

Electronic Supporting Information

Functional Poly(dihalopentadiene)s: Stereoselective Synthesis, Aggregation- Enhanced Emission and Sensitive Detection of Explosives

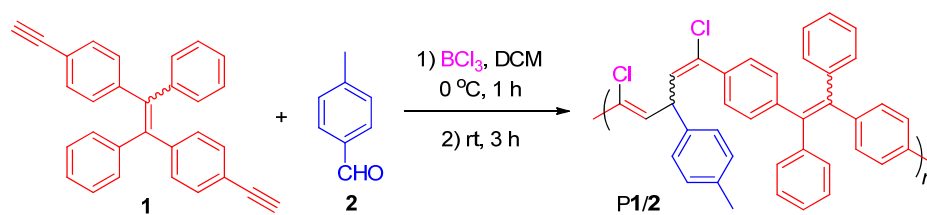
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Scheme S1. Synthetic route of P1/2.

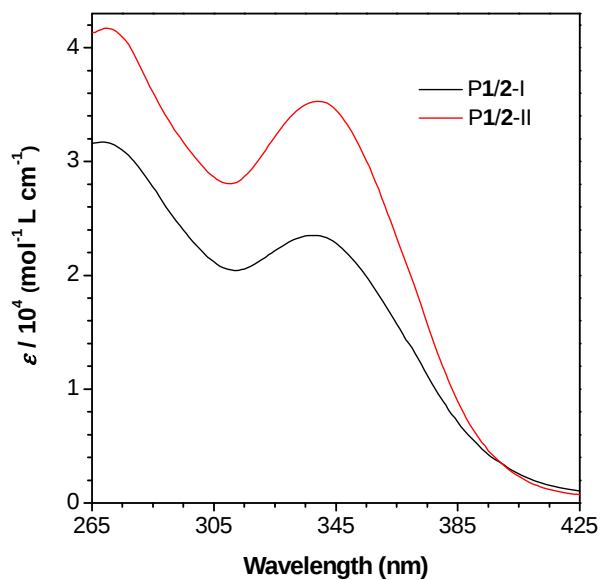


Figure S1. Absorption spectra of P1/2-I and P1/2-II in THF solutions. Solution concentration: 10 μM .

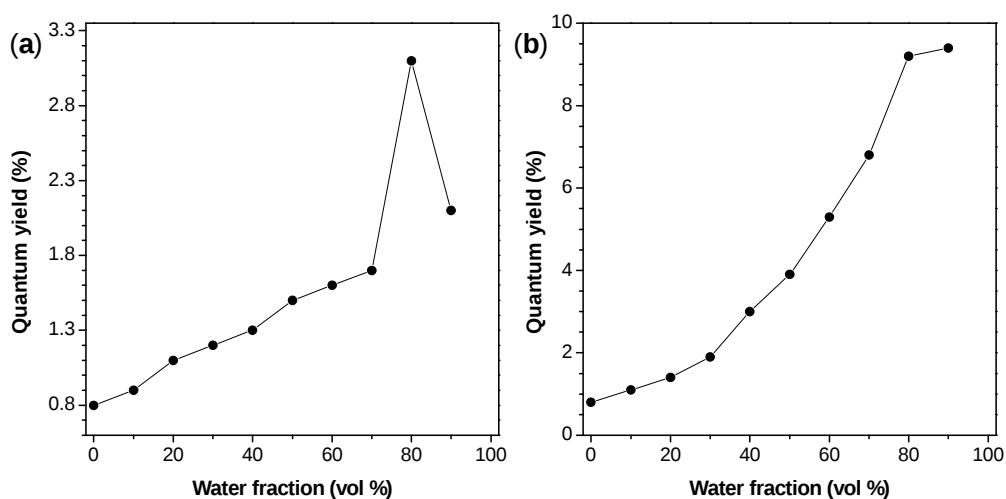


Figure S2. Plot of fluorescence quantum yield of (a) P1/2-I and (b) P1/2-II versus water fraction in THF/water mixtures. Solution concentration: 10 μM ; excitation wavelength: 350 nm.

