



Article The Structural, Electronic, Magnetic and Elastic Properties of Full-Heusler Co₂CrAl and Cr₂MnSb: An Ab Initio Study

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Abstract: In this paper, the full-potential, linearized augmented plane wave (FP-LAPW) method was employed in investigating full-Heusler Co₂CrA1's structural, elastic, magnetic and electronic properties. The FP-LAPW method was employed in computing the structural parameters (bulk modulus, lattice parameters, c/a and first pressure derivatives). The optimized structural parameters were determined by generalized gradient approximation (GGA) for the exchange-correlation potential, V_{xc}. Estimating the energy gaps for these compounds was accomplished through modified Becke–Johnson potential (mBJ). It was found that the conventional Heusler compound Co₂CrA1 with mBJ and CGA approaches had a half-metallic character, and its spin-down configuration had an energy gap. It was also found that the conventional and inverse Heusler Cr₂MnSb and tetragonal (139) (Co₂CrA1, Cr₂MnSb) compounds with a half-metallic character had direct energy gaps in the spin-down configuration. To a certain degree, the total magnetic moments for the two compounds were compatible with the theoretical and experimental results already attained. Mechanically, we found that the conventional and inverse full-Heusler compound Co₂CrA1 was stable, but the inverse Cr₂MnSb was unstable in the ferromagnetic state. The conventional Heusler compound Cr₂MnSb was mechanically stable in the ferromagnetic state.

Keywords: full-Heusler compound; electronic band structure; magnetic order; elastic properties; FP-LAPW

1. Introduction

Since their discovery in 1903, Heusler compounds have found many applications including spintronics [1], shape-memory devices [2] and thermoelectric power generators [3]. Heusler compounds have a type of face-centered cubic (fcc) crystal structure. These compounds can be categorized into two classes: XYZ (half-Heuslers), which consist of three FCC sub-lattices, and X₂YZ (full-Heuslers), which have four FCC sub-lattices, where transition elements are represented by X and Y, and the s, p ... elements are represented by Z [4].

Full-Heusler X₂YZ compounds crystallize in two kinds of inverse and conventional forms. Conventional Heusler compounds crystallize in a Cu₂MnAl structure with a space group of Fm-3m (space group number 225) having atomic positions of X₂ (1/4,1/4,1/4), (3/4,3/4,3/4), Y (1/2,1/2,1/2) and Z (0,0,0). Inverse Heusler compounds crystallize in a Hg₂CuTi structure with space group of F-43m (space group number 216) having atomic positions of X₂ (1/4,1/4,1/4), (1/2,1/2,1/2), Y (3/4,3/4,3/4) and Z (0,0,0) [5]. Some Heusler compounds have a half-metallic (HM) character [5–18], where only a single conduction spin channel exists for half metals. For one spin channel, the spin-polarized band structure



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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). shows metallic behavior. On the other hand, at the Fermi level, the other spin band structure shows a gap. Therefore a 100% spin polarization is exhibited by half-metallic materials.

Certain studies have explored the electronic, magnetic, elastic and structural properties of these compounds using different methods. Zhang et al. [19] focused on the Co₂CrAl Heusler compound's electronic band structure and transport properties. They measured the lattice parameter, magnetic moment and indirect band gap and found those to be 5.74, 3 μ_B and 0.475 eV, respectively. Hakimi et al. [20] conducted an experimental study of the Co₂CrAl compound's magnetic and structural properties. They found that the conventional Co₂CrAl's total magnetic moment was 2 μ_B .

Ozdogan and Galanakis [21] determined half-metallic antiferromagnetic Cr₂MnSb's magnetic and electronic properties for both conventional and inverse structure types. They found that for both structural types, Cr₂MnSb is a half-metallic ferrimagnetic compound for a broad array of lattice constants.

Heusler alloys are well-known for their potential application in the spin-transfer torque (STT) sector. These materials crystallize in multifaceted structures in both cubic and tetragonal symmetries with multiple magnetic sublattices. Galanakis [22] conducted research on the magnetic and electronic properties of both full-Heusler and half-Heusler alloys. The full-Heusler alloys investigations included Co₂MnSi and Co₂MnGe, and the half-Heusler alloys included PtMnSb, CoMnSb and NiMnSb.

Atsufumi Hirohata et al. [23] reviewed the development of anti-ferromagnetic (AFM) Heusler alloys for the replacement of iridium as a critical raw material (CRMs). They established correlations between the crystalline structure of these alloys and the magnetic properties, i.e., antiferromagnetism. This study revealed that the Heusler alloys consisting of elements with moderate magnetic moments require perfectly or partially ordered crystalline structures to exhibit AFM behavior. Using elements with large magnetic moments, a fully disordered structure was found to show either AFM or ferrimagnetic (FIM) behavior. The considered alloys may become useful for device applications due to the additional increase in their anisotropy and grain volume being able to sustain AFM behavior above room temperature.

