

SUPPORTING INFORMATION

Table S1. Crystallographic data and refinement parameters for crystals **1–6**

Parameter	1 · 2H ₂ O	2 · 2H ₂ O	3 · 2CH ₃ CN	4	5 · 2CH ₃ CN	6 · 0.5H ₂ O
Empirical formula	C ₃₀ H ₅₅ B ₂₀ N ₉ O ₂	C ₂₄ H ₃₄ B ₁₂ N ₄ O ₂	C ₂₀ H ₄₂ B ₁₀ N ₆	C ₂₄ H ₂₈ B ₁₀ N ₄	C ₆₀ H ₈₀ B ₁₂ N ₆ O ₆	C ₅₄ H _{48.5} B ₁₀ N ₉ O _{0.5}
Fw	790.30	540.27	474.69	480.60	1111.02	940.62
Crystal system, space group	Triclinic, <i>P</i> $\bar{1}$	Monoclinic, <i>P</i> 2 ₁ / <i>m</i>	Orthorhombic, <i>Pna</i> 2 ₁	Monoclinic, <i>P</i> 2 ₁ / <i>n</i>	Monoclinic, <i>C</i> 2/ <i>c</i>	Orthorhombic, <i>P</i> 2 ₁ 2 ₁ 2
T (K)	173(2)	120.0(2)	173.0(2)	100.0(2)	120.0(2)	100.0(2)
a (Å)	7.437(4)	7.9096(3)	15.893(2)	20.430(16)	24.465(3)	16.586(2)
b (Å)	17.777(10)	18.9027(6)	8.7035(13)	8.611(7)	14.7934(17)	25.190(3)
c (Å)	17.866(9)	9.8240(4)	20.298(3)	30.41(2)	17.787(2)	11.5646(14)
α (°)	115.710(8)	90	90	90	90	90
β (°)	90.072(9)	108.169(1)	90	109.200(14)	108.284(2)	90
γ (°)	93.744(9)	90	90	90	90	90
V (Å ³)	2123(2)	1395.58(9)	2807.8(7)	5052(7)	6112.5(12)	4831.7(10)
Z	2	2	4	8	4	4
μ (cm ⁻¹)	0.071	0.570	0.063	0.069	0.074	0.075
No. of measured, independent and observed [<i>I</i> > 2σ(<i>I</i>)] reflections	14362, 6680, 2714	15786, 2451, 2115	10279, 5122, 2876	21203, 5884, 3247	39755, 9349, 6367	28518, 7690, 4297
<i>R</i> _{int}	0.0879	0.0607	0.0523	0.1258	0.0691	0.1072
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.0873, 0.2000, 1.046	0.0848, 0.1765, 1.009	0.0548, 0.1371, 1.008	0.0541, 0.1175, 0.962	0.0576, 0.1654, 1.014	0.1149, 0.2505, 1.133
Δρ _{max} , Δρ _{min} (e Å ⁻³)	0.45, -0.30	0.64, -0.24	0.23, -0.20	0.20, -0.23	0.61, -0.35	0.82, -0.32

Table S2. Maxima of selected absorption bands in IR spectra of compounds **1–6** as compared to alkali metal *clos*o-borates (cm⁻¹).

	$\nu(\text{NH})$	$\nu(\text{BH})$	$\nu(\text{CN}), \nu(\text{CC})$ heterocycle	$\pi(\text{CH})$ heterocycle
K ₂ [B ₁₀ H ₁₀]	—	2529, 2461	—	—
Phen	—	—	1586	854, 739
4	3095	2510, 2477, 2451, 2421	1658, 1631, 1616, 1595	855, 844, 719, 715
6 · 0.5H ₂ O	3164, 3152	2495, 2476, 2414	1656, 1651, 1618, 1610, 1543	824, 761
BPA	3255, 3181, 3102	—	1599, 1566, 1531	769, 735
1 · 2H ₂ O	3286, 3236, 3204, 3136	2543, 2492, 2444, 2420	1641, 1591, 1528	787, 781, 773
3 · 2CH ₃ CN	3467, 3316, 3255, 3212, 3145, 3111	2499, 2466, 2448, 2404	1648, 1612	773, 721
Cs ₂ [B ₁₂ H ₁₂]	—	2465	—	—
5 · CH ₃ CN	3410, 3361	2484, 2461, 2447, 2422	1648, 1607, 1549, 1531, 1500	847, 812, 785
2 · 2H ₂ O	3098, 3063	2487, 2443	1629, 1608, 1586	851, 783, 738, 719

