



Article Three-Dimensional MoS₂ Nanosheet Structures: CVD Synthesis, Characterization, and Electrical Properties

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Abstract: The proposed study demonstrates a single-step CVD method for synthesizing three-dimensional vertical MoS₂ nanosheets. The postulated synthesizing approach employs a temperature ramp with a continuous N₂ gas flow during the deposition process. The distinctive signals of MoS₂ were revealed via Raman spectroscopy study, and the substantial frequency difference in the characteristic signals supported the bulk nature of the synthesized material. Additionally, XRD measurements sustained the material's crystallinity and its 2H-MoS₂ nature. The FIB cross-sectional analysis provided information on the origin and evolution of the vertical MoS₂ structures and their growth mechanisms. The strain energy produced by the compression between MoS₂ islands is assumed to primarily drive the formation of vertical MoS₂ nanosheets. In addition, vertical MoS₂ structures that emerge from micro fissures (cracks) on individual MoS₂ islands were observed and examined. For the evaluation of electrical properties, field-effect transistor structures were fabricated on the synthesized material employing standard semiconductor technology. The lateral back-gated field-effect transistors fabricated on the synthesized material showed an n-type behavior with field-effect mobility of 1.46 cm² V^{-1} s⁻¹ and an estimated carrier concentration of 4.5×10^{12} cm⁻². Furthermore, the effects of a back-gate voltage bias and channel dimensions on the hysteresis effect of FET devices were investigated and quantified.

Keywords: molybdenum disulfide; vertical nanosheet; field-effect transistors; hysteresis area

1. Introduction

Transition metal dichalcogenides (TMDs) have received much attention in advanced material science and technology due to their unique material properties. Among this material class, molybdenum disulfide (MoS₂) has been extensively used in many device applications [1,2]. Moreover, its unique semiconductor properties have gained much recognition among 2D materials for various promising applications. As a result, many bottom-up approaches have been developed to synthesize large-scale MoS₂, including sulfurization of the Mo layer [3–7] and chemical vapor deposition (CVD) [8–13]. However, among the numerous vapor phase reactions, researchers mainly emphasized horizontally grown 2D materials on planar substrates rather than the vertical formation of MoS₂ structures.

Nevertheless, some research has recently been conducted on the vertical formation of MoS_2 nanostructures [14–17]. In addition, several studies have been made to analyze the growth mechanism of this vertical formation of MoS_2 [14,17]. In vertical MoS_2 , the physical properties may vary due to its growth direction, making it a promising candidate for diverse applications. For example, the high aspect ratio, presence of dangling bonds, and extensively exposed edges with a high absorption rate of vertical MoS_2 nanosheets show



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). outstanding gas-sensing properties [18,19]. Furthermore, vertically exposed structures with chemically active sites and catalytically active zones make them suitable for photocatalytic, hydrogen evolution reactions (HER) and many other promising applications [16,20,21].

Additionally, the vertically formed MoS₂ nanoflakes exhibit excellent optical absorption with enhanced photoresponsivity and interlayer transport, making them a highly efficient material for photodetection, optoelectronic applications [22,23], and field emission applications [24]. However, the studies based on the formation and electrical properties of the vertical MoS₂ structures have not received much attention in research. Although numerous investigations on horizontal MoS₂-based planar devices with multi-terminal designs have been conducted, understanding of the structural development and electrical characterization of devices based on vertical MoS₂ nanosheets remains sporadic.

This study presents the development and investigation of a reliable, single-step CVD method for synthesizing three-dimensional (3D) vertical MoS₂ nanosheets. The developed synthesizing approach uses the vapor phase reaction of solid precursors with a ramping temperature and a constant N₂ gas flow for the vertical growth of MoS₂ nanosheets. In addition, the time-dependent growth morphologies of MoS₂ were analyzed using scanning electron microscope (SEM) and focused ion beam (FIB), and the growth mechanism and evolution of the vertical structures was examined. Employing standard semiconductor technology, several lateral backgated field effect transistors were fabricated using the synthesized material. Furthermore, the electrical properties were determined by measuring the FET devices in ambient environment.

