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Regiodivergent Organocatalytic Reactions

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Abstract: Organocatalysts are abundantly used for various transformations, particularly to obtain highly enantio- and diastereomeric pure products by controlling the stereochemistry. These applications of organocatalysts have been the topic of several reviews. Organocatalysts have emerged as one of the very essential areas of research due to their mild reaction conditions, cost-effective nature, non-toxicity, and environmentally benign approach that obviates the need for transition metal catalysts and other toxic reagents. Various types of organocatalysts including amine catalysts, Brønsted acids, and Lewis bases such as N-heterocyclic carbene (NHC) catalysts, cinchona alkaloids, 4-dimethylaminopyridine (DMAP), and hydrogen bond-donating catalysts, have gained renewed interest because of their regioselectivity. In this review, we present recent advances in regiodivergent reactions that are governed by organocatalysts. Additionally, we briefly discuss the reaction pathways of achieving regiodivergent products by changes in conditions such as solvents, additives, or the temperature.

Keywords: organocatalysts; regiodivergent; metal-free; Lewis base; NHC; amine catalyst; Brønsted acid; hydrogen bond-donating catalysts; solvent control; temperature control

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1. Introduction

Over the past two decades, reactions that rely on organocatalysts have emerged as important catalytic systems in asymmetric and conventional synthesis [1–14]. The utilization of organocatalysts has garnered interest because they are robust, inexpensive, environmentally benign, and easily recoverable, among other advantageous characteristics. Moreover, the use of chiral organic molecules as catalysts enables the synthesis of highly enantio- and diastereomeric pure products, which are of great importance in medicinal and pharmaceutical chemistry [15–18].

Controlling the selectivity of the reactions is one of the popular fields of the research area in synthetic organic chemistry. Organocatalysts have made it possible to develop a large number of reactions to synthesize stereoselective [19–29], regioselective [30–40], and chemo-selective [41–48] products. Regiodivergent synthesis reactions, which enable two or more regioisomeric products to be synthesized from the same starting material, are controlled by various reaction parameters such as catalysts, additives, solvents, temperatures, ligands, and functional groups [49,50]. Although several metal-catalyzed regiodivergent reactions have been thoroughly studied [51–55], organocatalytic regiodivergent reactions is intended to fill this void.

Catalysts **2021**, 11, 1013 2 of 47

2. Lewis Base Catalysts

Lewis base catalysts, including various tertiary amines (1,8-diazabicyclo[5.4.0]undec-7-ene (DBU), 4-dimethylaminopyridine (DMAP), 1,4-diazabicyclo[2.2.2]octane (DABCO), etc.), cinchona alkaloids, and *N*-heterocyclic carbene (NHC) catalysts, are widely utilized for asymmetric transformations such as the aldol reaction, the Morita–Baylis–Hillman reaction, and cycloaddition. One primary application of this type of catalyst is formal cycloaddition reactions, in which this catalyst type combines with ketenes or alkynes to form in situ generated zwitterion. These intermediates subsequently undergo a cycloaddition reaction with several electrophilic moieties to yield the cyclized products. Consequently, various types of cycloaddition reactions have been developed such as [2+2], [2+3], and [2+4] [57–66].

2.1. Phosphine and Amine Bases

Recently, nucleophilic phosphine- and nitrogen-containing Lewis bases have emerged as a powerful tool for constructing carbo- and heterocyclic compounds under metal-free and mild reaction conditions. Several natural products and spirocyclic compounds were synthesized by using these catalysts via a cycloaddition reaction with allenoates or alkynones. In general, these reactions occur by way of cycloaddition of the allenoates with electron-deficient olefins or imines via [3+2] and [4+2] cycloadditions [57–66].

Shi's group developed a highly regioselective [3+2] cycloaddition reaction by using phosphine as a Lewis base, which afforded five-membered spiro compounds (Scheme 1) [67]. The cycloaddition reaction was conducted between an α -allenic ester (1-2) and α,β -unsaturated diesters (1-1) which derived from isatin, in the presence of PBu₃, and the reaction proceeded smoothly by way of a [3+2] cycloaddition to yield 1-3 and 1-4 in >20:1 regioselectivity. With different electron-donating groups (EDGs), or electron-withdrawing groups (EWGs), at the fifth, sixth, or seventh positions, the reaction proceeded without complication to furnish the products in good yields. In contrast, the DMAP-catalyzed reactions initiated [4+2] cycloadditions to yield the six-membered dihydropyranone products 1-5. Various isatins having different EDGs and EWGs on the benzene rings and different *N*-protecting groups underwent this cycloaddition reaction depending on the catalyst used to afford the desired cyclic products in good yield with good geometric selectivities.

Scheme 1. Lewis base-catalyzed [3+2] and [4+2] cycloaddition reactions.

Catalysts 2021, 11, 1013 3 of 47

The proposed reaction mechanism is depicted in Scheme 2. The reaction commences with the addition of a phosphine catalyst to the allenic ester (1-2) to produce a zwitterionic intermediate. The intermediate serves as a dipole for the subsequent [3+2] cycloaddition, which occurs at the C-3 position of isatin to produce intermediate C. Subsequently, 1,2-proton transfers followed by regeneration of the catalyst afford product 1-3a. In the case of DMAP, a zwitterionic intermediate forms by the addition of the base DMAP to the allenic ester, which reacts with 1-1a to produce intermediate D. Consecutive enolization, followed by cyclization and elimination of the catalyst, delivers the six-membered dihydropyranone 1-5a products.

Scheme 2. Proposed mechanism for [3+2] and [4+2] cycloaddition reactions.

In their continuous efforts in this study, Shi et al. demonstrated different cycloaddition reactions in which different regiodivergent spiro compounds were obtained depending on the Lewis bases involved in the reaction (Scheme 3) [68]. In the presence of nitrogen-containing base DABCO, isatin (2-1) reacts with butynone (2-2) to produce six-membered spiro compounds, the pyranones (2-3). With isatins bearing either EWG or EDG substituents at various positions on the benzene ring, the reaction proceeded smoothly to afford the products in good yield. On the other hand, the use of PPh2Me as a Lewis base promoted the formation of the five-membered spiro compound furanone (2-4). Optimized conditions applied with various EWG and EDG groups on the benzene ring had no influence on the reaction yields. Various protecting groups on nitrogen underwent the reaction smoothly to yield five-membered oxygen-containing spiro compounds in moderate to good yields. Focusing on the mechanistic study, initial deprotonation of butynone generates an enolate intermediate in the presence of DABCO (Scheme 4). Later, nucleophilic addition is followed by an intramolecular Michael addition of an O⁻ anion to the alkynyl group, and protonation to yield the required six-membered pyranone product (2-3). Using PPh₂Me with butynone initially generates a zwitterionic intermediate (C), which undergoes a 1,3-hydrogen shift to form an enolate intermediate (D). Nucleophilic addition to the carbonyl group followed by intramolecular addition of an O⁻ anion (E) to the alkenyl group creates the desired five-membered spiro products (2-4) by the liberation of the catalyst (Scheme 5).

Catalysts 2021, 11, 1013 4 of 47

Scheme 3. Synthesis of spiro-cyclohexaneoxindole and spiro-cyclopenteneoxindole.

 $Scheme \ 4. \ Plausible \ mechanism \ for \ the \ synthesis \ of \ pyranones \ (2-3).$

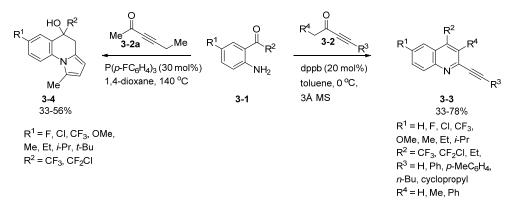
Scheme 5. Plausible mechanism for the formation of furanones (2-4).

Catalysts **2021**, 11, 1013 5 of 47

In 2019, Shi's group further explored a phosphine-catalyzed intermolecular annulation between *ortho*-aminoacetophenones and alkynones using two different regioselective approaches: [4+2] or [3+2] cycloaddition reactions (Scheme 6) [69]. Switchable transformations were achieved using different phosphine catalysts and temperatures.

Reaction between *ortho*-aminoacetophenones (**3-1**) and alkynones (**3-2**) in toluene at 0 °C in the presence of a bisphosphine catalyst such as 1,4-bis(diphenylphosphino)butane (dppb) delivered 2-alkynylquinolines (**3-3**) via a [4+2] cycloaddition reaction. This is because the lower temperatures favor the addition of phosphine to the alkynone to form zwitterionic intermediates, which then undergo an α , α -H shift, enolization, and proton abstraction to produce the α' -carbanionic species II" (Scheme 7). This intermediate reacts with *ortho*-aminoacetophenones followed by protonation, dehydration, and intramolecular condensation to afford the quinoline products (**3-3**). This methodology exhibited a broad substrate scope that efficiently proceeded with both electron-deficient and electronrich aromatic rings to afford the products in good yields.

