

Ru-catalyzed repetitive batch borylative coupling of olefins in ionic liquids or ionic liquids/scCO₂ systems

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1. General information:

1.1. NMR analyses

^1H and ^{13}C NMR, spectra were recorded at 25 °C on Bruker UltraShield 300 MHz. Chemical shifts were reported in ppm with the reference to the residue portion solvent peak. Chloroform- d_1 or acetone- d_6 were used as solvents and for internal deuterium lock. The multiplicities were reported as follows: singlet (s), doublet (d), a doublet of triplets (dt), multiplet (m), triplet (t) and broad resonances (br).

1.2. GC-MS analysis

The mass spectra of the products were obtained by GC-MS analysis on a Bruker Scion 436-GC with a 30m Varian DB-5 0.25mm capillary column and a Scion SQ-MS mass spectrometry detector. Two temperature programs were used a) 60 °C (3 min), 10°C/min, 250 °C (30 min), b) 100 °C (3 min), 10°C/min, 280 °C (44.5 min).

1.3. ICP-MS analysis

Caution: All manipulation should be carried out under the fume hood. Serious risk of chemical burns. 50 mg of product from the extraction was placed into 100 mL Teflon beaker equipped with a magnetic stirring bar and 30 mL of concentrated HNO_3 was added. The mixture was heated up to complete evaporation of the solvent. In the same manner, the mixture was treated with aqua regia and 20% solution of HF. Subsequently, 20 mL of concentrated sulphuric acid was added and the solution was heated up to partial evaporation of H_2SO_4 . Then, the mixture was diluted with deionized water, filtered (if necessary) and analyzed by inductively coupled plasma-mass spectroscopy (ICP-MS). The analysis was performed on a quadrupole mass spectrometer with excitation in an inductively coupled plasma (Perkin-Elmer Nexion 300D), which was optimized prior to measurements using the appropriate SETUP SOLUTION from Perkin-Elmer. Samples were analyzed in liquid form using an injection system consisting of an concentric nebulizer and cyclonic spray. Data were processed by NexION™ v. 1.0 software.

2. Catalyst leaching in selected catalytic runs

Table S1. Ruthenium content in selected extracts.

	Ru content in extract [ppm]						
	cat@[BMIm][TfO]		cat@[EMPy][Tf ₂ N]		cat@[BMIm][TfO]/scCO ₂ ^b		
Olefin	(2a)	(2d)	(2a)	(2d)	(2a)	(2d)	(2j)
Run 1	7.1 ^a , 0.77 ^b	6.8, 0.64 ^b	7.0 ^a , 0.69 ^b	6.9 ^a , 0.75 ^b	0.88	0.74	0.77
2	6.8 ^a	-	6.5 ^a	-	0.74	-	-
5	6.4 ^a	-	6.7 ^a	-	0.75	-	-
8	4.6 ^a , 0.68 ^b	5.1	4.7 ^a , 0.71 ^b	5.1 ^a	0.54	0.61	0.59

