

*Supporting Information*

# Comparing the Performance of Supported Ru Nanocatalysts Prepared by Chemical Reduction of RuCl<sub>3</sub> and Thermal Decomposition of Ru<sub>3</sub>(CO)<sub>12</sub> in the Sunlight-Powered Sabatier Reaction

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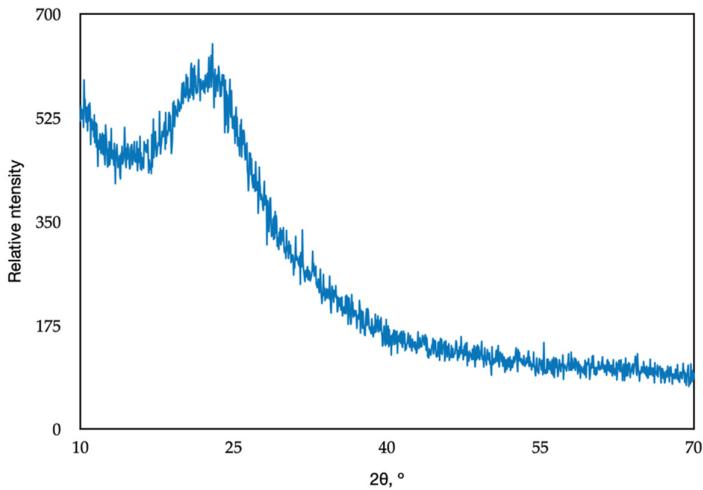
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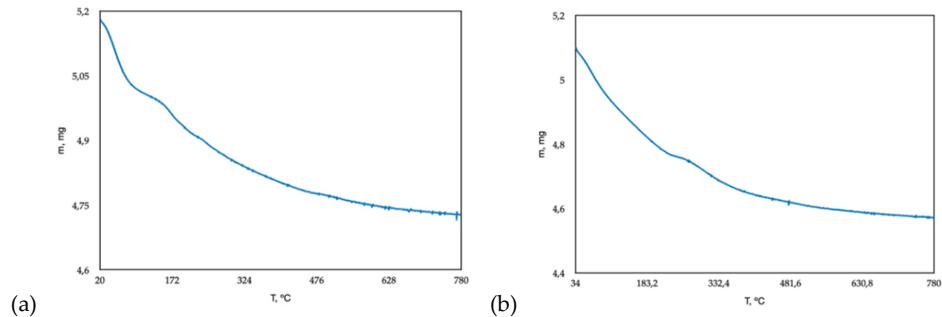
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**Figure S1.** X-ray diffractogram of  $\text{SiO}_2$  supported Ru catalyst prepared by reduction of  $\text{RuCl}_3$ .

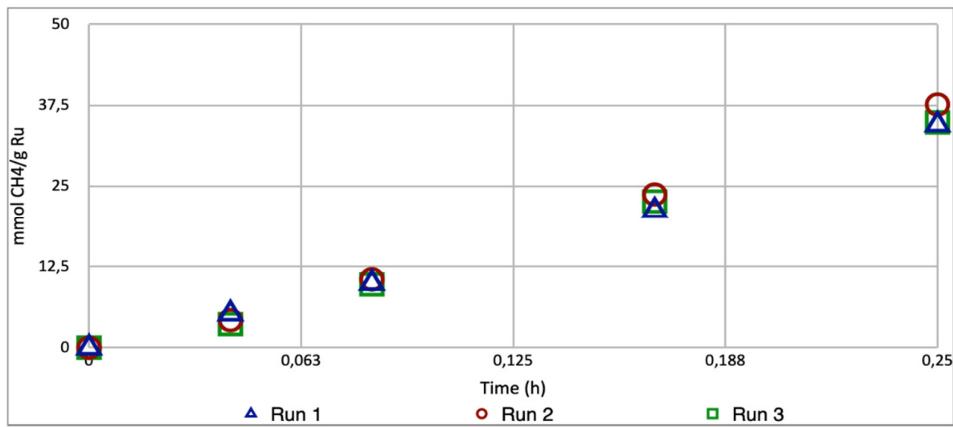
The powder X-ray diffraction pattern was recorded using a Bruker AXS D8 Discover diffractometer ( $\text{Cu K}\alpha$  radiation, LynxEye detector). The broad peak centered at  $2\theta = 23.8^\circ$  represents the short range order in amorphous  $\text{SiO}_2$ . The X-ray diffraction analysis neither shows diffraction peaks of crystalline Ru nanoparticles on the  $\text{SiO}_2$  support (expected at  $2\theta \approx 37^\circ$  and  $2\theta \approx 45^\circ$  [1]), nor reflections corresponding to crystalline  $\text{RuO}_2$  [1] (expected at  $2\theta \approx 28^\circ$ ,  $2\theta \approx 35^\circ$  and  $2\theta \approx 54^\circ$ ). Ergo, due to the low Ru loading and small particle size, Ru reflections could not be detected.

[1] JCPDS reference patterns: Ru 00-006-0663;  $\text{RuO}_2$  00-040-1290.

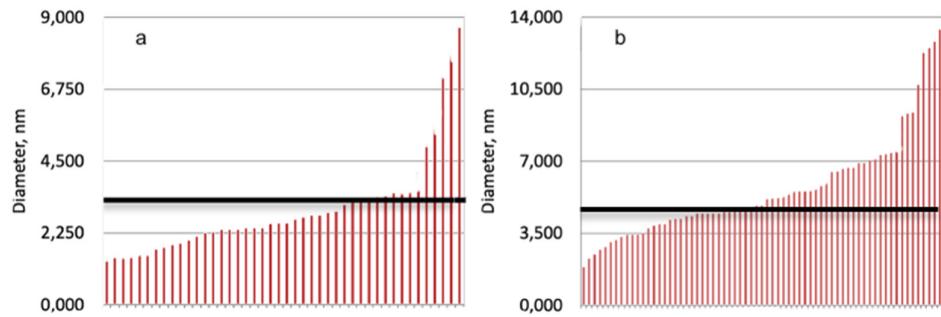


**Figure S2.** Thermogravimetric analyses of (a)  $\text{SiO}_2$ -supported and (b)  $\text{Al}_2\text{O}_3$ -supported Ru catalyst prepared by reduction of  $\text{RuCl}_3$ , under air.