Recently, Abu Baker et al. [24] investigated the elastic, electronic, magnetic and structural characteristics of half-metallic ferromagnetic full-Heusler alloys, namely conventional Co₂TiSn and inverse Zr₂RhGa, employing the FP-LAPW technique. The lattice parameters for the conventional Co₂TiSn and inverse Zr₂RhGa were found to be 6.094 A⁰ and 6.619 A⁰, respectively. In addition, the total magnetic moments for these compounds were recorded as 1.9786 μ_B and 1.99 μ_B , respectively. The conventional Co₂TiSn and inverse Zr₂,RhGa compounds had indirect energy gaps of 0.482 eV and 0.573 eV, respectively. From their electronic properties, it can be noted that the conventional full-Heusler Co₂TiSn compound and the inverse full-Heusler Zr₂RhGa compound had stability from a mechanical perspective.

Furthermore, Gupta et al. prepared Cr_2MnSb thin films on a MgO (001) substrate using the DC/RF magnetron sputtering method. The XRD analysis of the deposited films revealed that they crystalized in a cubic phase with full B2 and partial L2₁ ordering [25]. Previously, Dubowik et al. deposited 100 nm Co₂CrAl films on glass and NaCl substrates using the flash evaporation technique [26].

Paudel and Zhu [27] showed that the full Heusler alloy Co₂ScSb is stable at the ferromagnetic phase with an optimized lattice constant of 6.19. They confirmed the structural stability from the calculations of the negative cohesive, formation energy and real phonon frequency. Paudel and Zhu [28] also showed that the half-metallic ferromagnetic properties of a Fe₂MnP alloy have energy band gaps of 0.34 eV and half-metallic gaps of 0.09 eV at an optimized lattice parameter of 5.56.

In this article, the motivation for investigating the mechanical, electronic and magnetic characteristics of the full-Heusler compounds Co₂CrAl and Cr₂MnSb in both the conventional and inverse form was to study in detail their mechanical and structural stability and preferable magnetic phase, as well as to introduce their elastic properties and behaviors. The article is organized as follows: after the introduction and background review, the

computational methods and model are introduced. This is followed by a presentation of the results, discussion and conclusion.

2. Computational Method

In the current study, the calculations were accomplished using the full-potential, linearized augmented plane wave procedure executed in the WIEN2k [29] suite. Generalized gradient approximation (GGA) was used to calculate the structural parameters, i.e., the lattice parameters and bulk modulus. The GGA method depends on the local gradient of the electronic density in addition to the value of the density, giving a more accurate description of variations in the electron–electron interactions. GGA functionals provide a severe underestimation of the energy band gaps, so a modified Becke–Johnson (mBJ-GGA) functional was used to improve the energy band gaps. For the compound Co_2CrAl , the muffin-tin radii (R_{MT}) of the Co, Cr and Al atoms were taken to be 2.1, 2.05 and 1.95 a.u., respectively, and for the compound Cr_2MnSb , the R_{MT} of Cr, Mn and Sb atoms were 2.14, 2.2 and 2.2 a.u., respectively. Moreover, 35 special k-points in the irreducible Brillion Zone (IBZ) with a grid size of $10 \times 10 \times 10$ (equal to 1000 k-points in the Full Brillion Zone (FBZ)) [30] were employed in obtaining self-consistency calculations for the Co₂CrAl and Cr₂MnSb compounds. In addition, the plane waves quantity was limited as $K_{MAX} \times R_{MT} = 8$, and the wave functions' expansions was set by l = 10 inside the muffin-tin spheres. Furthermore, the self-consistent computations were only perceived as well-converged when the computed aggregate crystal energy converged to lower than 10^{-5} Ry. In addition, the cubic phase's elastic constants were computed using the second-order derivatives within the WIEN2k-code-contained formalism.

3. Results and Discussion

3.1. Structural Properties

By fitting the total energy to Murnaghan's equation of state (EOS) [31], the optimized lattice constant (a), bulk modulus (B) and its pressure derivative (B') were computed as given below:

$$E(\nu) = E_0 + \frac{BV}{B'} \left| \frac{\left(\frac{V_0}{V}\right)^{B'}}{B' - 1} + 1 \right| - \frac{B'V_0}{B' - 1}$$
(1)

where *B* represents the bulk modulus at the equilibrium volume, *B*' is the pressure derivative of the bulk modulus at the equilibrium volume and *E*₀ is the minimum energy. The bulk modulus (*B*) and the pressure (*P*) are given by $B = -V \frac{dP}{dV} = V \frac{d^2E}{dV^2}$ and $P = -\frac{dE}{dV}$.