a) *n*-Heptane extraction. b) ScCO₂ extraction

3. Spectra of isolated products

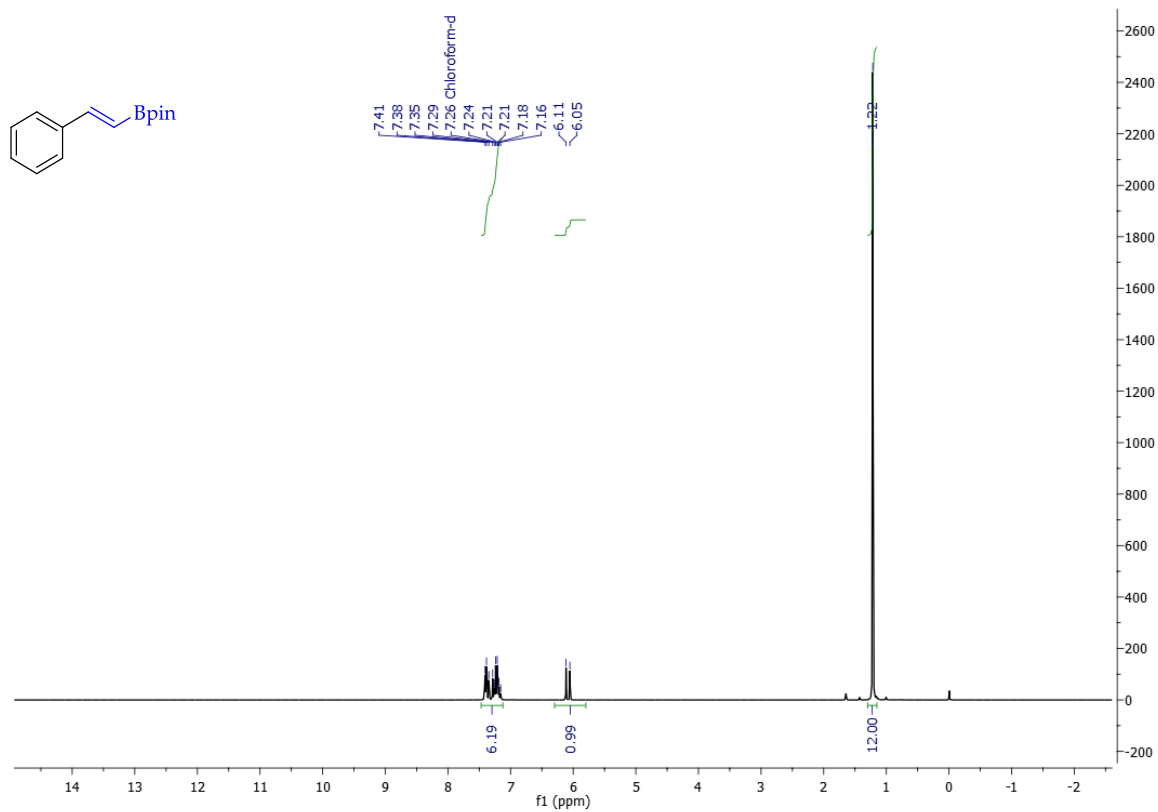


Figure S1. ¹H NMR spectrum of 3a

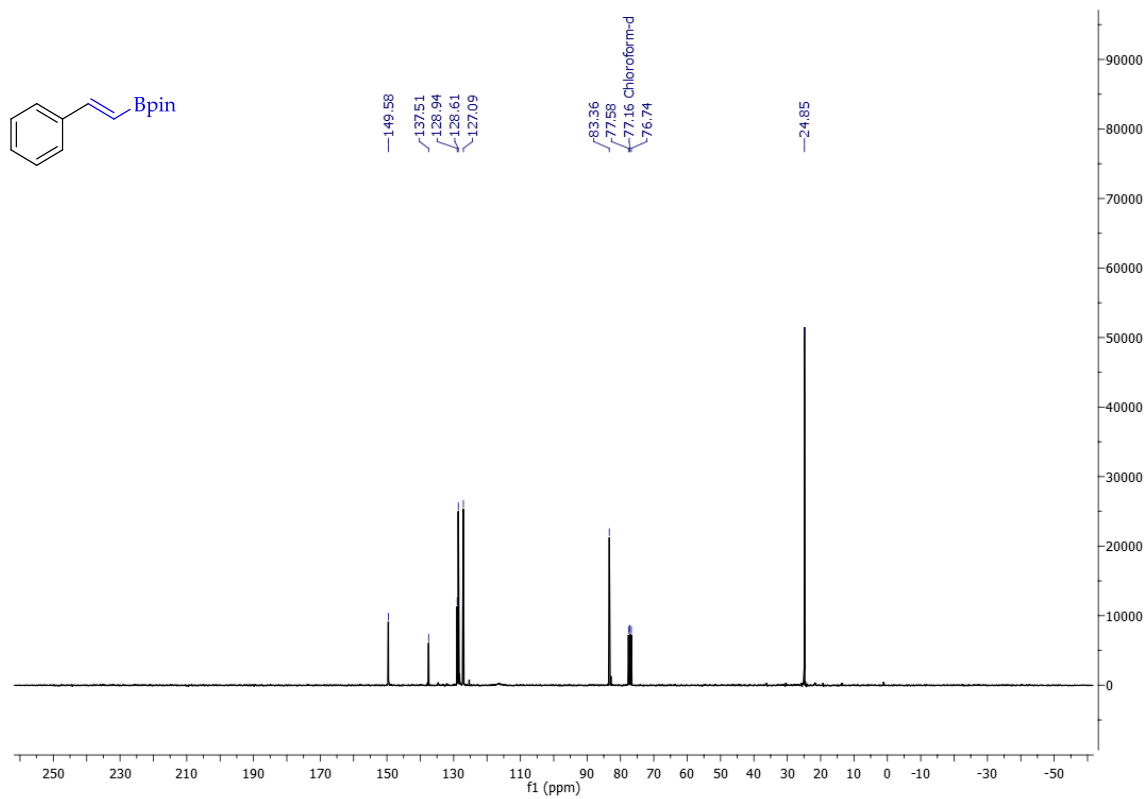


Figure S2. ¹³C NMR spectrum of 3a