Alternatively, using dioxane as the solvent and replacing the catalyst with $P(p-FC_6H_4)_3$ at an elevated temperature of 140 °C, the reaction proceeded via a [3+2] cycloaddition to provide the benzoindalizines (3-4). The optimized reaction conditions were affective, and the electronic factors did not have any significant impact on the reaction, which proceeded smoothly to produce the expected products in moderate yields. At a higher temperature, the zwitterionic intermediate initially underwent an α, γ -H shift followed by a proton shift to produce the δ -activated intermediate VII, which then reacted with *ortho*-aminoacetophenones followed by intramolecular proton migration, α, β -H-shift, and regeneration of the catalyst to afford the compound 3-5. Finally, the benzoindalizines (3-4) were obtained from compound 3-5 via the Knorr reaction methodology (Scheme 7). Both of the regioisomers were unambiguously confirmed by X-ray crystallography.

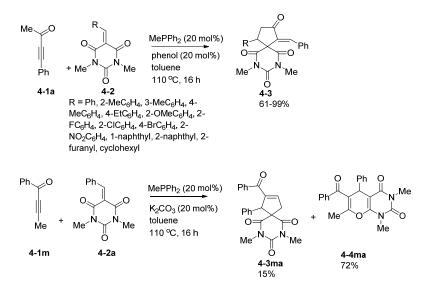


Scheme 6. Phosphine-catalyzed switchable [4+2] or [4+2]/[3+2] cycloaddition.

Catalysts **2021**, 11, 1013 6 of 47

Scheme 7. Proposed mechanism for the intermolecular annulation reaction.

Guo and co-workers illustrated that a [3+2] annulation reaction between barbiturate-derived alkenes (4-2) and ynone (4-1a) would offer spirobarbiturate-cyclopentanones (4-3) in the presence of MePPh2 using phenol as an acid additive (Scheme 8) [70]. Ynones containing alkyl, MeO, F, and Cl substituents were compatible with alkenes, producing the expected products in good to excellent yields, with excellent *E/Z* stereoselectivities. Similarly, the barbiturate-derived alkenes bearing various substituents including alkyl, OMe, F, Cl, Br, CF3, CN, and NO2 groups afforded the products in good to excellent yields. In addition to that, 1-naphthyl-, 2-naphthyl-, and 2-furanyl-derived barbiturates delivered spirobarbiturate-cyclopentanones in excellent yields. The reaction between the ynone and barbiturate-derived alkenes in the presence of MePPh2 with an inorganic base additive, such as K2CO3, involved [4+2] annulation to deliver 1,5-dihydro-2*H*-pyrano[2,3-*d*]pyrimidine-2,4(3*H*)-dione (4-4ma) as a major product along with a minor [3+2] cycloadduct. In the absence of the phosphine catalyst, no [4+2] cycloaddition product was obtained which clearly indicates that this reaction did not occur via the hetero-Diels–Alder reaction pathway.



Scheme 8. Phosphine-catalyzed [3+2] and [4+2] annulation reactions.

Catalysts 2021, 11, 1013 7 of 47

A possible reaction mechanism is presented in Scheme 9. The phosphine catalyst initally attacks the ynone (**4-1a**) to produce intermediate **A**. In a [3+2] annulation, intermediate **A** undergoes a proton shift to produce intermediate **B**. This is followed by a conjugate addition with a barbiturate-derived alkene followed by intramolecular nucleophilic addition of the carbanion (intermediate **C**) to the double bond, which delivers the cyclic intermediate **D**. The acid additive phenol promotes 1,2-proton migration followed by regeneration of the catalyst which yields the spirobarbiturate-cyclopentanone **4-3**. In a [4+2] annulation, intermediate **A** is stabilized by the base K_2CO_3 , which produces intermediate **F** by the conjugate addition with the barbiturate-derived alkene. Intermediate **F** undergoes subsequent enolization (**G**), intramolecular oxa-Michael addition, and elimination of the phosphine catalyst to furnish 1,5-dihydro-2*H*-pyrano[2,3-*d*]pyrimidine-2,4(3*H*)-dione (**4-4**).

Scheme 9. Plausible mechanism for phosphine-catalyzed [3+2] and [4+2] annulation reactions.

Christmann and colleagues described the catalysis of the bromocyclization/regiodivergent reaction of alkenes (5-1a-l) by chiral phosphoric acids to afford both bromohydrin products in excellent yields and with good enantioselectivities (Scheme 10) [71]. More specifically, the chiral phosphoric acids containing 9-anthracenyl (5-C5) delivered the best results to produce both the regioisomers 5-2 and 5-3 in excellent yields with enantioselectivities. A cinnamyl ester containing an EWG or EDG in the *para* or *meta* position produced the two corresponding isomers in excellent yields with high enantioselectivities. Derivatives with sterically bulky groups such as phenyl, and 1-naphthyl derivatives at the *ortho* position smoothly reacted to afford the expected isomers in excellent yields together with good enantioselectivities. Various homoallylic esters (5-1m-o) were utilized to produce 5-2m-o with excellent enantioselectivities, whereas 5-3m-o was obtained with lower enantioselectivities.

Catalysts **2021**, 11, 1013 8 of 47

Scheme 10. Enantioenriched bromohydrin synthesis by anchimeric oxygen borrowing.

A possible mechanism for this reaction is proposed as follows (Scheme 11). Initially, the chiral phosphoric acid activates NBS for halocyclization followed by nucleophilic attack of water, affording an oxocarbenium ion, which results in both of the cyclic hemiorthoesters (INT and *ent*-INT) having excellent diastereoselectivity. At this stage, chiral phosphoric acid then collapses these intermediates for regiodivergent RRM (reaction of a racemic mixture) to afford both constitutional isomers. The authors expected that the catalyst would activate different oxygen atoms to form the respective enantiomers. INT may collapse via pathway **C** which results in *ent*-5-3, which may be the reason for the lower enantioselectivity of the 5-3 isomer.

Scheme 11. Proposed mechanism for the synthesis of enantioenriched bromohydrins.

In 2016, Lu's group reported phosphine-catalyzed regiodivergent C-2- and C-4-selective γ -additions of oxazolones to 2,3- butadienoates (Scheme 12) [72]. Grafting suitable substituents on the oxazolones enabled the asymmetric C-2- and C-4-selective γ -additions of oxazolones to 2,3-butadienoates to be accomplished with excellent yields of 81 to 99% and with admirable enantioselectivities (as high as 96%) for a broad range of substrates. The C-4-selective γ -addition of oxazolones produced (6-3) in a highly enantioselective manner when 2-aryl-4-alkyloxazol-5-(4H)-ones (6-2) were employed as pronucleophiles,

Catalysts 2021, 11, 1013 9 of 47

furnishing enantioenriched α , α -disubstituted α -amino acid derivatives. The employment of the 2-alkyl-4- aryloxazol-5-(4H)-ones (6-4) as the donors resulted in the C-2-selective γ -addition of oxazolones (6-5) in a highly enantioselective manner which led to the facile synthesis of chiral N,O-acetals, and γ -lactols.

Scheme 12. Phosphine-catalyzed regiodivergent enantioselective C-2 and C-4 γ -additions.

The proposed mechanistic pathway involves the formation of zwitterionic intermediate **A** by the addition of catalyst **6-6c** to **6-1c**, which then abstracts the C-4-proton of 2-phenyl-4-methyloxazol-5(4*H*)-one **6-2a**′ (Scheme 13). Further oxazolide **C1** reacts with phosphonium **B** at the C-4 position of **C1** via the transition state **TS-C4**, followed by a hydride shift to generate addition product **6-3a**′ by the elimination of the catalyst. In contrast, the use of 2-methyl-4- phenyloxazol-5(4*H*)-one **6-4a** generates the oxazolide **C2**, which favors the C-2-selective addition to eventually lead to the formation of C-2-selective product **6-5a**, via a key transition state, **TS-C2**. Theoretical studies (DFT calculations) suggested that the origin of the observed regioselectivity was the distortion energy that resulted from the interaction between the nucleophilic oxazolones and the electrophilic phosphonium intermediate.

Scheme 13. Proposed mechanism for γ -additions of oxazolone.

Recently, in 2020, Lin's group developed a regiodivergent cascade reaction between 3-homoacylcoumarin (7-1) and the 1,3-indanedione-derived 1,6-acceptor (7-2) to construct spirocyclohexene indane-1,3-diones (7-3) and coumarin-fused cyclopentanes (7-4) catalyzed by bases such as DMAP or Et₃N, respectively (Scheme 14) [73]. In the presence of

Catalysts 2021, 11, 1013 10 of 47

DMAP as the base, the reaction afforded spirocyclohexene indane-1,3-diones (7-3) via 1,6-addition followed by a regio- and chemoselective aldol cascade reaction. Electronic factors did not influence the yields when 3-homoacylcoumarin containing various group such as 6-Cl, 6-Br, and 6-methoxy was used, and all derivatives afforded the expected products in good yields. In the case of the indanedione-derived acceptors, EWGs provided the products in better yields than the methoxy group regardless of its position. Fused cyclopentanes (7-4) were obtained via 1,6-addition followed by a regio- and chemoselective vinylogous Michael addition by using Et₃N. The presence of an EWG on coumarins ensured good yields compared with an EDG irrespective of its position. This is because the EDG on coumarin enhanced the electron density at the reactive site to prevent an intramolecular Michael addition. Moreover, in the case of indanedione, sterically bulky EWGs afforded the products in good yields, whereas no product formed in the case of an EDG (OMe).

Scheme 14. Synthesis of spiro and fused systems via organobase-controlled cascade reaction.

