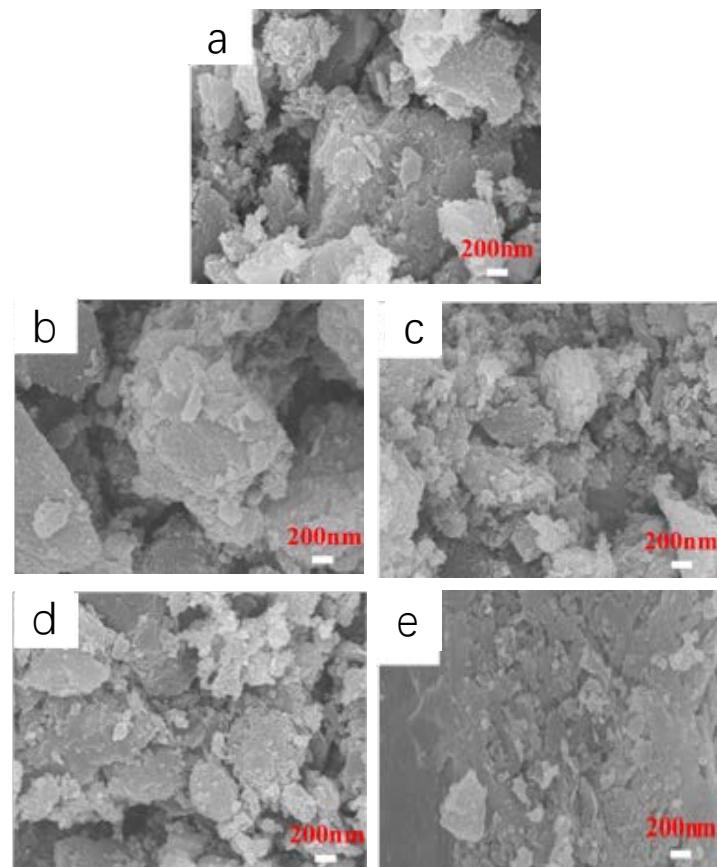
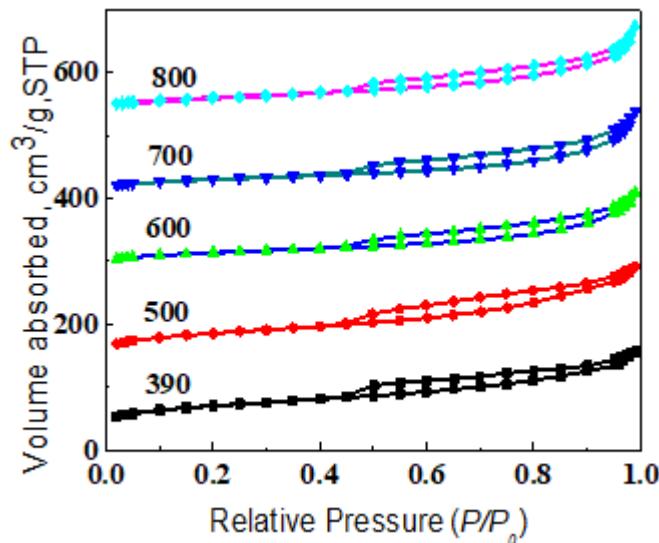


# Supplementary Materials: Selectivity Dependence of 1,1-Difluoro-1-Chloroethane Dehydrohalogenation on the Metal-Support Interaction over SrF<sub>2</sub> Catalyst

Wucan Liu, Yongnan Liu, Kaboza M. Mardochee, Zhikun Wang, Shucheng Wang, Wei Yu, Jianjun Zhang, Wenfeng Han