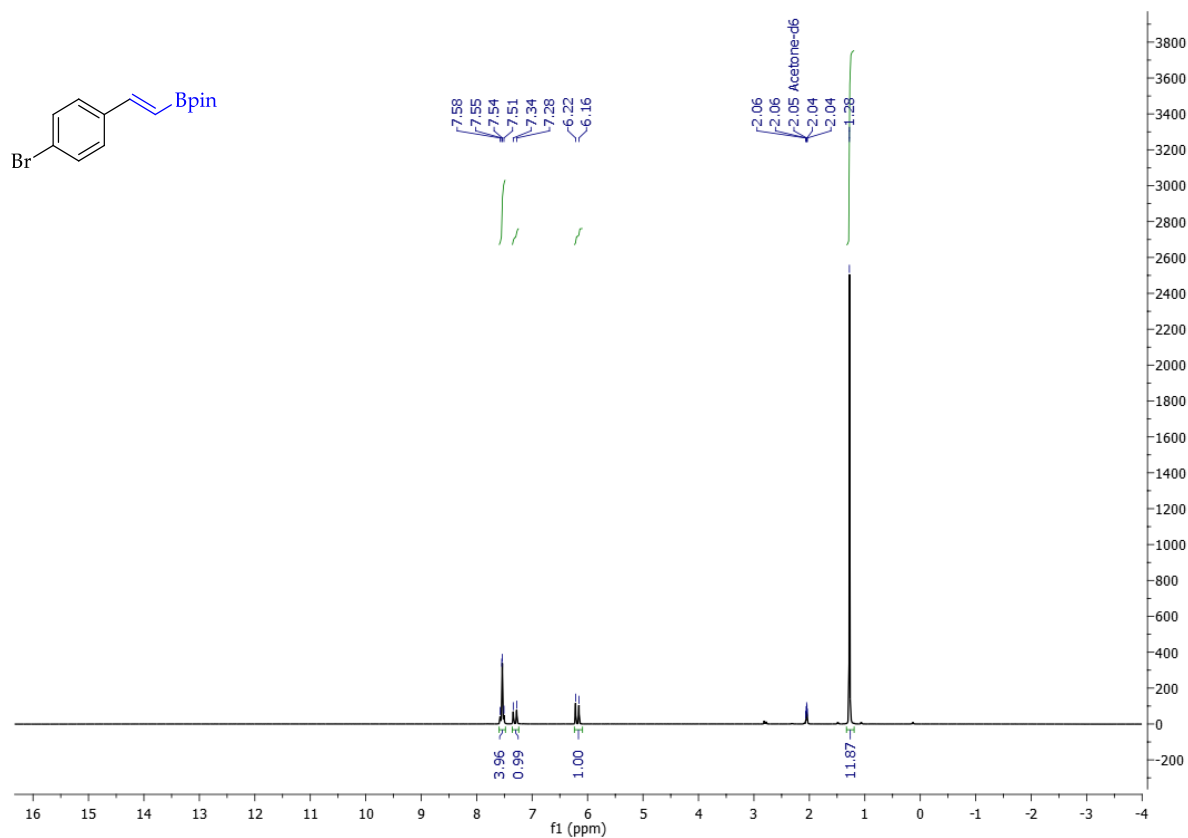


Figure S3. ^1H NMR spectrum of **3b**

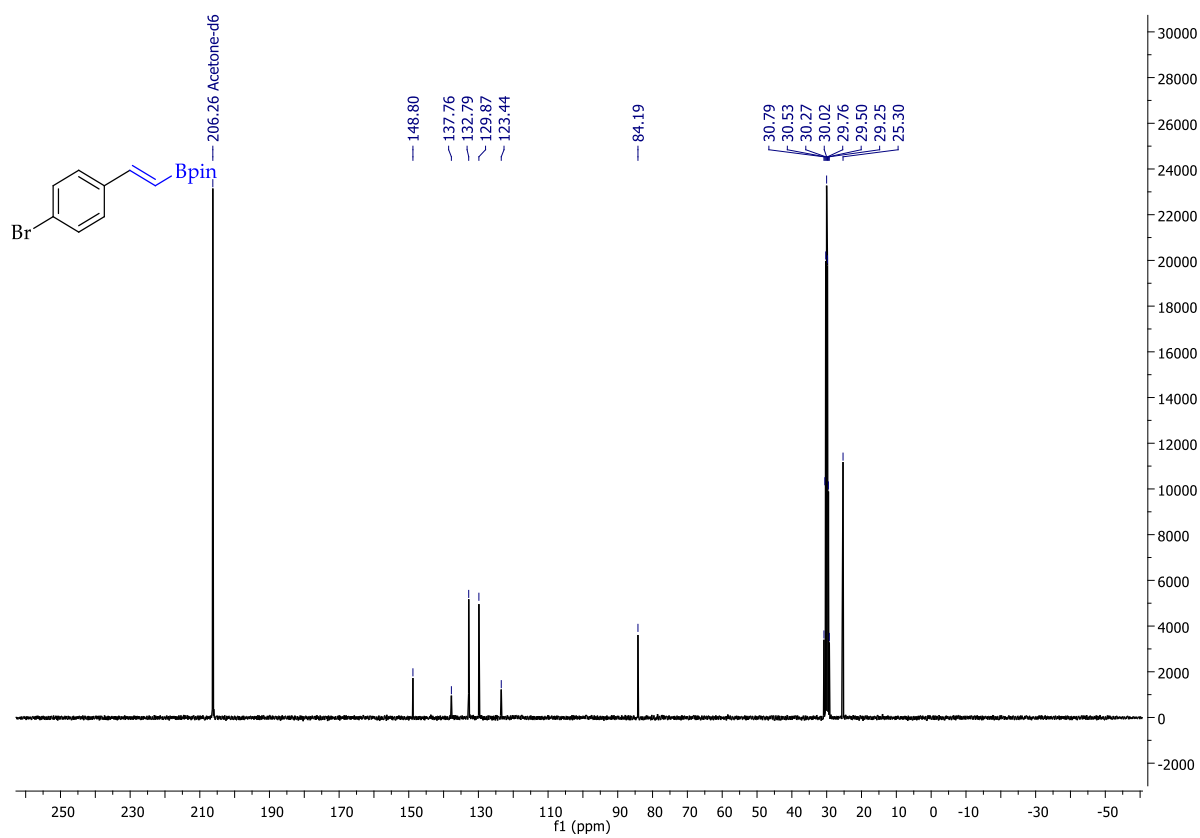


Figure S4. ^{13}C NMR spectrum of **3b**

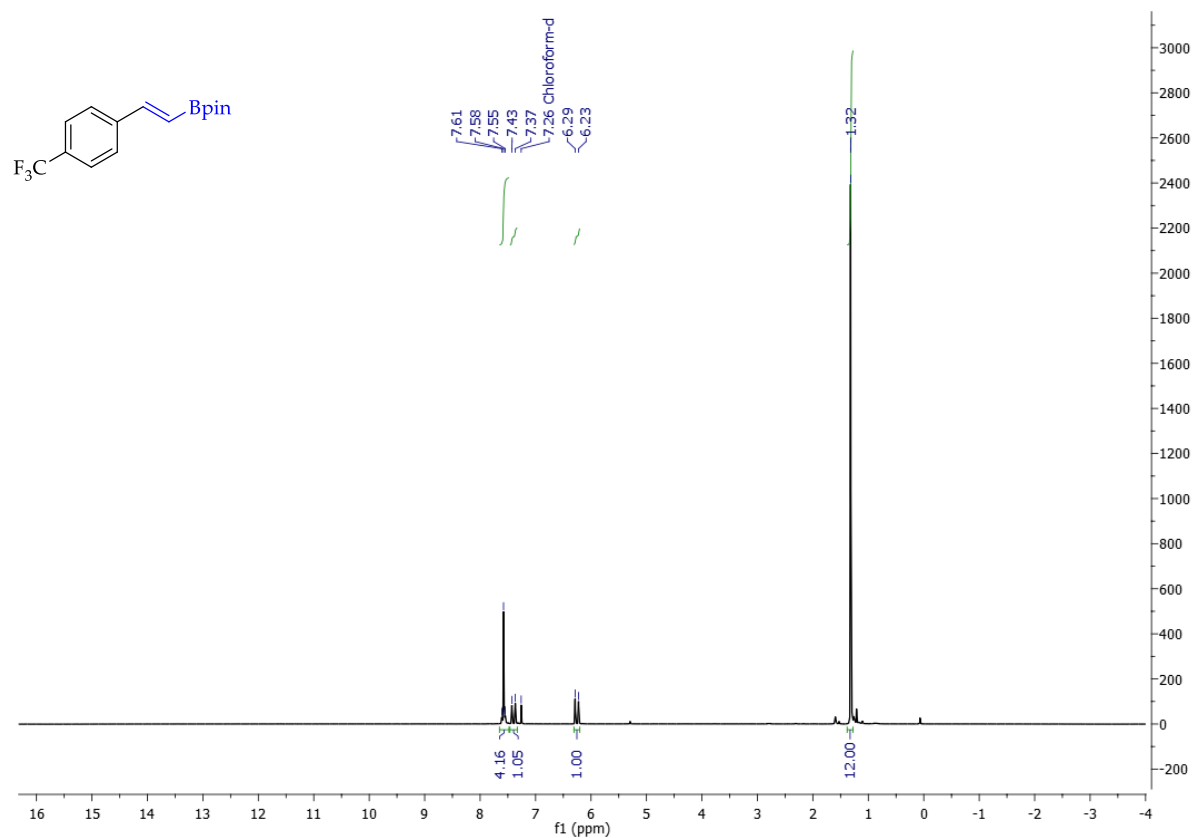


Figure S7. ¹H NMR spectrum of **3d**

