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High pO_2 Floating Zone Crystal Growth of the Perovskite Nickelate $PrNiO_3$

Hong Zheng ^{1,*}, Junjie Zhang ^{1,2} , Bixia Wang ¹, Daniel Phelan ¹, Matthew J. Krogstad ¹, Yang Ren ³, W. Adam Phelan ⁴, Omar Chmaissem ^{1,5}, Bisham Poudel ⁵ and J. F. Mitchell ¹

¹ Materials Science Division, Argonne National Laboratory, Argonne, IL 60439, USA

² Materials Science and Technology Division, Oak Ridge National Laboratory, Oak Ridge, TN 37831, USA

³ Advanced Photon Source, Argonne National Laboratory, Argonne, IL 60439, USA

⁴ Platform for the Accelerated Realization, Analysis and Discovery of Interface Materials (PARADIM), Department of Chemistry, The Johns Hopkins University, Baltimore, MD 21218, USA

⁵ Department of Physics, Northern Illinois University, DeKalb, IL 60115, USA

* Correspondence: zheng@anl.gov

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Abstract: Single crystals of $PrNiO_3$ were grown under an oxygen pressure of 295 bar using a unique high-pressure optical-image floating zone furnace. The crystals, with volume in excess of 1 mm^3 , were characterized structurally using single crystal and powder X-ray diffraction. Resistivity, specific heat, and magnetic susceptibility were measured, all of which evidenced an abrupt, first order metal-insulator transition (MIT) at $\sim 130\text{ K}$, in agreement with previous literature reports on polycrystalline specimens. Temperature-dependent single crystal diffraction was performed to investigate changes through the MIT. Our study demonstrates the opportunity space for high fugacity, reactive environments for single crystal growth specifically of perovskite nickelates but more generally to correlated electron oxides.

Keywords: high pressure crystal growth; floating zone method; perovskite; nickelate; metal-insulator transition

1. Introduction

The perovskite nickelates, $RNiO_3$ ($R = Y, La, Pr-Lu$), have received ongoing attention due to their complex magnetic, transport, and structural phase diagram [1–23]. Except for $LaNiO_3$ [1–3,6,24–30], all of the perovskite nickelates undergo a first-order metal-insulator transition (MIT) as a function of temperature. The transition temperature of the MIT depends on the size of the rare-earth ions, and correlates with the Ni-O-Ni bond angle [1–3]. The nature of the MIT has been discussed in the literature in terms of charge disproportionation [14,16,31–34], negative charge transfer [5], and bipolaron condensation [4], but no consensus has been reached. In addition to the MIT, antiferromagnetic order at T_N occurs concomitantly with MIT for $R = Pr^{3+}$ and Nd^{3+} , and below T_{MIT} for rare earth ions smaller than Nd. However, the magnetic structure of the ground state remains an open question. In particular, three magnetic structures have been proposed based on powder neutron diffraction that are consistent with an observed propagation vector $q = (\frac{1}{4}, \frac{1}{4}, \frac{1}{4})$; however, no experiments have been able to differentiate among them [15]. Polycrystalline powders and thin films provide an incomplete means to understand the nature of the structural and magnetic transitions in the perovskite nickelates [1,2]. Thus, single crystals of size and quality suitable for neutron diffraction are necessary.

It is well-established that high oxygen partial pressure (pO_2) is needed to prepare polycrystalline specimens of $RNiO_3$ [35–37]. For single crystal growth, it was reported that single crystals of $PrNiO_3$ with dimensions of $\sim 0.5\text{ mm}$ on an edge could be grown by slowly cooling the melt of $Pr_6O_{11} +$

$6\text{NiO} + 0.5 \text{KClO}_4 + 0.5 \text{KCl} + 0.5 \text{NaClO}_4 + 0.5 \text{NaCl}$ at 4.5 GPa by Saito et al. in 2003 [38,39]. Micrometer-sized single crystals of NdNiO_3 (~100 μm on an edge) were grown in a belt-type press at 4 GPa by Alonso et al. [40]. Recently, we reported the successful single crystal growth of LaNiO_3 using a high pressure floating zone furnace [27], and subsequently so did Guo et al. [26], reviving interest in this strongly correlated metallic oxide. The required $p\text{O}_2$ for stabilizing PrNiO_3 is expected to exceed that of the LaNiO_3 case, as is true for polycrystalline synthesis of nickelates of the smaller rare-earth ions [36].

In this paper, we report successful crystal growth of PrNiO_3 at 295 bar of oxygen pressure via the floating-zone technique. Single crystal X-ray scattering demonstrates the quality of the single crystals and reveals temperature-induced changes near T_{MIT} . Electrical resistance and heat capacity measurements on the grown single crystals evidence MITs and show how the MITs and the low-temperature states respond to oxygen stoichiometry.