2. Materials and Methods

2.1. Growth Method

For the preferential growth of vertical MoS_2 thin films, we employed the vapor phase reactions of the precursor materials molybdenum (MoO_3) and sulfur (S). The developed process was performed in a 1.2 m long quartz tube with inner and outer diameters of 4.6 and 5 cm, respectively, as shown in Figure S1 (supplementary information). In a typical growth process, silicon substrates (1 \times 1 cm) with a 90 nm thick SiO₂ layer were initially cleaned in acetone, isopropanol, and a deionized water bath for 10 min, and subsequently dried using nitrogen gas. The substrates were then inserted into the quartz tube and positioned in the center of a single-zone horizontal tube furnace. After that, we placed a ceramic boat containing the precursor MoO_3 powder (0.3 mg) (99%, Sigma-Aldrich) in the center of the quartz tube (40 cm from the inlet). The sulfur (S, 0.6 mg) (99% Sigma-Aldrich) source was positioned at a lower temperature position (27 cm upstream from the inlet). The furnace was then purged with 500 sccm of nitrogen for one hour. After purging, the furnace's temperature was increased to a predetermined level of 850 °C using a linear heating ramp that increased by 28 °C per minute with a flow of 500 sccm of nitrogen gas. The heating ramp was slowed down once 850 °C was reached. The temperatures were monitored using thermocouples. The sulfur was kept at a constant temperature of 240 °C. The growth process was carried out for different growth times that varied from 10 min to 30 min. Finally, a 500 sccm nitrogen flow operating at atmospheric pressure brought the system down to ambient temperature. The temperature-time profile of the whole CVD process is given in Figure S1 (supplementary information).

2.2. Characterization of Synthesized 3D Verical MoS₂ Nanosheet

The surface morphology and topographical details of the vertical MoS₂ nanosheets were analyzed using optical microscopy and SEM. A cross-sectional study also has been performed with a FIB device (Zeiss Auriga 60 dual beam) to analyze the inner morphology of the vertical structures. Before cutting, protective carbon and platinum layers were deposited sequentially via an electron beam. A gallium ion beam was used to realize the cuts. Furthermore, we investigated the material properties with Raman spectroscopy and X-ray diffraction (XRD). The Raman spectra were obtained using a WiTec Alpha 300R with

532 nm excitation. X-ray diffraction (XRD) was performed using a Siemens D5000 with a copper anode (radiation: 0.15406 nm for K α 1 (Bragg—Brentano mode)).

2.3. Device Fabrication

The devices were fabricated using a maskless aligner (MLA) photolithography tool from the Heidelberg Instrument. First, the CVD-grown MoS₂ samples were cleaned, prebaked, and coated with an adhesion promoter, hexamethyldisilazane (HMDS). Later, the samples were spin-coated at 4000 rpm with a positive resist (AZ1505). After the material was structured by image inversion using MLA, the exposed structures were developed in AZ 351-B for 30 s. Subsequently, the samples were etched using Cl₂ and O₂ plasma for 1 min, and the resist was removed by acetone. After etching, the samples were again coated with HMDS and later spin-coated with another positive resist (AZ 1518) at 4000 rpm to pattern the metal contacts. Then, the patterned resist was developed in AZ 315-B and metalized with Titanium (Ti-10 nm) and gold (Au-80 nm) using electron beam evaporation. The metal deposited on the resist was removed in a lift-off step in dimethyl sulfoxide (DMSO) for 30 min.

3. Results and Discussion

3.1. Formation and Characterization of the Three-Dimensional MoS₂ Nanosheet

A typical optical micrograph image obtained from the MoS_2 nanosheet sample surface is shown in Figure 1a. The homogeneous interference color demonstrates a homogeneous deposition on the substrate. Figure 1b,c show SEM images. In them, a three-dimensional (3D) MoS_2 nanosheet structure can be observed.



Figure 1. (a) Optical microscope image of the material surface. (b) Typical SEM image of a vertical MoS₂ nanosheet. (c) Randomly grown vertical MoS₂ nanosheet. (d) Cross-sectional FIB image of vertical MoS₂.