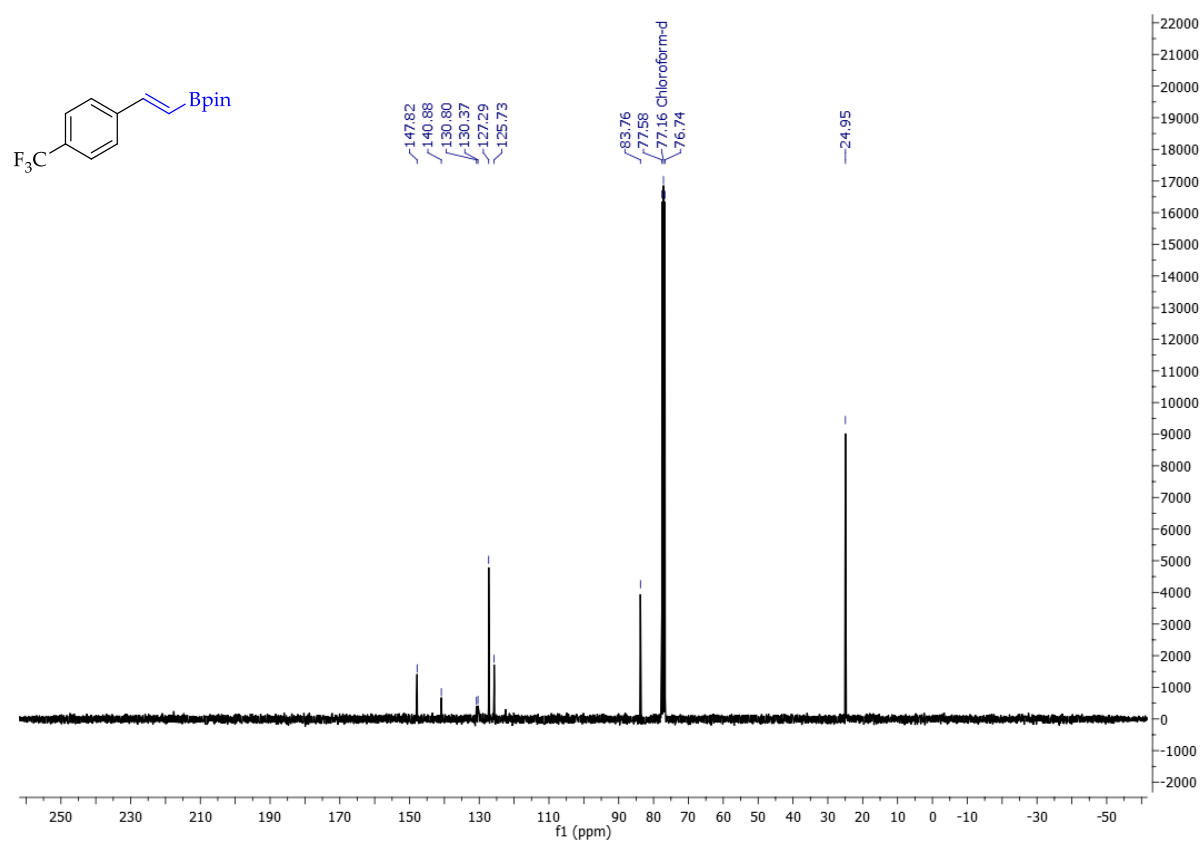


Figure S8. ¹³C NMR spectrum of **3d**

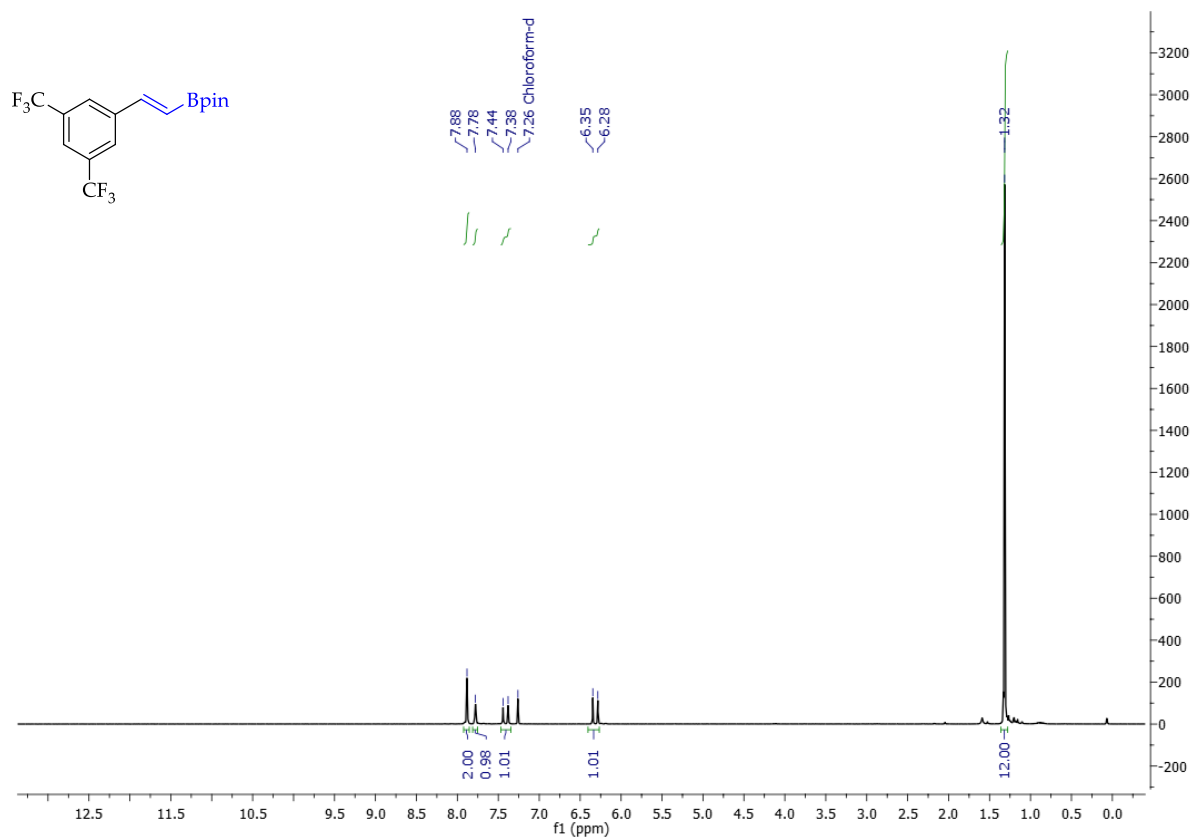


Figure S9. ¹H NMR spectrum of **3e**

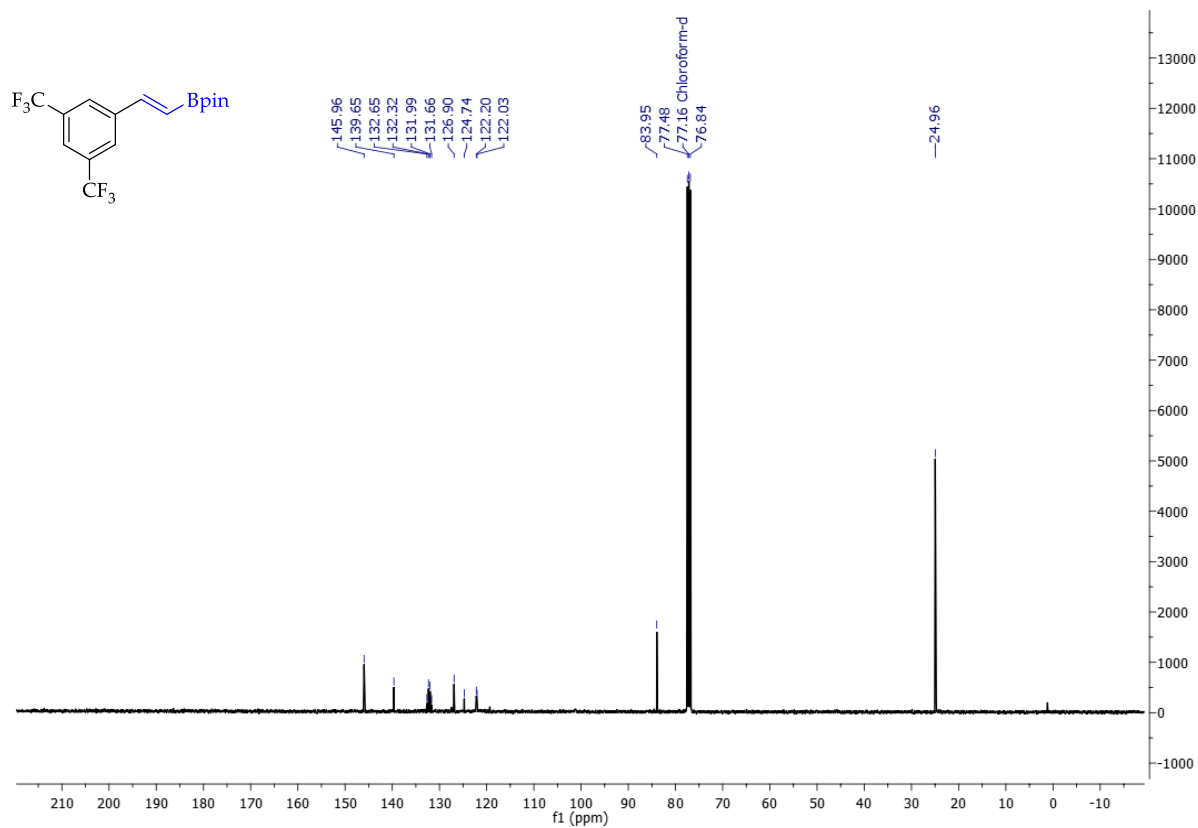


