

# SUPPLEMENTARY INFORMATION

## Radical C–H <sup>18</sup>F-Difluoromethylation of Heteroarenes with [<sup>18</sup>F]Difluoromethyl Heteroaryl-Sulfones by Visible Light Photoredox Catalysis

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1. NMR spectra of the compounds 4a-4f, 5a-5f, and 6a-6f

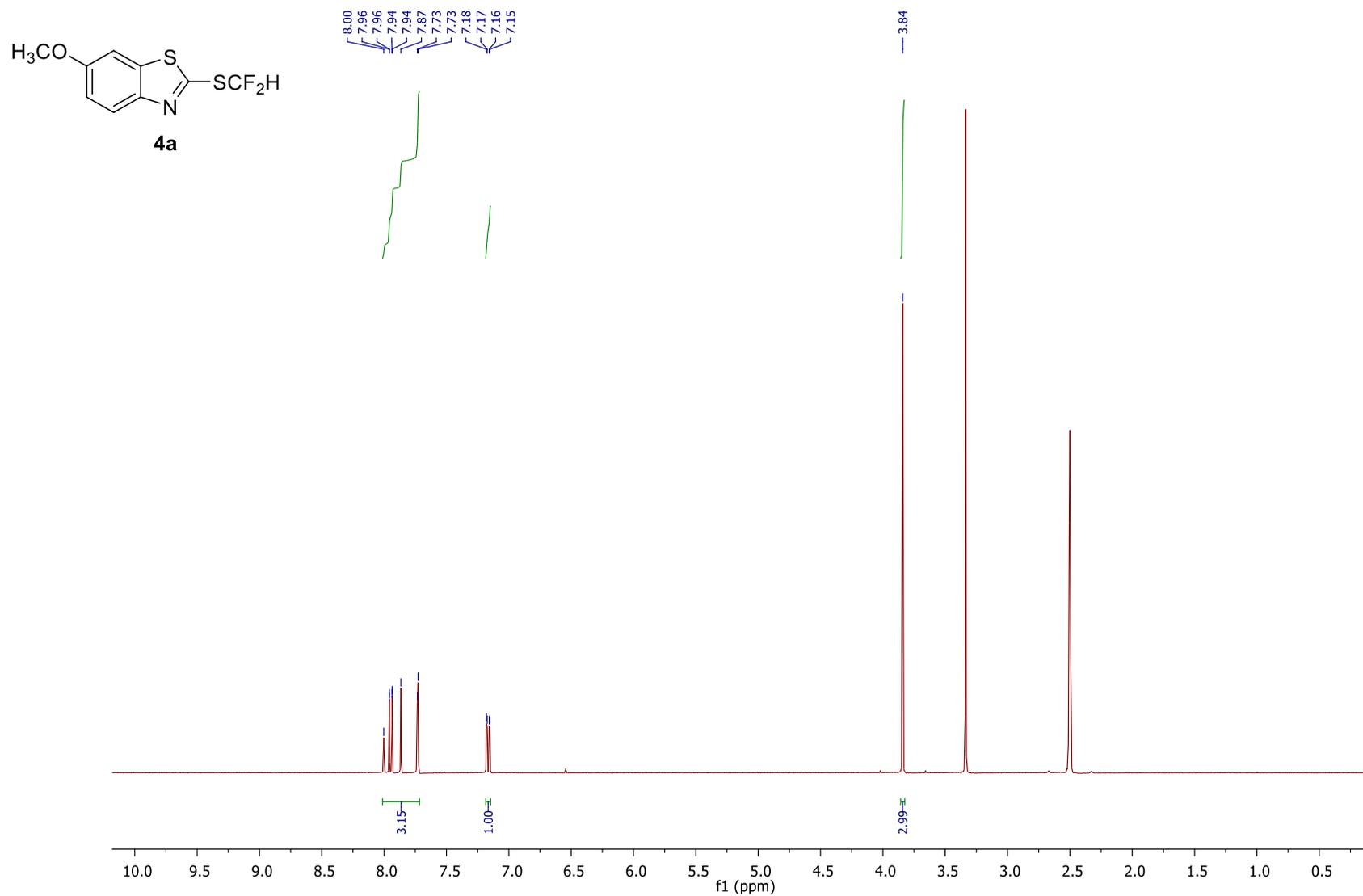
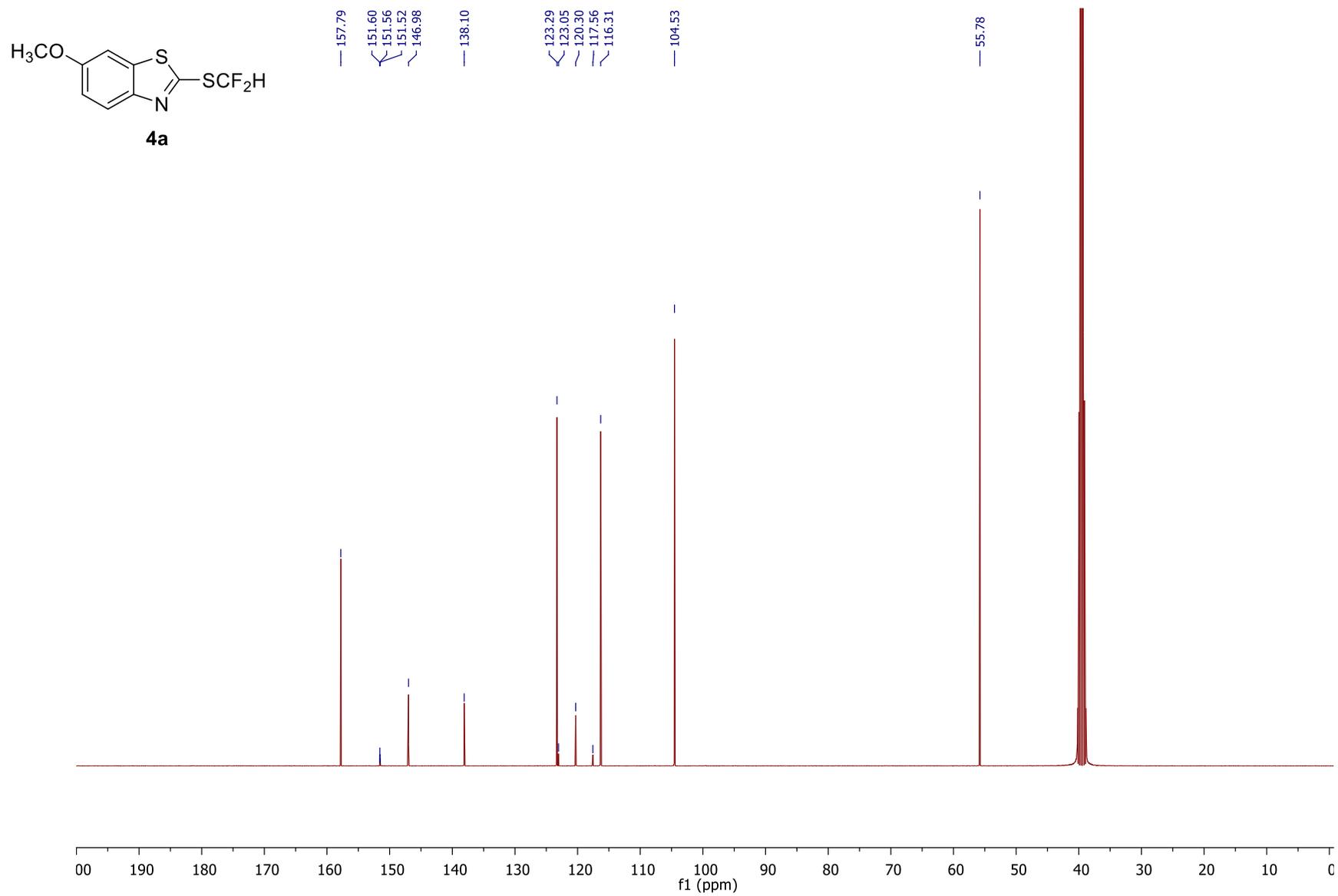
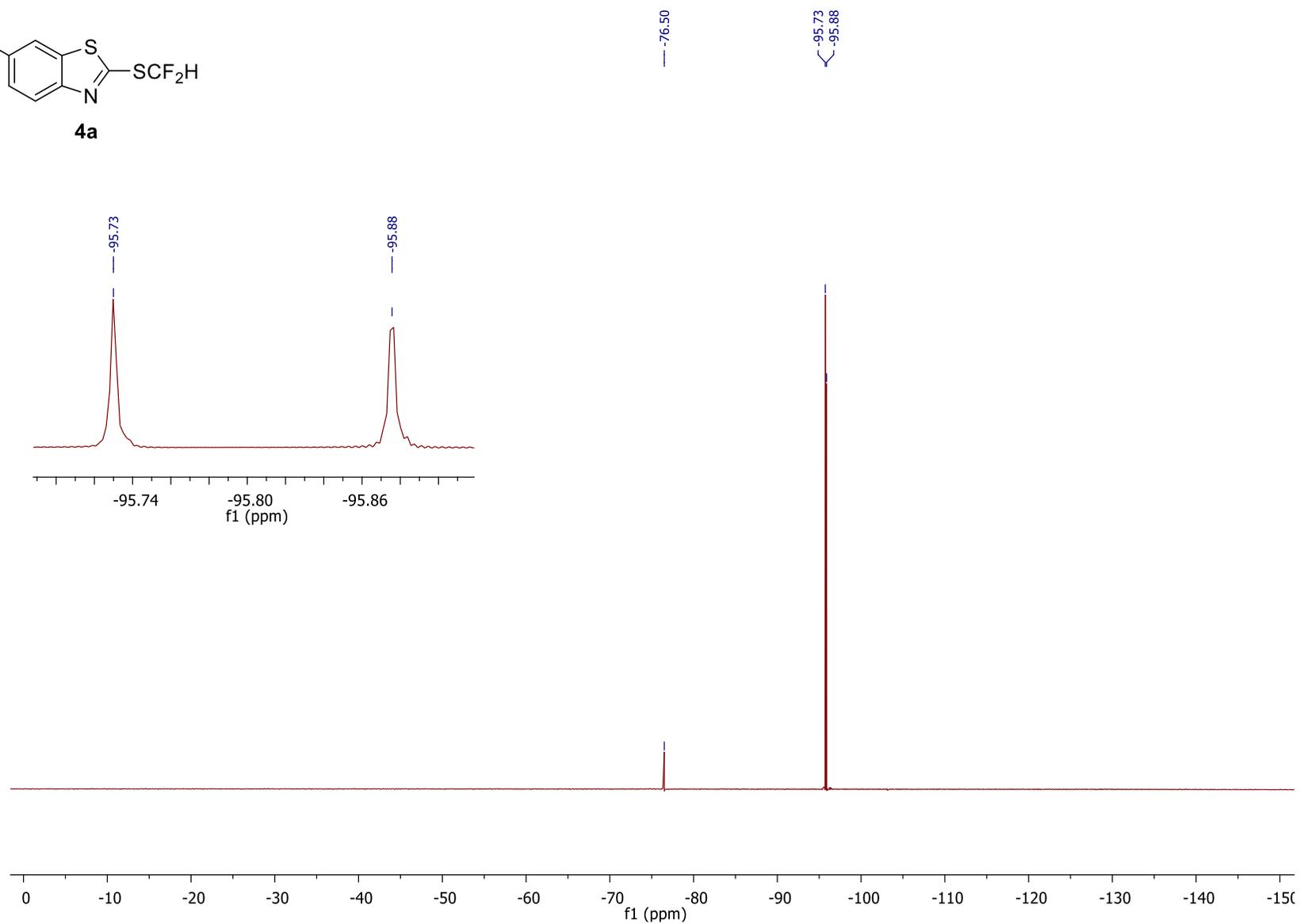
