



## Chemical-physical characterization of binary mixtures of a twist bend nematic liquid crystal with smectogens

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**Figure S1.** Polarizing optical micrographs of CB7CB observed at different temperatures in a planar cell (EHC, 10µm). (a) and (b) Nematic phase at 105°C. (c) and (d) Multiples domains of NTB phase appearing in the N phase during the N/NTB transition at 104°C. The difference between the two phases is quite visible. (e) and (f) NTB phase observed at 102°C. Crossed polarizers, r indicates the rubbing direction.



**Figure S2.** (a) Growth of needle crystals in the N<sub>TB</sub> phase of CB7CB at 95°C in a planar cell (EHC, thickness 10µm). The crystallization was induced by a CB7CB crystal seed in contact with the periphery of the cell (at a distance about 10mm). (b) Image of the same region taken 1 min later. The crystal growth velocity is thus about 100µm.min<sup>-1</sup> and the cell totally crystallizes in several hours. Crossed polarizers, r indicates the rubbing direction.



**Figure S3.** X-rays diffracted intensities of 8CB/CB7CB mixtures as a function of wave vector q. (a) Mixture at a concentration  $\phi = 10.3$  wt% of CB7CB in SmA phase at 10°C and in N phase at 40°C. (b) Mixture at a concentration  $\phi = 54.9$  wt% of CB7CB in N<sub>TB</sub> phase at 20°C and N phase at 50°C.



**Figure S4.** POM of the NTB phase of a 8CB/CB7CB mixture ( $\phi = 25 \text{ wt\%}$ ) at -20°C. Crossed polarizers.



**Figure S5** Contact experiments at room temperature of various smectogens -(a) 8OCB, (b) 10CB, (c) 8CB- with CB7CB. In the vicinity of the contact we never observed a NTB/SmA front interface but the sequence SmA-N-NTB phases (POM).



Figure S6 Partial phase diagram of the 8OCB/CB7CB binary system.

## Crystals, Supplementary Materials



**Figure S7.** POM of an 8CB/CB7CB mixture (50/50 wt%) in a10µm thick cell. **(a,b)** Destabilization at 26°C of the N<sub>TB</sub> phase towards needle crystals and a N phase. The destabilization was induced by a crystal seed of CB7CB momentarily in contact with the periphery of the cell (at a distance of a few mm of the observed zone). **(c)** After a few hours at 26°C, an equilibrium state is reached. **(d,e,f)** Under rapid heating conditions (5 °C.min<sup>-1</sup>), the N phase of this zone transformed into the isotropic phase at about 56°C **(d)**. The melting of the crystals was observed at higher temperatures (76°C in **(e)**) and was almost complete when reaching 86°C **(f)**. Micrographs obtained with crossed polarizers. The double arrow indicates the rubbing direction of the cell.



**Figure S8. (a)** Heat flow curve of pure CB7CB is characterized by two distinct peaks of similar magnitude corresponding respectively to the Iso/N and N/NTB phase transitions. **(b)** A mixture at concentration  $\phi$ =43 wt% still displays a strong Iso/N phase transition peak but the N/NTB one is barely visible, in the shape of a broad weak hump.



**Figure S9.** POM texture at room temperature of the 8CB/CB7CB binary system at a concentration  $\phi = 50$  wt%. The N<sub>TB</sub> phase shows the typical stripes and ropes textures. Planar anchoring: the arrow indicates the rubbing direction of the PVA layers.