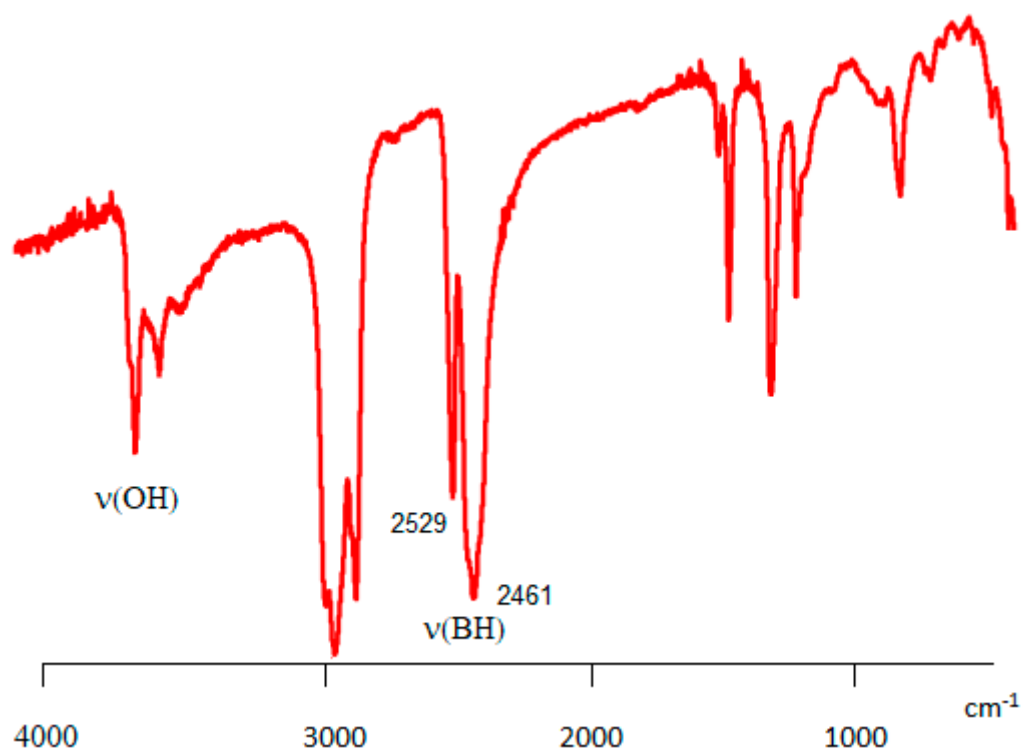


Fig. S1. IR spectrum of $K_2[B_{10}H_{10}] \cdot nH_2O$ (Nujol mull).

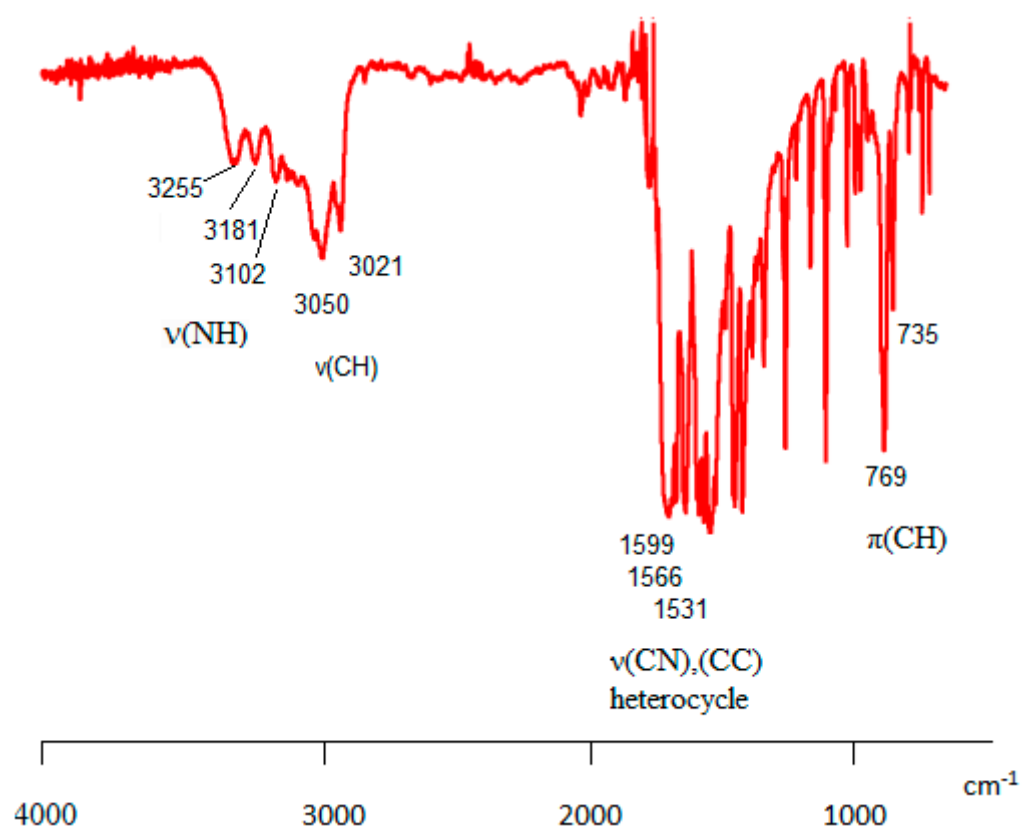


Fig. S2. IR spectrum of bpa (thin layer).

