



Article Photocatalytic Synthesis of 6-Phosphorylated Phenanthridines from 2-Isocyanobiphenyls via Radical C—P and C—C Bond Formation

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Abstract: A mild, efficient, photocatalytic synthesis of 6-phosphorylated phenanthridines via tandem radical addition/cyclization/aromatization of 2-isocyanobiphenyls with diarylphosphine oxides is reported. The method features operational simplicity in metal-free conditions, using low-cost Rose Bengal as a catalyst and sustainable air as a terminal oxidant at room temperature and providing the desired products in moderate to good yields.

Keywords: phenanthridine; 2-isocyanobiphenyls; diarylphosphine oxides; radical cyclization; metal-free

1. Introduction

Phenanthridine is a significant fused *N*-heterocycle, which is ubiquitous in many bioactive alkaloids, pharmaceuticals, natural products, and functional material molecules [1–3]. For example, nitidine chloride (Figure 1) is an active, natural anticancer product with a phenanthridine motif that has been found to inhibit topoisomerase I and topoisomerase II [4]. NK109 (Figure 1) has been found to exhibit antitumor effects against a number of human cancer cell lines [5]. Fagaronine (Figure 1) is an active antileukemic and antimalarial alkaloid [6]. In addition, trispheridine [7], *N*-methylcrinasiadine [8], and lycobetaine [9] are natural alkaloids with excellent bioactivities (Figure 1).



Figure 1. Representative bioactive phenanthridine derivatives.



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). Extensive studies and improvements upon the synthesis of highly functionalized phenanthridine derivatives have been conducted due to their important biological activities. Great progress and developments have been made in the synthetic strategies of phenanthridine derivatives via C-H bond activation, radical addition cyclization, photochemical catalysis, isocyanide chemistry, etc. [10–12]. The transformation proceeded in the absence or presence of various transition-metal catalysts, such as Pd, Mn, Cu, Ru, Ir, Rh, Fe, Ni, and so on. The functionalization mainly focused on the 6-position of phenanthridine [10–12], including trifluoromethylation, difluoromethylation, arylation, alkylation, alkynylation, sulfonylation, phosphorylation, difluoromethylphosphonation, benzylation, trichooromethylation, thiolation, etc.

C-P bond construction has attracted dramatic attention due to the potential applications of organophosphorous compounds in pharmaceutical chemistry and materials science [13,14]. In 2014, several methods for the synthesis of 6-phosphorylated phenanthridines initiated by Ag(I) or Mn(III) salts were reported via radical cascade cyclization (Scheme 1, eq(a–e)) [15–19]. In 2016, Lakhdar and coworkers described the 6-phosphorylation of phenanthridines in the presence of diphenyliodonium salt and triethy-lamine under metal-free conditions (Scheme 1, eq(f)) [20]. Subsequently, several photocatalytic methods were reported in the presence of photocatalyst [Ir(ppy)₂(dtbpy)]PF₆ [21]/2D-COF-1 [22]/4CzIPN-^tBu [23] using K₂S₂O₈ or TBHP as an oxidant (Scheme 1, eq(g–i)).



Conditions:

a) AgOAc (3.0 equiv), DMF, Ar, 100 °C (Studer et al);

b) Mn(OAc)₃ (3.0 equiv), toluene, N₂, 40 °C (Our previous work);

c) Mn(OAc)₃·2H₂O (3.0 equiv), NMP, N₂, 80 °C (Tang et al);

d) AgOAc (0.2 equiv), PhI(OAc)₂ (3.0 equiv), DMF, Ar, 100 °C (Wang, Ji et al);

e) AgNO₃ (2.0 equiv), MeCN, Ar, 60 °C (Yang et al);

f) Ph₂IOTf (1.3 equiv), Et₃N (2.0 equiv), CH₂Cl₂, 40 $^{\circ}$ C (Lakhdar et al);

g) Ir-PC, hv, K₂S₂O₈ (3.0 equiv), CsF (2.0 equiv), DMF, rt (Yan, Lu et al);

h) 2D-COF-1, K₂S₂O₈ (1.5 equiv), CsF (1.0 equiv), EA, N₂, 34 W Blue LED

(Yu, Xu, Yang, et al);

i) 4CzIPN-^fBu (5 mol%), TBHP (2.0 equiv), NaHCO₃ (2.0 equiv), MeCN, N₂, Blue LED (Chen, Yu et al);

j) Rose Bengal (5 mol%), DBU (3.0 equiv), MeCN/H₂O (1.0/0.18,v/v), air, rt, 30 W Blue LED (Metal-free conditions, *this work*).

Scheme 1. Synthesis of 6-phosphorylated phenanthridines from 2-isocyanobiphenyls.

Recently, we have described the synthesis of 6-aroyl [24] and 6-benzylated [25] phenanthridines via the iron-catalyzed cascade radical addition/cyclization of 2-biphenyl isocyanides. In 2019, we reported a visible-light-induced $Mn(acac)_3$ -catalyzed method for the synthesis of 6- β -keto alkyl phenanthridines in the absence of an extended oxidant [26]. In addition, we also described a Mn(III)-catalyzed radical process of 2-isocyanobiphenyls for the synthesis of 6-phosphorylated phenanthridines [16]. Herein, we report another alternative method for the preparation of 6-phosphorylated phenanthridines using low-cost Rose Bengal as a catalyst and sustainable air as a terminal oxidant under metal-free conditions at room temperature.

2. Results and Discussion

Initially, 2-isocyano-5-methyl-1,1'-biphenyl, 1a, and diphenylphosphine oxide, 2a, were chosen as model reaction substrates for the optimization conditions. The desired product, **3aa**, was isolated in 15% yield using Rose Bengal (2 mol%) as a catalyst and DBU as a base in the co-solvent MeCN/ H_2O at room temperature under an air atmosphere (Table 1, entry 1). Increasing the loading of catalyst to 5–10 mol% improved the yields of **3aa** (Table 1, entries 2 and 3). With an increase in the diphenylphosphine oxide (**2a**) amount from 1.5 equiv to 3.5–4.5 equiv, the results showed that the yields of target product **3aa** raised obviously (Table 1, entries 4 and 5). Then, we screened several other organic photocatalysts, and the results showed that Eosin Y could promote the reaction smoothly and provide 3aa in a 61% yield, while only trace amounts of the desired product 3aa were observed when Fluorescein and Rhodamine B were used as photocatalysts (Table 1, entries 6–8). In addition, a series of bases involving Et₃N, DABCO, Na₂CO₃, and K₂CO₃ were surveyed, but no better yield was obtained (Table 1, entries 9–12). Then we attempted to change the amount of base, and the best result (84%) was obtained in the presence of a 3.0 equivalent of DBU (Table 1, entries 13–15). The examination of the reaction medium showed that the effect of the solvent is obvious. The reaction could not offer a better yield in pure acetonitrile or other co-solvents, such as MeOH/H₂O, EtOH/H₂O, and THF/H₂O (Table 1, entries 16–19). The control experiment indicated that the irradiation of blue LED lights is important because no reaction was carried out in the dark (Table 1, entry 20). Unfortunately, the present method was not suitable for the multigram scale. When the reaction carried out on 6 mmol scale, only 15% of the desired product 3aa was isolated (Table 1, entry 14).