The conventional Heusler Co_2CrAl and Cr_2MnSb compounds had a space group Fm-3m L21 (225) and the inverse Heusler Co_2CrAl and Cr_2MnSb compounds had a space group F-43m Xa (216) [5], while tetragonal crystal lattices were the result of stretching of the cubic lattice along with one of its vectors. This resulted in the cube taking the shape of a rectangular prism whose base was a square (a by a), and the height (c) was different from the base edge a. Therefore, the tetragonal Heusler Co_2CrAl and Cr_2MnSb compounds had space groups of I4/mmm (139) and I-4m2 (119). The full-Heusler structural properties of the Co_2CrAl and Cr_2MnSb compounds were calculated. Figure 1 presents the crystal structures of the full-Heusler Co_2CrAl and Cr_2MnSb compounds. The aggregate energy as a function of the volume for the Heusler Co_2CrAl and Cr_2MnSb compounds are presented in Figures 2 and 3. Moreover, the state (EOS) was used to compute the optimized structural parameters, presented in Tables 1 and 2.



Figure 1. Different crystal structures of Heusler Co₂CrAl. (**a**) Conventional Heusler structure Co₂CrAl Fm-3m L21 (225), (**b**) inverse Heusler Co₂CrAl structure F-43m X (216), (**c**) tetragonal structure I4/mmm (139) and (**d**) tetragonal structure I-4m2 (119).

Tables 1 and 2 show our computed lattice parameters compared with other theoretical and experimental lattice parameters of conventional and inverse Heusler Co_2CrAl and Cr_2MnSb compounds. The calculated lattice parameters for the conventional Heusler Co_2CrAl compound deviated from the measured one within 0.38% [19]. The calculated lattice parameters for the conventional and inverse Heusler Cr_2MnSb compounds perfectly agreed with the theoretical outcomes [21]. As far as we know, comparable experimental results for conventional and inverse Heusler Cr_2MnSb compounds are not available. These results ensured the reliability of the present first-principle computations.



Figure 2. The total energy (Ry) versus volume (a.u.³) for different crystal structures of Heusler Co_2CrAl .



Figure 3. The total energy (Ry) versus volume (a.u.³) for different crystal structures of Heusler Cr_2MnSb .

Structure	Space Group	Magnetic Phase	Reference	Lattice Parameter a (Å)	B (GPa)	E _{total} (Ry)/f.u	c/a
	Conventional Fm-3m (225)	NM	Present	5.6830	213.468	-8161.3694	1
			Present	5.7082		-8161.4115 [19]	1
	Conventional Fm-3m (225)	FM	M Experimental 5.7 5.7	5.74 [19] 5.70 [26]	206.811 [19]		
		Theoretical 5.73 [20]	_				
Co ₂ CrAl	Inverse F-43m (216)	NM	Present	5.6936	212.056	-8161.3309	1
	Inverse F-43m (216)	FM	Present	5.7398	169.0397	-8161.3454	1
	I4/mmm (139)	FM	Present	4.011	202.24	-8161.41148	1.4175
	I4/mmm (139)	AFM	Present	4.0239	202.133	-8161.4095	1.4175
	I-4m2 (119)	FM	Present	3.9382	166.847	-8161.34608	1.5253
	I-4m2 (119)	AFM	Present	3.9441	163.781	-8161.3458	1.5253

Table 1. Calculated lattice parameter (a), bulk modulus (B), and total energy (E_{tot}) for Heusler Co_2CrAl compound.

Table 2. Calculated lattice parameter (a), bulk modulus (B), and total energy (E_{tot}) for Heusler Cr_2MnSb compound.

Structure	Space Group	Magnetic Phase	Reference	Lattice Parameter a (Å)	B (GPa)	E _{total} (Ry)/f.u	c/a
-	Conventional Fm-3m (225)	NM	Present	6.1116	201.77	-19,487.9803	1
			Present	6.1381		-19,487.98197 1 -19,487.9644 1 -19,487.9683 1	
	Conventional Fm-3m (225)	FM	Theoretical Experimental	6.0 [21] 5.95 [25]	248.45		
	Inverse F-43m (216)	NM	Present	6.0724	220.322	-19,487.9644	1
CraMnSh			Present	6.0571			
	Inverse F-43m (216)	FM	Theoretical Result	5.9 [21]	149.5381 –19 5.9 [21]		1
	I4/mmm (139)	FM	Present	4.3337	296.852	-19,487.982	1.4158
	I4/mmm (139)	AFM	Present	4.2637	386.728	-19,487.9823	1.4158
	I-4m2 (119)	FM	Present	4.0617	142.459	-19,487.9394	1.6513
	I-4m2 (119)	AFM	Present	4.0459	153.8705	-19,487.9394	1.6513