2. Experimental Section

Crystal Growth. As-purchased Pr_6O_{11} (Alfa Aesar, 99.99%) was baked at 600 °C for at least 48 h before use. A stoichiometric ratio of Pr_6O_{11} and NiO (Alfa Aesar, 99.99%) powder was weighed, and the mixture was thoroughly ground, loaded into a Pt crucible, heated in flowing O_2 to 1050 °C at a rate of 3 °C/min, allowed to dwell for 24 h, and then furnace-cooled to room temperature. The solid was then reground and fired twice at 1050 °C, with intermediate grinding, using the same atmosphere and heating protocol. The powder was then hydrostatically pressed into polycrystalline rods (length ~80 mm, diameter ~6 mm) at 15,000 psi and sintered at 950 to 1050 °C for 24 h for crystal growth. PrNiO_3 crystals were grown using the 300 bar high pressure floating zone furnace (SciDre GmbH) at the Platform for the Accelerated Realization, Analysis and Discovery of Interface Materials (PARADIM) user facility at Johns Hopkins University [41]. A 5 kW Xenon arc lamp was utilized as a heating source at pressure of 295 bar with an oxygen flow rate of 0.2 L/min. Feed and seed rods were counter-rotated at 20 and 15 rpm, respectively. The feed rod and the seed rod was advanced at rates of 10 to 20 mm/h and 5 mm/h respectively. Figure 1a shows crystal boules of PrNiO_{3-x} , and large single crystals can be seen in the polished cross section (see Figure 1b). The crystallinity of the of the as-grown boule was characterized using high-energy synchrotron X-ray diffraction at Beamline 11-ID-C ($\lambda = 0.11165 \text{ \AA}$, beam size = $0.5 \times 0.5 \text{ mm}^2$) of the Advanced Photon Source at Argonne National Laboratory (Details of this beamline and its capabilities can be found at: https://www.aps.anl.gov/Beamlines/Directory/Details?beamline_id=15), shown in Figure 1c.

Powder X-ray Diffraction. Powder X-ray diffraction data were collected at room temperature on pulverized single crystals using a PANalytical X'Pert Pro powder diffractometer with $\text{Cu K}\alpha$ radiation (Malvern PANalytical, Royston, United Kingdom) ($\lambda = 1.5418 \text{ \AA}$) in the 2θ range of 5–80°. The data were analyzed using GSAS-II software [42] using a structural model from the literature [35]. Refined parameters include scale factor, sample displacement, background (8th-degree polynomial), lattice parameters, isotropic domain size, isotropic microstrain, and isotropic atomic displacement parameters, U_{iso} (constrained to be the same).

Single Crystal X-ray Diffraction. Single crystal X-ray diffraction was performed at Beamline 6-ID-D at the Advanced Photon Source using a wavelength of 0.14238 Å and a 2M CdTe Pilatus detector. Data collection was performed under continuous rotation, and data were transformed from detector space to reciprocal space using NeXpy. Symmetrization was not applied to the data shown in Figure 5, but symmetrization was applied to the data shown in Figure 6.

Electrical Transport. The resistivity of PrNiO_3 single crystals was measured on a Quantum Design Physical Properties Measurement System (PPMS, Quantum Design, San Diego, CA, USA) in the temperature range 1.8–300 K using the van der Pauw method with indium contacts.

Specific Heat. Specific heat measurements were performed on a Quantum Design PPMS in the temperature range 2 to 260 K. Apiezon-N vacuum grease was employed to fix ~20 mg crystals to the sapphire sample platform. Addenda were subtracted.

Magnetic Susceptibility. Magnetic susceptibility measurements were performed using a Quantum Design MPMS-3 SQUID magnetometer (Quantum Design, San Diego, CA, USA). A small piece of PrNiO_3 crystal cut from the crystal boule, was attached to a quartz rod using a minimum amount of Apiezon grease. Zero-field cooled (ZFC) and field cooled (FC) data were collected under a magnetic field of 0.1 T. The sample was cooled in zero field to 2 K at a rate of 35 K/min, after which the field was applied, and DC magnetization measurement was performed on warming at 2 K/min (ZFC). The sample was then cooled in field at 35 K/min to 2 K, and then DC magnetization data were collected on warming at 2 K/min (FC).

