



Electronic Supplement Information (ESI) for

## Preparations of Tough and Conductive PAMPS/PAA Double Network Hydrogels Containing Cellulose Nanofibers and Polypyrroles

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## **Captions:**

Table S1. Summary of tensile testing results of T–Py–DN samples.

- **Fig. S1.** (a) UV–Vis spectra of Fe(III)EDTA solutions with different concentrations and (b) linear fitting of the UV–Vis absorbance at 260 nm versus the concentration of Fe(III)EDTA aqueous.
- **Fig. S2.** SEM images of (a) original cellulose and (b) TOCN. The counts of fiber numbers versus the width of (c) cellulose fibers and (d) nanofibrils from the SEM images of (a) and (b), respectively.
- Fig. S3. UV–Vis spectra of the Fe(III)EDTA solutions extracted from T<sub>2</sub>–(Fe–)DN hydrogels.
- **Fig. S4.** Apparent color changes of various T–Fe–DN hydrogels after (a) immersing in various FeCl<sub>3(aq)</sub> with different time and (b) removing ferric ions with EDTA solutions.

## 1. Estimation of the resistance change of hydrogels.

Assuming constant resistivity and volume for T<sub>2</sub>–Py<sub>5</sub>–DN hydrogel, a relationship between  $\Delta R$  (%) and the strain ( $\varepsilon$  in %) can be derived for the case of ideal homogeneous deformation of the sample. Because the total volume is constant during stretching, the value of cross-section area (A) multiples length (l) of the hydrogel specimen is constant (i.e.  $l \times A$ =  $l_0 \times A_0$ ). Where  $l_0$  and  $A_0$  are the original length and cross-section area of the hydrogel specimen. The resistance change could be derived to the fallowing equation:

$$\Delta R \ (\%) = \left(\frac{l^2}{{l_0}^2} - 1\right) \times 100\%$$

Replace  $l/l_0$  by  $(1+\varepsilon/100)$  and the resistance change can further derive to:

$$\Delta R (\%) = \frac{\varepsilon^2}{100} + 2\varepsilon$$

Thus, the resistance change was estimated and expected to be observed at strain ratio of 0–25–50 %, respectively. For the first elongation cycle at strain ratio of 25 and 50 %, slightly deviations around 46 and 86% of resistance change were measured. However, the resistance changes at 0–25–50 % strain ratio slightly increase at 2–5 cycles. The final electrical resistance changes during releasing at 5<sup>th</sup> cycle are around 111, 56, and 6 % relevant to the strain ratio of 50–25–0 %, respectively, which were close to the abovementioned estimated results of resistance change. The data of electrical resistance changes over elongation cycles indicated the hydrogel was isotopic.

Sample	Stress	Strain	Modulus	Toughness*
	(kPa)	(%)	(MPa)	(kJ/m³)
T <sub>0</sub> –DN	310.6	131.3	0.93	248.4
T2-Py5-DN	451.1	106.7	0.96	249.1
T2-Py5(16h)-DN	354.4	67.8	0.96	112.9
To-Py5-DN	386.1	85.8	1.15	170.2
T1-Py5-DN	239.6	83.6	0.68	104.8
T3-Py5-DN	434.6	76.0	1.34	145.8
T2-Py50-DN	418.9	86.6	0.96	172.8
T2-Py100-DN	295.2	54.1	0.98	63.8
T2-Py200-DN	295.6	54.0	0.94	69.9

Table S1. Summary of tensile testing results of T–Py–DN samples.

\* Toughness was estimated from the area of the obtaining stress-strain curve.



**Figure S1.** (a) UV–Vis spectra of Fe(III)EDTA solutions with different concentrations and (b) linear fitting of the UV–Vis absorbance at 260 nm versus the concentration of Fe(III)EDTA aqueous.



**Figure S2.** SEM images of (a) original cellulose and (b) TOCN. The counts of fiber numbers versus the width of (c) cellulose fibers and (d) nanofibrils from the SEM images of (a) and (b), respectively.



Figure S3. UV–Vis spectra of the Fe(III)EDTA solutions extracted from  $T_2$ –(Fe–)DN hydrogels.



**Figure S4.** Apparent color changes of various T–Fe–DN hydrogels after (a) immersing in various FeCl<sub>3(aq)</sub> with different time and (b) removing ferric ions with EDTA solutions.



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