Supplementary Materials for:

Gradient of Segmental Dynamics in Stereoregular Poly(methyl methacrylate) Melts Confined Between Pristine or Oxidized Graphene Sheets

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S1 Force field

The force field parameters for the model graphene-based sheets and for PMMA are provided in Table S1 and Table S2. Both PMMA and graphene force fields exclude non-bonded interactions between the first and second chemically bonded neighbors. The Lorentz–Berthelot combination rule was used for calculating the Van der Waals interactions between dissimilar atoms (including interactions between PMMA and the model graphene-based nanosheets). More details about the force fields and their references are provided in the main text.

Table S1: Force field parameters for pristine and oxidized graphene sheets.

non-bonded, $U_{\rm LJ}(r) =$	$4\varepsilon[(\frac{\sigma}{r})^{12} - (\frac{\sigma}{r})^6] + \text{Coulomb}$	mass (g mol ^{-1})	σ (nm)	$\varepsilon (\mathrm{kJ} \mathrm{mol}^{-1})$	q(e)				
CGR	3		0.3470	0.2750	0				
CEP		12.011	0.3473	0.3979	0.2				
СОН		12.011	0.3473	0.3979	0.265				
OE		15.9994	0.3033	0.4004088	-0.4				
OH		15.9994	0.3033	0.4004088	-0.683				
Н		1.008	0.2846	0.063597	0.418				
bond stretching, Morse $U_{\text{bond}} = D[1 - \exp(\beta(r - b_0))]^2$		$b_0 (nm)$	D (kJ :	mol^{-1}) β (nr	$n^{-1})$				
CGR-CGR	CGR-CGR		478	8.9	21.867				
bond stretching, Harmonic $U_{\text{bond}} = \frac{1}{2}k_{\text{bond}}(r-b_0)^2$		b_0 (nm	$b_0 \text{ (nm)}$		$k_{\rm bond} \ (\rm kJ \ mol^{-1} \ nm^{-2})$				
CEP-CEP		0.153	0.153		292880.0				
CEP-CGR		0.146	0.146		292880.0				
CEP-COH	CEP-COH		0.153		292880.0				
CEP-OE		0.142	0.142		292880.0				
CGR-COH		0.146	0.146		292880.0				
COH-COH		0.153		292880.0					
COH-OH	COH-OH		0.142		292880.0				
H-OH		0.098	0.098		292880.0				
angle bending, $U_{\text{angle}} = \frac{1}{2} k_{\text{angle}} (\cos \theta - \cos \theta_0)^2$		θ_0 (deg	$\theta_0 (\mathrm{deg})$		$k_{\rm angle} \; (\rm kJ \; mol^{-1})$				
CGR-CGR-CGR		120	120		562.2				
-CGR- <i>a</i>		120		557	.9				
-CEP-		109.47	109.471		470.7				
-COH-		109.47	109.471		470.7				
CEP-OE-CEP		104.51	104.51		446.4				
СОН-ОН-Н	ОН-ОН-Н		104.51		446.4				
dihedral angle potential, $U(\phi) = k(1 + \cos(n\phi - \phi_0))$									
	$\phi_0 ({ m deg})$	$k \; (kJ \; mol^{-1})$	n						
-CGR-CGR-	180	12.56	2						
-CEP-CEP-	0	0.465	3						
-CEP-CGR-	180	0.349	6						
-CEP-COH-	0	0.465	3						
-CEP-OE-	0	1.39	3						
-CGR-COH-	180	0.349	6						
-COH-COH-	0	0.465	3						
-COH-OH-	0	1.395	3						

CGR: graphene sp² carbon atom. CEP: epoxide-group carbon atom. COH: hydroxyl-group carbon atom. OE: epoxide-group oxygen atom. OH: hydroxyl-group oxygen atom. H: hydroxyl-group hydrogen atom. a: except CGR-CGR