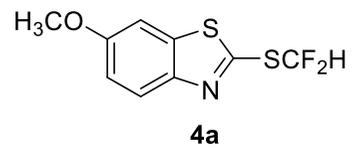


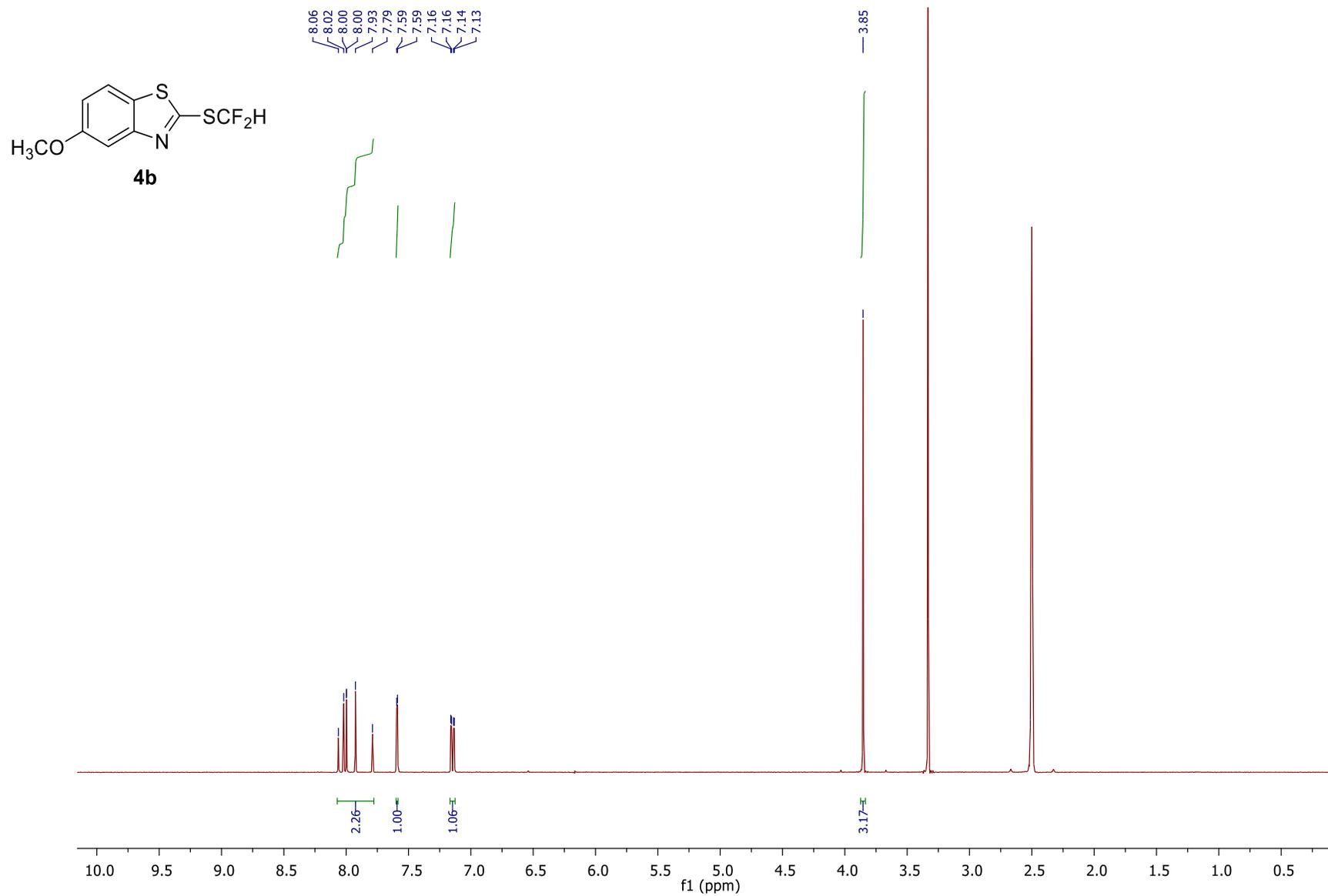
Figure S1. <sup>1</sup>H-NMR spectrum of 2-((difluoromethyl)thio)-6-methoxybenzo[d]thiazole (4a).



