

Supplementary Material: Conversion of Carbon Monoxide into Methanol on Alumina-Supported Cobalt Catalyst: Role of the Support and Reaction Mechanism—A Theoretical Study

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Tables:

Table S1. The calculated energy (E_{cluster}) of Co_4 clusters with different number of unpaired electrons

Number of unpaired electrons	E_{cluster} , eV
10	-2871.125081
12	-2873.933653
14	-2872.605828

Table S2. The calculated interaction energy (E_{int}) between Co_4 cluster and Al_2O_3 .

The interaction energy E_{int} is calculated by formula: $E_{int} = E(Co_4-Al_2O_3) - E(Co_4) - E(Al_2O_3)$.

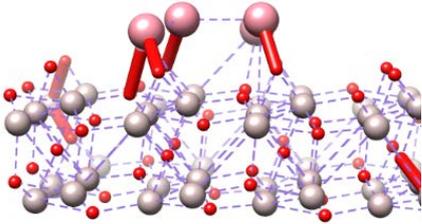
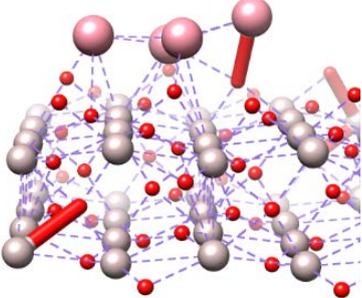
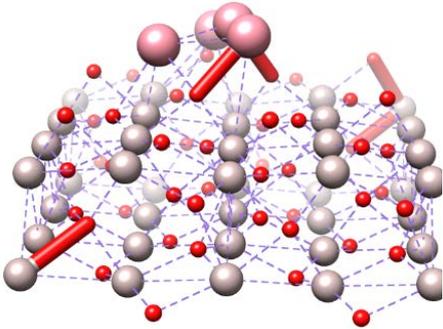
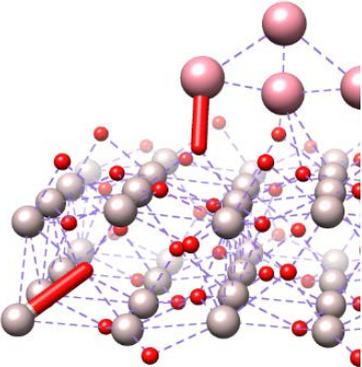
Structure	E_{int} , $kJ.mol^{-1}$	Structure	E_{int} , $kJ.mol^{-1}$
	-1027.3		-1023.3
	-1023.5		-1050.4

Table S3. The calculated adsorption energies (E_{ads}) for the adsorption of CO on Co_4 and $\text{Co}_4/\text{Al}_2\text{O}_3$ systems.

$$E_{\text{ads}} = E(\text{CO-Adsorbent}) - E(\text{CO}) - E(\text{Adsorbent}), \quad \text{where adsorbent: } \text{Co}_4 \text{ or } \text{Co}_4/\text{Al}_2\text{O}_3.$$

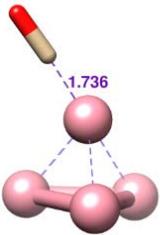
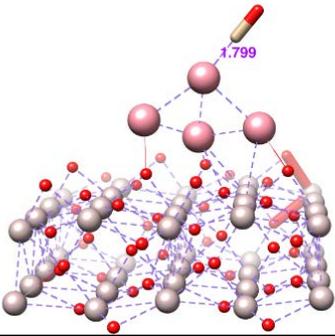
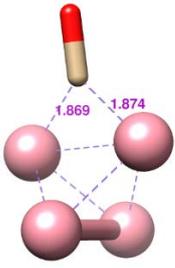
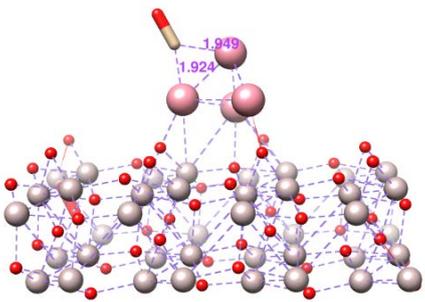
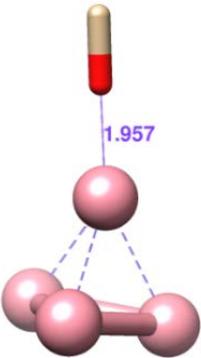
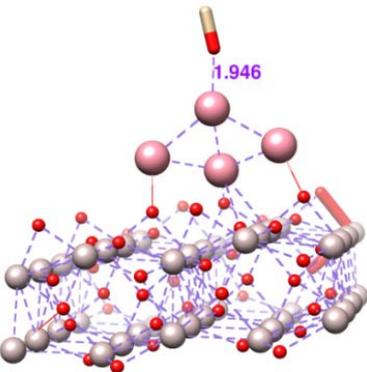
Adsorption of CO on Co_4			
C–Co Configurations			
Structure	E_{ads} , $\text{kJ}\cdot\text{mol}^{-1}$	Structure	E_{ads} , $\text{kJ}\cdot\text{mol}^{-1}$
d-1 configurations			
	-186.9		-231.9
d-2 configurations			
	-201.4		-237.3
O–Co Configurations			
Structure	E_{ads} , $\text{kJ}\cdot\text{mol}^{-1}$	Structure	E_{ads} , $\text{kJ}\cdot\text{mol}^{-1}$
	+21.8		-58.8

