

Influence of Selectively Localised Nanoclay Particles on Non-Isothermal Crystallisation and Degradation Behaviour of PP/LDPE Blend Composites

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Table S1. Amount of polymers, compatibilizers and clay present in the extruded samples.

Sample	Mass of PP (g)	Mass of LDPE (g)	Mass of PE-g-MA (g)	Mass of PP-g-MA (g)	Mass of Clay (g)	Mass of PP (wt %)	Mass of LDPE (wt %)	Mass of PE-g-MA (wt %)	Mass of PP-g-MA (wt %)	Mass of Clay (wt %)	Mass Fraction of PP	Mass Fraction of LDPE	TOTAL Mass Fraction of PP in the Sample	TOTAL Mass Fraction of LDPE in the Sample
100/0/0/0/0	2000	0	0	0	0	100	0	0	0	0	1	0	1	0
96/0/0/0/4	1920	0	0	0	80	96	0	0	0	4	0.96	0	0.96	0
0/100/0/0/0	0	2000	0	0	0	0	100	0	0	0	0	1	0	1
0/96/0/0/4	0	1920	0	0	80	0	96	0	0	4	0	0.96	0	0.96
80/20/0/0/0	1600	400	0	0	0	80	20	0	0	0	0.8	0.2	0.8	0.2
80/20/0/0/4	1536	384	0	0	80	76.8	19.2	0	0	4	0.768	0.192	0.768	0.192
80/20/5/0/4	1456	364	0	100	80	72.8	18.2	0	5	4	0.728	0.182	0.778	0.182
80/20/0/5/4	1456	364	100	0	80	72.8	18.2	5	0	4	0.728	0.182	0.728	0.232
80/20/5/5/4	1376	344	100	100	80	68.8	17.2	5	5	4	0.688	0.172	0.738	0.222