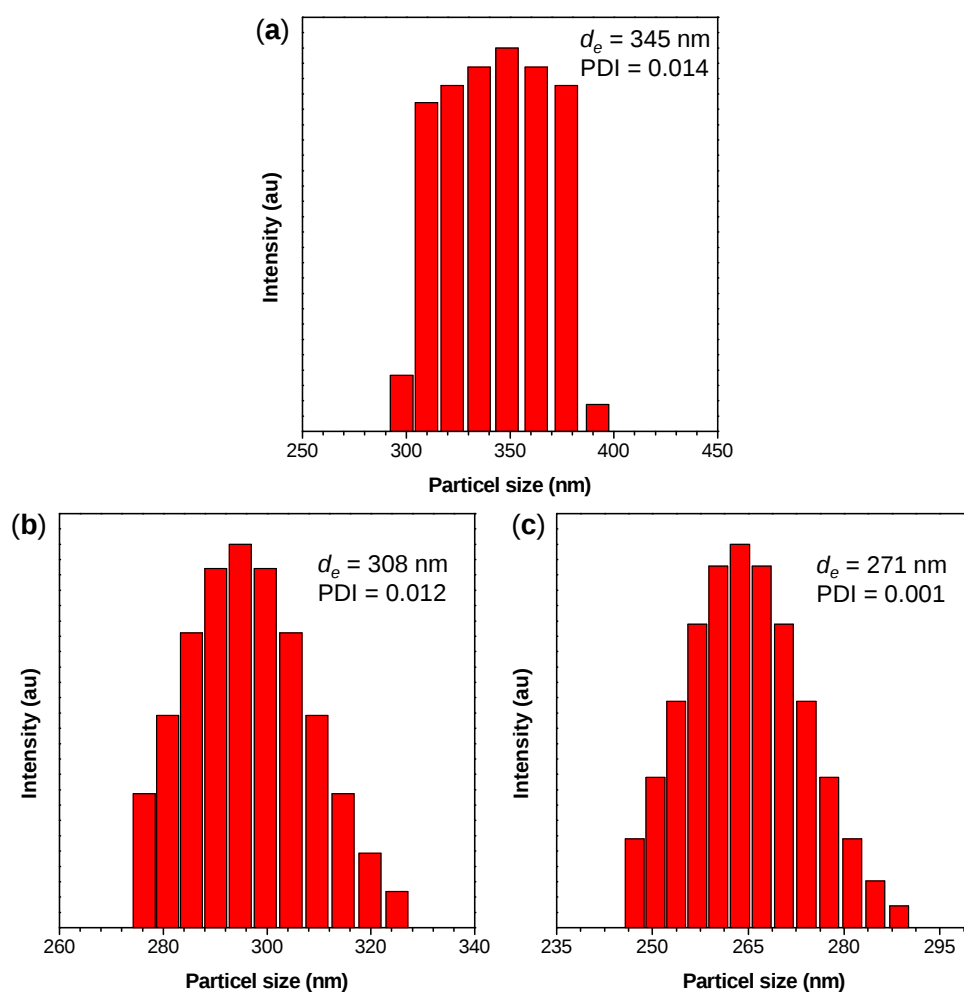


Figure S3. Particle size distributions of the nanoparticles of P1/2-I in THF/water mixtures with (a) 70%, (b) 80%, and (c) 90% water fraction measured by dynamic light scattering. Abbreviation: d_e = effective diameter, PDI = polydispersity index.

