

Supplementary Materials

CO total and preferential oxidation over stable Au/TiO₂ catalysts derived from preformed Au nanoparticles

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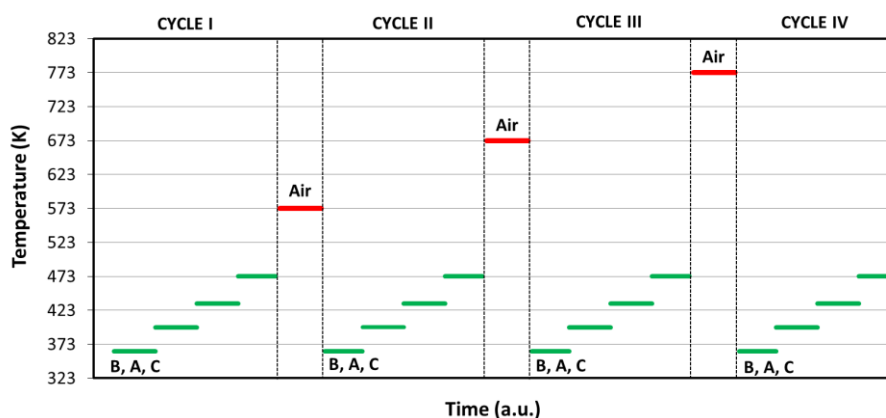


Figure S1. Sequence of experimental conditions performed, highlighting the order of the reaction mixtures, temperatures and details of the oxidations performed.

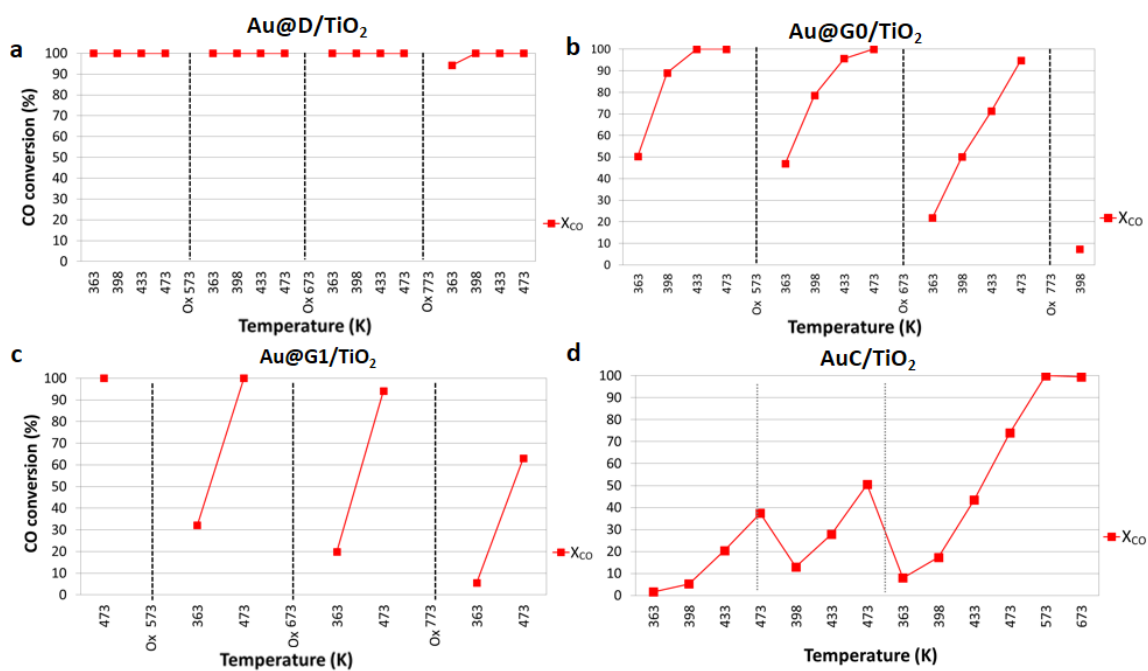


Figure S2. Temperature dependence performance of (a) Au@D/TiO₂, (b) Au@G0/TiO₂, (c) Au@G1/TiO₂, and (d) AuC/TiO₂ catalysts for mixture A. Black vertical lines in (a)-(c) indicate the oxidations performed; grey lines in (d) show when mixture B was dosed at a higher temperature.