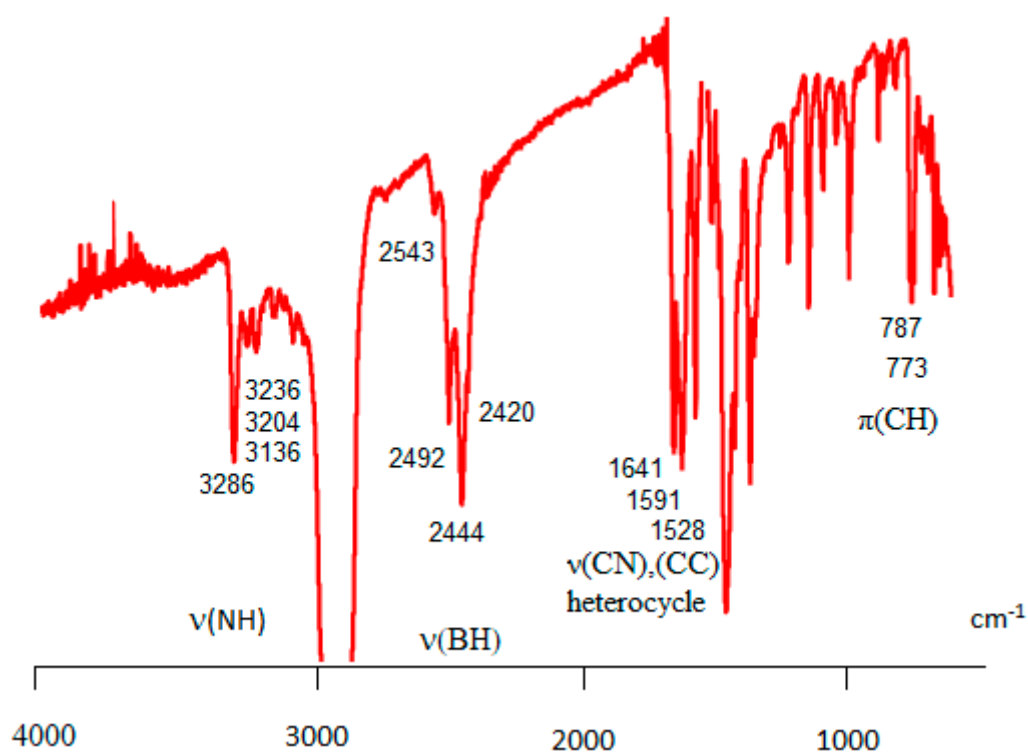


Fig. S3. IR spectrum of (Hbpa)₂(H₂bpa)[B₁₀H₁₀]₂ (1 · 2H₂O) (Nujol mull).

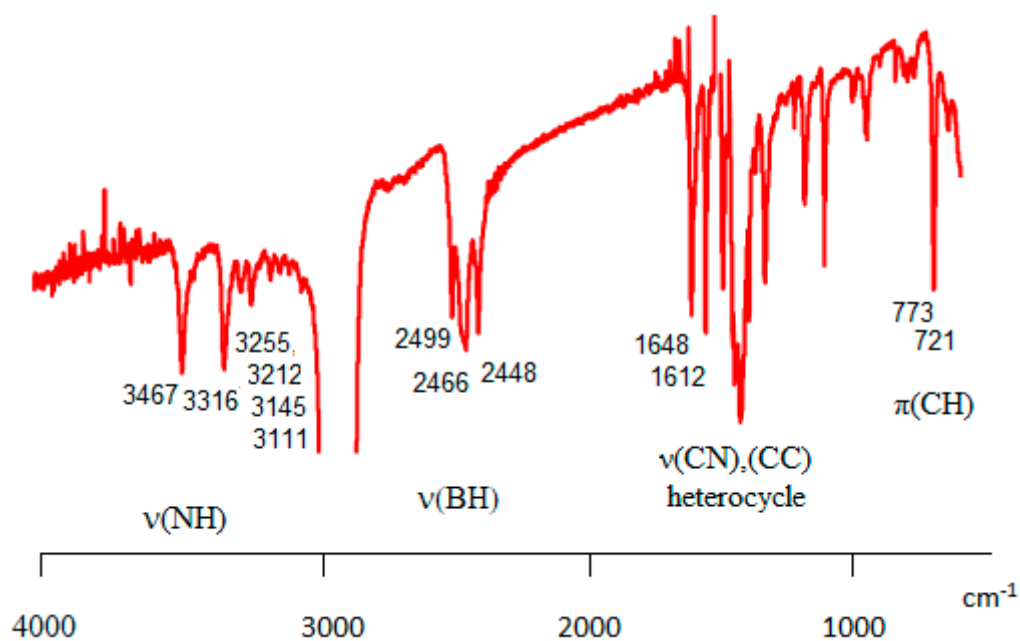


Fig. S4. IR spectrum of (NH₄)₃(Hbpa)[B₁₀H₁₀] (3 · 2CH₃CN) (Nujol mull).

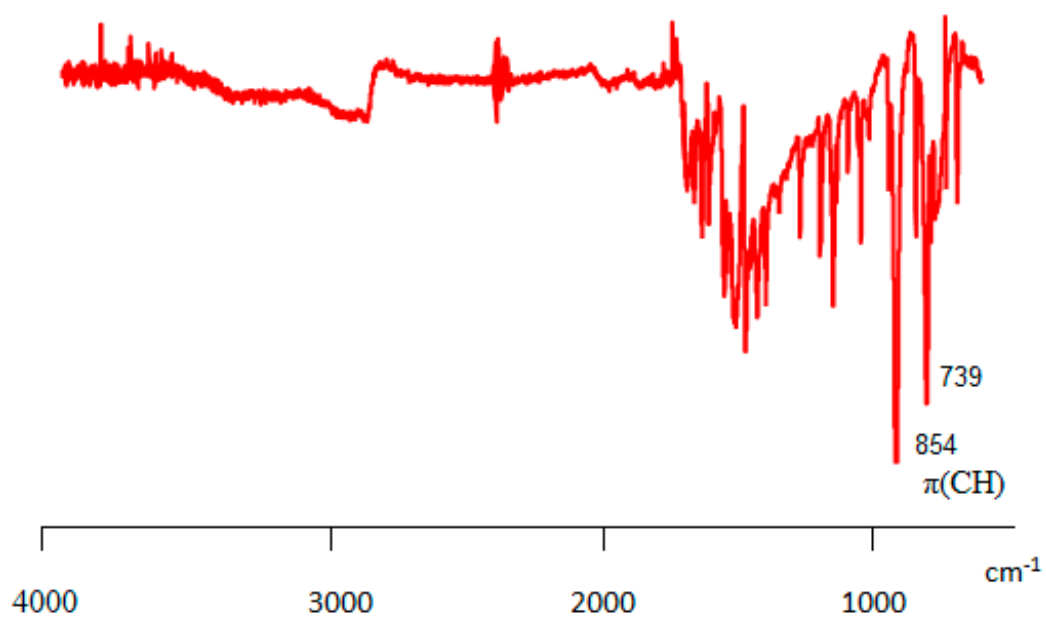


Fig. S5. IR spectrum of phen (thin layer).