**Figure S2.**  $^{13}\text{C}$ -NMR spectrum of 2-((difluoromethyl)thio)-6-methoxybenzo[*d*]thiazole (**4a**).



**Figure S3.**  $^{19}\text{F}$ -NMR spectrum of 2-((difluoromethyl)thio)-6-methoxybenzo[*d*]thiazole (**4a**).



**Figure S4.** <sup>1</sup>H-NMR spectrum of 2-((difluoromethyl)thio)-5-methoxybenzo[d]thiazole (**4b**).

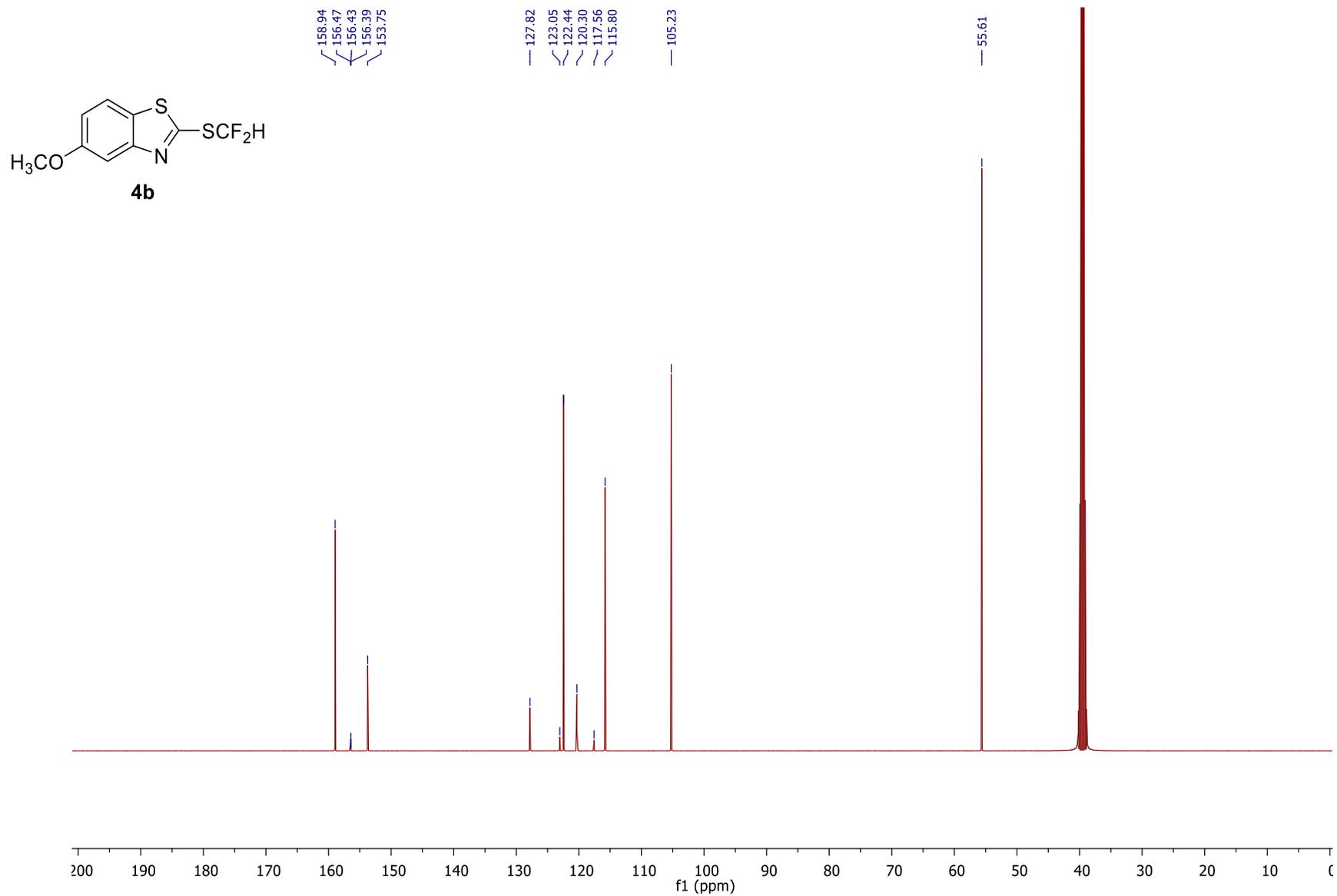
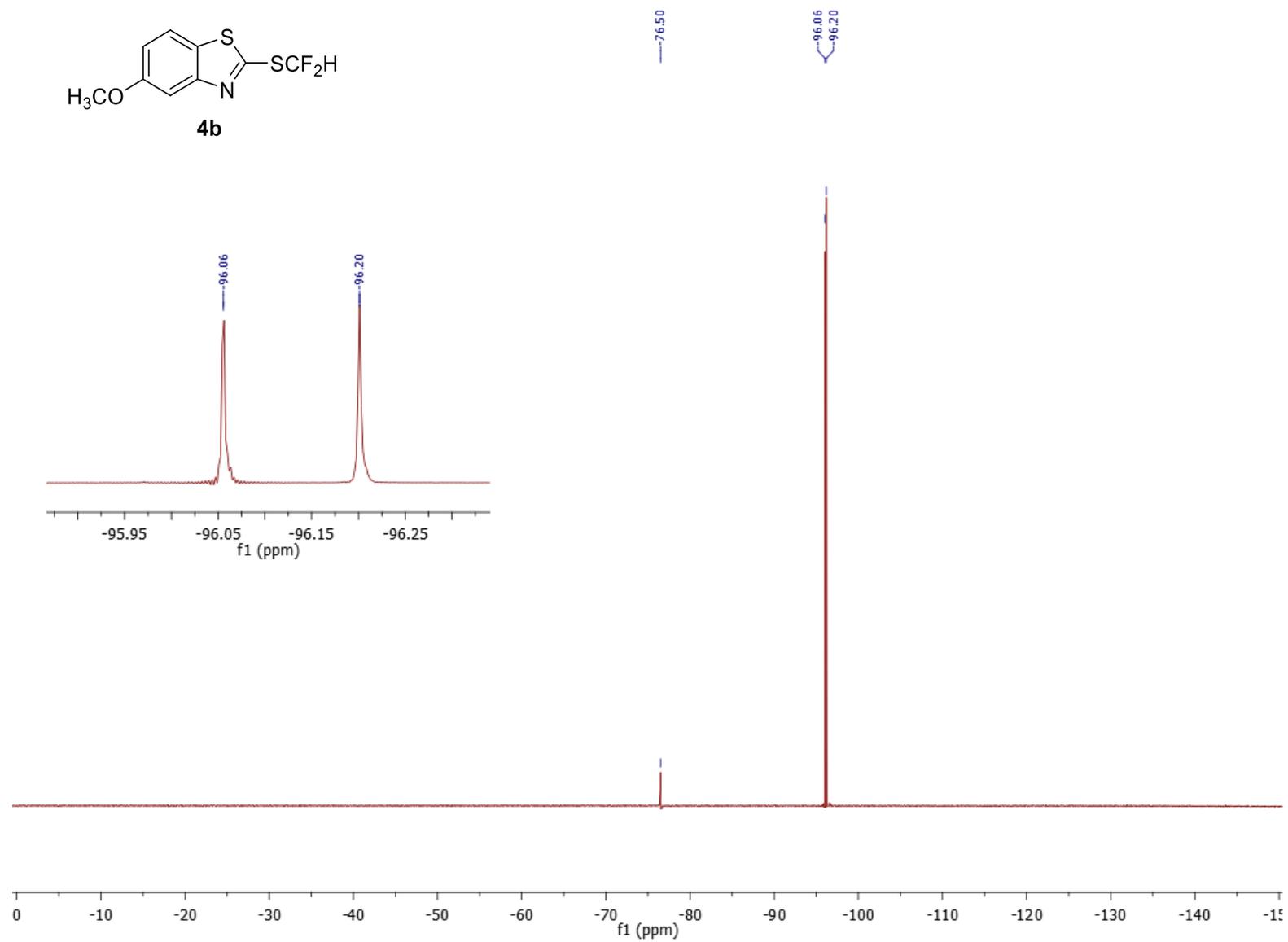
