## Growth of Ga<sub>2</sub>O<sub>3</sub> Nanowires via Cu-As-Ga Ternary Phase Diagram

Hang Wang<sup>1,2,+</sup>, Ying Wang<sup>2,3,+</sup>, Shuyan Gong<sup>2</sup>, Xinyuan Zhou<sup>2</sup>, Zaixing Yang<sup>4</sup>, Jun Yang<sup>1+</sup>, Ning Han<sup>2,3+</sup>, and Yunfa Chen<sup>2,3+</sup>

- <sup>1</sup> School of Metallurgical Engineering, Xi'an University of Architecture and Technology, Xi'an 710055, P. R. China
- <sup>2</sup> State Key Laboratory of Multiphase Complex Systems, Institute of Process Engineering, Chinese Academy of Sciences, Beijing 100190, P. R. China
- <sup>3</sup> Center for Excellence in Regional Atmospheric Environment, Institute of Urban Environment, Chinese Academy of Sciences, Xiamen 361021, P. R. China
- <sup>4</sup> Center of Nanoelectronics and School of Microelectronics, Shandong University, Jinan 250100, P. R. China

\*Correspondence: vi-yangjun@xauat.edu.cn, Tel.: 86-13152420820;nhan@ipe.ac.cn, Tel.:

+86-010-6255-8356; <u>chenyf@ipe.ac.cn</u>;

Synthesis of Cu<sub>2</sub>O nanoparticles: a NaOH aqueous solution (25.0 mL, 1.5 M) was added to an aqueous solution of CuSO<sub>4</sub> (100.0 mL, 0.01 M) under magnetic stirring at room temperature. After stirring for >1 min, blue Cu(OH)<sub>2</sub> sol is formation. In this blue sol, ascorbic acid (AA) aqueous solution with a concentration of 0.1M and 25 ml was added rapidly, and stirred at room temperature for 15-30 minutes. The solution gradually turned to orange-red precipitation. Subsequently, the as obtained product was centrifuged and washed several times with deionized water and ethanol to remove the residual inorganic ions, and finally dried in vacuo at 80°C overnight. Other Cu<sub>2</sub>O particle sizes samples were obtained by changing the concentration of added NaOH. Specific test conditions are as follows table S1.

Number	CuSO <sub>4</sub> (M)	AA(M)	NaOH(M)	Cu <sub>2</sub> O
				Sizes(nm)
1	0.01	0.1	0.25	25.36
2	0.01	0.1	0.5	33.25
3	0.01	0.1	0.6	51.47
4	0.01	0.1	0.75	99.52
5	0.01	0.1	0.8	177.63

Table S1. Synthesis conditions of Cu<sub>2</sub>O using ascorbic acid as reductive agent.

Cu <sub>2</sub> O size (nm)	25.36	33.25	51.47	99.52	177.63
Standard	4.17	6.49	10.04	17.68	22.86
Deviation					
NW	18.82	24.33	39.69	69.87	67.704
diameter(nm)					
Standard	5.65	5.46	8.24	19.82	21.49
Deviation					

Table S2. Catalyst size, NW diameter and corresponding standard deviation



**Figure S1** The broad views of several different sizes Cu<sub>2</sub>O cubes synthesized by chemical methods. (a) 51nm and (b) 25nm are TEM images, (c) and (d) are corresponding SEM images.



**Figure S2a-c:** HRTEM of catalyst heads of different Ga2O3 NWs and d is the TEM image and elemental atomic ratio of tip.