

Synthesis of Brominated Alkanes via Heterogeneous Catalytic Distillation over $\text{Al}_2\text{O}_3/\text{SO}_4^{2-}/\text{ZrO}_2$

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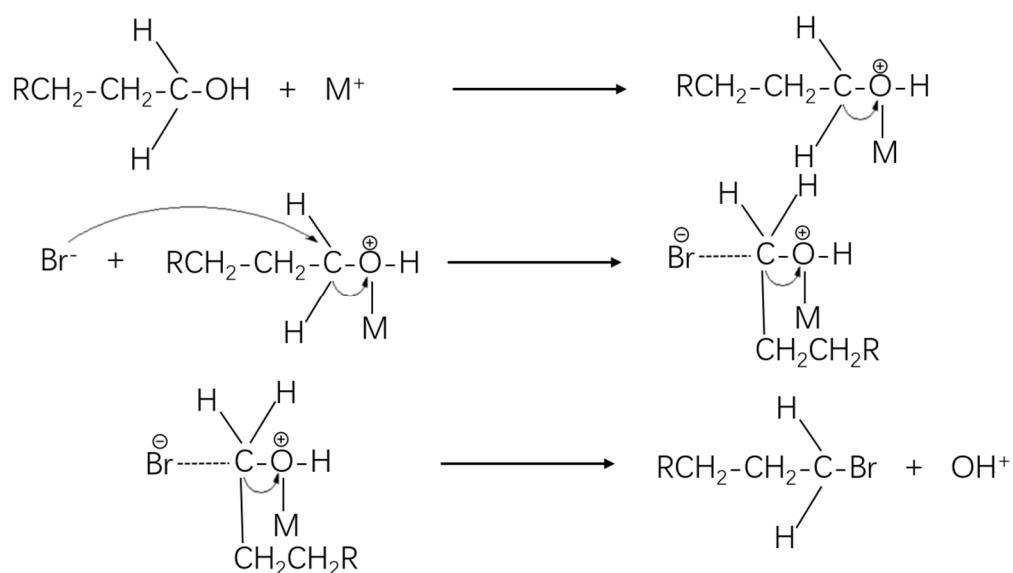


Figure S1. $\text{SO}_4^{2-}/\text{M}_x\text{O}_y$ catalytic reaction mechanism

Table S1. TOF of the $\text{Al}_2\text{O}_3/\text{SO}_4^{2-}/\text{ZrO}_2$ catalyst and comparison with other catalysts.

Catalyst	T/°C	TOF/h ⁻¹	Yield/%	Ref.
$\text{H}_3\text{PW}_{12}\text{O}_{40}\text{-}[\text{bmim}][\text{FeCl}_4]$	135	12.5	83	[1]
$\text{SO}_4^{2-}/\text{ZrO}_2\text{-ZnO}$	130	1.79	70.22	[2]
methyl trioctyl ammonium chloride	110	0.83	72.8	[3]
H_2SO_4	70-130	4.17	93.2	[4]
$\text{Al}_2\text{O}_3/\text{SO}_4^{2-}/\text{ZrO}_2$	110	13.33	96.1	This work

Table S2. Conversion substrate application of the $\text{Al}_2\text{O}_3/\text{SO}_4^{2-}/\text{ZrO}_2$ catalyst

Substrate	T/°C	Time/h	Yield/%
Ethanol	90	3	93.72
N-propanol	110	3	96.18
Isopropanol	110	3	95.55
N-butanol	120	4	92.26

References:

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