

Article

Fagonia cretica-Mediated Synthesis of Manganese Oxide (MnO₂) Nanomaterials Their Characterization and Evaluation of Their Bio-Catalytic and Enzyme Inhibition Potential for Maintaining Flavor and Texture in Apples

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Abstract: The apple is the most widely used fruit globally. Apples are more prone to fungal spoilage, which leads to browning and subsequent changes in their flavor and texture. Browning is also caused by the tyrosinase enzyme. By inhibiting tyrosinase initiation and fungal spoilage in fruits, the natural flavor and texture of fruits can be maintained. Biogenic NPs can act as antioxidants to inhibit tyrosinase and due to oxidative stress, it also catalyzes the deformation of fungal hyphae and spores. Nanotechnology is a research hotspot that has gained considerable interest due to its potential inferences in biosciences and food preservation technology. The present study aims to use biomass from the *Fagonia cretica* to create bio-inspired manganese oxide MnO₂ NPs and to evaluate its bio-catalytic potential for antifungal anti-browning through the inhibition of tyrosinase and its antioxidant potential for preserving apple flavor and texture. The green synthesized nanoparticles were extensively analyzed using UV spectroscopy, XRD, SEM, EDX, and FTIR techniques. Moreover, the synthesized manganese oxide nanoparticles (MnO₂ NPs) were evaluated for their bio-catalytic potential as anti-fungal and anti-spoiling agents. The values of antifungal activity among all the samples were 14.2 ± 86 mm, 8.9 ± 6.0 mm, 17.7 ± 1.26, and 20.7 ± 4.38 mm for *Penicillium expansum*, *Monilinia fructigena*, *Penicillium chrysogenum*, and *Aspergillus oryzae* at 200 µg/well, respectively. Moreover, the biogenic NPs were evaluated for their anti-browning potential through the inhibition of tyrosinase. MnO₂ NPs have been shown to have considerable inhibitory effects on tyrosinase up to 64.8 ± 0.16 at 200 µg/mL and (27.2 ± 0.58) at 25 µg/mL. Biogenic MnO₂ NPs can also act as antioxidants to inhibit tyrosinase and fungal growth by the formation of free radicals that damage the fungal hyphae and, as a result, slow down browning. The maximum DPPH free radical scavenging activity was 74.5 ± 0.39% at 200 µg/mL, and the minimum was 12.4 ± 0.27 at 25 µg/mL. The biogenic MnO₂ NPs are biocompatible and play a potent role in maintaining the flavor and texture of apples.

Keywords: MnO₂ NPs; *Fagonia cretica*; FTIR; apple; flavor; texture

1. Introduction

Fruits are the outcome of fertilized ovaries, and, as such, they have long been a valuable source of sustenance for humans. Despite the multiple benefits they bring, fruits survival and longevity are threatened by a range of factors, the most serious of which are fungal attacks [1]. Fungi may induce fruit spoilage, which refers to a number of changes that make the fruits less pleasant or desirable in terms of flavor and scent, as well as appearance and texture. As a result of the discovery of a huge number of toxigenic fungi in damaged fruits, rotting microorganisms are often viewed as poisonous or pathogenic [1,2]. Even when food is refrigerated, microbes, such as molds and other fungi, may produce mycotoxins in different forms, posing a health concern to consumers. Fruits with preservatives have a longer shelf life because they prevent them from rotting. Antimicrobials, such as nitrogen oxides, nitrogen trioxides (NO_x), benzoates, and sulfur dioxide, destroy or inhibit bacteria, yeast, and mold development in fruits [2]. However, since certain synthetic food and cosmetic ingredients have been shown to be carcinogenic and very dangerous, it is best to avoid using them. Synthetic chemical additives and preservatives should be avoided due to a lack of sufficient testing [1].

The plant-based synthesis of NPs has revolutionized drug discovery. *Fagonia cretica* contains many important bioactive elements that play a promising role in the synthesis of NPs. The chemical constituents are saponins, saponinins, alkaloids, terpenoids, sterols, flavonoids, proteins, amino acids, coumarins, vitamins, and trace elements [3]. Biosynthesized nanoparticles, on the other hand, are safer and more effective in slowing the degradation of fruits and preserving their texture and flavor. Biogenic MnO₂ NPs have the ability to act as a biocatalyst while simultaneously being poisonous to cells and damaging membranes [4]. According to one study, membrane rupture and free radical production are two of the most prevalent mechanisms of NP-induced cellular damage [5]. The interaction of nanoparticles with fungal hyphae and spores has been proven to inhibit fungal growth.

Fungi may also cause fruit to become brown, altering the flavor and texture of the fruit as a consequence of the browning. Browning is the gradual change in the color of food from white to brown or dark brown over time, which may have a positive or negative influence on the nutritional content of the meal. Browning may be classed as enzymatic or non-enzymatic browning in food, depending on the method involved. During the harvesting, transportation, storage, and processing phases of FV goods, enzymatic browning is the most common. Enzymatic browning produces a significant quantity of waste. Tyrosinase is also responsible for enzymatic browning in apples, which affects the fruits' flavor and texture [6]. The oxidative process is supposed to diminish the flavor and nutritional value of fruits and vegetables due to the browning caused by tyrosinase. As a consequence, a variety of tyrosinase inhibitors have been produced, including various glycyl dipeptides, L-ascorbic acid, ellagic acid, and others [7]. These commercially available chemical inhibitors, on the other hand, have a variety of deleterious effects on apple quality. As a consequence, novel biocompatible tyrosinase inhibitors, capable of lowering tyrosinase activity in apples while maintaining their flavor and texture, have been developed.

