Supplementary Materials: Isotope Labelling for Reaction Mechanism Analysis in DBD Plasma Processes

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2. Results and Discussion

2.1. Analysis of Plasmas Induced Isotope Exchange Reactions



Figure S1. MS spectra (zone m/z from 13 to 20) taken for the ternary mixture N₂+D₂+NH₃ before (black line) and after (red line) switching on the plasma. The residual intensity at m/z=18 is due to the residual water always present in the MS chambers. After application of plasma there is a change in the relative intensities of m/z peaks at 16, 17, 18 and 19, while the m/z peak at 14 remains constant. These changes are attributed to isotope exchange processes affecting to some of the initially detected NH₃ molecules that become transformed into NH₂D, NHD₂ and ND₃ (see the text).



Figure S2. FTIR spectra recorded for binary N₂+D₂ (black line) and ternary N₂+D₂+NH₃ (red line) mixtures before plasma ignition. This analysis disregards the presence of water in the reactor chamber, since the zone 1300-2000 cm⁻¹ only presents bands for the ternary mixture, which correspond to NH₃.



Figure S3. Evolution of the intensity of the IR absorption bands attributed to different ammonia species during isotope labelling processes induced by DBD plasma.



Figure S4. Scheme of the experimental set-up. A more detailed description of the reactor and the electrical supply can be found in our previous works [1–3]. Outlet gas flow is represented with the point line.



Figure S5. Squared AC curves for the different maximum to minimum voltage difference (3.5, 6.7, 8.7, 8.7 and 7.9 kV) and power (4.4, 20.4, 40.5, 62.0 and 81.3 W) in the case of the NH₃/D₂/N₂ mixture. Squared signals are expected to provide a higher efficiency that sinusoidal ones due to their higher V_{rms} value.

References

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