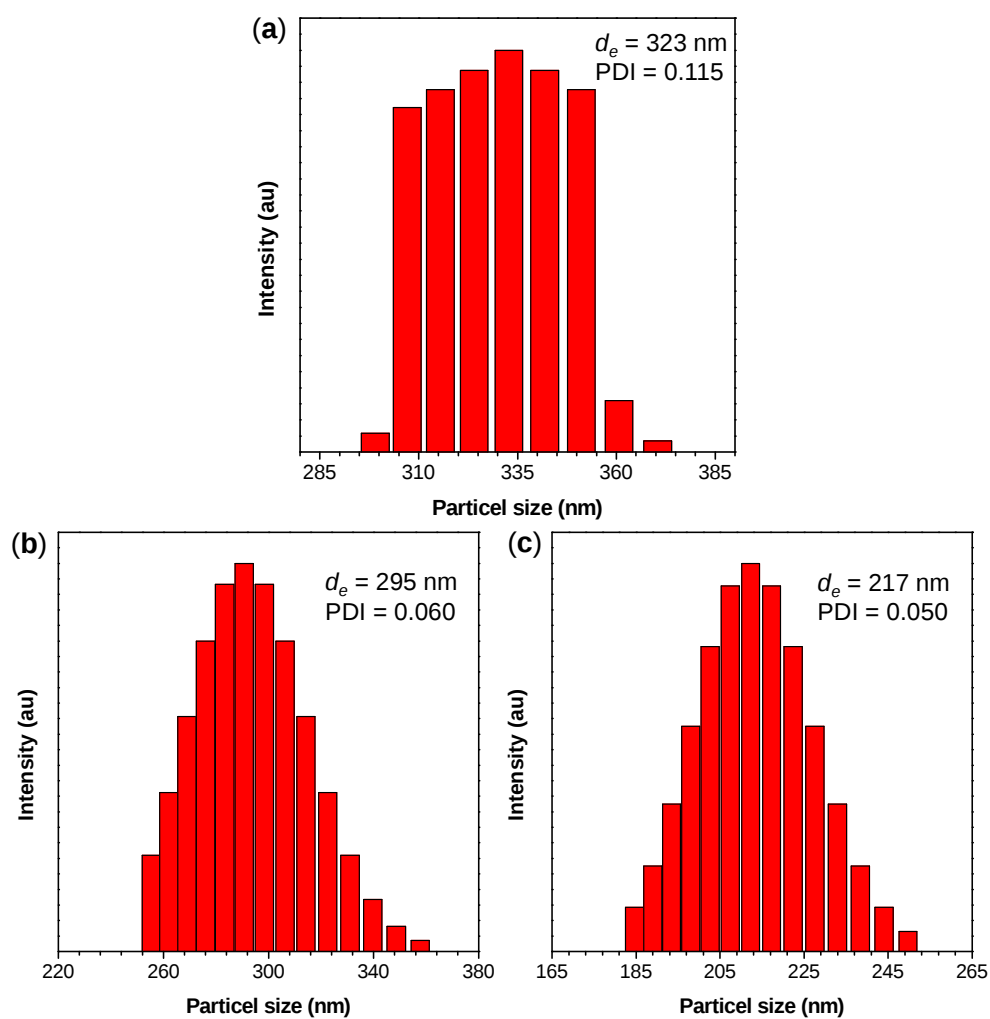


Figure S4. Particle size distributions of the nanoparticles of P1/2-II in THF/water mixtures with (a) 70%, (b) 80%, and (c) 90% water fraction measured by dynamic light scattering. Abbreviation: d_e = effective diameter, PDI = polydispersity index.