Metal oxide nanoparticles, on the other hand, could not be used by fruits owing to the use of harmful substances that act as reducing and capping agents in the chemical synthesis of these nanoparticles [8]. These novel materials are nontoxic and safer thanks to the use of plant metabolites that act as reducing and capping agents in the biogenesis of metal oxide nanoparticles. There have been many reports of tyrosinase enzyme inhibitory nanoparticles [5,6]. Nanoparticles have greater anti-tyrosinase action than equivalent plant metabolites, according to research [5]. Biogenic NPs may also act as antioxidants by inhibiting the initiation of tyrosinase, which prevents browning in fruits [8]. The capacity of biogenic nanoparticles to inhibit tyrosinase, as well as their bio-catalytic activity and mechanisms of action, is seldom recorded [9].

In the present study, we attempted to synthesize MnO₂ using *Fagonia cretica* as a reducing agent. The synthesized MnO₂ were characterized by UV spectroscopy, XRD, SEM, EDX, and FTIR techniques to visualize their morphology, involvement of bioactive compounds,

stability, crystallinity of the particles, etc. Furthermore, the tyrosinase inhibitory and antioxidant potential of MnO₂ nanoparticles for slowing down browning and its bio-catalytic potential as an antifungal agent against spoilage causing fungi was also evaluated.

2. Results

2.1. Synthesis Mechanism

The leaf extract of *Fagonia cretica* was used as a reducing and capping agent in the synthesis of MnO₂ NPs. The synthesis of MnO₂ NPs was visually monitored by detecting the color change generated by the addition of a precursor to the leaf extraction. The color of the reaction mixture changed from yellowish green to brownish, indicating the production of manganese dioxide NPs. The color change was caused by the surface plasmon resonance activity of the nanoparticle. Several studies have shown that the *Fagonia cretica* leaf's extract is a rich source of biogenic phytochemicals such as alkaloids, flavonoids, tannins, phenolic compounds, saponins, and triterpenoids. These phytochemicals may work during the biosynthesis process by converting manganese ions to zero-valent species through a reduction and oxidation mechanism, resulting in keto form products. Other secondary metabolites (surfactants, proteins, alkaloids, and so on) found in the *Fagonia cretica* leaf extract also stabilized and capped the zero-valent Mn⁰ species. The zero-valent species of Mn⁰ would quickly oxidize and transform into MnO₂ nanoparticles capped with phytochemicals of plant leaf extract during air drying and calcination at 200 °C. (See Figure 1).

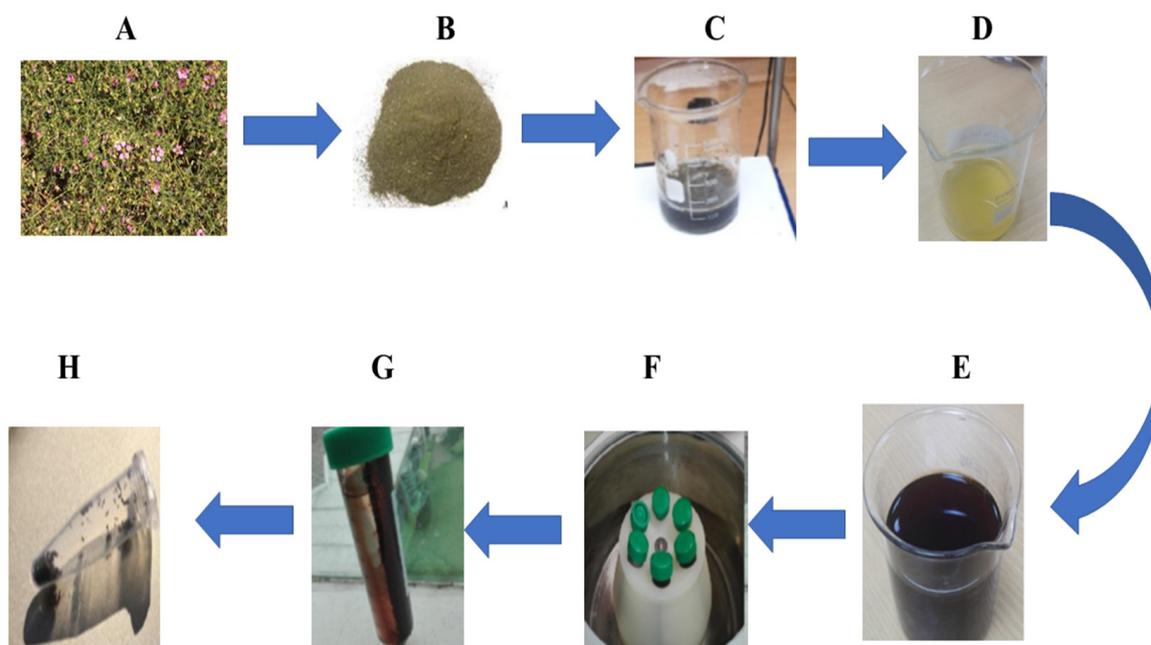


Figure 1. The entire process of making MnO₂ NPs is depicted in this schematic picture. (A) *Fagonia cretica* plant, (B) *Fagonia cretica* powder, (C) *Fagonia cretica* extract, (D) filtrate, (E) reduction and capping of MnO ions by *Fagonia cretica* extract, (F) centrifugations, (G) pellets of NPs, and (H) purified MnO₂ NPs.

