

## Supplementary Information

### **Enhanced hydrocarbons biodegradation at deep-sea hydrostatic pressure with microbial electrochemical snorkels**

Federico Aulenta<sup>1\*</sup>, Enza Palma<sup>1</sup>, Ugo Marzocchi<sup>2,3</sup>, Carolina Cruz Viggi<sup>1</sup>, Simona Rossetti<sup>1</sup>, Alberto Scoma<sup>4,5\*</sup>

<sup>1</sup> Water Research Institute (IRSA), National Research Council (CNR), Monterotondo, Italy

<sup>2</sup> Center for Electromicrobiology, Section for Microbiology, Department of Bioscience, Aarhus University, Aarhus, Denmark

<sup>3</sup> Integrative Marine Ecology Department, Stazione Zoologica Anton Dohrn, National Institute of Marine Biology, Ecology and Biotechnology, Napoli, Italy

<sup>4</sup> Section of Microbiology, Department of Biology, Aarhus University, Aarhus, Denmark

<sup>5</sup> Engineered Microbial Systems (EMS) Lab, Department of Biological and Chemical Engineering (BCE), Aarhus University, Aarhus, Denmark

\*Correspondence to:

Assoc. Prof. Alberto Scoma

Head of the Engineered Microbial Systems (EMS) Lab,

Section of Biological and Chemical Engineering (BCE), Department of Engineering

Hangøvej 2, 8200, Aarhus University, Aarhus, Denmark

email: as@bce.au.dk

Dr. Federico Aulenta

Head of the Laboratory for Microbial Bioprocesses

Water Research Institute (IRSA), National Research Council (CNR)

Via Salaria km 29,300 - 00015, Monterotondo (RM), Italy

email: aulenta@irsa.cnr.it

## **Estimates on the respiration of O<sub>2</sub> in seawater owing to the electrochemical oxidation of H<sub>2</sub>S in sediments through snorkels**

Sediments were placed in cylinders with the following dimensions: internal diameter 3.8 cm; height, 33.5 cm. The internal volume was therefore: 379.74 cm<sup>3</sup> (or mL) This volume was occupied by sediment, water and rubber stoppers (on top and at bottom). In particular, the sediment had a volume of 110 mL and was about 9.7 cm high. The water had a volume of 200 mL, and was about 17.6 cm high. These dimensions were used to estimate the total amount of O<sub>2</sub> dissolved in water, and the total amount of H<sub>2</sub>S accumulated in sediments. In all experiments carried out with Duc oil, H<sub>2</sub>S concentration was equal to zero at a specific depth below sediment surface level (bssl) (namely, between 0.1 and 0.9 cm, according to experimental condition; Table S2) and always increased linearly until 2 cm bssl. As H<sub>2</sub>S increase was linear, a homogeneous average concentration along the whole section of sediment rich in H<sub>2</sub>S can be calculated as:

$$\text{Eq. 1} \quad (\text{Max H}_2\text{S concentration} - \text{Min H}_2\text{S concentration}) / 2$$

The exact volume of sediment rich in H<sub>2</sub>S is calculated by the specific height (in cm) multiplied by the internal surface area of the glass cylinder (in cm<sup>2</sup>), which is equal in all conditions to 11.335 cm<sup>2</sup>:

$$\text{Eq. 2} \quad (\text{Onset H}_2\text{S accumulation} - \text{End of H}_2\text{S accumulation}) \cdot 11.335$$

The average concentration of H<sub>2</sub>S (in  $\mu\text{M}$  or nmoles mL<sup>-1</sup>) multiplied for the volume of sediment section (in cm<sup>3</sup> or mL) gives the amount of H<sub>2</sub>S in the sediment section (in nmoles). A summary of all the data for the experiments with Duc oil is reported in Table S2. For experiments conducted with Statfjord oil the same procedure was followed, with the only exception of glass control experiments at both 10 and 0.1 MPa (HPC and APC, respectively). Here, the H<sub>2</sub>S concentration pattern followed a first linear increase (to a depth of 1.5 cm bssl) followed by a somewhat linear decrease (to a depth of 3.5 cm bssl). As such, the sediment section rich in H<sub>2</sub>S in HPC and APC with Statfjord oil was divided in two, and two average H<sub>2</sub>S concentrations were assessed (all data in Table S2).