Table 1. Optimization of the reaction conditions^{*a*}.

Me Ph +	0 H	cat. / base	Me
L NC	Ph ^{-P} -Ph H	30 W blue LEDs	
1a	2a	solvent, air, rt, 20 h	3aa Ph

Entry	1a/2a	Catalyst	Base	Solvent	Yield (%) ^b
1	1/1.5	Rose Bengal (2 mol%)	DBU	MeCN/H ₂ O (1.0/0.18 mL)	15
2	1/1.5	Rose Bengal (5 mol%)	DBU	MeCN/H ₂ O (1.0/0.18 mL)	36
3	1/1.5	Rose Bengal (10 mol%)	DBU	MeCN/H ₂ O (1.0/0.18 mL)	32
4	1/3.5	Rose Bengal (5 mol%)	DBU	MeCN/H ₂ O (1.0/0.18 mL)	78
5	1/4.5	Rose Bengal (5 mol%)	DBU	MeCN/H ₂ O (1.0/0.18 mL)	77
6	1/3.5	Eosin Y (5 mol%)	DBU	MeCN/H ₂ O (1.0/0.18 mL)	61
7	1/3.5	Fluorescein (5 mol%)	DBU	MeCN/H ₂ O (1.0/0.18 mL)	trace
8	1/3.5	Rhodamine B (5 mol%)	DBU	MeCN/H ₂ O (1.0/0.18 mL)	trace
9	1/3.5	Rose Bengal (5 mol%)	NEt ₃	MeCN/H ₂ O (1.0/0.18 mL)	69
10	1/3.5	Rose Bengal (5 mol%)	DABCO	MeCN/H ₂ O (1.0/0.18 mL)	trace
11	1/3.5	Rose Bengal (5 mol%)	Na ₂ CO ₃	MeCN/H ₂ O (1.0/0.18 mL)	trace
12	1/3.5	Rose Bengal (5 mol%)	K_2CO_3	MeCN/H ₂ O (1.0/0.18 mL)	trace
13	1/3.5	Rose Bengal (5 mol%)	DBU ^c	MeCN/H ₂ O (1.0/0.18 mL)	38
14	1/3.5	Rose Bengal (5 mol%)	DBU d	MeCN/H2O (1.0/0.18 mL)	84 (15) ^g
15	1/3.5	Rose Bengal (5 mol%)	DBU ^e	MeCN/H2O (1.0/0.18 mL)	80
16	1/3.5	Rose Bengal (5 mol%)	DBU d	MeCN	50
17	1/3.5	Rose Bengal (5 mol%)	DBU d	EtOH/H ₂ O (1.0/0.18 mL)	43
18	1/3.5	Rose Bengal (5 mol%)	DBU d	$MeOH/H_2O$ (1.0/0.18 mL)	56
19	1/3.5	Rose Bengal (5 mol%)	DBU d	$THF/H_2O(1.0/0.18 \text{ mL})$	61
20 ^f	1/3.5	Rose Bengal (5 mol%)	DBU	$THF/H_2O(1.0/0.18 \text{ mL})$	NR

^{*a*} Reaction conditions: **1a** (0.2 mmol), base (2.0 equiv) and reaction were irradiated by 30 W blue LEDs at room temperature, stirring for 20 h under air atmosphere. ^{*b*} Isolated yields of **3aa**. ^{*c*} DBU (1.0 equiv). ^{*d*} DBU (3.0 equiv). ^{*e*} DBU (4.0 equiv). ^{*f*} The reaction was carried out in the dark. ^{*g*} The reaction was proceeded on 6 mmol scale.

We next examined the generality and the substrate scope of 2-isocyanobiphenyls under the optimized reaction conditions (Scheme 2). Initially, the electronic effect of the substituents R^2 was investigated. The reaction could tolerate both electron-donating groups and electron-withdrawing groups, such as methyl, methoxy, halogen (F, Cl), and trifluoromethyl on the ortho-, meta-, or para-position of the phenyl ring, providing the corresponding products in moderate to good yields. For instance, ortho-methyl-substituted substrate 1b gave the target product 3ba in 74% yields. Meta-methyl-substituted substrate 1c provided two isomers, 3ca and 3ca', in 60% total yields in a ratio of about 1:1. Paratrifluoromethyl-substituted substrate 1i provided the product 3ia in a 53% yield under standard conditions. Unfortunately, when substrate **1n** was used, bearing a nitro group, only a trace amount of desired product was observed under the same conditions. Then, 2-(2-isocyanophenyl)naphthalene was applied to the reaction with **2a**, and the product **3pa** was obtained in a low yield (38%). The electronic effect of substituent \mathbb{R}^1 was significant, and the substrates with electron-withdrawing groups of halogens (F, Cl) provided the corresponding products 3ra-3ta in moderate yields of 37-46%. The reaction did not work when substate 1u bore a strong electron-withdrawing group (-CF₃). In addition, ortho-methyl (R¹)-substituted 2-isocyanobiphenyl **1c** also did not work, maybe due to the steric hindrance.



Scheme 2. Substrate scope of 2-isocyanobiphenyls.

In addition, the substrate scope of diphenylphosphine oxides (2) were also examined under the optimal conditions (Scheme 3). The results shown that the electronic effect was not obvious; diaryl-substituted phosphine oxides (2) with both electron-donating groups and electron-withdrawing groups were proceeded smoothly, providing the corresponding phosphorylation phenanthridines in good yields. Product **3ad** was isolated only in moderate yield (46%) due to the steric effect. However, no target product was observed in the case of diisopropyl phosphine oxide **2i**.



Scheme 3. Substrate scope of diphenylphosphine oxides.

A control experiment was executed to investigate the possible reaction mechanism (Scheme 4). The reaction was totally inhibited in the presence of the radical scavenger of 2,2,6,6-tetramethyl-1-piperidinyloxy (TEMPO, 3.5 equiv) under standard reaction conditions. The results indicated that the transformation probably underwent cascade radical addition/cyclization processes [10–12,24–26].



Scheme 4. Control experiment in the presence of TEMPO.