2.2. Characterization

The dark brownish hue is a marker of nanoparticle production, according to UV–visible spectroscopy in the 300 nm to 800 nm region. An examination of zinc nanoparticle absorption at 2.25 au shows a distinct peak at 410 nm, indicating that they were manufactured. This is seen in Figure 2a. In metallic nanoparticles, there is only one SPR peak, which is detected in MnO₂ NPs. Infrared spectroscopy confirms the creation of the produced nanoparticles. Figure 2b depicts the essential infrared stretching vibrations. The absorption band at 562.2 cm^{−1} was ascribed to the normal stretching collision of O–Mn–O, indicating that MnO₂ NPs were present in the produced sample. Aromatic unsaturation (C=C) of

stabilized *Fagonia cretica* is represented by absorption bands at 1616.1 and 1115.4 cm^{-1} , respectively, whereas C–O stretching in the *Fagonia cretica* molecule is represented by an absorption band at 628.4 cm^{-1} . The wide band at 3385.9 cm^{-1} is the typical absorption for O–H–O of water present in the solution, which may play a role in manganese nanoparticle production and stability in the aqueous medium. The crystallinity of MnO_2 NPs produced using the *Fagonia cretica* leaf extract was investigated using XRD analysis. The XRD pattern of MnO_2 NPs is shown in Figure 2c. The XRD pattern shows five distinct peaks at 2θ 28.78°, 37.66°, 42.14°, 49.90°, and 56.44°, corresponding to the crystal planes of MnO_2 NPs (310), (211), (301), (411), and (600) (JSPDF 44-0141). Furthermore, the strength of the peaks in the XRD pattern suggests that the MnO_2 NPs are very crystalline. The EDX analysis was then used to identify the chemical composition of the MnO_2 NPs. The EDX pattern is shown in Figure 3D. Four distinct peaks in the EDX spectra correspond to sodium, potassium, sulphur, chlorine, and manganese. In EDX spectra, a tiny peak of Mn can also be seen. The SEM pictures of MnO_2 NPs are shown in Figure 3A. The produced MnO_2 NPs are spherical with homogenous dispersity, as seen in the SEM picture. The particle size of the MnO_2 NPs produced was determined to be 15.5 ± 0.85 nm.

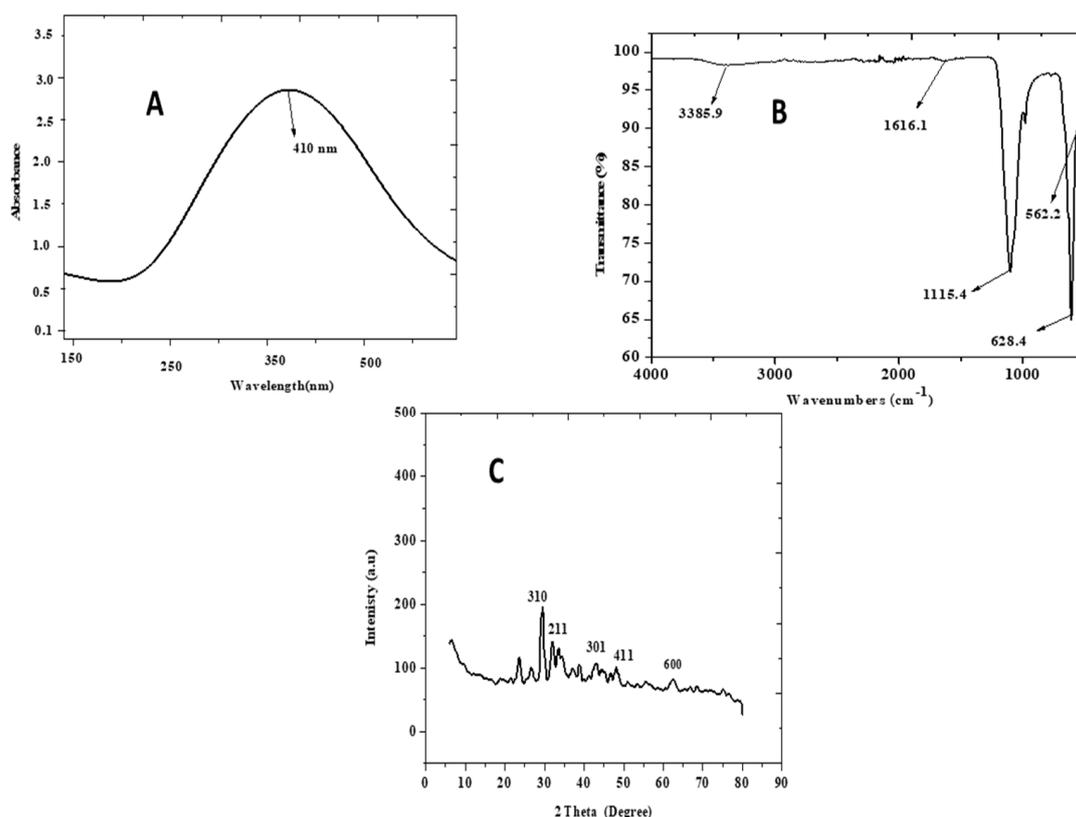


Figure 2. (A) UV–visible spectroscopy, (B) Fourier transform infrared (FTIR) spectroscopy, and (C) XRD pattern for the green synthesized MnO_2 NPs.

