

## Structural Analysis, Characterization and First-Principles Calculation on Bismuth Tellurium Oxides, $\text{Bi}_6\text{Te}_2\text{O}_{15}$

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Table S1. Atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ ) for  $\text{Bi}_6\text{Te}_2\text{O}_{15}$ .

Table S2. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for  $\text{Bi}_6\text{Te}_2\text{O}_{15}$ .

Table S3. Local dipole moment calculation

Table S4. Table of the calculation for BSI (Bond Stability Index) and GII (Global Instability Index)

Figure S1. Experimental and Calculated X-ray powder diffraction patterns for  $\text{Bi}_6\text{Te}_2\text{O}_{15}$ .

Figure S2. FT-IR Spectra of  $\text{Bi}_6\text{Te}_2\text{O}_{15}$ .

Figure S3. Thermogravimetric analysis (TGA) diagram for  $\text{Bi}_6\text{Te}_2\text{O}_{15}$

Figure S4. PXRD pattern for final residuals after TGA experiment.

Figure S5. Dependence of total energy versus the unit cell volume by selected potential

Figure S6. Other optical properties

**Table S1. Atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ ) for  $\text{Bi}_6\text{Te}_2\text{O}_{15}$ .**

atom	x	y	z	$U_{\text{(eq)}}$
Bi(1)	0.3798(1)	0.5061(1)	0.7172(1)	0.01985(1)
Bi(2)	0.6384(1)	0.7500	0.2767(1)	0.02068(1)
Bi(3)	0.3899(1)	0.7500	0.7316(1)	0.01973(1)
Bi(4)	0.6161(1)	0.3777(1)	0.7659(1)	0.02154(1)
Te(1)	0.6319(1)	0.6280(1)	0.7514(2)	0.01999(1)
O(1)	0.5777(9)	0.6801(4)	1.0138(17)	0.02498(2)
O(2)	0.6901(8)	0.5781(4)	0.4910(15)	0.01923(2)
O(3)	0.5023(9)	0.6624(4)	0.5422(15)	0.02428(2)
O(4)	0.7597(8)	0.5934(4)	0.9553(16)	0.01830(2)
O(5)	0.5105(9)	0.5734(4)	0.8916(17)	0.03174(2)
O(6)	0.7467(8)	0.6878(4)	0.6174(16)	0.02544(2)
O(7)	0.5649(8)	0.4648(4)	0.6506(17)	0.01591(2)
O(8)	0.8253(11)	0.7500	0.1470(20)	0.01687(3)

$U_{\text{(eq)}}$  is defined as one third of the trace of the orthogonalized  $U^{\text{ij}}$  tensor.

**Table S2. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for  $\text{Bi}_6\text{Te}_2\text{O}_{15}$ .**

atom	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Bi(1)	20(1)	17(1)	18(1)	0(1)	1(1)	-1(1)
Bi(2)	21(1)	16(1)	18(1)	0	0(1)	0
Bi(3)	20(1)	17(1)	18(1)	0	0(1)	0
Bi(4)	22(1)	16(1)	18(1)	0(1)	0(1)	0(1)
Te(1)	20(1)	14(1)	17(1)	1(1)	0(1)	0(1)
O(1)	25(5)	19(5)	21(5)	-4(4)	3(4)	-2(4)
O(2)	19(4)	27(5)	11(4)	-6(4)	-2(3)	-1(4)
O(3)	24(5)	27(5)	11(4)	0(4)	-2(4)	0(4)
O(4)	18(4)	21(5)	20(5)	4(4)	-3(4)	5(4)
O(5)	32(5)	17(5)	22(5)	-1(4)	0(4)	-10(4)
O(6)	25(5)	20(4)	13(4)	-1(4)	-2(4)	-4(4)
O(7)	16(4)	26(5)	22(5)	-2(4)	1(4)	-2(4)
O(8)	17(6)	24(7)	16(6)	0	1(5)	0

**Table S3. Local dipole moment calculation**

The local dipole moment of the square pyramidal of  $[\text{Bi}(\text{I})\text{O}_5]^{7-}$  without lone-pair electron or with lone-pair electron is calculated.

