

Supplementary Materials

The Impact of Pressure and Hydrocarbons on NO_x Abatement over Cu- and Fe-Zeolites at Pre-Turbocharger Position

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1. Tables

Table S1. Washcoat loading of Fe-ZSM-5 and Cu-SSZ-13 honeycombs.

	Fe-ZSM-5 [g]	Cu-SSZ-13 [g]
Sample #1 (C ₃ H ₆)	2.06	2.20
Sample #2 (C ₁₂ H ₂₆)	1.96	2.20
Sample #3 (C ₈ H ₁₀)	2.04	2.20

2. Figures

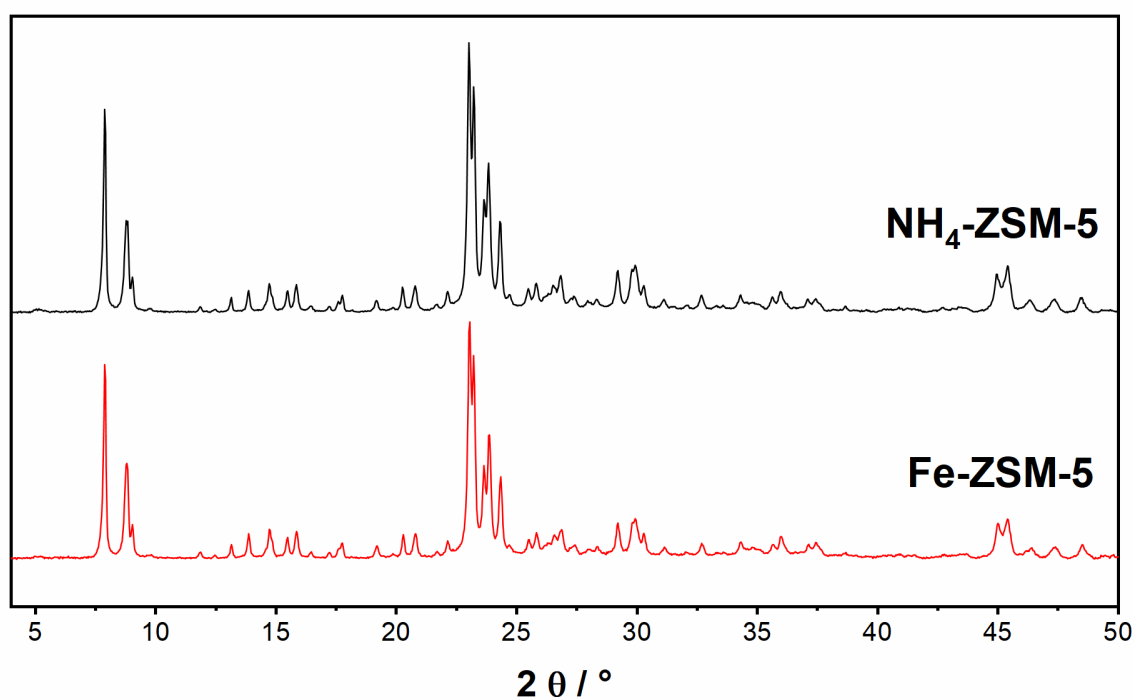


Figure S1. XRD patterns of NH₄-ZSM-5 (before ion exchange) and Fe-ZSM-5 (after ion exchange) show the characteristic pattern of the ZSM-5 framework.