2.3. Anti-Browning Activity of NPs

2.3.1. Tyrosinase Assay

The actual mechanism that causes browning in apples involves an enzyme called tyrosinase. The inhibitory ability of NPs against the production of tyrosinase was determined. MnO_2 NPs block the active site and cause enzyme inhibition; thus they bear anti-browning characteristics and may be employed as anti-browning agents. MnO_2 NPs were used as anti-tyrosinase agents in our work. MnO_2 NPs have been shown to have considerable inhibitory effects on tyrosinase up to 64.8 ± 0.16 at 200 $\mu\text{g}/\text{mL}$ and (27.2 ± 0.58) at 25 $\mu\text{g}/\text{mL}$, as shown in Figure 4.

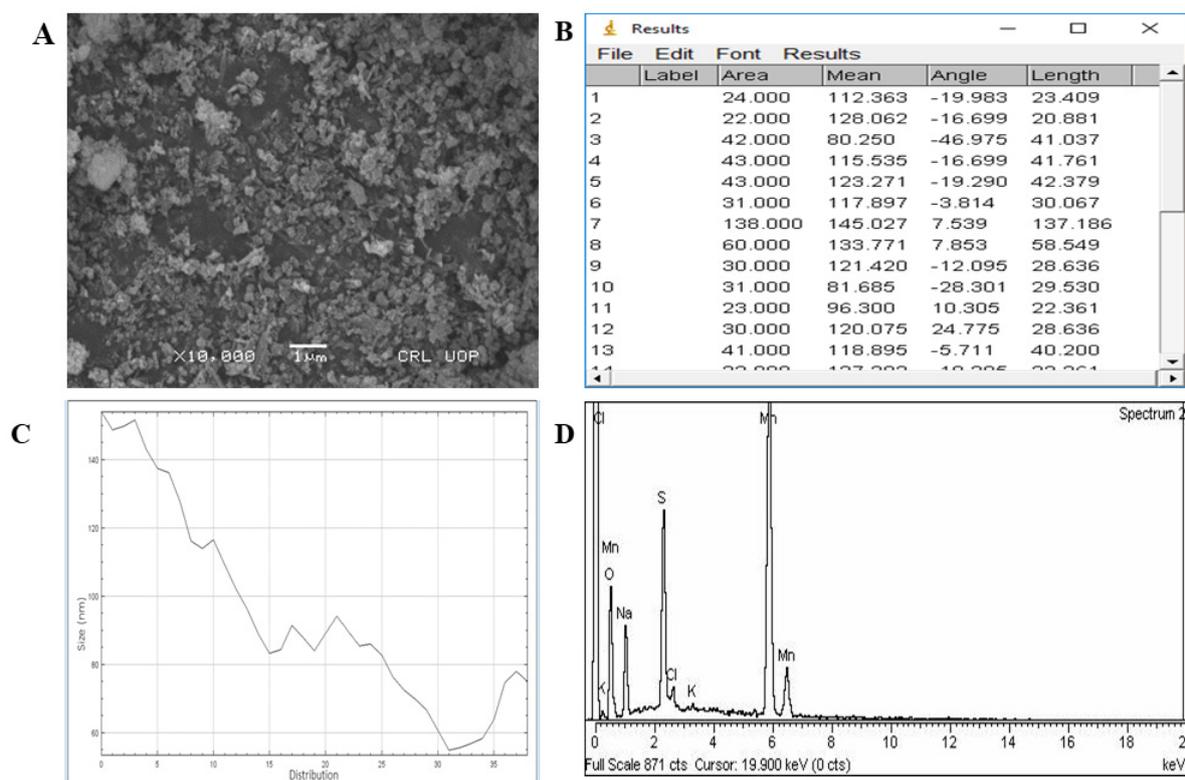


Figure 3. (A) Scanning electron microscopy (SEM), (B) image J analysis of SEM image, (C) size distribution, and (D) energy-dispersive X-ray (EDX) for the green synthesized MnO_2 NPs.

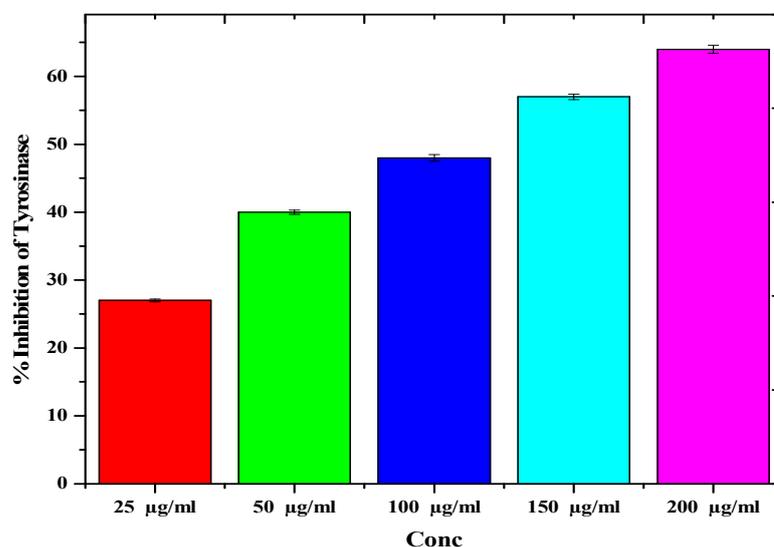


Figure 4. Anti-browning potential of biosynthesized MnO_2 NPs by inhibition of tyrosinase.