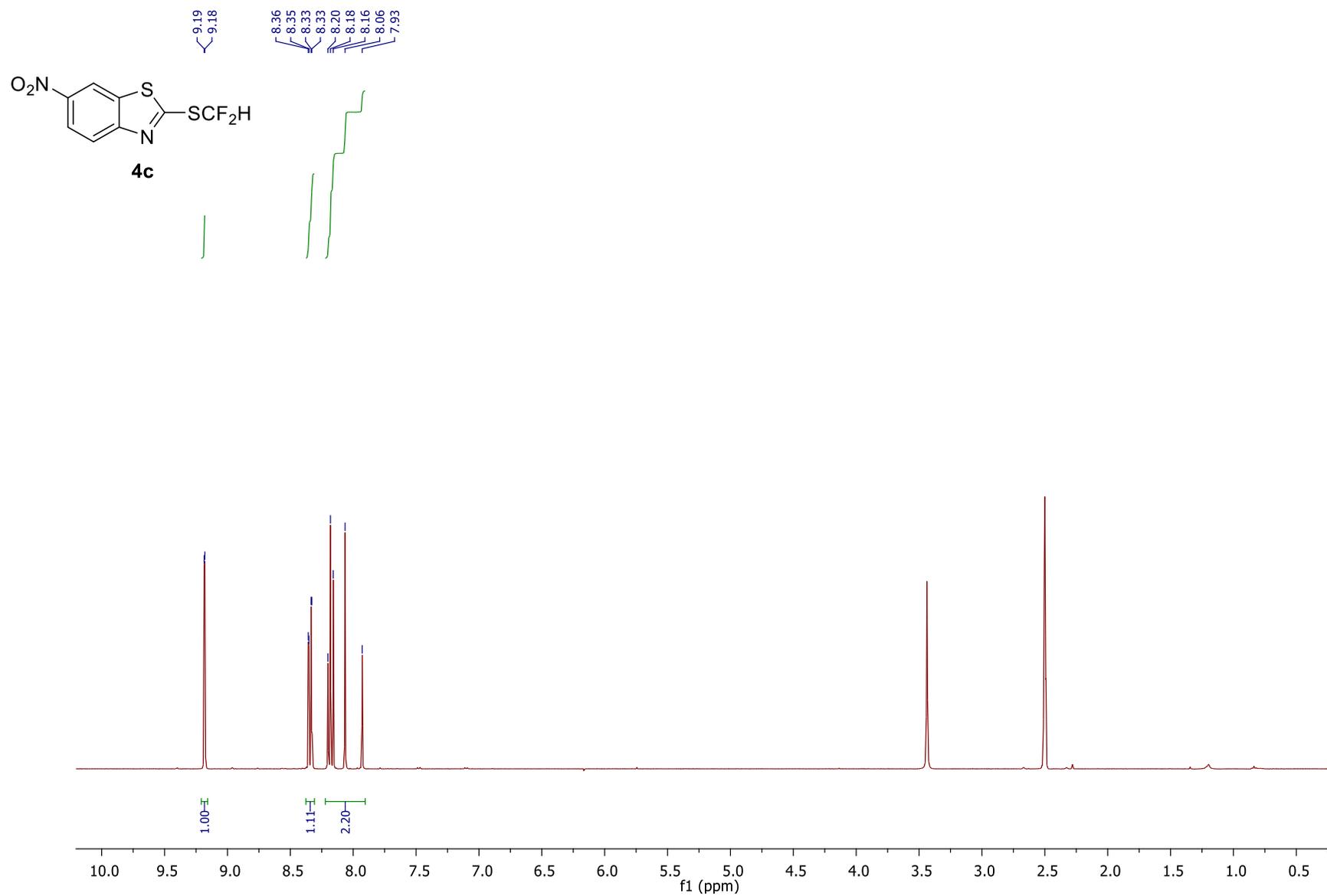


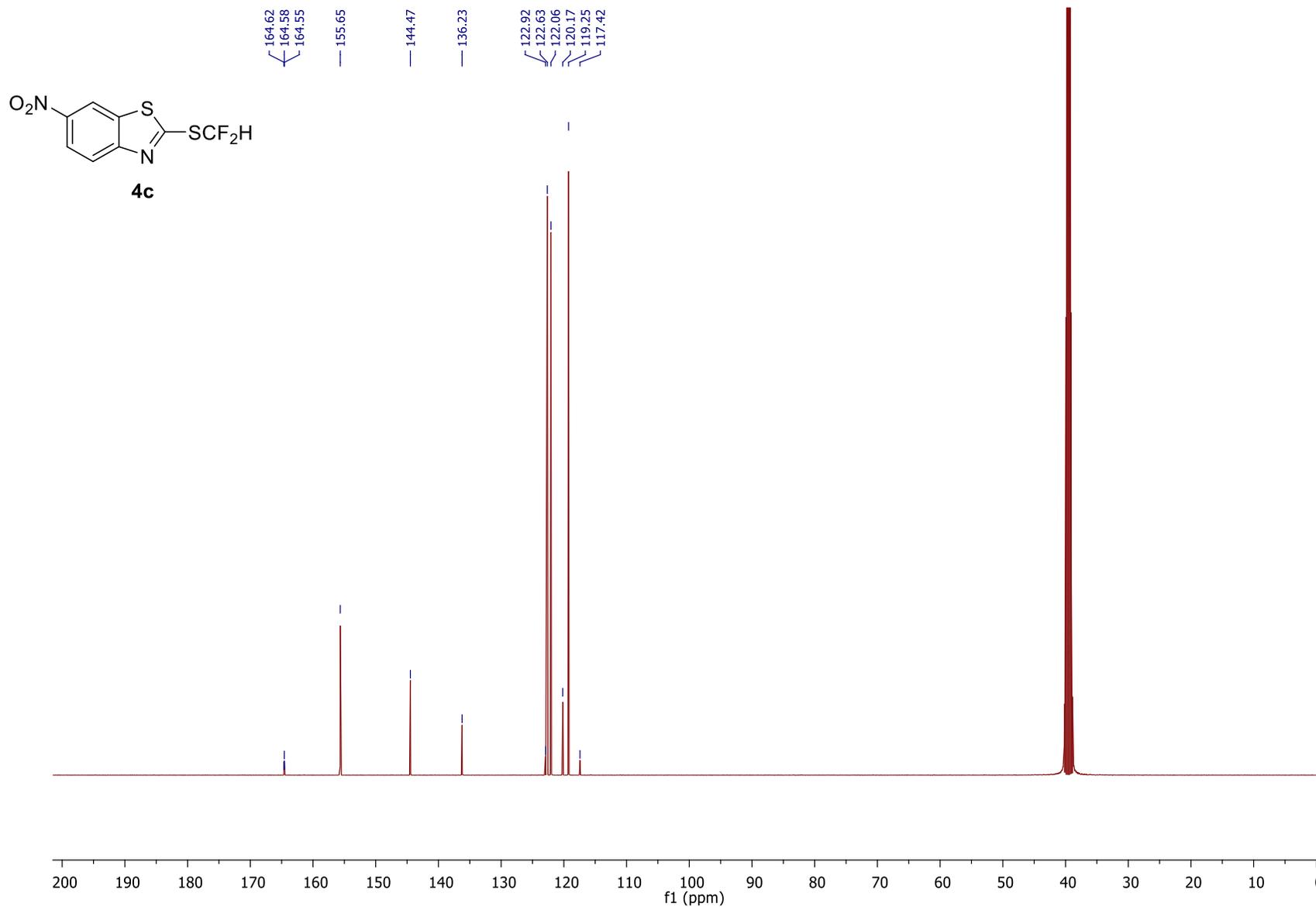
Figure S5. <sup>13</sup>C-NMR spectrum of 2-((difluoromethyl)thio)-5-methoxybenzo[d]thiazole (**4b**).



