub> -SC(S)SnPr	/	Figure S 2		Figure S 3	Figure S 3	-53.5±2.1
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-(St <sub>0.9</sub> -co-DPPS <sub>0.1</sub> ) <sub>300</sub> -SC(S)SnPr	/	Figure S 2		Figure S 3	Figure S 3	-50.6±3.9
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-(St <sub>0.8</sub> -co-DPPS <sub>0.2</sub> ) <sub>300</sub> -SC(S)SnPr	/	Figure S 2 Figure S 4	Figure S 4	Figure S 3	Figure S 3	-49.7±7.0
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-(St <sub>0.75</sub> -co-DPPS <sub>0.25</sub> ) <sub>300</sub> -SC(S)SnPr	/	Figure S 2		/	/	
<b>2. CCMs with a PSS<sup>-</sup>Na<sup>+</sup> shell and a mixed DEGDMA-co-St core</b>						
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-St <sub>50</sub> -b-(St <sub>0.95</sub> -co-DPPS <sub>0.05</sub> ) <sub>300</sub> -b-(St <sub>0.9</sub> -co-DEGDMA <sub>0.1</sub> ) <sub>150</sub> -SC(S)SnPr	/	Figure S 5 Figure S 9	Figure S 9	Figure 3 Figure S 6	Figure 3	-50.0±1.9
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-St <sub>50</sub> -b-(St <sub>0.9</sub> -co-DPPS <sub>0.1</sub> ) <sub>300</sub> -b-(St <sub>0.9</sub> -co-DEGDMA <sub>0.1</sub> ) <sub>150</sub> -SC(S)SnPr	/	Figure S 5 Figure S 9	Figure S 9	Figure 3 Figure S 7	Figure 3	-50.5±0.6
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-St <sub>50</sub> -b-(St <sub>0.8</sub> -co-DPPS <sub>0.2</sub> ) <sub>300</sub> -b-(St <sub>0.9</sub> -co-DEGDMA <sub>0.1</sub> ) <sub>150</sub> -SC(S)SnPr	/	Figure S 5 Figure S 9	Figure S 9	Figure 3 Figure S 8	Figure 3	-51.8±1.6
<b>3. CCMs with a PSS<sup>-</sup>Na<sup>+</sup> shell and a neat DEGDMA core</b>						
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-(St <sub>0.95</sub> -co-DPPS <sub>0.05</sub> ) <sub>300</sub> -b-DEGDMA <sub>15</sub> -SC(S)SnPr	/	Figure S 10 Figure S 11	Figure S 11	Figure 4 Figure S 12	Figure 4	-51.6±1.2
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-(St <sub>0.9</sub> -co-DPPS <sub>0.1</sub> ) <sub>300</sub> -b-DEGDMA <sub>15</sub> -SC(S)SnPr	/	Figure S 10 Figure S 11	Figure S 11	Figure 4 Figure S 13	Figure 4	-54.0±2.3
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-(St <sub>0.8</sub> -co-DPPS <sub>0.2</sub> ) <sub>300</sub> -b-DEGDMA <sub>15</sub> -SC(S)SnPr	/	Figure S 10 Figure S 11	Figure S 11	Figure 4 Figure S 14	Figure 4	-69.8±2.3
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-(St <sub>0.75</sub> -co-DPPS <sub>0.25</sub> ) <sub>300</sub> -b-DEGDMA <sub>15</sub> -SC(S)SnPr	/	Figure S 10 Figure S 11	Figure S 11	Figure 5 Figure S 15	Figure 5	-49.7±6.8
<b>4. NGs with PSS<sup>-</sup>Na<sup>+</sup> shell</b>						
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-(St <sub>285</sub> -co-DPPS <sub>15</sub> -co-DEGDMA <sub>15</sub> )-SC(S)SnPr	/	Figure S 16	Figure S 16	Figure 5	Figure 5 Figure S 17	-51.2±1.2
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> ) <sub>140</sub> -b-St <sub>50</sub> -b-(St <sub>425</sub> -co-DPPS <sub>30</sub> -co-DEGDMA <sub>15</sub> )-SC(S)SnPr	/	Figure S 16	Figure S 16	Figure 5	Figure 5 Figure S 18	-52.6±2.2

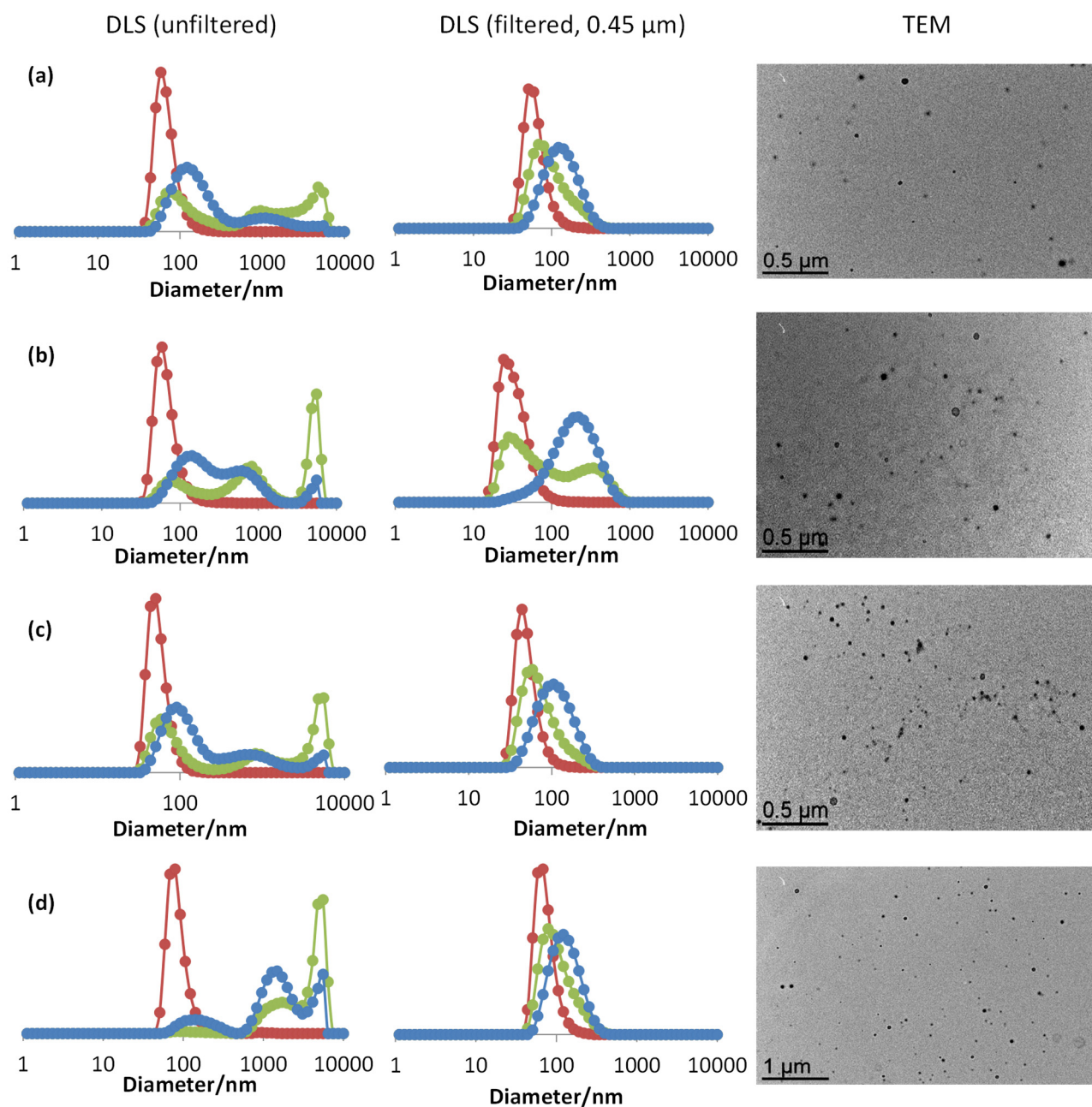




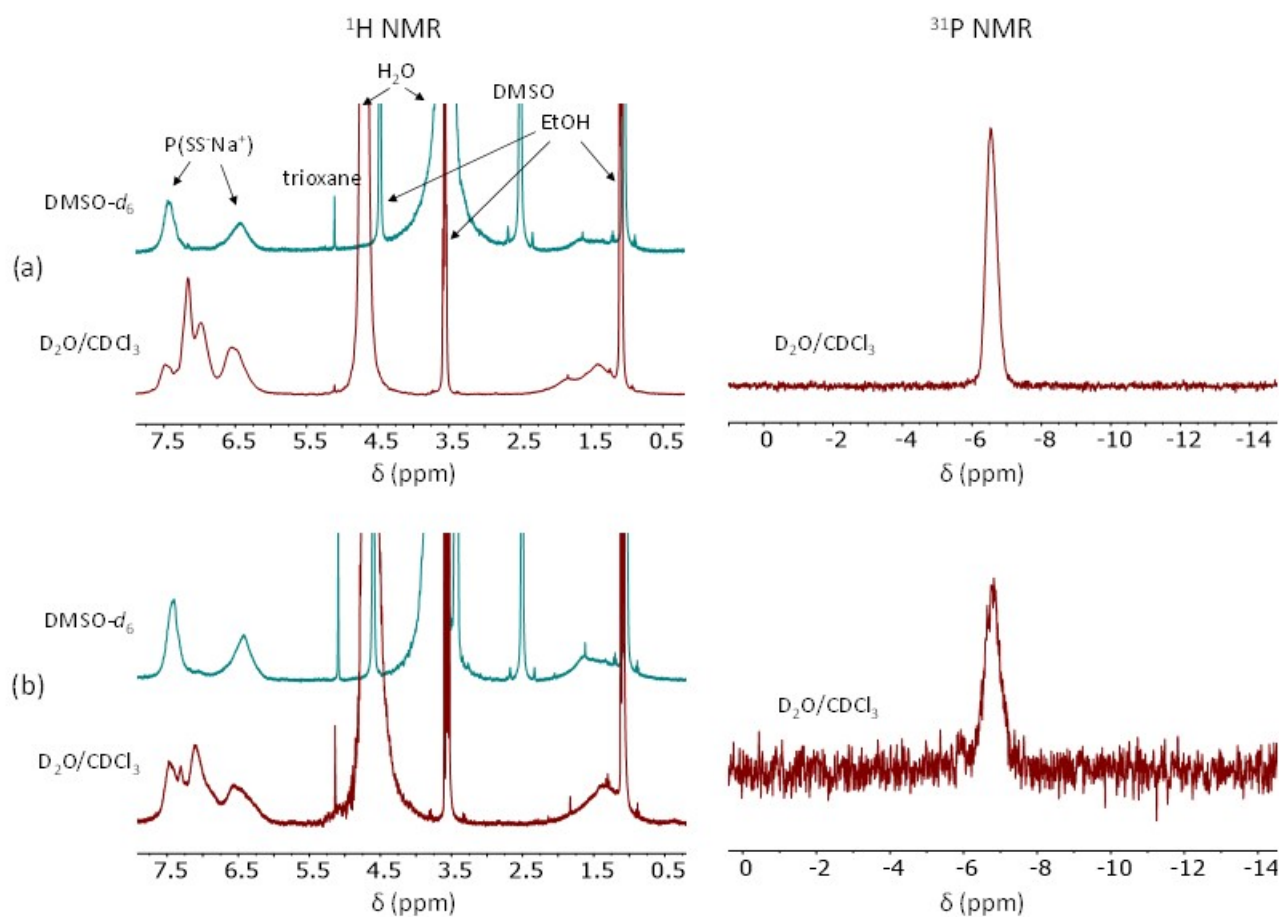
**Figure S 1.**  $^1\text{H}$  NMR monitoring of the St/DPPS copolymerization for the chain extension of  $\text{R}_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-St}_{50}\text{-SC(S)SnPr}$  to yield  $\text{R}_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-St}_{50}\text{-}b\text{-(St}_{1-y}\text{-co-DPPS}_y\text{)}_{300}\text{-SC(S)SnPr}$ . (a)  $y = 0.05$ ; (b)  $y = 0.10$ ; (c)  $y = 0.20$ . All NMR samples were prepared by adding a drop of the reaction mixture directly to the  $\text{DMSO-}d_6$  solvent in the NMR tube.



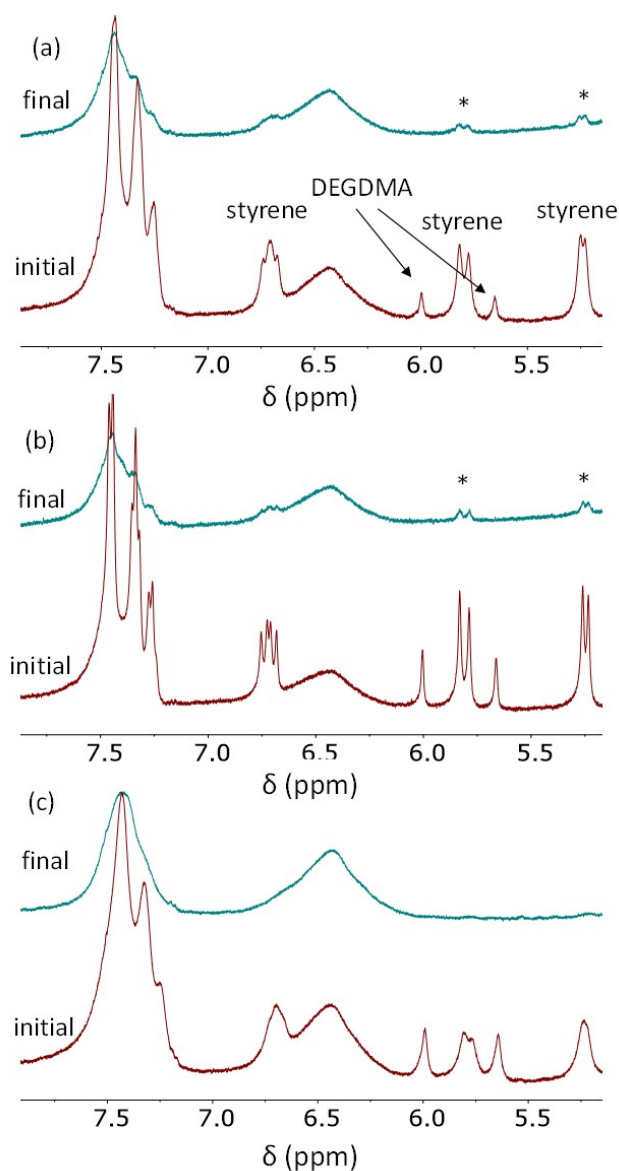
**Figure S 2.**  $^1\text{H}$  NMR monitoring of the St/DPPS copolymerization for the chain extension of  $\text{R}_0\text{-(SS}^-\text{Na}^+)_{140}\text{-SC(S)SnPr}$  to yield  $\text{R}_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{1-y}\text{-co-DPPS}_y\text{)}_{300}\text{-SC(S)SnPr}$ . (a)  $y = 0.05$ ; (b)  $y = 0.10$ ; (c)  $y = 0.20$ ; (d)  $y = 0.25$ . The starred resonances correspond to residual styrene ( $< 1\%$  by integration against the trioxane standard). All NMR samples were prepared by adding a drop of the reaction mixture directly to the  $\text{DMSO-}d_6$  solvent in the NMR tube.



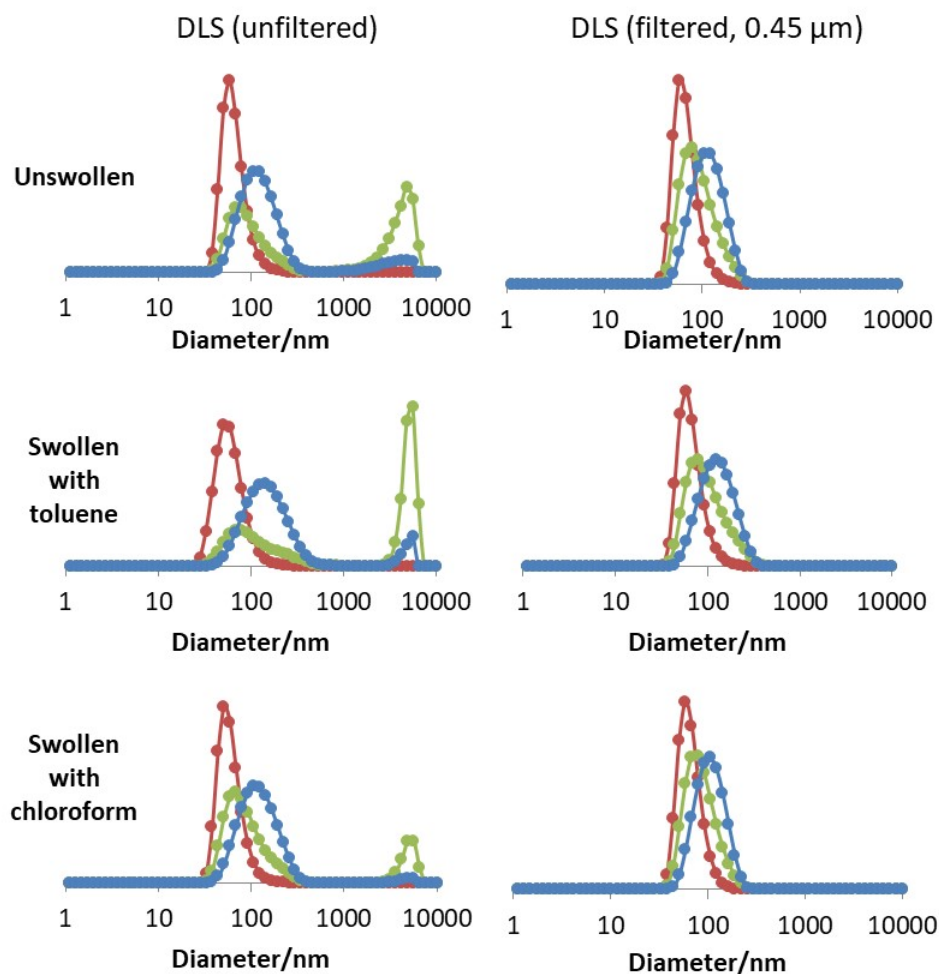
**Figure S 3.** DLS and TEM characterization of the amphiphilic di(tri)-block copolymers,  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-St}_x\text{-}b\text{-(St}_{1-y}\text{-}co\text{-DPPS}_y)_{300}\text{-SC(S)SnPr}$ . (a)  $x = 50$ ,  $y = 0.20$ ; (b)  $x = 0$ ,  $y = 0.05$ ; (c)  $x = 0$ ,  $y = 0.10$ ; (d)  $x = 0$ ,  $y = 0.20$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



**Figure S 4.**  $^1\text{H}$  and  $^{31}\text{P}$  NMR spectra of (a)  $\text{R}_0\text{-(SS}^-\text{Na}^+)_{140}\text{-b-(St}_{0.8}\text{-co-DPPS}_{0.2})_{300}\text{-SC(S)SnPr}$  and (b)  $\text{R}_0\text{-(SS}^-\text{Na}^+)_{140}\text{-b-St}_{50}\text{-b-(St}_{0.8}\text{-co-DPPS}_{0.2})_{300}\text{-SC(S)SnPr}$  latexes in  $\text{D}_2\text{O}/\text{CDCl}_3$  (3:1 v/v). The  $^1\text{H}$  NMR spectra in DMSO- $d_6$  are also shown for comparison.