Bi(I)O <sub>5</sub> in Bi <sub>6</sub> Te <sub>2</sub> O <sub>15</sub> without lone pair			Cart. Coord.			Distance	Unit vector				Dipole moment between two atoms		
Z	atom	charges	x	y	z	Å	x	y	z	Unitized Vector	C_grav	C_charge	u_debye
83	Bi(1)	3.029	4.019	-0.138	3.862								
8	O2	-0.521	3.280	1.772	2.741	2.335	-0.317	0.818	-0.480	1.000	0.088	0.096	8.309
8	O4	-0.308	2.749	-2.119	2.933	2.530	-0.502	-0.783	-0.367	1.000	0.088	0.094	6.645
8	O5	-0.630	5.403	-1.666	4.800	2.265	0.611	-0.675	0.414	1.000	0.088	0.097	9.142
8	O7A	-0.819	4.605	-0.798	1.881	2.169	0.270	-0.305	-0.913	1.000	0.088	0.099	10.547
8	O7B	-0.751	5.978	0.798	3.503	2.200	0.890	0.425	-0.163	1.000	0.088	0.099	10.046
			Cell Volume			1293.613				Dipole Moment			
			Z			4.000				x	y	z	Magnitude
			Total Dipole Moments =Polarization(P) =			0.057	Debye/Å <sup>3</sup>			11.406	-3.510	-13.914	18.331 Debye
Bi(I)O <sub>5</sub> (Lp) in Bi <sub>6</sub> Te <sub>2</sub> O <sub>15</sub> with lone pair			Cart. Coord.			Distance	Unit vector				Dipole moment between two atoms		
Z	atom	charges	x	y	z	Å	x	y	z	Unitized Vector	C_grav	C_charge	u_debye
83	Bi(1)	3.029	0.000	0.000	0.000								
8	O2	-0.521	0.740	-1.909	1.121	2.335	0.317	-0.818	0.480	1.000	0.088	0.096	8.309
8	O4	-0.308	1.271	1.981	0.929	2.530	0.502	0.783	0.367	1.000	0.088	0.094	6.645
8	O5	-0.630	-1.384	1.528	-0.938	2.265	-0.611	0.675	-0.414	1.000	0.088	0.097	9.142
8	O7A	-0.819	-0.586	0.660	1.981	2.169	-0.270	0.305	0.913	1.000	0.088	0.099	10.547
8	O7B	-0.751	-1.958	-0.936	0.359	2.200	-0.890	-0.425	0.163	1.000	0.088	0.099	10.046
2	LP	-2.000	0.451	-0.311	-0.812	0.980	0.460	-0.318	-0.829	1.000	0.024	0.048	9.524
			-4.248										
			Cell Volume			1293.613				Dipole Moment			
			Z			4.000				x	y	z	Magnitude
			Total Dipole Moments =Polarization(P) =			0.029	Debye/Å <sup>3</sup>			-7.020	0.484	6.021	9.261 Debye

The local dipole moment of the square pyramidal of  $[\text{Bi}(2)\text{O}_5]^{7-}$  without lone-pair electron or with lone-pair electron is calculated.

Bi(2)O <sub>5</sub> in Bi <sub>6</sub> Te <sub>2</sub> O <sub>15</sub> without lone pair			Cart. Coord.			Distance	Unit vector				Dipole moment between two atoms		
Z	atom	charges	x	y	z	Å	x	y	z	Unitized Vector	C_grav	C_charge	u_debye
83	Bi(2)	2.942	3.82678	5.6735	-6.8742								
8	O1A	-0.709	4.46908	4.08735	-5.45836	2.221	0.289	-0.714	0.637	1.000	0.088	0.098	9.651
8	O1B	-0.709	4.46908	7.25965	-5.45836	2.221	0.289	0.714	0.637	1.000	0.088	0.098	9.651
8	O6A	-0.267	2.68091	4.2613	-8.70858	2.583	-0.444	-0.547	-0.710	1.000	0.088	0.094	6.226
8	O6B	-0.267	2.68091	7.0857	-8.70858	2.583	-0.444	0.547	-0.710	1.000	0.088	0.094	6.226
8	O8	-0.989	1.84904	5.6735	-6.17416	2.098	-0.943	0.000	0.334	1.000	0.088	0.101	11.689
			Cell Volume			1293.613				Dipole Moment			
			Z			4.000				x	y	z	Magnitude
			Total Dipole Moments =Polarization(P) =			0.041	Debye/Å <sup>3</sup>	-10.960	0.000	7.362	13.203	Debye	
Bi(2)O <sub>5</sub> (Lp) in Bi <sub>6</sub> Te <sub>2</sub> O <sub>15</sub> with lone pair			Cart. Coord.			Distance	Unit vector				Dipole moment between two atoms		
Z	atom	charges	x	y	z	Å	x	y	z	Unitized Vector	C_grav	C_charge	u_debye
83	Bi(2)	2.942	0.000	0.000	0.000								
8	O1A	-0.709	-0.642	1.586	-1.416	2.221	-0.289	0.714	-0.637	1.000	0.088	0.098	9.651
8	O1B	-0.709	-0.642	-1.586	-1.416	2.221	-0.289	-0.714	-0.637	1.000	0.088	0.098	9.651
8	O6A	-0.267	1.146	1.412	1.834	2.583	0.444	0.547	0.710	1.000	0.088	0.094	6.226
8	O6B	-0.267	1.146	-1.412	1.834	2.583	0.444	-0.547	0.710	1.000	0.088	0.094	6.226
8	O8	-0.989	1.978	0.000	-0.700	2.098	0.943	0.000	-0.334	1.000	0.088	0.101	11.689
2	LP	-2.000	-0.979	0.000	-0.045	0.980	-0.999	0.000	-0.046	1.000	0.024	0.048	9.509
			-3.050										
			Cell Volume			1293.613				Dipole Moment			
			Z			4.000				x	y	z	Magnitude
			Total Dipole Moments =Polarization(P) =			0.025	Debye/Å <sup>3</sup>	1.461	0.000	-7.798	7.934	Debye	

The local dipole moment of the square pyramidal of  $[\text{Bi}(3)\text{O}_5]^{7-}$  without lone-pair electron or with lone-pair electron is calculated.

