

Supporting Information

# Synthesis, Crystal Structures and Spectroscopic Characterization of *bis*-aldehyde Monomers and Their Electrically Conductive Pristine Polyazomethines

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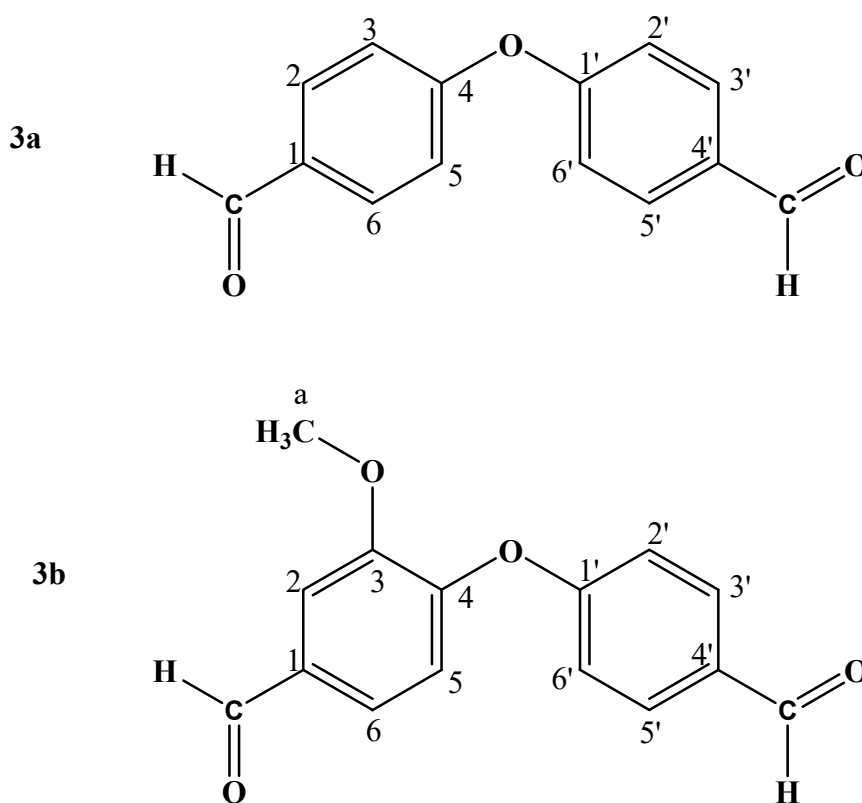
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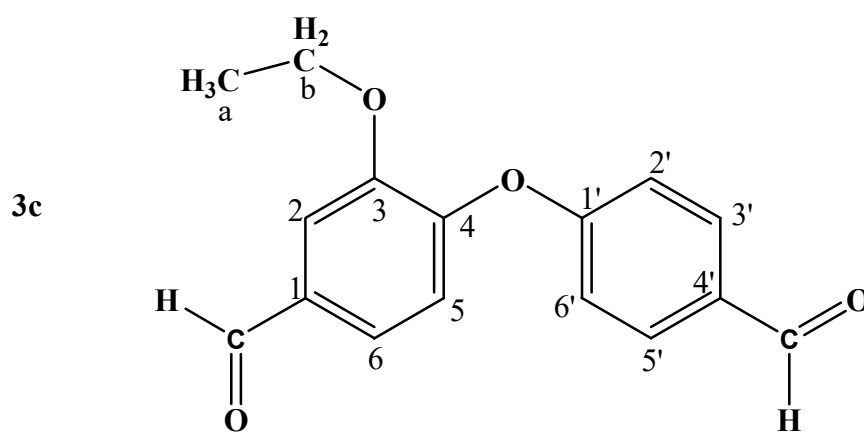


Figure S1. Labelling of monomers.