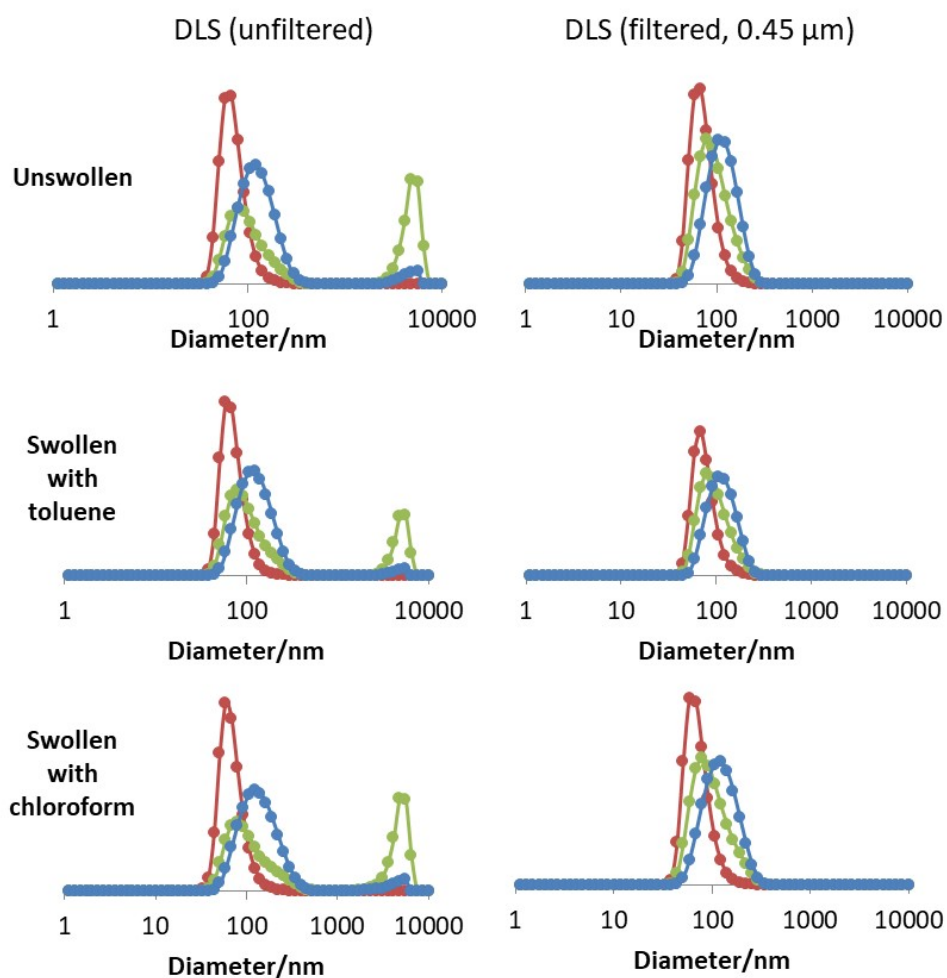


**Figure S 5.**  $^1\text{H}$  NMR monitoring of the St/DEGDMA copolymerization for the crosslinking of  $\text{R}_0\text{-(SS}^+\text{Na}^+)_{140}\text{-}b\text{-St}_{50}\text{-}b\text{-(St}_{1-y}\text{-co-DPPS}_y\text{)}_{300}\text{-SC(S)SnPr}$  to yield the CCMs  $\text{R}_0\text{-(SS}^+\text{Na}^+)_{140}\text{-}b\text{-St}_{50}\text{-}b\text{-(St}_{1-y}\text{-co-DPPS}_y\text{)}_{300}\text{-}b\text{-(St}_{0.9}\text{-co-DEGDMA}_{0.1}\text{)}_{150}\text{-SC(S)SnPr}$ . (a)  $y = 0.05$ ; (b)  $y = 0.10$ ; (c)  $y = 0.20$ . The starred resonances correspond to residual styrene ( $< 1\%$  by integration against the trioxane standard). All NMR samples were prepared by adding a drop of the reaction mixture directly to the  $\text{DMSO-}d_6$  solvent in the NMR tube.

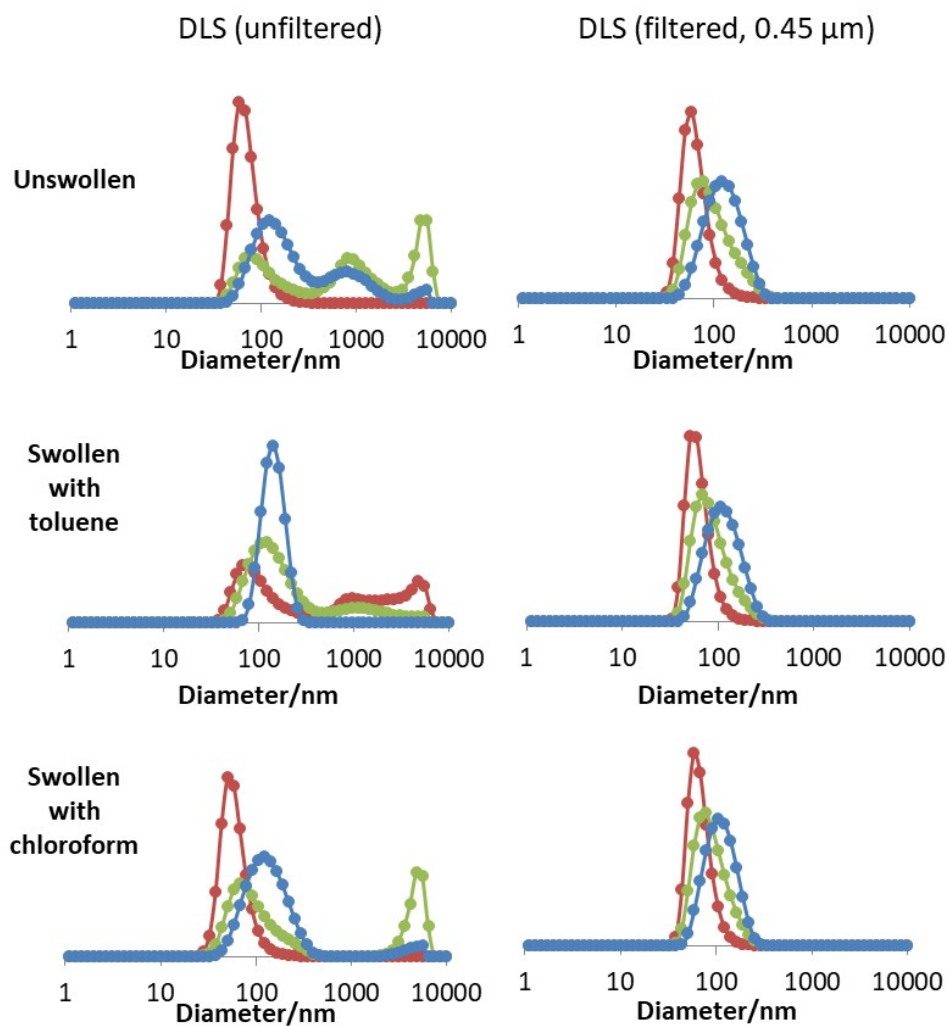


**Figure S 6.** Comparison of unfiltered and filtered DLS traces of the aqueous dispersions of the CCMs  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-St}_{50}\text{-}b\text{-(St}_{0.95}\text{-co-DPPS}_{0.05})_{300}\text{-}b\text{-(St}_{0.90}\text{-co-DEGDMA}_{0.10})_{150}\text{-SC(S)SnPr}$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).

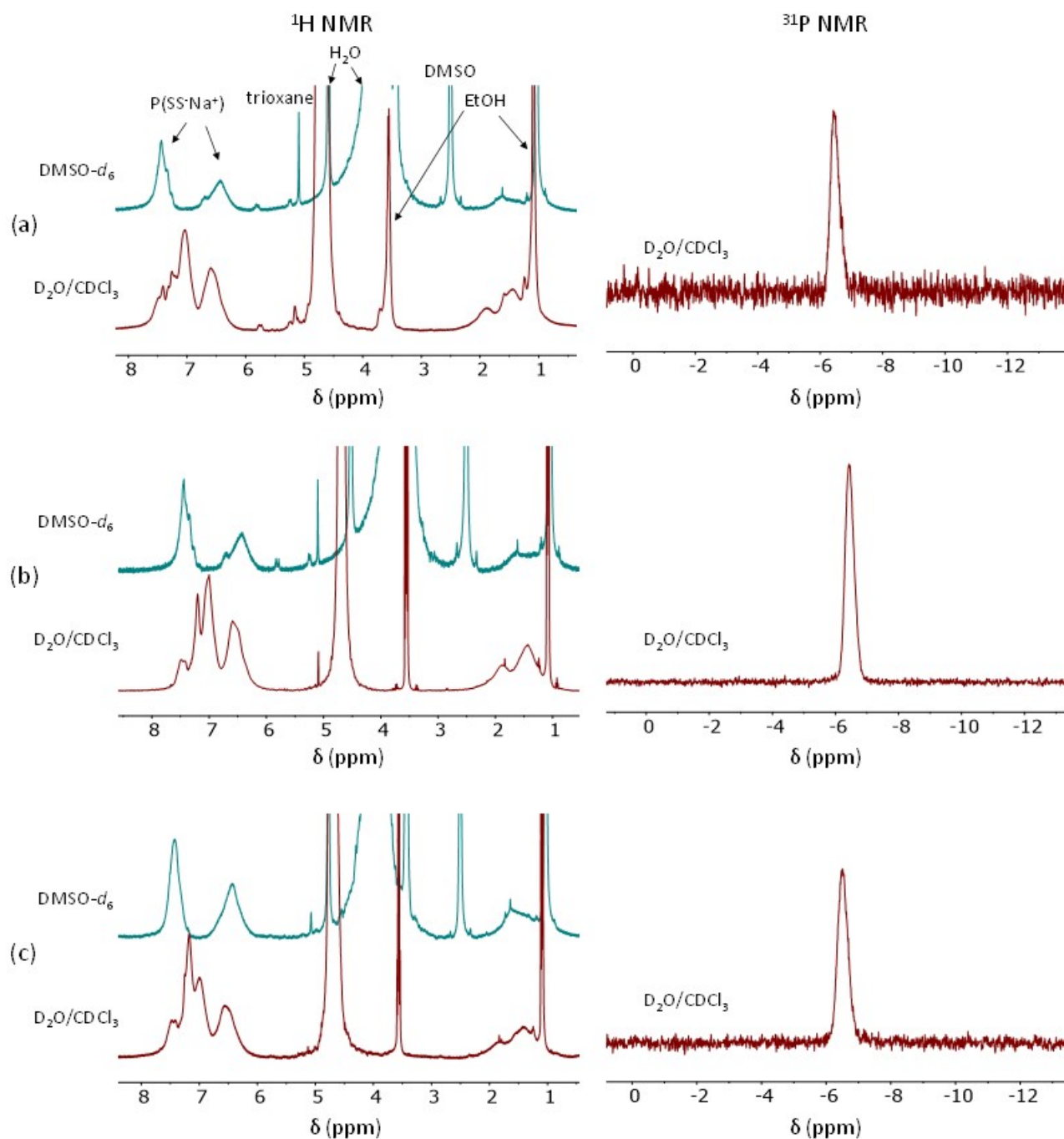




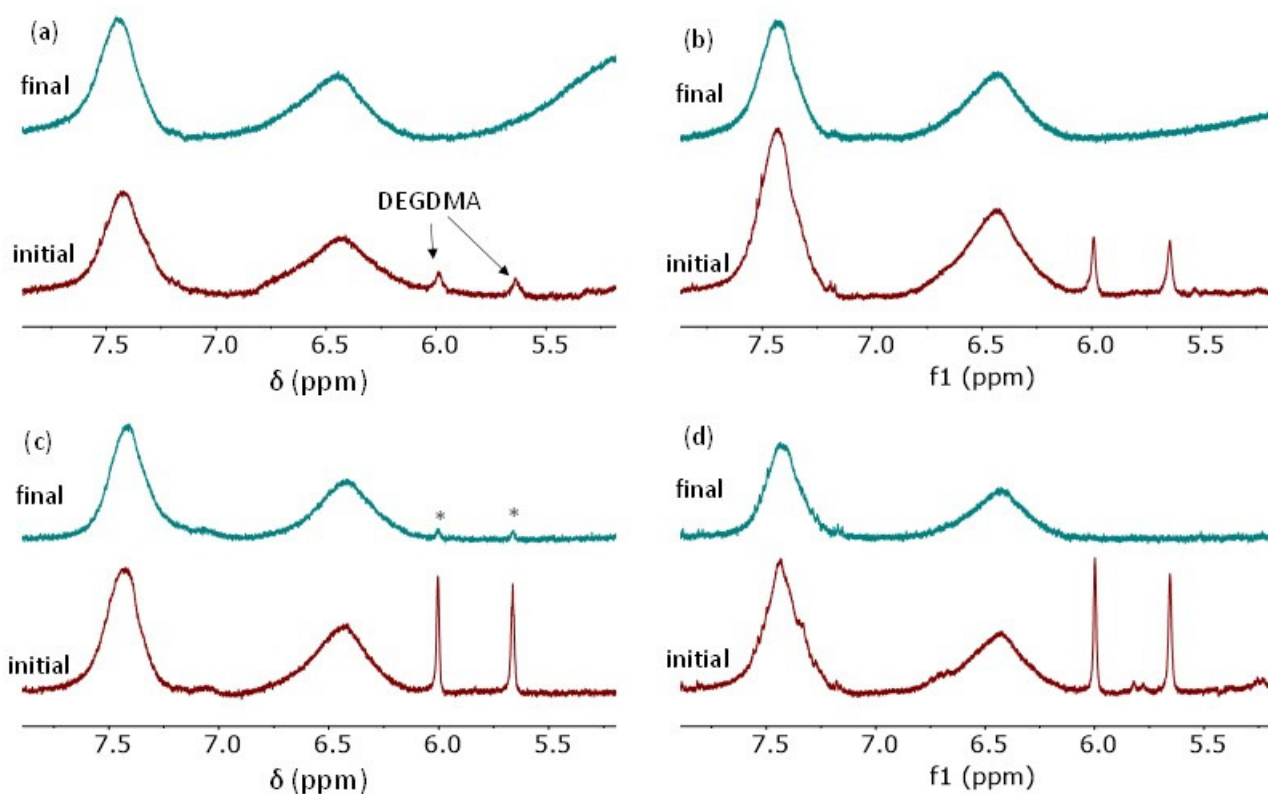
**Figure S 7.** Comparison of unfiltered and filtered DLS traces of the aqueous dispersions of the CCMs  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{50}\text{-}b\text{-(St}_{0.90}\text{-}co\text{-DPPS}_{0.10})_{300}\text{-}b\text{-(St}_{0.90}\text{-}co\text{-DEGDMA}_{0.10})_{150}\text{-SC(S)SnPr}$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