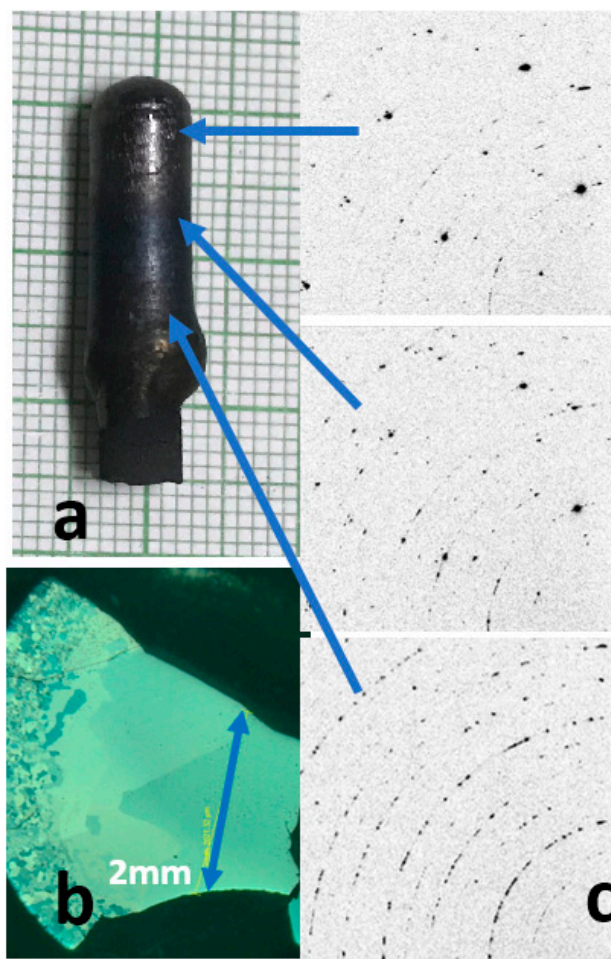


Figure 1. (a) PrNiO_3 crystal boule grown under 295 bar of oxygen. Grid size is 1 mm. (b) A polished surface of a slice from the PrNiO_3 boule. (c) X-ray single crystal diffraction patterns measured at various locations along the length of the boule at 11-ID-C.

3. Results and Discussion

High oxygen fugacity environments are essential to prepare bulk RNiO_3 specimens [35,36], motivating attempts to grow single crystals under extreme high oxygen pressure. For instance, ~ 0.5 mm-size single crystals of PrNiO_3 were grown by Saito et al. at 4.5 GPa [38], but the nature of the high pressure flux growth experiment put significant constraints on the obtained crystal size. Recently, we and others have found that RNiO_3 crystals can be grown at significantly lower $p\text{O}_2$ using a floating-zone approach at high oxygen fugacity. For example, LaNiO_3 crystals were grown by our group [27] and subsequently by the MPI-Dresden group [26] using high oxygen pressure ($p\text{O}_2 \sim 30\text{--}150$ bar). Unfortunately, our attempts to grow PrNiO_3 using the 150 bar floating zone furnace in our Laboratory were unsuccessful. However, crystal growth was successful using the 300 bar floating

zone furnace installed at the NSF-sponsored PARADIM crystal growth facility located at Johns Hopkins University. Indeed, success in this case was serendipitous, as we observed the appearance of PrNiO_3 as a biproduct of growth of the trilayer Ruddlesden-Popper phase $\text{Pr}_4\text{Ni}_3\text{O}_{10}$ [43,44]. As shown in Figure 2, PrNiO_3 was observed as a minor phase at 220 bar $p\text{O}_2$, but as a significant component at 260 bar. The PrNiO_3 phase was not observed in growth at or below 180 bar. Thus, increasing $p\text{O}_2$ favors the perovskite phase (nominal Ni^{3+}) at the expense of the layered phase (nominal $\text{Ni}^{2.67+}$), consistent with our work on the La-Ni-O system [43].