3. Materials and Methods

3.1. General Information

Nuclear magnetic resonance (NMR) spectra were recorded in parts per million from internal tetramethylsilane on the δ scale. ¹H and ¹³C NMR spectra were recorded on a Bruker AV-400 spectrometer operating at 400 MHz and 100 MHz (Bruker, Ettlingen, Germany), respectively. All chemical shift values are quoted in ppm and coupling constants are quoted in Hz. High-resolution mass spectrometry (HRMS) spectra were obtained on a micrOTOF II instrument (Bruker Daltonik GmbH, Bremen, Germany). The characterization (¹H- and ¹³C-NMR and ³¹P-NMR) for products **3** are provided (see Support Information).

3.2. General Procedure for the Photocatalytic Synthesis of 6-Phosphorylated Phenanthridines from 2-Isocyanobiphenyls

Added to a 25 mL quartz test tube containing a magnetic stir bar were 2-biphenyl isocyanides (1) (0.2 mmol), Rose Bengal (0.01 mmol, 5 mol%), DBU (3.0 equiv) under air, diphenylphosphine oxides (2) (0.6 mmol, 3.0 equiv), and MeCN/H₂O (1.0/0.18 mL). The resulting mixture was stirred at room temperature under 30 W blue LED irradiation for 5 h. After completion, monitored by TLC, evaporation of the solvent under reduced pressure followed purification by silica gel chromatography using ethyl acetate–petroleum ether (1:3) as an eluent to provide the desired products (3).

3.3. Characterization Data for Products 3

(2-Methylphenanthridin-6-yl)diphenylphosphine oxide [16,18] (**3aa**): Isolated ($R_f = 0.6$, EtOAc—petroleum ether = 1:3) as a white solid (66.1 mg, 84% yield), mp: 222–223 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.48 (d, *J* = 8.4 Hz, 1H), 8.57 (d, *J* = 8.4 Hz, 1H), 8.31 (s, 1H), 8.05–7.84 (m, 5H), 7.76 (t, *J* = 7.6 Hz, 1H), 7.63 (t, *J* = 7.6 Hz, 1H), 7.52–7.37 (m, 7H), 2.57 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 155.6 (d, *J*_{C-P} = 128.7 Hz), 141.3 (d, *J*_{C-P} = 23.3 Hz), 139.1, 133.1 (d, *J*_{C-P} = 104.0 Hz), 132.4, 132.3 (d, *J*_{C-P} = 9.1 Hz), 131.6 (d, *J*_{C-P} = 2.5 Hz), 130.9, 130.8, 130.5, 128.5, 128.2 (d, *J*_{C-P} = 12.1 Hz), 127.9, 127.7, 124.2 (d, *J*_{C-P} = 2.4 Hz), 122.1, 121.7, 22.2. ³¹P NMR (162 MHz, CDCl₃) δ 28.2. HRMS (ESI): *m*/*z* [M + H]⁺ calcd for C₂₆H₂₁NOP: 394.1355, found: 394.1358.

(2-Methylphenanthridin-6-yl)di-p-tolylphosphine oxide (**3ab**): Isolated ($R_f = 0.4$, EtOAc—petroleum ether = 1:3) as a white solid (64.1 mg, 76% yield), mp: 249–251 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.45 (d, *J* = 8.0 Hz, 1H), 8.63 (d, *J* = 8.4 Hz, 1H), 8.37 (s, 1H), 7.95 (d, *J* = 8.4 Hz, 1H), 7.87–7.73 (m, 5H), 7.65 (t, *J* = 7.6 Hz, 1H), 7.52 (dd, *J* = 8.4, 0.8 Hz, 1H), 7.26–7.21 (m, 4H), 2.63 (s, 3H), 2.37 (s, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 156.0 (d, *J*_{C-P} = 128.4 Hz), 141.9 (d, *J*_{C-P} = 2.7 Hz), 141.2 (d, *J*_{C-P} = 23.4 Hz), 138.9, 132.3 (d, *J*_{C-P} = 9.5 Hz), 130.9, 130.4, 130.0 (d, *J*_{C-P} = 106.4 Hz), 128.9 (d, *J*_{C-P} = 12.5 Hz), 128.6, 127.9 (d, *J*_{C-P} = 23.0 Hz), 127.7, 124.2, 122.0, 121.6, 22.2, 21.6. ³¹P NMR (162 MHz, CDCl₃) δ 28.8. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₈H₂₅NOP: 422.1668, found: 422.1670.

(2-Methylphenanthridin-6-yl)di-m-tolylphosphine oxide (**3a**c): Isolated ($R_f = 0.4$, EtOAc—petroleum ether = 1:3) as a white solid (56.5 mg, 67% yield), mp: 237–239 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.38 (d, *J* = 8.4 Hz, 1H), 8.50 (d, *J* = 8.4 Hz, 1H), 8.24 (s, 1H), 7.85 (d, *J* = 8.4 Hz, 1H), 7.76–7.64 (m, 3H), 7.62–7.51 (m, 3H), 7.41 (d, *J* = 8.3 Hz, 1H), 7.23–7.17 (m, 4H), 2.50 (s, 3H), 2.25 (s, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 155.8 (d, *J*_{C-P} = 128.1 Hz), 141.3 (d, *J*_{C-P} = 23.2 Hz), 139.0, 138.0 (d, *J*_{C-P} = 11.9 Hz), 133.0 (d, *J*_{C-P} = 104.4 Hz), 132.6 (d, *J*_{C-P} = 8.9 Hz), 132.4 (d, *J*_{C-P} = 2.7 Hz), 132.3 (d, *J*_{C-P} = 6.8 Hz), 130.9, 130.7, 130.4, 129.5 (d, *J*_{C-P} = 9.3 Hz), 128.6, 128.1, 128.0 (d, *J*_{C-P} = 12.8 Hz), 127.7, 124.2 (d, *J*_{C-P} = 2.4 Hz), 122.1, 121.7, 22.2, 21.5. ³¹P NMR (162 MHz, CDCl₃) δ 28.6. HRMS (ESI): *m*/*z* [M + H]⁺ calcd for C₂₈H₂₅NOP: 422.1668, found: 422.1671.