Based on the experimental results, the authors proposed a plausible mechanistic pathway (Scheme 15). Initially, in the presence of organocatalytic bases (DMAP or Et₃N), 1,6-addition occurs between 3-homoacylcoumarin 7-1 and the indanedione 7-2 to provide a common intermediate, **A**. The conjugate acid formed by deprotonation of intermediate **A** by DMAP interacts with the dienolate to form intermediate **B**, which undergoes an intramolecular aldol reaction to deliver the desired spirocyclic product 7-3. On the other hand, exposure of the common intermediate **A** to Et₃N forms the dienolate, which then coordinates with the respective conjugate acid to yield intermediate **C**. Finally, the intramolecular vinylogous Michael addition results in the regio- and chemoselective coumarin-fused cyclopentane 7-4. Although the authors did not provide mechanistic proof, the selectivity was attributed to the different hydrogen bonding interactions complemented by the steric interactions with the dienolate intermediate. These interactions give rise to different transition states which result in the formation of divergent products.

Catalysts **2021**, 11, 1013

Scheme 15. Plausible mechanism for organobase-controlled regiodivergent cascade reaction.

Shi and co-workers developed regioselective trifluoromethylthiolation between Morita–Baylis–Hillman (MBH) carbonates (8-2) and Zard's trifluoromethylthiolation reagent (8-1a) in the presence of DABCO (Scheme 16) [74]. A primary allylic trifluoromethylthiolate (SCF₃) was obtained as the favored product (8-3) when the authors used tetrahydrofuran as the solvent at room temperature, whereas when using chloroform at 0 °C, this produced a secondary allylic trifluoromethylthiolate as a major product (8-4). The optimized reaction conditions with THF enabled the reaction to smoothly proceed with various MBH carbonates containing 4-Cl, 4-CN, 4-Br, and heterocyclic compounds such as 2-thienyl, 2-furyl, and 2-pyridyl to afford the primary allylic trifluoromethylthiolation products in good yields with good regioselectivities. Similarly, MBH carbonates containing 4-MeO, 3-Me, 4-NO₂, and 2-pyridyl produced a secondary allylic trifluoromethylthiolate in the CHCl₃ solvent with good yields and high selectivities.

Scheme 16. Solvent-controlled nucleophilic trifluoromethylthiolation of MBH carbonates.

In the mechanism depicted in Scheme 17, this reaction proceeded with an initial nucleophilic addition of DABCO to **8-1a** and **8-2a**, forming the ammonium salt intermediates **A** and **B**, respectively. Then, the exchange of SCF₃- and ¹BuO- afforded the other intermediates **C** and **D**. The secondary allylic trifluoromethylthiolation product **8-4a** was obtained

Catalysts 2021, 11, 1013 12 of 47

by an intermolecular S_N2′ reaction of intermediate **D**. In the presence of the catalyst DABCO, the secondary product was readily converted to the primary product in THF, whereas the conversion in chloroform was difficult. This is attributed to electrostatic interaction between [¬]SCF₃ and CHCl₃, weakening the nucleophilicity of the SCF₃ anion. The catalyst DABCO was regenerated by the nucleophilic addition of water to intermediate **C**.

Scheme 17. Plausible reaction mechanism for trifluoromethylthiolation.

A divergent strategy was developed by Liu and Chen et al. for the modular synthesis of various enantioenriched phenylthio-substituted lactones from the thiolation of homoallylic acids via regiodivergent cyclization (Scheme 18) [75]. The authors developed Lewis base/Brønsted acid co-catalyst-controlled regio- and enantioselective thiolactonizations of a variety of homoallylic acid derivatives with different electrophilic SAr reagents (6-endo vs. 5-exo). The homoallylic acid (9-1) underwent 6-endo cyclization using N-phenyl thiosaccharin (9-2) as the sulfenylating agent, and chiral BINOL-derived selenide ((S)-9-5a) as the Lewis base. Various styrene-based carboxylic acids afforded the products (9-3) in excellent yields with high enantioselectivity, which was affected by the position of the substrate. For instance, a fluorine substituent in the para position resulted in high enantioselectivity compared with the meta or ortho positions. Moreover, 2-naphthyl, 2-thienyl, 3furyl, and various substituted ethynylbenzenes underwent this reaction to afford δvalerolactones in good yield with good enantioselectivity. In the presence of 1.0 equivalent of EtSO3H using N-phenylthiotphthalimide as the sulfenylating agent, and chiral BINOL-derived selenide as the Lewis base, homoallylic acids afforded various γ -butyrolactones (9-4). Styrene derivatives with various functional groups, irrespective of their position, underwent 5-exo cyclization to yield the corresponding five-membered ring products in good yields with high enantioselectivities. 2-Naphthyl and various unbiased alkylsubstituted alkenes proceeded smoothly to afford the products in good yields with high Catalysts 2021, 11, 1013 13 of 47

enantioselectivities, whereas an ethynylbenzene-substituted alkene afforded the product in poor yield with moderate enantioselectivity.

Scheme 18. Acid-controlled asymmetric thiolation of alkenes.

These researchers conducted experimental and computational studies to elucidate the origins of the regio- and enantioselectivity. The results of kinetic control experiments to acquire mechanistic information suggested that the 6-endo product (9-3a) could isomerize into a thermodynamic 5-exo product (9-4a) via the configurationally stable thiiranium intermediate under strongly acidic conditions (Scheme 19), which was further supported by the reaction between 6-endo and 5-exo with 100 mol% of EtSO₃H to afford a 5-exo product with retention of ee. The combination of DFT calculation results suggested that C-O and C-S bond formation might occur simultaneously, without formation of a commonly supposed catalyst-coordinated thiiranium ion intermediate. The potential π - π stacking between the substrate and SPh is an important factor in the enantio-determining step.

Scheme 19. Isomerization process under acidic conditions.

In 2017, Lu *et al.* developed the catalyst-controlled regioselective synthesis of spirocyclic benzofuranones via regiodivergent [3+2] annulations of aurones and allenoates (Scheme 20) [76]. The use of the L-thr-D-thr-derived chiral phosphine catalyst **10-5g** in an ether in an annulation reaction produced the α -isomer **10-3** in moderate to good yield with excellent enantioselectivities. On the other hand, the L-thr-L-thr-derived chiral phosphine catalyst **10-5b** yielded the γ -isomer **10-4** in good yield with high enantioselectivities (96–99%). Under the optimized reaction conditions, various substituted aurones afforded α -or γ -selective spiro benzofuranones with excellent enantio- and regioselectivities depending on the catalyst present.

Catalysts **2021**, 11, 1013

Scheme 20. Catalyst-controlled synthesis of spirocyclic benzofuranones.

Mechanistic studies suggested that the phosphine catalyst attacked the allene (10-2) to form zwitterionic intermediate $\bf B$, in which the negative charge may be delocalized either on the α -carbon or the γ -carbon (Scheme 21). Then, the aurone (10-1a) underwent [3+2] annulation with the putative intermediates, delivering intermediates $\bf E$ or $\bf I$. Proton transfer followed by elimination of the phosphine catalyst furnished the α - and γ -selective products.

Scheme 21. Plausible mechanism for phosphine-catalyzed [3+2] annulation of aurones with allenoates.

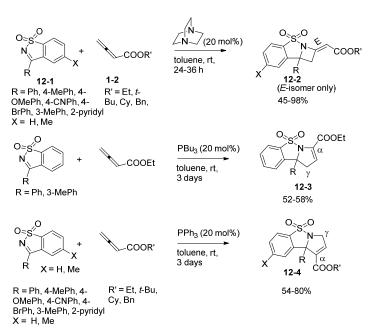
Cahard *et al.* reported the synthesis of primary and secondary allylic SCF₃ compounds in the presence of DABCO with Morita–Baylis–Hillman (MBH) carbonates (Scheme 22) [77]. The combination of CF₃SiMe₃/S₈/KF in DMF as the solvent afforded the primary product (11-2). Regardless of whether an EWG or EDG was present on the MBH carbonate, the primary allylic SCF₃ products formed in excellent yields. Sterically bulky groups such as 1- and 2- naphthyls and 2-thienyl were well tolerated in this trifluoromethyl thiolation reaction, furnishing the equivalent products in good yields. On the other hand, Zard's reagent (CF₃SCO₂C₁₈H₃₇) afforded the secondary allylic SCF₃ product (11-3i) when the reaction was conducted in the THF solvent at room temperature. The authors expected that the base DABCO would activate both Zard's reagent and the MBH carbonate and provide the secondary allylic trifluoromethyl thiolation product 11-3i (kinetic

Catalysts 2021, 11, 1013 15 of 47

product) within 5 minutes. Upon extension of the reaction time to 30 min, the kinetic product (secondary) was rapidly isomerized into a thermodynamic product (**11-2i**, primary allylic trifluoromethyl thiolation product), as monitored by ¹⁹F NMR.

Scheme 22. Regio- and stereo-controlled nucleophilic trifluoromethylthiolation of MBH carbonates.

Ye's group established sultam-fused azetidines and dihydropyrroles via two different cycloadditions ([2+2] and [3+2]) from cyclic sulfonamide ketimines (12-1) and allenoates (1-2). These compounds are formed by involving Lewis bases in the reaction (Scheme 23) [78]. In the toluene solvent at room temperature, PPh₃, as a catalyst, underwent a [3+2] cycloaddition to produce 12-4 as the product via α -addition. The regioselectivity was switched in the case of PBu₃, which led to a γ -cycloadduct (12-3). A completely different cycloaddition product was formed with the DABCO catalyst, delivering a [2+2] cycloadduct (12-2). Ketimines with an EDG or EDG worked well. Similarly, various allenoates were also found to be suitable under optimized conditions.