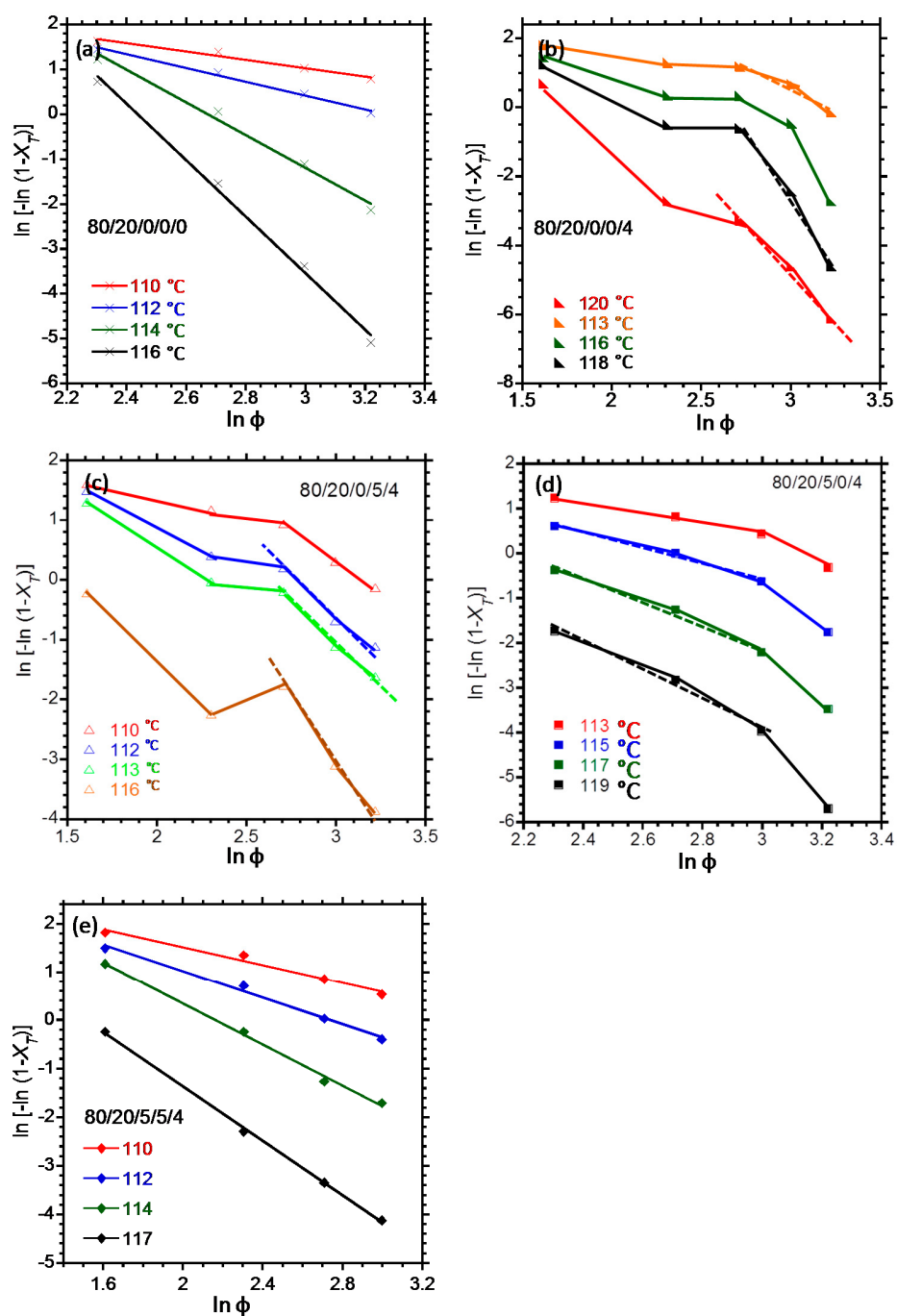


Figure S1. $\ln[-\ln(1 - X_T)]$ versus $\ln \phi$ plots for: (a) neat blend and (b–e) blend composites.

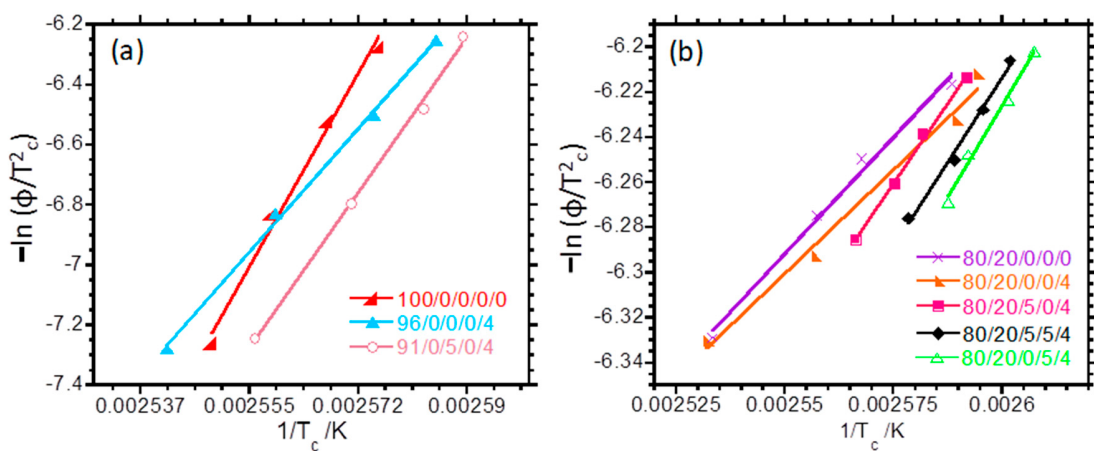


Figure S2. Determination of the activation energy, ΔE describing the nonisothermal crystallization process of PP in neat PP, blend, PP containing binary and ternary composites.