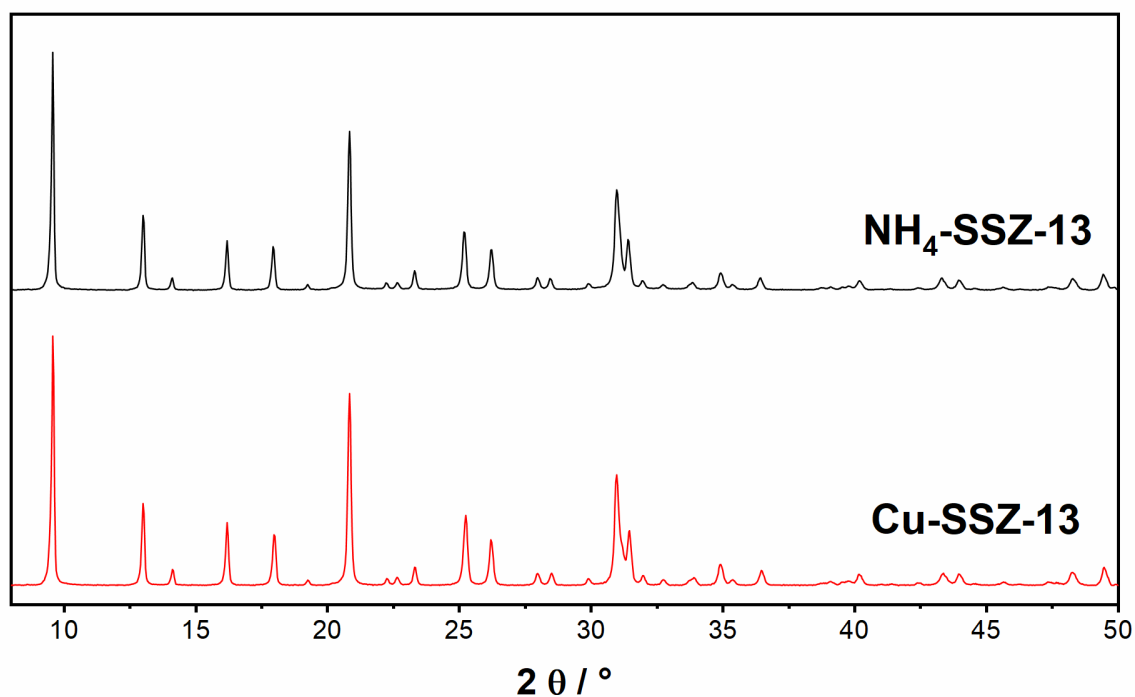


Figure S2. XRD patterns of NH₄-SSZ-13 (before ion exchange) and Cu-SSZ-13 (after ion exchange) show the characteristic pattern of the SSZ-13 framework.

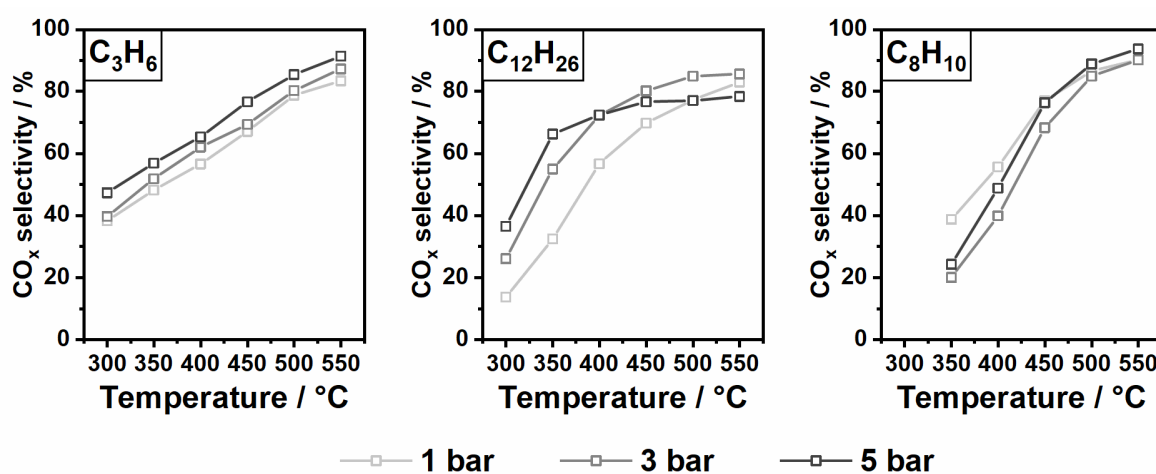


Figure S3. CO_x selectivity (CO + CO₂) during C_xH_y oxidation (14% O₂ and 4.5% H₂O in N₂) over Fe-ZSM-5 at 1, 3 and 5 bar pressure.

