

Non-Solvent Synthesis of a Robust Potassium-Doped PdCu-Pd-Cu@C Nanocatalyst for High Selectively Tandem Reactions

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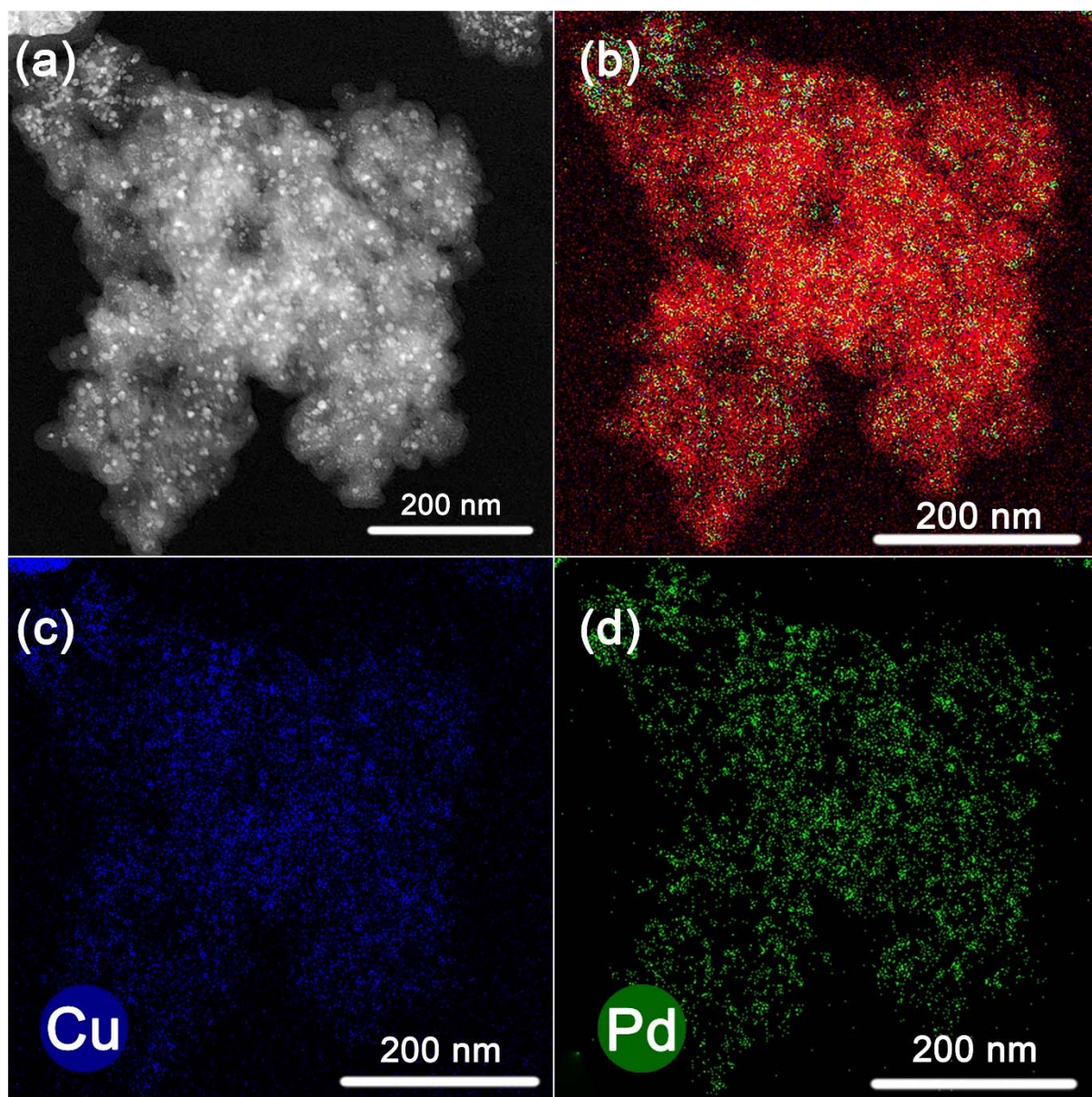


Figure S1. HAADF-STEM images of (a) PdCu@C nanocatalyst. Elemental mapping image of (b) total element (red: carbon), (c) copper (blue), and (d) palladium (green), respectively.