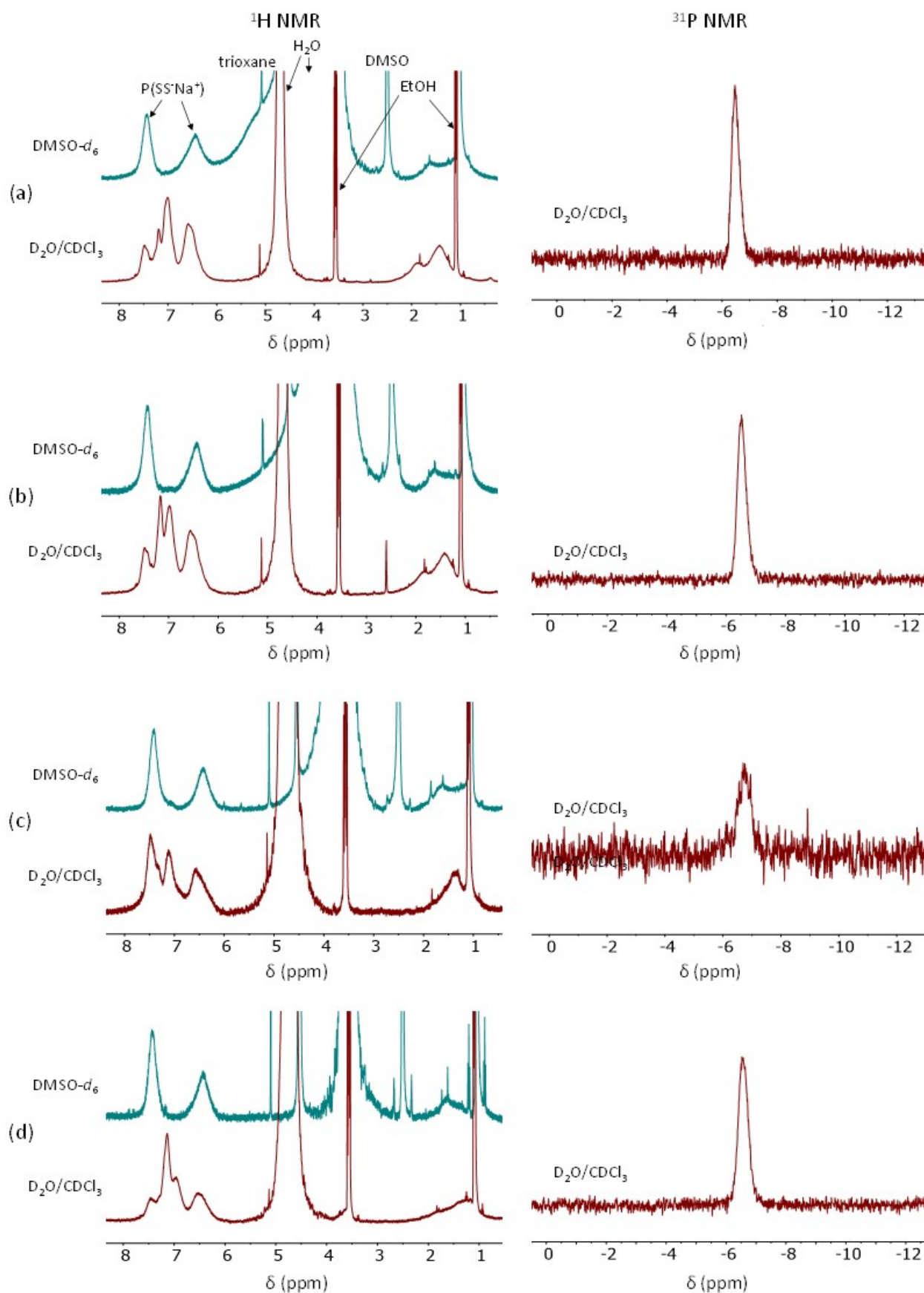
**Figure S 8.** Comparison of unfiltered and filtered DLS traces of the aqueous dispersions of the CCMs  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{50}\text{-}b\text{-(St}_{0.80}\text{-}co\text{-DPPS}_{0.20})_{300}\text{-}b\text{-(St}_{0.90}\text{-}co\text{-DEGDMA}_{0.10})_{150}\text{-SC(S)SnPr}$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



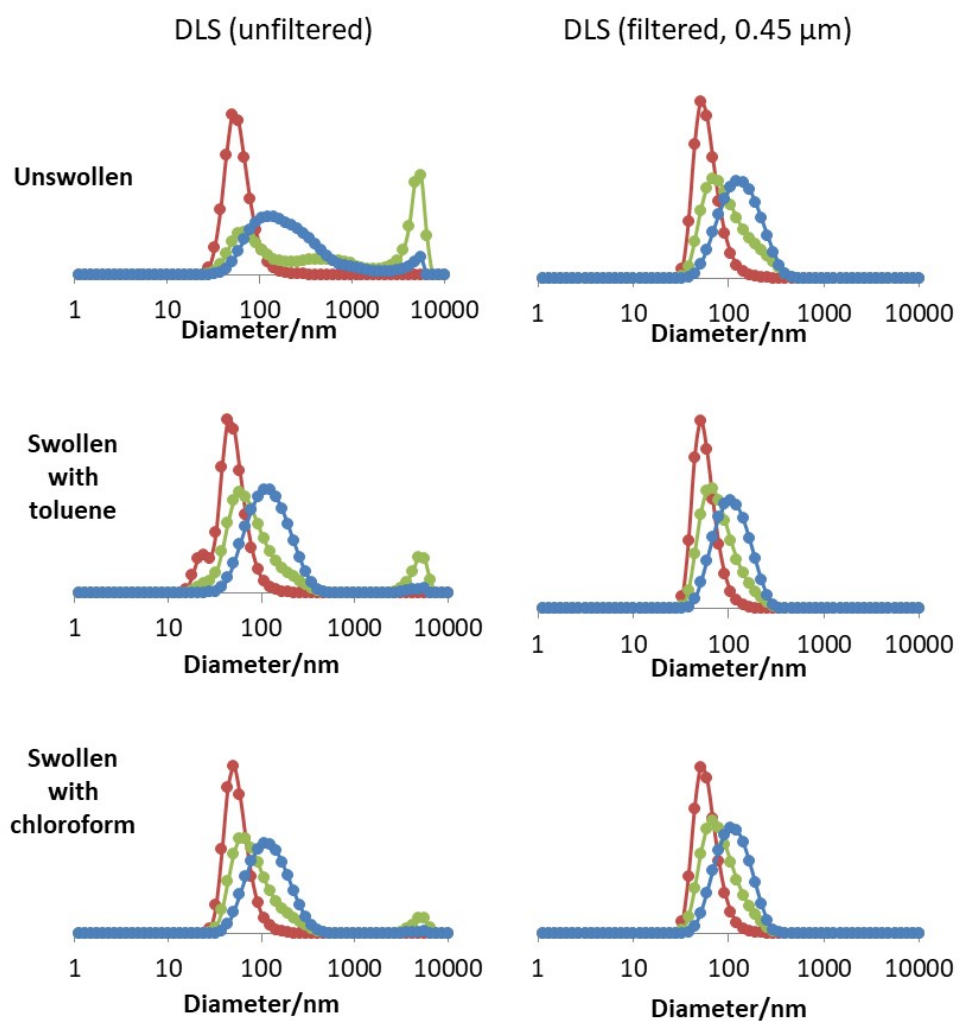
**Figure S 9.**  $^1\text{H}$  and  $^{31}\text{P}$  NMR spectra of  $\text{R}_0\text{-(SS-Na}^+\text{)}_{140}\text{-}b\text{-St}_{50}\text{-}b\text{-(St}_{1-y}\text{-co-DPPS}_y\text{)}_{300}\text{-}b\text{-(St}_{0.9}\text{-co-DEGDMA}_{0.1}\text{)}_{150}\text{-SC(S)SnPr}$  latexes in  $\text{D}_2\text{O/CDCl}_3$  (3:1 v/v). (a)  $y = 0.05$ ; (b)  $y = 0.10$ ; (c)  $y = 0.20$ . The  $^1\text{H}$  NMR spectra in  $\text{DMSO-}d_6$  are also shown for comparison.



**Figure S 10.**  $^1\text{H}$  NMR monitoring of the DEGDM polymerization for the crosslinking of  $\text{R}_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{1-y}\text{-co-DPPS}_y)_{300}\text{-SC(S)SnPr}$  to yield the CCMs  $\text{R}_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{1-y}\text{-co-DPPS}_y)_{300}\text{-}b\text{-DEGDMA}_{15}\text{-SC(S)SnPr}$ . (a)  $y = 0.05$ ; (b)  $y = 0.10$ ; (c)  $y = 0.20$ ; (d)  $y = 0.25$ . The starred resonances correspond to residual DEGDM ( $< 1\%$  by integration against the trioxane standard). All NMR samples were prepared by adding a drop of the reaction mixture directly to the  $\text{DMSO-}d_6$  solvent in the NMR tube.

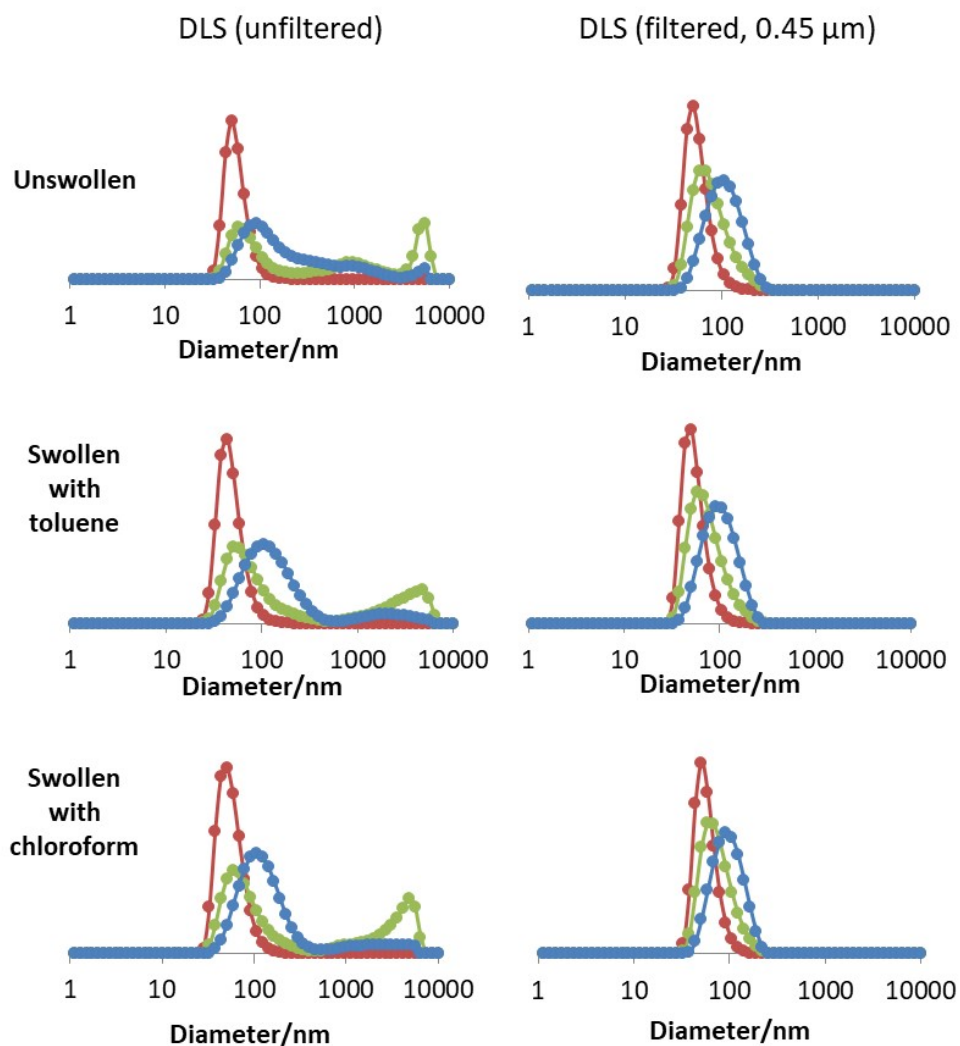


**Figure S 11.**  $^1\text{H}$  and  $^{31}\text{P}$  NMR spectra of  $\text{R}_0\text{-(SS-Na}^+\text{)}_{140}\text{-}b\text{-(St}_{1-y}\text{-co-DPPS}_y\text{)-}b\text{-DEGDMA}_{15}\text{-SC(S)SnPr}$  latexes in  $\text{D}_2\text{O/CDCl}_3$  (3:1 v/v). (a)  $y = 0.05$ ; (b)  $y = 0.1$ ; (c)  $y = 0.2$ ; (d)  $y = 0.25$ . The  $^1\text{H}$  NMR spectra in  $\text{DMSO-}d_6$  are also shown for comparison.

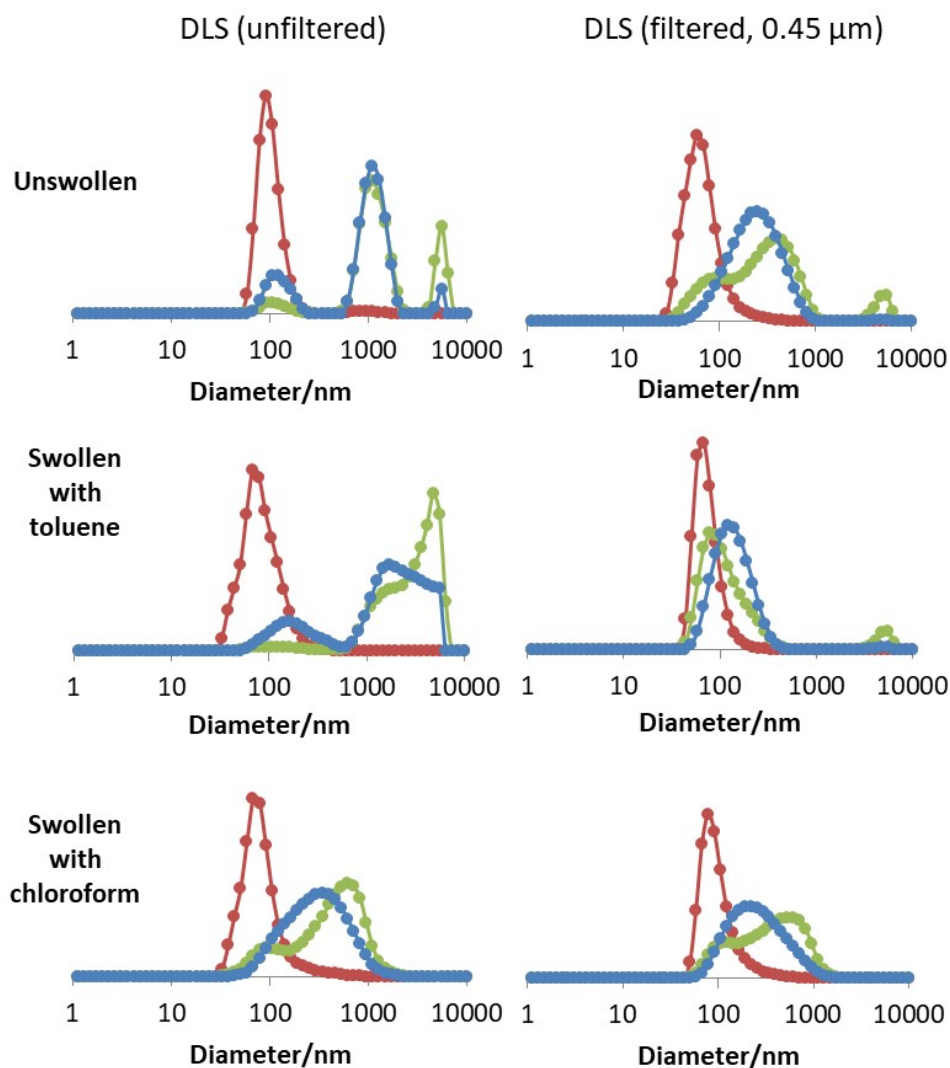


**Figure S 12.** Comparison of unfiltered and filtered DLS traces of the aqueous dispersions of the CCMs  $R_0\text{-(SS}^+\text{Na}^+)_{140}\text{-}b\text{-(St}_{0.95}\text{-co-DPPS}_{0.05})_{300}\text{-}b\text{-DEGDMA}_{15}\text{-SC(S)SnPr}$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).