According to the results obtained in this study, our volume optimization results showed that AFM tetragonal distortion (No. 139) was more preferred than FM cubic L2₁ for the Cr₂MnSb compound with a slightly small energy difference $\Delta E_{tet-cubic} = 0.0047 R_{f.u}^{Ry}$ (see Equation (2)). On the other hand, FM cubic L2₁ was more preferred than tetragonal distortion for the Co₂CrAl case, with an energy difference $\Delta E_{tet-cubic} = 0.002 R_{f.u}^{Ry}$. To make the AFM tetragonal phase stable, the energy difference with a cubic structure should be greater than 0.1 eV/f.u. As reported previously, Cr₂MnSb crystallizes in a cubic L2₁ structure with a fully compensated ferrimagnetic configuration, where the magnetic moment of Cr and Mn are dominated by antiparallel exchange [21].

$$\Delta E_{tet-cubic} = E_{tet} - E_{cubic} \tag{2}$$

3.2. Magnetic Properties

This part involved the calculation of the inverse, conventional and tetragonal I4/mmm (139) Heusler Co_2CrAl and Cr_2MnSb compounds' partial and total magnetic moments. The results obtained were compared with other theoretical values as shown in Tables 3 and 4.

Compounds -		Magnetic Moment in µ _B							
		Со	Co Co Cr Al Interstitial				Total Magnetic Moment (M^{tot}) in μ_B		
Inverse Co ₂ CrAl	Present	0.96069	1.36311	-1.26906	-0.00841	-0.21497	0.83116		
Conventional Co ₂ CrAl	Present	1.01815	1.01815	1.32570	-0.06082	-0.30118	3		
	Theoretical Result	0.650 [19]	0.650 [19]	1.745 [19]	-0.045 [19]	_	3 [19]		
	Theoretical Result	_	_	_	_	_	2.96 [22]		
Tetragonal I4/mmm (139) Co ₂ CrAl	Present	-0.04084	-0.04084	0.81717	1.45741	-0.05097	2.99994		

Table 3. Total magnetic moment for inverse, conventional and tetragonal I4/mmm (139) Heusler Co₂CrAl compound.

Table 4. Total magnetic moment for inverse, conventional and tetragonal I4/mmm (139) Heusler Cr_2MnSb compound.

Compounds		Magnetic Moment in µ _B						
		Cr	Cr	Mn	Sb	Interstitial	Total Magnetic Moment (M^{tot}) in μ_{B}	
Inverse Cr ₂ MnSb	Present	-1.72053	2.68899	-1.05810	0.04777	0.04187	0	
	Theoretical Result	1.96 [21]	-3.18 [21]	1.29 [21]	-	-	0 [21]	
Conventional Cr ₂ MnSb	Present	-1.51854	-1.51854	3.21064	0.06167	-0.23513	0.00011	
	Theoretical Result	1.77 [21]	1.77 [21]	-3.44 [21]	-	-	0.01 [21]	
Tetragonal I4/mmm (139) Cr ₂ MnSb	Present	0.05294	0.05294	-1.46141	3.08253	-0.20953	0.00312	

We found that the conventional and tetragonal Heusler Co₂CrAl compounds were ferromagnetic compounds. Furthermore, the total magnetic moment for the inverse Co₂CrAl compound was $M^{tot} = 0.83116 \mu_B$, while it was $M^{tot} = 3 \mu_B$ for the conventional Co₂CrAl compound. The physics interpretation behind this huge difference between the total spin magnetic moment of the conventional and inverse Co₂CrAl was due to antiparallel exchange interactions between the Cr atom and Co atom in the case of inverse Co₂CrAl, whereas it was a direct interaction in the case of the conventional phase. Therefore, it can be noted from the results produced here that the conventional Co₂CrAl compound's computed total magnetic moment perfectly matched with the prior theoretical results [19,22], as Table 3 shows. Theoretically, a compound with a total magnetic moment M^{tot} with an integer value means it is a half-metallic material.

Table 4 shows the results for the inverse, conventional and tetragonal I4/mmm (139) Heusler Cr_2MnSb compounds. Table 4 shows that the inverse Heusler Cr_2MnSb had a negative spin moment for one Cr atom and one Mn atom and a positive spin moment for the second Cr atom. The conventional Heusler Cr_2MnSb had a negative spin moment for its Cr atoms and a positive spin moment for its Mn atom. The Sb atom's spin moment was extremely tiny in both structural types of Cr_2MnSb . The electronic configurations in the Mn and Cr atoms were similar, and they had one electron difference. Consequently, their exchange maintained the compound's ferromagnetic character, which led to small variations in the spin moments per site.