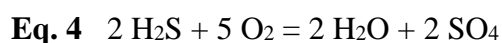
The total amount of H<sub>2</sub>S accumulated in all sediments varied between 14 to 1596 nmoles. H<sub>2</sub>S microprofiles in sediments were not assessed throughout the whole sediment depth. This

means that the total H<sub>2</sub>S accumulation (therefore, the O<sub>2</sub> respiration owing to its electrochemical oxidation) may have been underestimated. Such underestimation is not expected to be substantial as: 1) in Duc oil, H<sub>2</sub>S concentrations at the deepest level analyzed (2 cm bssl) were never higher than 60  $\mu$ M; 2) in Statfjord oil, H<sub>2</sub>S levels at the deepest point analyzed were higher than in Duc only when testing glass controls at both 10 and 0.1 MPa (HPC and APC; above 80  $\mu$ M); however, their concentration pattern was consistently decreasing at that depth, indicating that the bulk of H<sub>2</sub>S accumulation had been detected.

To determine the net amount of H<sub>2</sub>S oxidized electrochemically (Table S3), the difference between sediments incubated with snorkels and their respective glass controls was assessed as:

**Eq. 3** H<sub>2</sub>S concentration in HPS - H<sub>2</sub>S concentration in HPC, or  
H<sub>2</sub>S concentration in APS - H<sub>2</sub>S concentration in APC.

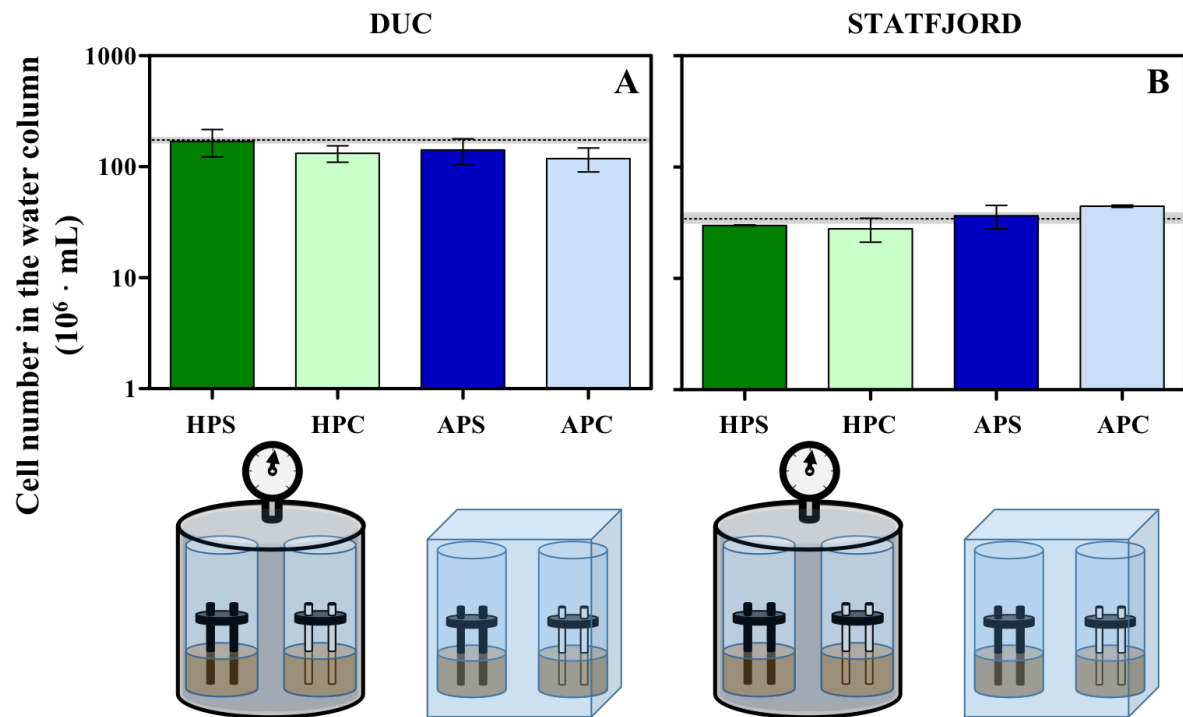
The stoichiometric electrochemical oxidation of H<sub>2</sub>S (in sediments) using O<sub>2</sub> (in seawater) as terminal acceptor follows the equation:



As it takes 2.5 moles of O<sub>2</sub> to completely oxidize 1 mole of H<sub>2</sub>S, the net amount of H<sub>2</sub>S oxidized electrochemically (in nmoles) was multiplied by 2.5 to yield the amount of O<sub>2</sub> oxidized (in nmoles). Provided that O<sub>2</sub> molecular weight is about 32 g per mole, the net amount of O<sub>2</sub> respired electrochemically by snorkels (in  $\mu$ g) was determined. The volume of seawater in glass cylinders was always 200 mL. Thus, the concentration of O<sub>2</sub> (in mg/L) respired electrochemically could be estimated.

The total O<sub>2</sub> consumption in seawater had been assessed in all experiments. The difference between O<sub>2</sub> respiration in cylinders incubated with snorkels and their respective glass control was calculated using the same approach for Eq. 3. As such, the relative amount of O<sub>2</sub> respiration due to H<sub>2</sub>S electrochemical oxidation could be estimated. This ranged between 42 and 7% across all experiments (Table S3).

**Figure S1.** Cell number in the water column at the end of 7 weeks of incubation in sediments contaminated with either Duc (A) or Statfjord (B) oil (n=3; bars represent standard errors). Cell numbers at time zero are reported as horizontal dotted lines, with the grey areas representing the standard error (n=3). Keys reported in the graph are: HPS, high pressure with snorkels; HPC, high pressure with glass controls; APS, ambient pressure with snorkels; APC, ambient pressure with glass controls. No statistical difference was observed between the different reactors contaminated with the same crude oil ( $p>0.05$ ).



**Table S1. Duc and Statfjord crude oil *n*-alkanes profiles in contaminated sediments at Time zero. Keys: DW, dry weight; s.d., standard deviation.**