(2-Methylphenanthridin-6-yl)di-o-tolylphosphine oxide (**3ad**): Isolated ($R_f = 0.4$, EtOAc—petroleum ether = 1:3) as a white solid (38.8 mg, 46% yield), mp: 228–230 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.11 (d, *J* = 8.4 Hz, 1H), 8.58 (d, *J* = 8.4 Hz, 1H), 8.31 (s, 1H), 7.73 (d, *J* = 8.0 Hz, 2H), 7.54 (t, *J* = 7.6 Hz, 1H), 7.42 (d, *J* = 8.0 Hz, 1H), 7.36–7.26 (m, 4H), 7.24–7.18 (m, 2H), 7.07 (t, *J* = 7.2 Hz, 2H), 2.55 (s, 3H), 2.36 (s, 6H). ¹³C NMR (100 MHz,

CDCl₃) δ 156.1 (d, J_{C-P} = 128.3 Hz), 143.4 (d, J_{C-P} = 7.8 Hz), 141.3 (d, J_{C-P} = 23.5 Hz), 139.1, 133.1 (d, J_{C-P} = 11.9 Hz), 132.4 (d, J_{C-P} = 6.5 Hz), 131.8 (d, J_{C-P} = 2.5 Hz), 131.6, 131.5, 131.1, 130.7, 130.5, 130.4, 128.8, 127.7, 127.6 (d, J_{C-P} = 23.0 Hz), 125.3 (d, J_{C-P} = 12.8 Hz), 124.3, 122.2, 121.6, 22.2, 22.0, 22.0. ³¹P NMR (162 MHz, CDCl₃) δ 38.2. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₈H₂₅NOP: 422.1668, found: 422.1673.

Bis(3,5-dimethylphenyl)(2-methylphenanthridin-6-yl)phosphine oxide (**3ae**): Isolated ($R_f = 0.5$, EtOAc—petroleum ether = 1:3) as a white solid (54.8 mg, 61% yield). mp: 265–268 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.46 (d, J = 8.4 Hz, 1H), 8.62 (d, J = 8.4 Hz, 1H), 8.36 (s, 1H), 7.97 (d, J = 8.4 Hz, 1H), 7.80 (t, J = 7.4 Hz, 1H), 7.65 (t, J = 7.4 Hz, 1H), 7.55–7.50 (m, 5H), 7.12 (s, 2H), 2.62 (s, 3H), 2.30 (s, 12H). ¹³C NMR (100 MHz, CDCl₃) δ 156.0 (d, $J_{C-P} = 127.6$ Hz), 141.2 (d, $J_{C-P} = 23.2$ Hz), 138.9, 137.7 (d, $J_{C-P} = 12.6$ Hz), 133.4 (d, $J_{C-P} = 2.8$ Hz), 132.9 (d, $J_{C-P} = 104.5$ Hz), 132.3 (d, $J_{C-P} = 6.8$ Hz), 131.0, 130.5 (d, $J_{C-P} = 30.6$ Hz), 129.9 (d, $J_{C-P} = 9.2$ Hz), 128.2 (d, $J_{C-P} = 100.6$ Hz), 128.0 (d, $J_{C-P} = 23.0$ Hz), 124.2 (d, $J_{C-P} = 2.4$ Hz), 122.0, 121.6, 22.2, 21.4. ³¹P NMR (162 MHz, CDCl₃) δ 29.0. HRMS (ESI): m/z [M + H]⁺ calcd for C₃₀H₂₉NOP: 450.1981, found: 450.1986.

Bis(3-methoxyphenyl)(2-methylphenanthridin-6-yl)phosphine oxide (**3a**f): Isolated ($R_f = 0.6$, EtOAc—petroleum ether = 1:3) as a yellow solid (60.7 mg, 67% yield), mp: 241–243 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.33 (d, J = 8.0 Hz, 1H), 8.51 (d, J = 8.2 Hz, 1H), 8.25 (s, 1H), 7.85 (d, J = 8.4 Hz, 1H), 7.69 (t, J = 7.6 Hz, 1H), 7.55 (t, J = 7.6 Hz, 1H), 7.49–7.40 (m, 3H), 7.36 (d, J = 7.6 Hz, 1H), 7.33 (d, J = 8.0 Hz, 1H), 7.27–7.18 (m, 2H), 6.93 (d, J = 7.6 Hz, 2H), 3.68 (s, 6H), 2.51 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 159.3 (d, $J_{C-P} = 15.0$ Hz), 155.4 (d, $J_{C-P} = 129.3$ Hz), 141.2 (d, $J_{C-P} = 23.5$ Hz), 139.1, 134.2 (d, $J_{C-P} = 103.4$ Hz), 132.3 (d, $J_{C-P} = 6.8$ Hz), 130.9, 130.8, 130.5, 129.3 (d, $J_{C-P} = 14.3$ Hz), 128.4, 127.9 (d, $J_{C-P} = 23.5$ Hz), 127.7, 124.7 (d, $J_{C-P} = 9.2$ Hz), 124.3, 122.1, 121.7, 118.0 (d, $J_{C-P} = 2.5$ Hz), 117.0 (d, $J_{C-P} = 10.1$ Hz), 55.4, 22.2. ³¹P NMR (162 MHz, CDCl₃) δ 28.2. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₈H₂₅NO₃P: 454.1567, found: 454.1568.

Bis(3-fluorophenyl)(2-methylphenanthridin-6-yl)phosphine oxide (**3ag**): Isolated ($R_f = 0.5$, EtOAc—petroleum ether = 1:3) as a yellow solid (58.4 mg, 68% yield), mp: 257–259 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.44 (d, J = 8.4 Hz, 1H), 8.67 (d, J = 8.4 Hz, 1H), 8.40 (s, 1H), 7.98 (d, J = 8.4 Hz, 1H), 7.86 (t, J = 7.6 Hz, 1H), 7.77–7.62 (m, 5H), 7.58 (d, J = 8.0 Hz, 1H), 7.48–7.40 (m, 2H), 7.22 (td, J = 8.4, 2.0 Hz, 2H), 2.65 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 161.4 (dd, $J_{C-F} = 247.9$, $J_{C-P} = 17.0$ Hz Hz), 160.0, 153.4 (d, $J_{C-P} = 131.6$ Hz), 140.1 (d, $J_{C-F} = 23.9$ Hz), 138.5, 134.4 (dd, $J_{C-P} = 103.5$, $J_{C-F} = 5.6$ Hz Hz), 131.4 (d, $J_{C-P} = 7.1$ Hz), 130.0, 129.8, 129.7, 129.2 (d, $J_{C-F} = 7.3$ Hz), 129.1 (d, $J_{C-F} = 7.3$ Hz), 128.3, 128.1, 127.1, 127.0 (d, $J_{C-P} = 3.0$ Hz), 126.9, 126.8, 123.3, 121.2, 120.7, 118.3 (d, $J_{C-F} = 9.9$ Hz), 118.1, 118.0 (d, $J_{C-F} = 9.1$ Hz), 117. 9 (d, $J_{C-P} = 2.6$ Hz), 21.2. ³¹P NMR (162 MHz, CDCl₃) δ 24.9. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₁₉F₂NOP: 430.1167, found: 430.1169.

