

SUPPLEMENTARY INFORMATION

Photocatalytic-Fenton Process under Simulated Solar Radiation Promoted by a Suitable Catalyst Selection

Aida M. Díez ^{1,*}, Helen E. Valencia ^{2,3}, Maria Meledina ^{2,3}, Joachim Mayer ^{2,3} and Yury V. Kolen'ko ¹

¹ Nanochemistry group, International Iberian Nanotechnology group, Avda. Mestre José Veiga s/n, Braga 4715-330, Portugal; Yury.Kolenko@inl.int

² Central Facility for Electron Microscopy (GFE), RWTH Aachen University, D-52074 Aachen, Germany; valencia@gfe.rwth-aachen.de (H.E.V.); meledina@gfe.rwth-aachen.de (M.M.); j.mayer@fz-juelich.de (J.M.)

³ Forschungszentrum Jülich GmbH, Ernst Ruska-Centre (ER-C), D-52425 Jülich, Germany

* Correspondence: adiez@uvigo.es

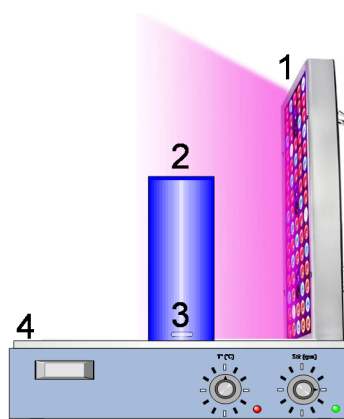


Fig. S1. Set-up schema. 1: Simulated visible lamp, 2: Cylindrical glass reactor, 3: Stirring bar, 4: Magnetic stirrer.

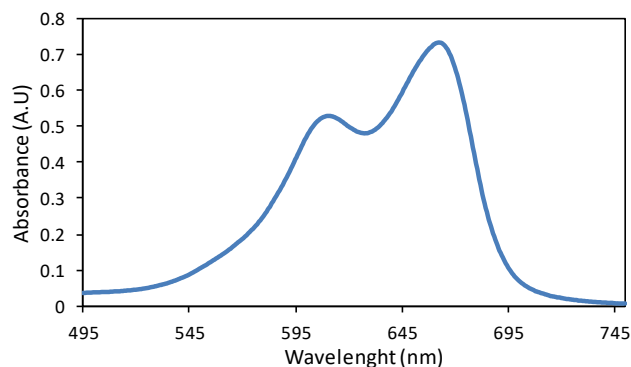


Fig. S2. Initial MB spectra attained by UV-Vis spectrophotometry.

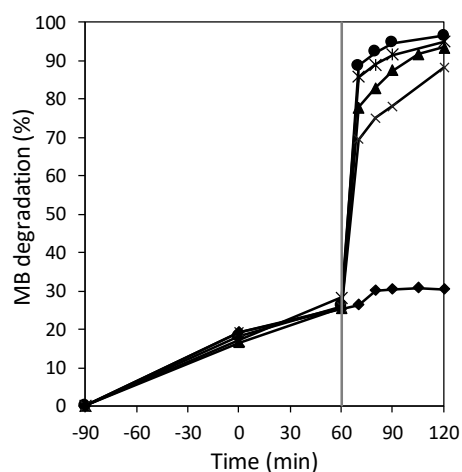
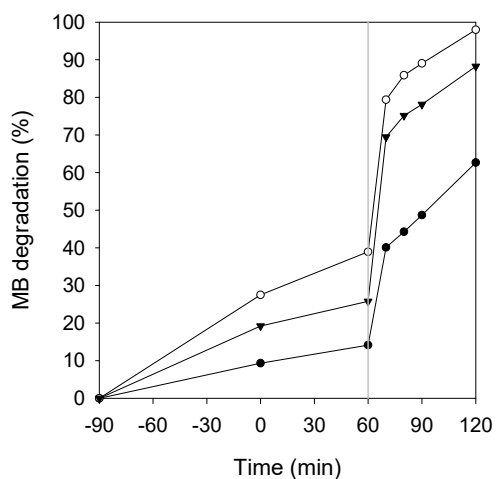


Fig. S3: MB degradation at different H_2O_2 dosages: 6g/L (circles), 3g/L (asterisk), 1.5 g/L (triangles), 0.6 g/L (crosses), 0 g/L (rhombus). The vertical grey line represents the moment of the H_2O_2 addition

A



B

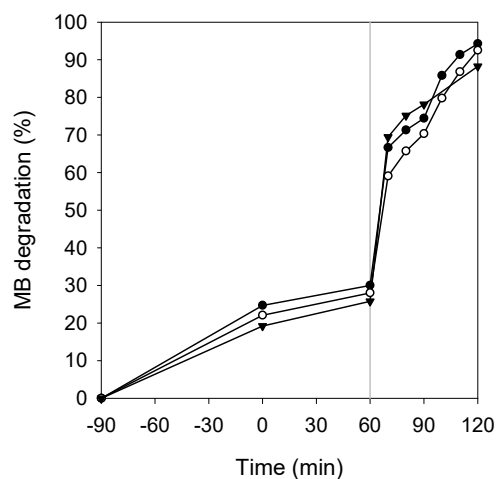


Fig. S4. A: MB degradation at MB dosage of 10 ppm (white cycles), 20 ppm (triangles) and 40 ppm (black circles) with H_2O_2 0.6 g/L and 0.8 g/L catalyst dosage B: MB degradation at same concentrations when keeping the ratio $\text{MB}/\text{H}_2\text{O}_2$ and $\text{MB}/\text{catalyst}$ dosage at 0.025 and 33.33, respectively. The vertical grey line represents the moment of the H_2O_2 addition.