non-bonded, $U_{\rm LJ}(r)$	$= 4\varepsilon[(\frac{\sigma}{r})^{12} - (\frac{\sigma}{r})^6] +$	- Coulomb	mass (g mol ^{-1})	σ (nm)	$\varepsilon (\mathrm{kJ} \mathrm{mol}^{-1})$	q(e)			
CX (-CH ₃ and -CH ₂	-)		12.011	0.339967	0.457730	0.0051			
CX (α carbon)			12.011	0.339967	0.457730	0.0189			
CX (ester-methyl)			12.011	0.339967	0.457730	0.2801			
HC			1.008	0.2649533	0.065689	0			
С			12.011	0.339967	0.359824	0.7464			
0			15.9994	0.2959922	0.878640	-0.5939			
OS			15.999	0.3000012	0.711280	-0.4617			
H1			1.008	0.2471353	0.065689	0			
bond stretching, $U_{\text{bond}} = \frac{1}{2}k_{\text{bond}}(r-r_0)^2$		$r_0 (\text{nm})$	$r_0 (\text{nm})$		$k_{\rm bond} \; (\rm kJ \; mol^{-1} \; nm^{-2})$				
CX-CX		0.152		259408.0					
HC-CX			0.1095		301248.0				
CX-C			0.15		276144.0				
C-O	C-O		0.1204	0.1204		835963.2			
C-OS	C-OS		0.1343	0.1343		344175.9			
OS-CX	DS-CX		0.141	0.141		267776.0			
CX-H1	ζ-H1		0.1092	0.1092		343088.0			
angle bending, $U_{\text{angle}} = \frac{1}{2} k_{\text{angle}} (\theta - \theta_0)^2$		$\theta_0 \ (deg)$	$\theta_0 \; (\text{deg})$		$k_{\text{angle}} \text{ (kJ mol}^{-1} \text{ rad}^{-2})$				
HC-CX-HC			109.5		334.7				
HC-CX-CX	HC-CX-CX		112.6		376.6				
CX-CX-CX	CX-CX-CX		113.5		376.6				
CX-CX-C	CX-CX-C		111.5		334.7				
CX-C-O		125.4		493.7					
CX-C-OS		111.0		418.4					
C-OS-CX	C-OS-CX		114.0		423.4				
O-C-OS			122.5	122.5		861.9			
OS-CX-H1			110.0		502.1				
H1-CX-H1			109.5		376.6				
dihedral angle potential, $U(\phi) = \sum_{n=1}^{3} k_n (1 + \cos n\phi)$									
	$k_1 \; (\text{kJ mol}^{-1})$	k_2	k_{3}	3					
CX-CX-CX-CX	1.8828	0	0						
CX-CX-CX-HC	0	0	0.41	0.4184					
CX-CX-CX-C	0	0	0	0					
HC-CX-CX-C	0	0	-0.12	-0.12552					
CX-CX-C-O	0	-1.715	-1.2134						
CX-CX-C-OS	0	-1.882	8 -1.3	807					
O-C-OS-CX	0	-17.614	64 -3.05	5432					
H1-CX-OS-C	0	0	0.62	276					
CX-C-OS-CX	-2.7196	-11.50	6 0						

CX: sp^3 carbon atom. HC: hydrogen (except ester group hydrogen). C: sp^2 carbon atom. O: carbonyl-group oxygen. OS: ester-group oxygen. H1: ester-side-group hydrogen.

S2 Layer resolved TACF curves

The layer resolved torsional autocorrelation function (TACF) curves of the confined i-PMMA systems at T = 550 K are provided in Figure S1 and the curves for some of the model confined s-PMMA systems are provided in Figure S1.



Figure S1: Layer resolved torsional autocorrelation function (TACF) curves for (a) i-PMMA/PG, (b) i-PMMA/RGO, and (c) i-PMMA/GO at T = 550 K.



Figure S2: Layer resolved torsional autocorrelation function (TACF) curves for (a) s-PMMA/PG, (b) s-PMMA/RGO, (c) s-PMMA/GO at T = 580 K, and (d) s-PMMA/GO at T = 520 K.

S3 Relaxation times of the confined s-PMMA systems

The layer resolved relaxation times of the s-PMMA/PG, s-PMMA/RGO. and s-PMMA/GO systems at different temperatures are provided in Figure S3. The relaxation times are normalized with the respective bulk values.



Figure S3: (a), (b), and (c) show the layer resolved values of the normalized relaxation times $(\frac{\tau_{\text{seg}}(d)}{\tau_{\text{seg}}^{\text{bulk}}})$ for the s-PMMA/PG, s-PMMA/RGO, and s-PMMA/GO interfacial systems, at different temperatures, respectively. The error bars at T = 580 K have been estimated based on block averaging; at lower temperatures larger error bars are expected.