(2-Methylphenanthridin-6-yl)di(naphthalen-1-yl)phosphine oxide (**3ah**): Isolated ($R_f = 0.5$, EtOAc—petroleum ether = 1:3) as a yellow solid (66.1 mg, 67% yield), mp: 293–296°C. ¹H NMR (400 MHz, CDCl₃) δ 9.41 (dd, J = 8.2, 2.0 Hz, 1H), 8.49–8.44 (m, 3H), 8.20 (s, 1H), 7.88–7.79 (m, 3H), 7.75–7.72 (m, 4H), 7.69 (d, J = 8.0 Hz, 2H), 7.64 (t, J = 7.6 Hz, 1H), 7.51 (t, J = 7.6 Hz, 1H), 7.41–7.31 (m, 5H), 2.45 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 155.7 (d, $J_{C-P} = 129.5$ Hz), 141.3 (d, $J_{C-P} = 23.4$ Hz), 139.2, 134.7 (d, $J_{C-P} = 2.3$ Hz), 134.0 (d, $J_{C-P} = 8.9$ Hz), 132.5 (d, $J_{C-P} = 7.3$ Hz), 132.4 (d, $J_{C-P} = 6.9$ Hz), 130.9, 130.8, 130.5, 130.4 (d, $J_{C-P} = 103.8$ Hz), 129.1, 128.5, 128.1, 127.9, 127.8 (2C), 127.7, 127.6, 126.7, 124.3, 122.2, 121.7, 22.2. ³¹P NMR (162 MHz, CDCl₃) δ 28.9. HRMS (ESI): m/z [M + H]⁺ calcd for C₃₄H₂₅NOP: 494.1668, found: 494.1672.

(2,10-Dimethylphenanthridin-6-yl)diphenylphosphine oxide (**3ba**): Isolated ($R_f = 0.5$, EtOAc—petroleum ether = 1:3) as a white solid (60.3 mg, 74% yield), mp: 222–225 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.32 (d, *J* = 8.0 Hz, 1H), 8.52 (s, 1H), 7.89–7.73 (m, 5H), 7.55 (d, *J* = 7.2 Hz, 1H), 7.46 (t, *J* = 8.0 Hz, 1H), 7.43–7.37 (m, 3H), 7.37–7.30 (m, 4H), 3.01 (s, 3H), 2.52 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 156.1 (d, *J*_{C-P} = 129.0 Hz), 142.3 (d, *J*_{C-P} = 23.8 Hz), 138.0, 135.3, 135.0, 133.2 (d, *J*_{C-P} = 104.3 Hz), 132.3 (d, *J*_{C-P} = 9.1 Hz), 131.9 (d, *J*_{C-P} = 6.7 Hz),

131.6 (d, $J_{C-P} = 2.5$ Hz), 131.2, 129.5, 129.3 (d, $J_{C-P} = 23.4$ Hz), 128.1 (d, $J_{C-P} = 12.1$ Hz), 127.2, 127.0, 126.4, 125.7 (d, $J_{C-P} = 2.4$ Hz), 27.1, 22.5. ³¹P NMR (162 MHz, CDCl₃) δ 29.4. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₇H₂₃NOP: 408.1512, found: 408.1515.

(2,8-Dimethylphenanthridin-6-yl)diphenylphosphine oxide (**3da**): Isolated ($R_f = 0.5$, EtOAc—petroleum ether = 1:3) as a white solid (48.1 mg, 64% yield), mp: 231–233 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.28 (s, 1H), 8.48 (dd, *J* = 8.4, 1.2 Hz, 1H), 8.29 (s, 1H), 7.97–7.87 (m, 5H), 7.61 (dd, *J* = 8.4, 1.2 Hz, 1H), 7.53–7.36 (m, 7H), 2.58 (s, 3H), 2.53 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 155.0 (d, *J*_{C-P} = 129.0 Hz), 140.9 (d, *J*_{C-P} = 23.4 Hz), 139.0, 137.9, 133.6 (d, *J*_{C-P} = 104.0 Hz), 132.7, 132.4 (d, *J*_{C-P} = 9.1 Hz), 131.5 (d, *J*_{C-P} = 2.6 Hz), 130.8, 130.2 (d, *J*_{C-P} = 6.9 Hz), 130.0, 128.4, 128.1 (d, *J*_{C-P} = 12.0 Hz), 127.6, 124.3 (d, *J*_{C-P} = 2.5 Hz), 122.0, 121.5, 22.2, 21.9. ³¹P NMR (162 MHz, CDCl₃) δ 28.0. HRMS (ESI): *m*/*z* [M + H]⁺ calcd for C₂₇H₂₃NOP: 408.1512, found: 408.1516.

Diphenyl(2,7,9-trimethylphenanthridin-6-yl)phosphine oxide (**3ea**): Isolated ($R_f = 0.5$, EtOAc—petroleum ether = 1:3) as a white solid (62.3 mg, 74% yield), mp: 255–257 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.21 (s, 2H), 7.72–7.65 (m, 4H), 7.51 (d, *J* = 8.0 Hz, 1H), 7.40–7.35 (m, 2H), 7.34–7.27 (m, 5H), 7.17 (s, 1H), 2.84 (s, 3H), 2.48 (s, 3H), 2.43 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 155.4 (d, *J*_{C-P} = 129.9 Hz), 140.5, 139.8 (d, *J*_{C-P} = 24.1 Hz), 138.9, 137.4, 135.1 (d, *J*_{C-P} = 106.9 Hz), 134.2 (d, *J*_{C-P} = 6.7 Hz), 133.3, 132.0 (d, *J*_{C-P} = 8.9 Hz), 131.1 (d, *J*_{C-P} = 2.5 Hz), 130.3, 130.2, 128.0 (d, *J*_{C-P} = 12.1 Hz), 125.7 (d, *J*_{C-P} = 23.7 Hz), 124.1 (d, *J*_{C-P} = 2.6 Hz), 121.8, 120.0, 25.0, 22.2, 21.9. ³¹P NMR (162 MHz, CDCl₃) δ 36.2. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₈H₂₅NOP: 422.1668, found: 422.1672.

(2-Methyl-[1,3]dioxolo[4,5-j]phenanthridin-6-yl)diphenylphosphine oxide (**3fa**): Isolated ($R_f = 0.4$, EtOAc—petroleum ether = 1:1) as a white solid (53.3 mg, 63% yield), mp: 262–265 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.98 (s, 1H), 8.11 (s, 1H), 7.98–7.84 (m, 6H), 7.54–7.38 (m, 7H), 6.08 (s, 2H), 2.58 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 153.4 (d, $J_{C-P} = 131.3$ Hz), 151.1, 148.2, 141.0 (d, $J_{C-P} = 23.2$ Hz), 138.5, 133.8, 132.2 (d, $J_{C-P} = 103.9$ Hz), 132.3 (d, $J_{C-P} = 9.1$ Hz), 131.6, 130.7 130.5 (d, $J_{C-P} = 7.4$ Hz), 129.9, 128.1 (d, $J_{C-P} = 12.0$ Hz), 125.2 (d, $J_{C-P} = 23.4$ Hz), 124.4, 121.4, 105.6, 102.0, 99.8, 22.1. ³¹P NMR (162 MHz, CDCl₃) δ 28.1. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₇H₂₁NO₃P: 4038.1254, found: 438.1257.

