

Structure and Photocatalytic Activity of PdCrO_x Cocatalyst on SrTiO₃ for Overall Water Splitting

Tomoki Kanazawa ^{1,2}, Shunsuke Nozawa ³, Daling Lu ⁴ and Kazuhiko Maeda ^{1,*}

¹ Department of Chemistry, School of Science, Tokyo Institute of Technology, 2-12-1-NE-2 Ookayama, Meguro-ku, Tokyo 152-8550, Japan

² Japan Society for the Promotion of Science, Kojimachi Business Center Building, 5-3-1 Kojimachi, Chiyoda-ku, Tokyo 102-0083

³ Institute of Materials Structure Science, High Energy Accelerator Research Organization, 1-1 Oho, Tsukuba, Ibaraki 305-0801, Japan

⁴ Suzukakedai Materials Analysis Division, Technical Department, Tokyo Institute of Technology 4259 Nagatsuta-cho, Midori-ku, Yokohama 226-8503, Japan

*Correspondence: maedak@chem.titech.ac.jp; Tel.: +81- 3-5734-2239

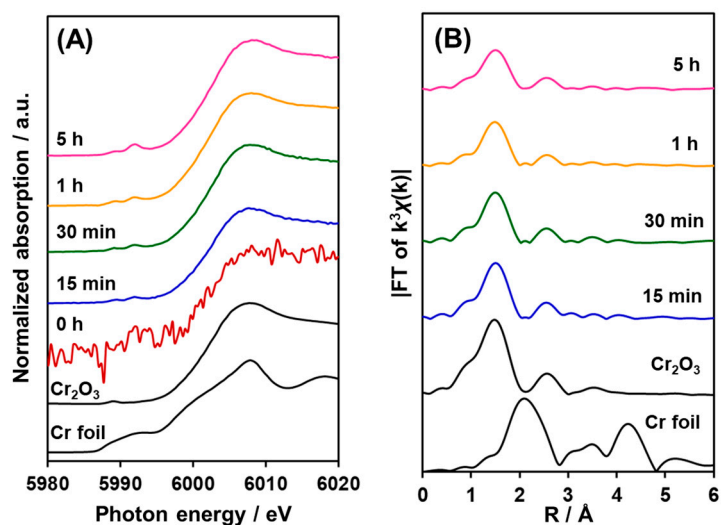


Figure S1. (A) XANES spectra and (B) Fourier transforms of k^3 -weighted Cr K-edge of PdCrO_x nanoparticles (0–5 h) on SrTiO₃. Cr foil and Cr₂O₃ are shown as references.

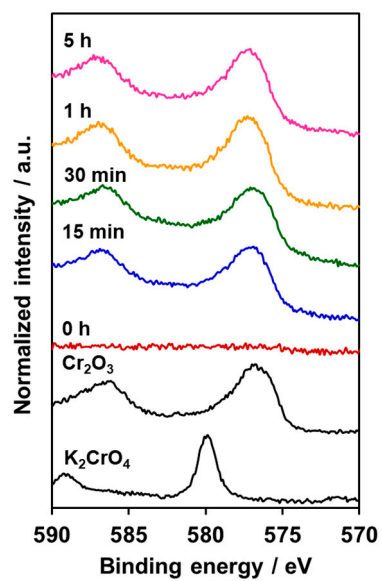


Figure S2. Cr 2p XPS spectra of PdCrO_x/SrTiO₃ prepared for various time periods (0–5h). Data for K₂CrO₄ and Cr₂O₃ are shown as references.

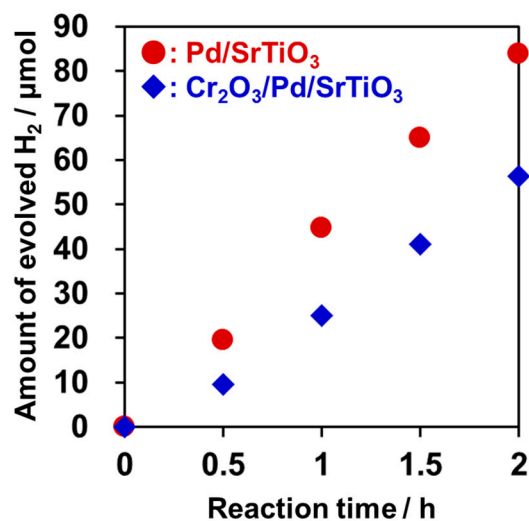


Figure S3. Time course of H₂ evolution over Pd or Cr₂O₃/Pd nanoparticle loaded SrTiO₃. Reaction conditions: catalyst, 100 mg; 10 vol% aqueous methanol solution, 140 mL; metal precursors, Pd 0.5 wt%, Cr 1.0 wt%; light source, a 300 W xenon lamp ($\lambda > 300$ nm).