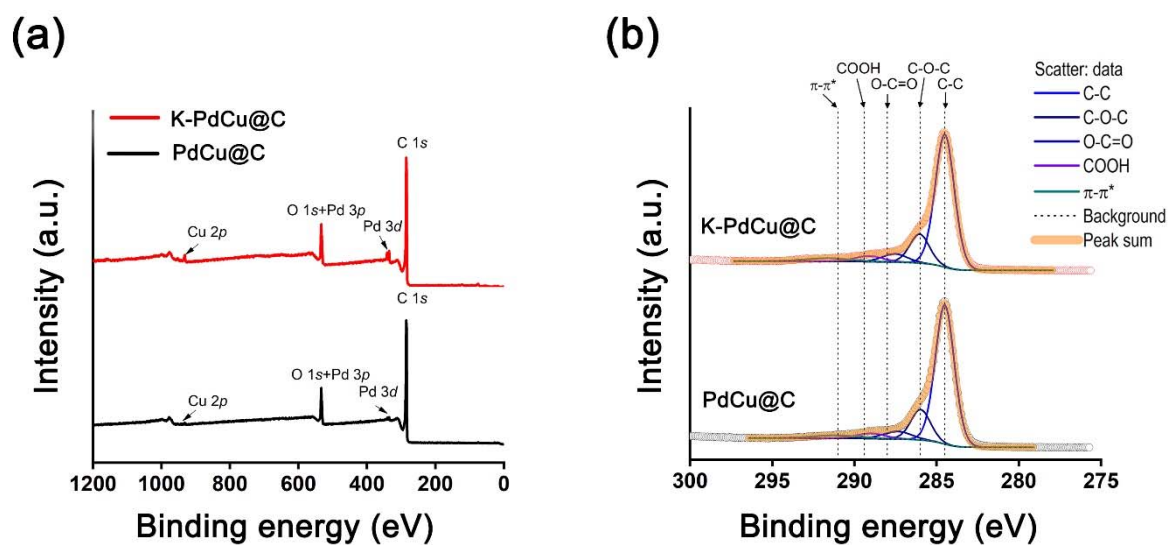


Figure S2. Core level X-ray photoelectron spectrum of (a) survey scan and (b) C 1s. The vertical dotted lines indicate the reference peak positions. The solid curves indicate deconvolution results.

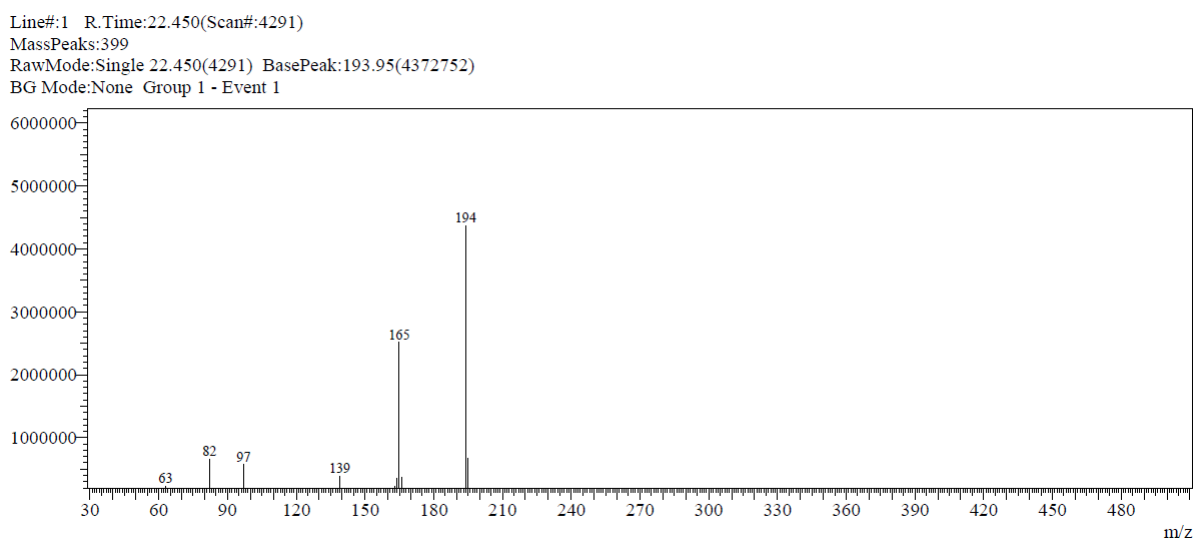


Figure S3. Mass spectra of 2-phenylbenzofuran.