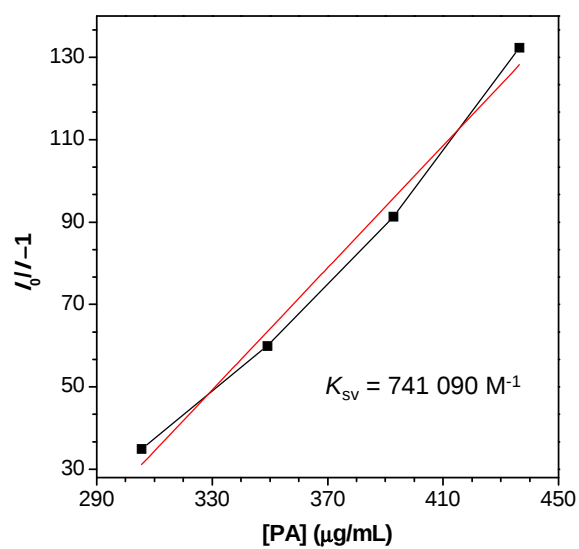


Figure S5. Fitting curve of P1/2-I for the calculation of the quenching constant.

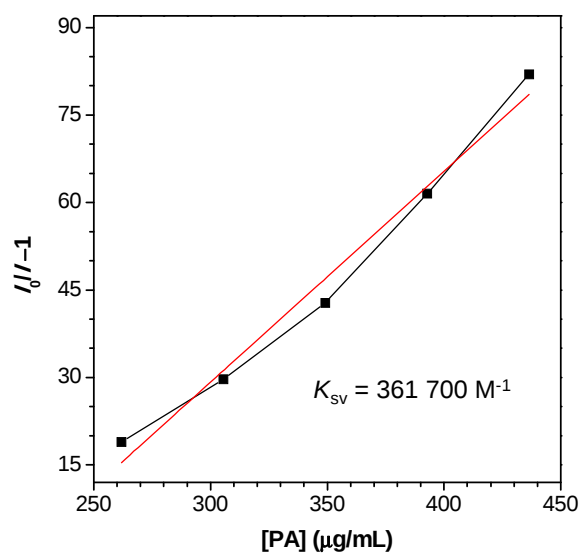


Figure S6. Fitting curve of P1/2-II for the calculation of the quenching constant.

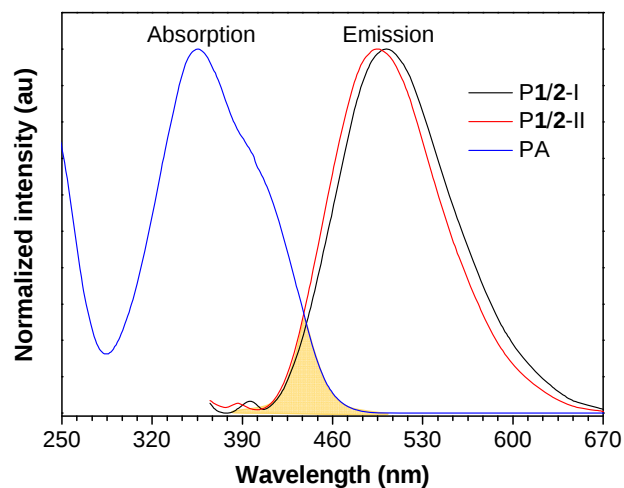


Figure S7. Normalized absorption spectrum of PA in water and PL spectra of P1/2-I and P1/2-II nanoaggregates in THF/water mixtures with 80% water fraction.

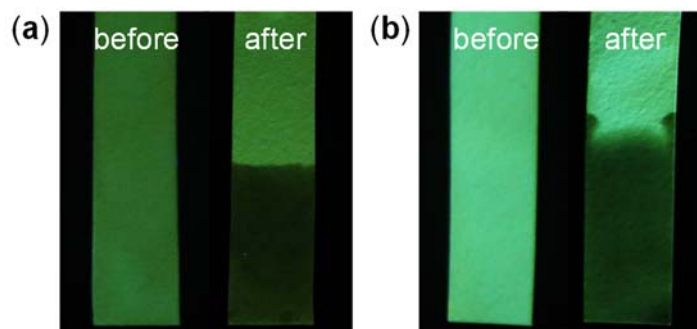


Figure S8. Fluorescence images of test strips coated with (a) P1/2-I and (b) P1/2-II before and after being partially dipped into an aqueous PA solution.