2.3.2. Antioxidant Assay

Antioxidants can react with oxygen to suppress the initiation of browning. They are also able to react with the intermediate products, thereby breaking the chain reaction and inhibiting tyrosinase formation. 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radicals were exposed to test samples of varying concentrations in order to evaluate the antioxidant potential of MnO_2 NPs, as shown in Figure 5. The maximum DPPH free radical scavenging activity was $74.5 \pm 0.39\%$ at (200 $\mu\text{g}/\text{mL}$), and the minimum was 12.4 ± 0.27 at (25 $\mu\text{g}/\text{mL}$).

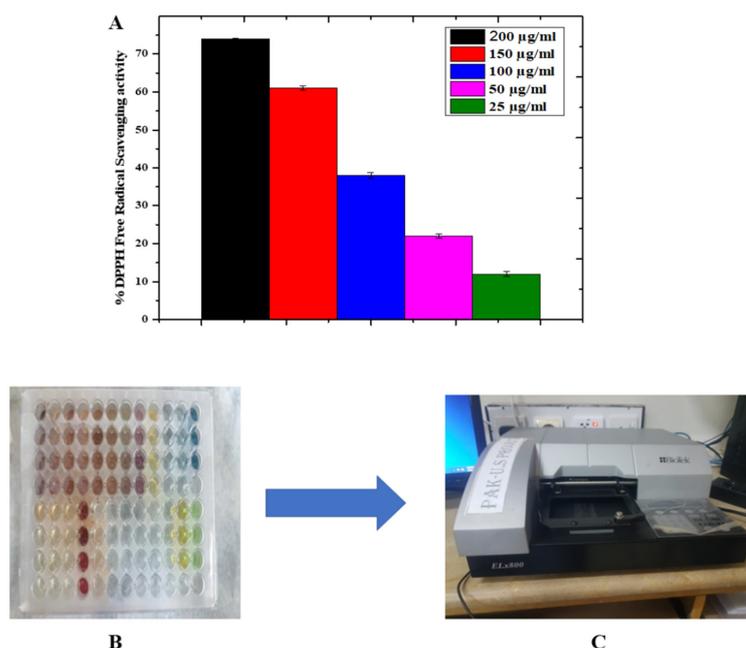


Figure 5. (A) Antioxidant potential of biosynthesized MnO₂ NPs, (B) Test solution in the 96-well plates and (C) microplate reader.

2.3.3. Anti-Spoil Activity of NPs against Spoil-Causing Fungal Isolate Species

Characterization of fungal flora: The characterization of fungal flora from fruits obtained in the Peshawar market was shown in Figure 6A. The results show the occurrence of four different fungal species distributed among the spoiled fruits: *Penicillium expansum*, *Monilinia fructigena*, *penicillium chrysogenum*, and *Aspergillus oryzae*. Figure 6 shows the identification of fungus. Figure 7 show the inhibition zone of MnO₂ NPs. The outcomes of MnO₂ NPs antifungal potential are depicted in Figure 8. In this study, MnO₂ was added to each well. The values of antifungal activity of all the samples were 14.2 ± 86 mm, 8.9 ± 6.0 mm, 17.7 ± 1.26 mm, and 20.7 ± 4.38 mm for *Penicillium expansum*, *Monilinia fructigena*, *Penicillium chrysogenum*, and *Aspergillus oryzae* at 200 µg/mL in each well, respectively. According to several studies, the most prevalent routes of NP-induced cellular toxicity include membrane disruption and free radical production. Fungal growth is inhibited by the interaction of nanoparticles with fungal hyphae and spores.

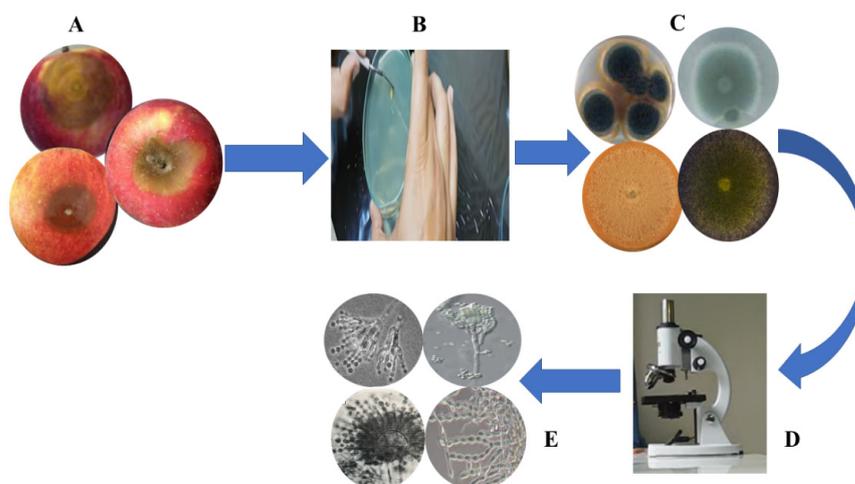


Figure 6. (A) Spoilt apple fruits, (B) culture on PDA media, (C) growth of fungus on PDA media, (D) microscope, and (E) microscopic identification of fungus.