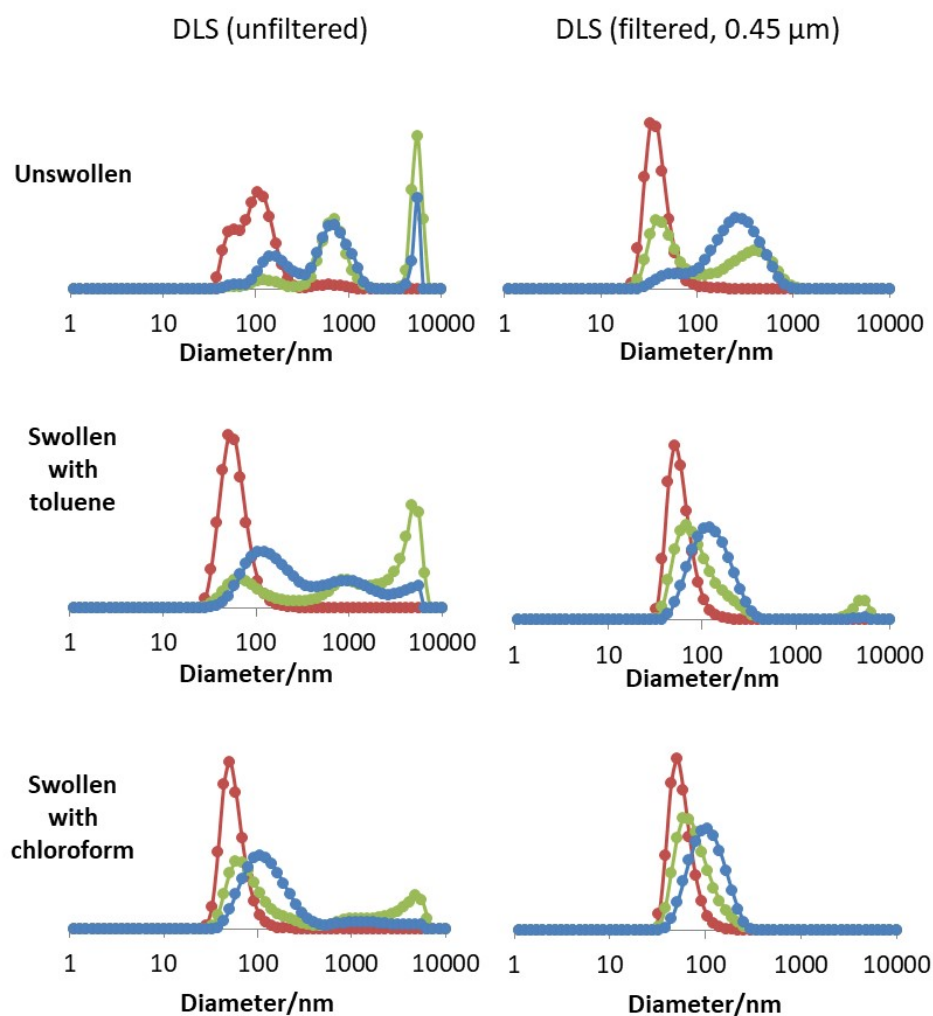




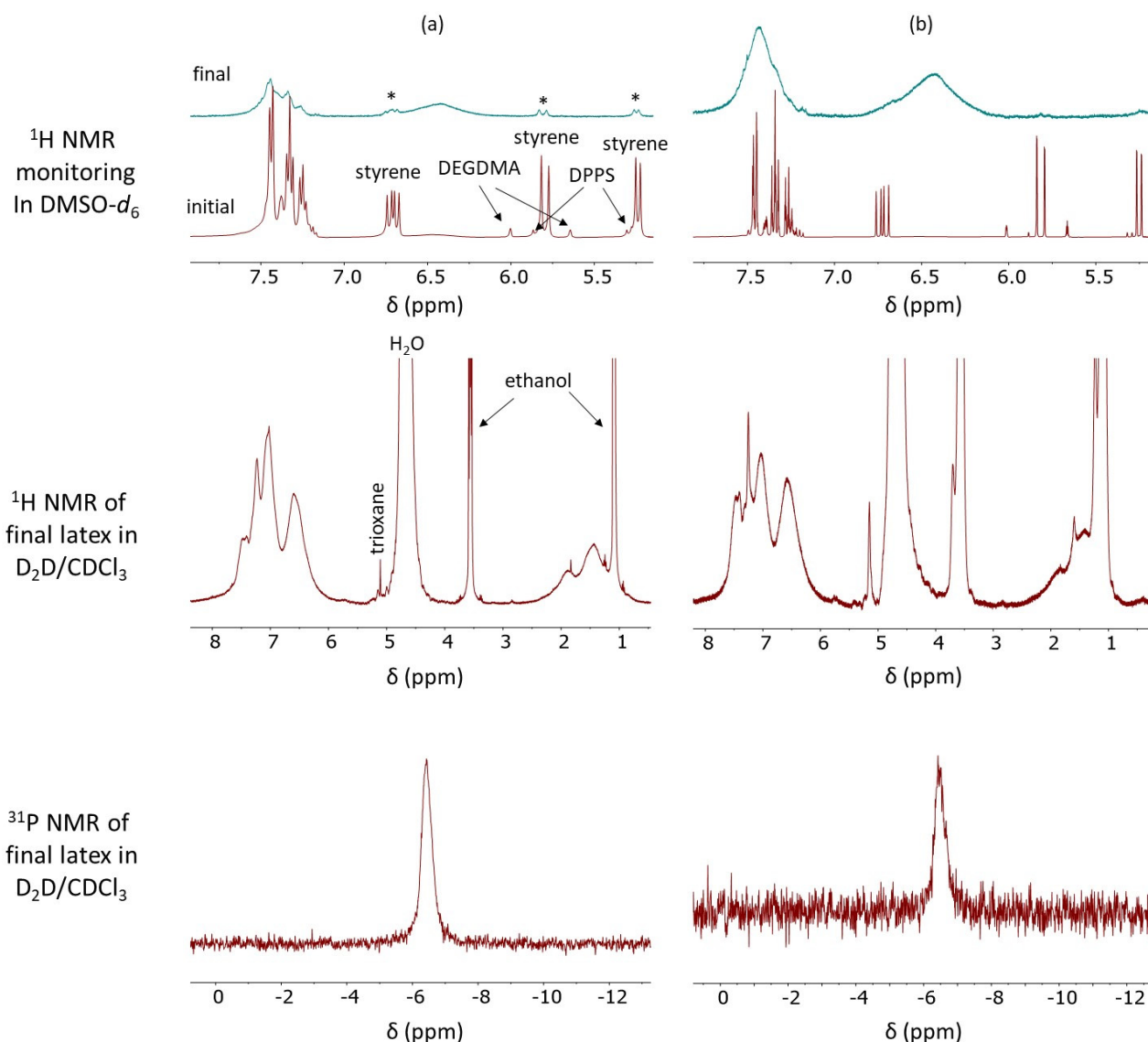
**Figure S 13.** Comparison of unfiltered and filtered DLS traces of the aqueous dispersions of the CCMs  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{0.90}\text{-co-DPPS}_{0.10})_{300}\text{-}b\text{-DEGDMA}_{15}\text{-SC(S)SnPr}$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



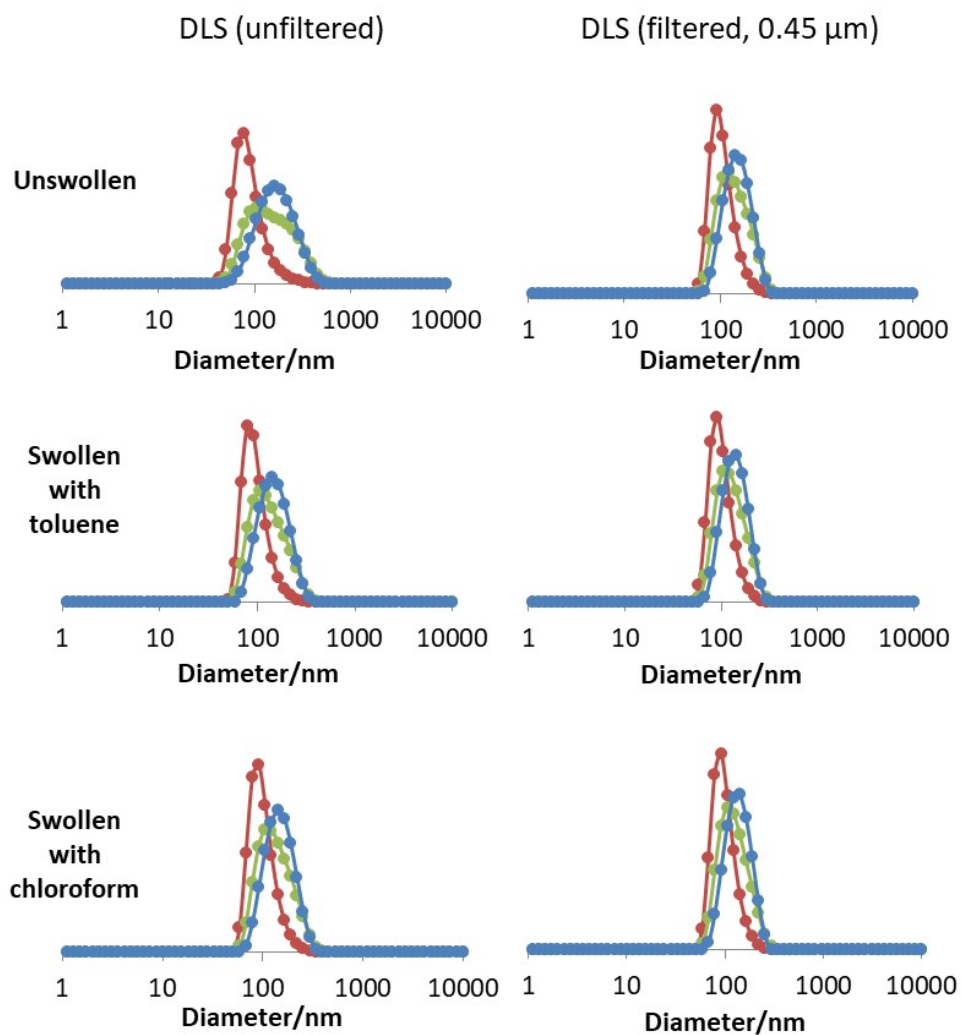
**Figure S 14.** Comparison of unfiltered and filtered DLS traces of the aqueous dispersions of the CCMs  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{0.80}\text{-co-DPPS}_{0.20})_{300}\text{-}b\text{-DEGDMA}_{15}\text{-SC(S)SnPr}$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



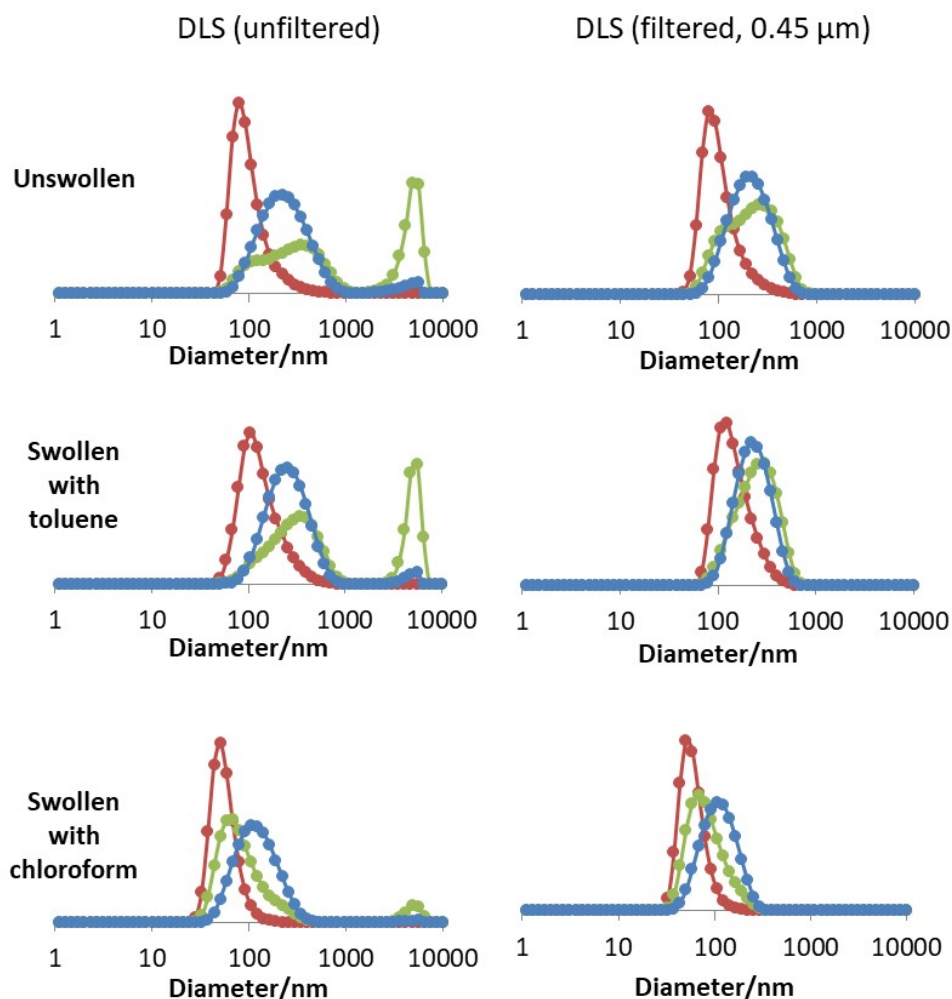
**Figure S 15.** Comparison of unfiltered and filtered DLS traces of the aqueous dispersions of the CCMs  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{0.75}\text{-co-DPPS}_{0.25})_{300}\text{-}b\text{-DEGDMA}_{15}\text{-SC(S)SnPr}$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



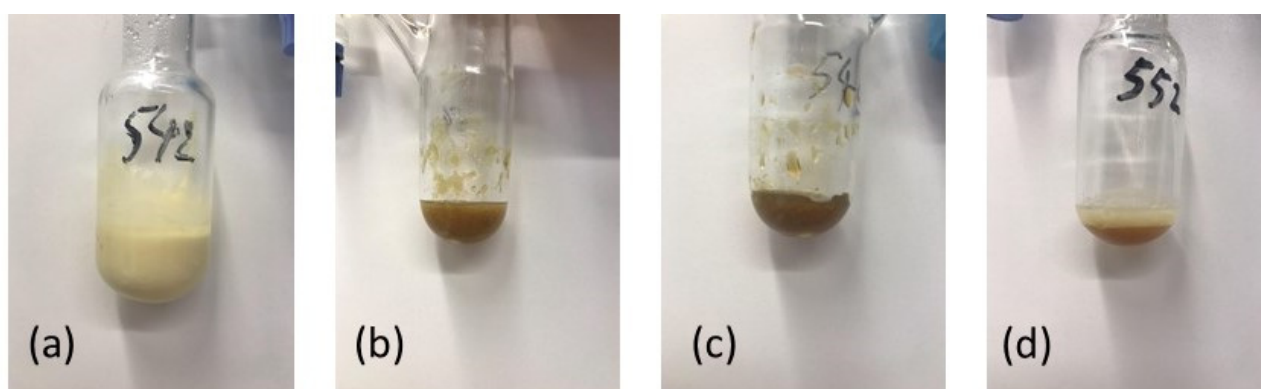
**Figure S 16.** NMR data for the synthesis of the NGs R<sub>0</sub>-(SS<sup>-</sup>Na<sup>+</sup>)<sub>140</sub>-*b*-St<sub>*x*</sub>-*b*-(St<sub>*y*</sub>-*co*-DPPS<sub>*z*</sub>-*co*-DEGDMA<sub>15</sub>)-SC(S)SnPr. (a) *x* = 0, *y* = 285, *z* = 15; (b) *x* = 50, *y* = 425, *z* = 30. The starred resonances correspond to residual styrene [2.2% for (a), < 1% for (b), by integration against the trioxane standard]. All NMR samples were prepared by adding a drop of the reaction mixture directly to the deuterated solvent or solvent mixture in the NMR tube.



**Figure S 17.** Comparison of unfiltered and filtered DLS traces of the aqueous dispersions of the NG  $\text{R}_0\text{-(SS-Na}^+\text{)}_{140}\text{-}b\text{-(St}_{285}\text{-co-DPPS}_{15}\text{-co-DEGDMA}_{15}\text{)-SC(S)SnPr}$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



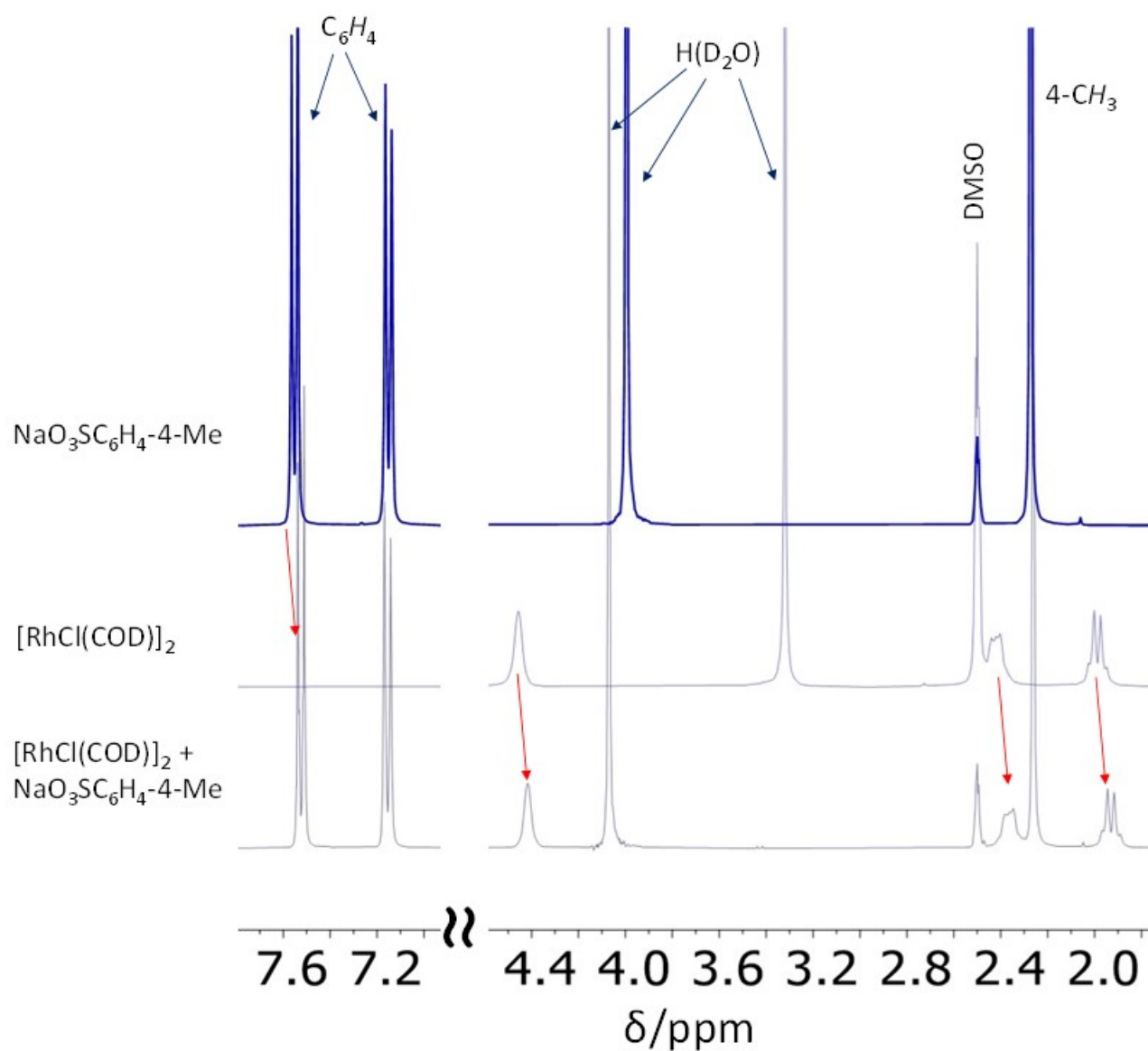
**Figure S 18.** Comparison of unfiltered and filtered DLS traces of the aqueous dispersions of the NG  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{50}\text{-}b\text{-(St}_{425}\text{-}co\text{-DPPS}_{30}\text{-}co\text{-DEGDMA}_{15})\text{-SC(S)SnPr}$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