Figure S10. ¹³C NMR spectrum of **3e**

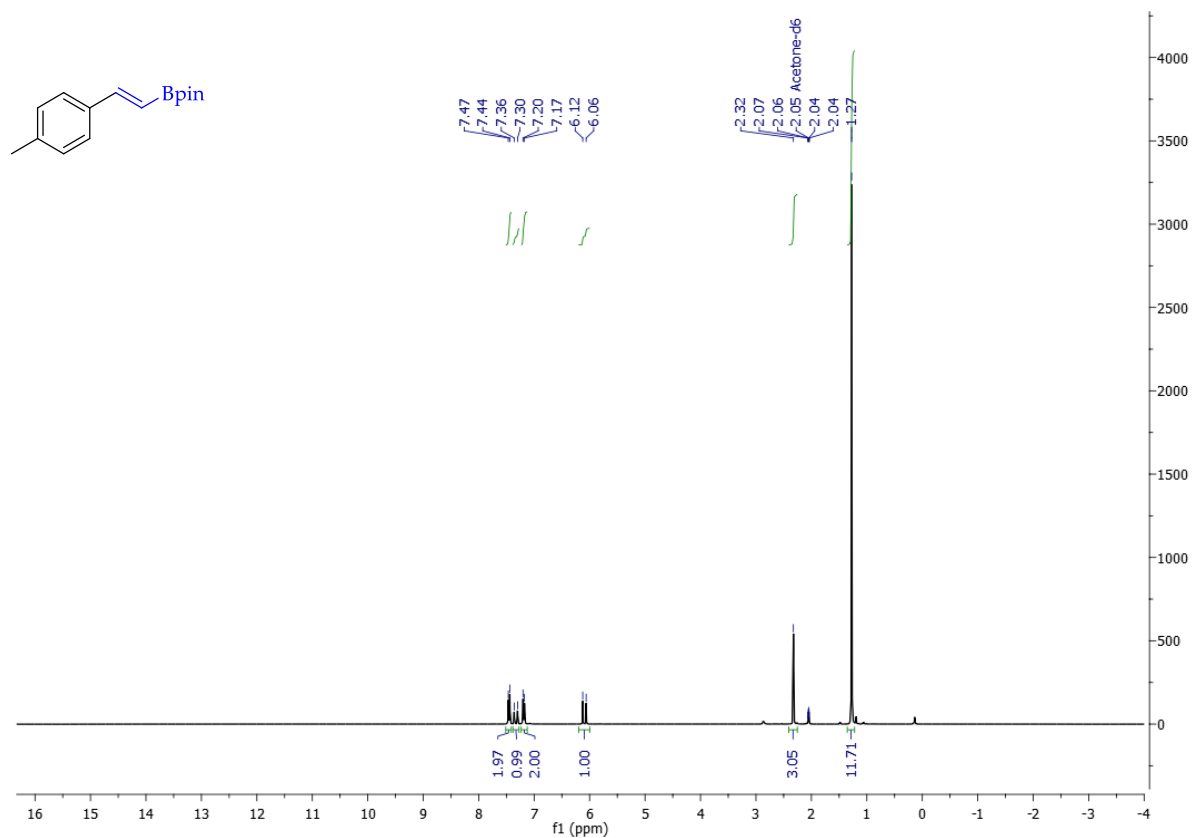


Figure S11. ¹H NMR spectrum of 3f

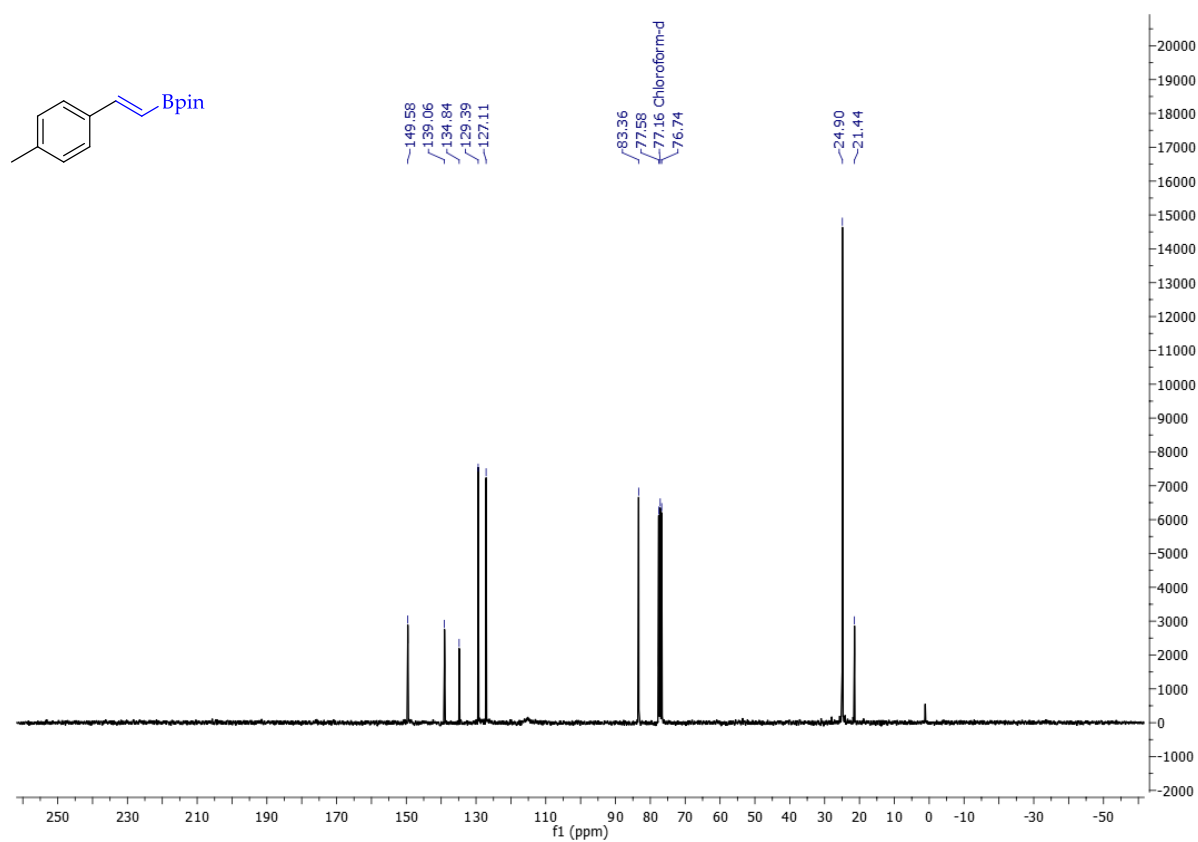
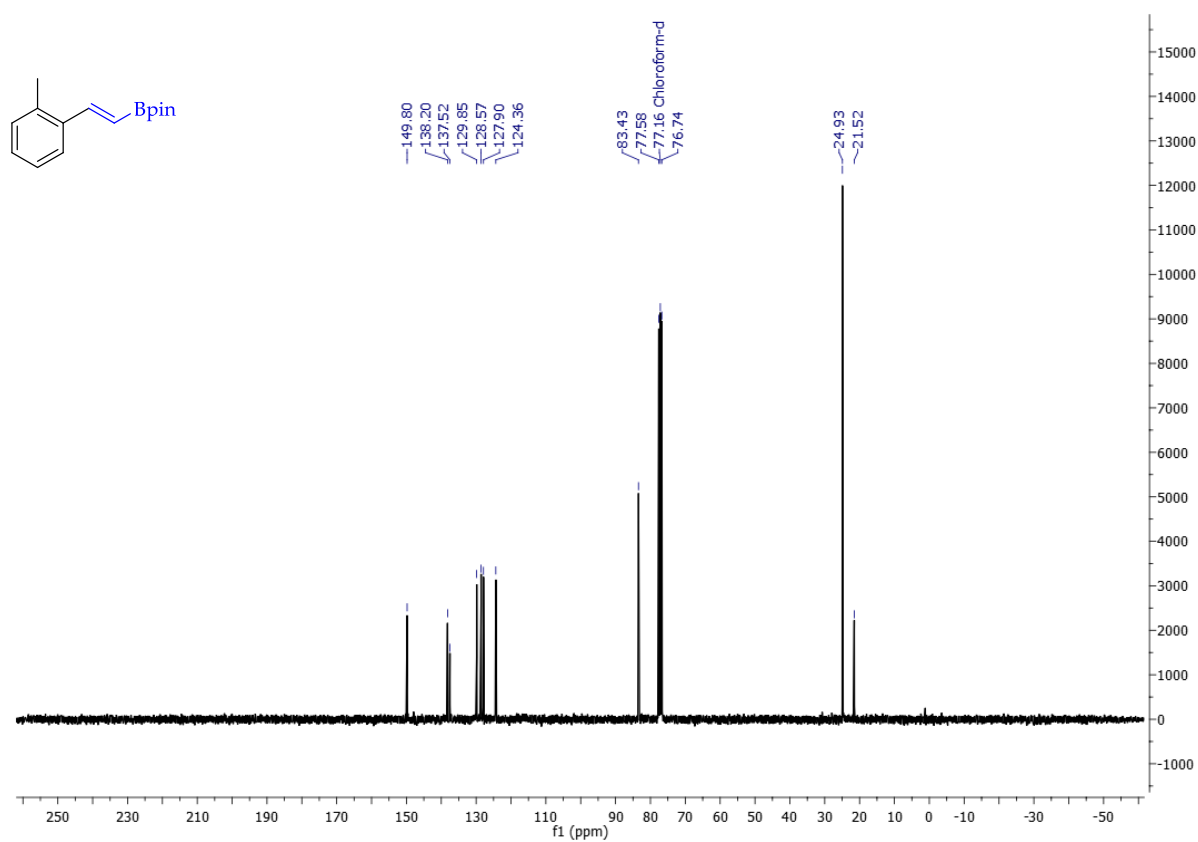
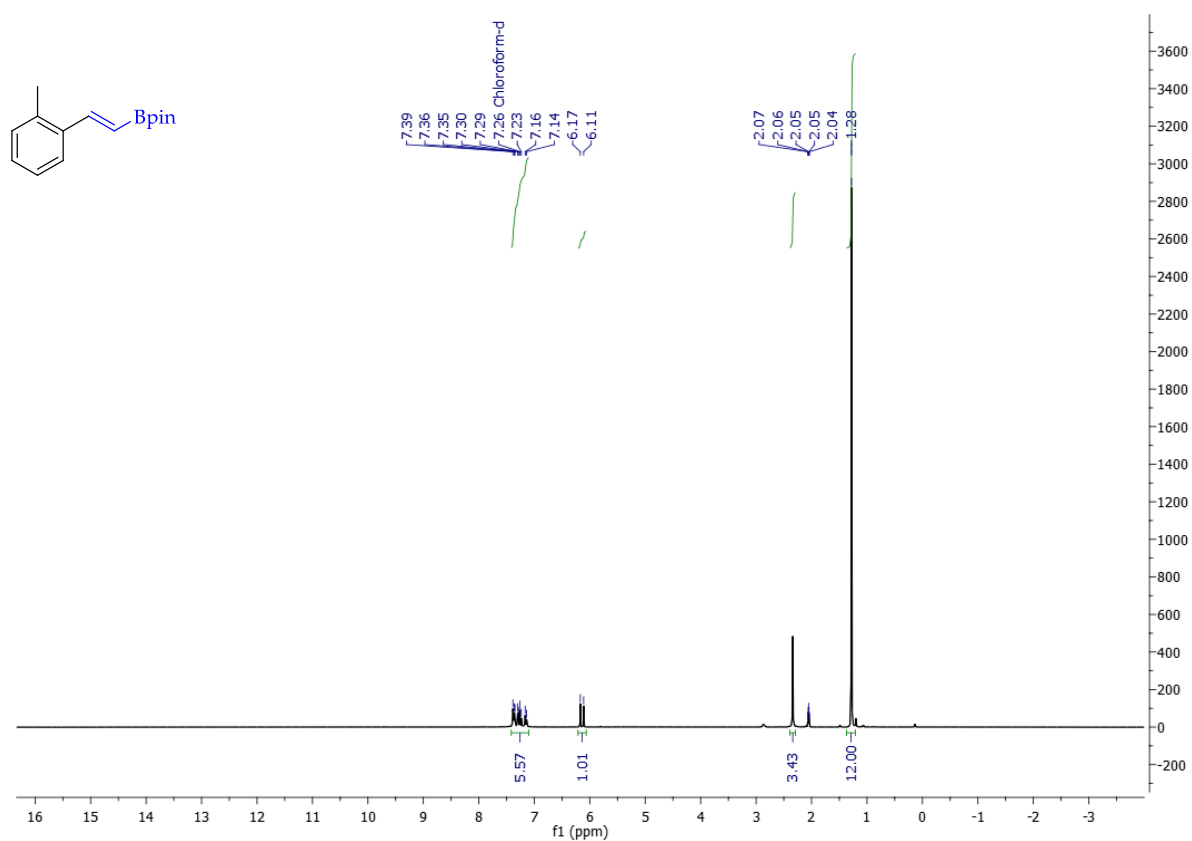