The MoS_2 has randomly grown vertically from the substrate surface, forming three-dimensional vertical nanostructures representing a quasiperiodic modulation of the material surface. Each nanosheet has a flower petal-like shape with sub-20 nm dimension thin edges, as illustrated in Figure 1c. Furthermore, these vertical structures have nano-sized edge sites with multiple grain boundaries. On the substrates, these nano platelets are grown dominantly in the vertical direction, exposing their edge sites rather than their basal planes. Moreover, these structures often possess a high aspect ratio with more edge sites suitable for many applications.

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A cross-sectional examination was conducted to better understand the early stages of the vertical MoS_2 formation. The cross-section samples were prepared using focused ion beam (FIB) technology. The general morphology of vertically free-standing MoS₂ nanosheets grown on a SiO₂/Si substrate can be observed in the cross-section FIB picture shown in Figure 1d. Most of the vertical structures are oriented perpendicular to the substrate surface. The basal plane of MoS_2 is a bulk layer and has a height of 32 nm. The bulk layer is buckled. The buckling is highlighted in Figure S2 with arrows. The buckling is evidence of compressive stress in the deposited polycrystalline 2H-MoS₂ layer. Each of these vertical structures possesses a size up to 1 µm and they are formed at the coalescence point of two misoriented MoS₂ flakes. Additionally, the thickness of the MoS₂ nanosheet is not constant along the vertical nanosheet height. Instead, it has a tapered morphology with a larger basal size with decreasing size towards the end. Detailed morphological attributes of these features are given in Figure S2. It is worth noting that these vertical features have different dimensions and grow randomly upwards with varying thicknesses. Furthermore, the basal plane of the vertical structures has a tetrahedral shape. This tetrahedral layer is a three-dimensional bulk layer of MoS₂, as will be discussed later.

The material properties of the vertical MoS₂ nanosheet layers were further investigated using X-ray diffraction analysis and Raman spectroscopy measurements. Raman measurements were performed at room temperature under ambient conditions and unpolarized detection. Figure 2 is the corresponding Raman spectrum obtained from the material surface.



Figure 2. Raman spectra obtained from different points on the 3D MoS₂ nanosheet. Inset shows a photograph of the 1×1 cm growth substrate with Raman measurement points and the N₂ flow direction (arrow mark points to the outlet).

The Raman measurements were carried out on the sample surface at three different places. The insets in Figure 2 show the positions where the Raman spectra were obtained. Each spectrum color corresponds to the color assigned to the respective circle shown in the measurement area of the inset in Figure 2. The two dominant Raman modes of MoS₂ are revealed by the Raman spectra, which are the in-plane (E^{1}_{2g}) and the out-of-plane (A_{1g}) vibrations of the sulfur atoms at 382.5 and 409.0 cm⁻¹, respectively [25]. The study reveals further corresponding Raman peaks of MoS₂ attributable to the following two phonon modes such as 2LA (M) at 454 cm⁻¹, [25–28]; E_{2g} + LA at 600 cm⁻¹; $2E_{2g}$ at 764 cm⁻¹; and $2A_{1g}$ at 820 cm⁻¹. As shown in the insets in Figure 2, the top edge is near to the precursor and the bottom edge is facing the outlet of the furnace. The flow direction of N₂ is shown

with an arrow mark pointing towards the outlet. The Raman measurements obtained from three points show a variation in the intensities of the E^{1}_{2g} and A_{1g} modes. This variation can reveal different attributes such as crystallinity, thickness, and defect density of the deposited material [14,29,30]. The intensities of the E^{1}_{2g} and A_{1g} peaks are decreasing from the top edge to the bottom edge of the sample. The sample surface near to the precursor has a higher thickness of MoS₂ deposition as compared to the other end and a higher defect density of exposed edges of the vertical structures [29,30]. Moreover, most of the peak shapes remain unchanged without any significant peak shift, providing the stable crystal morphology of the MoS₂ sand flower-like structures at different positions of the sample. Furthermore, we could not observe any additional peaks at different measurement positions of the sample surface. The wavenumber difference Δ between the E_{2g} and A_{1g} Raman modes is 25 cm⁻¹, confirming the bulk nature of the material [28,30,31]. A weak Raman signal of MoO₃ is visible, which can be attributed to partial oxidation at growth or after exposure to air [32].