We found that the conventional Heusler Cr_2MnSb had a small total magnetic moment (non-zero total magnetization) due to the decrease in the atomic disorder in the Mn–Sb sublattice. This implied that the conventional Heusler Cr_2MnSb compound had a ferrimagnetic order. On the other hand, we found that the inverse Heusler Cr_2MnSb had a zero total magnetic moment, which meant that this compound had an antiferromagnetic magnetic order. The tetragonal Cr_2MnSb had a small total magnetic moment, which meant that this compound was ferrimagnetic.

3.3. Electronic Properties

In this section, the partial and total density of states and the band structure for the inverse and conventional Heusler (Co₂CrAl, Cr₂MnSb) compounds were investigated. An analysis of the density of states and band structure showed that the conventional Co₂CrAl, conventional Cr₂MnSb and inverse Cr₂MnSb Heusler compounds exhibited a half-metallic conduct in a ferromagnetic state. This implied that the spin-up electrons in the materials had a metallic behavior; when they behaved as semiconducting with a spin-down direction, the materials had a semiconducting behavior. On the other hand, the inverse Heusler Co₂CrAl, compound had a metallic behavior. The tetragonal I4/mmm (139) Heusler (Co₂CrAl, Cr₂MnSb) compounds had a half-metallic behavior in the antiferromagnetic state.

Figure 4a,b shows the metallic behavior of the spin up and spin down band structures within the PBE-GGA method for the inverse Heusler Co₂CrAl compound with zero energy gap. Figure 5a,b also shows the metallic behavior of the spin-up and spin-down band structures within the mBJ-GGA method for the inverse Heusler Co₂CrAl compound with zero energy gap.



Figure 4. The band structure for the inverse Heusler Co₂CrAl compound by employing the PBE-GGA technique for (**a**) spin-up inverse Heusler Co₂CrAl compound and (**b**) spin-down inverse Heusler Co₂CrAl compound.



Figure 5. The band structure for the inverse Heusler Co_2CrAl compound by employing the mBJ-GGA technique for (**a**) spin-up inverse Heusler Co_2CrAl compound (**b**) spin-down inverse Heusler Co_2CrAl compound.

From Figure 6a,b, the conventional Heusler Co₂CrAl compound's band structure had an indirect energy band gap (spin down) using the PBE-GGA technique. In addition, Figure 7a,b shows that the conventional Heusler Co₂CrAl compound's band structure had an indirect energy band gap (spin-down) when using the mBJ-GGA technique. As indicated in Table 5, the indirect energy gaps within PBE-GGA and mBJ-GGA were 0.6 eV and 0.9 eV, respectively.

Table 5. The energy band gaps for conventional and inverse Co_2CrAl compound using PBE-GGA and mBJ methods.

Compounds	Band Gap Type	High Symmetry Lines	<i>Eg</i> -PBE-GGA (eV)	<i>E_g</i> -mBJ-GGA (eV)	
Conventional -Co ₂ CrAl	Indirect	$\Gamma - X$	0.6	0.9	
$Inverse-Co_2CrAl$	Metallic	-	-	-	

From Figure 8a,b, the band structure (spin-down) of the inverse Heusler Cr_2MnSb compound had a direct energy band gap using the PBE-GGA technique. In addition, Figure 9a,b shows that the inverse Heusler Cr_2MnSb compound's band structure (spin-down) had a direct energy band gap when the mBJ-GGA technique was used. As can be seen in Table 6, the direct energy gaps within the PBE-GGA and mBJ-GGA methods were 0.8 eV and 0.9 eV, respectively.



Figure 6. The band structure for the conventional Heusler Co₂CrAl compound by employing the PBE-GGA technique for (**a**) spin-up conventional Heusler Co₂CrAl compound and (**b**) spin-down conventional Heusler Co₂CrAl compound.



Figure 7. The band structure for the conventional Heusler Co₂CrAl compound by employing the mBJ-GGA technique for (**a**) spin-up conventional Heusler Co₂CrAl compound and (**b**) spin-down conventional Heusler Co₂CrAl compound.



Figure 8. The band structure for the inverse Heusler Cr_2MnSb compound by employing the PBE-GGA technique for (**a**) spin-up inverse Heusler Cr_2MnSb compound and (**b**) spin-down inverse Heusler Cr_2MnSb compound.



Figure 9. The band structure for the inverse Heusler Cr_2MnSb compound by employing the mBJ-GGA technique for (**a**) spin-up inverse Heusler Cr_2MnSb compound and (**b**) spin-down inverse Heusler Cr_2MnSb compound.