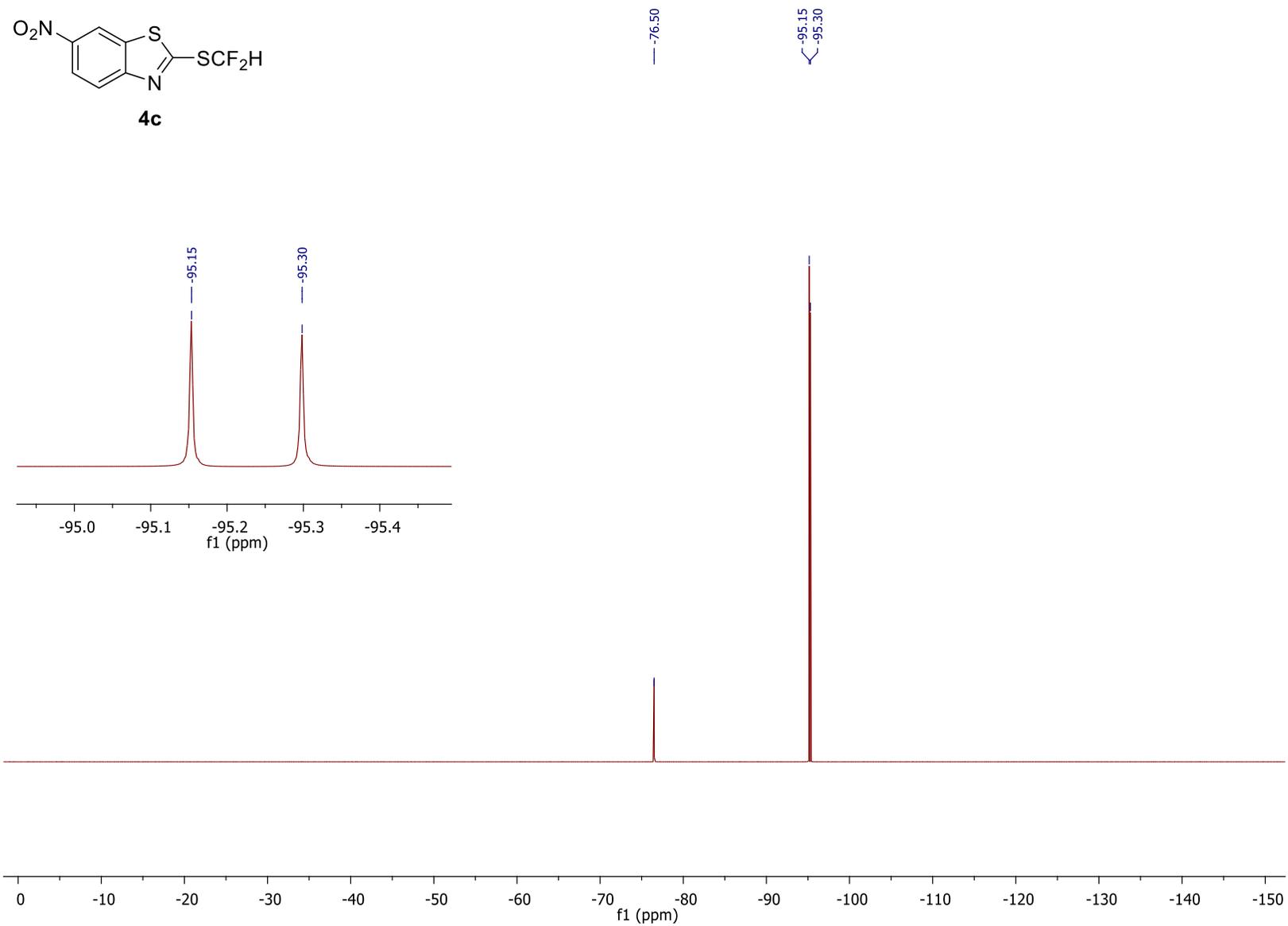
**Figure S6.**  $^{19}\text{F}$ -NMR spectrum of 2-((difluoromethyl)thio)-5-methoxybenzo[*d*]thiazole (**4b**).