(9-Chloro-2-methylphenanthridin-6-yl)diphenylphosphine oxide (**3ga**): Isolated ($R_f = 0.4$, EtOAc—petroleum ether = 1:3) as a white solid (48.8 mg, 57% yield), mp: 211–214 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.50 (d, *J* = 8.8 Hz, 1H), 8.57 (s, 1H), 8.27 (s, 1H), 7.97–7.86 (m, 5H), 7.60 (dd, *J* = 8.8, 1.6 Hz, 1H), 7.58–7.48 (m, 3H), 7.47–7.41 (m, 4H), 2.62 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 155.3 (d, *J*_{C-P} = 130.1 Hz), 141.5 (d, *J*_{C-P} = 22.9 Hz), 139.5, 137.4, 133.8 (d, *J*_{C-P} = 6.7 Hz), 132.8 (d, *J*_{C-P} = 104.4 Hz), 132.3 (d, *J*_{C-P} = 9.2 Hz), 131.8, 131.2, 130.9, 130.2, 128.4, 128.2 (d, *J*_{C-P} = 12.0 Hz), 126.3 (d, *J*_{C-P} = 23.4 Hz), 123.2, 121.7 (d, *J*_{C-P} = 12.1 Hz), 21.2. ³¹P NMR (162 MHz, CDCl₃) δ 27.9. HRMS (ESI): *m*/*z* [M + H]⁺ calcd for C₂₆H₂₀ClNOP: 428.0966, found: 428.0972.

(8-Chloro-2-methylphenanthridin-6-yl)diphenylphosphine oxide (**3ha**): Isolated ($R_f = 0.5$, EtOAc—petroleum ether = 1:3) as a white solid (56.5 mg, 66% yield), mp: 229–231 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.62 (d, J = 2.0 Hz, 1H), 8.46 (dd, J = 8.8, 1.2 Hz, 1H), 8.23 (s, 1H), 8.06–7.90 (m, 5H), 7.69 (dd, J = 8.8, 2.0 Hz, 1H), 7.57–7.47 (m, 3H), 7.46–7.39 (m, 4H), 2.58 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 154.5 (d, $J_{C-P} = 128.6$ Hz), 141.1 (d, $J_{C-P} = 22.7$ Hz), 139.7, 133.8, 132.9 (d, $J_{C-P} = 104.6$ Hz), 132.3 (d, $J_{C-P} = 9.1$ Hz), 131.8 (d, $J_{C-P} = 2.6$ Hz), 131.5, 130.9, 130.8, 130.7 (d, $J_{C-P} = 6.8$ Hz), 128.8 (d, $J_{C-P} = 23.1$ Hz), 128.2 (d, $J_{C-P} = 12.1$ Hz), 127.5, 123.7, 123.6 (d, $J_{C-P} = 2.4$ Hz), 121.5, 22.2. ³¹P NMR (162 MHz, CDCl₃) δ 27.2. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₂₀ClNOP: 428.0966, found: 428.0969.

(2-Methyl-8-(trifluoromethyl)phenanthridin-6-yl)diphenylphosphine oxide (**3ia**): Isolated ($R_f = 0.6$, EtOAc—petroleum ether = 1:3) as a yellow solid (48.9 mg, 53% yield), mp: 199–201 °C. ¹H NMR (400 MHz, CDCl₃) δ 10.00 (s, 1H), 8.70 (d, *J* = 8.8 Hz, 1H), 8.35 (s, 1H), 8.05–7.95 (m, 6H), 7.60 (d, *J* = 8.4 Hz, 1H), 7.55–7.48 (m, 2H), 7.47–7.41 (m, 4H), 2.63 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 155.8 (d, *J*_{C-P} = 127.7 Hz), 141.8 (d, *J*_{C-P} = 12.5 Hz), 139.9, 134.4 (d, *J*_{C-P} = 6.2 Hz), 132.8 (d, *J*_{C-P} = 105.4 Hz), 132.3 (d, *J*_{C-P} = 9.1 Hz), 131.8 (d, *J*_{C-P} = 2.6 Hz), 131.7, 131.0, 129.4 (q, *J*_{C-F} = 32.6 Hz), 128.3 (d, *J*_{C-P} = 12.1 Hz), 127.3 (d, *J*_{C-P} = 22.9 Hz), 126.6

(q, ${}^{1}J_{C-F} = 3.1$ Hz), 126.2 (d, $J_{C-P} = 4.3$ Hz), 123.9 (q, $J_{C-F} = 270.8$ Hz), 123.2 (d, $J_{C-P} = 2.2$ Hz), 123.1, 122.0, 22.2. 31 P NMR (162 MHz, CDCl₃) δ 26.8. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₇H₂₀F₃NOP: 462.1229, found: 462.1232.

(10-Fluoro-2-methylphenanthridin-6-yl)diphenylphosphine oxide (**3ja**): Isolated ($R_f = 0.3$, EtOAc—petroleum ether = 1:3) as a white solid (55.1 mg, 67% yield), mp: 192–194 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.35 (d, J = 8.2 Hz, 1H), 8.78 (s, 1H), 7.96–7.86 (m, 5H), 7.65–7.58 (m, 1H), 7.57–7.48 (m, 4H), 7.47–7.42 (m, 4H), 2.61 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 160.5 (d, ¹ $J_{C-F} = 253.5$ Hz), 155.0 (d, $J_{C-P} = 129.1$ Hz), 141.5 (d, $J_{C-P} = 23.0$ Hz), 139.8, 132.8 (d, $J_{C-P} = 104.8$ Hz), 132.3 (d, $J_{C-P} = 9.1$ Hz), 131.7 (d, $J_{C-P} = 2.5$ Hz), 130.7 (d, $J_{C-F} = 17.7$ Hz), 129. 9 (d, $J_{C-F} = 24.3$), 128.2 (d, $J_{C-P} = 12.1$ Hz), 128.1 (d, $J_{C-P} = 8.9$ Hz), 126.6 (d, ² $J_{C-F} = 23.0$ Hz), 124.6 (d, $J_{C-P} = 4.0$ Hz), 121.8 (m), 117.3 (d, ² $J_{C-F} = 23.2$ Hz), 22.4. ³¹P NMR (162 MHz, CDCl₃) δ 28.6. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₂₀FNOP: 412.1261, found: 412.1267.