Table S4. The calculated adsorption energies (E_{ads}) for the adsorption of H_2 on Co_4 and $\text{Co}_4/\text{Al}_2\text{O}_3$ system.

$$E_{\text{ads}} = E(\text{H}_2\text{-adsorbent}) - E(\text{CO}) - E(\text{adsorbent}), \text{ where adsorbent: } \text{Co}_4 \text{ or } \text{Co}_4/\text{Al}_2\text{O}_3.$$

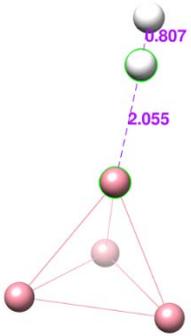
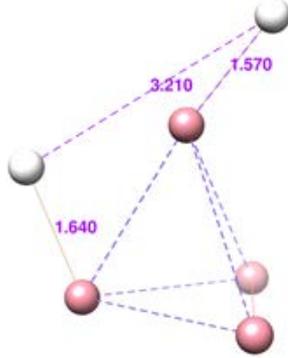
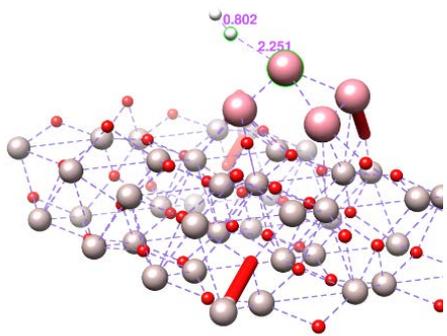
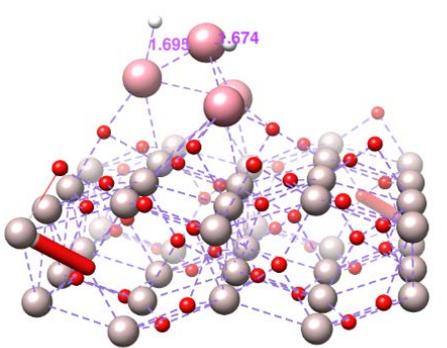
Structure	E_{ads} , $\text{kJ}\cdot\text{mol}^{-1}$	Structure	E_{ads} , $\text{kJ}\cdot\text{mol}^{-1}$
Molecular adsorption		Dissociative chemisorption	
	+32.0		-99.9
	-110.4		-280.7

Table S5. The calculated adsorption energies (E_{ads}) for the adsorption of CO and H_2 on Al_2O_3 system.