The X-ray diffraction spectra measured in Bragg–Brentano configuration from the synthesized MoS₂ at different deposition times is shown in Figure 3. Here, for each deposition time, the diffraction peak at 14.4° can be attributed to the 2H-MoS₂ structure, which corresponds to (002) crystal planes in 2H-MoS₂ [33] that are orientated along its c-axis [33,34]. The corresponding higher-order diffraction peaks (004), (006), and (008) are given at their corresponding characteristic designations. While for the deposition time of 10 min we could observe a broader peak with lower intensity at diffraction peak 14.4°, other higher order diffraction peaks are not visible. This might be due to discontinuous and scattered deposition of MoS₂. Moreover, the prominent Si (400), compared to other deposition times, further supports the non-uniformity of the MoS₂ deposition. However, it is evident that, with an increase in the deposition time, the intensity of the diffraction peaks became prominent and further give significance to the crystallinity of the synthesized MoS₂. Unfortunately, diffraction peaks of non-basal planes are not observed. This might be caused by the orientations of these sheets not fulfilling the diffraction.



Figure 3. XRD analysis of synthesized MoS₂ at different deposition times.

3.2. Evolution and Growth Mechanism of the Three-Dimensional MoS₂ Nanosheet

Several samples with varying growth durations were examined to understand the vertical growth mechanism of the MoS_2 nanosheets. The growth morphological evolution of the three-dimensional vertical MoS_2 nanosheet are summed up in the SEM images in Figure 4. Initially in the growth process, the precursor MoO_3 is partially reduced,

resulting in the formation of the volatile MoO_{3-x} species [35,36]. The carrier gas transports these under-stoichiometric MoO_{3-x} species to the substrate, where they absorb and diffuse, forming the initial MoS_2 nucleation sites. These nucleation sites react with incoming sulfur molecules to form the randomly oriented initial MoS_2 grains (flakes) with even-sided triangular shape (Figure 4a). More MoO_{3-x} gets absorbed as the growth time increases, and the nucleation domains enlarge due to the steady supply of reactants [35,37]. Due to the enlarging terrace dimension, secondary nucleation of the next MoS_2 layer occurs. This is evidenced by the different grey scale in Figure 4a. In Figure 4b, taken from a sample with a deposition time of 10 min, a random distribution of few-layered MoS_2 triangular structures is seen. In this pattern, two types of light grey areas can be noticed: (1) triangularand (2) non-triangular-like. The triangular-like areas stem from substrate surface uncovered with MoS_2 . To reveal the nature of the non-triangular feature, higher magnification images were taken. A typical image is shown in Figure 4c.



Figure 4. Growth morphology evolution of synthesized vertical MoS₂ nanosheets. (**a**) SEM images of the initial formation of nucleation domains of MoS₂ after 5 min of growth time. (**b**) Triangular-shaped MoS₂ formation after 10 min. (**c**) A magnified SEM image of triangular MoS₂ structures with micro spikes. (**d**) Formation of MoS₂ multilayered islands which cover the substrate surface after 20 min. (**e**) Formation of a three-dimensional tetrahedral MoS₂ thin film along with numerous vertical flower-like formations. (**f**,**g**) Tilted and magnified SEM images of material surface after 25 min. (**h**) Formation of dense flower-like vertical formations of MoS₂ structures after 30 min.

A closer observation revealed submicron vertical spikes platelets that appeared to have formed at the grain boundaries of the thin layers of MoS_2 (see Figure 4c). These the edges cause a higher secondary electron emission. After 15 min of growth time, most triangular structures grow laterally into larger domain sizes and join into a uniform film, as presented in Figure 4d. In addition, it was observed that MoS_2 was distributed across the whole substrate surface with several scattered multi-stacked triangular islands (see Figure 4d).