Scheme 23. Lewis base-catalyzed [2+2] and [3+2] cycloaddition reactions.

Catalysts 2021, 11, 1013 16 of 47

The authors also proposed a reaction mechanism (Scheme 24). They suggested that the reaction was initiated by adding Lewis bases to the allenoate (1-2) to generate two zwitterionic intermediates, **A** and **A'**, which react with the cyclicimines (12-1) to form intermediate **B** or **B'**. The carbanion **A'** was stabilized by the electron-poor nucleophile PPh₃ which then produced the thermodynamically favored α -addition intermediate **B'**, and elimination of the catalyst delivered 12-4. In the case of DABCO and PBu₃, which are relatively electron-rich nucleophiles, they provided kinetically favored intermediate **B** via γ -addition. Later, these intermediates underwent ring closure, followed by the release of the catalysts, to afford the expected cycloaddition products (12-2 and 12-3).

Scheme 24. Plausible catalytic cycles for cycloadditions.

Zhong and co-workers reported [3+2] annulation between γ -substituted allenoates (13-1) and unsaturated pyrazolones (13-2) to furnish spirocyclopentene-pyrazolones (13-3) when the reaction was performed in PPh₃ and K₂CO₃ (Scheme 25) [79]. In terms of the scope of the substrates, pyrazolones with an aryl ring bearing an EDG at the *para* position afforded higher yields than those bearing EWGs. Similarly, halogens such as Cl and Br, and 1- and 2-naphthalenes were compatible with the substrate to afford products. Allenoates containing *tert*-butyl instead of ethyl delivered spirocyclopentene-pyrazolones in lower yield owing to the steric hindrance.

Catalysts **2021**, 11, 1013 17 of 47

Scheme 25. Lewis base-controlled [3+2] and [4+2] annulation reactions.

On the other hand, in the presence of DBU as a base, pyrano[2,3-*c*]pyrazoles (13-4) were obtained via [4+2] annulations. Pyrazolones containing F, Cl, Br, Me, ^{*i*}Bu, ^{*i*}Bu, 2-naphthyl, and 2-thienyl all reacted smoothly to afford pyrano[2,3-*c*]pyrazoles in good to excellent yields. With regard to the mechanism, the Lewis base catalyst attacks the allenoate to afford zwitterionic intermediate **I**, which then α-attacks the pyrazolones to form intermediate **II** via 1,4-addition (Scheme 26). The reaction can proceed along path **A**, in which case an intramolecular Michael addition takes place in the presence of PPh₃ and K₂CO₃ to afford intermediate **III**. Next, proton transfer and regeneration of the catalyst (PPh₃) furnish the [3+2] annulated product spirocyclopentene-pyrazolone (13-3). Path **B** involves elimination of the catalyst DBU to afford intermediate **II''** followed by an *O*-Michael addition to provide intermediate **IV**. Subsequently, 1,3-proton transfer of intermediate **IV** delivers the [4+2] annulated product pyrano[2,3-*c*]pyrazoles (13-4).

Scheme 26. Possible reaction mechanism for Lewis base-controlled annulation reactions.

In 2019, Sun *et al.* developed a regiodivergent allylation of *N*-acylhydrazones (NAHs, **14-1**) with Morita–Baylis–Hillman (MBH) carbonates (**14-2**), selectively affording α (**14-4**)-

Catalysts 2021, 11, 1013 18 of 47

or γ (14-3)-allylated products (Scheme 27) [80]. The regioselectivity of the above methodology was precisely regulated by an expedient alternation of the catalysts to afford α - and γ -allylated N-acylhydrazone derivatives selectively in excellent yields.

Scheme 27. Proposed reaction pathways for base-promoted regiodivergent allylation.

The authors screened a wide range of base catalysts and identified tBuOK and DABCO as the optimal catalysts to promote the formation of α - or γ -allylated products, respectively. In DABCO, the optimized conditions were compatible with a broad range of MBH carbonates having various EWGs and EDGs either at the *ortho* or *para* position of the phenyl ring and were tolerated. Similarly, NAHs having various functional groups in their aryl ring including Me, Cl, MeO, and F all afforded γ -allylated products (14-3) in good to excellent yields. On the other hand, due to the strong electron-withdrawing nature of the nitro group, it afforded the product, albeit in a lower yield. In a similar fashion, the substrate scope of tBuOK -catalyzed α -allylation was explored (14-4). Various electron-donating and withdrawing groups were incorporated in both MBH carbonates and NAHs. The electronic effect or the bulkiness of the substituents did not affect the efficiency of the α -allylation, affording the products in good to excellent yields. However, an MBH carbonate derived from aliphatic aldehydes afforded the corresponding allylated product in the DABCO base, whereas this failed to occur in tBuOK .

In the proposed mechanism, the regiodivergent allylation proceeds through the key step involving deprotonation of N-Acylhydrazone (14-1) by 'BuO- to produce nucleophilic intermediate II (Scheme 28). Intermediate II participates in the further reaction divergently in the presence of different catalysts to yield either α - or γ -allylated products. In path a, the attack of the DABCO catalyst on the α -position of the MBH carbonate (14-2) in an SN'2 fashion affords intermediate I. The 'BuO- released in due course subsequently deprotonates 14-1, leading to intermediate II. The key intermediate II reacts with intermediate I via the SN'2 pathway to produce the γ -allylated products (14-3). When 'BuOK is used as a catalyst, intermediate II approaches the α -position of 14-2, leading to the α -allylated products (14-4) through the SN'2 pathway (path b).

Catalysts 2021, 11, 1013 19 of 47

Scheme 28. Proposed reaction pathways for base-promoted regiodivergent allylation.

2.2. Cinchona Alkaloids

Both naturally occurring and modified cinchona alkaloids (quinine, quinidine, cinchonidine, cinchonidine, cinchonine) are widely used in asymmetric synthesis. These alkaloids offer various important features such as numerous chiral centers, structural rigidity, multiple donors in the form of hydrogen bonds, and facile conversion into different functional groups including chiral quaternary ammonium salts [81–90]. Apart from this, they have various applications such as utilization in chiral ligands in the preparation of metal complexes, in NMR as chiral agents, resolving agents, chiral stationary phase in HPLC, electrolytic additives, and chiral solvating agents. They have been successfully used in various important asymmetric transformations such as the Mannich reaction [91–93], Michael addition [94–96], aza-Henry reactions [97,98], and epoxidation [99,100], in order to promote a highly enantio- and diastereoselective outcome.

Cheng et al. developed a Lewis base-catalyzed cycloaddition between allene ketones or α -methyl allene ketones and pyrazolones to produce tetrahydropyrano[2,3-c] pyrazoles in moderate to good yields via a [4+2] cycloaddition (Scheme 29) [101]. The annulation of benzylidenepyrazolones (15-2) with allene ketones (15-1) proceeded smoothly via either an α - or γ -selective pathway, and the desired products were obtained in good yields with high regioselectivities. The use of quinine as the catalyst favored the formation of an α adduct (15-3) with high regioselectivity in a 99% yield. After optimizing the conditions, the authors examined various substrates (neutral groups, EWGs, and EDGs as substituents at the ortho, meta, or para position on benzylidene pyrazolones) and found that they are capable of delivering the expected products in good yields with excellent regioselectivities. Interestingly, pyrazolone containing α -naphthyl, β -naphthyl, and 2-furyl groups reacted smoothly and furnished the anticipated product in good yield with high regioselectivity. On the other hand, DMAP produced the γ -selective cycloaddition products as the major regioisomers in good yield (15-4). This γ -selective [4+2] annulation of various substrates with DMAP was then investigated. Pyrazolone with α -naphthyl-, β -naphthyl-, 2-thienyl-, and 2-furyl-containing substrates efficiently reacted under the standard conditions to produce the expected products in good yield with high regioselectivity.

Catalysts **2021**, 11, 1013 20 of 47

Scheme 29. Lewis base-catalyzed regioselective [4+2] cycloaddition.

According to the proposed reaction mechanism (Scheme 30), first, the Lewis bases undergo addition with the allene ketones (15-1) to generate zwitterionic intermediates, which then undergo nucleophilic addition with the unsaturated pyrazolones (15-2) to form intermediates $\bf C$ and $\bf D$ via α - or γ -addition. An electron-poor nucleophile such as quinine may stabilize carbanion $\bf A$ and lead to a thermodynamically feasible α -addition, whereas a kinetically favored γ -addition could occur at carbanion $\bf B$ in the case of the electron-rich nucleophile DMAP. Further, a proton shift and subsequent ring closure of the intermediates via an oxygen anion or in reverse mode would then generate the cyclic adducts $\bf F$ and $\bf H$, respectively. These adducts result in the formation of regiodivergent products after elimination of the Lewis bases from the cyclic adducts.

Scheme 30. Proposed reaction mechanism of cycloaddition catalyzed by Lewis bases.