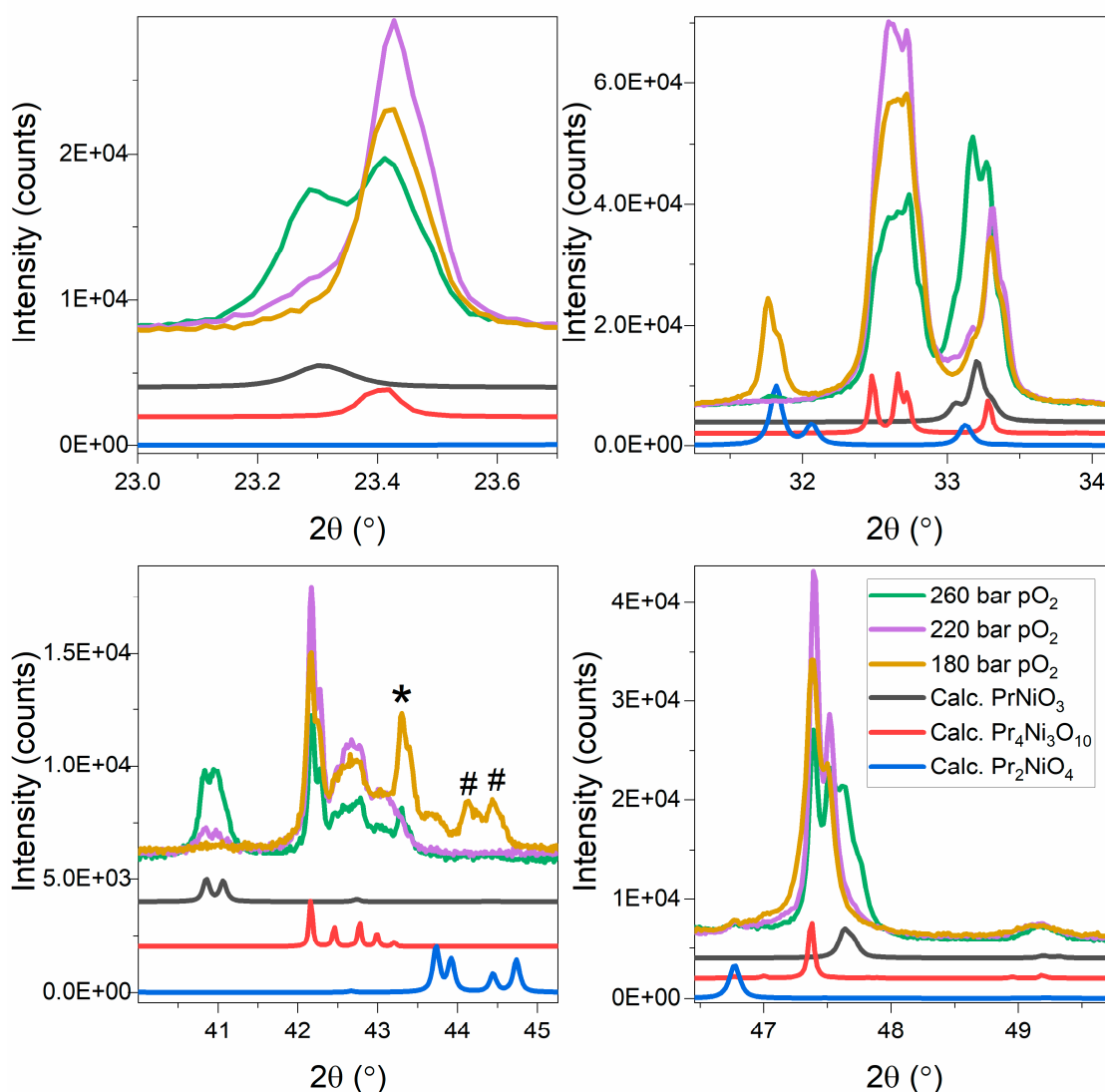


Figure 2. X-ray powder diffraction patterns of the phases obtained during floating zone growth under different $p\text{O}_2$ with a feed rod composition of Pr: Ni = 4:3. The PrNiO_3 phase is observed to increase in volume fraction upon increasing $p\text{O}_2$. Bottom three curves in each panel are calculated patterns for PrNiO_3 , $\text{Pr}_4\text{Ni}_3\text{O}_{10}$, and Pr_2NiO_4 . The * and # mark reflections from NiO and Pr_6O_{11} second phases, respectively.

We then attempted to grow PrNiO_3 with a molar ratio of Pr:Ni = 1:1 at $p\text{O}_2 = 295$ bar of oxygen, which was close to the upper limit of the furnace (300 bar). The starting rod itself was not a single PrNiO_3 phase. Rather, it consisted of a mixture of PrO_x , NiO, and Pr_2NiO_4 because the PrNiO_3 phase is formed only at elevated oxygen pressure. During initial growth efforts, we found that stability of the zone was difficult to maintain both because of cracks that formed on the feed rod itself prior to melting, and because of the low viscosity of the molten liquid in the zone, which led to frequent zone

loss. To improve the stability and prevent cracking in the feed rod, we found that preparing the rods at low sintering temperatures (950 °C) was beneficial, as severe cracks were less likely to appear when the feed rods were less dense. We surmise that small amounts of molten liquid penetrate into the feed rod and seal the cracks before they can expand catastrophically. Fortunately, excessive penetration of the liquid into the feed rod was not observed, which would have led to a deleterious ‘pre-melt’ zone often found with low-density feed rods.

Figure 1a shows a PrNiO₃ crystal boule grown under 295 bar pO₂ at a traveling rate of 5 mm/h for seed rod, and Figure 1b shows a part of polished cross section surface from the boule. It is apparent that the inside core of the boule (right side of Figure 1b) consists of several ~ mm size single crystals, whereas the outside of the boule consists of a polycrystalline shell. Despite several attempts, we have so far been unable to prepare single crystal boule. Both the single crystals on the inside core and the polycrystalline shell on the outside consist of the PrNiO₃ phase without any second phases, as shown by laboratory X-ray powder diffraction. The mechanism leading to this polycrystalline shell is not understood. It is not unreasonable to speculate that this effect is due to the surface contact with the supercritical O₂ fluid growth environment, which could increase rapid cooling at the surface related to the bulk and/or dissolution of PrNiO₃ and the surface [41]. However, verifying this will require further experiments. Figure 1c shows single crystal diffraction patterns measured in transmission mode along the length of the boule at the beamline 11-ID-C of the APS. The evolution from a ceramic rod to single crystals can be seen by the transition from a textured ring of scattering to well-defined Bragg reflections along the growth direction. The constellation of diffraction spots reflects the fact that the crystal boule consists of multiple grains, as shown in Figure 1b.

