



Editorial Editorial: Semiconductor Photocatalysts

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Since the discovery of the photocatalytic ability of TiO₂ electrodes to decompose water [1], techniques of semiconductor photocatalysis, including photocatalytic energy conversion and photocatalytic pollution treatment, have achieved rapid development. In the contemporary world, these concerns animate the interdisciplinary fields of semiconductor physics, materials science, environmental and energy science, computational chemistry and many others [2–9]. The overall photocatalysis process typically incorporates light absorption, charge generation/separation/transfer, and surface redox reactions [5,6]. These three steps are complementary and indispensable, and only when they executed with a high degree of efficiency can the overall photocatalytic efficiency be high. A series of strategies, such as foreign element doping [10,11], metal loading [12], heterojunction construction [13,14], strain and external electric field regulation [15,16], have been developed to overcome the bottlenecks of each step in order to improve the photocatalytic efficiency. In this Special Issue of *Crystals*, entitled "Semiconductor Photocatalysis", we have collected a total of 4 articles about the recent progress in semiconductor photocatalysis. Next, we briefly outline the research highlights of these studies.

Alhalili et al. [17] studied the effect of calcination time on TiO₂ nanoparticles (NPs), prepared from Aloe vera leaf extract, when the temperature was maintained at 500 °C and discussed the relationship between calcination time and the size of NP. The size of synthesized TiO₂ NP decreases with the increase in calcination time. The NP with small size possesses improved optical and photocatalytic activity. The visible light photocatalytic ability for RR180 degradation varies with time, from 1 h with TiO₂ NP (23 \pm 2 nm) to 2 h with TiO₂ NP (83 \pm 5 nm).

Xiao et al. [18] used molten salt and ultrasound-assisted liquid-phase exfoliation methods to successfully prepare a black phosphorus-/heptazine-based crystalline carbon nitride (BP/KPHI) composite. The photocatalytic hydrogen production performance of BP/KPHI composites can be tuned by altering the mass ration of BP. The 10% BP/KPHI composite shows the highest photocatalytic hydrogen production rate of 4.3 mmol·g⁻¹·h⁻¹, which is about three times higher than that of KPHI. The excellent photocatalytic performance of BP/KPHI composite can primarily be attributed to the stellar photoinduced carrier separation and excellent visible light harvesting capacity.

First-principles calculation based on density functional theory (DFT) plays an undeniable role in developing and designing novel semiconductor photocatalysts. Li et al. [19] adopted first-principles calculation methods to investigate the geometric, electronic, optical properties as well as the hydrogen evolution reaction (HER) and carrier mobility of SiP₂ monolayers (MLs) in order to explore their potential use in photocatalytic hydrogen production. SiP₂ MLs are indirect bandgap semiconductors with 2.277 bandgaps that are still stable at 1200 K. The suitable band edge alignment and strong light absorption ability



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). make SiP₂ MLs into potential water-splitting photocatalyst. In addition, SiP₂ MLs have excellent electron mobility of 33,153 cm²·V⁻¹·S⁻¹. The calculated hydrogen adsorption free energy shows that SiP₂ ML possesses better HER ability compared to that of graphene.

Recently, Ren et al. [20] constructed different stacked MoTe₂/PtS₂ van der Waals heterostructures (vdWHs) and explored their electronic characteristics by using first-principles calculation. By changing stacking configurations, the MoTe₂/PtS₂ vdWHs could achieve a transition from type-I to type-II, which respectively possess potential applications in light emitting diode and photocatalysis. The type-II MoTe₂/PtS₂ vdWH has adequate bandedge alignment for overall water-splitting when pH is 0. In addition, MoTe₂/PtS₂ vdWHs possess excellent visible light absorption capacity. All the results show that MoTe₂/PtS₂ vdWHs are promising candidates as photocatalytic and photovoltaic devices.

We hope that this Special Issue of *Crystals*, entitled "Semiconductor Photocatalysts", will be able to provide assistance and guidance for the development and design of novel semiconductor photocatalysts.

