

Supplementary material of the Article in *Crystals*

A tetranuclear Ni(II)-cubane cluster molecule build by four μ_3 -O-methanolate (MeO) ligands, externally cohesive by four unprecedented bridging μ_2 -N7,O6-acyclovirate (acv-H) anions.

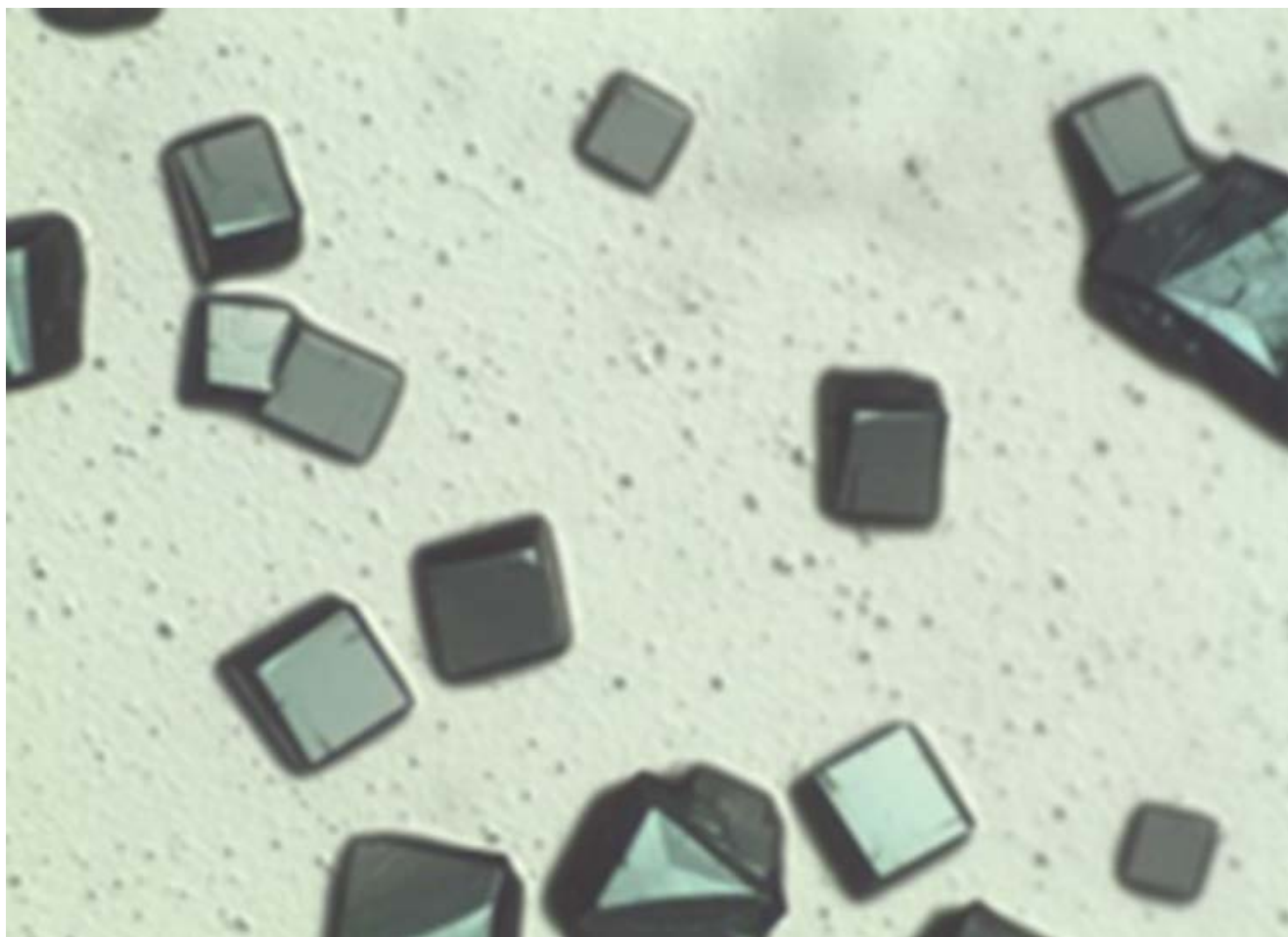
by

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Figure S1. Crystals of compound $[\text{Ni}(\mu_2\text{-N7,O6-acv-H})(\mu_3\text{-MeO})(\text{H}_2\text{O})]_4 \cdot 8\text{H}_2\text{O}$ (1).



S2. Supplementary structural information of compound 1.

Table A. Selected bond lengths [Å] and angles [deg] for [Ni(acv)(MeO)(H₂O)]4.8H₂O.

Ni(1)-O(1)	2.038(2)
Ni(1)-O(1)#1	2.041(3)
Ni(1)-O(1)#2	2.049(2)
Ni(1)-O(16)	2.054(2)
Ni(1)-N(17)	2.069(3)
Ni(1)-O(2)	2.131(3)
Ni(1)-Ni(1)#3	3.0527(10)
Ni(1)-Ni(1)#1	3.0527(10)
Ni(1)-Ni(1)#2	3.0657(11)
O(1)-O(1)#2	2.694(5)
O(1)-O(1)#3	2.701(4)
O(1)-O(1)#1	2.701(4)
O(1)-Ni(1)-O(1)#1	82.94(10)
O(1)-Ni(1)-O(1)#2	82.44(10)
O(1)#1-Ni(1)-O(1)#2	82.66(11)
O(1)-Ni(1)-O(16)	175.74(10)
O(1)#1-Ni(1)-O(16)	96.29(10)
O(1)#2-Ni(1)-O(16)	93.30(10)
O(1)-Ni(1)-N(17)	94.82(10)
O(1)#1-Ni(1)-N(17)	95.31(11)
O(1)#2-Ni(1)-N(17)	176.76(12)
O(16)-Ni(1)-N(17)	89.43(11)
O(1)-Ni(1)-O(2)	91.31(10)
O(1)#1-Ni(1)-O(2)	171.96(11)
O(1)#2-Ni(1)-O(2)	91.04(10)
O(16)-Ni(1)-O(2)	89.03(10)
N(17)-Ni(1)-O(2)	90.75(12)
Ni(1)-O(1)-Ni(1)#3	96.90(11)
Ni(1)-O(1)-Ni(1)#2	97.18(10)
Ni(1)#3-O(1)-Ni(1)#2	96.54(10)

Symmetry transformations used to generate equivalent atoms:

#1 -z+1/4, -y+3/4, x #2 -x+1/4, y, -z+1/4

#3 z, -y+3/4, -x+1/4

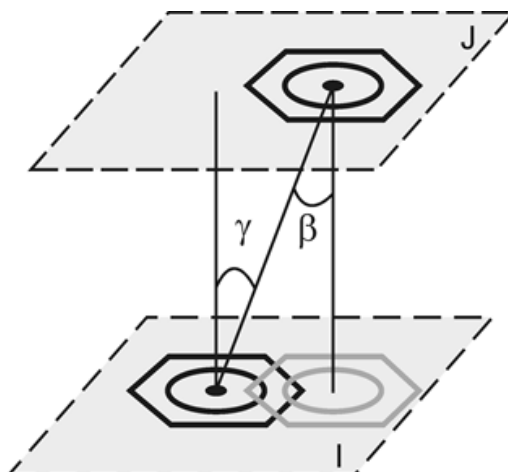
Table B. Hydrogen bonds for [Ni(acv)(MeO)(H₂O)]4.8H₂O [Å and deg.].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O(2)-H(2A)...O(4)#4	0.98	1.77	2.714(6)	159.8
N(12)-H(12A)...O(4)#5	0.84	2.15	2.935(6)	156.4

Symmetry transformations used to generate equivalent atoms:

#1 -z+1/4, -y+3/4, x #2 -x+1/4, y, -z+1/4 #3 z, -y+3/4, -x+1/4

#4 y, z, x #5 -x, -y+1/2, -z+1/2

Table C. Selected data for π -sacking interactions between the six-membered ring of guanine moieties of acv-H.

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Analysis of Short Ring-Interactions with Cg-Cg Distances < 6.0 Angstrom and Beta < 60.0Deg.

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- Cg(I) = Plane number I (= ring number in () above)
 - Alpha = Dihedral Angle between Planes I and J (Deg)
 - Beta = Angle Cg(I)-->Cg(J) or Cg(I)-->Me vector and normal to plane I (Deg)
 - Gamma = Angle Cg(I)-->Cg(J) vector and normal to plane J (Deg)
 - Cg-Cg = Distance between ring Centroids (Ang.)
 - CgI_Perp = Perpendicular distance of Cg(I) on ring J (Ang.)
 - CgJ_Perp = Perpendicular distance of Cg(J) on ring I (Ang.)
 - Slippage = Distance between Cg(I) and Perpendicular Projection of Cg(J) on Ring I (Ang).
 - P,Q,R,S = J-Plane Parameters for Carth. Coord. (Xo, Yo, Zo)
- =====

6-Membered Ring (21) N(11) --> C(12) --> N(13) --> C(14) --> C(15) --> C(16)a -->

6-Membered Ring (22) C(16) --> N(11)c --> C(12)c --> N(13)c --> C(14)c --> C(15)c -->

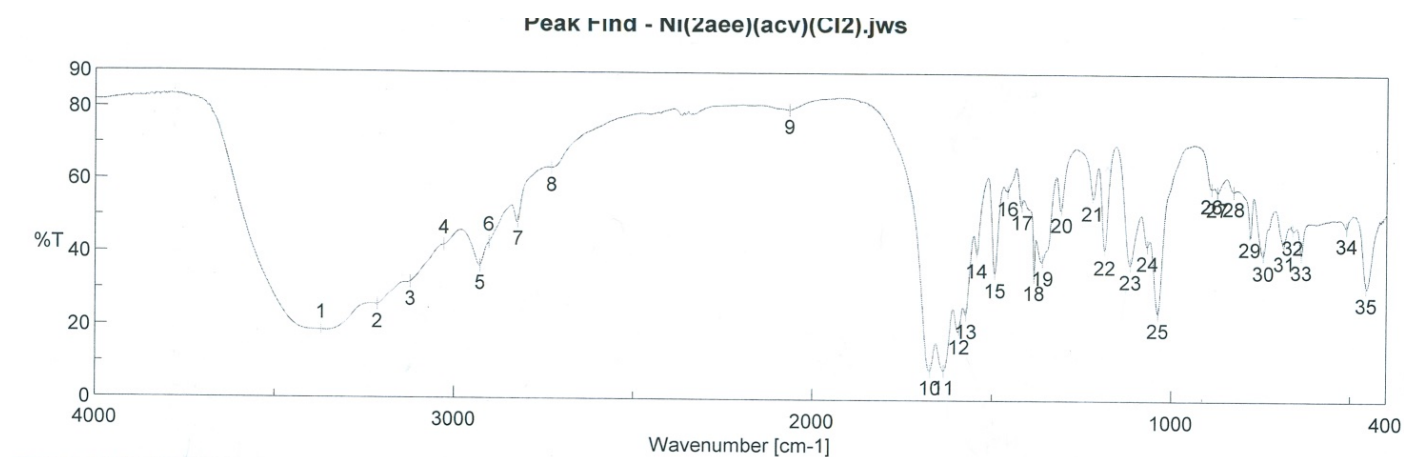
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Cg(I)	Res(I)	Cg(J)	[ARU(J)]	Cg-Cg Transformed J-Plane P, Q, R, S				Alpha	Beta	Gamma	CgI_Perp	CgJ_Perp	Slippage
Cg(21)	[1]	-> Cg(21)	[10554.10]	3.7561	-0.2580	0.6911	-0.6752	10.2640	1	25.32	25.32	-3.3954	-3.3954
Cg(22)	[1]	-> Cg(22)	[6455.10]	3.7561	-0.6752	0.6911	0.2580	6.8318	1	25.32	25.32	3.3954	3.3954
				Min or Max	3.745				0.67	1.98	89.44	-3.421	-5.352

[1555] = X,Y,Z
 [103555] = 1/4-X,Y,1/4-Z
 [6455] = -1/4+Y,1/4+X,-Z

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S3. FT-IR spectrum (KBr disk) of compound 1.