Compounds	Band Gap Type	High Symmetry Lines	<i>Eg</i> -PBE-GGA (eV)	<i>Eg-</i> mBJ-GGA (eV)
Conventional- Cr ₂ MnSb	Direct	Γ	0.9	1
Inverse-Cr ₂ MnSb	Direct	Г	0.8	0.9

Table 6. The energy band gaps for conventional and inverse Cr_2MnSb compound using PBE-GGA and mBJ methods.

Figure 10a,b shows that the band structure of the conventional Heusler Cr_2MnSb compound had a direct energy band gap (spin-down) using the PBE-GGA technique. In addition, Figure 11a,b shows that the Heusler Cr_2MnSb compound's band structure had a direct energy band gap (spin-down) using the mBJ-GGA technique. As indicated in Table 6, the indirect energy gaps within PBE-GGA and mBJ-GGA methods were 0.9 eV and 1 eV, respectively.



Figure 10. The band structure for the conventional Heusler Cr_2MnSb compound by employing the PBE-GGA technique for (**a**) spin-up conventional Heusler Cr_2MnSb compound and (**b**) spin-down conventional Heusler Cr_2MnSb compound.

Figure 12a,b presents the Heusler Co₂CrAl compound band structure for tetragonal I4/mmm (139) in the AFM state. Figure 12a shows that the tetragonal I4/mmm (139) Heusler Co₂CrAl compound's spin-up band structure had a metallic character, while Figure 12b indicates that the tetragonal I4/mmm (139) Heusler Co₂CrAl compound's spin-down band structure had a direct energy band gap. The direct energy gap was found to be 0.8 eV, as shown in Table 7. Figure 13a,b shows the tetragonal I4/mmm (139) Heusler Cr₂MnSb compound's band structure in the AFM state. Figure 13a indicates that the tetragonal I4/mmm (139) Heusler Cr₂MnSb compound's spin-up band structure had a metallic nature, while Figure 13b illustrates that the tetragonal I4/mmm (139) Heusler Cr₂MnSb compound's spin-down band structure had a direct energy band gap. The direct energy band gap. The direct energy band gap. The direct energy band structure had a metallic nature, while Figure 13b illustrates that the tetragonal I4/mmm (139) Heusler Cr₂MnSb compound's spin-down band structure had a direct energy band gap. The direct energy gap was found to be 0.9 eV, as shown in Table 7.



Figure 11. The band structure for the conventional Heusler Cr_2MnSb compound by employing the mBJ-GGA technique for (**a**) spin-up conventional Heusler Cr_2MnSb compound and (**b**) spin-down conventional Heusler Cr_2MnSb compound.



Figure 12. The band structure for the tetragonal I4/mmm (139) Heusler Co₂CrAl compound in AFM state. (**a**) Spin-up tetragonal I4/mmm Heusler (139) Co₂CrAl compound and (**b**) spin-down tetragonal I4/mmm Heusler (139) Co₂CrAl compound.

Compounds	Band Gap Type	High Symmetry Lines	$E_g(eV)$
Co ₂ CrAl	Direct	Γ	0.8
Cr ₂ MnSb	Direct	Γ	0.9

Table 7. The energy band gaps for the tetragonal I4/mmm (139) Heusler Co_2CrAl and Cr_2MnSb compounds in AFM state.



Figure 13. The band structure for the tetragonal I4/mmm (139) Heusler Cr₂MnSb compound in AFM state. (a) Spin-up tetragonal I4/mmm Heusler (139) Cr₂MnSb compound and (b) spin-down tetragonal I4/mmm Heusler (139) Cr₂MnSb compound.

For the conventional and inverse Heusler Co_2CrAl and Cr_2MnSb compounds, the partial and total density of states for the spin-down, spin-up and inverse Heusler Co_2CrAl and Cr_2MnSb compounds are presented in Figures 14–17. The density of states in Figures 14–17 also show half-metallic behaviors for the conventional Heusler Co_2CrAl and the inverse and conventional Heusler Cr_2MnSb compounds with a minor energy band gap in the spin-down segment. This implied that the behavior of these compounds was half- metallic.

In the conventional Co₂CrAl/s (Figure 15) spin-down segment, the valence band resulted from the d-state of Co, the d-state of Cr and the tiny effect of the Al in the s-state. The d-state of Co, the d-state of Cr and the tiny effect of the Al in the s-state were accredited to the conduction band. In the spin-down channel of the conventional Co₂CrAl, the valence band was attributed to the d-state of Co, the d-state of Cr and the minor effect of Al in the s-state of Cr and the minor effect of Al in the s-state of Cr and the minor effect of Al in the s-state of Cr and the minor effect of the Al in the s-state of Cr and the minor effect of Al in the s-state of Cr and the minor effect of the Al in the s-state.