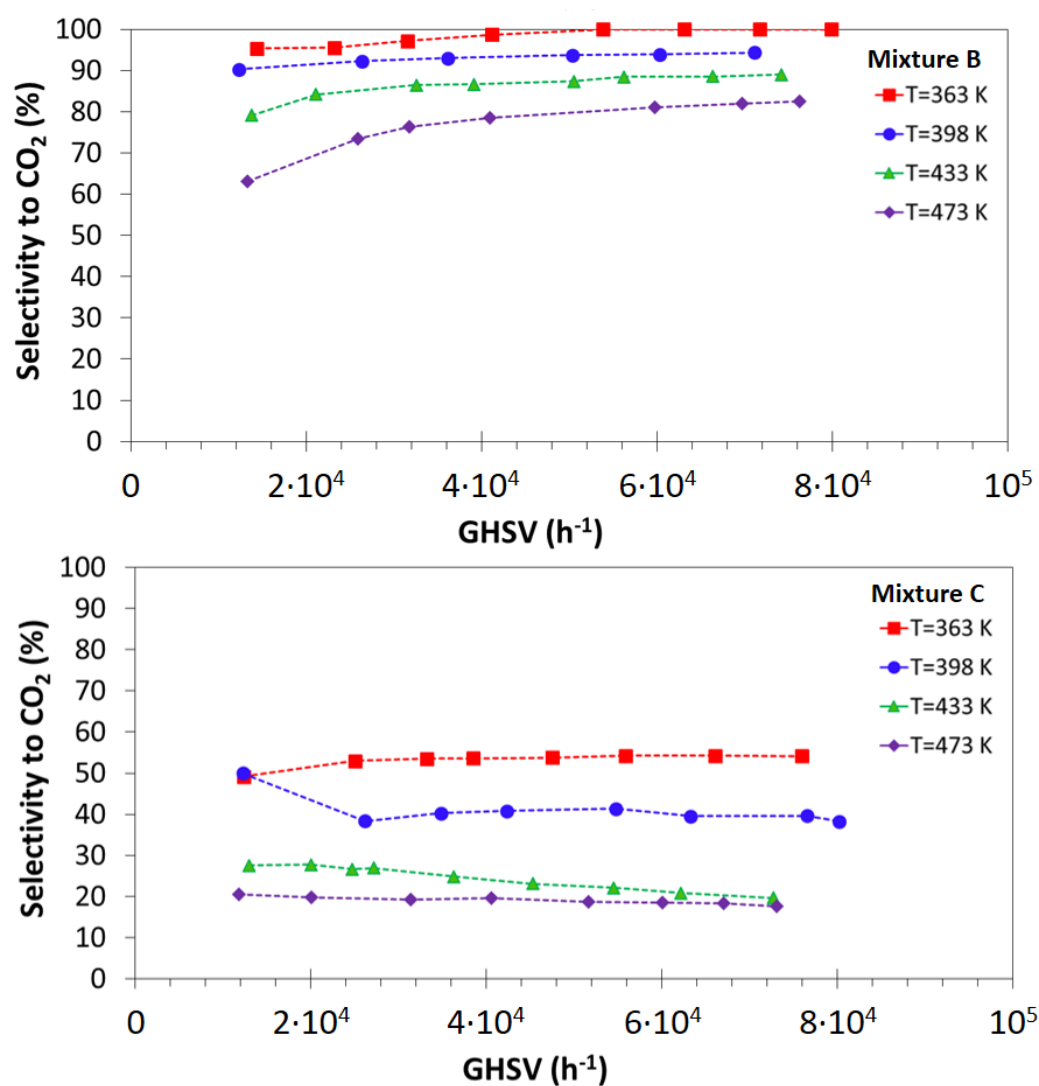


Figure S3. CO₂ selectivity for the feed load tests carried out for mixtures B and C on the microreactor functionalized with Au@D/TiO₂.

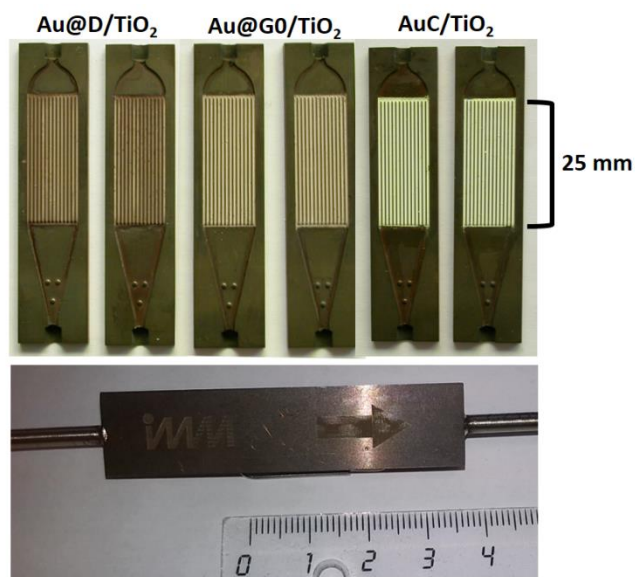


Figure S4. Upper photographs: Detail of the inner part of the microreactors showing the microchannels coated with TiO₂ and three different Au precursors: Au@D, Au@G0, and HAuCl₄·3H₂O. Lower photograph: detail of an assembled microreactor. The scale is shown in cm.



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