Supplementary Information

Halogen and Hydrogen Bonding in Multicomponent Crystals of Tetrabromo-1H-Benzotriazole

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Table S1. Crystallographic data for 1 and 2.

	1	2
	Crystal data	
Chemical formula	C6HBr4N3·CH3OH	$C_6H_{16}N_2 \cdot 2(C_6HBr_4N_3)$
Mr	466.78	985.68
Crystal system, space group	Monoclinic, P21/c	Monoclinic, P21/c
Temperature (K)	296	298
a, b, c (Å)	9.5254 (8), 12.4405 (10), 9.9466 (8)	9.2969 (12), 9.5812 (14), 15.468 (2)
β (°)	91.328 (10)	101.183 (12)
V (Å ³)	1178.36 (17)	1351.7 (3)
Z	4	2
Radiation type	Μο Κα	Μο Κα
μ (mm ⁻¹)	13.64	11.89
Crystal size (mm)	$0.40 \times 0.40 \times 0.12$	$0.34 \times 0.22 \times 0.12$
•	Data collection	
$T_{ ext{min}}$, $T_{ ext{max}}$	0.356, 0.745	0.427, 0.778
No. of measured, independent		
and observed $[I > 2\sigma(I)]$	8790, 2340, 1720	18998, 4483, 3326
reflections		
R_{int}	0.046	0.040
$(\sin heta/\lambda)_{ ext{max}}(\mathring{ m A}^{-1})$	0.619	0.763
	Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.031, 0.067, 1.01	0.032, 0.073, 1.03
No. of reflections	2340	4483
No. of parameters	145	160
No. of restraints	1	-
	H atoms treated by a mixture of	H atoms treated by a mixture of
H-atom treatment	independent and constrained	independent and constrained
	refinement	refinement
Δǫmax, Δǫmin (e Å–3)	0.83, -0.48	0.68, -0.86
CCDC number	1574367	1574368

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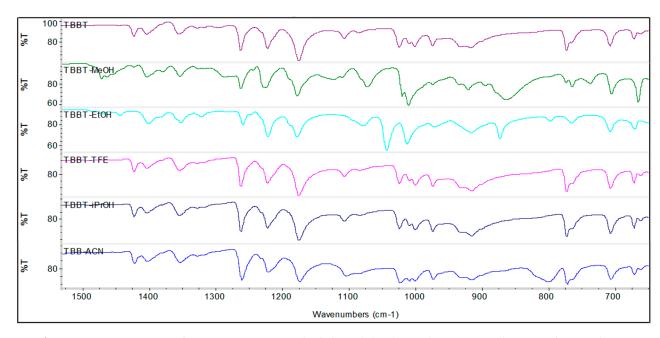


Figure S1. FTIR spectra of pure TBBT (top) and of the solids obtained on recrystallization of TBBT from methanol, ethanol, trifluoroethanol, *i*-propanol, and acetonitrile (from second to sixth spectrum in the order). Only in the case of methanol were crystals suitable for single crystal X-ray analyses obtained. It seems that pure TBBT was obtained in all cases, except that of ethanol, which might have formed a solvate, or a polymorph, of TBBT.