[Result of Peak Picking]

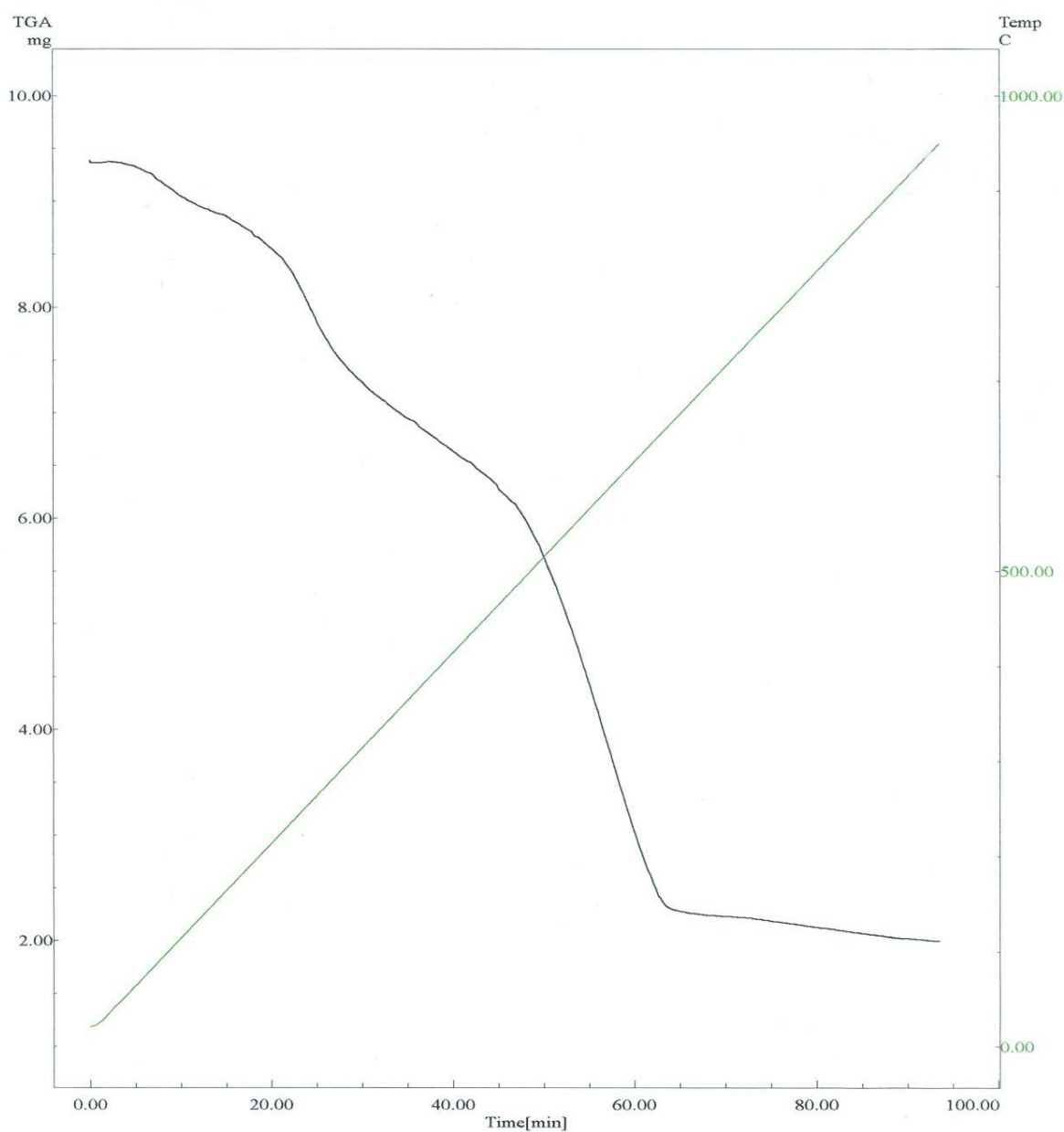
No.	Position	Intensity	No.	Position	Intensity	No.	Position	Intensity
1	3369.51	18.54	2	3212.34	25.6522	3	3122.19	31.8255
4	3028.66	41.9277	5	2927.89	36.3618	6	2903.79	42.8008
7	2823.76	48.4978	8	2730.71	63.3809	9	2065.87	79.8039
10	1671.98	8.20386	11	1634.38	8.29311	12	1591.95	19.491
13	1573.63	23.7284	14	1544.22	40.2333	15	1493.6	34.8934
16	1458.89	57.4153	17	1418.87	53.5897	18	1384.64	34.2486
19	1360.53	38.286	20	1307.5	52.9018	21	1222.17	56.0861
22	1187.45	41.2979	23	1115.62	37.3534	24	1067.89	42.4665
25	1038.48	24.0948	26	889.505	58.3826	27	873.596	57.4321
28	828.277	57.6666	29	782.958	46.4963	30	745.835	40.0812
31	687.498	42.7729	32	663.393	47.1947	33	639.287	40.338
34	511.526	47.619	35	456.082	31.374			

S4. TGA study of compound 1 with FT-IR spectra to identification of evolved gases.**S4.1. TGA curve versus time (abscissas) and temperature (right ordinates).**

File Name: 18C-0055.D20
Detector Type: Shimadzu TGA-50H
Acquisition Date: 11/11/30
Acquisition Time: 08:45:37
Sample Name: C-884
Weight: 9.382[mg]
Cell: Alumina
Atmosphere: Air
Rate Flow: 100.0[ml/min]
Operator: MASP