Additional micro spike-like structures also grew along the substrate surface. Then, with the subsequent increase in growth time and constant supply of reactants, more multilayered triangular features developed and started to merge. Eventually, after 25 min of growth, most of the multilayered triangular islands joined and formed a bulk tetrahedral MoS₂ thin film, as shown in Figure 4e. Here, the MoS₂ thin film exhibits a tetrahedral geometrical shape. The grain boundaries of these structures have a triangular steps decrease from the bottom to the apex, forming a tetrahedral shape. A magnified SEM image of the tetrahedral structure is shown in Figure 4g. Figure 4f shows the tilted image of those layers with several sand flower-like MoS₂ formations. The synthesis of tetrahedral MoS₂ thin films was reported previously [9,38–40]. This type of structural formation could be referred to as the Stranski–Krastanov-like growth mechanism [41–43], which might also stem from Ehrlich–Schwoebel barriers [44].

In Figure 4e, f, a significant number of vertical sand flower-like structures emerged above the bulk layer. Notably, the developed bulk layer formed the base for the vertical MoS₂ structures. Many vertical structures were finally formed densely covering the substrate surface after 30 min of growth, as illustrated in Figure 4h. To further clarify the layer transformation of MoS₂, Raman measurements were performed on the samples with different growth times. The wavenumber difference between E^{1}_{2g} and A_{1g} peaks for various growth times was analyzed and plotted as shown in the supplementary Figure S3. The frequency difference of the prominent peaks of Raman spectra of MoS₂ increases with increasing growth time, indicating a transformation of MoS₂ structures from a two-dimensional to a three-dimensional structure.

For further elucidation of the formation of vertical MoS₂ nanosheets, we have analyzed more plane view and cross-section SEM images at different growth intervals. The so-called extrusion growth model [14] is currently the most acceptable growth mechanism among the numerous growth models for vertically standing MoS_2 nanosheets presented in the literature [14,45]. According to this growth model, several multi-layered MoS₂ islands are initially developed individually on the substrate surface. Then, once these layers have reached a specific thickness, provided with more supply of reactants and an increase in growth time, these individual islands start to merge to form a thick layer of MoS₂. During this process, edge-oriented, vertically standing MoS_2 seedlings of the sand flower-like structures are originated. A similar formation of vertical MoS₂ seedlings is presented in Figure 5a. As speculated above, the seedlings stemmed from the aggregation zone of different MoS_2 islands. A three-dimensional representation of the SEM image shown in Figure 5b confirms the interaction in the aggregation zone between different multilayered MoS₂ islands. A comparable formation has already been reported [14,45]. Therefore, FIB cross sections were prepared to gather more information on these MoS_2 vertical structures, shown in Figure 5c–e, from different 3D vertical MoS₂ structures. The FIB images indicate that the vertical structures are induced by the distorted growth of planar bulk MoS_2 layers.

Two types of vertical morphology could generally be observed based on the crosssectional observations. Figure 5c,e show that most vertical structures originate from the base bulk layer of MoS_2 , tend to bend or slip with the adjacent MoS_2 islands, and grow upwards. Another alternative would be that the vertical structure was formed by converging two separate MoS_2 films, creating a void space between them, as shown in Figure 5d. These two inferences provide direct evidence of forming a vertical MoS_2 nanosheet consistent with the known growth model. Thus, the formation of vertical



standing MoS₂ seedlings could be attributed to the compressive force created by developing deformations and strain energy [14,45,46].

Figure 5. Typical growth morphological evolution of vertical MoS_2 structures. (a) Vertical seedlings originate from the aggregation zone of two MoS_2 islands due to the compressive forces between them (b) A three-dimensional representation of aggregation zone between different multi-layered MoS_2 islands, the light blueish green color represents the multilayered structures and the red color represent the vertical sheet. (**c**–**e**) Cross-sectional FIB image of vertical MoS_2 structures with different morphological features. (**f**,**g**) The formation of vertical seedlings from the micro fissure appears on individual MoS_2 islands.

However, based on our experimental results, we could also observe a seedling of vertical MoS_2 structures that originates from the micro fissures (cracks) appearing on individual MoS_2 islands, as shown in Figure 5f,g. Furthermore, it could be observed that the vertically standing MoS_2 seedlings appear from the micro fissures on separate MoS_2 islands with different geometrical features. Therefore, we believe that these self-formative micro cracks arise during the growth and vertical nanosheets extrude from these cracks due to the strain energy inside the cracks.