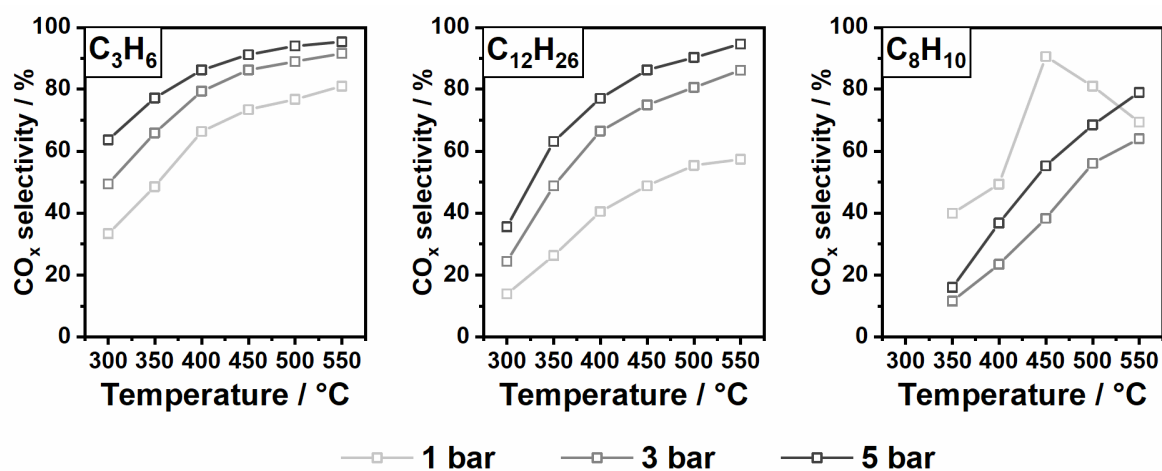


Figure S4. CO_x selectivity (CO + CO₂) during C_xH_y oxidation (14% O₂ and 4.5% H₂O in N₂) over Cu-SSZ-13 at 1, 3 and 5 bar pressure.

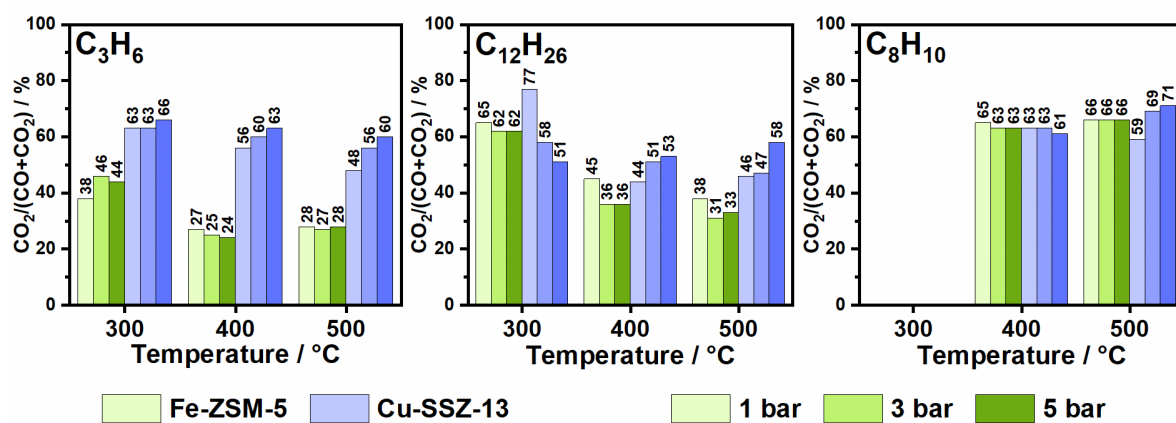


Figure S5. CO₂ share of formed CO_x (CO + CO₂) via hydrocarbon oxidation during standard SCR over Fe-ZSM-5 and Cu-SSZ-13.