C-884

Temp Program		
Rate	Hold Temp	Hold Time
[C/min]	[C]	[min]
10.0	950.0	0.0



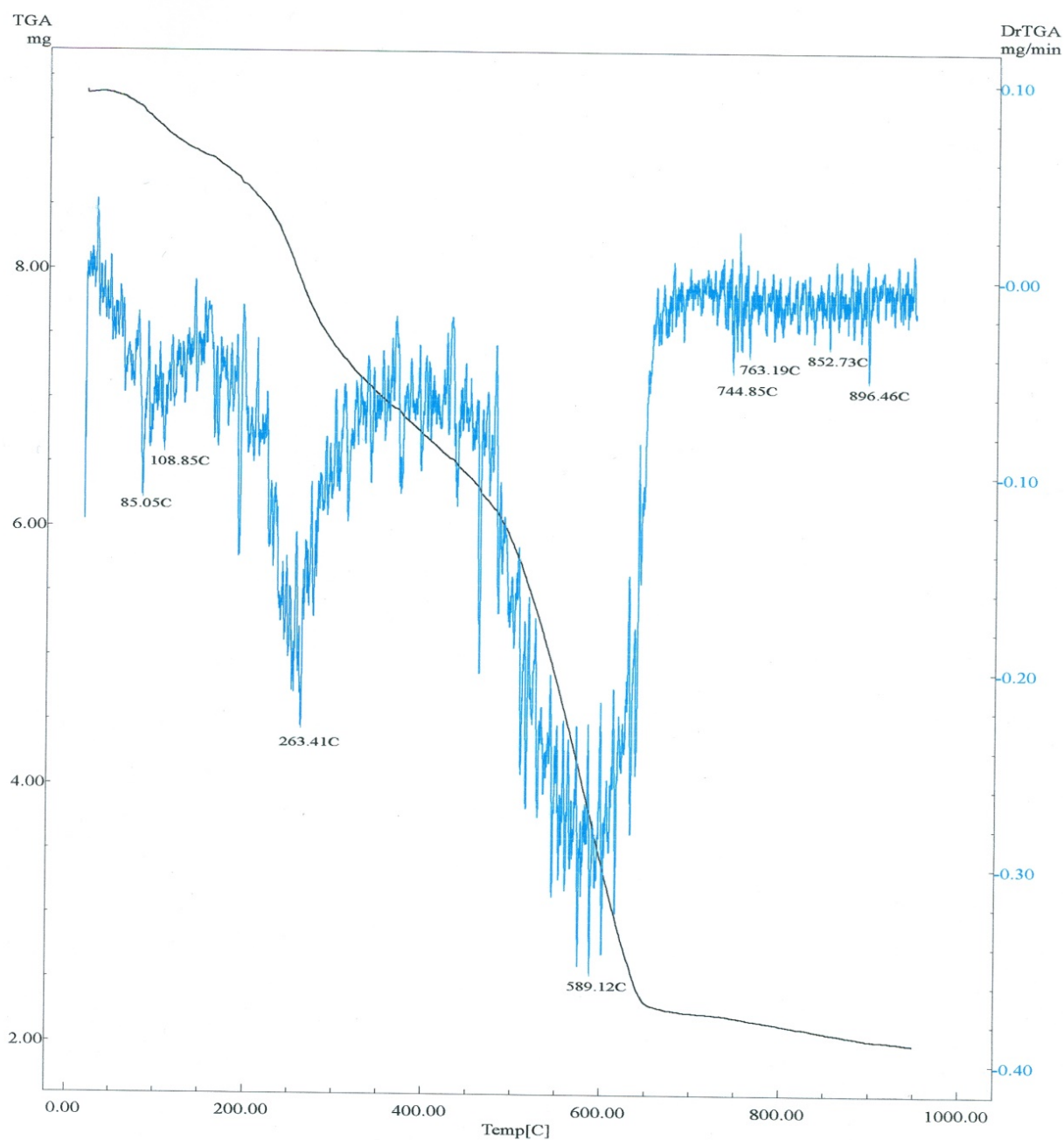
S4.2. First derivative of TGA curve, to decide the beginning and end of steps.

Unidades de TA y UVNIR
Centro de Instrumentación Científica
Universidad de Granada

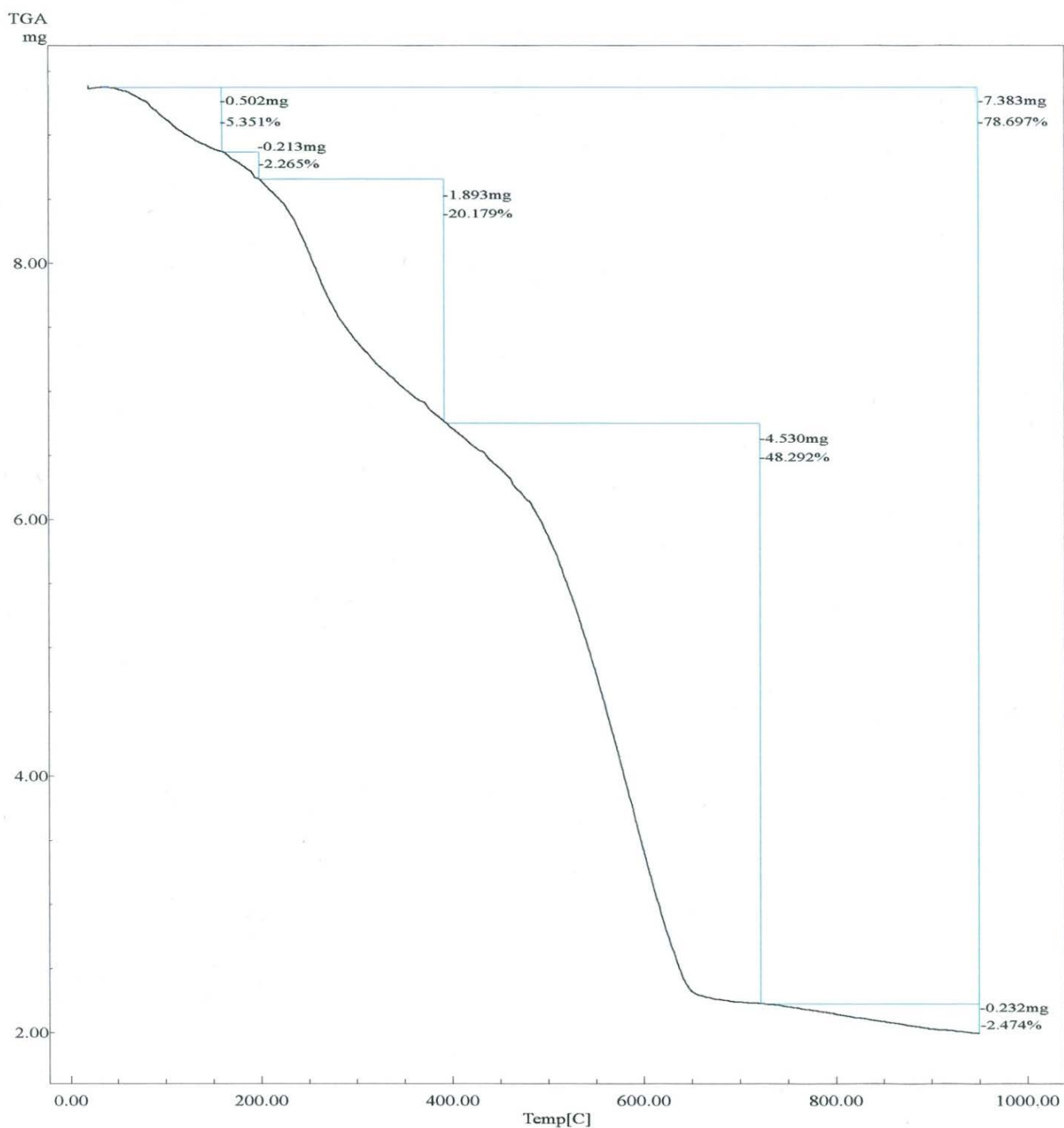
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Operator: MASP