Figure S12. ¹³C NMR spectrum of 3f



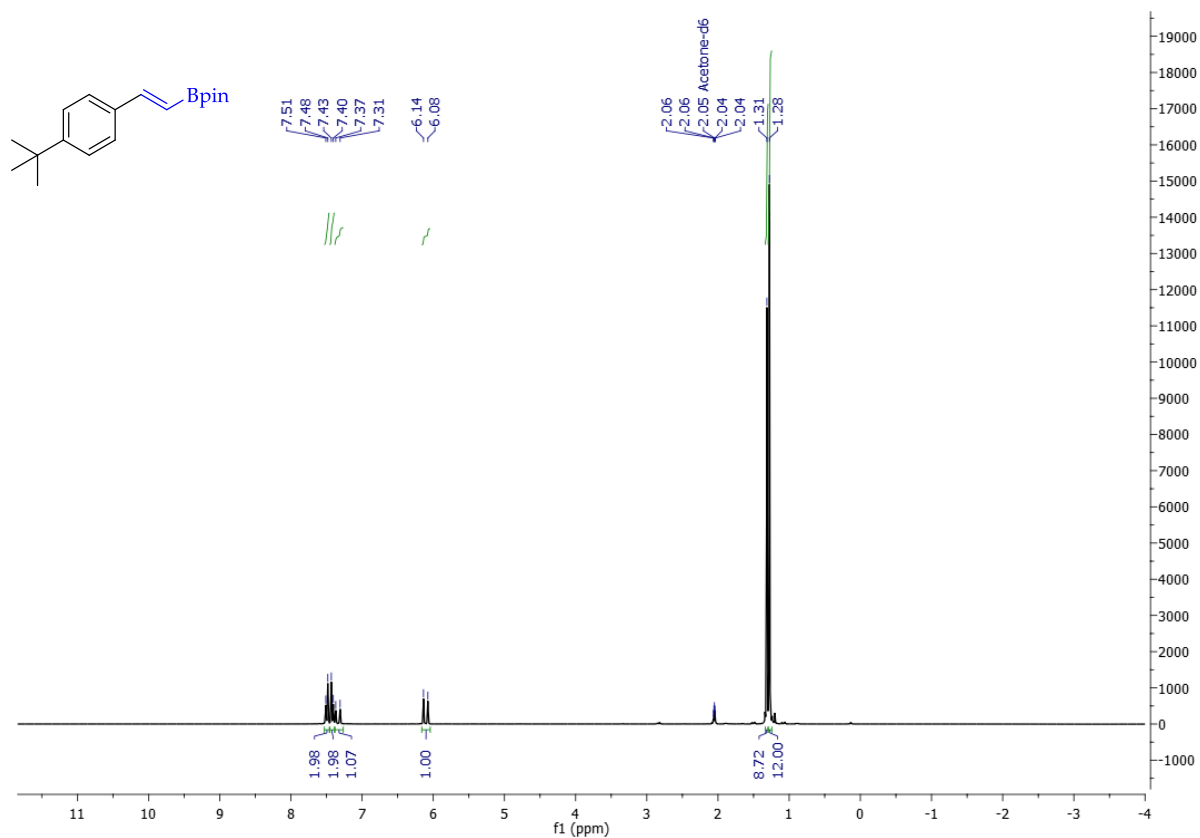


Figure S15. ¹H NMR spectrum of 3h

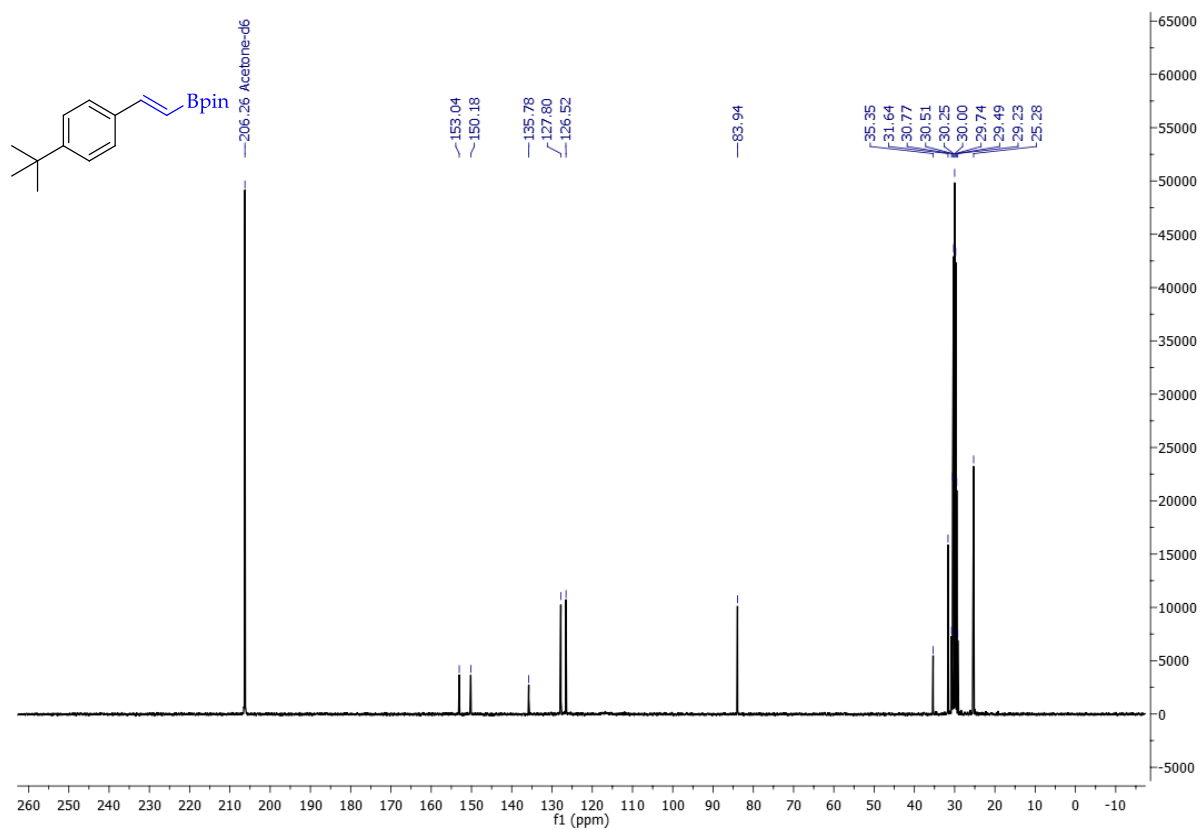


Figure S16. ¹³C NMR spectrum of 3h

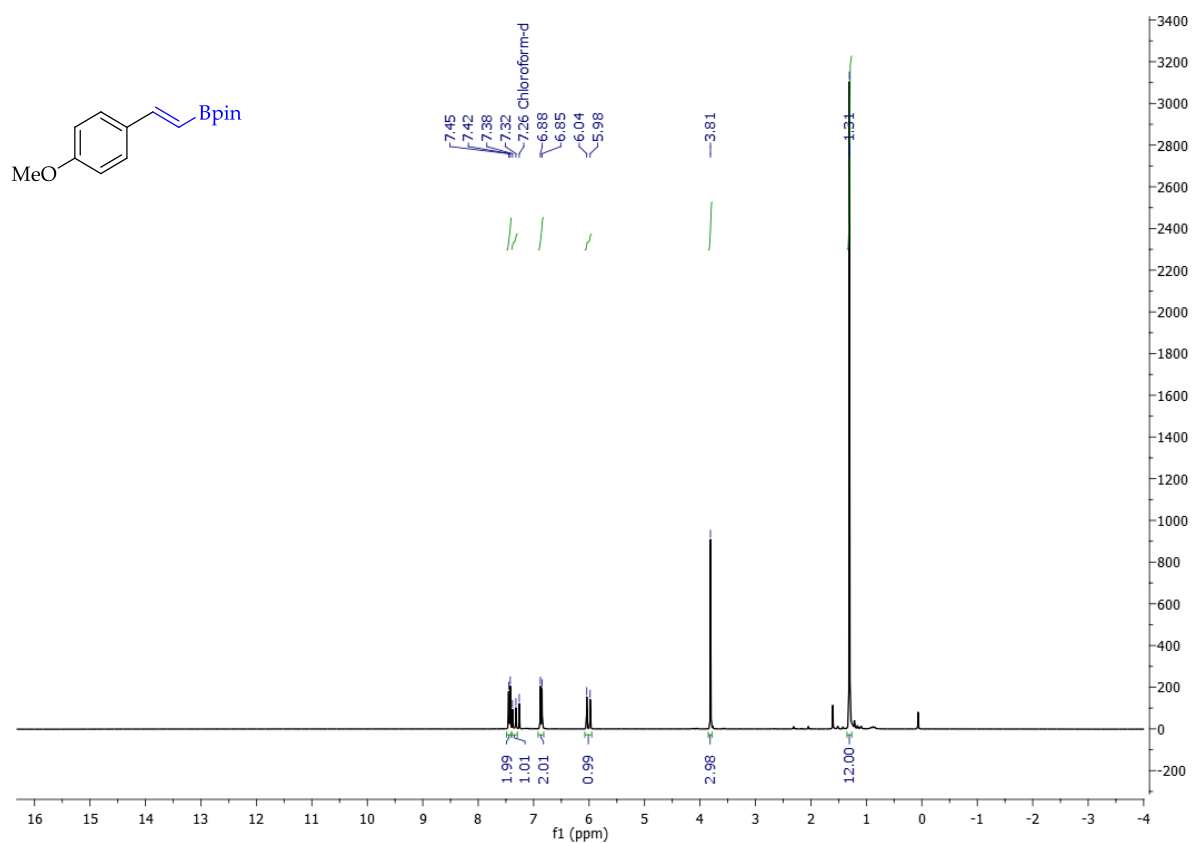


Figure S17. ¹H NMR spectrum of 3i