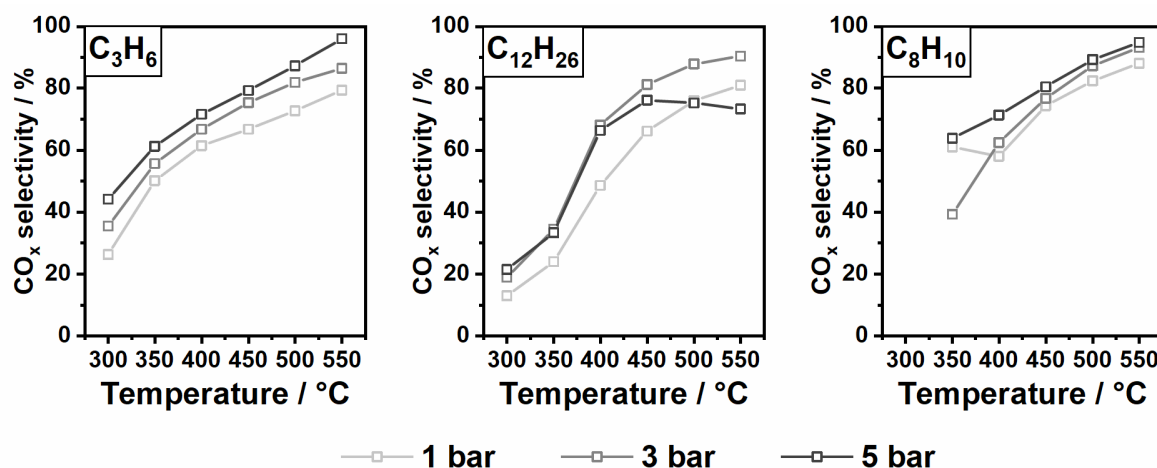


Figure S6. CO_x selectivity (CO + CO₂) during C_xH_y oxidation in standard SCR gas mixture (350 ppm NO, 350 ppm NH₃, 14% O₂ and 4.5% H₂O in N₂) over Fe-ZSM-5 at 1, 3 and 5 bar pressure.