A regiodivergent 1,3-dipolar cycloaddition of azomethine ylides (16-1) and 2-hydoxybenzylidene indandiones (16-2) was developed by Lin *et al.* in 2018 (Scheme 31)

Catalysts **2021**, 11, 1013 21 of 47

[102]. The (3+2) cycloaddition, which involved the reversal of the nucleophilic site in azomethine ylides, was controlled by choosing suitable base catalysts, DMAP and 1,1,3,3-tetramethylguanidine (TMG), which subsequently resulted in two different cascade processes to generate the diverse chromenopyrrolidines 16-3 and 16-4, respectively. The azomethine ylide was stabilized by the conjugate acids of the bases in two different conformations via hydrogen bonding, which afforded regiodivergent (3+2) cycloadditions. Subsequent cyclization delivered the above products in moderate to good yields (as high as 84%) and with excellent diastereo- and enantioselectivity (as high as 96%).

NMe2
$$(3+2)$$
 cycloaddition/ lactonization chloroform, $(3+2)$ cycloaddition/ lactonization $(3+2)$ cycloaddition/ lactonization $(3+2)$ cycloaddition/ $(3+2)$ cycloaddition/ lactonization $(3+2)$ c

Scheme 31. Base-controlled regiodivergent [3+2] cycloaddition.

According to the plausible mechanistic pathway (Scheme 32), initially, the iminodiester (16-1a) is deprotonated in the presence of bases to form the equivalent conjugate acids, which then subsequently participate in hydrogen bonding with the azomethine ylide. The use of the electrophile 2-hydroxybenzylidene indan-1,3-dione (16-2a) introduces steric hindrance and leads to two different transition states. In the presence of DMAP as the base, a [3+2] cycloaddition followed by cascade lactonization affords the expected product, chromeno[3,4-b]pyrrolidines (16-3a). The unanticipated chromeno[3,4-c]pyrrolidine adduct 16-4a is obtained when TMG is used as the base. This is the consequence of the opposite regioselectivity during the initial (3+2) cycloaddition, subsequent acetalization, and lactonization. Both the regiodivergent adducts 16-3a and 16-4a were further confirmed by X-ray diffraction analysis. The steric hindrance on the azomethine ylide resulting from TMG exceeded that introduced by DMAP, which led to the regioselective reversal in the (3+2) cycloaddition. The control experiments and NMR studies of the deprotonation of the iminodiester (16-1) by DMAP and TMG were in alignment with the proposed mechanism.

Catalysts **2021**, 11, 1013 22 of 47

Scheme 32. Possible reaction mechanism for base-controlled regiodivergent [3+2] cycloaddition.

2.3. NHC Catalysts

The widespread use of NHC has revealed it to be an important organocatalyst that has been used in many synthetic strategies. Catalysts based on NHC are widely utilized for the synthesis of various biologically important natural products as well as pharmaceutical drugs [31]. Usually, these catalysts are used in C-C and C-heteroatom bond formation reactions and involve various cycloaddition reactions such as [2+2], [2+2+2], and [2+4]. These cycloadditions are generally achieved by the catalytic ability of NHC, which alters the polarity of a carbonyl compound via NHC-linked intermediates such as the Breslow, azolium enolate, acylazolium, and homoenolate intermediates (Scheme 33) [103–113].

Scheme 33. NHC-linked intermediates.

In 2014, Smith and co-workers described a regiodivergent *O*- to *C*- or *N*-carboxyl transfer of pyrazolyl carbonates (17-1) by the choice of catalyst (Scheme 34) [114]. Specifically, DMAP in dichloromethane delivered kinetically favored *O*- to *N*-carboxyl transfer with good regioselectivity (as high as 99%) and low to good yields (17-2, 10–80%), whereas triazolinylidene NHC in toluene afforded thermodynamically favored *O*- to *C*-carboxyl transfer with good regioselectivity (as much as 99%) and low to good yields (17-3, 12–84%). In addition to that, the chiral triazolium NHC catalyst promoted enantioselective

Catalysts **2021**, 11, 1013 23 of 47

(as high as 92%) and regioselective (as high as 99%) *O*- to *C*-carboxyl transfer products in good to excellent yields (17-4).

Scheme 34. Selective regiodivergent *O*- to *C*- or *N*-carboxyl transfer of pyrazolyl carbonates catalyzed by Lewis bases.

Further mechanistic experiments led to the conclusion that *O*- to *C*- or *N*-carboxyl transfer in pyrazolyl carbonates with DMAP was irreversible because the formation of the *N*-carboxylation product is kinetically favored. Contrary to this, *N*- to *C*-carboxyl transfer is not possible with DMAP. The *O*- to *C*- or *N*-carboxyl transfer with triazolinylidene NHCs is reversible because the formation of the *C*-carboxylation product is thermodynamically favored, and *N*- to *C*-carboxyl transfer is also considered to be feasible. On the other hand, *O*- to *C*- or *N*-carboxyl transfer is irreversible, with the chiral NHC catalyst exercising good enantiocontrol, although *N*- to *C*-carboxyl transfer is allowed with high enantiocontrol. Further, DFT studies supported the proposed mechanistic pathway shown in Scheme 35. Initially, the catalyst attacks the *O*-carboxylate to form the tetrahedral intermediate **TS(IV)**. Then, consecutive collapse of **TS(IV)** produces two common intermediates, enolate and a carboxylated catalyst, after which the carboxylated catalyst could be recaptured by the enolate either at *C*(4) or *N*(1) to produce (**TS-VII**). Finally, regeneration of the catalyst from the tetrahedral intermediate (**TS-VIII**) affords the two regiodivergent products (**17-2**, **17-3**).

Catalysts 2021, 11, 1013 24 of 47

Scheme 35. Proposed reaction mechanism for *O*- to *C*- and *N*-carboxylation.

In 2015, Smith *et al.* reported regioselective carboxylation either at the γ - or α -position depending on the Lewis base involved (Scheme 36) [115]. Treatment of a furanyl carbonate (18-1) with the triazolinylidene NHC catalyst produced a γ -isomer with regioselectivity as high as 99:1 (18-2). Under optimal conditions, phenyl, trichloro ethyl, and certain sterically hindered substrates were well tolerated to afford the corresponding γ -C(5) carboxylation product in good to high yields. In contrast, the α -isomer product was generated by changing the catalyst to DMAP with moderate regiocontrol (60:40) to produce the α -C(3)-carboxylate as the major product (18-3). Individual treatment of the α - and γ isomers with DMAP did not result in transformation, and the starting material was recovered even though the reaction time was prolonged. The α -carboxyl product underwent regioisomeric exchange in the presence of the NHC catalyst to afford the α/γ products in a 16:84 ratio, with the γ derivative as the major product. Similar results were obtained when the γ -regioisomer was reacted in the presence of the NHC catalyst to afford a 14:86 ratio of α/γ . These results revealed that C-carboxylation with DMAP is irreversible to preferentially yield the α -regioisomer. However, in the case of the NHC catalyst, C-carboxylation resulted in the formation of the γ -isomer as the major product followed by subsequent equilibration to form a mixture of α/γ products.

Catalysts **2021**, 11, 1013 25 of 47

Scheme 36. Lewis base-promoted *O*- to *C*-carboxyl transfer of furanyl carbonates.

Yao *et al.* demonstrated the regioselective synthesis of 3-pyrazolidinones via NHC-catalyzed [3+2] annulation of α -bromoenals (19-1) with hydrazine (19-2) in the presence of a base (Scheme 37) [116]. A regioselective methodology was devised by carefully adjusting the NHC catalysts, i.e., the imidazolium NHC precursor produced the 1,5-disubstituted 3-pyrazolidinone (19-3), whereas the triazolium NHC precatalyst was able to drive the reaction to completion to furnish the 2,5- difunctionalized isomer (19-4). Specifically, the regioselective Michael addition of the key intermediate to phenylhydrazine followed by subsequent lactamization afforded the regiodivergent products (19-3, 19-4). This protocol was an attractive strategy for the assembly of biologically significant 3-pyrazolidinones in moderate to high yields (as high as 84%), under mild reaction conditions, and with good regioselectivity.

$$R^{1} \stackrel{\longleftarrow}{ \coprod} R^{1}$$

$$R^{1} \stackrel{\longleftarrow}{ \coprod} R^{2} \stackrel{\longleftarrow}{ \coprod} N$$

$$R^{1} \stackrel{\longleftarrow}{ \coprod} R^{2} \stackrel{\longleftarrow}{ \coprod} N$$

$$R^{2} \stackrel{\longrightarrow}{ \coprod} N$$

$$R^{2} \stackrel{\longleftarrow}{ \coprod} N$$

$$R^{2} \stackrel{\longleftarrow}{ \coprod} N$$

$$R^{2} \stackrel{\longrightarrow}{ \coprod}$$

Scheme 37. NHC-catalyzed regiodivergent synthesis of 3-pyrazolidinones.

The authors proposed a plausible mechanistic pathway to explain the formation of product **19-3** (Scheme 38). Initially, the addition of the NHC catalyst to the α -bromoenal forms the Breslow intermediate **I**, which is further debrominated into the α , β -unsaturated acylazolium intermediate **II** followed by Michael addition with phenylhydrazine, producing intermediate **III**. This intermediate then undergoes lactamization to afford the target compound **19-3** and the regenerated catalyst. Although the origin of the regioselectivity aided by the tuning of the catalyst remains uncertain, the authors suggested that a computational study on the relationship between the structure of the catalysts and the regioselectivities would aid further understanding.