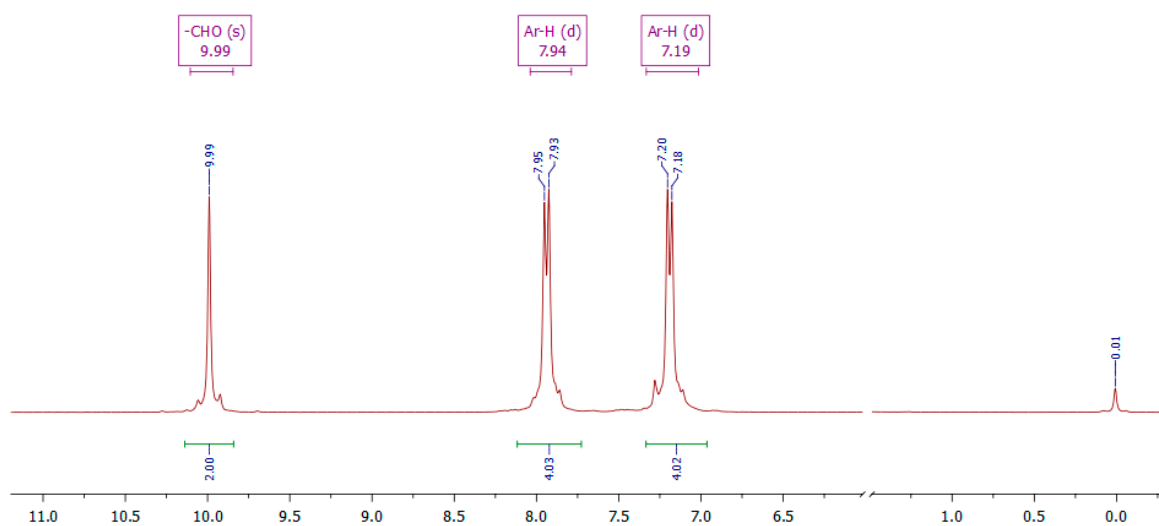


Figure S2. Proton NMR of 3a monomer.

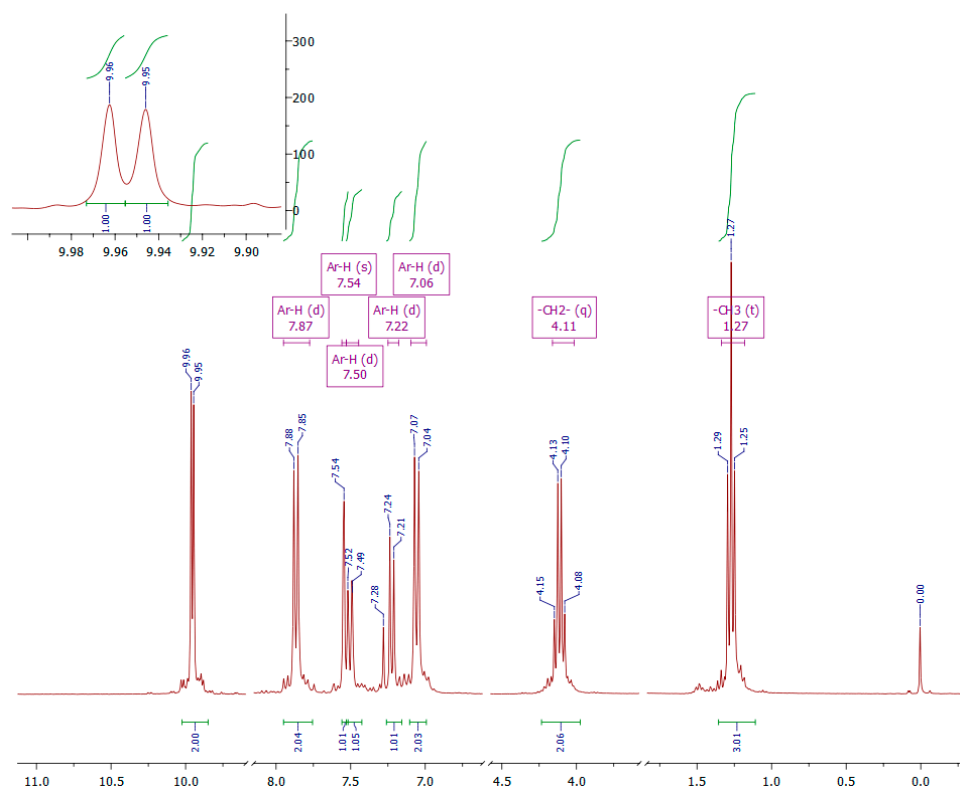


Figure S3. Proton NMR of 3c monomer.