Bi(3)O <sub>5</sub> in Bi <sub>6</sub> Te <sub>2</sub> O <sub>15</sub> without lone pair			Cart. Coord.			Distance	Unit vector				Dipole moment between two atoms		
Z	atom	charges	x	y	z	Å	x	y	z	Unitized Vector	C_grav	C_charge	u_debye
83	Bi(3)	2.878	1.165	5.674	-4.138								
8	O3A	-0.709	-0.024	7.662	-5.157	2.531	-0.470	0.786	-0.403	1.000	0.088	0.098	10.931
8	O3B	-0.709	-0.024	3.685	-5.157	2.531	-0.470	-0.786	-0.403	1.000	0.088	0.098	10.931
8	O6A	-0.267	2.681	4.261	-3.324	2.226	0.681	-0.634	0.365	1.000	0.088	0.094	5.305
8	O6B	-0.267	2.681	7.086	-3.324	2.226	0.681	0.634	0.365	1.000	0.088	0.094	5.305
8	O8	-0.989	1.849	5.674	-6.174	2.148	0.318	0.000	-0.948	1.000	0.088	0.101	11.912
			Cell Volume			1293.613					Dipole Moment		
			Z			4.000					Magnitude		
			Total Dipole Moments =Polarization(P) =			0.050	Debye/Å <sup>3</sup>	0.751	0.000	-16.223	16.240	Debye	
Bi(3)O <sub>5</sub> in Bi <sub>6</sub> Te <sub>2</sub> O <sub>15</sub> with lone pair			Cart. Coord.			Distance	Unit vector				Dipole moment between two atoms		
Z	atom	charges	x	y	z	Å	x	y	z	Unitized Vector	C_grav	C_charge	u_debye
83	Bi(3)	2.878	0.000	0.000	0.000								
8	O3A	-0.709	1.189	-1.988	1.020	2.531	0.470	-0.786	0.403	1.000	0.088	0.098	10.931
8	O3B	-0.709	1.189	1.988	1.020	2.531	0.470	0.786	0.403	1.000	0.088	0.098	10.931
8	O6A	-0.267	-1.516	1.412	-0.813	2.226	-0.681	0.634	-0.365	1.000	0.088	0.094	5.305
8	O6B	-0.267	-1.516	-1.412	-0.813	2.226	-0.681	-0.634	-0.365	1.000	0.088	0.094	5.305
8	O8	-0.989	-0.684	0.000	2.037	2.148	-0.318	0.000	0.948	1.000	0.088	0.101	11.912
2	LP	-2.000	0.470	0.000	-0.860	0.980	0.479	0.000	-0.878	1.000	0.024	0.048	9.510
			-2.847										
			Cell Volume			1293.613					Dipole Moment		
			Z			4.000					Magnitude		
			Total Dipole Moments =Polarization(P) =			0.027	Debye/Å <sup>3</sup>	3.808	0.000	7.877	8.750	Debye	

The local dipole moment of the square pyramidal of  $[\text{Bi}(4)\text{O}_5]^{7-}$  without lone-pair electron or with lone-pair electron is calculated.

Bi(4)O <sub>5</sub> in Bi <sub>6</sub> Te <sub>2</sub> O <sub>15</sub> without lone pair			Cart. Coord.			Distance	Unit vector				Dipole moment between two atoms		
Z	atom	charges	x	y	z	Å	x	y	z	Unitized Vector	C_grav	C_charge	u_debye
83	Bi(4)	2.770	1.229	2.775	3.953								
8	O2	-0.709	3.280	1.772	2.741	2.585	0.793	-0.388	-0.469	1.000	0.088	0.098	11.045
8	O3	-0.709	-0.024	3.685	5.611	2.269	-0.552	0.401	0.731	1.000	0.088	0.098	9.695
8	O4	-0.267	2.543	2.119	5.625	2.226	0.590	-0.295	0.751	1.000	0.088	0.093	5.203
8	O5	-0.267	-0.112	1.666	2.108	2.536	-0.529	-0.437	-0.727	1.000	0.088	0.093	5.927
8	O7	-0.989	0.686	0.798	4.573	2.142	-0.253	-0.923	0.290	1.000	0.088	0.101	11.776
			Cell Volume			1293.613				Dipole Moment			
			Z			4.000				x	y	z	Magnitude
			Total Dipole Moments=			0.050	Debye/Å <sup>3</sup>			0.364	-15.394	4.918	16.165 Debye
			Polarization(P) =										
Bi(4)O <sub>5</sub> in Bi <sub>6</sub> Te <sub>2</sub> O <sub>15</sub> with lone pair			Cart. Coord.			Distance	Unit vector				Dipole moment between two atoms		
Z	atom	charges	x	y	z	Å	x	y	z	Unitized Vector	C_grav	C_charge	u_debye
83	Bi(4)	2.770	0.000	0.000	0.000								
8	O2	-0.709	-2.051	1.003	1.212	2.585	-0.793	0.388	0.469	1.000	0.088	0.098	11.045
8	O3	-0.709	1.253	-0.910	-1.659	2.269	0.552	-0.401	-0.731	1.000	0.088	0.098	9.695
8	O4	-0.267	-1.314	0.656	-1.672	2.226	-0.590	0.295	-0.751	1.000	0.088	0.093	5.203
8	O5	-0.267	1.340	1.109	1.845	2.536	0.529	0.437	0.727	1.000	0.088	0.093	5.927
8	O7	-0.989	0.543	1.977	-0.621	2.142	0.253	0.923	-0.290	1.000	0.088	0.101	11.776
2	LP	-2.000	0.057	-0.953	0.222	0.980	0.058	-0.972	0.227	1.000	0.024	0.047	9.495
			-4.025										
			Cell Volume			1293.613				Dipole Moment			
			Z			4.000				x	y	z	Magnitude
			Total Dipole Moments			0.021	Debye/Å <sup>3</sup>			0.187	6.163	-2.764	6.757 Debye
			=Polarization(P) =										