Catalysts **2021**, 11, 1013 26 of 47

Scheme 38. Proposed catalytic cycle for the synthesis of 3-pyrazolidinones.

Glorius *et al.* devised a scheme for the synthesis of 1,2-diazepines (20-3) via formal [4+3] annulation and the synthesis of pyrazoles (20-4) via formal [4+1] annulation reactions along highly regio- and enantioselective pathways (Scheme 39) [117]. The reaction between enals (20-1) and hydrazones (20-2) in the presence of the chiral triazolium NHC catalyst 20-5c afforded 1,2-diazepine derivatives through the homoenolate intermediate along a [4+3] annulation pathway. Various substituted enals containing both an EWG and EDG on the aromatic ring afforded the expected diazepine products in good yields with excellent enantioselectivities (99% *ee*). Similarly, hydrazones with different substituents reacted with enals via formal [4+3] annulation to form 1,2-diazepines in high yields with excellent enantioselectivities (99% *ee*). The use of the NHC catalyst with a morpholine backbone (20-5i) afforded the pyrazole derivatives with high regioselectivity (<1:20) via a Stetter reaction and subsequent cyclization reaction. This reaction occurred though the acyl anion intermediate initiated by the 20-5i NHC catalyst which suppressed the homoenolate reactivity of enals to produce the pyrazoles.

cat. 20-5c (10 mol%)

$$K_2CO_3$$
 (2.5 equiv.)

 $EVACO_3$ (2.5 equiv.)

 $EVACO_$

Catalysts 2021, 11, 1013 27 of 47

Scheme 39. NHC-catalyzed formal [4+3] and [4+1] annulations for the synthesis of 1,2-diazepines and pyrazoles.

According to the proposed reaction mechanism (Scheme 40), the chiral NHC catalyst initially undergoes addition to the enal cinnamaldehyde (**20-1**) to form two Breslow intermediates (**II** and **V**). The structure of the NHC is suggested to play a crucial role in determining the reaction pathways to form either a haloenolate or acyl anion. Specifically, *N-Mes* containing NHC catalyst **20-5c** preferentially forms a homoenolate intermediate (**II**), whereas the reaction pathway via the acyl anion (**V**) predominantly occurs with the NHC-based catalyst *N-2,6-(OMe)*² (NHC **20-5i**). Then, the homoenolate intermediate **II** undergoes conjugate addition with the in situ formed azoalkene, followed by C-C bond formation. Subsequent *N*-acylation delivers [4+3] annulated product 1,2-diazepine (**20-3**) after regeneration of the NHC catalyst **20-5c**. On the other hand, the NHC **20-5i**-bound acyl anion intermediate **V** undergoes a Stetter reaction with the in situ generated azoalkene to afford adduct **VII**. The release of NHC **20-5i** followed by intramolecular cyclization and dehydration affords the final [4+1] annulation pyrazole product (**20-4**).

Scheme 40. Proposed catalytic cycle for the synthesis of 1,2-diazepines and pyrazoles.

Glorius and co-workers also reported the NHC-catalyzed regiodivergent synthesis of pyridazino[6,1-*a*]isoquinoline and pyrazolo[5,1-*a*]isoquinolines by formal [3+3] and [3+2] annulations via a homoenolate intermediate and an enol intermediate, respectively (Scheme 41) [118]. The reaction between enals (21-1) and *N*-iminoisoquinolinium ylides (21-2) produced the above products in good to high yields with high enantiomeric excess. The formation of regiodivergent products was governed by the NHC precatalyst, base,

Catalysts **2021**, 11, 1013 28 of 47

and solvent of the reaction. Initially, the homoenolate intermediate formed by the reaction between α,β -unsaturated aldehydes and the NHC catalyst was converted into an enol intermediate by subsequent protonation at the β -position. The conjugate acid of the catalytic base was generated from the azolium salt by deprotonation, depending on whether this was sufficiently acidic to protonate the homoenolate, to afford the [3+2] annulation product via the formation of the enol intermediate. The authors also concluded that the addition of a base (DBU) would limit the formation of the enol intermediate, whereas increasing the proton concentration by the addition of an acid (acetic acid) would produce a greater amount of 21-4 by promoting the formation of the enol intermediate. The optimized conditions were compatible with various enals and N-iminoisoquinolinium ylides containing both an EWG and EDG, which reacted to produce the formal [3+3] annulated products in good yield with high ee when the 21-D NHC catalyst was employed. In the presence of NHC catalyst 21-E, formation of the NHC-enolate intermediates was predominant to afford the pyrazolo[5,1-a]isoquinoline product via formal [3+2] annulation by suppressing the homoenolate intermediate. Under the optimized conditions, various enals and N-imino-3-phenylisoquinolinium ylides delivered the expected products in good yields with excellent *ee* and *dr* (20:1).

Scheme 41. NHC-catalyzed regiodivergent dearomatizing annulation reaction.

In the proposed reaction mechanism (Scheme 42), addition of the NHC precatalyst to the α , β -unsaturated aldehydes (21-1) produces the common Breslow intermediate II. Under strongly basic conditions, the Breslow intermediate reacts with N-iminoisoquinoliunium ylide (21-2) to afford the acyl azolium V intermediate via a homoenolate intermediate. Regeneration of the catalyst from the acyl azolium affords the [3+3] annulated product (21-3). On the other hand, the Breslow intermediate undergoes β -protonation to form the enol equivalent VI under weakly basic reaction conditions. The subsequent addition of intermediate VI forms the acyl azolium VIII intermediate, which, on N-acylation, delivers the formal [3+2] annulated product (21-4) followed by regeneration of the NHC precatalyst.

Catalysts **2021**, 11, 1013 29 of 47

Scheme 42. Proposed reaction mechanism for NHC-catalyzed switchable annulation reaction.

3. Amine Catalysts

In the past two decades, L-proline and its derivatives have found rapidly growing application in various transformations to yield products with excellent *ee* and *dr* [119–126]. Remarkable advances have been made after the seminal work of List [127,128], Córdova [129,130], Barbas [131,132], and many other research groups. The discovery that a simple and effective catalyst such as L-proline could be put to effective use was a landmark achievement in this century and opened a new avenue for asymmetric synthesis. Despite the development of several modified proline catalysts, proline is still placed at the top of the list in terms of its performance. An enormous number of chemical transformations have been conducted by using derivatives of chiral organocatalysts including Aldol, Mannich, Michael addition, and Diels–Alder reaction, and if required, these catalysts are able to induce remarkable stereoselectivity. Importantly, several natural products and drugs have been synthesized by using these L-proline-derived catalysts [133–142].

Chen et al. disclosed switchable intermolecular regioselective [6+2] and [4+2] cycloadditions of α' -benzylidene-2-cyclopentenones with activated alkenes in the presence of a chiral primary amine catalyst and co-catalyst in high yields with high enantioselectivity (Scheme 43) [143]. The asymmetric intermolecular γ , β' -regioselective [6+2] cycloaddition of α' -benzylidene-2-cyclopentenones (22-1) with 3-olefinic 7-azaoxindoles (22-2), driven by the catalytic activity of the 22-C1 or 22-C3 chiral amine, with salicylic acid (A1) as the co-catalyst, provided thermodynamically stable fused bicyclic compounds with five contiguous stereogenic centers in toluene with excellent enantioselectivities (22-3). The cycloaddition proceeds through the in situ generated formal 4-amino fulvene, which served as a 6st partner. Interestingly, the cycloaddition in the presence of the chiral amine 22-C2 and co-catalyst 2-mercapto benzoic acid (22-A2) switched to an $\alpha_{i}\gamma$ -regioselective [4+2] cycloaddition with the generation of a dienamine intermediate which reacted with the alkene to afford bridged bicyclo[2.2.1]heptane derivatives (22-4). The proposed mechanism (Scheme 44) whereby these cycloadditions occur involves the formation of an imine intermediate with α' -benzylidene-2-cyclopentenones (22-1) with the aid of the chiral primary amine catalyst 22-C1. This iminium intermediate is then converted into a 4-aminofulvene (cross-conjugated trienamine) intermediate. Then, a [6+2] cycloaddition with an alkene affords the bicyclic γ , β' -regioselective product 22-3 by the elimination of the chiral amine catalyst **22-C1**. For the [4+2] cycloaddition reaction, initially, β' -regionelective sulfur addition takes place with the benzylidene-2-cyclopentenones, which then underCatalysts 2021, 11, 1013 30 of 47

goes C=C bond isomerization to produce an enone with a sulfide intermediate (22-5). Addition of the amine catalyst and alkene to intermediate 22-5 forms the corresponding product (22-4) via a dienamine-mediated [4+2] cycloaddition, followed by the elimination of mercaptobenzoic acid (22-A2).

Scheme 43. Chiral amine-catalyzed [6+2] and [4+2] cycloaddition reactions.

Scheme 44. Possible reaction mechanism for chiral amine-catalyzed asymmetric cycloadditions.