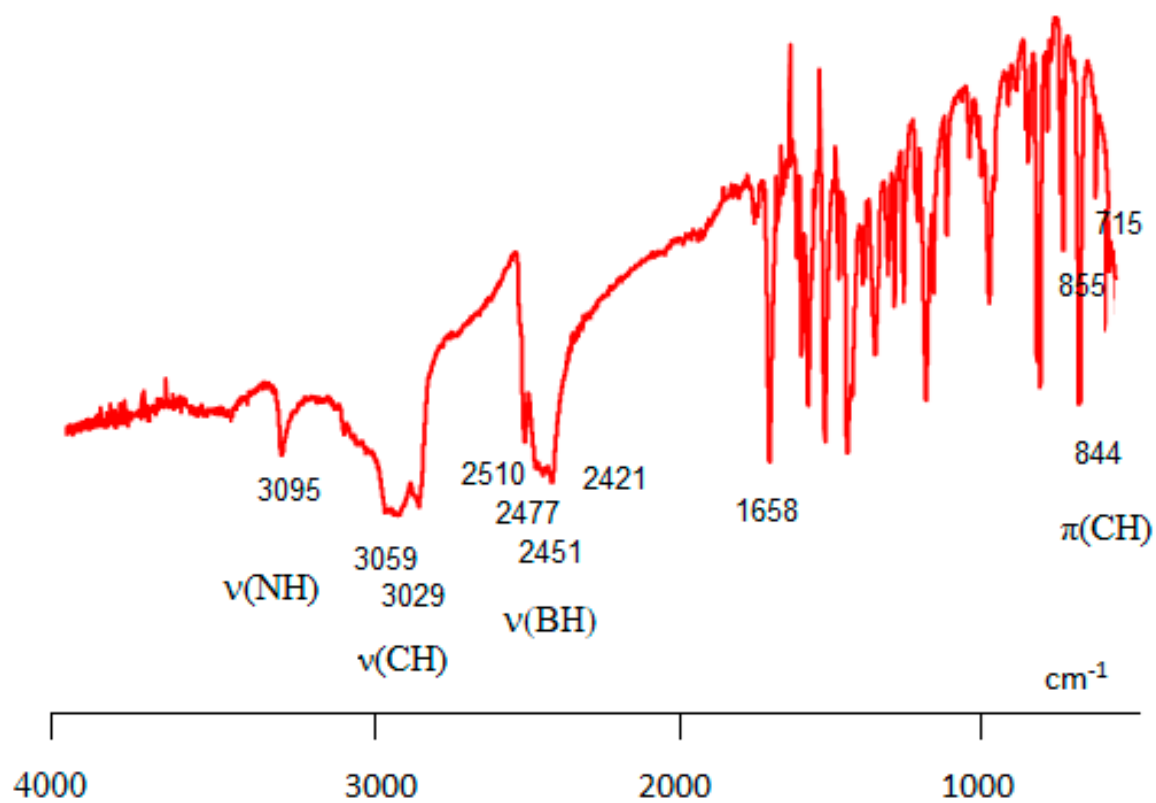


Fig. S6. IR spectrum of [Hphen)₂[B₁₀H₁₀] (4) (thin layer).

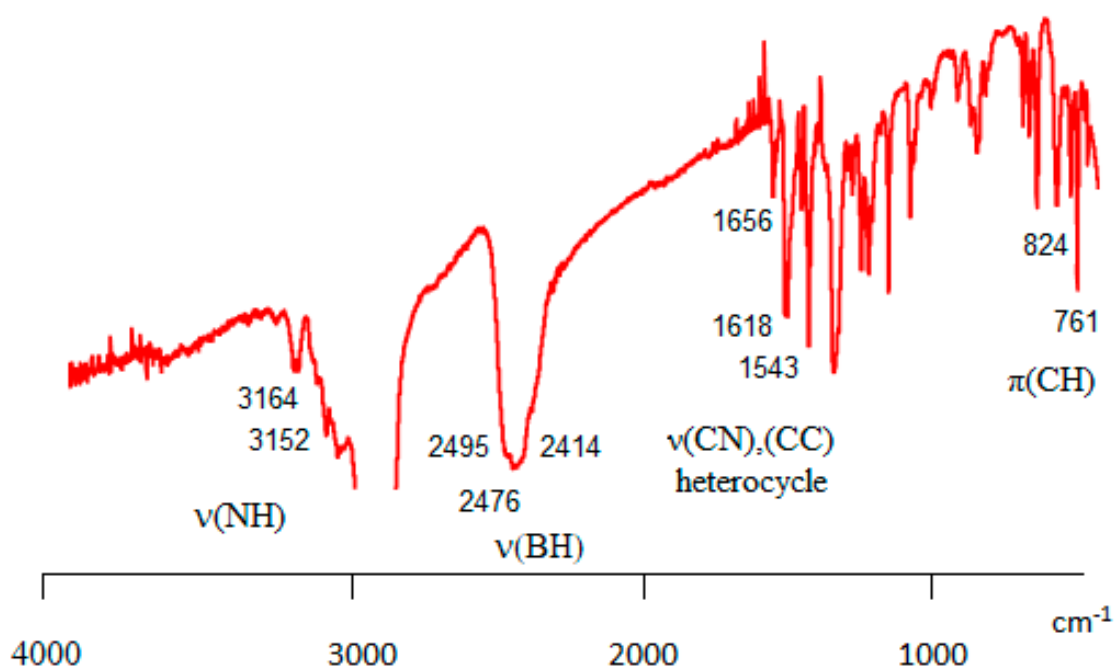


Fig. S7. IR spectrum of $[\text{Hphen}]_2(\text{phen})_{2.5}[\text{B}_{10}\text{H}_{10}] (6 \cdot 0.5\text{H}_2\text{O})$ (Nujol mull).