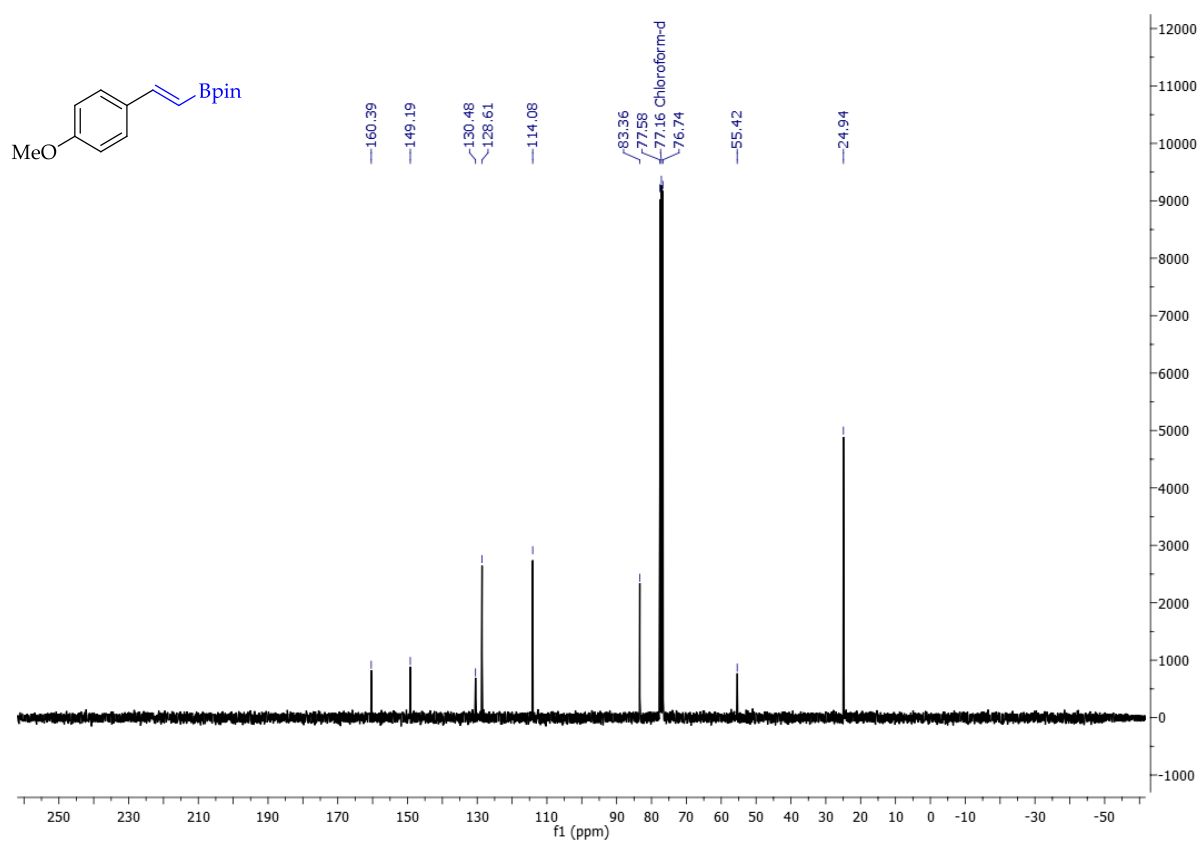


Figure S18. ¹³C NMR spectrum of 3i

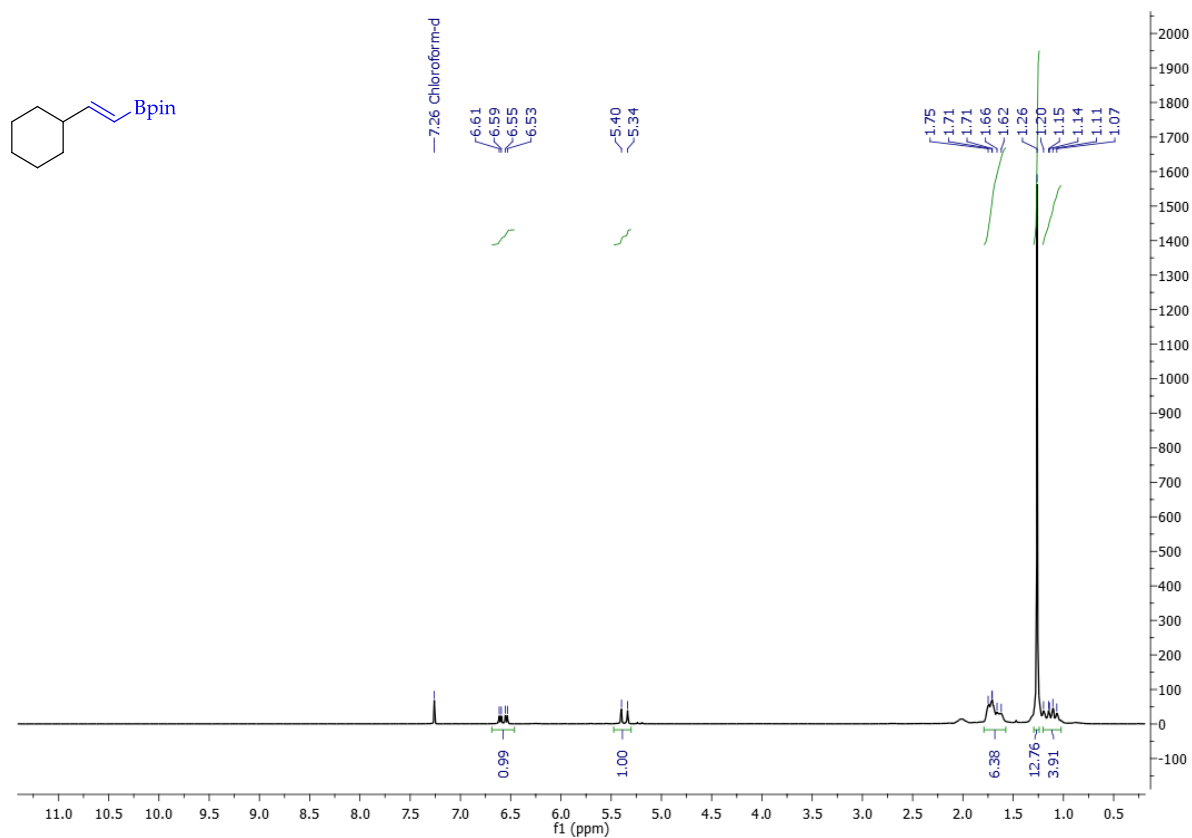


Figure S19. ¹H NMR spectrum of 3j

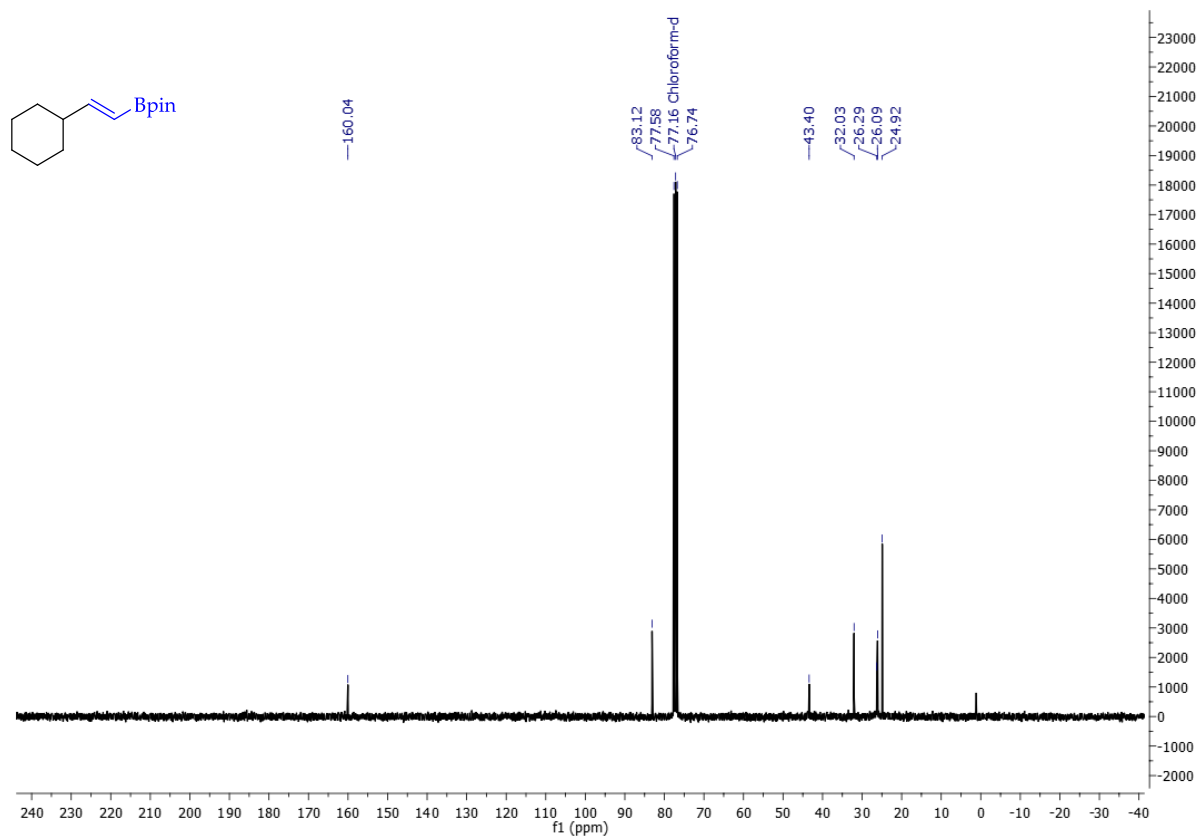


Figure S20. ¹³C NMR spectrum of 3j

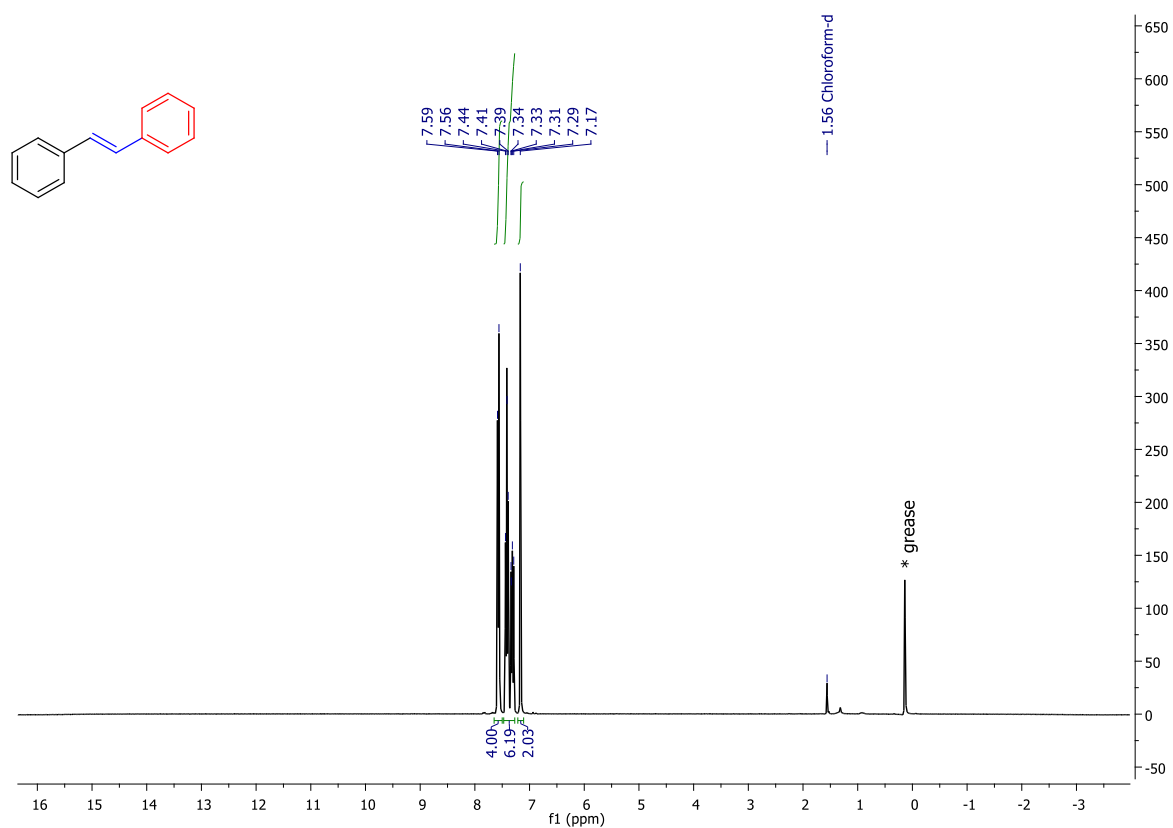