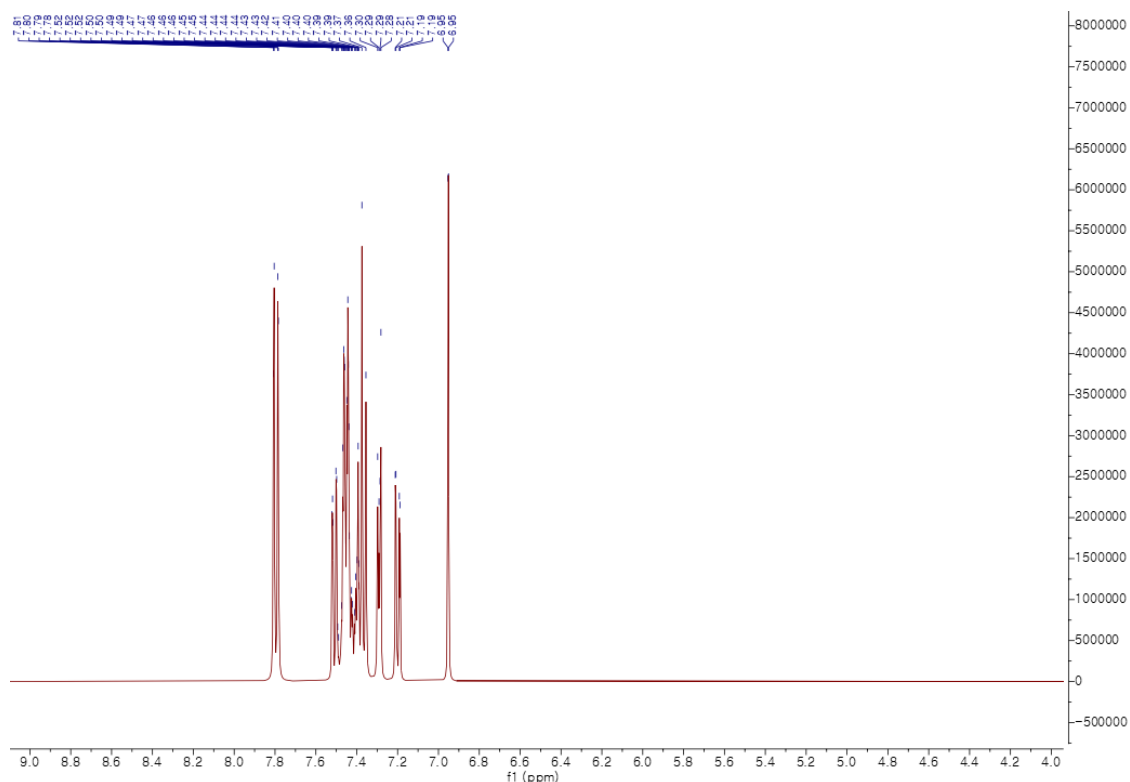


Figure S4. Proton NMR spectra of 2-phenylbenzofuran.

Table S1. Complementary analysis of elements in PdCu@C and K doped PdCu@C nanocatalysts using inductively coupled plasma optical emission spectrometers.

	Palladium (wt%)	Copper (wt%)	Potassium (wt%)
PdCu@C	21.8	13.5	0
K doped PdCu@C	19.9	13.3	0.47

Table S2. Complementary analysis of elements in after 5 times recovered K doped PdCu@C nanocatalyst using inductively coupled plasma optical emission spectrometers

	Palladium (wt%)	Copper (wt%)	Potassium (wt%)
Reused K doped PdCu@C	23.1	17.4	0.59