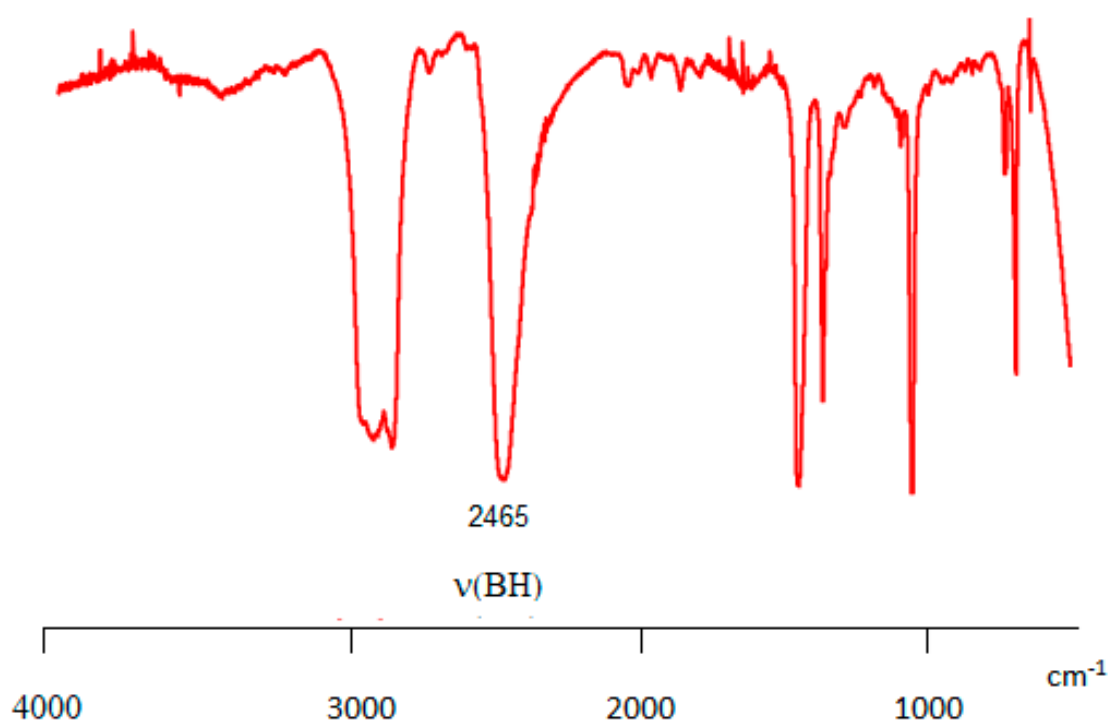


Fig. S8. IR spectrum of $\text{Cs}_2[\text{B}_{12}\text{H}_{12}]$ (Nujol mull).

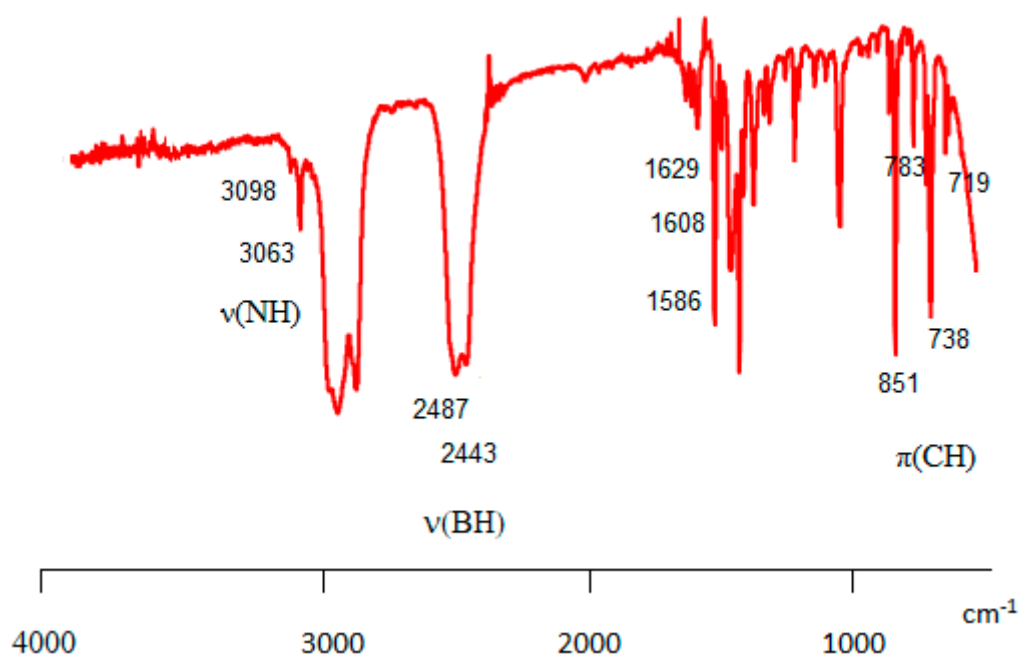


Fig. S9. IR spectrum of $[\text{PhenH}]_2[\text{B}_{12}\text{H}_{12}] (2 \cdot 2\text{H}_2\text{O})$ (Nujol mull).