The local dipole moment of the octahedral of [Te(1)O<sub>6</sub>]<sup>6-</sup> is also calculated.

Te(1)O <sub>6</sub> in Bi <sub>6</sub> Te <sub>2</sub> O <sub>15</sub>			Cart. Coord.			Distance	Unit vector				Dipole moment between two ato		
Z	atom	charges	x	y	z	Å	x	y	z	Unitized Vector	C_grav	C_charge	u_debye
52	Te	5.797	-1.395	2.906	-4.031								
8	O1	-0.968	-0.822	4.087	-2.618	1.929	0.297	0.613	0.732	1.000	0.133	0.163	14.922
8	O2	-1.030	-2.012	1.772	-5.433	1.906	-0.323	-0.595	-0.736	1.000	0.133	0.163	15.236
8	O3	-0.945	-0.024	3.685	-5.157	1.938	0.708	0.402	-0.581	1.000	0.133	0.162	14.811
8	O4	-1.014	-2.749	2.119	-2.933	1.912	-0.708	-0.412	0.574	1.000	0.133	0.163	15.159
8	O5	-0.945	-0.112	1.666	-3.276	1.938	0.662	-0.640	0.389	1.000	0.133	0.162	14.806
8	O6	-0.895	-2.611	4.261	-4.752	1.958	-0.621	0.692	-0.368	1.000	0.133	0.161	14.558
			Cell Volume			1293.613	Dipole Moment				Magnitude		
			Z			4.000							
			Total Dipole Moments =Polarization(P) =			0.001	Debye/Å^3	0.030	0.392	0.219	0.450	Debye	



Table S4. Table of the calculation for BSI (Bond Stability Index) and GII (Global Instability Index)