Figure 3 shows Rietveld refinement of laboratory powder x-ray diffraction collected from a pulverized quarter disc cut from the crystal boule, representing a bulk average of the resultant growth. No secondary phases were found. All peaks can be indexed in the space group *Pbnm*, with lattice parameters $a = 5.4242(3)$ Å, $b = 5.3781(3)$ Å, $c = 7.6323(5)$ Å, which are comparable to the powder results reported by Lacorre et al. [35] for a polycrystalline specimen, $a = 5.4154(1)$ Å, $b = 5.3755(1)$ Å, $c = 7.6192(2)$ Å.

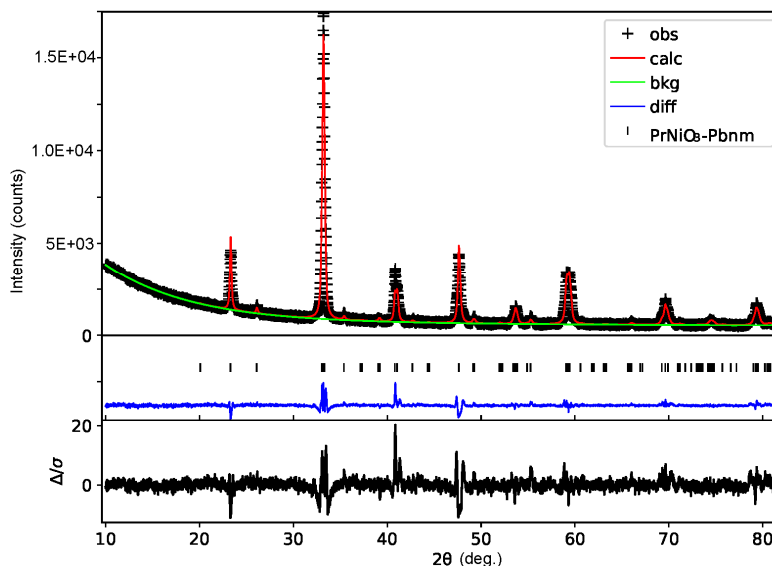


Figure 3. Rietveld refinement of laboratory-based x-ray powder diffraction of PrNiO₃ measured at room temperature.

Physical Properties. We explored the physical properties of several specimens extracted from the boule. Among the samples studied, we found two classes of temperature-dependent transport behavior in the crystals. Data are presented here for three such samples, labeled A, B, and C. Electrical transport and specific heat data are plotted in Figure 4a for sample A. Because of the irregularity in the shape

of the samples, we plot resistance rather than resistivity. Sample A exhibits a sharp metal-insulator transition (MIT) at $T_{MIT} \sim 128$ K spanning more than three orders of magnitude in resistance. This behavior is consistent with literature data reported on single crystals [38], although there the dynamic range is smaller ($\sim 10^2$). Correspondingly, the specific heat exhibits one sharp first-order-like peak at 128 K as shown in Figure 4a. The temperature dependence of both DC magnetic susceptibility (χ) and the inverse magnetic susceptibility ($1/(\chi - \chi_0)$) are shown in Figure 4b. No anomaly, discontinuity, or change in slope, which would mark the onset of magnetic ordering, was observed down to 2 K as Saito et al. mentioned [38]. This is likely because the susceptibility is dominated by the paramagnetic signal from Pr^{3+} . A Curie-Weiss fit to the data in the range from 150 K to 300 K, $\chi = \chi_0 + C/(T - \theta)$, where χ_0 accounts for T -independent contributions, and C and θ are the Curie and Weiss constants, respectively, yields $\theta = -48$ K and $C = 1.89$ emu K mole $^{-1}$ Oe $^{-1}$. The derived effective moment is $\mu_{\text{eff}} = 3.9$ $\mu_B/\text{f.u.}$, which is close to the 3.58 $\mu_B/\text{f.u.}$ expected for a free ion Pr^{3+} [45]. Other crystals, as exemplified by Sample B in Figure 4c, exhibit a clear transition in the transport around the MIT and then a re-entrance to a metallic behavior ($dR/dT > 0$) at lower temperatures. The DC magnetic susceptibility for sample B agrees with that of sample A. Samples exemplified by B showed a large temperature hysteresis, the width of which exhibited sample to sample variation. Such re-entrant behavior has previously been reported in polycrystalline samples, induced by high pressure, and attributed to oxygen non-stoichiometry [46]. To test whether oxygen deficiency was the cause of this behavior, we annealed Sample C, which showed a hysteresis in resistance, at $p\text{O}_2 = 200$ bar at 600 °C, followed by a slow cooling 0.5 °C/min to room temperature under 200 bar pressure. The resistance before and after annealing treatment are plotted in Figure 4d, which shows that the annealing yields a resistance akin to that exhibited by Sample A: a sharp transition with narrow hysteresis. This response suggests that the observed re-entrant behavior was a consequence of oxygen vacancies in the as-grown sample and that post-growth annealing reduces the vacancy concentration. However, we cannot rule out other possibilities, for instance, that microstructural defects are removed by annealing.