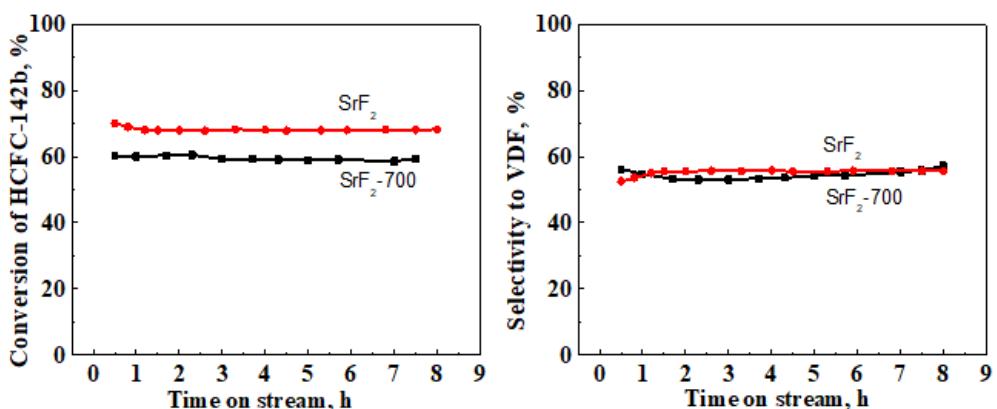
**Figure S1.** SEM images of SrF<sub>2</sub>@C catalysts calcined at different temperatures. (a) SrF<sub>2</sub>@C (390), (b) SrF<sub>2</sub>@C-500, (c) SrF<sub>2</sub>@C-600, (a) SrF<sub>2</sub>@C-700, (e) SrF<sub>2</sub>@C-800.



**Figure S2.** N<sub>2</sub> adsorption-desorption isotherms of SrF<sub>2</sub>@C.

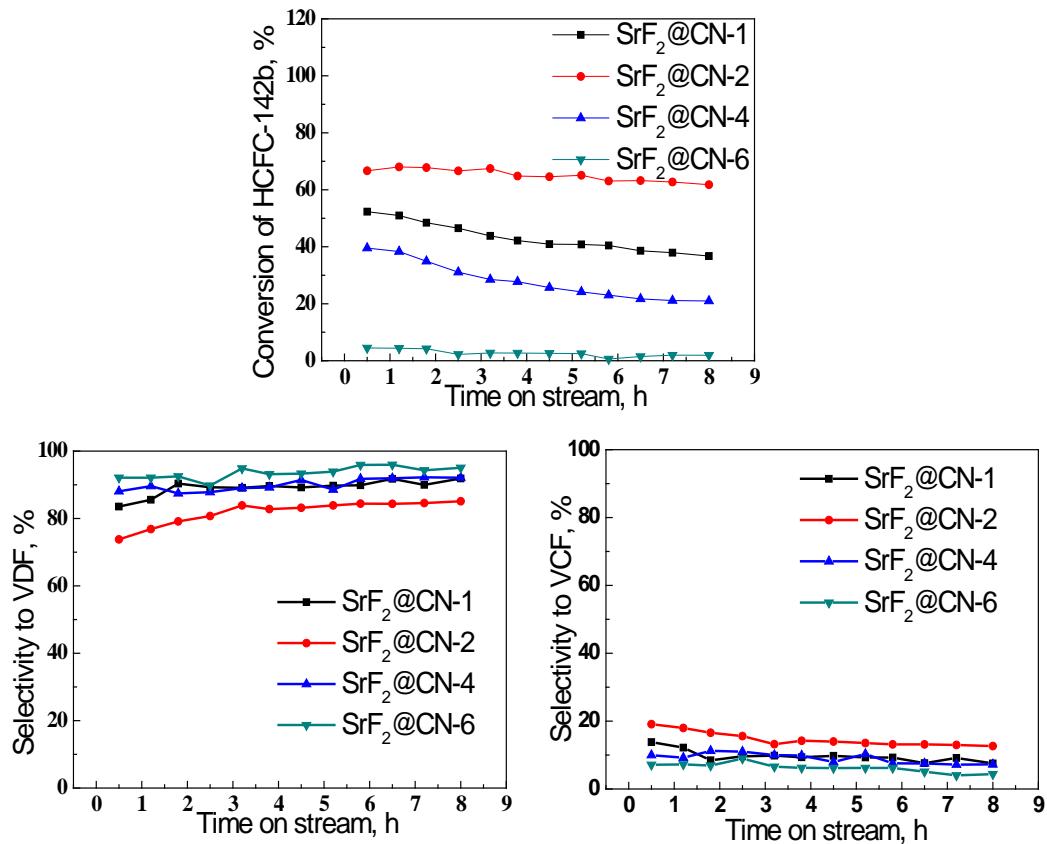
**Table S1.** Surface area and porous parameters of SrF<sub>2</sub>@C samples.