(9-Fluoro-2-methylphenanthridin-6-yl)diphenylphosphine oxide (**3ka**): Isolated ($R_f = 0.4$, EtOAc—petroleum ether = 1:3) as a white solid (32.1 mg, 39% yield), mp: 205–208 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.60 (dd, J = 9.2, 6.0 Hz, 1H), 8.23 (s, 1H), 8.21 (d, J = 9.2 Hz, 1H), 7.98–7.89 (m, 5H), 7.57–7.49 (m, 3H), 7.47–7.36 (m, 5H), 2.62 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 163.8 (d, ¹ $J_{C-F} = 251.6$ Hz), 155.1 (d, ² $J_{C-P} = 128.9$ Hz), 141.3 (d, ² $J_{C-F} = 23.2$ Hz), 139.2, 135.0 (d, $J_{C-P} = 9.3$ Hz), 134.9 (d, $J_{C-P} = 9.3$ Hz), 132.9 (d, $J_{C-P} = 104.4$ Hz), 132.4, 132.3, 131.7 (d, $J_{C-P} = 2.7$ Hz), 131.6, 131.2, 130.9, 128.2 (d, $J_{C-P} = 12.1$ Hz), 125.1 (d, ² $J_{C-P} = 24.3$ Hz), 123.8, 121.8, 116.9 (d, $J_{C-F} = 23.3$ Hz), 107.2 (d, $J_{C-F} = 22.2$ Hz), 22.1.³¹P NMR (162 MHz, CDCl₃) δ 27.9. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₂₀FNOP: 412.1261, found: 412.1265.

(8-Fluoro-2-methylphenanthridin-6-yl)diphenylphosphine oxide (**3la**): Isolated ($R_f = 0.5$, EtOAc—petroleum ether = 1:3) as a yellow solid (50.2 mg, 61% yield), mp: 216–219 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.29 (dd, J = 10.2, 2.6 Hz, 1H), 8.61–8.54 (m, 1H), 8.27 (s, 1H), 7.98–7.90 (m, 5H), 7.56–7.47 (m, 4H), 7.46–7.40 (m, 4H), 2.60 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 161.2 (d, ¹ $J_{C-F} = 247.4$ Hz), 154.6 (d, $J_{C-P} = 129.2$ Hz), 140.9 (d, ² $J_{C-F} = 23.5$ Hz), 132.8 (d, $J_{C-P} = 104.3$ Hz), 132.3 (d, $J_{C-P} = 9.1$ Hz), 131.7 (d, $J_{C-P} = 2.5$ Hz), 130.9, 130.4, 129.3 (d, ³ $J_{C-F} = 9.3$ Hz), 129.0 (d, ³ $J_{C-F} = 8.8$ Hz), 128.2 (d, $J_{C-P} = 12.1$ Hz), 124.6 (d, $J_{C-P} = 8.5$ Hz), 123.9, 121.4 120.3 (d, $J_{C-F} = 24.2$ Hz), 113.1 (d, ² $J_{C-F} = 23.1$ Hz), 22.2. ³¹P NMR (162 MHz, CDCl₃) δ 27.4. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₂₀FNOP: 412.1261, found: 412.1263.

(7,9-Difluoro-2-methylphenanthridin-6-yl)diphenylphosphine oxide (**3ma**): Isolated ($R_f = 0.5$, EtOAc—petroleum ether = 1:1) as a white solid (42.9 mg, 50% yield), mp: 245–247 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.20 (s, 1H), 8.08 (d, J = 9.6 Hz, 1H), 7.80–7.70 (m, 5H), 7.56–7.51 (m, 3H), 7.48–7.43 (m, 4H), 7.04 (t, J = 9.6 Hz, 1H), 2.62 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 163.6 (d, ¹ $J_{C-F} = 252.5$ Hz), 160.2 (d, ¹ $J_{C-F} = 246.6$ Hz), 141.2, 140.1, 136.1, 133.1 (d, $J_{C-P} = 109.2$ Hz), 131.9, 131.8, 131.4, 131.3, 130.9, 128.8, 128.0 (d, $J_{C-P} = 12.2$ Hz), 122.5, 121.9, 104.3 (d, $J_{C-F} = 27.2$ Hz), 103.9, 22.2. ³¹P NMR (162 MHz, CDCl₃) δ 34.8 (d, J = 4.1 Hz). HRMS (ESI): m/z [M + H]⁺ calcd for C₂₆H₁₉F₂NOP: 430.1167, found: 430.1171.

(3-Methylphenanthridin-6-yl)diphenylphosphine oxide [23] (**3oa**): Isolated ($R_f = 0.4$, EtOAc—petroleum ether = 1:3) as a white solid (45.6 mg, 58% yield), mp: 197–200 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.39 (d, *J* = 8.0 Hz, 1H), 8.46 (d, *J* = 8.4 Hz, 1H), 8.32 (dd, *J* = 8.2, 2.4 Hz, 1H), 7.88–7.80 (m, 4H), 7.72 (s, 1H), 7.68 (t, *J* = 7.2 Hz, 1H), 7.53 (t, *J* = 7.6 Hz, 1H), 7.42–7.37 (m, 3H), 7.36–7.30 (m, 4H), 2.42 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 156.7 (d, *J*_{C-P} = 128.2 Hz), 142.9 (d, *J*_{C-P} = 23.1 Hz), 139.0, 133.1 (d, *J*_{C-P} = 104.1 Hz), 132.7 (d, *J*_{C-P} = 6.9 Hz), 132.3 (d, *J*_{C-P} = 9.1 Hz), 131.6 (d, *J*_{C-P} = 2.6 Hz), 130.9, 130.6, 130.5, 128.5, 128.2 (d, *J*_{C-P} = 5.0 Hz), 21.4. ³¹P NMR (162 MHz, CDCl₃) δ 28.1. HRMS (ESI): *m*/*z* [M + H]⁺ calcd for C₂₆H₂₁NOP: 394.1355, found: 394.1358.

Benzo[i]phenanthridin-5-yldiphenylphosphine oxide (**3pa**): Isolated ($R_f = 0.5$, EtOAc—petroleum ether = 1:1) as a white solid (32.6 mg, 38% yield), mp: 221–223 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.40 (d, *J* = 8.8 Hz, 1H), 9.09–9.06 (m, 1H), 9.05–9.00 (m, 1H), 8.16–8.12

(m, 1H), 8.04–8.00 (m, 1H), 7.98–7.90 (m, 5H), 7.77–7.66 (m, 4H), 7.54–7.40 (m, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 155.3 (d, J_{C-P} = 128.2 Hz), 144.8 (d, J_{C-P} = 23.3 Hz), 133.1 (d, J_{C-P} = 104.6 Hz), 132.4 (d, J_{C-P} = 9.1 Hz), 131.9 (d, J_{C-P} = 6.8 Hz), 131.7, 131.6, 130.9, 128.9, 128.8, 128.7, 128.5, 128.4 (d, J_{C-P} = 2.7 Hz), 128.2 (d, J_{C-P} = 12.0 Hz), 128.1, 127.4 (d, J_{C-P} = 22.6 Hz), 127.2, 126.7, 124.6, 124.0. ³¹P NMR (162 MHz, CDCl₃) δ 29.1. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₉H₂₁NOP: 430.1355, found: 430.1359.