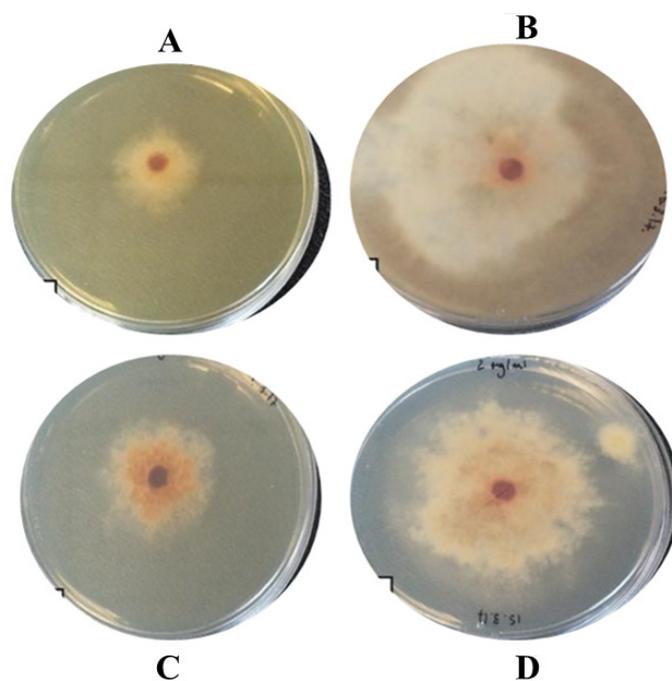


Figure 7. Shows the zone of inhibitions of NPs against (A) *Penicillium expansum*, (B) *Monilinia fructigena*, (C) *Penicillium chrysogenum*, and (D) *Aspergillus oryzae*.

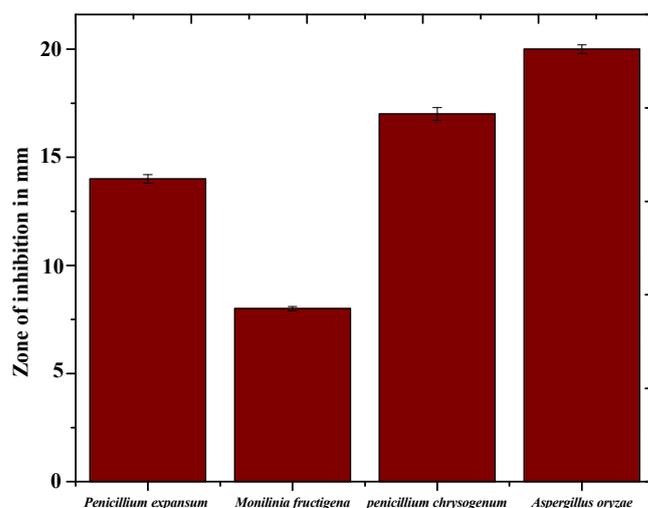


Figure 8. Anti-fungal potential of MnO_2 NPs against spoil-causing fungal isolates.

3. Discussion

Because of its simplicity of use and cost efficiency, as well as its potential for large-scale manufacturing, the NP synthesis approach has piqued the scientific community's attention in recent years. We used a biosynthesis approach to make MnO_2 NPs from *Fagonia cretica*, and we tested NPs as bio-catalysts for antifungal agents and enzyme inhibitors in food preservation and flavor maintenance. During the first steps, the MnO_2 NPs were explored for their morphological features by employing diverse analytical tools, including UV spectroscopy, FTIR, XRD, SEM, and EDX. According to UV spectroscopy, the sample absorbed energy at 410 nm, which is a sign of typical peak value for MnO_2 NPs. Aside from that, an absorption peak at 410 nm with no other peak demonstrated the NPs' exceptional purity. Many investigations revealed a significant absorption peak of MnO_2 NPs 410 nm wavelengths, related to the samples' red shift at 500 °C and 700 °C [10]. FTIR analysis was used to observe the stretching vibration of the different functional groups, which