Samples	Specific surface area (m <sup>2</sup> /g)	Total pore volume (cm <sup>3</sup> /g)	Micropore volume (cm <sup>3</sup> /g)	Mesopore volume (cm <sup>3</sup> /g)
SrF <sub>2</sub> @C-390	257	0.24	0.06	0.18
SrF <sub>2</sub> @C-500	250	0.27	0.07	0.20
SrF <sub>2</sub> @C-600	161	0.21	0.04	0.17
SrF <sub>2</sub> @C-700	156	0.23	0.06	0.17
SrF <sub>2</sub> @C-800	170	0.25	0.04	0.21

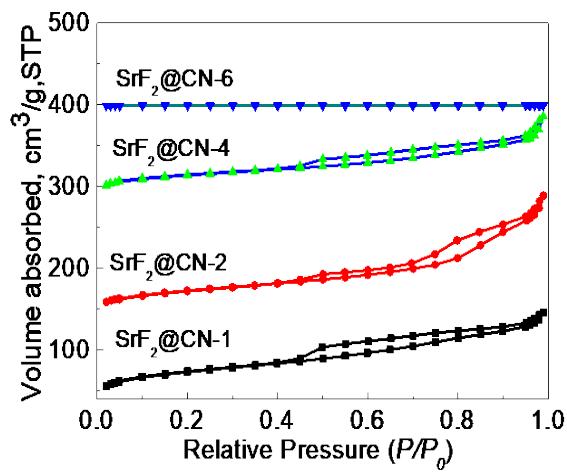


**Figure S3.** Catalytic performance of SrF<sub>2</sub> prepared by precipitation and calcined at 700 °C for the decomposition of HCFC-142b. Reactions were

carried out at 350 °C, 1 bar, GHSV (HCFC-142b) of 600 h<sup>-1</sup> with N<sub>2</sub>:HCFC-142b of 1:1.



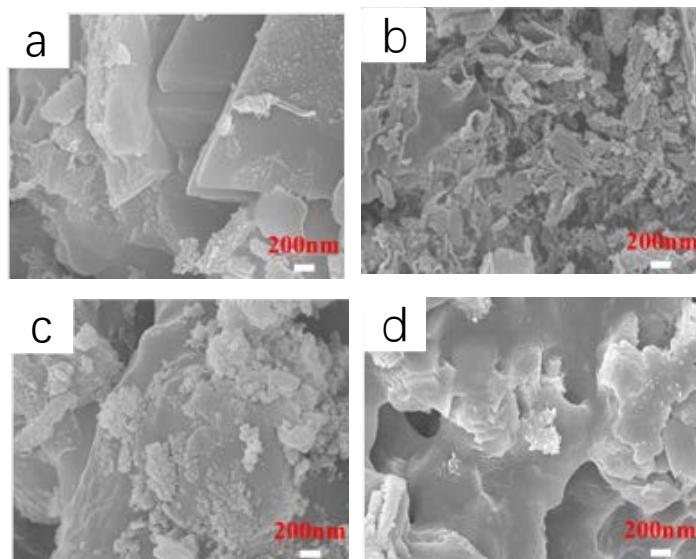
**Figure S4.** Catalytic performance of SrF<sub>2</sub>@CN with different doping amount of urea for the decomposition of HCFC-142b. Reactions were carried out at 350 °C, 1 bar, GHSV (HCFC-142b) of 600 h<sup>-1</sup> with N<sub>2</sub>:HCFC-142b of 1:1. SrF<sub>2</sub>@CN-1 with mass ratio of SrC<sub>2</sub>O<sub>4</sub>:PVDF:Urea=1:1:1, SrF<sub>2</sub>@CN-2 of 1:1:2, SrF<sub>2</sub>@CN-4 of 1:1:4, and SrF<sub>2</sub>@CN-6 of 1:1:6.