In 2018, we reported an L-proline-catalyzed, solvent-controlled regiodivergent Mannich reaction between cyclic imines and various ketones (Scheme 45) [144]. By utilizing this protocol, a wide range of ketones (23-2) and benzoxazinone cyclic imines (23-1) effi-

Catalysts 2021, 11, 1013 31 of 47

ciently underwent the Mannich reaction in a highly enantio- and diastereoselective manner. Later, a useful α -amino acid derivative was obtained after the removal of the aromatic auxiliary. The use of unsymmetrical ketones as nucleophilic partners, depending on the solvent, enabled different regioselective products to be obtained. Subsequently, highly enantioselective linear isomers were obtained as major products when the reaction was performed in chloroform (23-3). This may also proceed via the formation of the a less substituted enamine intermediate (TS1). On the other hand, we found that the polar solvent DMSO furnished the branch isomer as the major product, and that this reaction was highly enantio- and diastereoselective (23-4). The role of the solvent in the reaction remained unclear. However, other researchers also reportedly observed a similar transition state (TS2) when they utilized DMSO as the solvent. The XRD analysis (X-ray diffraction analysis) revealed that the obtained branch isomer was in fact an anti-Mannich adduct, suggesting that the enamine approaches the Re face of the benzoxazinone imine (Scheme 46).

Scheme 45. Solvent-controlled regiodivergent Mannich reaction.

Scheme 46. Plausible reaction mechanism for regiodivergent Mannich reaction.

Zanardi *et al.* reported the divergent regio- and stereoselective synthesis of spirodecanones and bicyclooctane derivatives via [3+2] and [4+2] cycloadditions, respectively (Scheme 47) [145]. The enolizable dicyanodienes (24-1) reacted with cinnamaldehyde (24-1)

Catalysts **2021**, 11, 1013 32 of 47

2) in the presence of an amine/NHC catalyst in a one-pot reaction to afford the spirodecanone (24-3) via a [3+2] cycloaddition reaction. On the other hand, the addition of 4-nitrophenol as a co-catalyst switched the reactivity to produce bicyclooctane carbaldehydes (24-4) by a [4+2] cycloaddition. A sequential C- ε regioselective bis-vinylogous Michael addition in the presence of a bulky TBS protected the prolinol catalyst, followed by an NHC-catalyzed 1,6-Stetter reaction involving C- δ [3+2] spiroannulation, producing ε , δ bonded spiro[4.5]decanones in the presence of potassium acetate as the base. Substrates of different sizes (including both EWGs and EDGs on the benzene ring) were well tolerated with complete diastereoselectivity (>20:1 dr) along with complete regioselectivity and a high enantiomeric excess. A two-step domino reaction sequence was utilized to synthesize bicyclo[2.2.2]octane carbaldehydes via a formal [4+2] cycloaddition reaction. Initially, γ' enolate was formed from the enolizable dicyanodienes and the enal, activated by the prolinol catalyst following which the subsequent intramolecular 1,6-Michael addition at the δ region afforded the expected product. The use of 10 mol% 4-nitrophenol as an additive in chloroform at room temperature afforded the product in good yields with 17:1 site selectivity along with high *ee* (96%) and *dr* (>20:1).

Scheme 47. Regioselective synthesis of enantioenriched carbocyclic compounds.

The reaction mechanism that was proposed (Scheme 48) involves the initial activation of the cinnamaldehyde (24-2) by the organocatalyst prolinol silyl ether by lowering the LUMO. Subsequently, the hydroxide ion deprotonates the cyclohexenylidene malononitriles at ε , δ' to yield both of the enolates **II** and **IV**, respectively. Coulombic interaction between the enal nitrogen atom and the nitrogen atoms of the cyano group initiates enantioselective attack of the Si face of the enal acceptor by the bis-vinylogous enolate **II**. Hydrolysis of enamine intermediate **III** produces 24-5 and, ultimately, the final product 24-3, and this is accompanied by the regeneration of the organocatalysts. For the [4+2] cycloaddition reaction, the δ' -enolate is not stabilized by the enal nitrogen; instead, it is stabilized by the addition of p-nitrophenol, which acts as a hydrogen bond donor. Under these circumstances, the attack of **IV** (from its Re face) to the Si face of the enal is more favorable, producing bicyclooctane carbaldehydes (24-4) upon hydrolysis of intermediate **VI** along with the regeneration of the catalyst.

Catalysts **2021**, 11, 1013 33 of 47

Scheme 48. Proposed reaction mechanism for regiodivergent ε - and γ' , δ -pathways.

4. Brønsted Acid Catalysts

4.1. Phosphoric Acid Catalysts

In recent years, the activation of carbonyl compounds by utilizing chiral Brønsted acids has received an enormous amount of attention, i.e., the activation of reactants by way of a hydrogen bonding connection, which is one of the fastest growing research areas. Chiral phosphoric acids have proven to be highly efficient catalysts for a wide range of asymmetric transformations under mild reaction conditions. In general, binaphthyl is used to synthesize chiral phosphoric acid derivatives. These catalysts have been involved in several reactions including the Diels–Alder, Nazarov, Mukaiyama Aldol, Mannich, Henry, Morita–Baylis–Hillman reactions, and 1,3-dipolar cycloadditions [146–154].

Tay and co-workers reported an efficient method for the regioselective synthesis of glycosides in macrolactone (Scheme 49) [155]. Chiral phosphoric acid-catalyzed selective glycosylation of complex phenols was achieved with excellent regiodivergence. Glycosylation of **6-dEB** (**25-6-DEB**) with 6-deoxyglucose (**25-1**) in the presence of BINOL-based chiral phosphoric acids led to glycosylation at the **C5** position of the macrolactone with a high r.r. (regiomeric ratio) (99:01) in toluene. However, the use of SPINOL as a chiral phosphoric acid in DCM resulted in glycosylation at the **C3** alcohol of the macrolactone with a 73:27 r.r. The **C11** hydroxyl was also selectively glycosylated in the presence of phenylboronic acid in toluene as the solvent. The **C3** and **C5** hydroxyls were present in a 1,3-syn relationship, which was masked by the formation of the boronic acid ester to allow formation of the glycoside at the **C11** position. The hydroxyl groups at **C3** and **C5** were regenerated after the boronate was cleaved during the subsequent workup with peroxide.

Catalysts **2021**, 11, 1013 34 of 47

Scheme 49. Regiodivergent glycosylation of 6-deoxy-erythronolide B.

In 2020, Wang's group developed aza- and oxo-[3+2] cycloadditions between α -enaminones (26-2) and quinones (26-1) in the presence of chiral phosphoric acids and 4Å molecular sieves (M.S.), respectively (Scheme 50) [156]. In the presence of the chiral phosphoric acid (26-(R)-CPA5), a wide range of N-substituted indoles (26-4) were obtained as the products of a formal aza-[3+2] cycloaddition. On the other hand, in the presence of bulky chiral phosphoric acid (26-(R)-CPA3) and 4Å molecular sieves as an additive, the product 2,3-dihydrobenzofuran (26-3) was produced in the highest yields with excellent enantioselectivities via an oxo-[3+2] cycloaddition. Various substituted quinones including different esters (Me, Et, and Bn), and α -enaminones containing an EDG and EWG reacted smoothly to produce the products in good yield with excellent enantioselectivities.

Scheme 50. Formal oxo- and aza-[3+2] reactions of α -enaminones and quinones.

In the absence of molecular sieves, this reaction proceeded to produce N-substituted indole derivatives in excellent yield with toluene as the solvent (26-4). Quinones containing different ester groups as well as α -enaminones bearing an EDG or EWG at the *para*

Catalysts 2021, 11, 1013 35 of 47

position of the aryl ring were compatible under the optimized conditions and delivered the indole derivatives in good yield.

The role of 4Å M.S. was important to obtain benzofuran derivatives. In the absence of molecular sieves, or when they are replaced by dry MgSO₄ or freshly activated 4Å M.S. and H₂O (2μL), this would slow down the formation of benzofuran derivatives (**26-3**) and slightly increase the formation of indole derivatives (**26-4**). These above data reveal that M.S. do not serve as a drying reagent in this reaction, and the addition of water relatively favors the indole formation. Further, the size and format of M.S. also influence the reaction outcome. In detail, for 3Å M.S. and 4Å M.S., their selectivity towards benzofuran and indole is >20:1 (**26-3:26-4**), whereas in the case of 5Å M.S., 4Å M.S. (beads), and wet 4Å M.S. (beads), it is about 11:1, 1:1, and 1:5.3, respectively. From ¹H NMR and ³¹P NMR studies of chiral phosphoric acid with and without M.S., the authors concluded that M.S. could affect the acidity of phosphoric acid and accelerate the interaction between the catalyst and the substrate.

According to the proposed mechanism (Scheme 51), the enamine of the α-enaminones (26-2) initially acts as a nucleophile to attack the *Re* face of the quinone (26-1) from the *Si* face to afford the intermediate int-A via TS-I. *O*-tautomerization from the quinone (enol/phenol) and *N*-tautomerization (enamine from the imine) afford int B and int C, respectively. These intermediates (int-B, int-C) are attached to the chiral phosphoric acid, as illustrated for TS-II and TS-III. The hydroxy group of TS-II then attacks the *Si* face of the imine to afford 2,3-dihydrobenzofuran (26-3a), whereas indole (26-4a) is obtained from TS-III via int D. Initially, the amine group attacks the *Si* face of the carbonyl group to afford int D, which then undergoes dehydroxylation to produce the indole derivative. As TS II is more polarized in nature than TS III, a polar solvent such as DCM stabilizes TS II, and the addition of 4Å molecular sieves (4Å M.S.) accelerates the proton transfer in TS II via absorbing/releasing a proton. Alternatively, TS-II is destabilized in the presence of toluene, a non-polar solvent, in which case the reaction preferentially proceeds via TS-III.

