

Supplementary Materials



## Development of Co(OH)<sub>x</sub>F<sub>2-x</sub> Nanosheets for Acetone Gas Sensor Applications: Material Characterization and Sensor Performance Evaluation

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Figure S1. EDS spectrum of Co(OH)xF2-x nanosheets with atomic percentages of each element.



**Figure S2.** Profile image of the interdigitated sensor device with coated Co(OH)<sub>x</sub>F<sub>2-x</sub> nanosheets powders.



Figure S3. N2 adsorption-desorption isotherm with calculated BET surface area (193.38 m<sup>2</sup>/g).



**Figure S4.** TEM bright-field images of the  $Co(OH)_xF_{2-x}$  nanosheets (a) before and after annealing at (b) 350, (c) 400, and (d) 450 °C for 2 h.



Figure S5. Responses of Co(OH) $_xF_{2-x}$  nanosheets-based sensor toward acetone gas of various concentrations at 200 °C.



**Figure S6.** Responses of a material synthesized without NH<sub>4</sub>F (other synthesis procedures were same with the Co(OH)<sub>x</sub>F<sub>2-x</sub> nanosheets) at various temperatures toward 4.5 ppm of acetone (Responses of this material at temperatures lower than 150 °C were immeasurable because of high resistance).



Figure S7. Responses of  $Co(OH)_xF_{2-x}$  without SiO<sub>2</sub> templates at various temperatures toward 4.5 ppm of acetone.



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