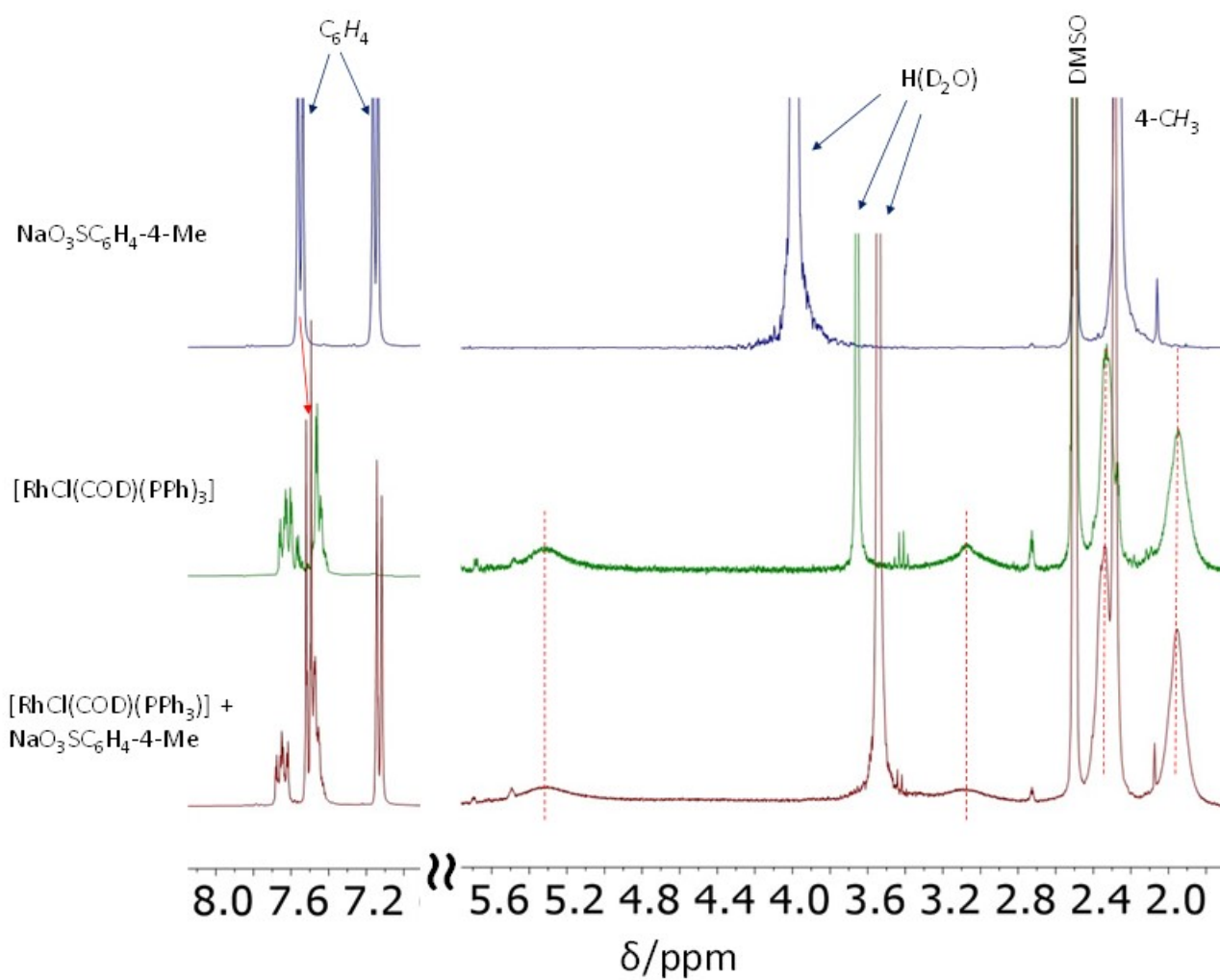
**Figure S 19.** Photos of the CCM and NG polymer latexes after treatment with a toluene solution of  $[\text{RhCl(COD)}]_2$  ( $\text{P/Rh} = 1:1$ ). (a)  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{0.75}\text{-}co\text{-DPPS}_{0.25})_{300}\text{-}b\text{-DEGDMA}_{15}\text{-SC(S)SnPr}$ ; (b)  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{0.9}\text{-}co\text{-DPPS}_{0.1})_{300}\text{-}b\text{-DEGDMA}_{15}\text{-SC(S)SnPr}$ ; (c)  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{50}\text{-}b\text{-(St}_{0.9}\text{-}co\text{-DPPS}_{0.1})_{300}\text{-}b\text{-(St}_{0.9}\text{-}co\text{-DEGDMA}_{0.1})_{150}\text{-SC(S)SnPr}$ ; (d)  $R_0\text{-(SS}^-\text{Na}^+)_{140}\text{-}b\text{-(St}_{285}\text{-}co\text{-DPPS}_{15}\text{-}co\text{-DEGDMA}_{15})\text{-SC(S)SnPr}$ .



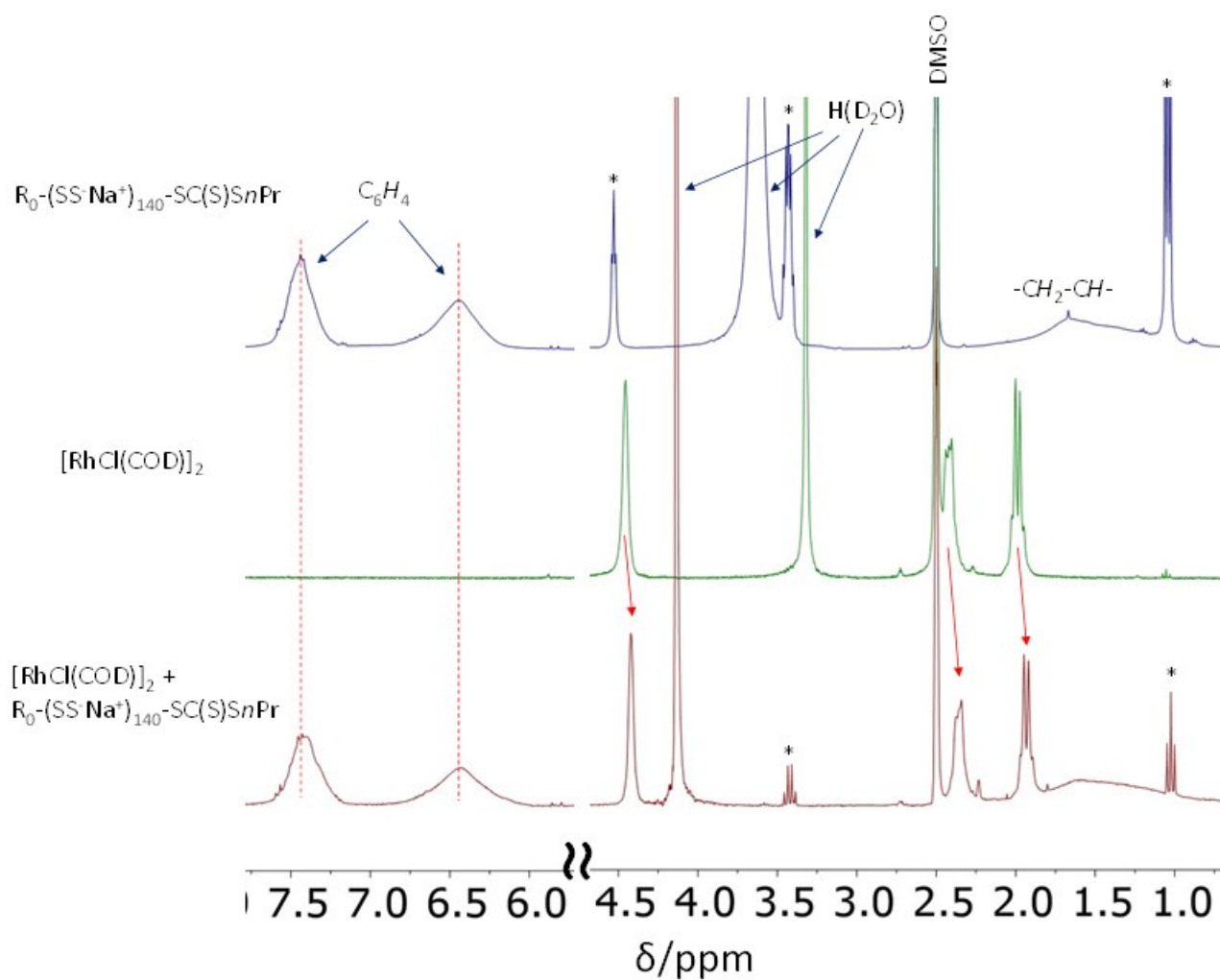
**(b)  $^1\text{H}$  NMR investigations of the Rh-sulfonate interaction**



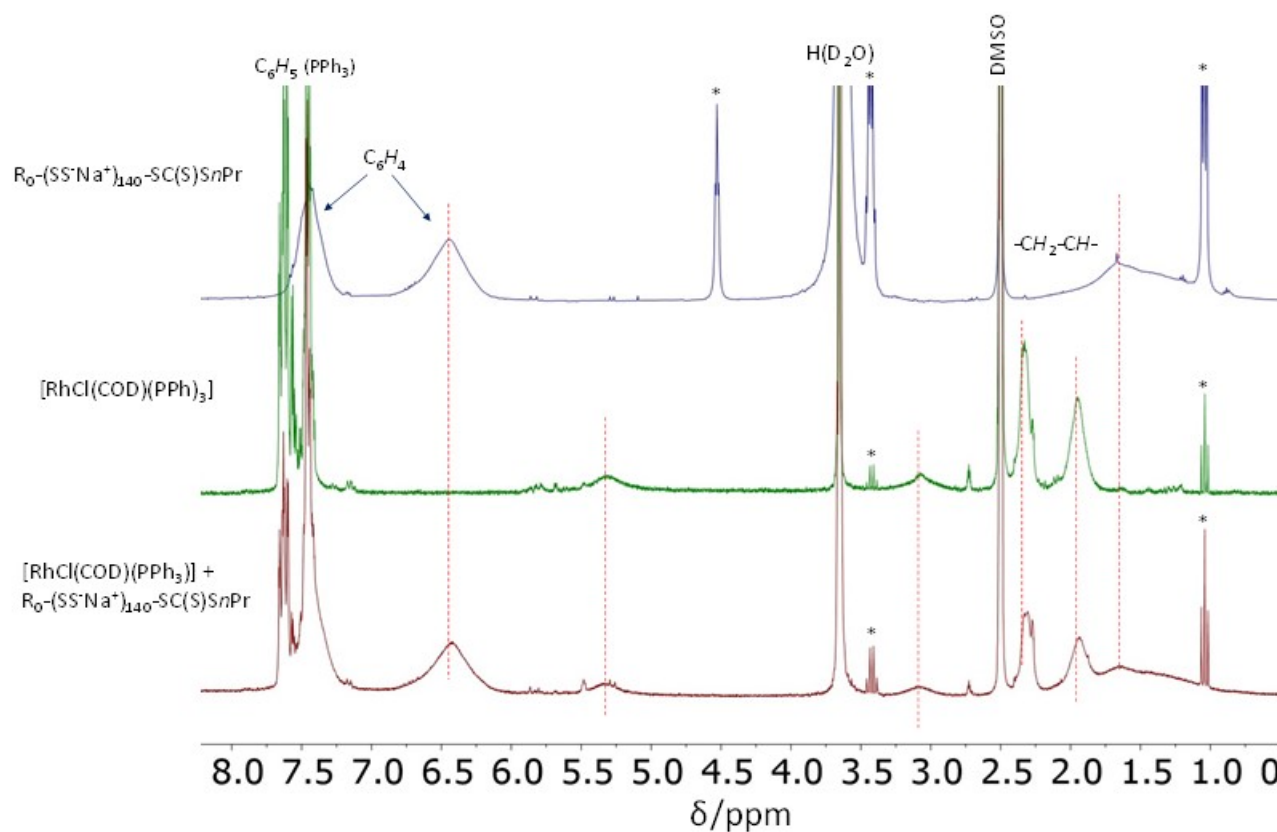
**Figure S 20.**  $^1\text{H}$  NMR spectrum in  $\text{DMSO-}d_6/\text{D}_2\text{O}$  (4:1 v/v) of the  $[\text{RhCl}(\text{COD})]_2/\text{NaO}_3\text{SC}_6\text{H}_4\text{-4-CH}_3$  mixture (Rh/sulfonate = ca. 1:5), in comparison with those of the separate reagents measured under the same conditions.



**Figure S 21.**  $^1\text{H}$  NMR spectrum in DMSO- $d_6$ /D $_2$ O (4:1 v/v) of the  $[\text{RhCl}(\text{COD})(\text{PPh}_3)]/\text{NaO}_3\text{SC}_6\text{H}_4\text{-4-CH}_3$  mixture (Rh/sulfonate = ca. 1:5), in comparison with those of the separate reagents measured under the same conditions.



**Figure S 22.**  $^1\text{H}$  NMR spectrum in DMSO- $d_6$ /D $_2$ O (6:1 v/v) of the  $[\text{RhCl}(\text{COD})]_2/\text{R}_0-(\text{SS}^-\text{Na}^+)_{140}-\text{SC}(\text{S})\text{SnPr}$  mixture (Rh/sulfonate = ca. 1:5), in comparison with those of the separate reagents measured under the same conditions. The starred resonances below to the residual ethanol in the macroRAFT polymer sample.

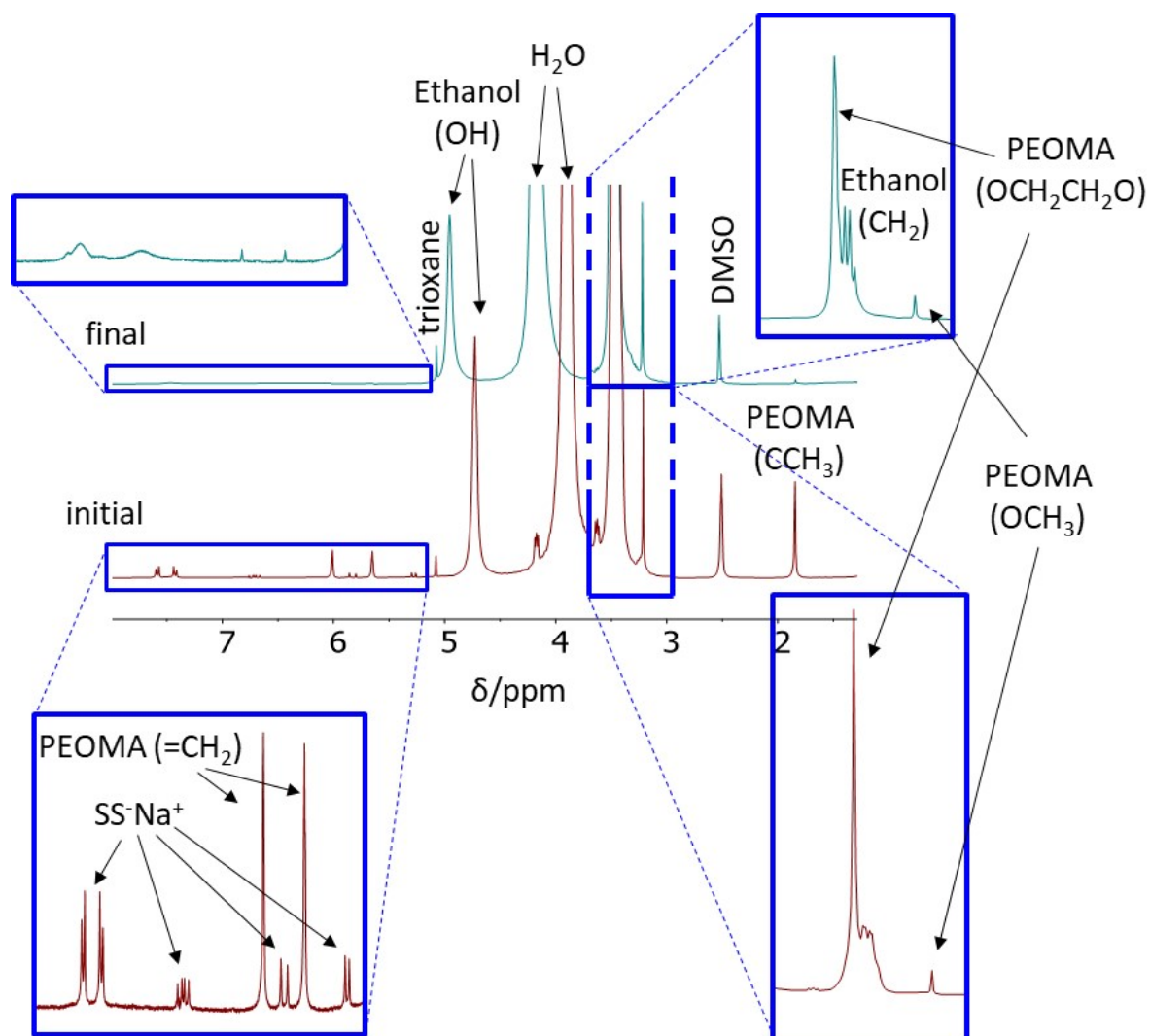


**Figure S 23.**  $^1\text{H}$  NMR spectrum in  $\text{DMSO-}d_6/\text{D}_2\text{O}$  (6:1 v/v) of the  $[\text{RhCl(COD)(PPh}_3)]/\text{R}_0\text{-(SS}^-\text{Na}^+)_{140}\text{-SC(S)SnPr}$  mixture ( $\text{Rh/sulfonate} = \text{ca. } 1:5$ ), in comparison with those of the separate reagents measured under the same conditions. The starred resonances below to the residual ethanol in the macroRAFT polymer sample.