3.3. Electrical Performance of FET Devices on the 3D Vertical MoS₂ Nanosheet

We performed electrical tests on fabricated field-effect transistor (FET) devices to analyze the electrical properties of the synthesized material. The measured characteristics of devices fabricated on vertical MoS_2 nanosheets are illustrated in Figure 6. The insets in Figure 6a are the SEM picture of the fabricated lateral device. The electrical measurements were carried out under a nitrogen environment at ambient temperature using a Keithley SCS 4200 system.



Figure 6. Electrical performance of a FET device on a vertical MoS_2 nanosheet. (**a**, **b**) Typical output and transfer characteristics of a lateral device; inset shows the SEM image of a fabricated device.

A back-gate voltages (V_{GS}) was applied to the Si-substrate and the SiO₂ layer of 90 nm thickness to modulate the Fermi level. Figure 6a shows a typical output characteristic of the fabricated device with channel dimensions of L= 9 μ m and W= 50 μ m. The back-gate voltage (V_{GS}) was varied in 10 V increments from 10 V to 40 V. The back-gate voltage (V_{GS}) was varied by 10 V from 10 V to 40 V. The FET shows an n-type behavior. The respective transfer characteristics of the device are given in Figure 6b. The device exhibits an on/off ratio greater than 10⁸ with minimum current (I_{off}) of 4.34 × 10⁻¹⁵ and maximum current (I_{on}) of 2.22 × 10⁻⁶ A at V_{DS} = 0.5 V and V_{GS} = 40 V. An exponential transition from the subthreshold to the conduction region was observed, corresponding to a subthreshold swing (SS) of 1.66 V/dec.

Furthermore, the transconductance extrapolation (GMLE) method was employed to calculate the threshold voltage from transfer characteristics [47,48]. According to this method, Equation (1) is:

$$\mu = \frac{L}{W} \frac{I_D}{C_{ox} V_{DS} (V_{GS} - V_T)} , \qquad (1)$$

where μ —low-field mobility of the carriers, L—the channel length, W—the channel width, I_D—the drain current, C_{OX}—the gate oxide capacitance per unit area, V_{DS}—the drain-to-source voltage, V_{GS}—the back-gate-to-source voltage, and V_T—the threshold voltage. Further, with the extracted threshold voltage V_T = 21 V, we have determined a low-field charge carrier mobility of 1.46 cm² V⁻¹ s⁻¹, and a channel carrier concentration of 4.5 × 10¹² cm⁻². The value obtained in this work is comparable to the values reported in related work [18]. The relatively low carrier mobility can be attributed to the high density of scatters, mainly grain boundaries, interface traps, and charged impurities [49–51]. In addition, vertical MoS₂ structures exhibit a lower mobility due to their high aspect ratio and edge sites, contributing also to the scattering [50,51]. However, the higher density of defects is advantageous for assessing the hysteresis of these devices on vertical MoS₂, which could further broaden its use in memristive applications. Figure 7a illustrates the log scale double sweep output curve of the MoS₂ FET. Here, the devices were measured in such a way that the drain–source voltage (V_{DS}) was varied from 0 V to 3.3 V, then from 3.3 V to -3.3 V, and back to 0 V. After each sweep, the gate voltage (V_{CS}) was increased by 10 V from 0 V to 40 V.



Figure 7. (a) Double-sweep output characteristics (logarithmic scale) of MoS_2 FET. (b) Dependence of hysteresis area with varying back-gate modulation. (c) Double-sweep output characteristics of MoS_2 FET with varying channel length-to-width ratio. (d) Hysteresis area dependence with length-to-width ratio of MoS_2 FET.