Table. 1-S. Textural and structural properties of $\text{Fe}_3\text{O}_4\text{-SiO}_2\text{-TiO}_2$.

Surface area (m^2/g)	30.992
External surface area (m^2/g)	29.127
Pore Volume (cm^3/g)	0.240
Pore Diameter (nm)	3.715
Micropore volume (cm^3/g)	0.001
Micropore area (m^2/g)	1.865

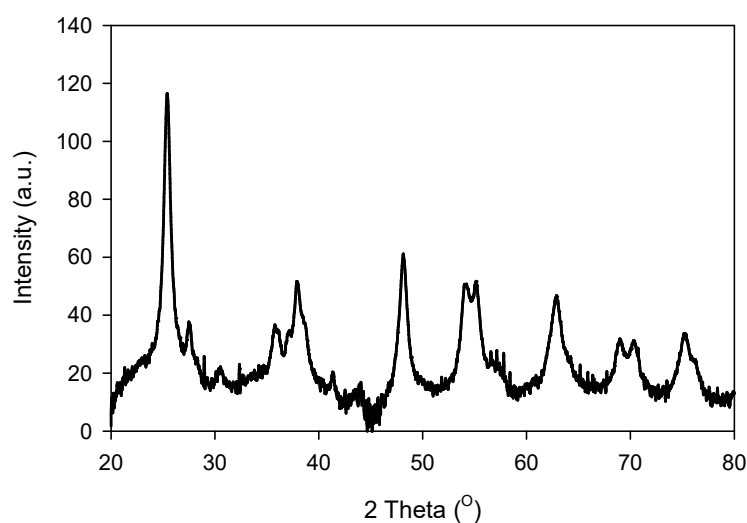


Fig. S5. XRD pattern of the synthesised $\text{Fe}_3\text{O}_4\text{-SiO}_2\text{-TiO}_2$ catalyst.

Fig. S5 shows XRD patterns of the $\text{Fe}_3\text{O}_4\text{-SiO}_2\text{-TiO}_2$ catalyst and the results revealed a set of different sharp peaks related to the conforming compounds, typical of a high crystalline structure. Indeed, peaks at 38.0° , 48.2° , 54.6° , 55.2° , 63.0° , 68.8° , 70.5° and 75.5° are related to the tetragonal structure of anatase TiO_2 (Bekena & Kuo, 2020). Moreover, the small peaks at 30.6° and 57.0° as well as the peak at 35.8° correspond to pure Fe_3O_4 (Lei et al., 2020). Moreover, the broad peak at 63.0° can be related not only to TiO_2 but also to Fe_3O_4 (Bekena & Kuo, 2020; Lei et al., 2020). SiO_2 has an amorphous structure although some small peaks found at $22\text{-}23^\circ$ have been related to its presence (Mohamed & Harraz, 2020).

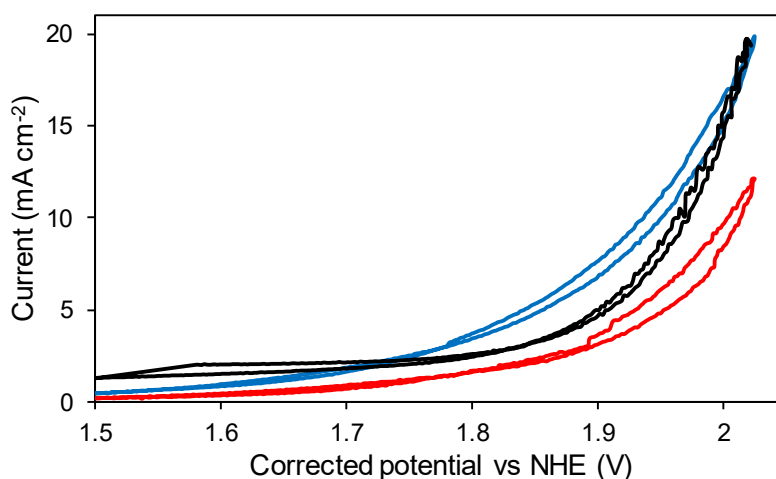


Fig. S6. Cyclic voltammetry of TiO_2 (red), $\text{Fe}_3\text{O}_4\text{-SiO}_2\text{-TiO}_2$ (blue) and Fe_3O_4 (black) catalysts.