BSI = 0.172 GII = 0.279		Bi(1)		Bi(2)	Bi(3)	Bi(4)		Te(1)		$\sum_i (S_{ij}-s_{ij})^2$	$V_i = \sum_j S_{ij}$ $z_i = \sum_j s_{ij}$	$V_i - z_i$	$(V_i - z_i)^2$
		1	1	1	1	1	1	1	1				
O(1)	Total Connectivity			2				1	1	0.289	3.355 4.000	-0.645	0.416
	Each Connection			2				1	1				
	$R_{ij}$			2.221				1.929	1.929				
	$S_{ij}$			0.709				0.968	0.968				
	$s_{ij}$			0.656				1.345	1.345				
	$S_{ij}-s_{ij}$			0.054				-0.376	-0.376				
	$(S_{ij}-s_{ij})^2$			0.003				0.142	0.142				
2													
O(2)	Total Connectivity	1	1			1	1	1	1	0.228	3.634 4.000	-0.366	0.134
	Each Connection	1	1			1	1	1	1				
	$R_{ij}$	2.335	2.335			2.585	2.585	1.906	1.906				
	$S_{ij}$	0.521	0.521			0.265	0.265	1.030	1.030				
	$s_{ij}$	0.557	0.557			0.565	0.565	0.878	0.878				
	$S_{ij}-s_{ij}$	-0.036	-0.036			-0.299	-0.299	0.152	0.152				
	$(S_{ij}-s_{ij})^2$	0.001	0.001			0.090	0.090	0.023	0.023				
2													
O(3)	Total Connectivity				2	1	1	1	1	0.025	3.750 4.000	-0.250	0.063
	Each Connection				2	1	1	1	1				
	$R_{ij}$				2.531	2.269	2.269	1.938	1.938				
	$S_{ij}$				0.307	0.623	0.623	0.945	0.945				
	$s_{ij}$				0.418	0.634	0.634	0.948	0.948				
	$S_{ij}-s_{ij}$				-0.111	-0.011	-0.011	-0.003	-0.003				
	$(S_{ij}-s_{ij})^2$				0.012	0.000	0.000	0.000	0.000				
2													
O(4)	Total Connectivity	1	1			1	1	1	1	0.198	4.043 4.000	0.043	0.002
	Each Connection	1	1			1	1	1	1				
	$R_{ij}$	2.530	2.530			2.226	2.226	1.912	1.912				
	$S_{ij}$	0.308	0.308			0.700	0.700	1.014	1.014				
	$s_{ij}$	0.557	0.557			0.565	0.565	0.878	0.878				
	$S_{ij}-s_{ij}$	-0.249	-0.249			0.135	0.135	0.136	0.136				
	$(S_{ij}-s_{ij})^2$	0.062	0.062			0.018	0.018	0.018	0.018				
2													
O(5)	Total Connectivity	1	1			1	1	1	1	0.157	3.755 4.000	-0.245	0.060
	Each Connection	1	1			1	1	1	1				
	$R_{ij}$	2.265	2.265			2.536	2.536	1.938	1.938				
	$S_{ij}$	0.630	0.630			0.303	0.303	0.945	0.945				
	$s_{ij}$	0.557	0.557			0.565	0.565	0.878	0.878				
	$S_{ij}-s_{ij}$	0.073	0.073			-0.262	-0.262	0.067	0.067				
	$(S_{ij}-s_{ij})^2$	0.005	0.005			0.069	0.069	0.004	0.004				
2													
O(6)	Total Connectivity			2	2			1	1	0.140	3.724 4.000	-0.276	0.076
	Each Connection			2	2			1	1				
	$R_{ij}$			2.583	2.226			1.958	1.958				
	$S_{ij}$			0.267	0.700			0.895	0.895				
	$s_{ij}$			0.384	0.543			1.073	1.073				
	$S_{ij}-s_{ij}$			-0.117	0.157			-0.178	-0.178				
	$(S_{ij}-s_{ij})^2$			0.014	0.025			0.032	0.032				
2													
O(7)	Total Connectivity	2	2			1	1			0.148	4.896 4.000	0.896	0.803
	Each Connection	2	2			1	1						
	$R_{ij}$	2.168	2.200			2.142	2.142						
	$S_{ij}$	0.819	0.751			0.878	0.878						
	$s_{ij}$	0.664	0.664			0.672	0.672						
	$S_{ij}-s_{ij}$	0.155	0.087			0.207	0.207						
	$(S_{ij}-s_{ij})^2$	0.024	0.008			0.043	0.043						
2													
O(8)	Total Connectivity			1	1					0.051	1.853 2.000	-0.147	0.021
	Each Connection			1	1								
	$R_{ij}$			2.098	2.148								
	$S_{ij}$			0.989	0.864								
	$s_{ij}$			0.921	1.079								
	$S_{ij}-s_{ij}$			0.069	-0.215								
	$(S_{ij}-s_{ij})^2$			0.005	0.046								
1													
$\sum_i (S_{ij}-s_{ij})^2$		0.117	0.084	0.038	0.120	0.219	0.219	0.219	0.219	1.236			
$V_i = \sum_j S_{ij}$		3.096	2.961	2.942	2.878	2.770	2.770	5.797	5.797				
$z_i = \sum_j s_{ij}$		3.000	3.000	3.000	3.000	3.000	3.000	6.000	6.000				
$V_i - z_i$		0.096	-0.039	-0.058	-0.122	-0.230	-0.230	-0.203	-0.203				
$(V_i - z_i)^2$		0.009	0.002	0.003	0.015	0.053	0.053	0.041	0.041	0.218			1.575

**Figure S1. Experimental and Calculated X-ray Powder Diffraction Patterns for  $\text{Bi}_6\text{Te}_2\text{O}_{15}$ .**