Figure 14. (a) Total density of states for the inverse Co_2CrAl compound and partial density of states for (b) Co1 atom, (c) Co2 atom, (d) Al atom and (e) Cr atom.

Figure 15. (a) Total density of states for the conventional Co₂CrAl compound and the partial density of states for (b) Co atom, (c) Cr atom and (d) Al atom.

Figure 16. (a) Total density of states for the inverse Cr_2MnSb compound and partial density of states for (b) Cr1 atom, (c) Cr2 atom, (d) Sb atom and (e) Mn atom.

Figure 17. (**a**) Total density of states for the conventional Cr_2MnSb compound and the partial density of states for (**b**) Cr atom, (**c**) Mn atom and (**d**) Sb atom.

In the spin-up sector of the inverse Cr_2MnSb , (Figure 16), the valence band was attributed to the d-state of Cr, the d-state of Mn, the s-state and the p-state of Sb. On the other hand, the conduction band resulted from the d-state of Cr, the d-state of Mn, the minor effect of the s-state and Sb's p-state. In the spin-down channel of the inverse Cr_2MnSb , the valence band was attributed to the minor contribution of the d-state of Cr, the d-state of Sb. Meanwhile, the conduction band was attributed to the d-state of Cr, the d-state of Sb.

In the conventional Cr_2MnSb (Figure 17) spin-up sector, the valence band was attributed to the d-state of Cr, the d-state of Mn, the s-state and the p-state of Sb. On the other hand, the conduction band was attributed to the d-state of Cr, the minor effect of the s-state and the p-state of Sb. In the spin-down channel of the conventional Cr_2MnSb , the valence band was attributed to the d-state of Cr, the d-state of Mn, the s-state and the p-state of Sb, while the conduction band was the result of the d-state of Cr, the minor effect of the s-state and the p-state of Sb.

3.4. Elastic Properties

This part presents the computation of the bulk modulus (*B*), the shear modulus (*S*), the elastic constants (C_{ij}), the *B*/*S* ratio, Poisson's ratio, Young's modulus (*Y*) and the anisotropic factor (**A**) of the inverse and conventional Heusler Cr₂MnSb and Co₂CrAl compounds. The standard mechanical stability condition or cubic crystal [32] was $C_{11} > 0$, $C_{11} + 2C_{12} > 0$, $C_{11} - C_{12} > 0$ and $C_{44} > 0$.

Table 8 presents our calculations for the inverse and conventional Heusler Cr_2MnSb and Co_2CrAl compounds. We concluded that the inverse and conventional Heusler Co_2CrAl were mechanically stable. The inverse Cr_2MnSb was found to be mechan-

ically unstable in the ferromagnetic state. On the other hand, the conventional Cr₂MnSb was mechanically stable in the ferromagnetic state.

Table 8. Reuss's bulk modulus (*B*), shear modulus (*S*), elastic constants (C_{ij}), *B*/S ratio, Poisson's ratio (v), Young's modulus (Y) and anisotropic factor (A) of the FM conventional and inverse Heusler (Co_2CrAl , Cr_2MnSb) compounds.

Materials	С ₁₁ (GPa)	С ₁₂ (GPa)	C ₄₄ (GPa)	B (GPa)	S (GPa)	B/S	Y (GPa)	υ	Α
Conventional Co ₂ CrAl	268.592	169.462	154.343	202.505	83.628	2.422	220.528	0.319	3.114
Inverse Co ₂ CrAl	275.6864	234.2043	138.9022	248.032	42.364	5.855	120.246	0.470	6.697
Inverse Cr ₂ MnSb	197.1538	138.7903	-488.5573	158.245	80.134	1.975	205.683	0.283	-16.7
Conventional Cr ₂ MnSb	267.3836	224.7424	100.9041	238.956	40.474	5.904	114.933	0.420	4.732

We used the Reuss approximation [33] to calculate the bulk and shear modulus. The following equation can be used to calculate the Reuss shear modulus S_R :

$$S_R = \frac{5C_{44}(C_{11} - C_{12})}{4C_{44} + 3(C_{11} - C_{12})}$$
(3)

The following equation gives the cubic structure's bulk modulus:

$$B = \frac{1}{3}(C_{11} + 2C_{12}) \tag{4}$$

The Young modulus (*Y*) is given by the following:

$$Y = \frac{9BS_R}{(S_R + 3B)} \tag{5}$$

The anisotropic factor and Poisson's ratio are given by the following:

$$A = \frac{2C_{44}}{C_{11} - C_{12}} \tag{6}$$

$$v = \frac{3B - 2S_R}{2(3B + S_R)}$$
(7)

Reuss's bulk modulus (*B*), the shear modulus (*S*), the elastic constants (C_{ij}), the B/S ratio, Poisson's ratio (v), Young's modulus (Y) and the anisotropic factor (A) of the FM conventional and inverse Heusler Co₂CrAl and Cr₂MnSb compounds are shown in Table 8.