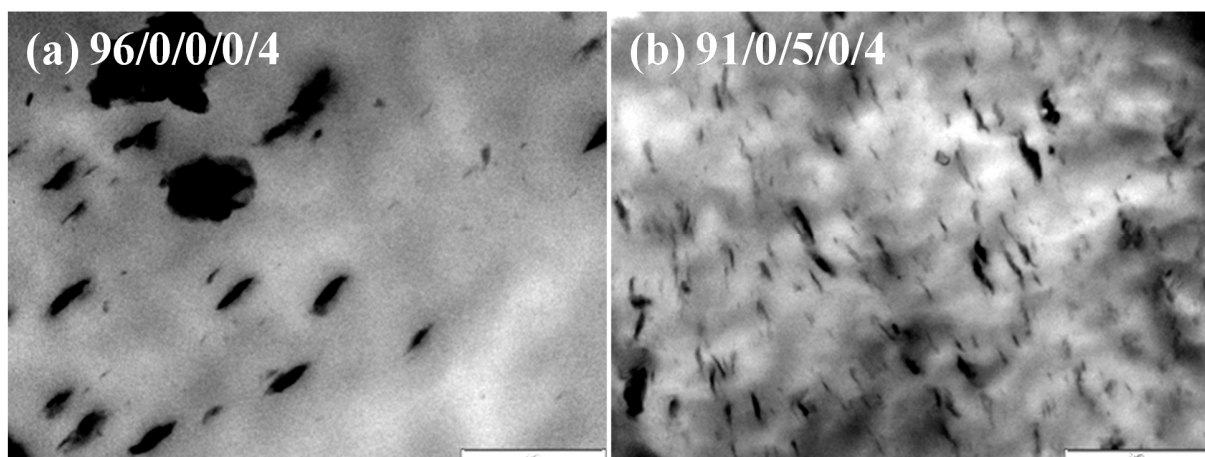


Figure S3. TEM micrographs of binary PP/LDPE/PP-g-MA/PE-g-MA/clay composites.

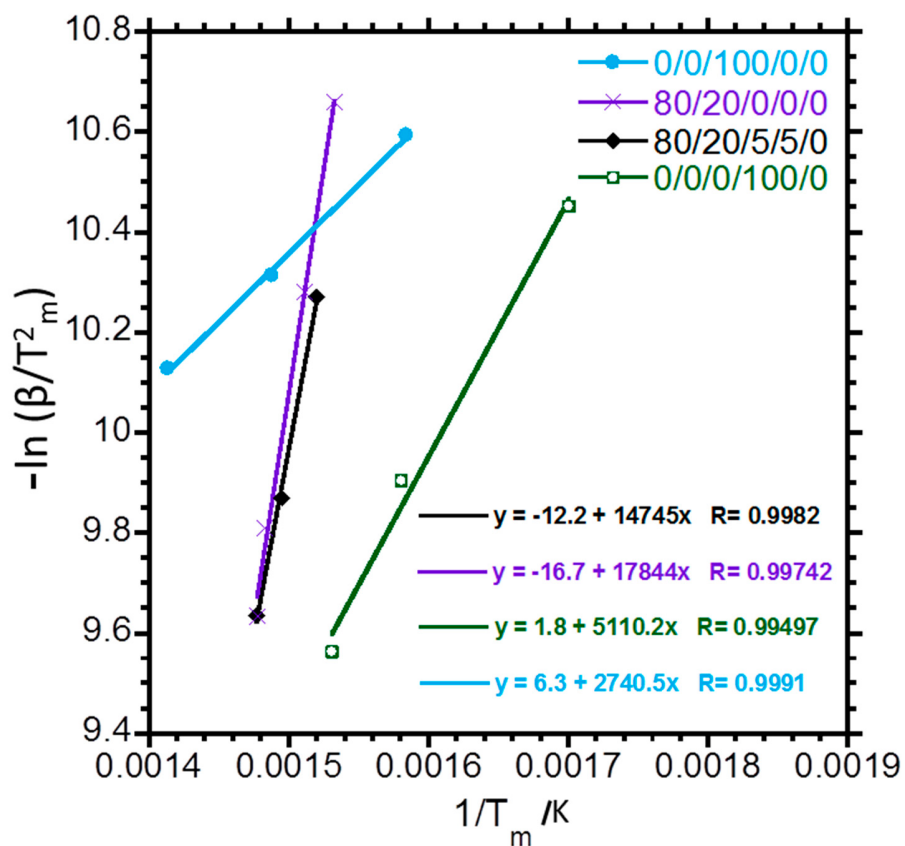


Figure S4. Determination of the activation energy, E_a describing the thermal degradation process of the neat blend, PP-g-MA, PE-g-MA, PP-g-MA and PE-g-MA containing blend.

Table S2. Activation energy for the overall non-isothermal crystallization of PP-g-MA, PE-g-MA, PP/LDPE, and PP/LDPE/ PP-g-MA/PE-g-MA blend.

PP/LDPE/PP-g-MA/PE-g-MA/clay	Activation energy (E_a)/kJ·mol ⁻¹
80/20/0/0/0	148.4
0/0/100/0/0	22.8
0/0/0/100/0	42.5
80/20/5/5/0	122.6