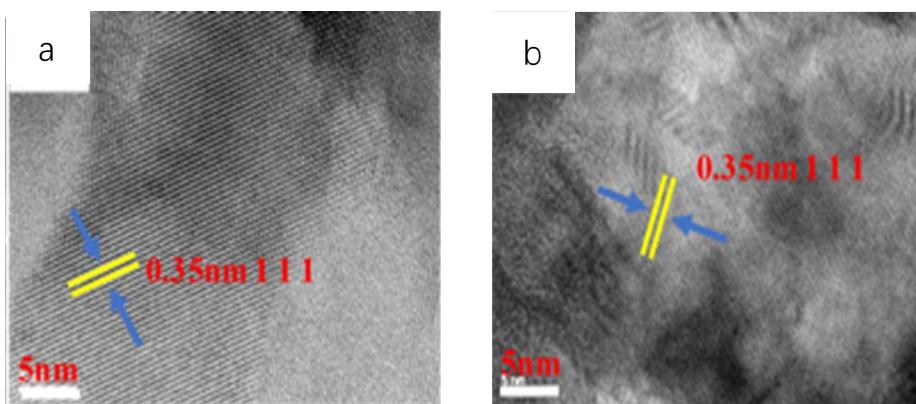
**Figure S5.** N<sub>2</sub> adsorption-desorption isotherms of SrF<sub>2</sub>@NC with different doping amount of urea during catalyst preparation. SrF<sub>2</sub>@CN-1 with mass ratio of SrC<sub>2</sub>O<sub>4</sub>:PVDF:Urea=1:1:1, SrF<sub>2</sub>@CN-2 of 1:1:2, SrF<sub>2</sub>@CN-4 of 1:1:4, and SrF<sub>2</sub>@CN-6 of 1:1:6.

**Table S2.** Surface area and porous system parameters of SrF<sub>2</sub>@NC catalyst prepared with different doping amount of urea.

Samples	Specific surface area (m <sup>2</sup> /g)	Total pore volume (cm <sup>3</sup> /g)	Micropore volume (cm <sup>3</sup> /g)	Mesopore volume (cm <sup>3</sup> /g)
SrF <sub>2</sub> @CN-1	267	0.23	0.08	0.15
SrF <sub>2</sub> @CN-2	325	0.31	0.1	0.21
SrF <sub>2</sub> @CN-4	190	0.19	0.06	0.13
SrF <sub>2</sub> @CN-6	1.2	0.02	--	0.02



**Figure S6.** SEM images of SrF<sub>2</sub>@NC catalysts prepared by doping different amounts of urea during catalyst preparation. SrC<sub>2</sub>O<sub>4</sub>:PVDF:Urea of (a) 1:1:1; (b) 1:1:2; (b) 1:1:4 and (d) 1:1:6 .



**Figure S7.** High resolution TEM images of SrF<sub>2</sub>@C and SrF<sub>2</sub>@NC-700 catalysts.

**Table S3.** Surface area and porous parameters of SrF<sub>2</sub>@C, SrF<sub>2</sub>@NC, SrF<sub>2</sub>@C-700 and SrF<sub>2</sub>@NC-700.

Samples	Specific surface area (m <sup>2</sup> /g)	Total pore volume (cm <sup>3</sup> /g)	Micropore volume (cm <sup>3</sup> /g)	Mesopore volume (cm <sup>3</sup> /g)
SrF <sub>2</sub> @C	257	0.24	0.06	0.18
SrF <sub>2</sub> @NC	325	0.31	0.1	0.21
SrF <sub>2</sub> @C-700	156	0.23	0.06	0.17
SrF <sub>2</sub> @NC-700	97	0.16	0.03	0.13

**Table S4.** Surface element content of SrF<sub>2</sub>@C, SrF<sub>2</sub>@NC, SrF<sub>2</sub>@C-700 and SrF<sub>2</sub>@NC-700 catalysts determined by XPS.

Catalysts	C	N	O	F	Sr
SrF <sub>2</sub> @C-390	63	0	9	22	5
SrF <sub>2</sub> @CN-390	63	6	4	20	7
SrF <sub>2</sub> @C-700	67	2	5	18	7
SrF <sub>2</sub> @NC-700	69	7	6	12	5