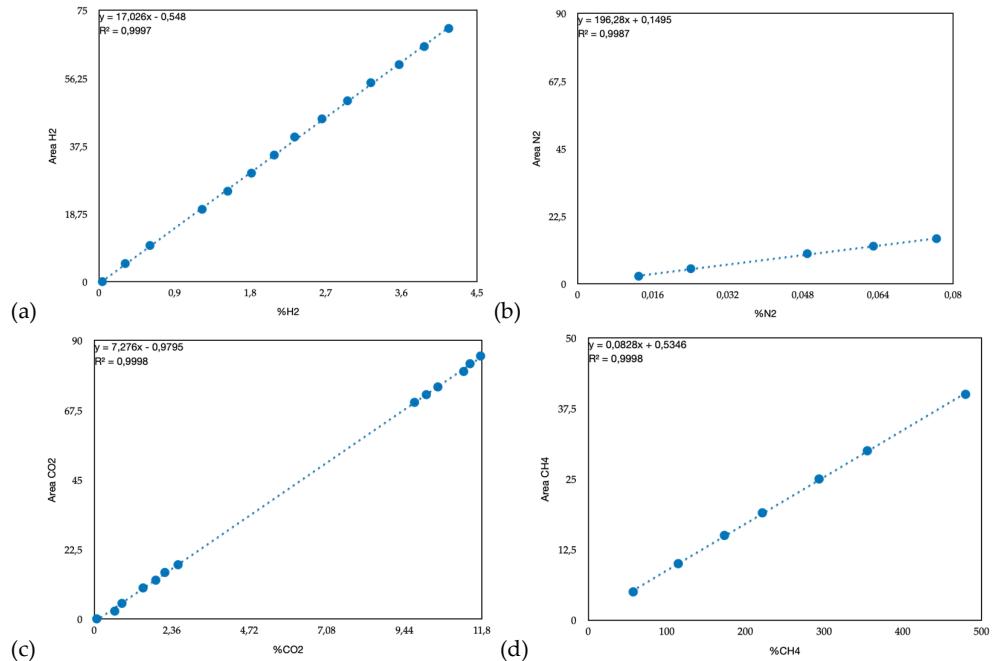
The thermal decomposition profiles of the samples were studied by thermogravimetric analysis (TGA, TA instruments Q500). The samples (~5 mg) were heated up to 800 °C at a heating rate of 10 °C/min under the air atmosphere. In both cases, the TGA analysis shows a mass loss of about 10%, which is likely due to loss of adsorbed water and progressive condensation of Si-OH and Al-OH groups.



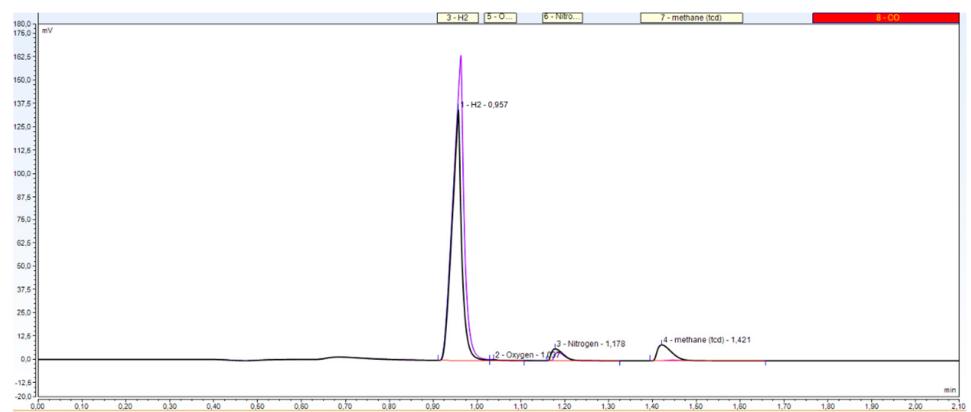
**Figure S3.** Conversion-time profile for three sequential runs of the sunlight-powered Sabatier reaction with Ru/SiO<sub>2</sub>-TD. Ru loading is 3.49% w/w. Reaction conditions for all experiments: reaction mixture of H<sub>2</sub>/CO<sub>2</sub>/N<sub>2</sub> (4.5:1:1) at 3.5 ± 0.2 bar pressure, 200 mg of Ru/SiO<sub>2</sub>-TD catalyst, light intensity of 6.66 suns, catalyst bed temperature approximately 220°C.



**Figure S4.** Comparison of mean diameter of Ru nanoparticles and numbers of agglomerates before (a) and after (b) reaction for Ru/SiO<sub>2</sub>-CR catalysts with a Ru loading on silica of 3.34% w/w. The black line in both diagrams indicates the border between single particles and agglomerates.



**Figure S5.** Calibration curves for GC detection of (a) H<sub>2</sub>, (b) N<sub>2</sub>, (c) CO<sub>2</sub> and (d) CH<sub>4</sub>.



**Figure S6.** Prototypical gas chromatogram for catalytic conversion of CO<sub>2</sub> and H<sub>2</sub> to CH<sub>4</sub> in a mixture diluted with N<sub>2</sub>.