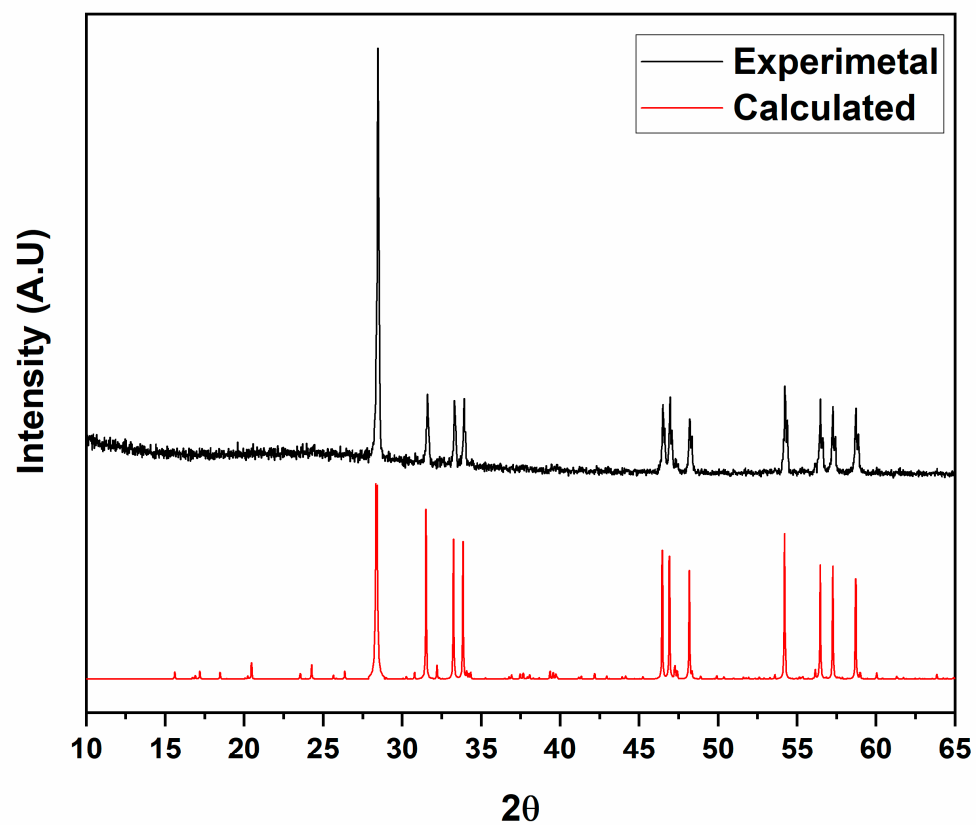
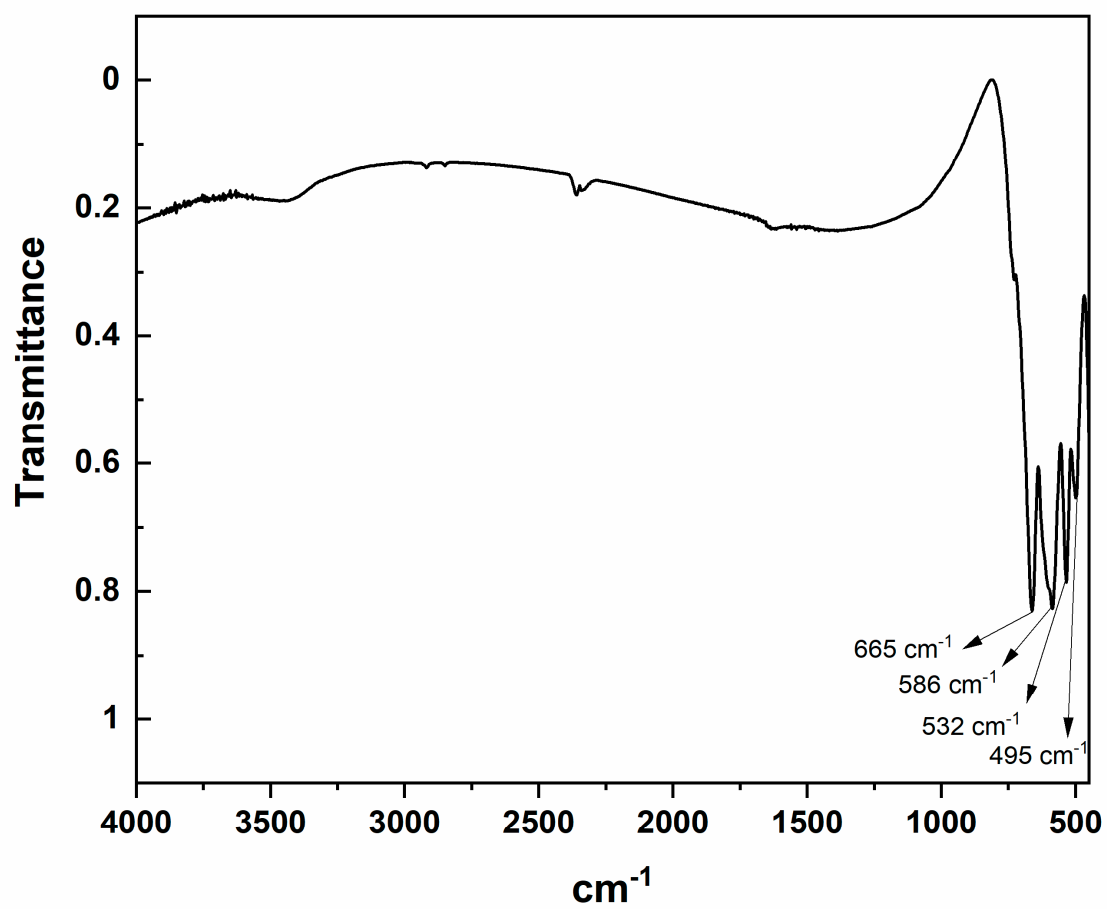


Figure S2. FT-IR Spectra of  $\text{Bi}_6\text{Te}_2\text{O}_{15}$ .



$\text{Bi}_6\text{Te}_2\text{O}_{15}$	
$\nu(\text{Bi-O})$	$\nu(\text{Te-O})$
532	665
495	586

Figure S3. Thermogravimetric analysis (TGA) diagram for  $\text{Bi}_6\text{Te}_2\text{O}_{15}$ .