(c) TPP-functionalized CCMs and NGs with P(SS<sup>-</sup>Na<sup>+</sup>-*co*-PEOMA) copolymer blocks in the hydrophilic shell

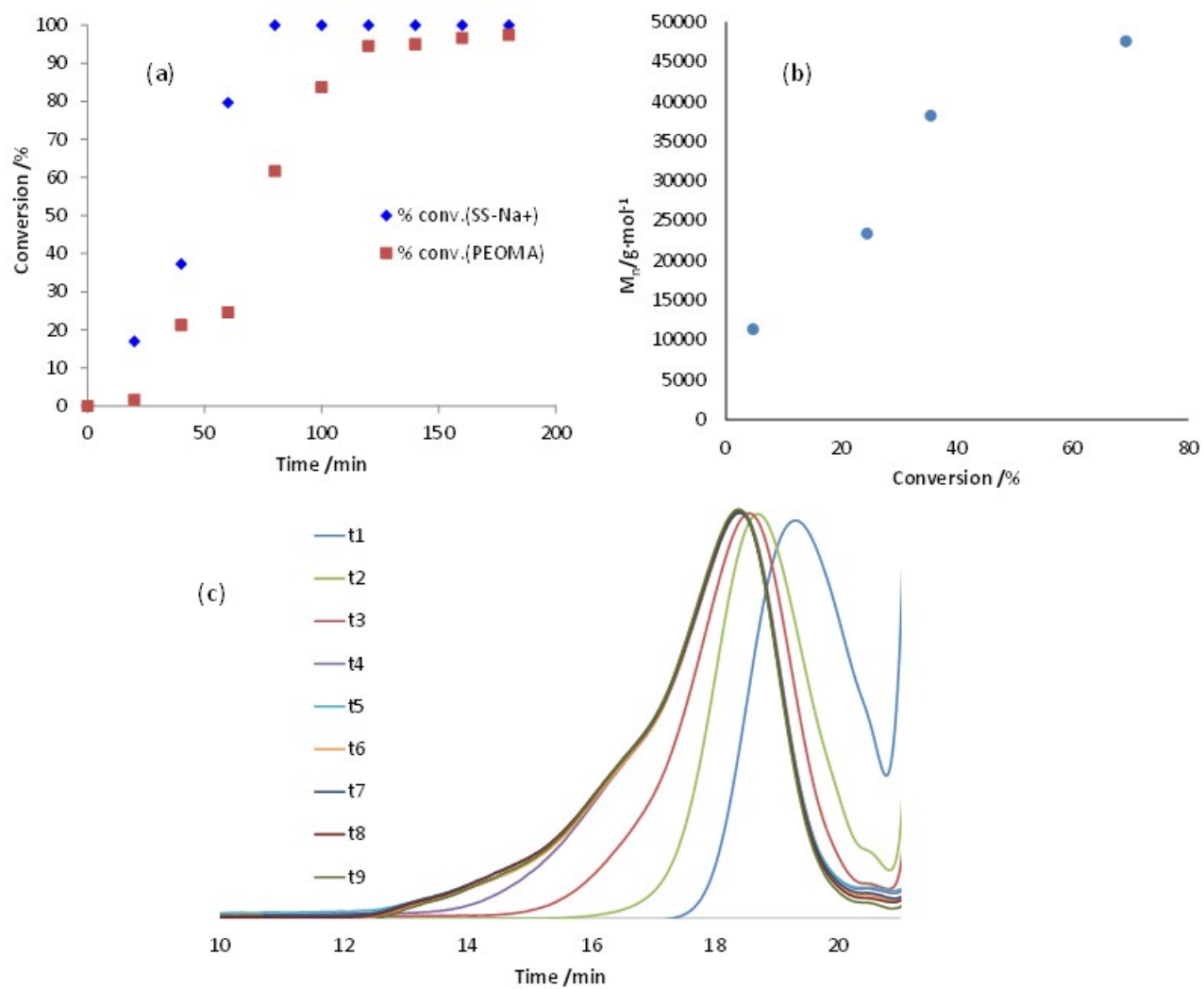
**Table S 2.** List of all polymers with a P(SS<sup>-</sup>Na<sup>+</sup>-*co*-PEOMA) shell synthesized in this study and reference to their characterization.

Formula <sup>a</sup>	SEC	<sup>1</sup> H NMR	<sup>31</sup> P NMR	DLS	TEM	ζ (mV)	Ref.
<b>1. Hydrosoluble R<sub>0</sub>-(SS<sup>-</sup>Na<sup>+</sup><sub>0.2</sub>-<i>co</i>-PEOMA<sub>0.8</sub>)<sub>x</sub>-SC(S)SPr macroRAFT agents</b>							
R <sub>0</sub> -[(SS <sup>-</sup> Na <sup>+</sup> ) <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ] <sub>50</sub> -SC(S)SnPr	Figure S 25	Figure S 24 Figure S 27		Figure S 26	Figure S 26	/	HW617
R <sub>0</sub> -[(SS <sup>-</sup> Na <sup>+</sup> ) <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ] <sub>140</sub> -SC(S)SnPr	/	Figure S 24 Figure S 27		Figure S 26	Figure S 26	/	HW627
<b>2. Diblock P(SS<sup>-</sup>Na<sup>+</sup>-<i>co</i>-PEOMA)-<i>b</i>-St<sub>50</sub> macroRAFT agents</b>							
R <sub>0</sub> -[(SS <sup>-</sup> Na <sup>+</sup> ) <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ] <sub>50</sub> - <i>b</i> -St <sub>50</sub> -SC(S)SnPr	/	/		Figure S 28	Figure S 28	/	HW688
R <sub>0</sub> -[(SS <sup>-</sup> Na <sup>+</sup> ) <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ] <sub>140</sub> - <i>b</i> -St <sub>50</sub> -SC(S)SnPr	/	/		Figure S 28	Figure S 28	/	HW691
<b>3. Diblock P(SS<sup>-</sup>Na<sup>+</sup>-<i>co</i>-PEOMA)-<i>b</i>-P(St-<i>co</i>-DPPS) micelles</b>							
R <sub>0</sub> -[(SS <sup>-</sup> Na <sup>+</sup> ) <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ] <sub>50</sub> - <i>b</i> -(St <sub>0.9</sub> - <i>co</i> -DPPS <sub>0.1</sub> ) <sub>300</sub> -SC(S)SnPr	/	Figure S 29	Figure S 29	Figure S 30 Figure S 35	Figure S 30	/	HW618H W701
R <sub>0</sub> -[(SS <sup>-</sup> Na <sup>+</sup> ) <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ] <sub>140</sub> - <i>b</i> -(St <sub>0.9</sub> - <i>co</i> -DPPS <sub>0.1</sub> ) <sub>300</sub> -SC(S)SnPr		Figure S 29	Figure S 29	/	/	/	HW629
<b>4. CCMs with a P(SS<sup>-</sup>Na<sup>+</sup>-<i>co</i>-PEOMA) shell and a mixed DEGDMA-<i>co</i>-St core</b>							
R <sub>0</sub> -[(SS <sup>-</sup> Na <sup>+</sup> ) <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ] <sub>50</sub> - <i>b</i> -(St <sub>0.9</sub> - <i>co</i> -DPPS <sub>0.1</sub> ) <sub>300</sub> - <i>b</i> -(St <sub>0.9</sub> - <i>co</i> -DEGDMA <sub>0.1</sub> ) <sub>150</sub> -SC(S)SnPr	/	Figure S 31	Figure S 31	Figure 10 Figure S 35	Figure 10	/	HW625
R <sub>0</sub> -[(SS <sup>-</sup> Na <sup>+</sup> ) <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ] <sub>140</sub> - <i>b</i> -(St <sub>0.9</sub> - <i>co</i> -DPPS <sub>0.1</sub> ) <sub>300</sub> - <i>b</i> -(St <sub>0.9</sub> - <i>co</i> -DEGDMA <sub>0.1</sub> ) <sub>150</sub> -SC(S)SnPr	/	Figure S 31	Figure S 31	Figure 10	Figure 10	/	HW632
<b>5. CCM with P(PEOMA-<i>co</i>-SS<sup>-</sup>Na<sup>+</sup>) shell and neat DEGDMA core</b>							
R <sub>0</sub> -(PSS <sup>-</sup> Na <sup>+</sup> <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ) <sub>50</sub> - <i>b</i> -(St <sub>0.9</sub> - <i>co</i> -DPPS <sub>0.1</sub> ) <sub>300</sub> - <i>b</i> -DEGDMA <sub>90</sub> -SC(S)SPr	/	Figure S 32	Figure S 32 Figure 11	Figure 10	Figure 10	-25.9±4,5	HW620H W657
<b>6. NG with P(PEOMA-<i>co</i>-SS<sup>-</sup>Na<sup>+</sup>) shell</b>							
R <sub>0</sub> -[(SS <sup>-</sup> Na <sup>+</sup> ) <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ] <sub>50</sub> - <i>b</i> -St <sub>50</sub> - <i>b</i> -(St <sub>425</sub> - <i>co</i> -DPPS <sub>30</sub> - <i>co</i> -DEGDMA <sub>15</sub> )-SC(S)SnPr	/	Figure S 33	Figure S 33	Figure S 34	Figure S 34	/	HW698
R <sub>0</sub> -[(SS <sup>-</sup> Na <sup>+</sup> ) <sub>0.2</sub> - <i>co</i> -PEOMA <sub>0.8</sub> ] <sub>140</sub> - <i>b</i> -St <sub>50</sub> - <i>b</i> -(St <sub>425</sub> - <i>co</i> -DPPS <sub>30</sub> - <i>co</i> -DEGDMA <sub>15</sub> )-SC(S)SnPr	/	Figure S 33	Figure S 33	Figure S 34	Figure S 34	/	HW692

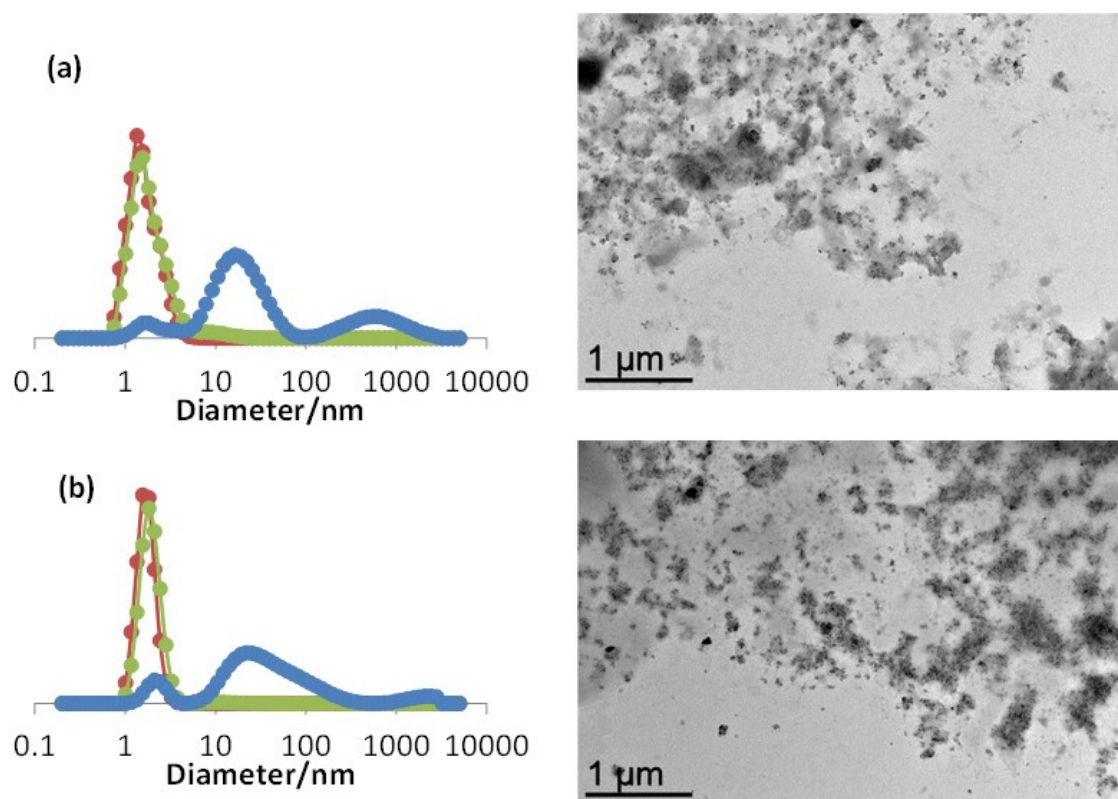


**Figure S 24.**  $^1\text{H}$  NMR monitoring of the  $\text{SS-Na}^+/\text{PEOMA}$  copolymerization for the synthesis of the  $\text{R}_0\text{-}[(\text{SS-Na}^+)_{0.2}\text{-co-PEOMA}_{0.8}]_{140}\text{-SC(S)SnPr}$  macroRAFT agents. All NMR samples were prepared by adding a drop of the reaction mixture directly to the  $\text{DMSO-}d_6$  solvent in the NMR tube.

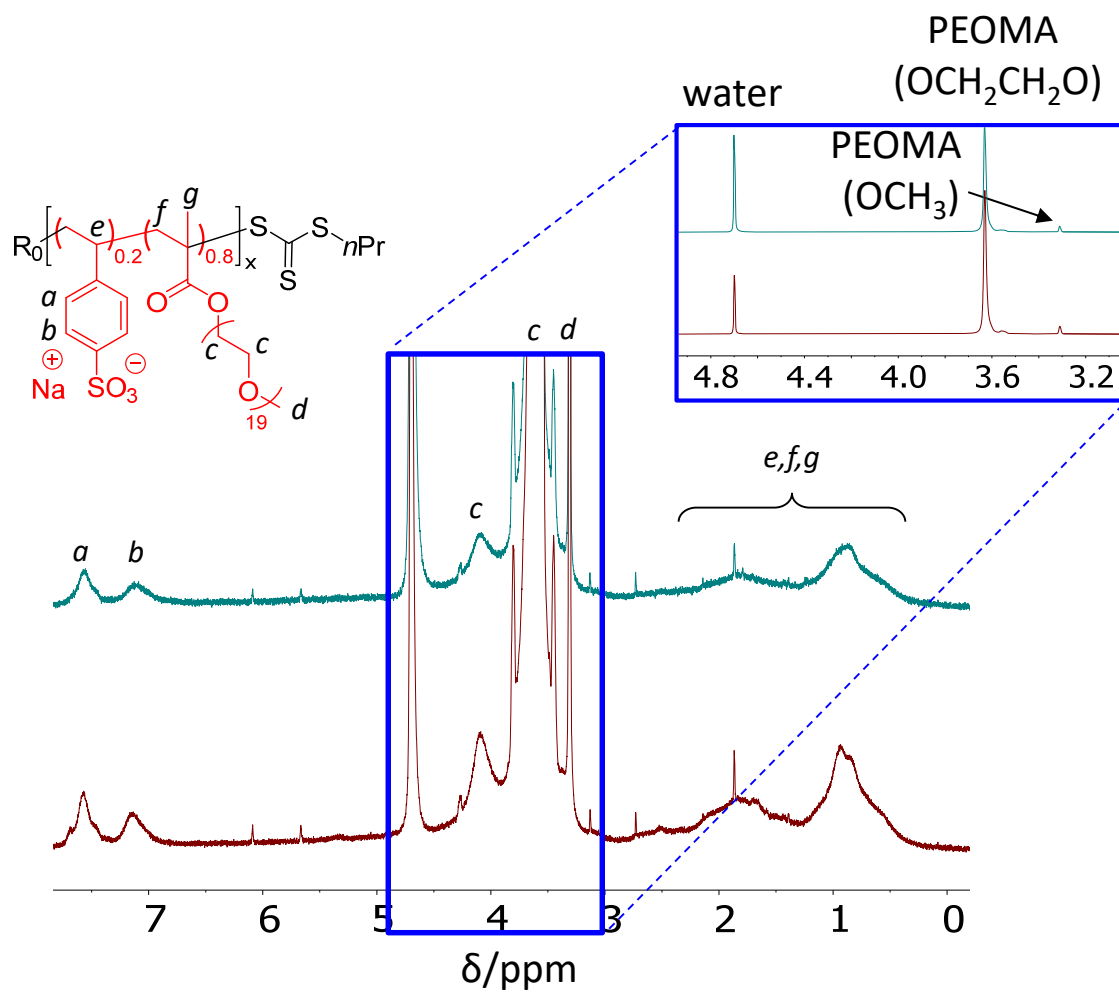




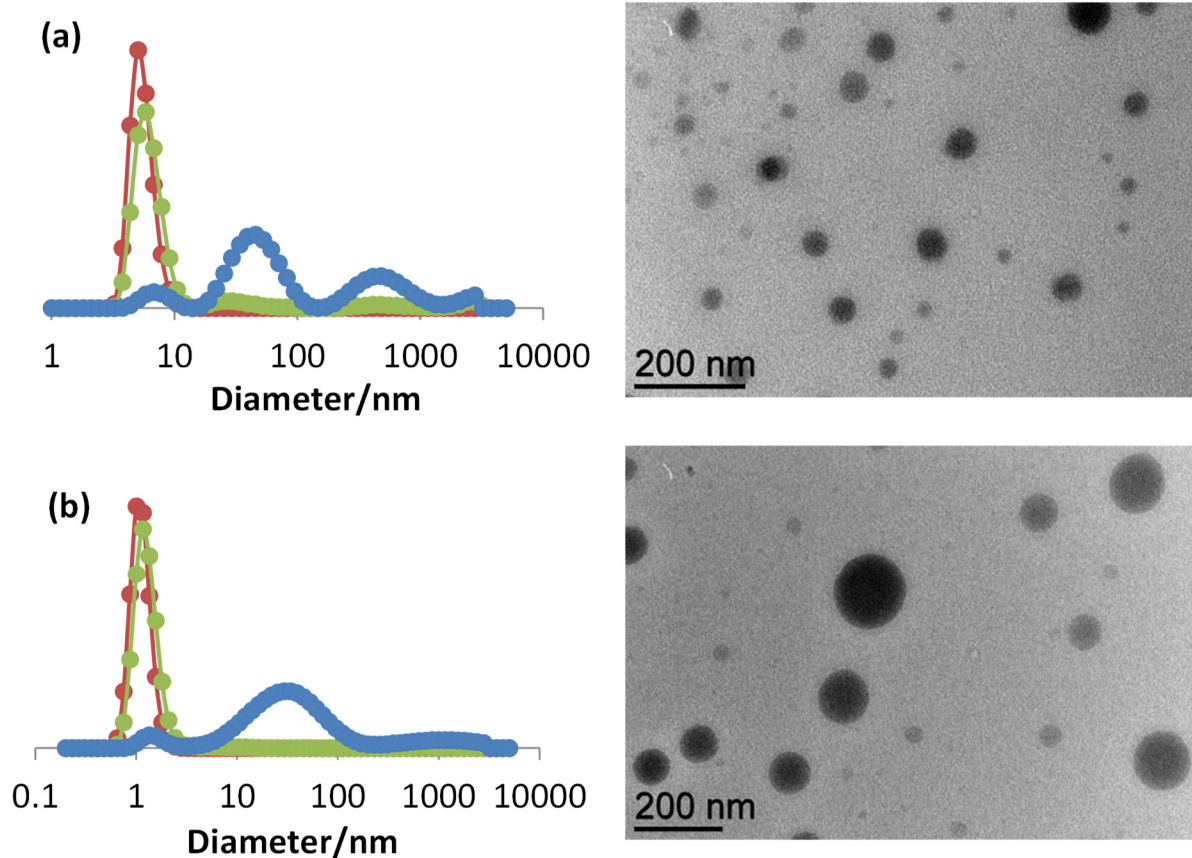
**Figure S 25.** <sup>1</sup>H NMR and SEC monitoring of the SS-Na<sup>+</sup>/PEOMA RAFT copolymerization: (a) individual monomer conversion vs. time; (b) evolution of the polymer molar mass with the global monomer conversion (c) GPC traces.