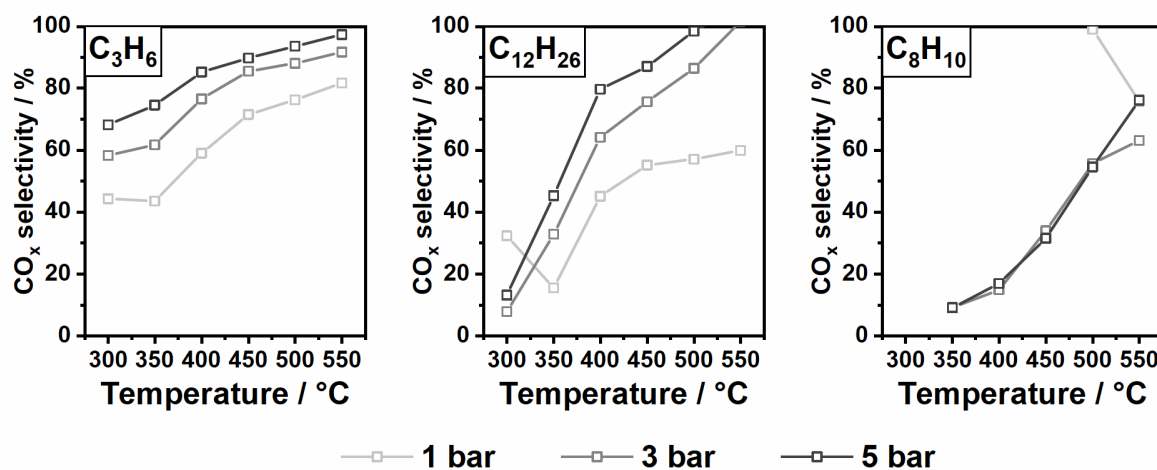


Figure S7. CO_x selectivity (CO + CO₂) during C_xH_y oxidation in standard SCR gas mixture (350 ppm NO, 350 ppm NH₃, 14% O₂ and 4.5% H₂O in N₂) over Cu-SSZ-13 at 1, 3 and 5 bar pressure.