Figure S21. ¹H NMR spectrum of 4

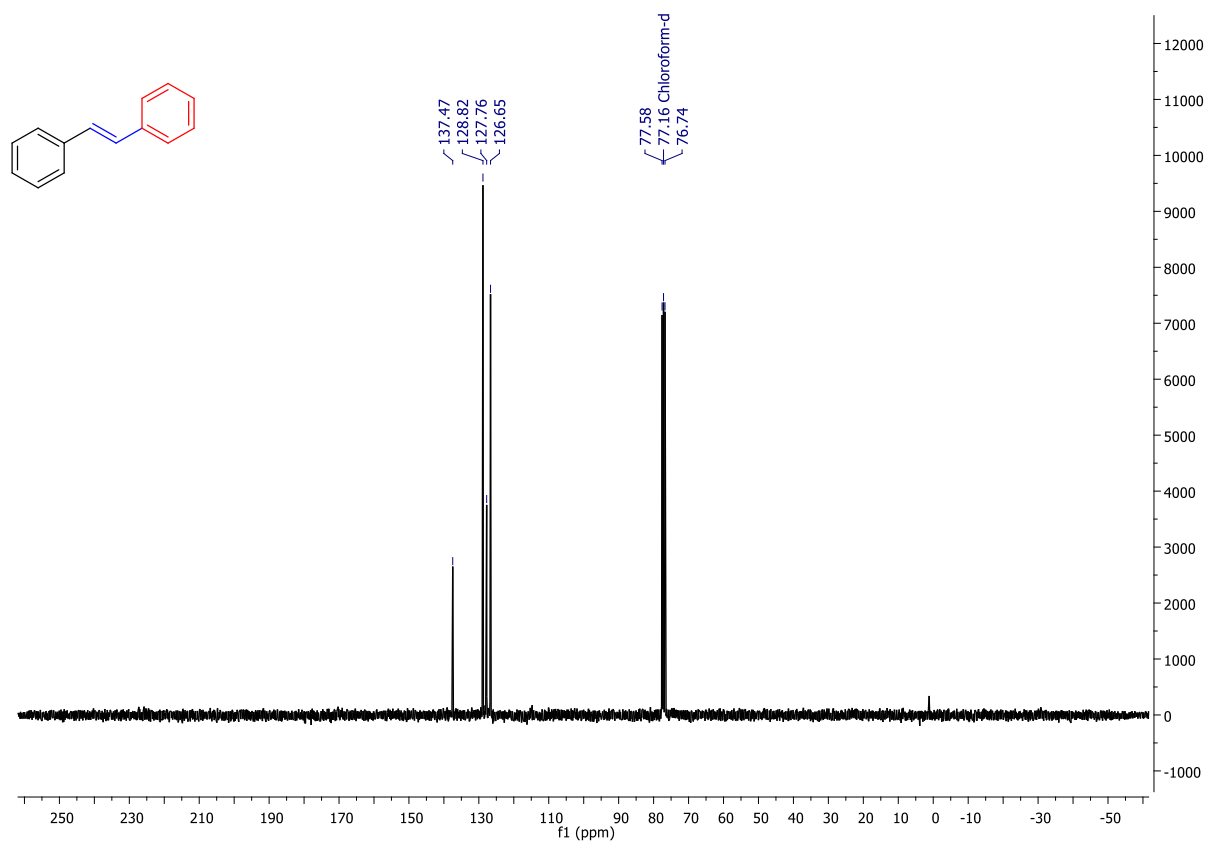


Figure S22. ¹³C NMR spectrum of 4

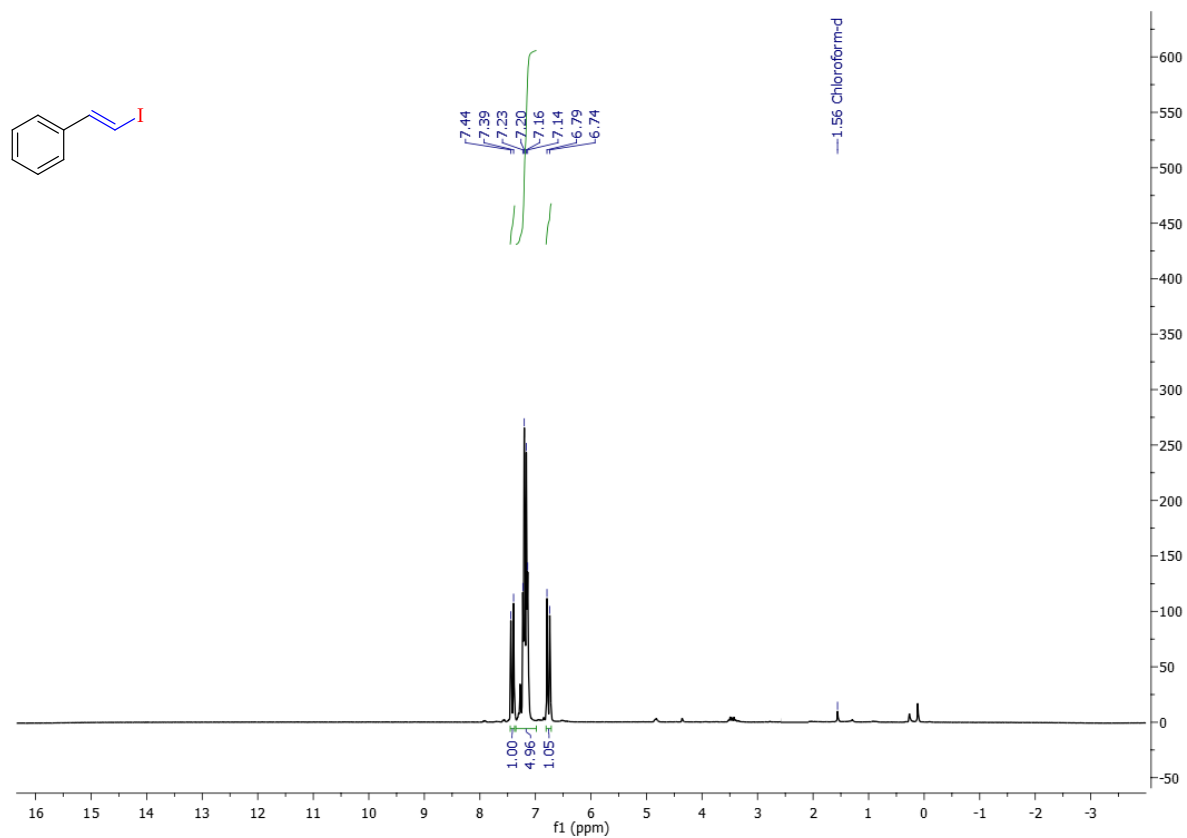


Figure S23. ¹H NMR spectrum of 5

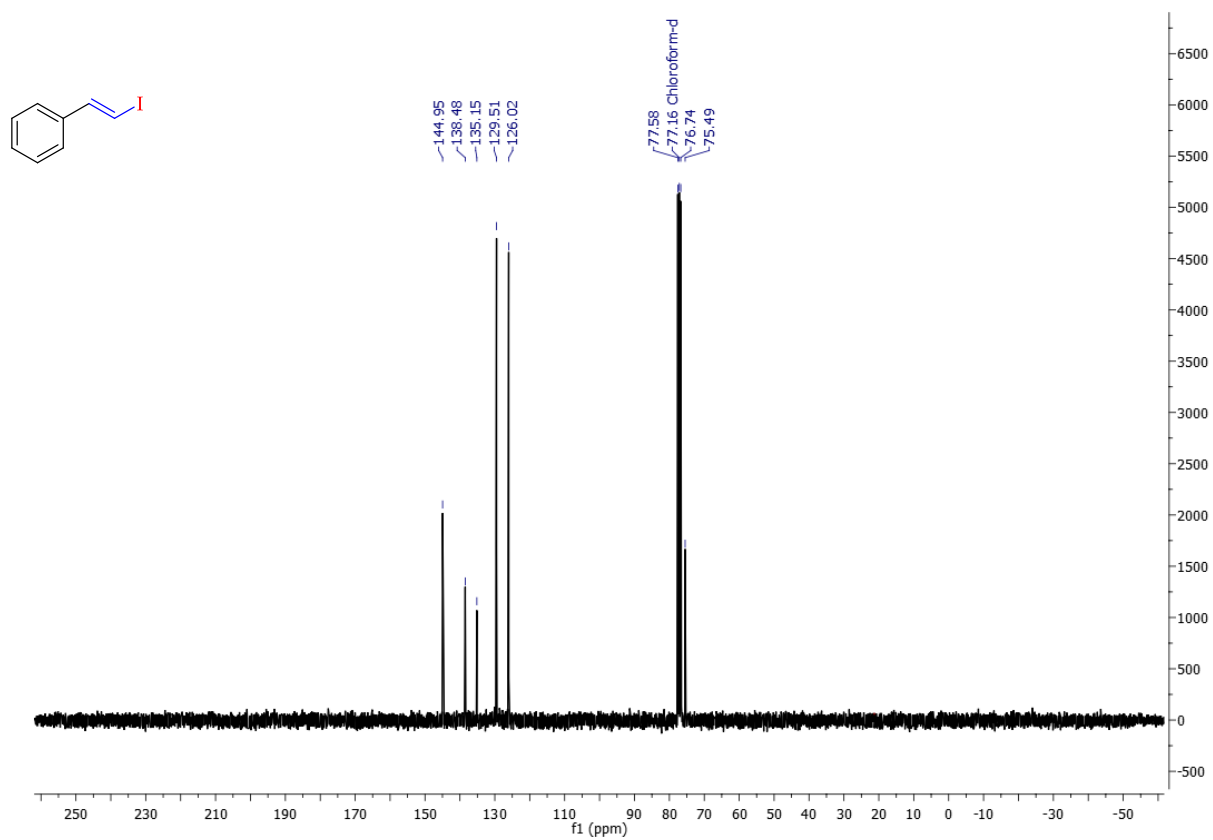


Figure S24. ¹³C NMR spectrum of 5