C-884

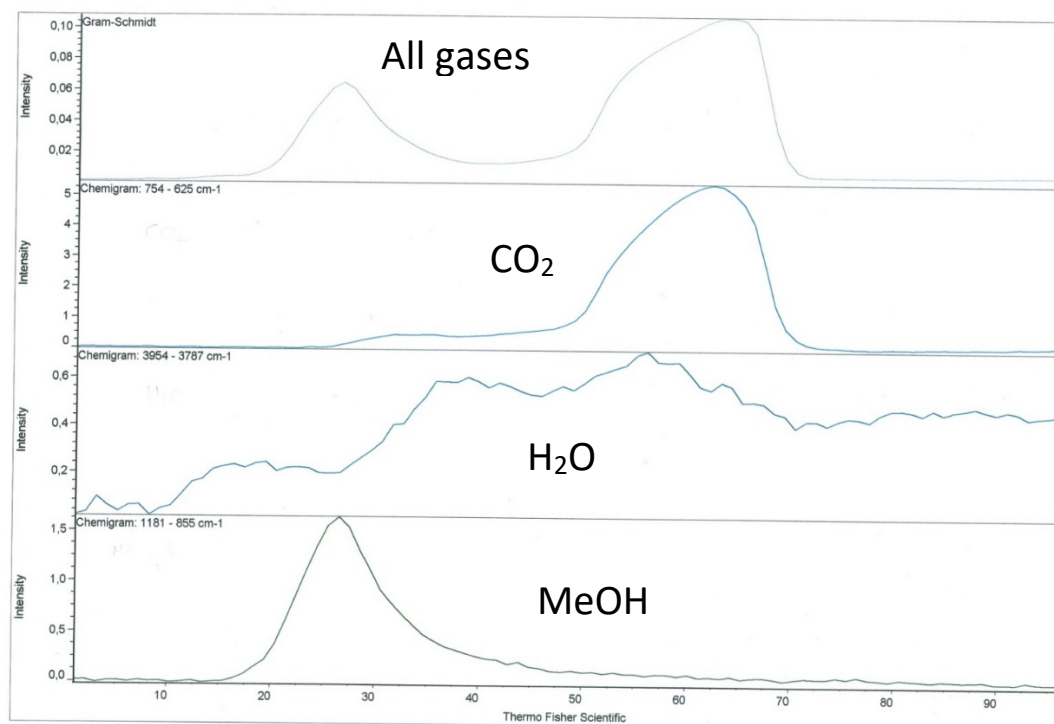
Temp Program
Rate Hold Temp Hold Time
[C/min] [C] [min]
10.0 950.0 0.0



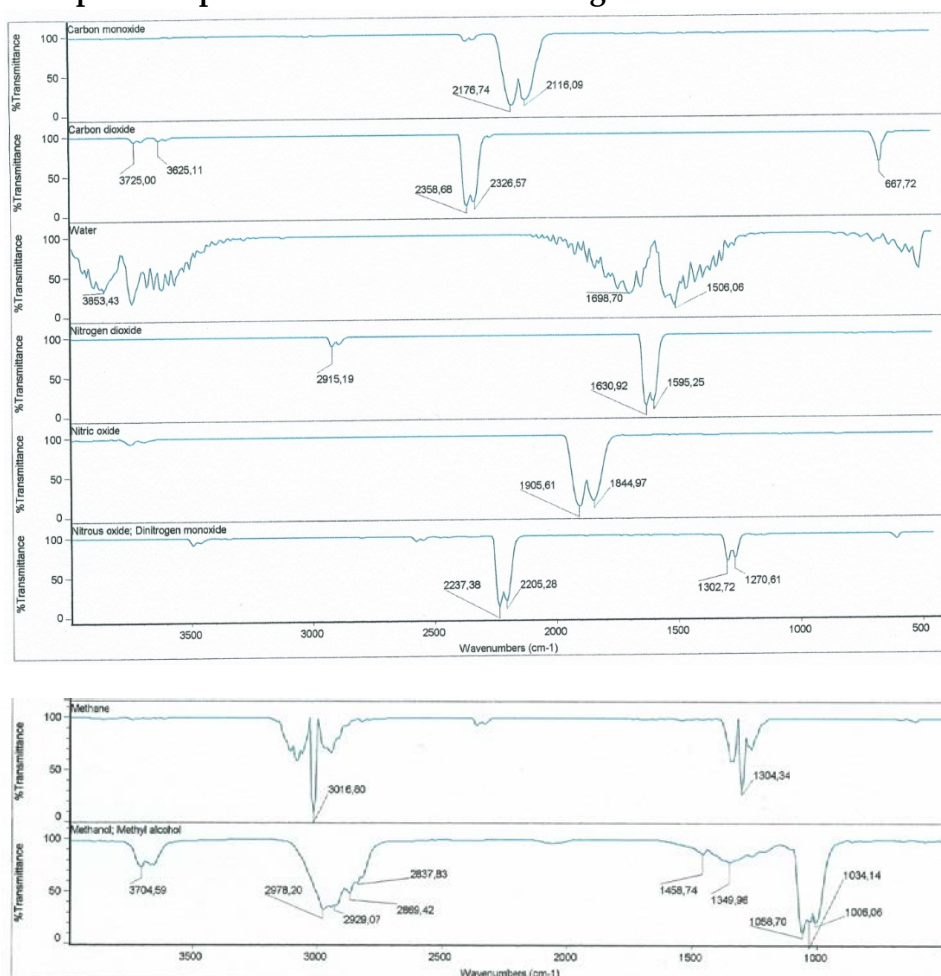
S4.3. TGA curve of weight versus temperature, with data of partial or total experimental weight loss (in mg or %).