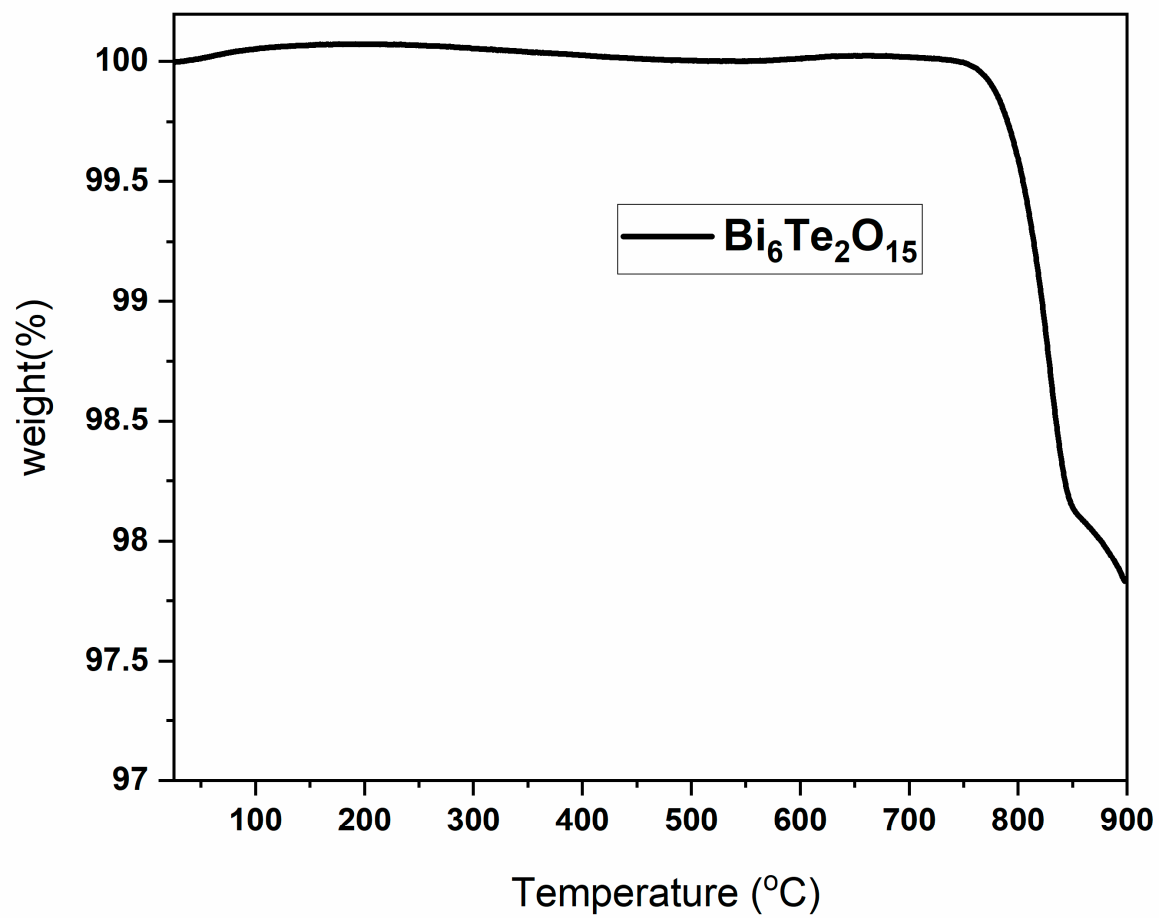
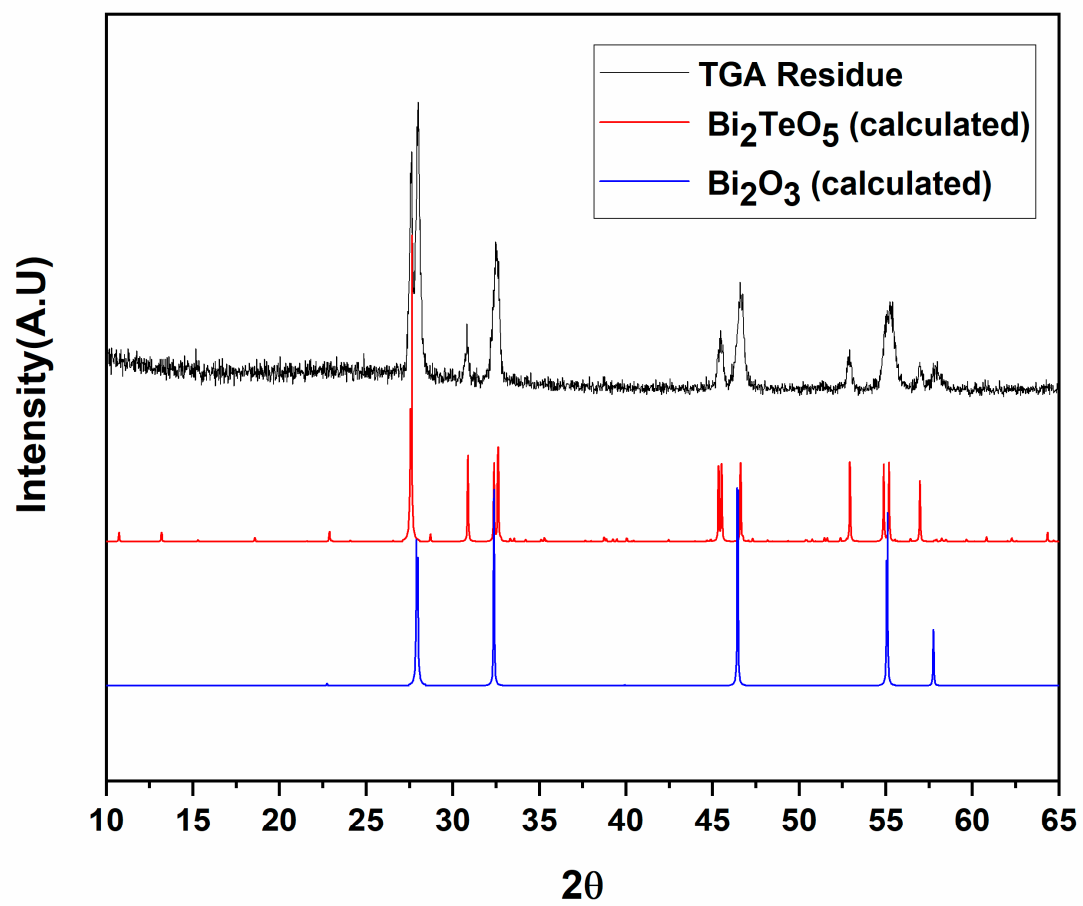
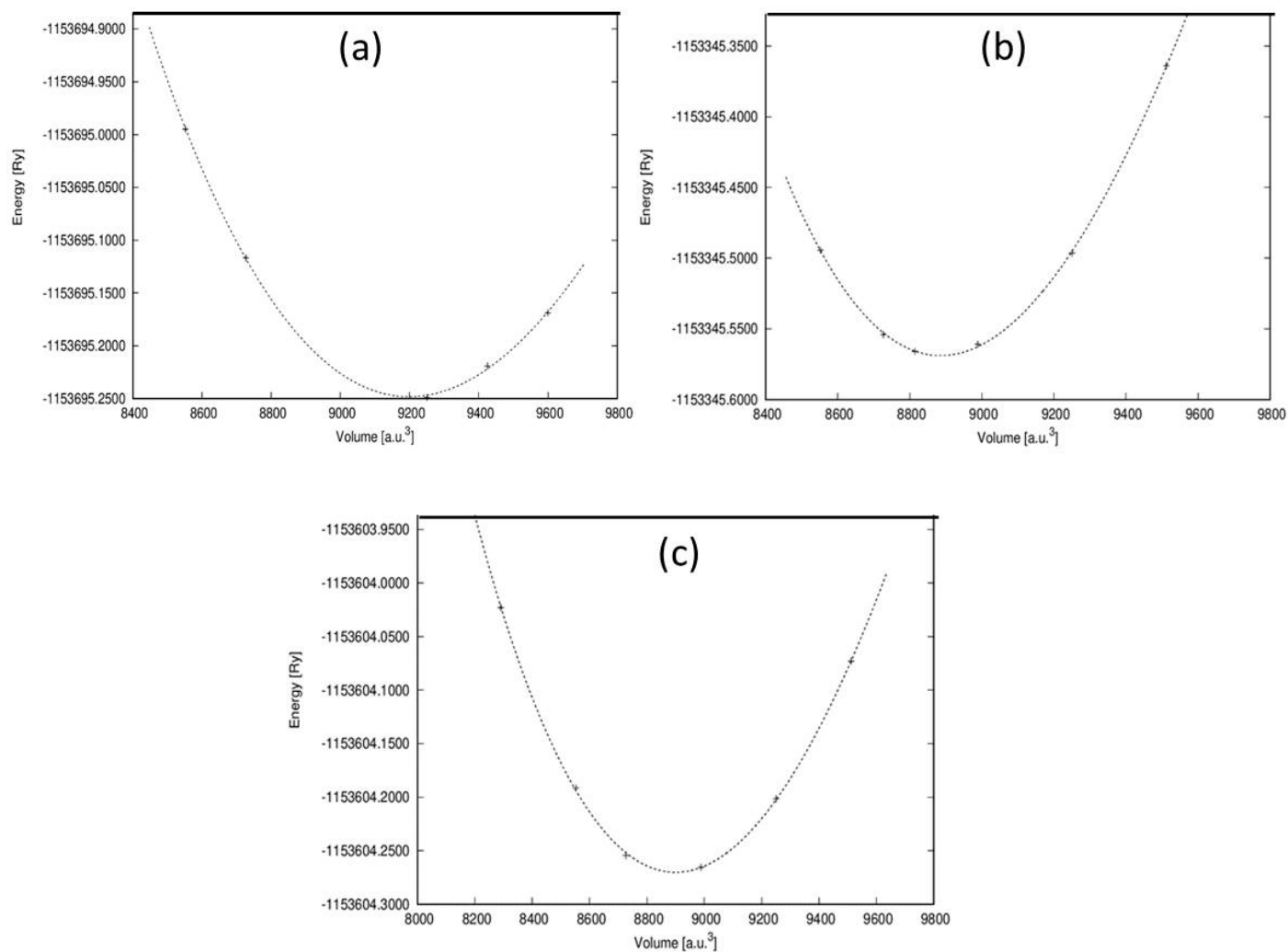


Figure S4. Powder XRD pattern for final residuals after TGA experiment



**Figure S5. Dependence of total energy *versus* the unit cell volume by selected potential**

(a) within GGA-PBE potential (b) within GGA-PBESol (c) within GGA-WC, respectively.



## Figure S6. Other optical properties

Reflectivity  $R(\omega)$ , Optical conductivity  $\sigma(\Omega\cdot\text{cm})^{-1}$ , Electron loss function  $L(\omega)$ , and Extinction coefficient  $k(\omega)$  are depicted.