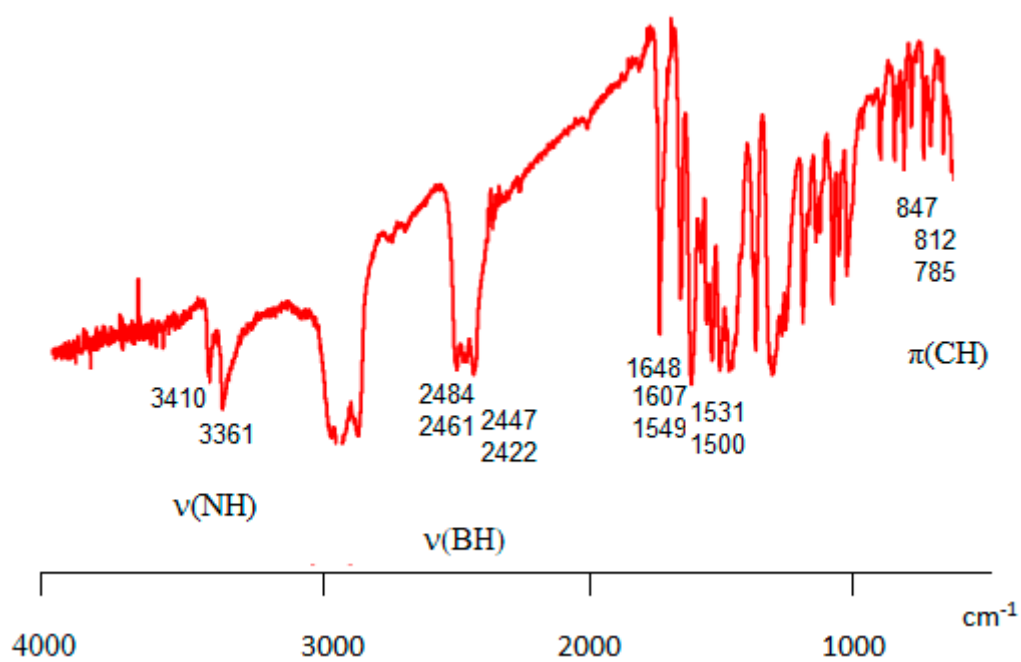


Fig. S10. IR spectrum of $[\text{Rh}_6\text{GH}]_2[\text{B}_{12}\text{H}_{12}] \cdot \text{CH}_3\text{CN} (5 \cdot \text{CH}_3\text{CN})$ (Nujol mull).