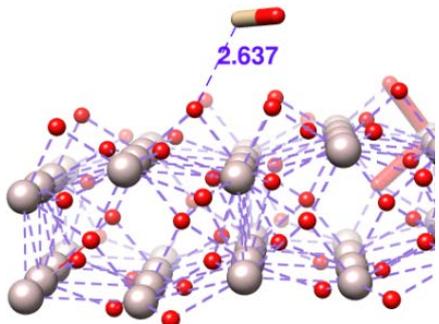
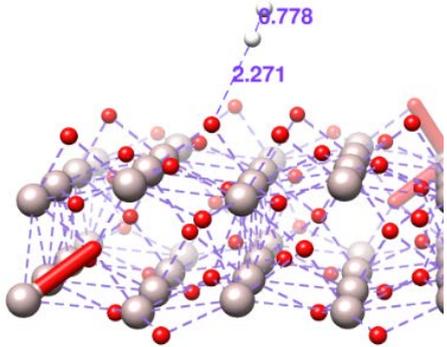
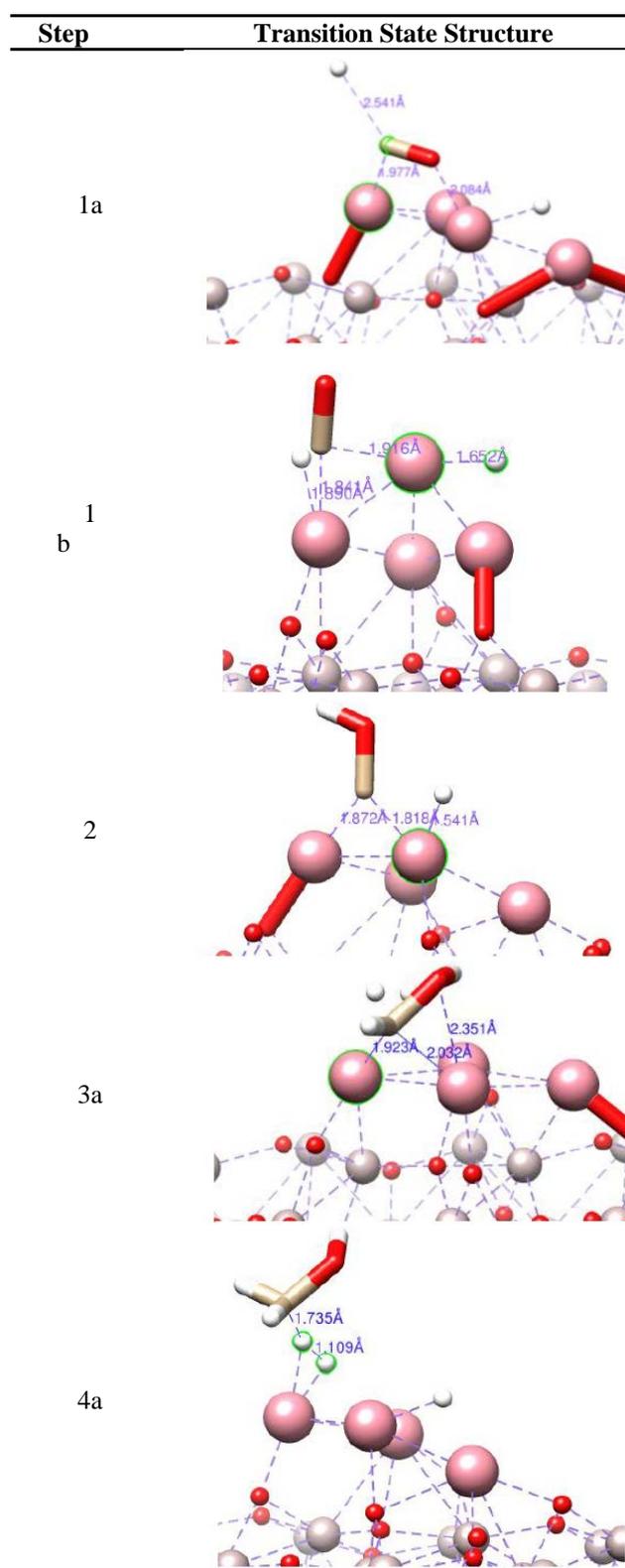
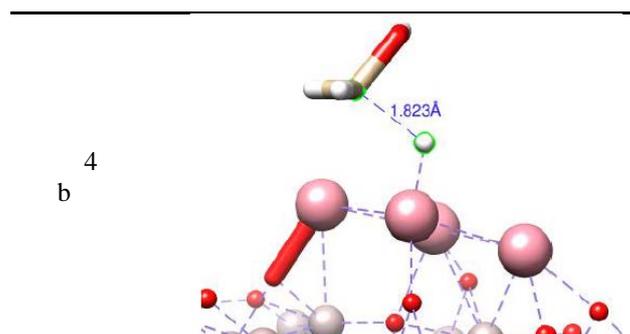
Structure	$E_{\text{ads}}(\text{CO})$, $\text{kJ}\cdot\text{mol}^{-1}$	Structure	$E_{\text{ads}}(\text{H}_2)$, $\text{kJ}\cdot\text{mol}^{-1}$
	-59.8		-32.7

Table S6. Transition state structures of CO hydrogenation on Co₄/Al₂O₃



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