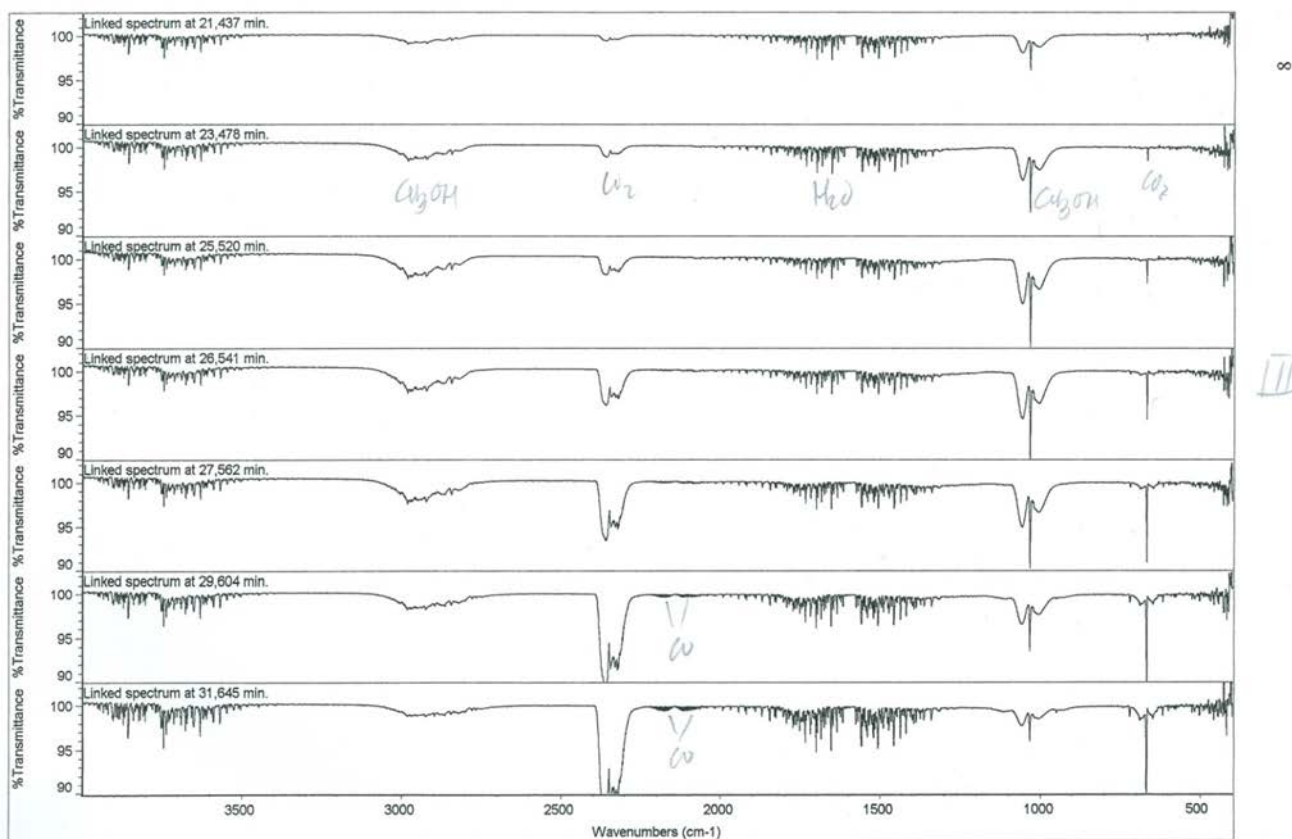
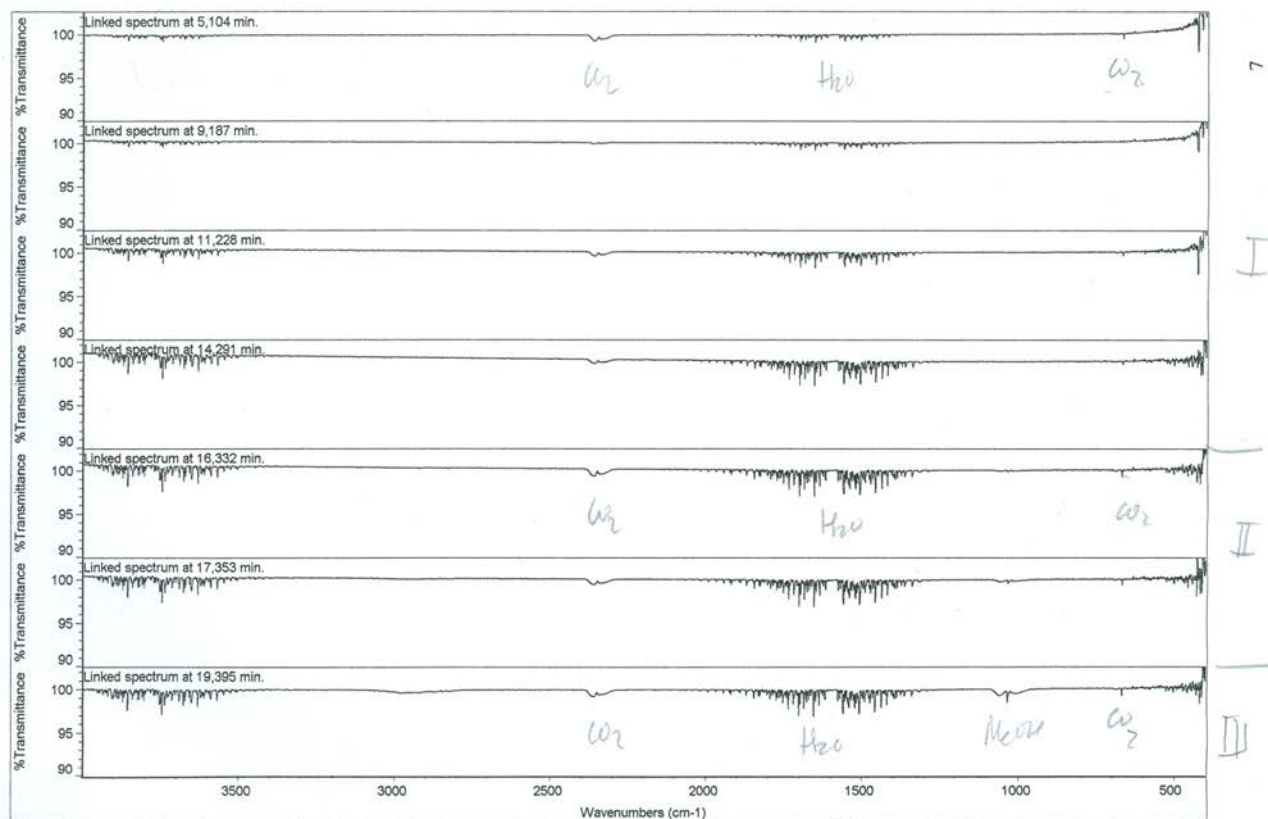
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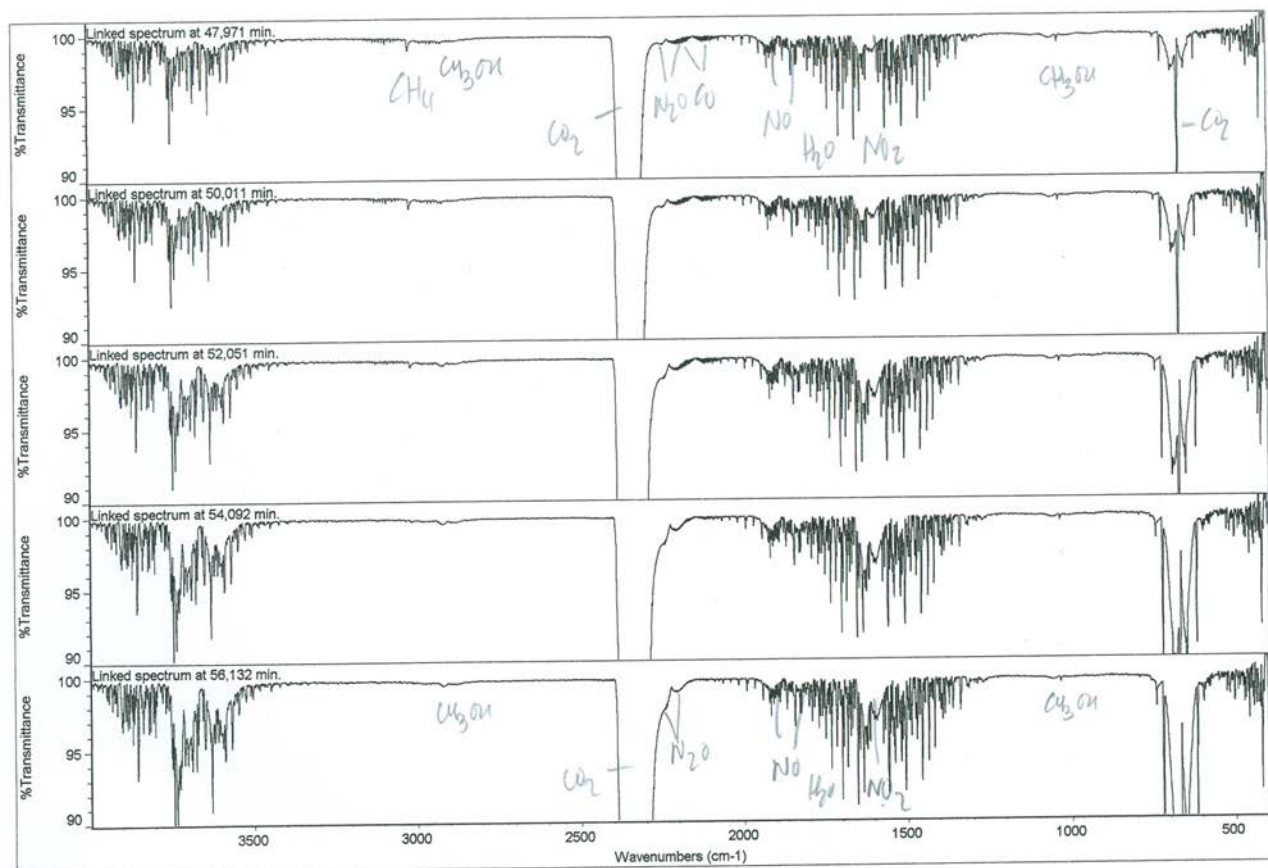
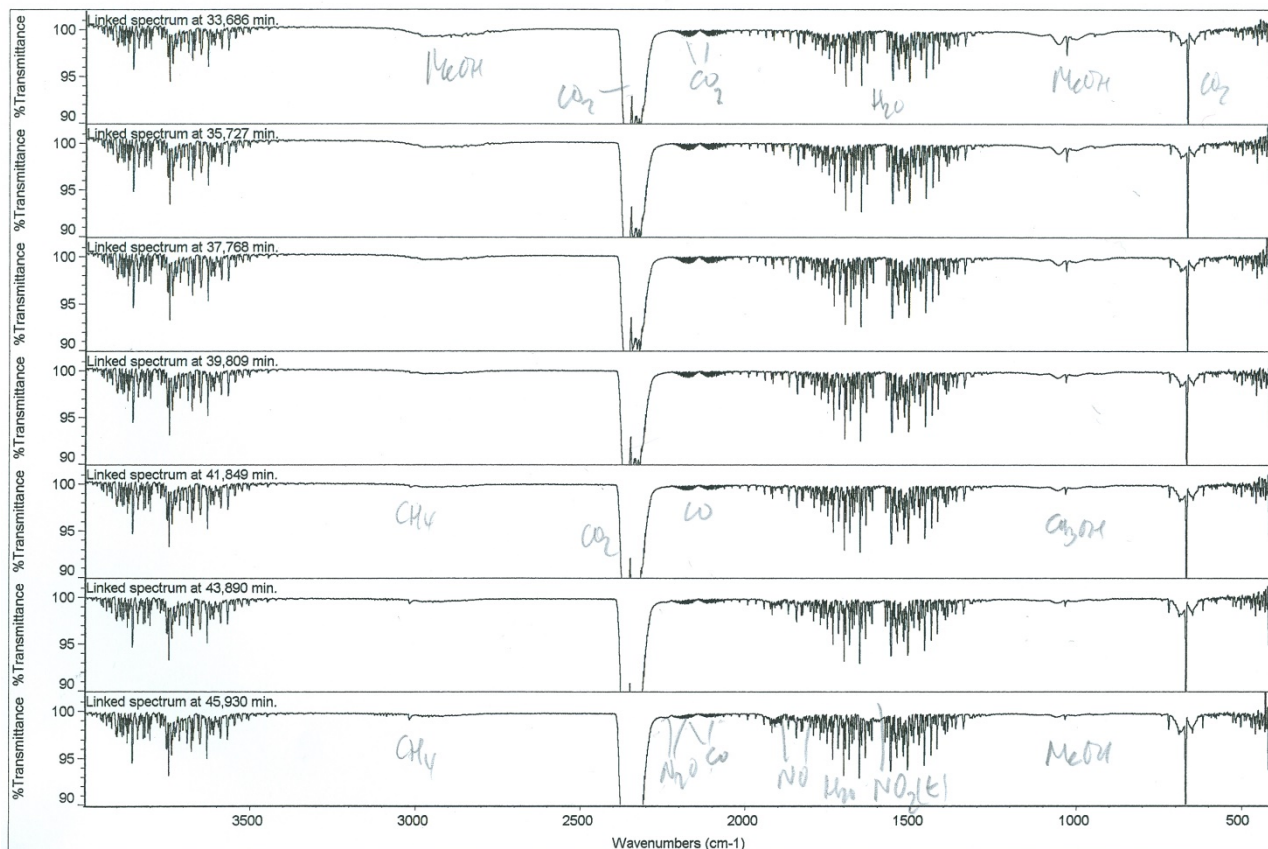