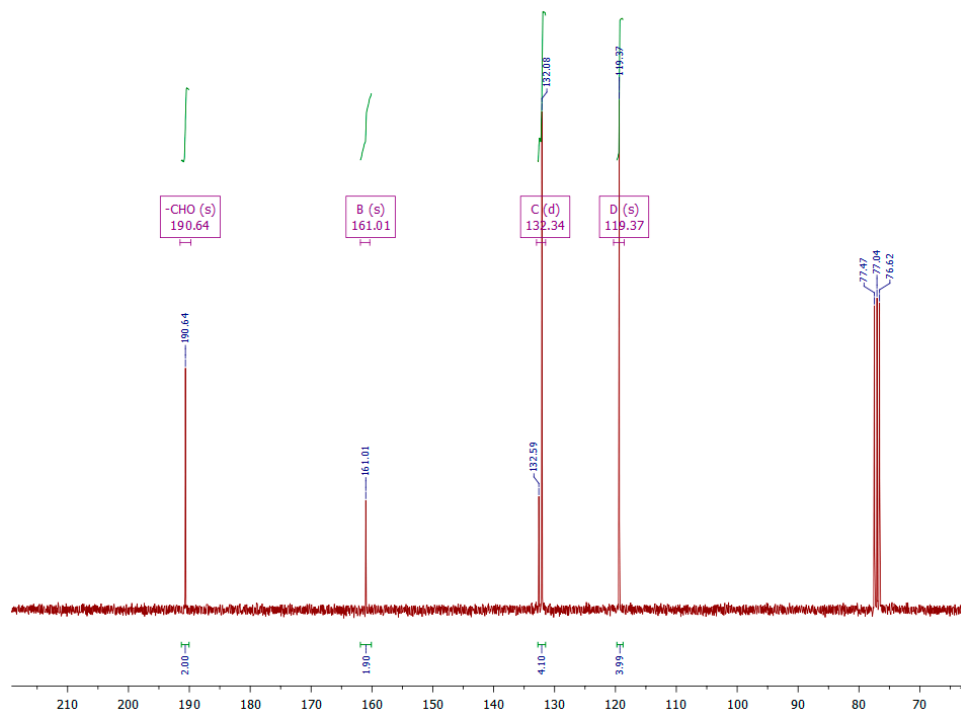


Figure S4. Carbon-13 NMR of 3a monomer.

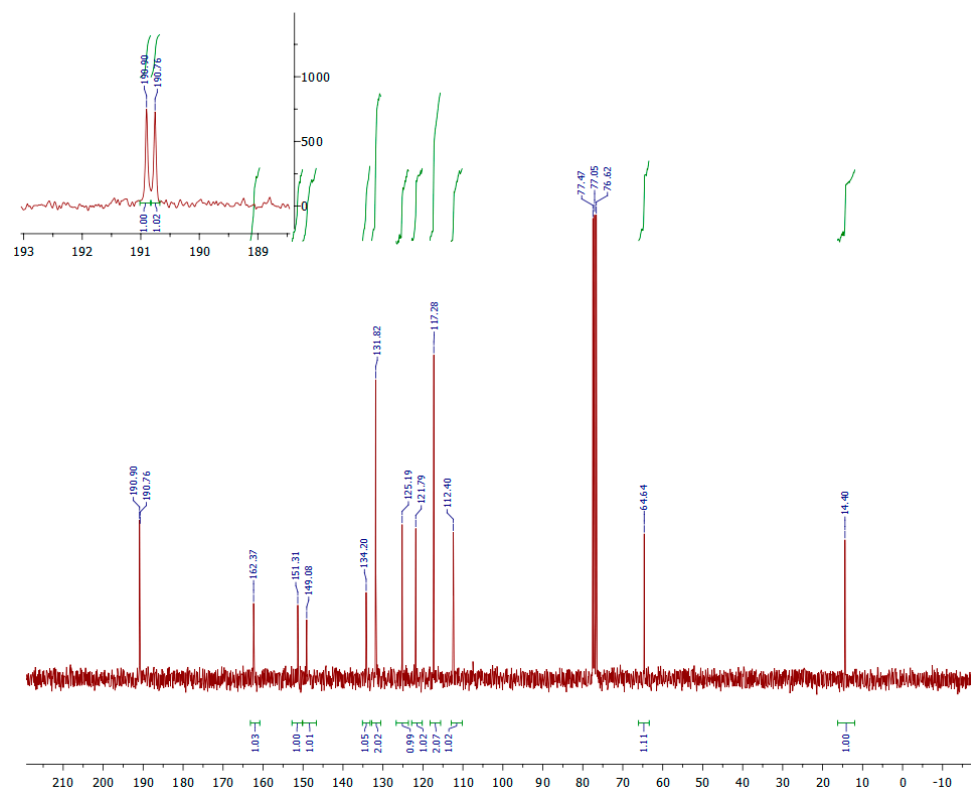


Figure S5. Carbon-13 NMR of 3c monomer.