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References

- 1. Fujishima, A.; Honda, K. Electrochemical photolysis of water at a semiconductor electrode. *Nature* **1972**, *238*, 37–38. [CrossRef] [PubMed]
- Chen, X.; Shen, S.; Guo, L.; Mao, S.S. Semiconductor-based photocatalytic hydrogen generation. *Chem. Rev.* 2010, 110, 6503–6570. [CrossRef] [PubMed]
- Iwashina, K.; Kudo, A. Rh-doped SrTiO₃ photocatalyst electrode showing cathodic photocurrent for water splitting under visible-light irradiation. *J. Am. Chem. Soc.* 2011, 133, 13272–13275. [CrossRef] [PubMed]
- 4. Schultz, D.M.; Yoon, T.P. Solar synthesis: Prospects in visible light photocatalysis. Science 2014, 343, 1239176. [CrossRef] [PubMed]
- Fu, C.F.; Wu, X.; Yang, J. Material design for photocatalytic water splitting from a theoretical perspective. *Adv. Mater.* 2018, 30, 1802106. [CrossRef] [PubMed]
- Wang, G.; Chang, J.; Tang, W.; Xie, W.; Ang, Y.S. 2D materials and heterostructures for photocatalytic water-splitting: A theoretical perspective. J. Phys. D Appl. Phys. 2022, 55, 293002. [CrossRef]
- Zhang, X.; Chen, A.; Chen, L.; Zhou, Z. 2D materials bridging experiments and computations for electro/photocatalysis. *Adv. Energy Mater.* 2022, 12, 2003841. [CrossRef]
- Sun, L.J.; Su, H.W.; Liu, Q.Q.; Hu, J.; Wang, L.L.; Tang, H. A review on photocatalytic systems capable of synchronously utilizing photogenerated electrons and holes. *Rare Met.* 2022, 41, 2387–2404. [CrossRef]
- Ye, L.; Peng, X.; Wen, Z.; Huang, H. Solid-state Z-scheme assisted hydrated tungsten trioxide/ZnIn₂S₄ photocatalyst for efficient photocatalytic H₂ production. *Mater. Futures* 2022, 1, 035103. [CrossRef]
- 10. Maeda, K.; Domen, K. Photocatalytic water splitting: Recent progress and future challenges. J. Phys. Chem. Lett. 2010, 1, 2655–2661. [CrossRef]
- 11. Tan, C.; Cao, X.; Wu, X.J.; He, Q.; Yang, J.; Zhang, X.; Chen, J.; Zhao, W.; Han, S.; Nam, G.-H.; et al. Recent advances in ultrathin two-dimensional nanomaterials. *Chem. Rev.* 2017, 117, 6225–6331. [CrossRef] [PubMed]
- 12. Bai, S.; Jiang, J.; Zhang, Q.; Xiong, Y. Steering charge kinetics in photocatalysis: Intersection of materials syntheses, characterization techniques and theoretical simulations. *Chem. Soc. Rev.* **2015**, *44*, 2893–2939. [CrossRef] [PubMed]
- Moniz SJ, A.; Shevlin, S.A.; Martin, D.J.; Guo, Z.-X.; Tang, J. Visible-light driven heterojunction photocatalysts for water splitting–a critical review. *Energy Environ. Sci.* 2015, *8*, 731–759. [CrossRef]
- 14. Wang, G.; Tang, W.; Xie, W.; Tang, Q.; Wang, Y.; Guo, H.; Gao, P.; Dang, S.; Chang, J. Type-II CdS/PtSSe heterostructures used as highly efficient water-splitting photocatalysts. *Appl. Surf. Sci.* **2022**, *589*, 152931. [CrossRef]
- 15. Wang, G.; Zhang, L.; Li, Y.; Zhao, W.; Kuang, A.; Li, Y.; Xia, L.; Li, Y.; Xiao, S. Biaxial strain tunable photocatalytic properties of 2D ZnO/GeC heterostructure. *J. Phys. D Appl. Phys.* **2020**, *53*, 015104. [CrossRef]
- 16. Li, X.; Wang, W.; Dong, F.; Zhang, Z.; Han, L.; Luo, X.; Huang, J.; Feng, Z.; Chen, Z.; Jia, G.; et al. Recent advances in noncontact external-field-assisted photocatalysis: From fundamentals to applications. *ACS Catal.* **2021**, *11*, 4739–4769. [CrossRef]

- 17. Alhalili, Z.; Smiri, M. The Influence of the Calcination Time on Synthesis of Nanomaterials with Small Size, High Crystalline Nature and Photocatalytic Activity in the TiO₂ Nanoparticles Calcined at 500 °C. *Crystals* **2022**, *12*, 1629. [CrossRef]
- Xiao, Z.; Wang, Y.; Chen, H.; Wang, H.; Li, Y.; Chen, Y.; Zheng, Y. Designing Black Phosphorus and Heptazine-Based Crystalline Carbon Nitride Composites for Photocatalytic Water Splitting. *Crystals* 2023, 13, 312. [CrossRef]
- 19. Li, J.; Pan, H.; Sun, H.; Zheng, R.; Ren, K. First-Principle Study of Two-Dimensional SiP2 for Photocatalytic Water Splitting with Ultrahigh Carrier Mobility. *Crystals* **2023**, *13*, 981. [CrossRef]
- Ren, K.; Zhu, Z.; Wang, K.; Huo, W.; Cui, Z. Stacking-mediated type-I/type-II transition in two-dimensional MoTe2/PtS2 heterostructure: A first-principles simulation. *Crystals* 2022, 12, 425. [CrossRef]

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