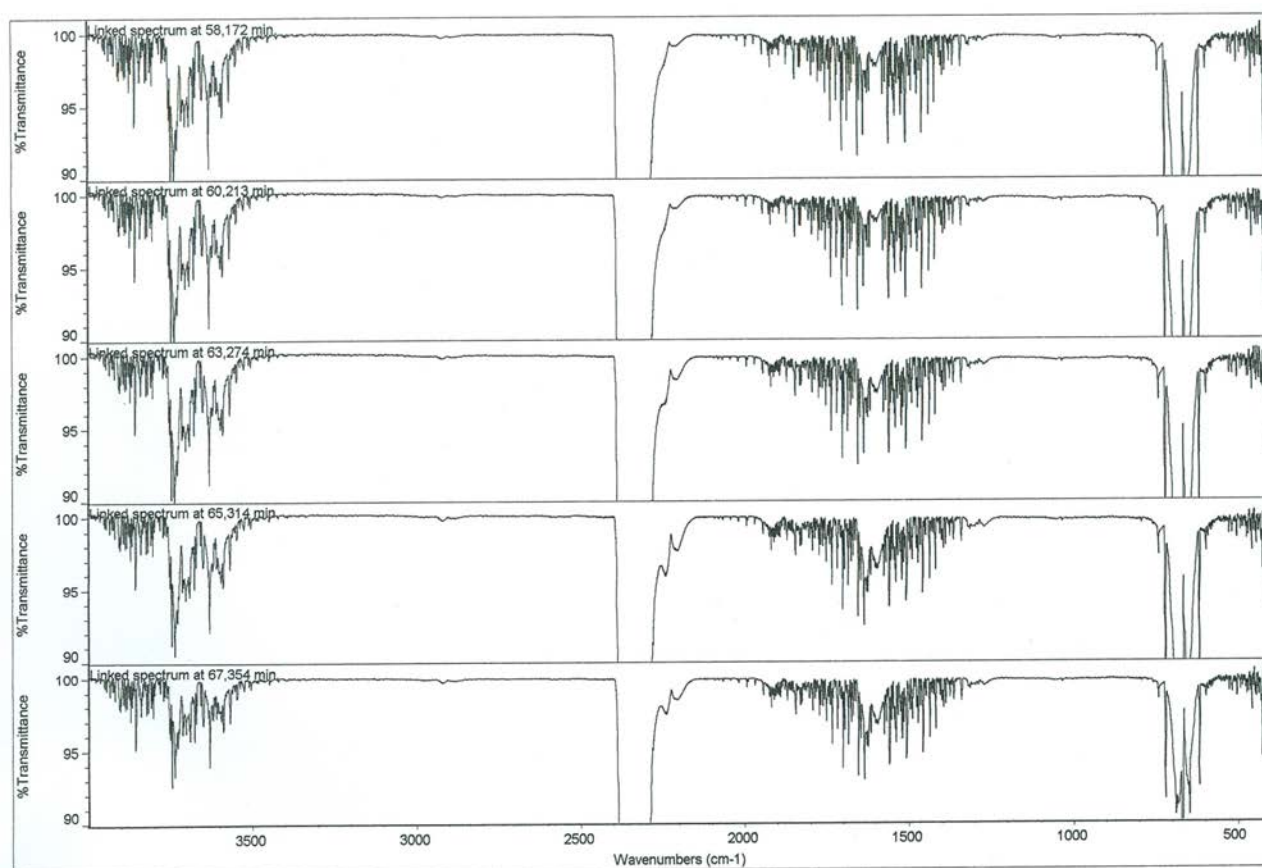
S4.5. FT-IR spectra of pure samples of the observed evolved gases.