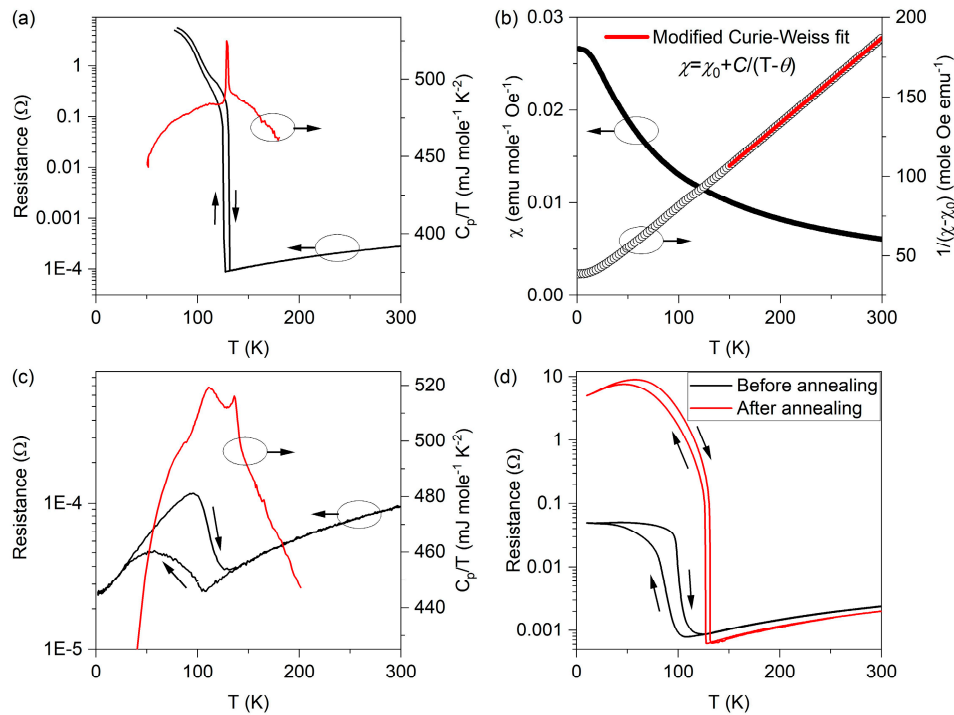


Figure 4. (a) Transport measurement (left) and specific heat (right) for sample A. (b) DC susceptibility χ (left) vs temperature and inverse magnetic susceptibility $1/(\chi - \chi_0)$ (right) vs temperature for sample A. Open and filled circles are experiment data; Solid line is fit. (c) Transport measurement (left) and specific heat (right) for sample B. (d) Transport measurements of as-grown and high pressure post annealing for sample C.

It is reasonable to question whether samples of type A or of annealed type C are indeed stoichiometric, PrNiO_3 . We do not have a direct measure of this oxygen stoichiometry; however, we note that the MIT temperature for sample A and for annealed sample C are identical, suggesting that they share a common stoichiometry. It strikes us as unlikely that these two samples, prepared in different ways, would coincidentally have the same oxygen stoichiometry unless this stoichiometry was that of a line phase. Of course, the transport measurement represents a bulk average, and some inhomogeneity may be present in one or both samples. Should this be the case, then the transport would be dominated by a fraction of the sample with composition PrNiO_3 .

X-ray Scattering. Motivated by the substantial interest in the community regarding the structural transitions in nickelates, we considered whether single crystal X-ray diffraction could be used to determine the symmetry of the low temperature and high temperature phases. This is complicated by the twinning of the crystal which causes the observed diffraction pattern to appear as a superposition of the patterns of all possible domains for a given distortion. It is thus most useful to view the structure in pseudocubic coordinates ($a = b = c \sim 3.9 \text{ \AA}$); the lower symmetries, which possess larger unit cells than the pseudocubic cell, then generate additional Bragg reflections that possess half-integer pseudocubic indices. Details of the transformation of systematic absences for twinned crystals can be found in the Supplementary Materials.

Figure 5 shows the observed $(hk0)$, $(hk1)$, and $(hk\frac{1}{2})$ planes, each at 100 K, 150 K, and 300 K for sample A. The points marked by circles are forbidden for both the $Pbnm$ and $P2_1/n$ space groups, the points marked by triangles are forbidden by $Pbnm$ but allowed for $P2_1/n$, and all other points indexed by integer/half-integer indices are allowed for both $Pbnm$ and $P2_1/n$. Points at $(n/2, n/2, 0)$, where n is an odd integer, are systematically forbidden for both $Pbnm$ and $P2_1/n$ (Figure 5a–c), though they are strongest at 100 K and significantly weaker at 300 K. In principle, this could mean that the symmetry is lower than $P2_1/n$ in both phases; however, other potential sources of these violators include multiple scattering (also known as Renninger reflections) and the possibility of a dilute presence of vacancy ordered superstructures of oxygen anions. Neither of these would be expected to have a strong temperature dependence, however, so it is useful to consider the temperature dependence of one of the stronger violators, such as $(-5/2, -5/2, 0)$, which is shown as one dimensional cuts along k in Figure 6a. The amplitude of $(-5/2, -5/2, 0)$ is comparable at 150 K and 300 K, but it is enhanced at 100 K. One possible interpretation of this temperature dependence is that the constant amplitude in the high temperature phase (150 and 300 K) is not intrinsic to the crystal (e.g., arising from multiple scattering), but that the enhancement in the low temperature phase is intrinsic. We note that recently Gawryluk et al. performed an analysis of the distortion modes of PrNiO_3 , which suggested that the symmetry of the low temperature phase may be lower than $P2_1/n$ [47]. We considered the effect of removing the n -glide, but because of the twinning the systematic absences are no different than for $P2_1/n$. Thus, removal of the n -glide would not explain the violators that we observed in the $(hk0)$ plane. On the other hand, removal of the 2_1 screw axis would allow the violators observed at $(n/2, n/2, 0)$.