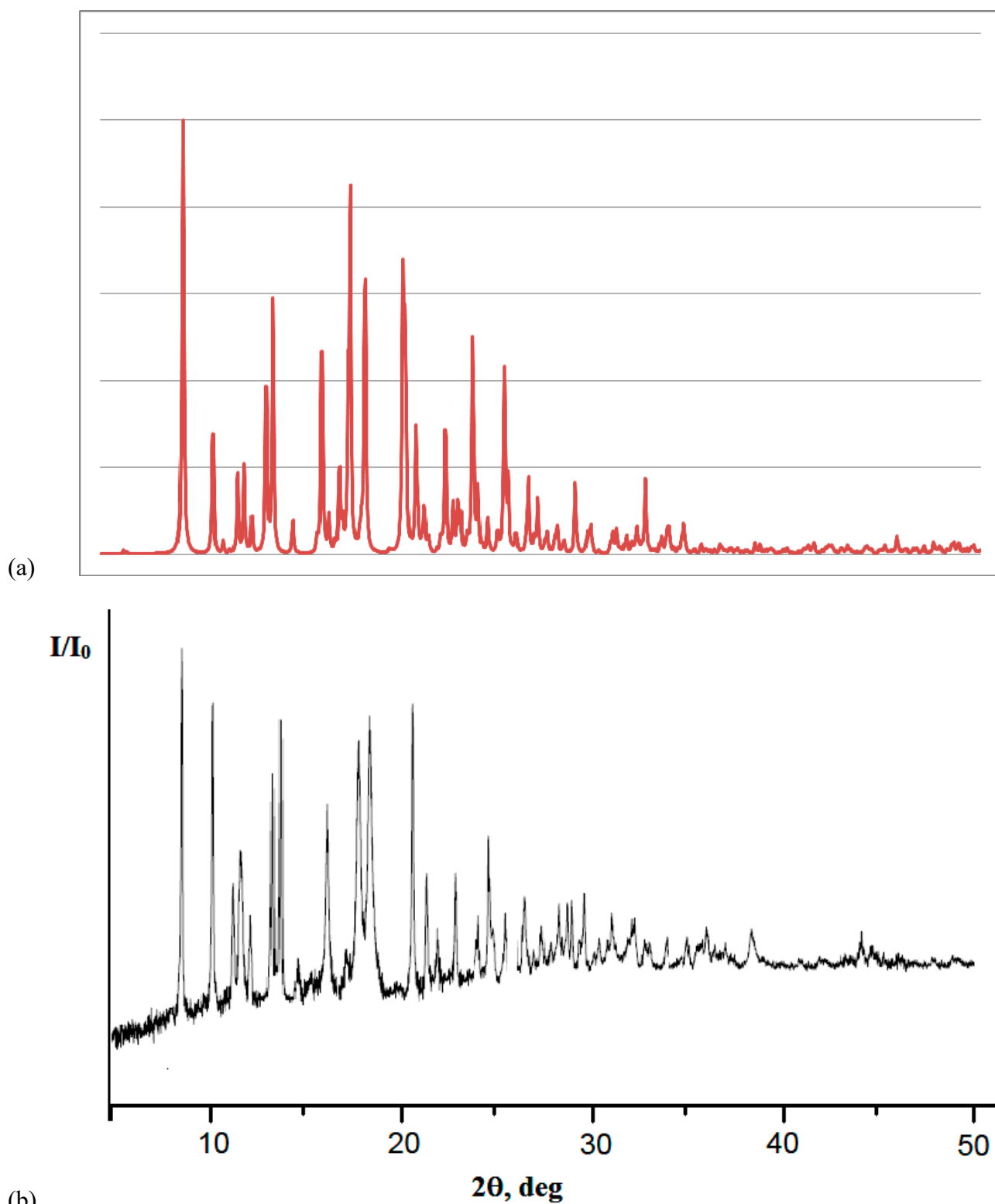


Fig. S11. (a) Experimental and (b) calculated X-ray diffraction patterns of $[\text{PhenH}]_2[\text{B}_{10}\text{H}_{10}]$ (**4**).

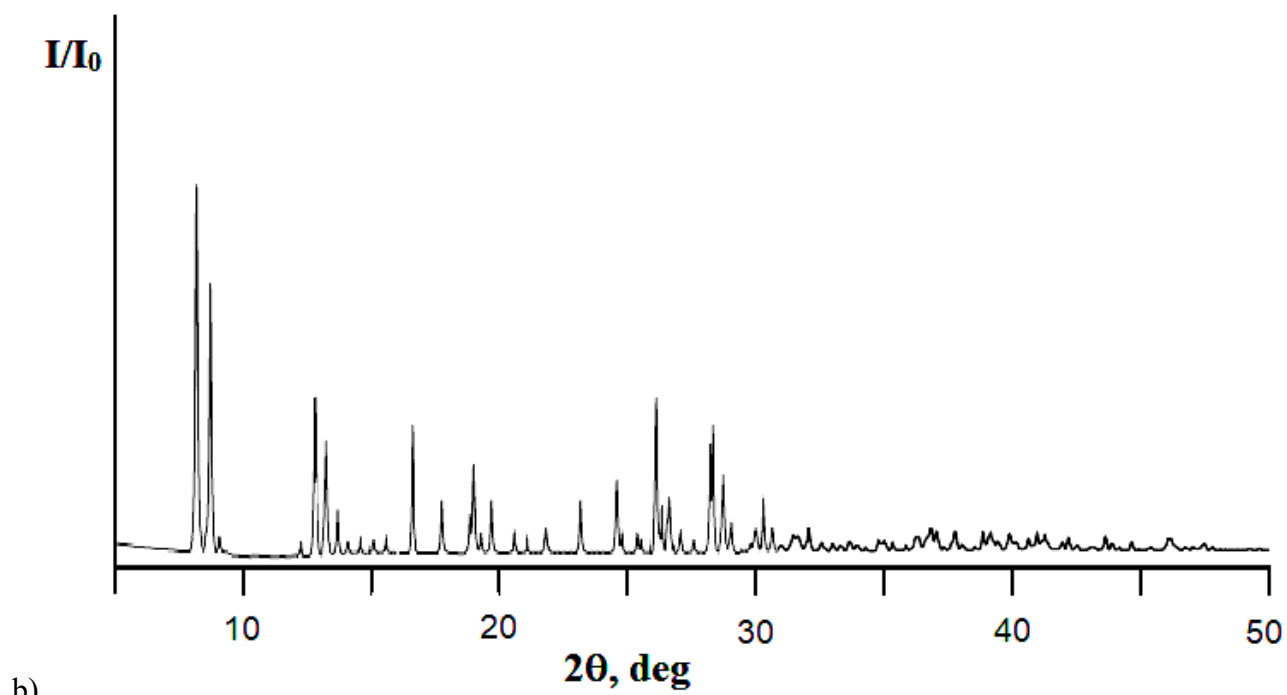
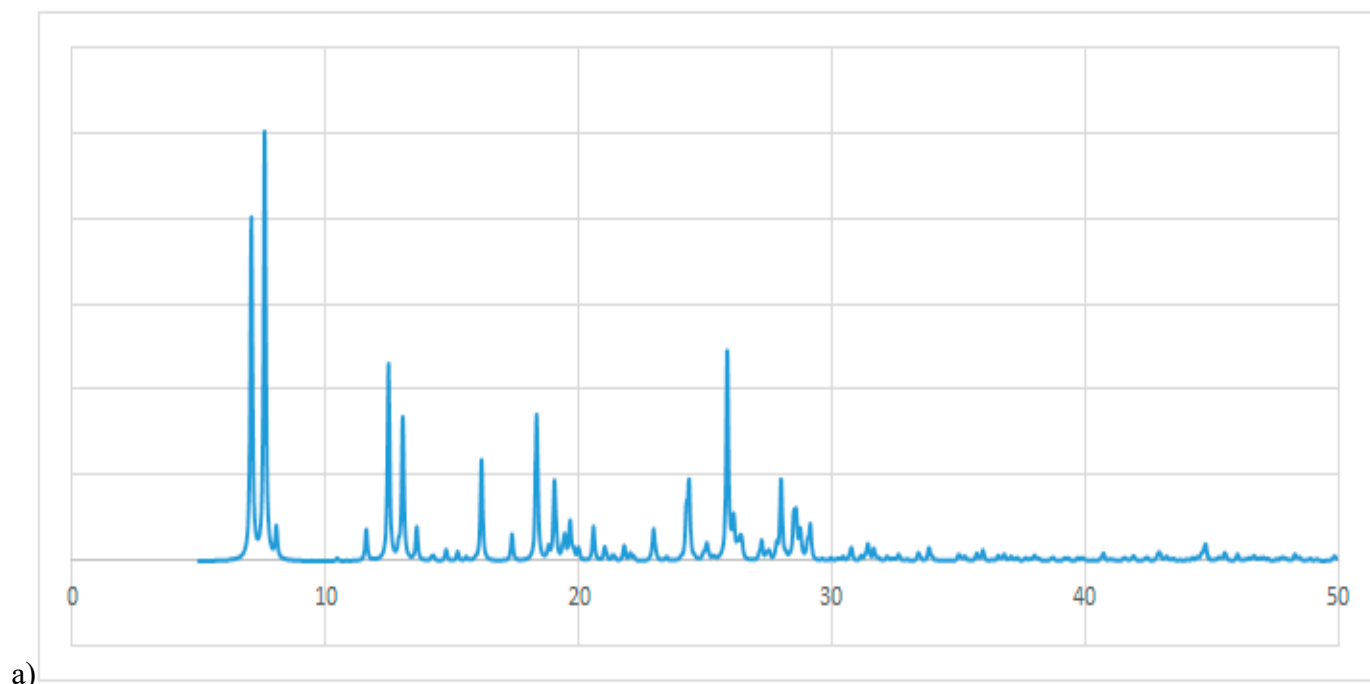