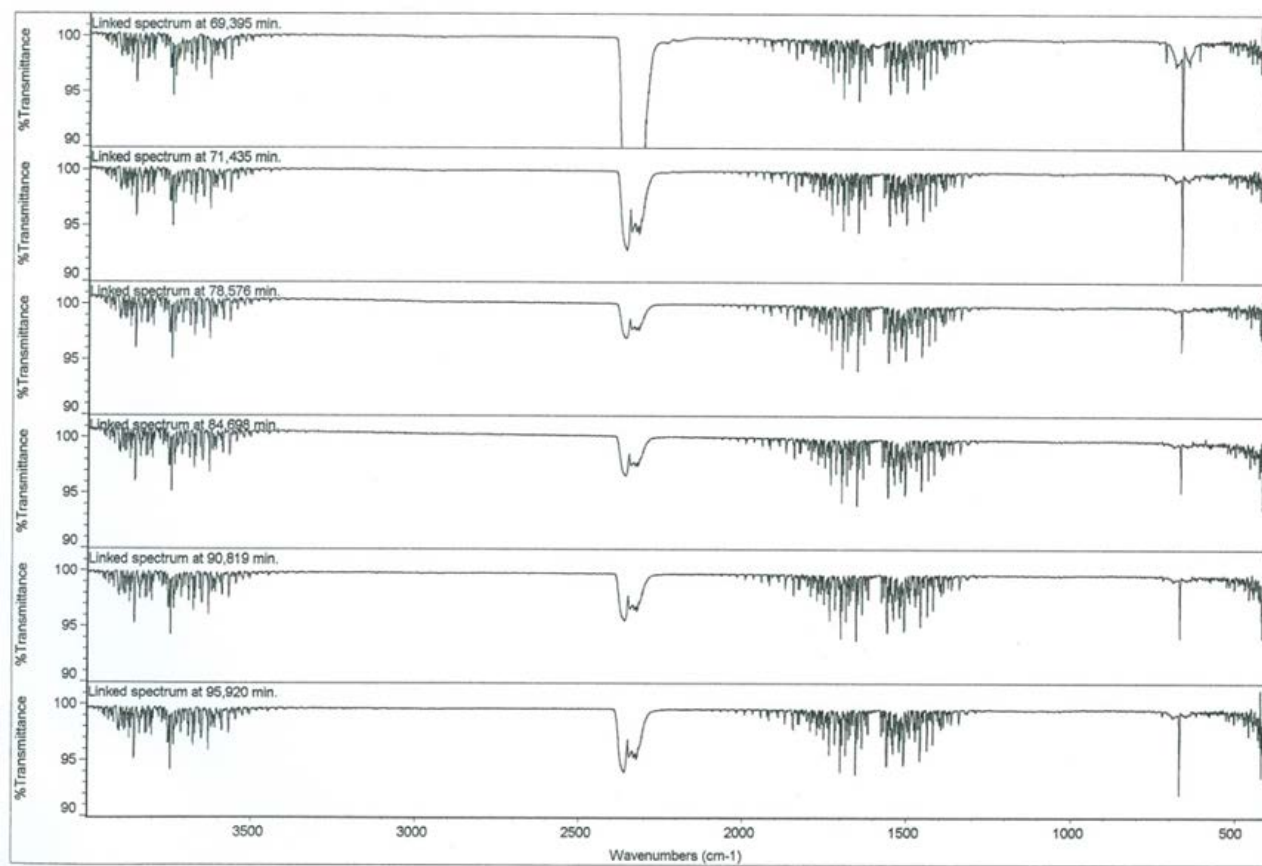
S4.6. Sheets with thirty-seven time-spaced IR spectra to identify evolved gases.







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S4.7. Selected of four time-spaced IR spectra enhancing the production of methanol, during the step 3 (first two spectra) and the step 4 (the two below).