Catalysts 2021, 11, 1013 36 of 47

Scheme 51. Proposed reaction mechanism for formal oxo- and aza-[3+2] reactions.

Recently, in 2020, the group of Li and Li reported ortho- and para-selective regiodivergent C-H functionalization between 1-naphthols (27-1) and 1-azadienes (27-2) via a Michael addition reaction (Scheme 52) [157]. The chiral squaramide catalyst afforded a product in which an ortho-selective C-H bond was constructed, whereas para-selective C-H bond formation occurred in the case of chiral phosphoric acid catalysts. Under the optimized reaction conditions, with the chiral squaramide catalyst (27-SA-1), 1-naphthol with different substituted 1-azadienes (F, Cl, Br, Me, and OMe) afforded the expected orthoselective Friedel-Crafts alkylation products in good yields with high enantioselectivities (27-3). Similar results were obtained with different substituted 1-naphthols (Br, OMe) which delivered the ortho-selective products in excellent yields with good ee. The use of 27-CPA-4 (1 mol%) as the catalyst resulted in regiodivergent para-selective C-H bond functionalization (27-4). Within the scope of this substrate, 1-azadienes containing various EWGs (F, Cl, Br) and EDGs (Me, OMe) on the aromatic ring could be well tolerated to offer para-selective Friedel-Crafts alkylation products in good yields and with high ee. Control experiments showed that the free hydroxy group of 1-napththol was essential to obtain the product in this Michael addition. Both the catalysts (27-SA-1, 27-CPA-4) failed to produce the product when 1-hydroxynaphthalene was protected with methyl (1-methoxynaphthalene, 27-1c), which reacts with 1-azadiene (27-2a).

Catalysts **2021**, 11, 1013 37 of 47

Scheme 52. Regiodivergent C-H functionalization of 1-naphthols with 1-azadienes.

4.2. p-Toluenesulfonic Acid Catalyst

Chen and co-workers reported the regiodivergent nucleophilic phosphorylation of indolylmethanols (28-1) with diaryl phosphine oxide (28-2) in the presence of a Brønsted acid catalyst (Scheme 53) [158]. The benzyl phosphorylated product (28-3) was obtained by the utilization of 10 mol% of TsOH.H₂O (*p*-toluenesulfonic acid monohydrate) in nitromethane at 25 °C with moderate to good yield. A variety of 2-indolylmethanols with an EDG produced comparably higher yields than those with an EWG. Similarly, diarylphosphine oxides with an EWG at the *para* position produced higher yields. On the other hand, the C-3 phosphorylation product (28-4) was obtained in good yield by using 20 mol% of the TfOH catalyst at 80 °C. Indolylmethanol containing both an EDG and EWG was well tolerated to afford the products in moderate yields.

Catalysts **2021**, 11, 1013 38 of 47

Scheme 53. Brønsted acid-catalyzed regiodivergent phosphorylation of 2-indolylmethanols.

In the proposed reaction mechanism (Scheme 54), the Brønsted acids generate a partial positive charge at the benzylic position of the nitrogen atom or the C3 position of the 2-indolylmethanols. Then, the diarylphosphine oxides attack the benzylic position to afford the benzylic phosphorylated product 28-3. In the presence of a strong acid such as TfOH and upon exposure to heat, the benzylic phosphorylated product may undergo a [1,3]-P migration to afford the thermodynamically stable product (28-4) via transient intermediate 28-5. Cross-over experiments concluded that the [1,3]-P migration entailed intramolecular migration. The acidity of the Brønsted acid was a crucial factor because the [1,3]-P migration hardly occurred in the presence of a weak acid.

Scheme 54. Possible reaction pathway for phosphorylation of 2-indolylmethanols.

5. Hydrogen Bond-Donating Catalysts

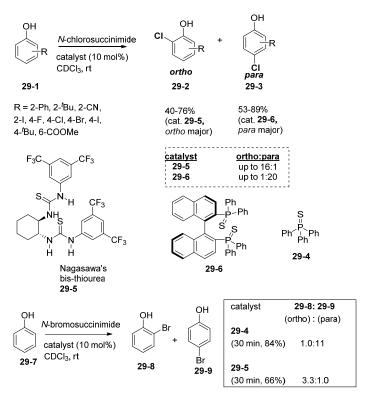
5.1. Thiourea Catalyst

In recent decades, thiourea derivatives have been commonly used as organocatalysts in organic and pharmaceutical chemistry. Moreover, these derivatives are also widely used as bifunctional catalysts in combinations such as amine–thiourea and phosphine–thiourea. Along with their catalytic activity, they are also involved as a component in various reactions including guanylation, thioarylation, and C–S cross-coupling reactions [159–166].

The regiodivergent chlorination of electron-rich phenols (**29-1**) established by Gustafson and co-workers is demonstrated in Scheme 55 [167]. Here, *ortho*-chlorination of the phenol (**29-2**) with *N*-chlorosuccinimide is promoted by 10 mol% of Nagasawa's bis-thiourea catalyst (**29-5**). The *meta*-substituted phenols (F, Cl, Br, I, and *t*-Bu) efficiently afforded good *ortho*-regioselectivity in the presence of Nagasawa's catalyst. The authors also

Catalysts 2021, 11, 1013 39 of 47

demonstrated the augmentation of the innate *para*-selectivity of phenols by using BINAP-derived phosphine sulfide as a catalyst (29-6). Phenols containing Ph, *t*-Bu, CN, and a halogen substituent afforded a *para*-selective chlorinated product as the major product (29-3). The authors also investigated the reaction conditions for regioselective bromination. Catalyst 29-4 afforded mainly the *para*-selective brominated product as the major regioisomer (29-9), whereas the presence of Nagasawa's bis-thiourea catalyst overcame the innate *para*-preference of the phenol to afford the *ortho*-brominated products (29-8) with good selectivity. The authors concluded that the regioselectivity mainly depends on the structure of the Lewis bases, and reversal of the regioselectivity by Nagasawa's bis-thiourea catalyst could promote chlorination via dual activation. That is, one of the thiourea moieties interacts with the phenol and the other activates NCS via a Lewis base or Brønsted acid manifold.



Scheme 55. Catalyst-controlled regiodivergent chlorination of phenols.

5.2. Squaramide Catalyst

Chiral squaramide, a bifunctional organocatalyst, is an effective alternative for urea/thiourea catalysts. Chiral squaramide has been shown to successfully catalyze several reactions including Michael additions, and the Mannich, aza-Henry, and Strecker reactions. Moreover, these catalysts have successfully produced enantio-enriched products in single and domino/cascade reactions in various asymmetric organic transformations [168–176].

In 2018, Xu and co-workers reported the organocatalytic, regiodivergent C-C bond cleavage of cyclopropenones (Scheme 56) [177]. Their efficient methodology involves a cascade cycloaddition followed by a regioselective cyclopropyl ring strain release process catalyzed by bifunctional squaramide catalysts. Aldimines (30-1) reacted with 2,3-diphenylcycloprop-2-enone (30-2) with 1 mol% of the catalyst to afford tetrahydrochromeno[4,3-b]pyrroles (30-3) as products in excellent yields with excellent *ee* and *dr* ratios (20:1). In contrast, completely different cyclized products, tetrahydrobenzofuro[3,2-b]pyridines (30-4), were obtained when methylphenylcyclopropenone (30-2') was used along with 20 mol% of the catalyst. The products were obtained in excellent yields with

Catalysts **2021**, 11, 1013 40 of 47

excellent enantioselectivities. The synergistic effect of hydrogen bonding activation and controlled ring strain release played a pivotal role in the generation of the two different ring systems. The "spring-loaded" intermediate with switchable C-C bond cleavages achieved by controllable ring strain release governed the regioselectivity of the reaction (Scheme 57). Nucleophilic addition to the hydroxy group at the carbonyl carbon (30-A) produced five-membered products, whereas six-membered cyclic products were obtained when the ring opening occurred at the α -site of the carbonyl carbon (30-B). This was substantiated by DFT studies. With this protocol, the authors were able to synthesize diverse heterocyclic frameworks with good enantioselectivity of 99% and an excellent yield (as high as 99%) for both regioisomers.

Scheme 56. Regiodivergent C-C bond cleavage of cyclopropenones.

Scheme 57. Results of the DFT calculation of the Mulliken charge distribution for 30-A and 30-B.

6. Conclusions

This review summarized the control of regiodivergent reactions by utilizing various organocatalysts. The use of several organocatalysts such as Lewis bases, amine bases,

Catalysts 2021, 11, 1013 41 of 47

Brønsted acids, and hydrogen bond-donating catalysts that were employed to deliver the regiodivergent products was described. The reactivity of various organocatalytic systems, the scope of the substrates, and their mechanistic studies were briefly discussed. The choice of the catalysts, additives, temperature, and solvents was found to play a crucial role in determining the regioselectivity of the reaction.

Although synthetic chemists have devoted lots of efforts to developing organocatalytic regiodivergent methods, in order to cater to the need for diverse molecules to access the chemical space, we need more regiodivergent methods by which we can synthesize a broad range of molecules easily.

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Catalysts 2021, 11, 1013 47 of 47

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