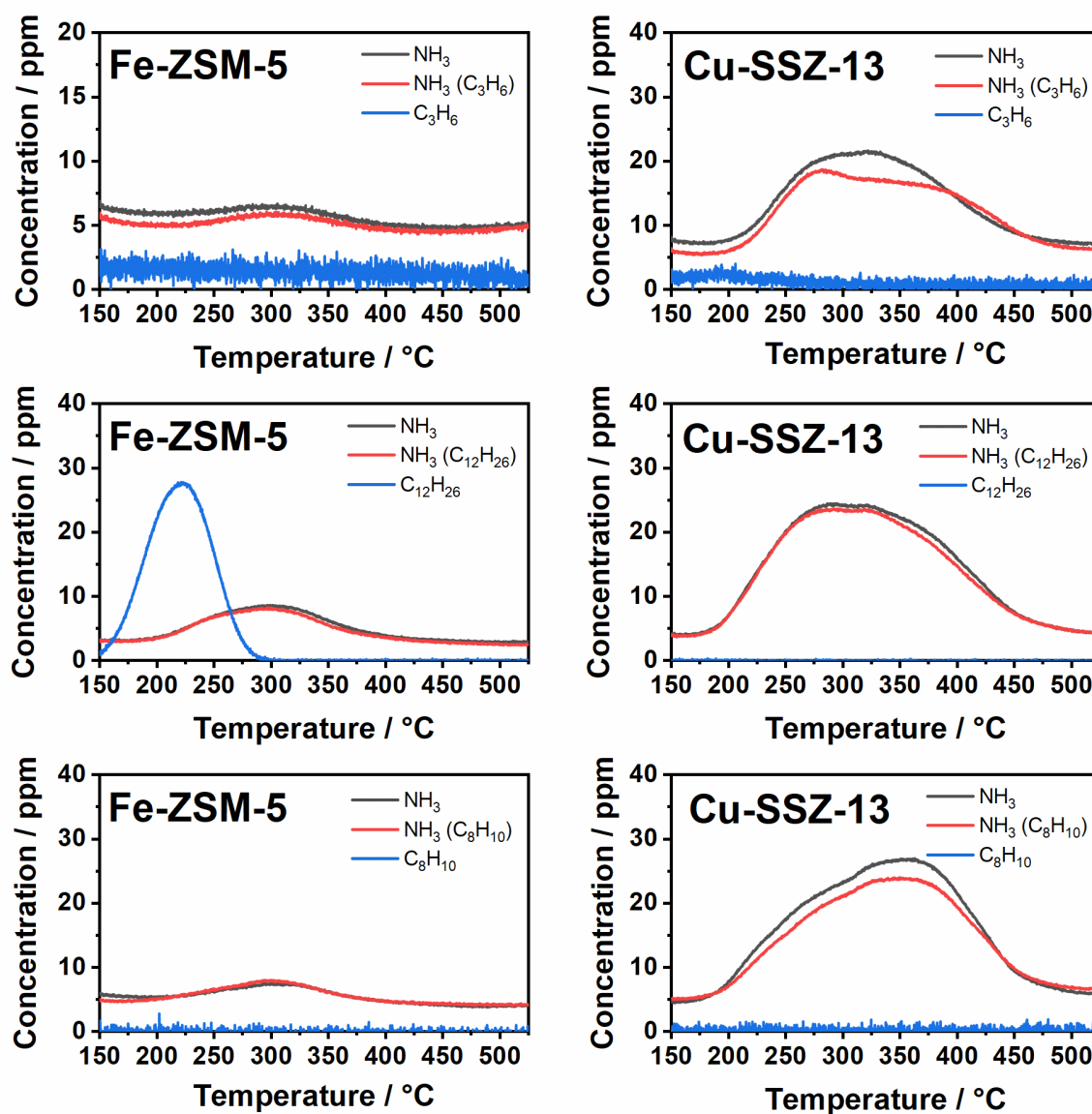


Figure S8. Comparison of the temperature programmed desorption of ammonia (NH₃-TPD) in presence and absence of C_xH_y of Fe-ZSM-5 and Cu-SSZ-13. All samples were saturated for half an hour with NH₃ or NH₃ + C_xH_y in a gas mixture consisting of 350 ppm NH₃, 200 ppm C₃H₆/50 ppm C₁₂H₂₆/75 ppm C₈H₁₀, 4.5% H₂O and N₂ at 150 °C and subsequently heated in N₂ at a heating rate of 5 K min⁻¹. Differences in the amount of stored NH₃ of the same catalyst type between the different hydrocarbons are due to a different amount of washcoat loading of the respective honeycomb used and the prior catalytic testing with the corresponding hydrocarbon.

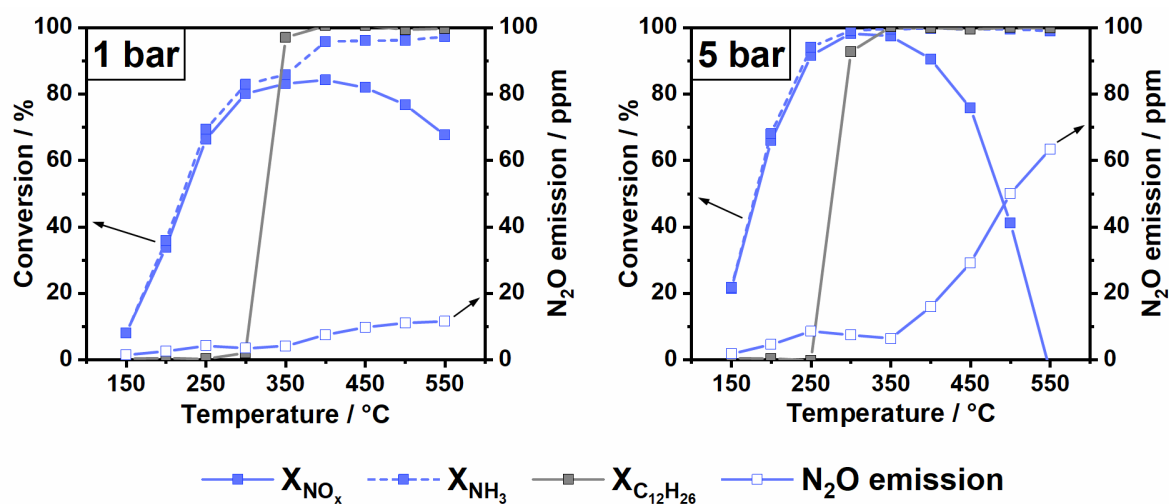


Figure S9. $\text{NO}_x/\text{NH}_3/\text{C}_{12}\text{H}_{26}$ conversion and N_2O emission over Cu-SSZ-13 for 1 bar (left) and 5 bar (right) in Standard SCR gas mixture (50 ppm $\text{C}_{12}\text{H}_{26}$, 350 ppm NO , 350 ppm NH_3 , 14% O_2 , 4.5% H_2O in N_2).