<i>n</i> -alkane chain length	DUC crude oil					Statfjord crude oil				
	$\mu\text{g g}^{-1}\text{DW}$			$\mu\text{g g}^{-1}\text{DW}$		$\mu\text{g g}^{-1}\text{DW}$			$\mu\text{g g}^{-1}\text{DW}$	
	Replicate 1	Replicate 2	Replicate 3	mean	s.d. error	Replicate 1	Replicate 2	Replicate 3	mean	s.d. error
C8	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
C9	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
C10	0.0	0.0	0.0	0.0	0.0	1.2	1.0	1.3	1.2	0.1
C11	0.0	0.0	0.0	0.0	0.0	2.5	0.3	2.5	1.8	0.7
C12	0.0	0.0	0.0	0.0	0.0	1.7	1.7	1.7	1.7	0.0
C13	0.0	0.0	0.0	0.0	0.0	2.7	2.5	2.3	2.5	0.1
C14	1.7	0.5	0.9	1.0	0.3	7.7	6.8	6.6	7.0	0.3
C15	9.9	6.4	4.1	6.8	1.7	10.1	13.0	11.3	11.5	0.8
C16	5.8	3.8	3.2	4.2	0.8	18.1	13.7	13.2	15.0	1.5
C17	8.7	6.5	4.5	6.6	1.2	12.8	10.7	11.2	11.6	0.6
Pristane	25.5	20.8	16.9	21.1	2.5	9.2	8.9	9.1	9.1	0.1
C18	8.5	7.0	4.6	6.7	1.1	18.1	15.8	17.8	17.2	0.7
Phytane	14.6	22.3	14.4	17.1	2.6	9.3	7.7	8.4	8.5	0.5
C19	6.5	7.2	5.5	6.4	0.5	9.9	9.2	9.3	9.5	0.2
C20	6.3	9.4	9.7	8.4	1.1	14.1	13.5	14.4	14.0	0.3
C21	7.0	4.5	4.3	5.3	0.9	9.2	8.4	8.8	8.8	0.2
C22	6.7	4.3	7.6	6.2	1.0	18.6	14.6	16.2	16.5	1.2
C23	8.0	6.6	14.5	9.7	2.4	14.5	15.1	22.6	17.4	2.6
C24	6.2	11.7	3.3	7.1	2.5	13.0	8.6	10.3	10.6	1.3
C25	12.4	14.9	14.6	14.0	0.8	16.9	14.9	18.8	16.9	1.1
C26	10.7	11.1	9.4	10.4	0.5	21.8	22.3	23.8	22.6	0.6
C27	18.5	9.6	17.1	15.1	2.8	16.5	18.1	15.3	16.6	0.8
C28	18.5	11.9	11.2	13.9	2.3	27.7	25.0	25.9	26.2	0.8
C29	18.2	20.5	12.9	17.2	2.3	26.2	22.5	21.6	23.4	1.4
C30	0.0	0.0	0.0	0.0	0.0	39.8	35.1	30.2	35.0	2.8

C31	0.0	0.0	0.0	0.0	0.0
C32	0.0	0.0	0.0	0.0	0.0
C33	0.0	0.0	0.0	0.0	0.0
	<b>Totals</b>			<b>mean</b>	<b>s.d. error</b>
<b>C14-C19</b>	31.3	41.1	22.8	31.8	5.3
<b>C20-C29</b>	104.5	112.6	104.5	107.2	2.7
<b>Total <i>n</i>-alkanes</b>	135.9	153.7	127.4	139.0	7.8
				<b>mean</b>	<b>s.d. error</b>
<b>Total Petroleum Hydrocarbons</b>	3064	3531	3057	3217	157
<b>Relative abundance of <i>n</i>-alkanes in crude oil (%)</b>	4.43	4.35	4.17	4.32	0.08
	<b>DUC crude oil</b>				

	29.7	26.3	27.4	27.8	1.0
	0.0	0.0	0.0	0.0	0.0
	0.0	0.0	0.0	0.0	0.0
	<b>Totals</b>			<b>mean</b>	<b>s.d. error</b>
<b>C10-C19</b>	84.8	74.5	77.3	78.8	3.1
<b>C20-C31</b>	248.0	224.4	235.4	235.9	6.8
<b>Total <i>n</i>-alkanes</b>	332.8	298.9	312.7	314.8	9.8
				<b>mean</b>	<b>s.d. error</b>
<b>Total Petroleum Hydrocarbons</b>	2612	1861	1960	2144	236
<b>Relative abundance of <i>n</i>-alkanes in crude oil (%)</b>	12.74	16.06	15.96	14.92	1.09
	<b>Statfjord crude oil</b>				

**Table S2. Estimation of the total H<sub>2</sub>S (nmoles) accumulated in sediments, which were incubated with either Duc or Statfjord crude oil, with snorkels (S) or glass controls (C), at 10 (HP) or 0.1 MPa (AP).**