show different absorption bands. The absorption band at 562.2 cm^{-1} shows a collision of O–Mn–O; 628.4 cm^{-1} shows C–O stretching; 1115.4 cm^{-1} , 1616.1 cm^{-1} indicates aromatic unsaturation (C=C) of the stabilized *Fagonia cretica* system; and 3385.9 cm^{-1} is the typical absorption for O–H–O of water present in the solution, respectively; our FTIR results are in correspondence with [11]. The produced MnO₂ NPs had crystalline in the range of $15.5 \pm 0.85\text{ nm}$, as estimated by Nano-measurer and ImageJ analysis, as confirmed by SEM micrographs. The size of NPs was larger in this work than in [12], which might be attributed to changes in synthesis settings such as temperature, incubation period, bacterial extract type, and handling applications. Furthermore, EDX analysis showed pure MnO₂ NPs phases and a strong peak in the EDX spectrum, showing that the test sample contained pure manganese. The EDX spectra of MnO₂ NPs were obtained using a simple precipitation process using manganese as the starting material. Pure MnO₂ NPs with substantial peaks were successfully synthesized, according to the EDX spectrum. However, additional peaks in the spectrum were detected, suggesting that bacterial biomolecules were involved in the nanoparticle synthesis. In comparison to other reports, we found the same EDX pattern of MnO₂ NPs with great purity [12]. The size and crystallinity of the bio-fabricated MnO₂ NPs were measured using an XRD profile. The XRD pattern shows five distinct peaks at $2\theta = 28.78^\circ$, 37.66° , 42.14° , 49.90° , and 56.44° , corresponding to the crystal planes of MnO₂ NPs (310), (211), (301), (411), and (600) (JSPDF 44-0141). Furthermore, the strength of the peaks in the XRD pattern suggests that the MnO₂ NPs are very crystalline [12]. After thorough morphological and chemical analysis, the produced NPs were tested for important applications, such as bio-catalysts as antifungal agents, and antioxidant and enzyme inhibitors in food preservation and flavor maintenance. We also checked the anti-browning mechanism that causes browning in apples involving an enzyme called tyrosinase. The inhibitory ability of NPs against the production of tyrosinase was determined. MnO₂ NPs block the active site and cause enzyme inhibition, which produce anti-browning characteristics and may be employed as anti-browning agents, which were used in our recent work. APE has been shown to have considerable inhibitory effects on tyrosinase up to 64.8 ± 0.16 at $200\text{ }\mu\text{g/mL}$ and (27.2 ± 0.58) at $25\text{ }\mu\text{g/mL}$. Similar results were reported by [13]. A variety of concentrations of synthesized NPs (25, 50, 100, 150, and 200 ppm) were evaluated as antioxidant agents. 2, 2-diphenyl-1-picrylhydrazyl (DPPH) free radicals were exposed to test samples of varying concentrations in order to evaluate the antioxidant potential of MnO₂ NPs. The maximum DPPH free radical scavenging activity was $74.5 \pm 0.39\%$ ($200\text{ }\mu\text{g/mL}$) and the minimum was 12.4 ± 0.27 at ($25\text{ }\mu\text{g/mL}$), respectively, in accordance with previous results [14]. We tested the NPs against spoil-causing fungal isolate in apples using well diffusion methods, and the antifungal potential of MnO₂ NPs was studied. The values of antifungal activity among all the samples were $14.2 \pm 86\text{ mm}$, $8.9 \pm 6.0\text{ mm}$, 17.7 ± 1.26 , and $20.7 \pm 4.38\text{ mm}$ for *Penicillium expansum*, *Monilinia fructigena*, *Penicillium chrysogenum*, and *Aspergillus oryzae* at $200\text{ }\mu\text{g/mL}$ in each well, respectively. Our results are congruent with earlier reports [15].

4. Materials and Methods

4.1. Chemicals

All of the chemicals used in this study were analytical grade and bought from Sigma Chemicals Co. (St. Louis, MO, USA) and Merck (Darmstadt, Germany). For NP synthesis, commercially available manganese acetate salt was purchased from US Research Nanomaterials, Inc. (3302 Twig Leaf Lane, Houston, TX 77084, USA).

4.2. Collection of the Plant Material

Fresh *Fagonia cretica* leaves were obtained in Peshawar, Pakistan's surrounding regions.

4.3. Preparation of Leaves Extract of *Fagonia cretica*

Using 20 g of fresh *Fagonia cretica* leaves, a *Fagonia cretica* leaf extract was made. To eliminate any pollutants or dust, the leaves were carefully washed with deionized (DI) water

and air-dried at 30 °C. The dried leaves were chopped into tiny bits and crushed in a professional blender before being transferred to a 500 mL beaker. After that, 150 mL of DI water was added and agitated for 60 min at 60 °C. After that, the *Fagonia cretica* leaves extract was cooled to room temperature and then filtered. For later usage, the filtrate was collected and kept at 4 °C in an airtight glass container.

4.4. Biogenic Synthesis of Manganese Dioxide NPs (MnO_2 NPs)

One millimeter of manganese acetate was added to twenty-five milliliters of *Fagonia cretica* leaves extract for the biogenic production of manganese dioxide NPs. At pH 7.15, the resultant liquid was heated for 70 min at 40 °C with continuous stirring. After that, the produced manganese dioxide NPs were isolated from the reaction mixture by centrifugation at 3000 rpm for 30 min. After centrifugation, the resulting NPs were washed three times with ethanol, dried at 40 °C, and then calcined for three hours in a muffle furnace at 200 °C. Finally, the green produced manganese dioxide NPs were designated MnO_2 NPs and kept in a glass container for further analysis.

4.5. Characterization of Biosynthesized MnO_2 NPs

Advanced tools were used to assess the physicochemical and morphological features of biosynthesized MnO_2 NPs [16]. The UV-vis-NIR spectrophotometer UV-3600 Plus Shimadzu was used for performing UV spectroscopy in the typical range of 200 to 700 nm to monitor the interaction between biomass and metallic salt. The crystal nature of biologically synthesized MnO_2 NPs was determined using the X-ray diffraction (XRD) profile. The Panalytical's X'Pert X-ray diffractometer was utilized to produce the XRD peaks at $CuK\alpha$ ($=1.54056 \text{ \AA}$). MnO_2 NPs were studied using the IRTracer-100 Fourier transform infrared (FTIR) spectrophotometer in the 400–4000 cm^{-1} spectrum region to reveal and assess related functional groups involved in their biosynthesis approach [17]. Scanning electron microscopy (SEM) was applied to measure the physical dimensions and morphological characteristics of biosynthesized MnO_2 NPs (JSM-5910, Tokyo, Japan). The biosynthesized MnO_2 NPs were subjected to energy-dispersive X-ray (EDX) spectroscopy to ascertain their elemental composition.

4.6. Anti-Browning Activity of NPs

4.6.1. Tyrosinase Assay

The actual mechanism that causes browning in apples involves an enzyme called tyrosinase. The inhibitory ability of MnO_2 NPs against the production of tyrosinase was determined. A tyrosinase test was performed using the previously reported technique of [18], which employed L-DOPA (5 mM; Sigma Aldrich, St. Louis, MO, USA). The test sample was combined with 10 mL of an L-DOPA diphenolase substrate and a sodium phosphate buffer (50 mM, pH 6.8). By adding 0.2 mg/mL of mushroom tyrosinase solution to the reaction mixture, the final volume increased to 200 mL (Sigma Aldrich). As a control, the DMSO solvent was utilized to replace the tested sample. At 475 nm, the reaction activities were monitored using a microplate reader (BioTek ELX800; BioTek Instruments, Bad Friedrichshall, Germany). The percent inhibition was calculated in comparison to the matching control tyrosinase effect.