(10-Methoxyphenanthridin-6-yl)diphenylphosphine oxide (**3qa**): Isolated ($R_f = 0.3$, EtOAc—petroleum ether = 1:1) as a white solid (45.9 mg, 56% yield), mp: 244–246 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.58–9.45 (m, 1H), 9.14 (d, *J* = 8.2 Hz, 1H), 8.03–7.98 (m, 1H), 7.93–7.86 (m, 4H), 7.72–7.64 (m, 2H), 7.62 (t, *J* = 8.0 Hz, 1H), 7.53–7.47 (m, 2H), 7.46–7.40 (m, 4H), 7.30 (d, *J* = 8.0 Hz, 1H), 4.10 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 158.1 (d, *J*_{C-P} = 2.7 Hz), 156.4 (d, *J*_{C-P} = 129.1 Hz), 143.3 (d, *J*_{C-P} = 23.2 Hz), 133.1 (d, *J*_{C-P} = 104.5 Hz), 132.3 (d, *J*_{C-P} = 104.5 Hz), 132.3 (d, *J*_{C-P} = 9.1 Hz), 131.6 (d, *J*_{C-P} = 5.6 Hz), 124.2, 123.1 (d, *J*_{C-P} = 7.0 Hz), 120.8, 112.1, 55.8. ³¹P NMR (162 MHz, CDCl₃) δ 29.2. HRMS (ESI): *m*/*z* [M + H]⁺ calcd for C₂₆H₂₁NO₂P: 410.1304, found: 410.1309.

(2-Chlorophenanthridin-6-yl)diphenylphosphine oxide [18] (**3ra**): Isolated ($R_f = 0.4$, EtOAc—petroleum ether = 1:3) as a white solid (38.1 mg, 46% yield), mp: 241–243 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.41 (d, *J* = 8.0 Hz, 1H), 8.48 (d, *J* = 8.0 Hz, 1H), 8.47 (s), 7.89 (d, *J* = 8.8 Hz, 1H), 7.86–7.80 (m, 4H), 7.77 (d, *J* = 8.0 Hz, 1H), 7.64 (t, *J* = 7.6 Hz, 1H), 7.56 (dd, *J* = 8.8, 2.0 Hz, 1H), 7.47–7.42 (m, 2H), 7.40–7.34 (m, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 156.3 (d, *J*_{C-P} = 128.2 Hz), 140.1 (d, *J*_{C-P} = 23.4 Hz), 138.2, 133.9, 131.6 (d, *J*_{C-P} = 104.7 Hz), 131.5, 131.3, 131.2, 130.8 (d, *J*_{C-P} = 2.6 Hz), 130.3, 128.3, 127.7, 127.5, 127.2 (d, *J*_{C-P} = 12.2 Hz), 124.4, 121.1, 120.7, 113.0. ³¹P NMR (162 MHz, CDCl₃) δ 28.5. HRMS (ESI): *m*/*z* [M + H]⁺ calcd for C₂₅H₁₈CINOP: 414.0809, found: 414.0813.

(2-Fluorophenanthridin-6-yl)diphenylphosphine oxide [18] (**3sa**): Isolated ($R_f = 0.3$, EtOAc—petroleum ether = 1:3) as a yellow solid (31.8 mg, 40% yield), mp: 235–237 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.41 (d, *J* = 8.4 Hz, 1H), 8.45 (d, *J* = 8.0 Hz, 1H), 8.13–8.09 (m, 1H), 7.99–7.93 (m, 1H), 7.86–7.76 (m, 5H), 7.64 (t, *J* = 7.6 Hz, 1H), 7.48–7.33 (m, 7H). ¹³C NMR (100 MHz, CDCl₃) δ 161.4 (d, *J*_{C-F} = 248.9 Hz), 155.1 (d, *J*_{C-P} = 131.7 Hz), 138.6 (d, *J*_{C-P} = 23.1 Hz), 132.5 (d, *J*_{C-P} = 9.3 Hz), 131.7 (d, *J*_{C-P} = 104.3 Hz), 131.2 (d, *J*_{C-P} = 9.1 Hz), 130.7, 130.1, 127.6 (d, *J*_{C-F} = 23.2 Hz), ³¹P NMR (162 MHz, CDCl₃) δ 28.5. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₅H₁₈FNOP: 398.1105, found: 398.1112.

(3-Chlorophenanthridin-6-yl)diphenylphosphine oxide (**3ta**): Isolated ($R_f = 0.4$, EtOAc—petroleum ether = 1:3) as a white solid (30.6 mg, 37% yield), mp: 194–197 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.44 (d, *J* = 8.4 Hz, 1H), 8.48 (d, *J* = 8.0 Hz, 1H), 8.40 (d, *J* = 8.8 Hz, 1H), 7.95 (d, *J* = 1.6 Hz, 1H), 7.87–7.80 (m, 4H), 7.76 (t, *J* = 7.6 Hz, 1H), 7.64–7.54 (m, 2H), T (m, 2H), 7.41–7.32 (m, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 158.5 (d, *J*_{C-P} = 125.7 Hz), 143.2 (d, *J*_{C-P} = 23.4 Hz), 134.5, 132.6 (d, *J*_{C-P} = 104.5 Hz), 132.2 (d, *J*_{C-P} = 9.1 Hz), 131.9, 131.5, 130.1, 129.3, 128.8, 128.3 (d, *J*_{C-P} = 12.3 Hz), 127.8 (d, *J*_{C-P} = 22.7 Hz), 123.6, 122.9 (d, *J*_{C-P} = 2.4 Hz), 122.0. ³¹P NMR (162 MHz, CDCl₃) δ 28.3. HRMS (ESI): *m*/*z* [M + H]⁺ calcd for C₂₅H₁₈ClNOP: 414.0809, found: 414.0816.

4. Conclusions

Rose Bengal was found to be an available photocatalyst for the cascade phosphorylation cyclization of 2-isocyanobiphenyls. A wide range of 6-phosphorylated phenanthridines was synthesized efficiently via visible-light-induced radical addition cyclization under metal-free conditions. A biological screening of these P-containing compounds is in progress in our laboratory.

Supplementary Materials: The following supporting information can be downloaded at: https://www.mdpi.com/article/10.3390/catal13071036/s1. The experimental procedures and characterization (¹H- and ¹³C-NMR and ³¹P-NMR) for all of the products are provided in the supporting information.

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