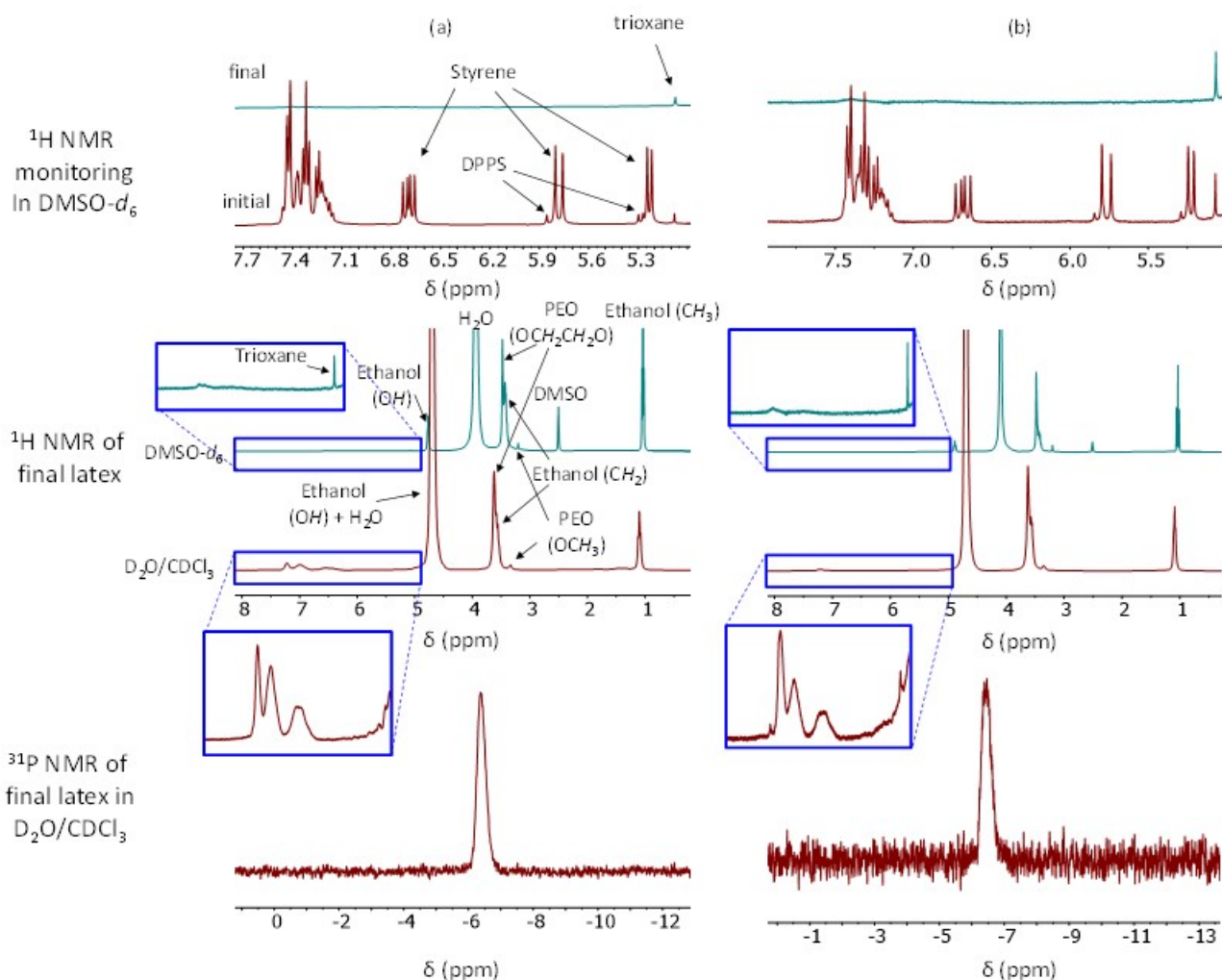
**Figure S 26.** DLS (left, unfiltered) and TEM (right) characterization of  $R_0-[(SS-Na^+)_{0.2}\text{-}co\text{-}PEOMA_{0.8}]_x\text{-SC(S)SnPr}$ . (a)  $x = 50$ ; (b)  $x = 140$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



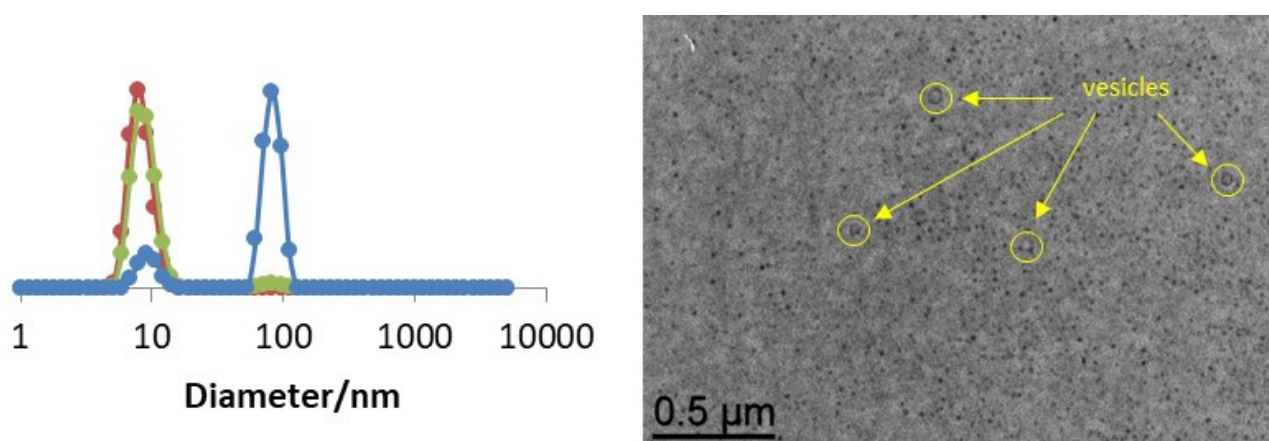
**Figure S 27.** <sup>1</sup>H NMR spectra of R<sub>0</sub>-[(SS-Na<sup>+</sup>)<sub>0.2</sub>-co-PEOMA<sub>0.8</sub>]<sub>x</sub>-SC(S)SnPr. (a) x = 50 (red); (b) x = 140 (cyan) in D<sub>2</sub>O.



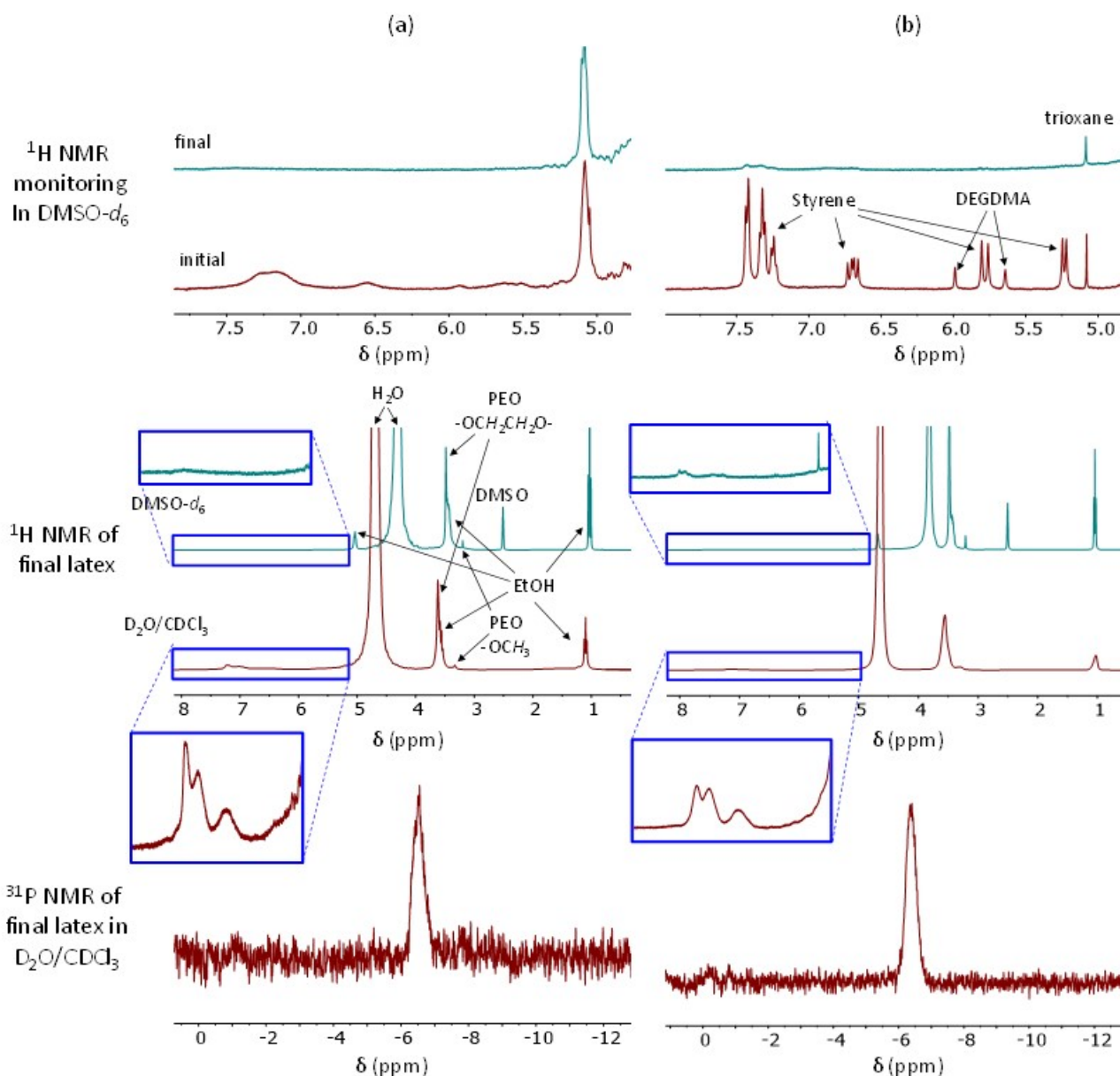
**Figure S 28.** DLS (left, unfiltered) and TEM (right) characterization of  $R_0-[(SS^-Na^+)_{0.2}\text{-}co\text{-}PEOMA_{0.8}]_x\text{-}b\text{-}St_{50}\text{-}SC(S)SnPr$ . (a)  $x = 50$ ; (b)  $x = 140$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



**Figure S 29.** NMR data for the synthesis of the  $R_0-[(SS^+Na^+)_{0.2-co-PEOMA_{0.8}}]_x-b-(St_{0.9-co-DPPS_{0.1}})_{300}-SC(S)SnPr$  diblock copolymer. (a)  $x = 50$ ; (b)  $x = 140$ . All NMR samples were prepared by adding a drop of the reaction mixture directly to the deuterated solvent or solvent mixture in the NMR tube.

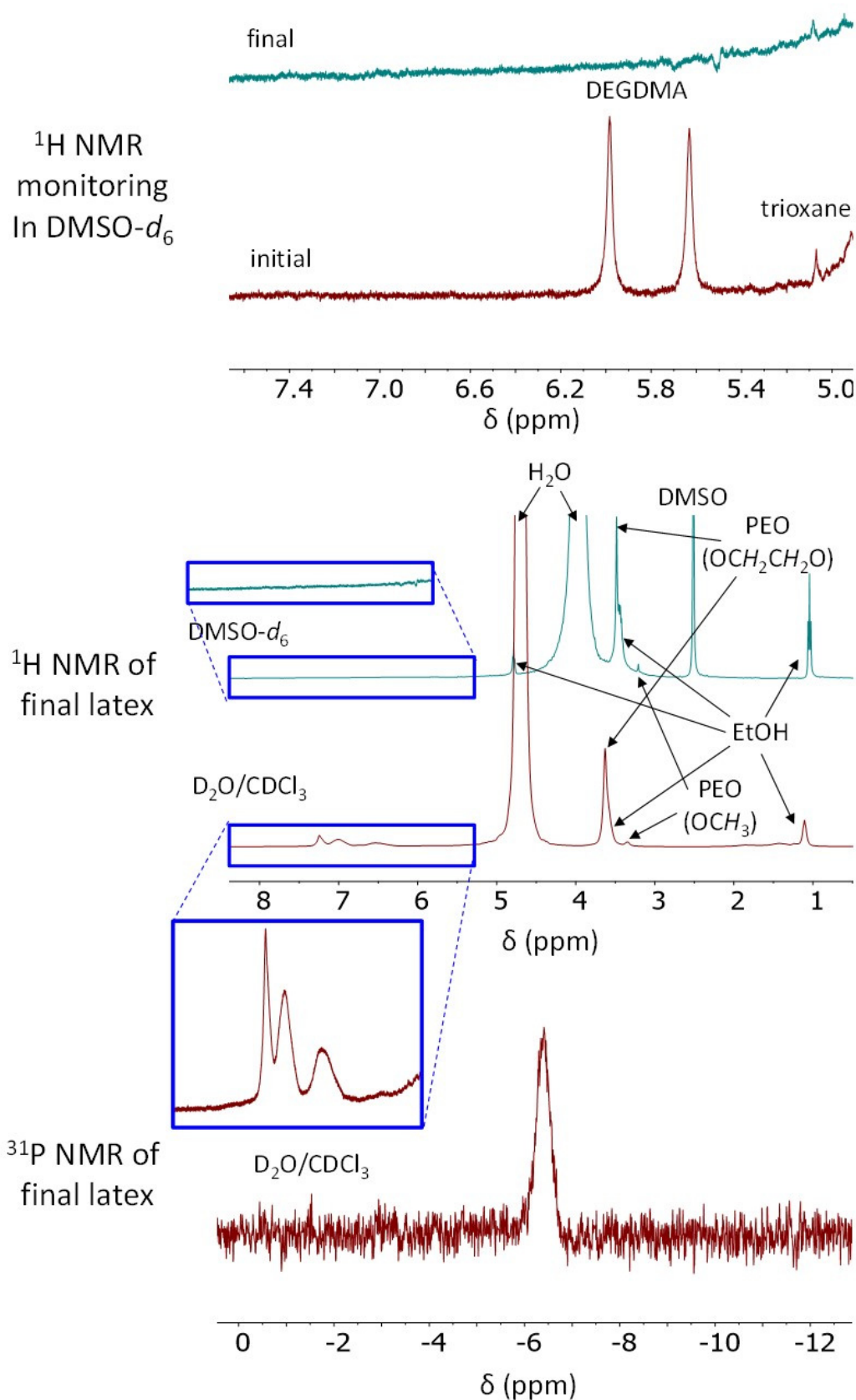


**Figure S 30.** DLS (left, unfiltered) and TEM (right) characterization of  $R_0-[(SS^+Na^+)_{0.2-co-PEOMA_{0.8}}]_{50}-b-(St_{0.9-co-DPPS_{0.1}})_{300}-SC(S)SnPr$ . Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).



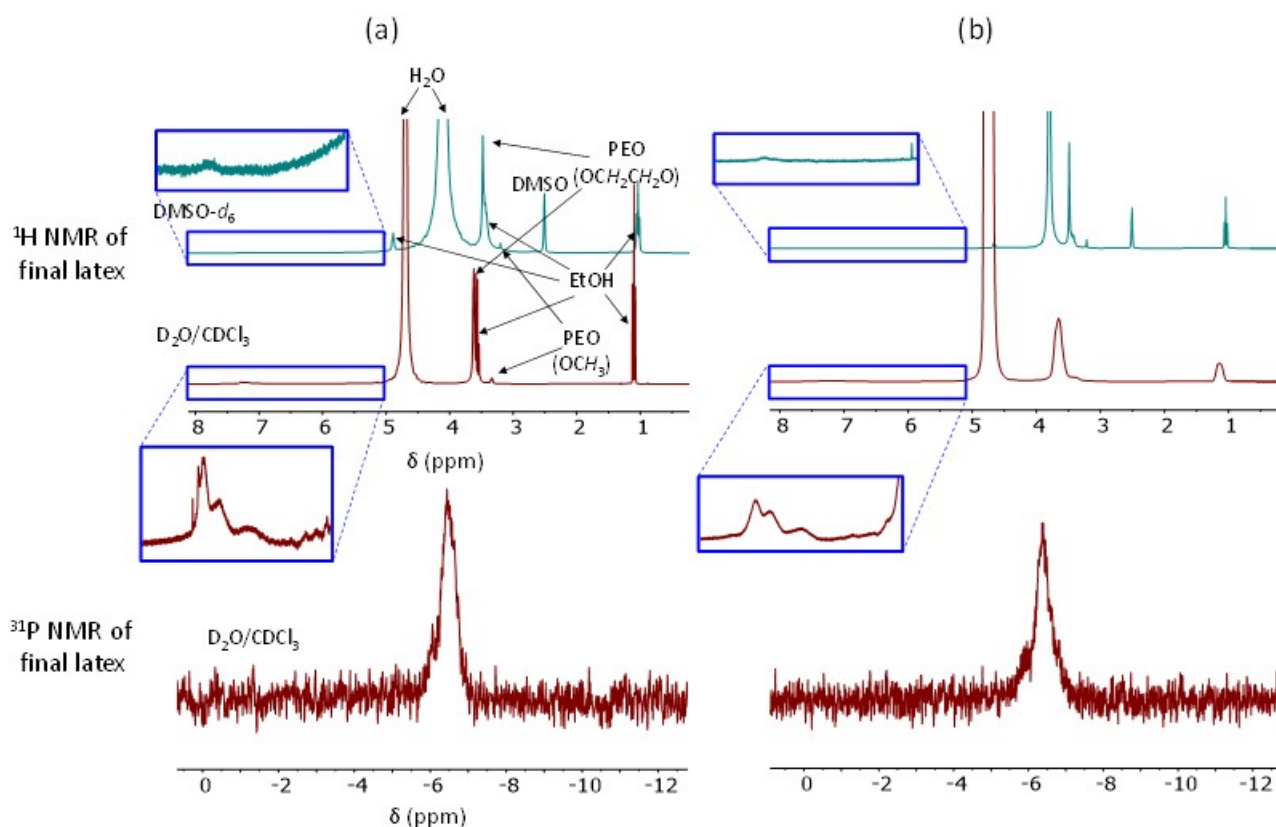
**Figure S 31.** NMR data for the synthesis of the  $R_0-[(SS^{\cdot-}Na^+)_{0.2}\text{-}co\text{-}PEOMA_{0.8}]_x\text{-}b\text{-}(St_{0.9}\text{-}co\text{-}DPPS_{0.1})_{300}\text{-}b\text{-}(St_{0.9}\text{-}co\text{-}DEGDMA_{0.1})_{150}\text{-}SC(S)SnPr$  CCMs. (a)  $x = 50$ ; (b)  $x = 140$ . All NMR samples were prepared by adding a drop of the reaction mixture directly to the deuterated solvent or solvent mixture in the NMR tube.