4.6.2. Antioxidant Assay

Antioxidants can react with oxygen to suppress the initiation of browning. They are also able to react with intermediate products, thereby breaking the chain reaction and inhibiting tyrosinase formation. 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radicals were exposed to test samples of varying concentrations in order to evaluate the antioxidant potential of MnO_2 NPs nanoparticles [19]. A DPPH solution was prepared and 180 μ L was poured into specified wells of a 96-well titer plate. Afterwards, 20 μ L of different concentrations (25–400 μ g/mL) of nanoparticles was added to each well and incubated for 1 h at 37 °C. DMSO was used as a negative while ascorbic acid was used as a positive

control. After incubation, the samples were exposed to absorbance at 517 nm under the microplate reader and absorbance was recorded for each well. The antioxidant potential was calculated using the following formula:

$$\text{DPPH}(\%) = \left(1 - \frac{\text{Absorbance of Sample}}{\text{Absorbance of Control}} \right) \times 100 \quad (1)$$

4.7. Anti-Spoil Activity of NPs against Spoil-Causing Fungal Isolate Species

4.7.1. Sample Collection of Spoilt Apple

Apple fruit samples were selected at random from both wholesalers and merchants in Peshawar. The ruined fruits were detected using the Bukar et al. [20] approach of morphological evaluation. Before mycological examination, the materials were maintained in the refrigerator at 4 °C.

4.7.2. Culture Media Preparation

Chloramphenicol (30 mg mL⁻¹) was added to potato dextrose agar (PDA). The culture media were produced according to the instructions provided by the manufacturer. The suitable medium or foundation medium was weighed in its various amounts. After that, the weighed quantity of media was suspended in 400 mL of distilled water. Over a Bunsen flame, the media were brought to a boil until the agar melted. The pH of the molten agar medium was adjusted according to the manufacturer's recommendations after cooling to 45 °C. The media were cotton-plugged and covered in aluminum foil before being autoclaved for 15 min at 121 °C. The media were aseptically distributed in 20 mL aliquots sterile Petri plates after sterilization, and then allowed to set on the flat before being aseptically dispensed into sterile Petri dishes. The Petri dishes were labelled and placed in the refrigerator to be used later.

4.7.3. Isolation of Fungi from Spoilt Fruits

Dashwood et al. [21] and Balali et al. [22] used the same procedure to isolate the mycological flora. The diseased fruits were surface-sterilized for 2 min with cotton wool soaked in 0.1 percent mercury chloride (HgCl₂), then washed three times in distilled water. To avoid bacterial development, a sterile blade and forceps were used to cut a tiny segment of tissue (3 mm diameter), including both the healthy and rotten portions, and then plated on solidified potato dextrose agar (PDA) with chloramphenicol (30 mg/mL). For 7 days, the infected plates were kept at room temperature (25 °C). As stated by Fawole and Oso [23], the various colonies detected on the plates were differentiated based on their cultural traits, such as colony size, shape, color, consistency, and hemolytic properties. To achieve pure isolates, the fungal isolates were subculture on PDA slants.

4.7.4. Identification of Fungal Isolates

Slide culture methods were used to identify fungal isolates collected from the slant based on their gross morphology, which included colony development pattern, conidial morphology, and pigmentation [24,25]. A small portion of aerial mycelia from the representative culture was picked with a sterile inoculating needle and inoculated on a slide containing a fraction of a prepared solidified potato dextrose agar and incubated for 48 h, after which it was viewed under a light microscope, first with a low resolution objective of ×10 and then with a high resolution objective of ×40 to detect spore, hyphae, and other special structures according to the mycological atlas of Domsch et al. [26]. We validated and verified the morphological traits and appearance of the fungal isolates from decaying fruits utilized in this investigation.

4.8. Antifungal Activity

It was determined that MnO₂ NPs have antifungal activity using the agar well diffusion assay [27] after they were generated in the laboratory. In this experiment, fungal

strains were disseminated on potato dextrose agar (PDA) plates, and uniform lawns were created using the spread plate method. In each plate, 5 mm wells were bored using a sterile well borer, and varied quantities of nanoparticles were put into each well. The plates were incubated at 28 °C for 72 h. Amphotericin B was utilized as a positive control, while DMSO was employed as a negative control in this study. The exercise was performed three times, and the mean inhibition zones were measured with the use of a Vernier caliper for each repetition.

5. Conclusions

In this study, the *Fagonia cretica* plant biomass-based MnO₂ NPs was prepared and characterized by various instrumental techniques and was then investigated as bio-catalytic for antifungal and anti-browning potential in order to preserve apple flavor and texture. The biomass of *Fagonia cretica* contains proteins, carbohydrates, and lipids—chemicals which helped in the capping and reduction of Mn⁰ into MnO₂ NPs. The prepared nanoparticles showed potent antifungal potential against spoil-causing fungal isolate species. DPPH and tyrosinase were potently inhibited by these nanoparticles, indicating that they could be effectively used as anti-browning agents for preserving apple flavor and texture.

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