The difference in the areas underneath the upper and lower branches of the hysteresis curves was used to quantify the hysteresis using the following equation:

$$H = \int_{V_{min}}^{V_{max}} I_d(V_d) dV_{dforward} - \int_{V_{min}}^{V_{max}} I_d(V_d) dV_{dbackward}$$
(2)

The hysteresis area obtained by varying the back-gate voltage is shown in Figure 7b. Notably, the hysteresis area increased with gate voltage variation from 10 V to 40 V. This implies that higher-gate electric field strength sweeps enhance the trapping and de-trapping of charges, contributing to a wider hysteresis [52,53]. Additionally, it is noted that the hysteresis area could also be modulated with an altered channel dimension of the device. Figure 7c shows the double-sweep output curves of the device with varying channel length-to-width ratio at the drain–source voltage ± 3.3 V at a constant gate–source voltage of 40 V. As seen in Figure 7c, increasing the channel width causes the surface-to-volume ratio and the number of charge trap states to increase for a given channel length, resulting in a broader hysteresis width. A calculated hysteresis area dependence on the channel length-to-width ratio is shown in Figure 7d.

Therefore, a change in the channel dimensions broadens the hysteresis area, suggesting the contribution of the MoS_2 surface and MoS_2/SiO_2 interface defects and their associated traps on hysteresis [52–54].

To further understand the relationship between structural properties and electrical properties of synthesized MoS_2 at different growth times, we have fabricated back-gate field effect devices and calculated the electrical parameters as shown in Table 1. The mobilities and sheet carrier concentrations of the devices decrease when the synthesized

 MoS_2 makes a transition from 2D phase to 3D phase. It further explains that higher mobilities are observed in few layered MoS_2 and decreases with an increase in the layer thickness, which is typical behavior of MoS_2 FET devices reported previously [49,50]. The reduced sheet carrier density leads to an increase of the ON/OFF ratio. With increasing thickness, a decrease in mobility is observed. This effect can contribute to an increase in defect density, grain boundaries, increasing surface-to-volume ratio, and significant phonon scattering at ambient conditions [5,49,50,55]. The change in the subthreshold swing (SS) is a consequence of the increased thickness of the MoS_2 layer at a constant thickness of the back-gate silicon dioxide.

Deposition Time	On/Off Ratio	SS (V/dec)	Mobility (cm ² V ⁻¹ s ⁻¹)	Carrier Density (cm ⁻²)
10 min	$1.8 imes10^4$	3.05	3.26	$1.01 imes 10^{13}$
25 min	$1.9 imes10^8$	1.9	0.25	$5.4 imes10^{12}$
30 min	$5.1 imes 10^8$	1.6	1.46	$4.5 imes10^{12}$

Table 1. Electrical properties of MoS₂ deposited at different growth times.

4. Conclusions

In conclusion, a reliable, single-step CVD process for synthesizing vertical MoS₂ nanosheets was developed and investigated. The morphological characteristics of the material surface were characterized by employing optical microscopy and SEM. Additionally, a Raman spectroscopy investigation uncovered the MoS_2 characteristic signals. The significant disparity in wavenumber between the active modes supported the bulk nature of the synthesized material. Moreover, XRD measurements verified the material crystallinity and its nature as 2H-MoS₂. The cross-sectional FIB images revealed the primary growth mechanism. It is assumed that the main driving force of vertical MoS_2 nanosheet growth is the high strain energy brought on by the compression between MoS2 islands. In addition to the known concept, we could also notice the sprouting of vertical MoS_2 structures that develop from the micro fissures (cracks) on individual MoS₂ islands. This might be due to the strain energy within the crack. Additionally, the lateral back-gated FET devices fabricated on the synthesized material showed an n-type behavior with field-effect mobility of 1.46 cm² V⁻¹ s⁻¹ and an estimated carrier concentration of 4.5×10^{12} cm⁻². Moreover, we have compared and evaluated the extracted low-field charge carrier mobilities and the electrical properties of the devices fabricated on MoS₂ with different growth times. Furthermore, the effects of back-gate voltage bias and channel dimensions on the hysteresis effect were investigated and quantified. Finally, it is noted that the high aspect ratio, defects on the vertical MoS_2 , and MoS_2/SiO_2 interface defects and their associated traps have a substantial influence on the hysteresis.

Supplementary Materials: The following supporting information can be downloaded at: https://www.mdpi.com/article/10.3390/cryst13030448/s1, Figure S1: Schematic illustration of the CVD setup. Figure S2: Cross-sectional FIB image of vertical MoS₂. Figure S3: Raman spectrum obtained with different growth times.

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