A material's hardness is measured by its shear modulus and bulk modulus [33]. Therefore, the ratio B/S measures a specific material's brittleness and ductility. A material is ductile when B/S < 1.75. Otherwise, it is brittle [34]. From the present calculations in Table 8, the B/S ratios of the inverse and conventional Heusler Co₂CrAl compounds were 5.855 and 2.422, respectively. Both the inverse and conventional Heusler compounds were ductile in nature, depending on the B/S ratio values. The B/S ratio values for the inverse and conventional Heusler Co₂CrAl, respectively. Both the inverse found to be 1.975 and 5.904, respectively. Both the inverse and conventional Heusler Cr₂MnSb compounds had a ductile character, depending on the B/S ratio values.

The stiffness of materials is measured using Young's modulus. Materials with a higher Young's modulus (Y) value are stiffer. Poisson's ratio can be employed for understanding the character of bonding and stability of a material. A Poisson's ratio value higher than

0.26 indicates that the material is ductile; otherwise, it is brittle [35]. From the present calculations summarized in Table 8, the Poisson's ratio values of the inverse and conventional Heusler Co₂CrAl compounds were 0.470 and 0.319, respectively. Depending on the Poisson's ratio values, both inverse and conventional Heusler compounds had a ductile nature. The Poisson's ratio values for the inverse and conventional Heusler Cr₂MnSb compounds were found to be 0.283 and 0.420, respectively. Depending on the Poisson's ratio values, both the inverse and conventional Heusler Cr₂MnSb compounds had a ductile nature. Poisson's ratio for compounds with covalent bonds is lower than 0.25, while for compounds with dominating ionic bonds, Poisson's ratio lies between 0.25 to 0.50. From Table 8, it appeared that the conventional and inverse Heusler Co₂CrAl and Cr₂MnSb compounds had prominent ionic bonds. In the same vein, elastic anisotropy is a crucial parameter for measuring the level of material's anisotropy [36]. The value of A is unity for an isotropic material. Otherwise, the elastic anisotropy of the material is elastic [37]. The present values of the anisotropy factor in Table 8 for the inverse and conventional Heusler Co₂CrAl and Cr₂MnSb compounds showed that these compounds were anisotropic elasticity.

4. Conclusions

This study focused on the elastic, magnetic, electronic and structural properties of inverse and conventional Heusler (Co₂CrAl, Cr₂MnSb) compounds. The results showed that the conventional Heusler Co₂CrAl, the conventional and inverse Heusler Cr₂MnSb and the tetragonal (139) Heusler Co₂CrAl and Cr₂MnSb compounds were half-metals. This halfmetallic character is a promising characteristic of materials for spintronic applications. The indirect energy gap of the conventional Heusler Co₂CrAl compound was 0.6 eV within the PBE-GGA scheme. The energy band gap within the mBJ-GGA scheme for the conventional Heusler Co₂CrAl compound was computed to be 0.9 eV. Within the PBE-GGA technique, the inverse and conventional Heusler Cr2MnSb compounds had a direct energy band gap of 0.8 eV and 0.9 eV, respectively. Within the mBJ-GGA method, the energy gaps for the inverse and conventional Heusler Cr₂MnSb compounds were 0.9 eV and 1 eV, respectively. The tetragonal Heusler Co₂CrAl and Cr₂MnSb compounds had direct energy band gaps and were 0.8 eV and 0.9 eV, respectively, within the PBE-GGA method. We discovered that the conventional Heusler Co₂CrAl compound was a ferromagnetic compound with a total magnetic moment of $M^{tot} = 3 \mu_B$. On the other hand, the total magnetic moment for the inverse Co₂CrAl compound was $M^{tot} = 0.831 \ \mu_B$. The conventional and tetragonal Heusler Cr₂MnSb compounds had a small total magnetic moment, which meant that these compounds were ferromagnetic. We found that the conventional and inverse Heusler Co₂CrAl compounds were mechanically stable. However, the inverse Cr₂MnSb compound was mechanically unstable in the ferromagnetic state. On the other hand, the conventional Cr₂MnSb is mechanically stable in the ferromagnetic state. The B/S values indicated that the inverse and conventional Heusler compounds for both Cr₂MnSb and Co₂CrAl had ductile characteristics. From the Poisson's ratio values, we found that the conventional and inverse Heusler Co₂CrAl and Cr₂MnSb compounds had dominant ionic bonds. Finally, the conventional and inverse Heusler Co₂CrAl and Cr₂MnSb compounds were anisotropy elastic.

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