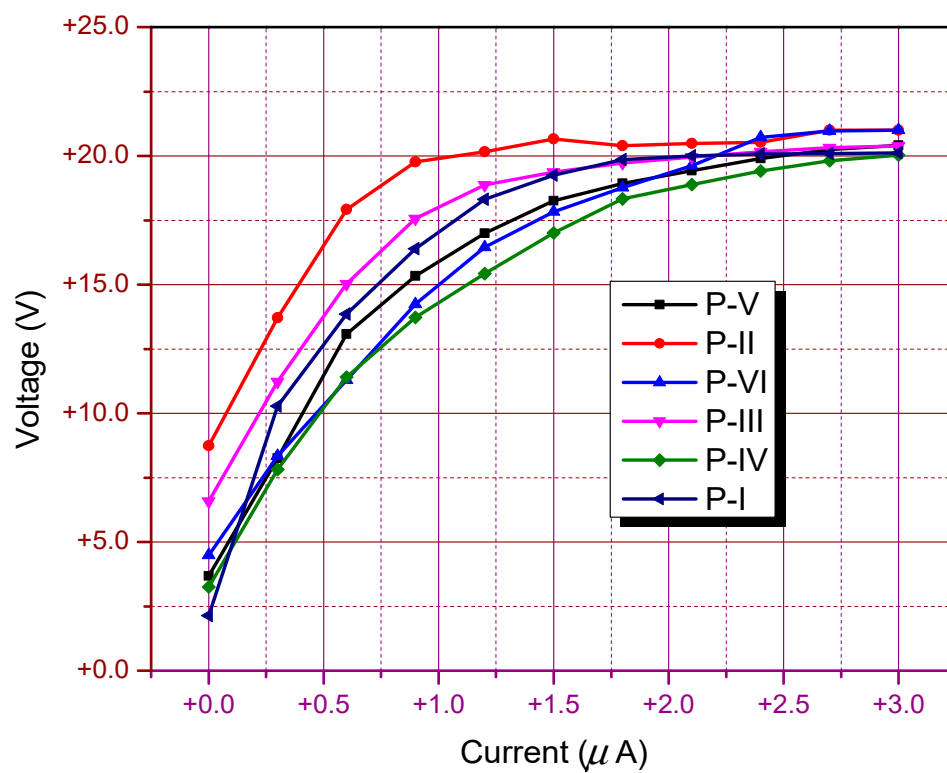


Figure S6. current vs voltage (I-V) curves of polyazomethines polymers.

### Organosolubility and inherent viscosities

The organosolubility of polymeric azomethines was checked at 30 °C in various solvents such as *n*-hexane, DMF, DMSO, NMP, CHCl<sub>3</sub> and sulfuric acid by dissolving 10 mg powdered polymer sample in 1 mL of solvent and the results are presented in Table S1. It was found that the polymers synthesized from non-substituted aromatic dialdehyde (*i.e.* 3a) were almost insoluble in most of the solvents whereas others having –OCH<sub>3</sub>, and –OC<sub>2</sub>H<sub>5</sub> showed reasonable solubility in all the solvents. The limited solubility of wholly aromatic polyazomethines may be attributed to the chain rigidity, regularity and inter-chain packing efficiency. It may also be noted that the polymer **P-I** and **P-IV** showed semi-crystalline characteristics in powder XRD which reinforces the fact that polymer chains stacked over one another. However, all the polymeric azomethines showed good solubility in *conc.* H<sub>2</sub>SO<sub>4</sub> at room temperature and Table S1 shows the respective solubility behavior of polyazomethines in aforementioned solvents. The polymers had viscosity in the range 1.98–1.72, as shown in Table S2.

**Table S1.** Organosolubility of polyazomethines in common organic solvents.

Polymer	<i>n</i> -hexane	DMF	DMSO	NMP	CHCl <sub>3</sub>	H <sub>2</sub> SO <sub>4</sub>
<b>P-I</b>	--h	--h	--h	--h	--h	+++
<b>P-II</b>	--h	+++	++h	++h	+++	+++
<b>P-III</b>	--h	+++	+++	+++	+++	+++
<b>P-IV</b>	--h	--h	--h	--h	--h	+++
<b>P-V</b>	--h	+–h	+–h	+–h	+–h	+++
<b>P-VI</b>	--h	++h	++h	+–h	+++	+++

+++ = soluble at room temperature, +–h = partially soluble on heating

--h = insoluble on heating, ++h = soluble on heating

**Table S2.** Viscometric Data of polyazomethine Polymers.

Code	$\eta_{rel}$	$\eta_{sp}$	$\eta_{red}$ (dl/g)	$\eta_{inh}$ (dl/g)
<b>P-I</b>	1.17	0.17	0.86	1.77
<b>P-II</b>	1.25	0.25	1.28	1.84
<b>P-III</b>	1.12	0.12	0.60	1.72
<b>P-IV</b>	1.136	0.14	0.68	1.74
<b>P-V</b>	1.45	0.45	2.25	1.98
<b>P-VI</b>	1.18	0.18	0.94	1.78