The $(hk1)$ and $(hk\frac{1}{2})$ planes possess points that differentiate the $Pbnm$ and $P2_1/n$ space groups. Thus, the phase transition should generate some additional peaks in these planes. However, like the case of the $(hk0)$ plane, the pattern of reflections appeared the same in both the low- and high-temperature phases, although enhancement of the peak amplitudes was observed in the low temperature phase. A particular example is the $(-5/2, -5/2, 1)$ point, which is shown as one-dimensional cuts along k in Figure 6b. Whereas the peak is weak at 300 K, it is somewhat stronger at 150 K, and significantly stronger at 100 K. Again, a potential interpretation is that the high temperature component is extrinsic, whereas the low temperature component is intrinsic, though if this interpretation is correct then the enhancement at 150 K, which should be well into the high temperature phase, is puzzling. Ultimately, we could not arrive at a firm conclusion regarding the intrinsic symmetries of the low and high temperature phases because of the combined effects of twinning and potential multiple scattering; diffraction from a single untwinned grain would be highly favorable for addressing the correct space group and structure.

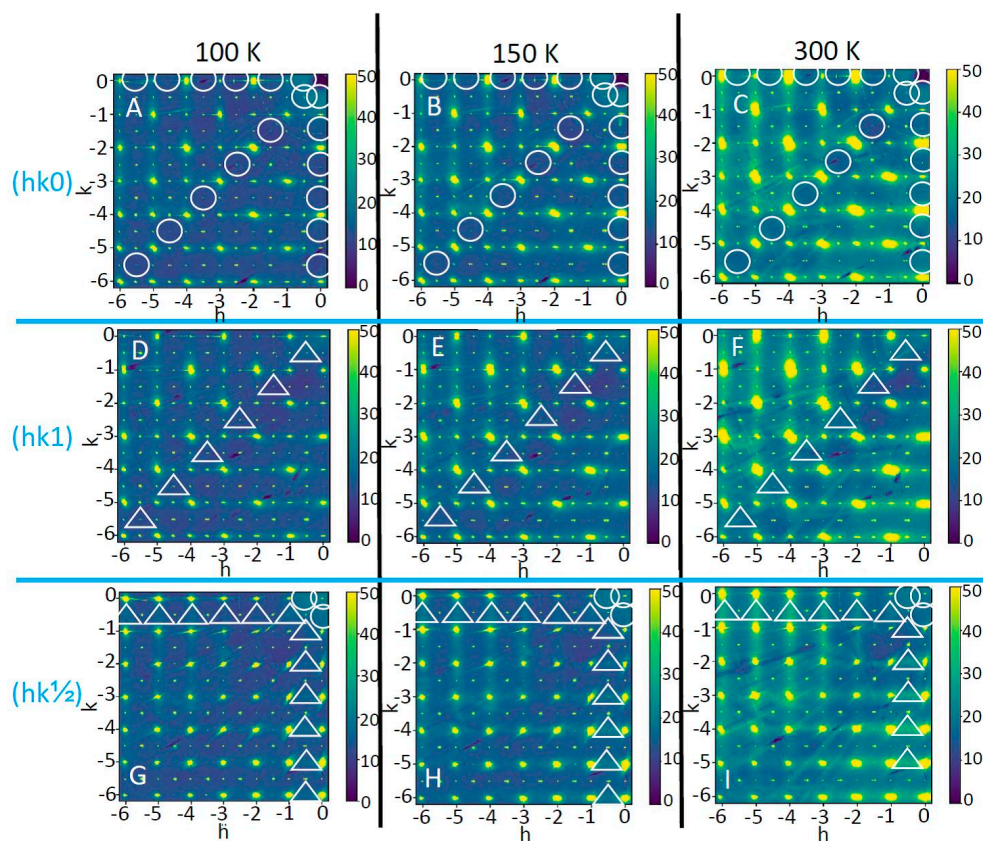


Figure 5. Pseudocubic (hk0), (hk1) and $(hk\frac{1}{2})$ planes measured by single crystal X-ray diffraction at 100 K, 150 K, and 300 K. Points that are forbidden by both $Pbnm$ and $P2_1/n$ space groups are circles, and points that are forbidden by $Pbnm$ but allowed by $P2_1/n$ are surrounded by triangles.