Fig. S12. (a) Experimental and (b) calculated X-ray diffraction patterns of $[\text{Rh}_6\text{GH}]_2[\text{B}_{12}\text{H}_{12}] \cdot 2\text{CH}_3\text{CN}$ (**5** · 2CH₃CN).

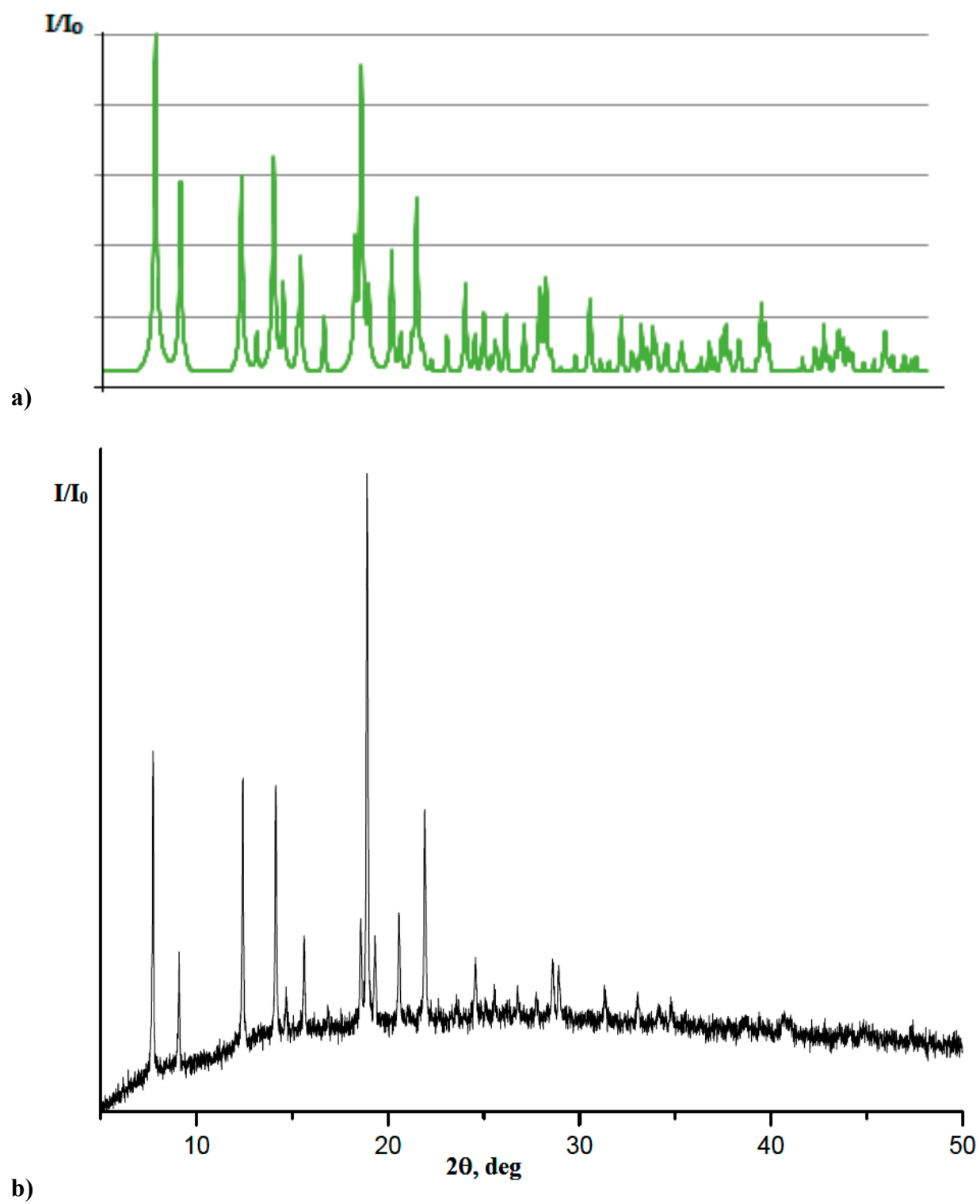


Fig. S13. (a) Experimental and (b) calculated (from ref. [36]) X-ray diffraction patterns of complex $[\text{Co}(\text{Phen})_3][\text{B}_{10}\text{H}_{10}]$ (**7**).