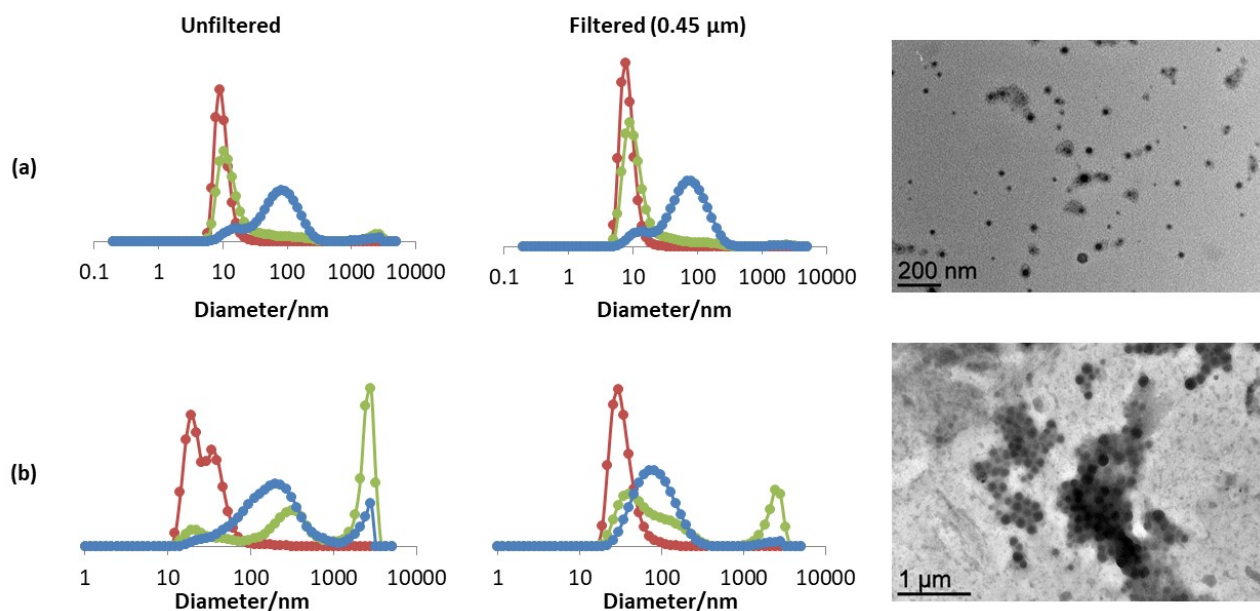


**Figure S 32.** NMR data for the synthesis of the  $\text{R}_0\text{-}[(\text{SS}^-\text{Na}^+)_{0.2}\text{-co-PEOMA}_{0.8}]_{50}\text{-b-(St}_{0.9}\text{-co-DPPS}_{0.1})_{300}\text{-b-DEGDMA}_{90}\text{-SC(S)SnPr}$  CCM. All NMR samples were prepared by adding a drop of the reaction mixture directly to the deuterated solvent or solvent mixture in the NMR tube.

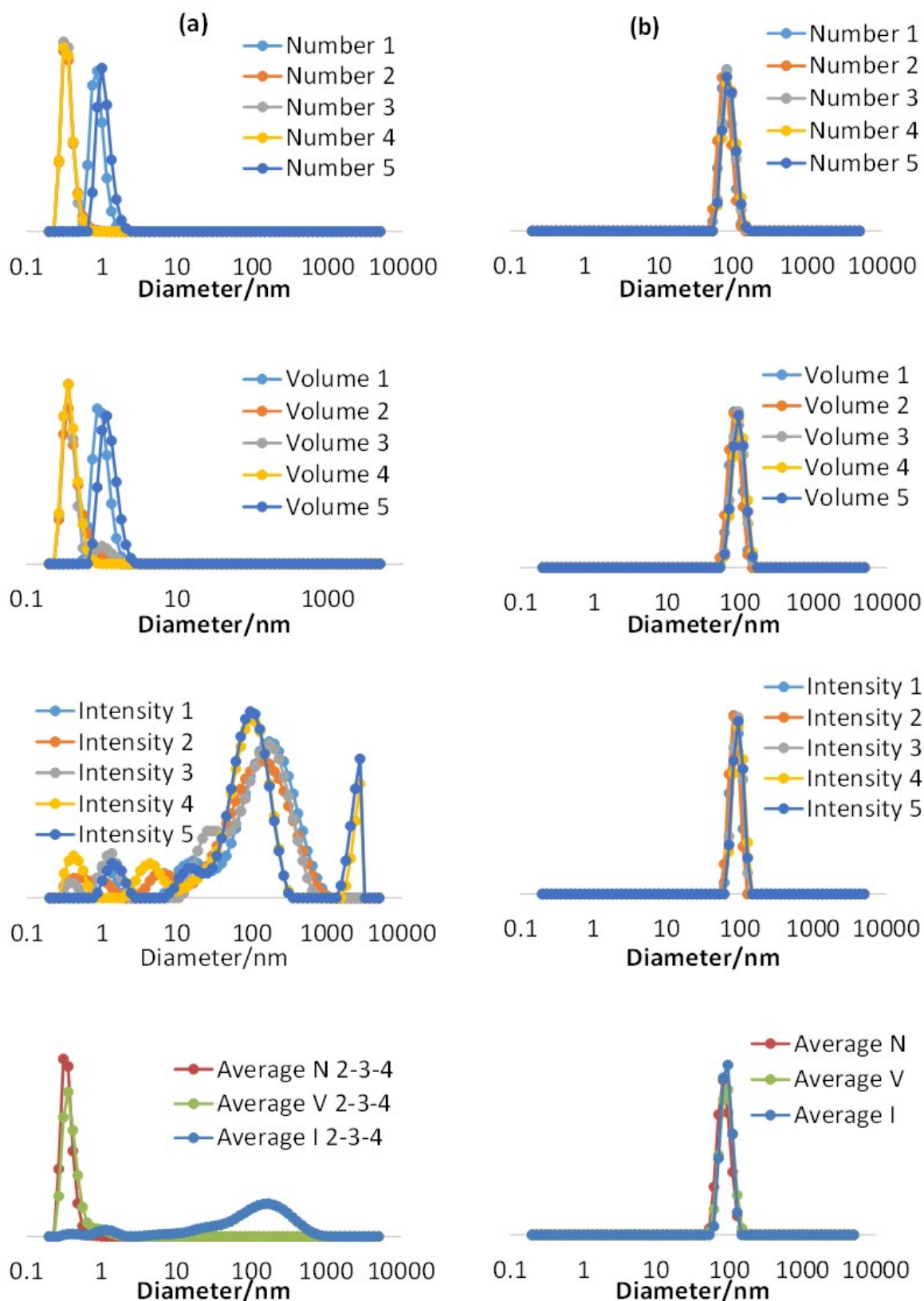




**Figure S 33.** NMR data for the synthesis of the  $R_0-[(SS^{Na^+})_{0.2}\text{-}co\text{-PEOMA}_{0.8}]_{50}\text{-}b\text{-St}_{50}\text{-}b\text{-}(\text{St}_{425}\text{-}co\text{-DPPS}_{30}\text{-}co\text{-DEGDMA}_{15})\text{-SC(S)SnPr}$  NGs. All NMR samples were prepared by adding a drop of the reaction mixture directly to the deuterated solvent or solvent mixture in the NMR tube.



**Figure S 34.** DLS (left) and TEM (right) characterization of the  $R_0-[(SS^{Na^+})_{0.2}\text{-}co\text{-PEOMA}_{0.8}]_{50}\text{-}b\text{-St}_{50}\text{-}b\text{-}(\text{St}_{425}\text{-}co\text{-DPPS}_{30}\text{-}co\text{-DEGDMA}_{15})\text{-SC(S)SnPr}$  NGs. Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).

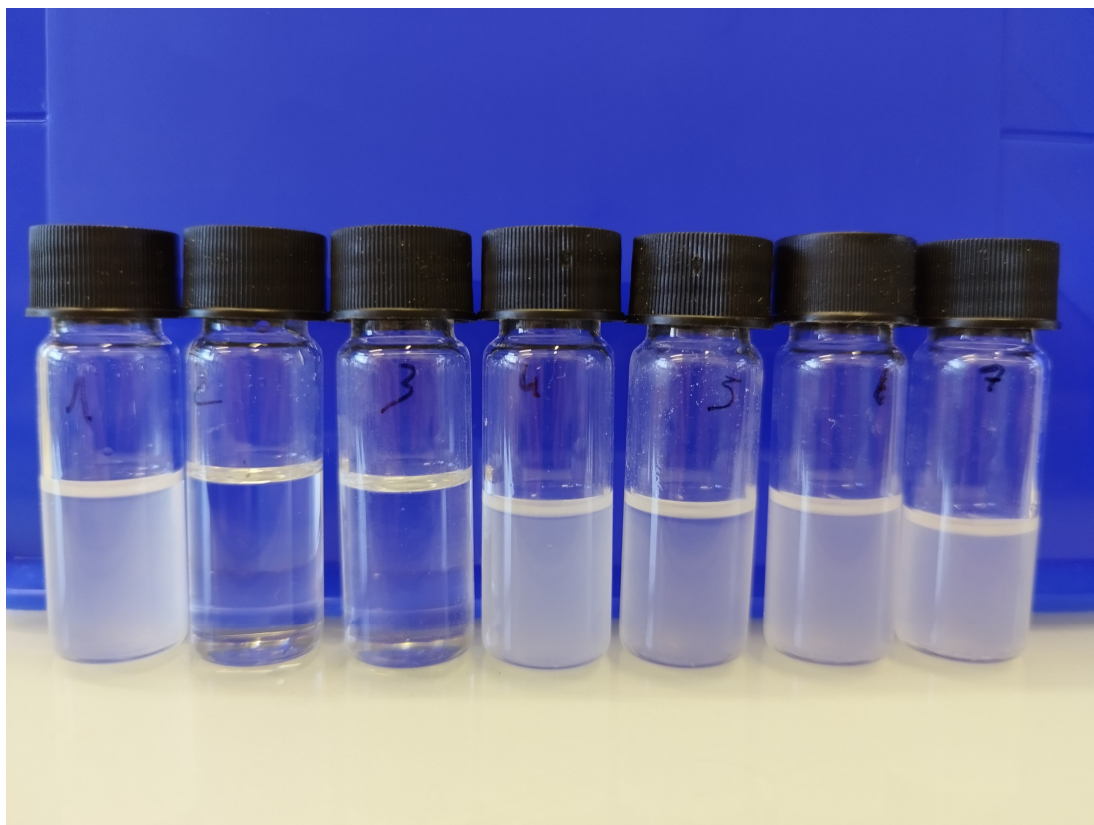


**Figure S 35.** DLS characterization of (a) the  $R_0$ -( $SS\text{-}Na^{+}_{0.2}\text{-co-PEOMA}_{0.8}$ ) $_{50}$ - $b$ -( $St_{0.9}\text{-co-DPPS}_{0.1}$ ) $_{300}$ -SC(S)SPr diblock copolymer and (b) the corresponding  $R_0$ -( $SS\text{-}Na^{+}_{0.2}\text{-co-PEOMA}_{0.8}$ ) $_{50}$ - $b$ -( $St_{0.9}\text{-co-DPPS}_{0.1}$ ) $_{300}$ - $b$ -( $St_{0.9}\text{-co-DEGDMA}_{0.1}$ ) $_{150}$ -SC(S)SPr CCM in a THF-water 60:40 mixed solvent. The dispersions were filtered through a 0.45  $\mu\text{m}$  pore filter and measurements were done in sequence on the same dispersion every 20 seconds.

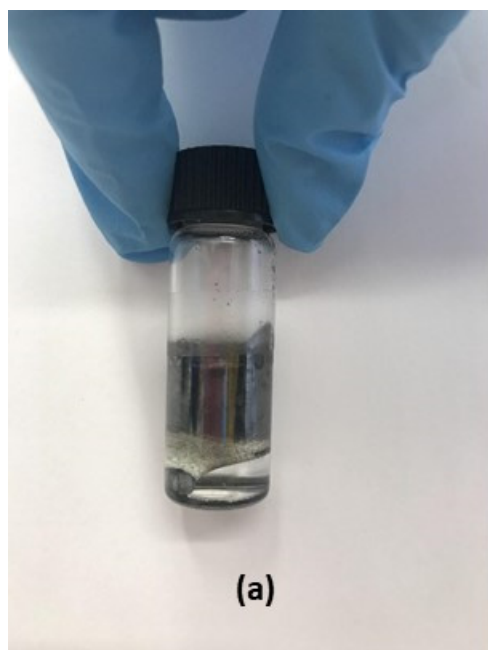
**Table S 3.** Hydrogenation of neat styrene with the [RhCl(COD)(TPP)] precatalyst embedded in CCM nanoreactors.<sup>a</sup>

Additive	Recycle	Conv./%	TON	[Rh]/ppm <sup>b</sup>
R <sub>0</sub> -[(SS·Na <sup>+</sup> ) <sub>0.2-co</sub> -PEOMA <sub>0.8</sub> ] <sub>140</sub> -b-(St <sub>0.9-co</sub> -DPPS <sub>0.1</sub> ) <sub>300</sub> -b-(St <sub>0.9-co</sub> -DEGDMA <sub>0.1</sub> ) <sub>150</sub> -SC(S)SnPr (Table 1, entry 1)	0	29.4	588	6.1
	1	3.52	70	23.2
	2	7.09	142	22.0
	3	6.67	133	26.6
	4	6.12	122	23.7
	5	4.90	98	14.1
	6	8.09	162	13.5
R <sub>0</sub> -[(SS·Na <sup>+</sup> ) <sub>0.2-co</sub> -PEOMA <sub>0.8</sub> ] <sub>50</sub> -b-(St <sub>0.9-co</sub> -DPPS <sub>0.1</sub> ) <sub>300</sub> -b-DEGDMA <sub>90</sub> -SC(S)SnPr (Table 1, entry 2)	0	26.3	526	2.4
	1	14.61	292	18.6
	2	4.56	91	1.5
	3	15.06	301	6.3
	4	14.46	289	24.0
	5	12.88	258	7.2
R <sub>0</sub> -[(SS·Na <sup>+</sup> ) <sub>0.2-co</sub> -PEOMA <sub>0.8</sub> ] <sub>140</sub> -b-St <sub>50</sub> -b-(St <sub>425-co</sub> -DPPS <sub>30-co</sub> -DEGDMA <sub>15</sub> )-SC(S)SnPr (Table 1, entry 3)	0	41.9	837	n.d.
	1	8.72	174	
	2	6.72	134	
	3	19.03	381	
	4	3.47	69	
	5	2.67	53	
	6	3.31	66	

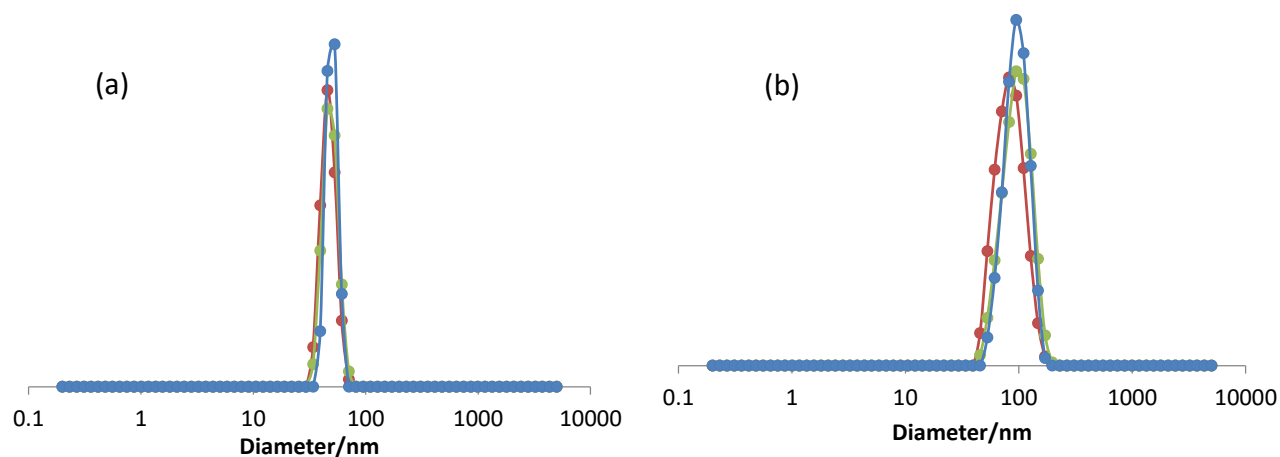
<sup>a</sup> Reaction conditions: styrene/Rh = 2000, P/Rh = 4, T = 25°C, p(H<sub>2</sub>) = 20 bar, stirring rate = 1200 rpm, reaction time: 2.5 h. <sup>b</sup> From the ICP-MS analysis of the recovered organic phase. n. d. = not determined.



**Figure S 36.** Recovered reaction mixtures after the recycling runs of Table 1 (entry 1) and Figure 13 (a).



**Figure S 37.** Recovered reaction mixtures after the catalytic runs of Table 1. (a) Entry 6. (b) Entry 7.



**Figure S 38.** DLS of the recovered organic phase from the recycling experiments of Table 1, measured after dilution in Et<sub>2</sub>O without filtration. (a) First recycle of entry 1. (b) First recycle of entry 2. Color coding for the DLS size distributions: number (red), volume (green) and intensity (blue).