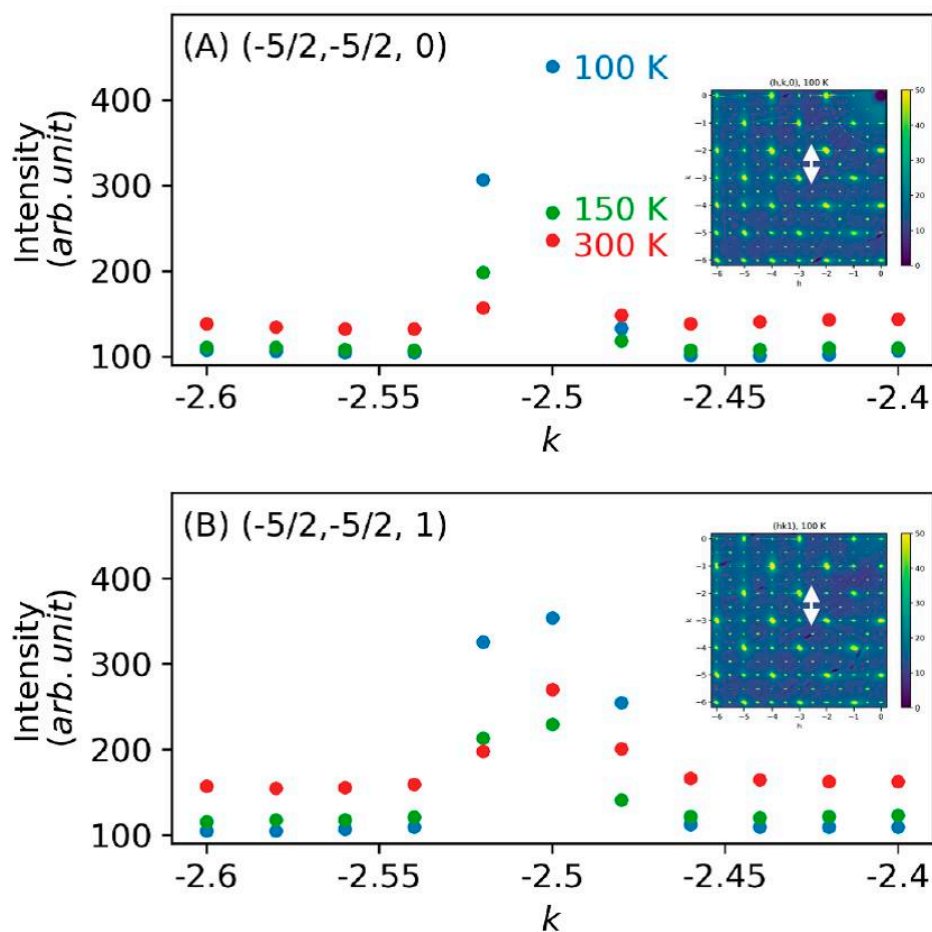


Figure 6. Cuts along k through the $(-5/2, -5/2, 0)$ and $(-5/2, -5/2, 1)$ points, shown in A and B respectively, at 100 K, 150 K, and 300 K. These cuts were made over integration ranges of 0.04 reciprocal lattice units in both h and l . The direction and center of the cuts in (A) and (B) are shown by the arrows in the insets.

4. Conclusions

Crystals of perovskite nickelate PrNiO_3 have been grown under an oxygen pressure of 295 bar using a unique high-pressure optical-image floating zone furnace. Single crystal and powder X-ray diffraction, resistivity, specific heat, and magnetic susceptibility measurements have confirmed the phase and quality. We found two types of behavior from the transport measurements, a MIT at $T_{\text{MIT}} \sim 128$ K and a clear transition in the transport around T_{MIT} and a re-entrance to a metallic state at lower temperatures. The re-entrant behavior can be modified by high pressure oxygen post-growth annealing, which we attributed to filling of oxygen vacancies in the as-grown sample. Single crystal X-ray scattering evidenced extremely weak peaks that violated the systematic absence condition from $Pbnm$ and $P2_1/n$ and others that violated only $Pbnm$; these peaks were enhanced in the low temperature phase. However, it is unclear whether or not they are related to an actual breaking of the symmetry of stoichiometric PrNiO_3 , and more work is needed on this front.

Supplementary Materials: The following are available online at <http://www.mdpi.com/2073-4352/9/7/324/s1>: Details of the transformation of systematic absences for twinned crystals.

Author Contributions: Conceptualization, J.F.M., H.Z. and J.Z.; methodology, H.Z., J.Z. and B.W.; validation, H.Z., J.Z. and D.P.; formal analysis, D.P., J.Z., and M.J.K.; investigation, W.A.P., O.C., H.Z. and J.Z.; resources, Y.R., O.M. and B.P.; writing—original draft preparation, H.Z.; writing—review and editing, D.P., J.Z., J.F.M.; supervision, J.F.M.

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Conflicts of Interest: The authors declare no conflict of interest.

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