Parameter	Unit of measurement	Duc oil				Statfjord oil			
		HPS	HPC	APS	APC	HPS	HPC	APS	APC
Onset of H <sub>2</sub> S increase	cm	0.9	0.3	0.1	0.1	0.0	0.0	0.0	0.0
End of H <sub>2</sub> S linear increase	cm	2.0	2.0	2.0	2.0	3.5	1.5	3.5	1.5
Net depth of the sediment rich in H <sub>2</sub> S	cm	1.1	1.7	1.9	1.9	3.5	1.5	3.5	1.5
Internal surface area of the glass cylinder	cm <sup>2</sup>	11.3	11.3	11.3	11.3	11.3	11.3	11.3	11.3
Volume of sediment rich in H <sub>2</sub> S	cm <sup>3</sup> or mL	12.5	19.3	21.5	21.5	39.7	17.0	39.7	17.0
Minimum H <sub>2</sub> S concentration	μM or nmoles/mL	0.0	0.0	0.0	0.0	5.7	8.3	15.4	28.7
Maximum H <sub>2</sub> S concentration	μM or nmoles/mL	2.3	15.6	19.4	57.7	15.3	83.8	52.6	136.2
Average H <sub>2</sub> S concentration in the sediment section	μM or nmoles/mL	1.2	7.8	9.7	28.8	4.8	37.7	18.6	53.7
Amount of H <sub>2</sub> S in the sediment section	nmoles	14.3	150.6	208.6	621.1	191.2	641.4	738.3	913.5
Onset of H <sub>2</sub> S increase	cm						1.5		1.5
End of H <sub>2</sub> S linear increase	cm						3.5		3.5
Net depth of the sediment rich in H <sub>2</sub> S	cm						2.0		2.0
Internal surface area of the glass cylinder	cm <sup>2</sup>						11.3		11.3
Volume of sediment rich in H <sub>2</sub> S	cm <sup>3</sup> or mL						22.7		22.7
Minimum H <sub>2</sub> S concentration	μM or nmoles/mL						38.4		75.9
Maximum H <sub>2</sub> S concentration	μM or nmoles/mL						83.8		136.2
Average H <sub>2</sub> S concentration in the sediment section	μM or nmoles/mL						22.7		30.1
Amount of H <sub>2</sub> S in the sediment section	nmoles						514.9		682.6
Estimated total amount of H <sub>2</sub> S accumulated in the sediment	nmoles	14.3	150.6	208.6	621.1	191.2	1156.3	738.3	1596.1

**Table S3. Estimates of the total (mg L<sup>-1</sup>) O<sub>2</sub> respiration in seawater due to electrochemical oxidation of H<sub>2</sub>S in sediments, and of the relative contribution (%) to the electrochemical O<sub>2</sub> respiration to the total, in marine sediments incubated with either Duc or Statfjord crude oil, with snorkels (S) or glass controls (C), at 10 (HP) or 0.1 MPa (AP).**

Parameter	Unit	Duc oil				Statfjord oil			
		HPS	HPC	APS	APC	HPS	HPC	APS	APC
Total amount of H <sub>2</sub> S in the sediment	nmoles	14.3	150.6	208.6	621.1	191.2	1156.3	738.3	1596.1
Net amount of H <sub>2</sub> S oxidized electrochemically	nmoles	136.3		412.6		965.1		857.8	
Net stoichiometric amount of O <sub>2</sub> respired electrochemically	μmoles	0.34		1.03		2.41		2.14	
Net stoichiometric amount of O <sub>2</sub> respired electrochemically	μg	10.9		33.0		77.2		68.6	
Volume of seawater in glass cylinders	mL	200		200		200		200	
Net stoichiometric O <sub>2</sub> concentration respired electrochemically	μg/mL or mg/L	0.05		0.17		0.39		0.34	
Total O <sub>2</sub> consumption in seawater	mg/L	40.5	40.4	39.6	38.1	33.1	29.4	37.2	32.4
Net amount of O <sub>2</sub> respired electrochemically	mg/L	0.13		1.50		3.71		4.84	
Relative amount of O <sub>2</sub> respired due to snorkels	%	41.9		11.0		10.4		7.1	