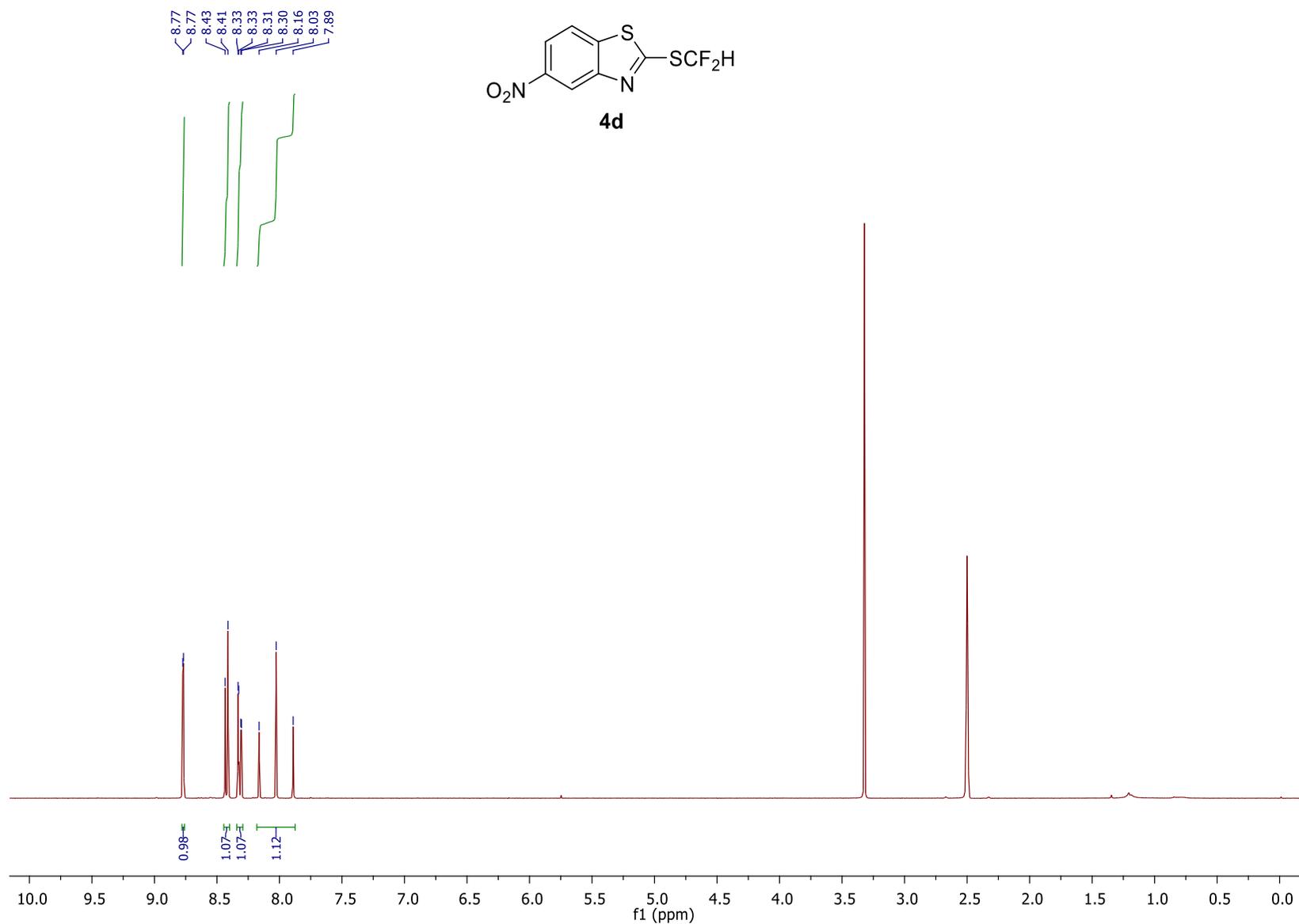
**Figure S7.** <sup>1</sup>H-NMR spectrum of 2-((difluoromethyl)thio)-6-nitrobenzo[*d*]thiazole (**4c**).



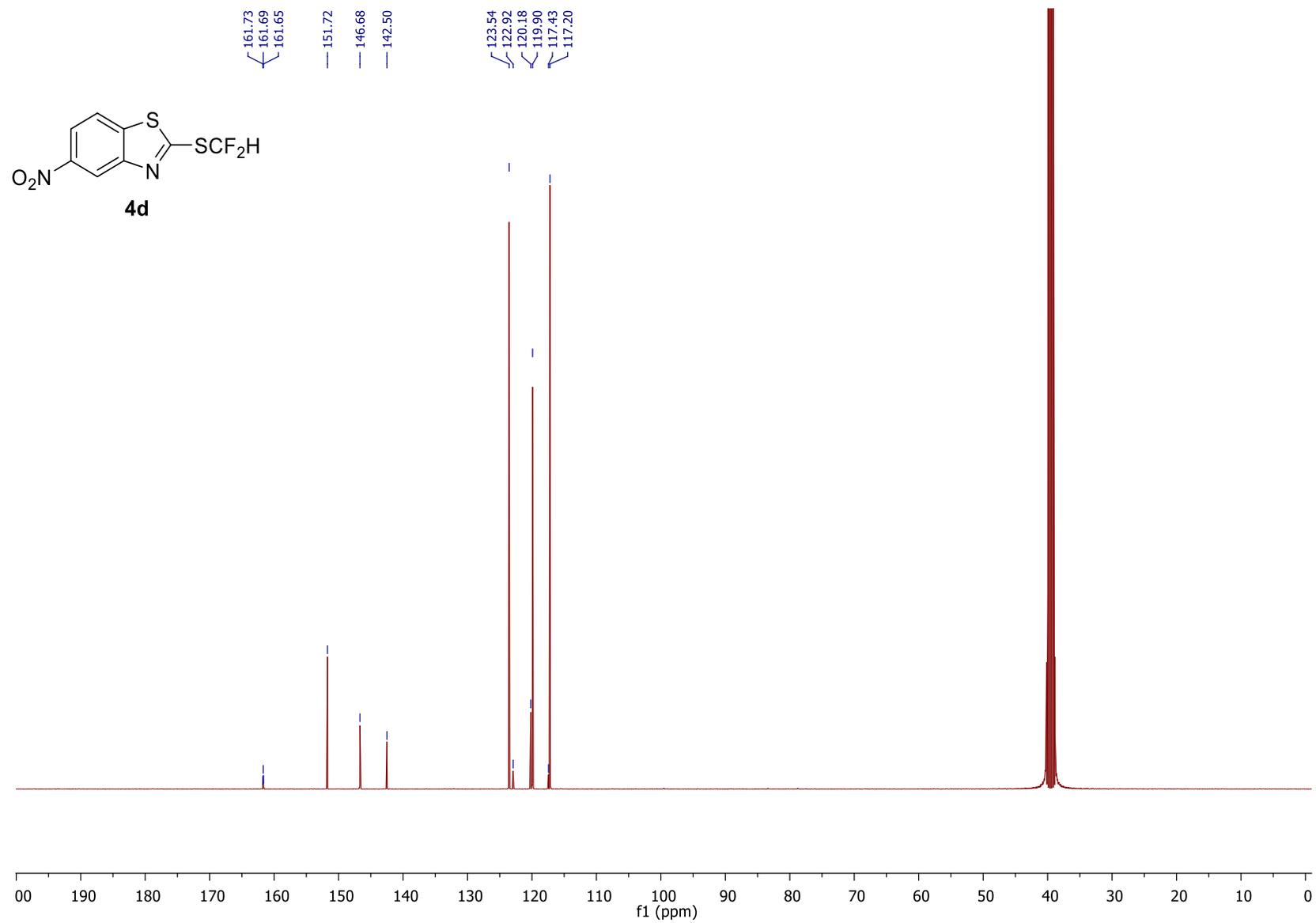
**Figure S8.** <sup>13</sup>C-NMR spectrum of 2-((difluoromethyl)thio)-6-nitrobenzo[d]thiazole (**4c**).



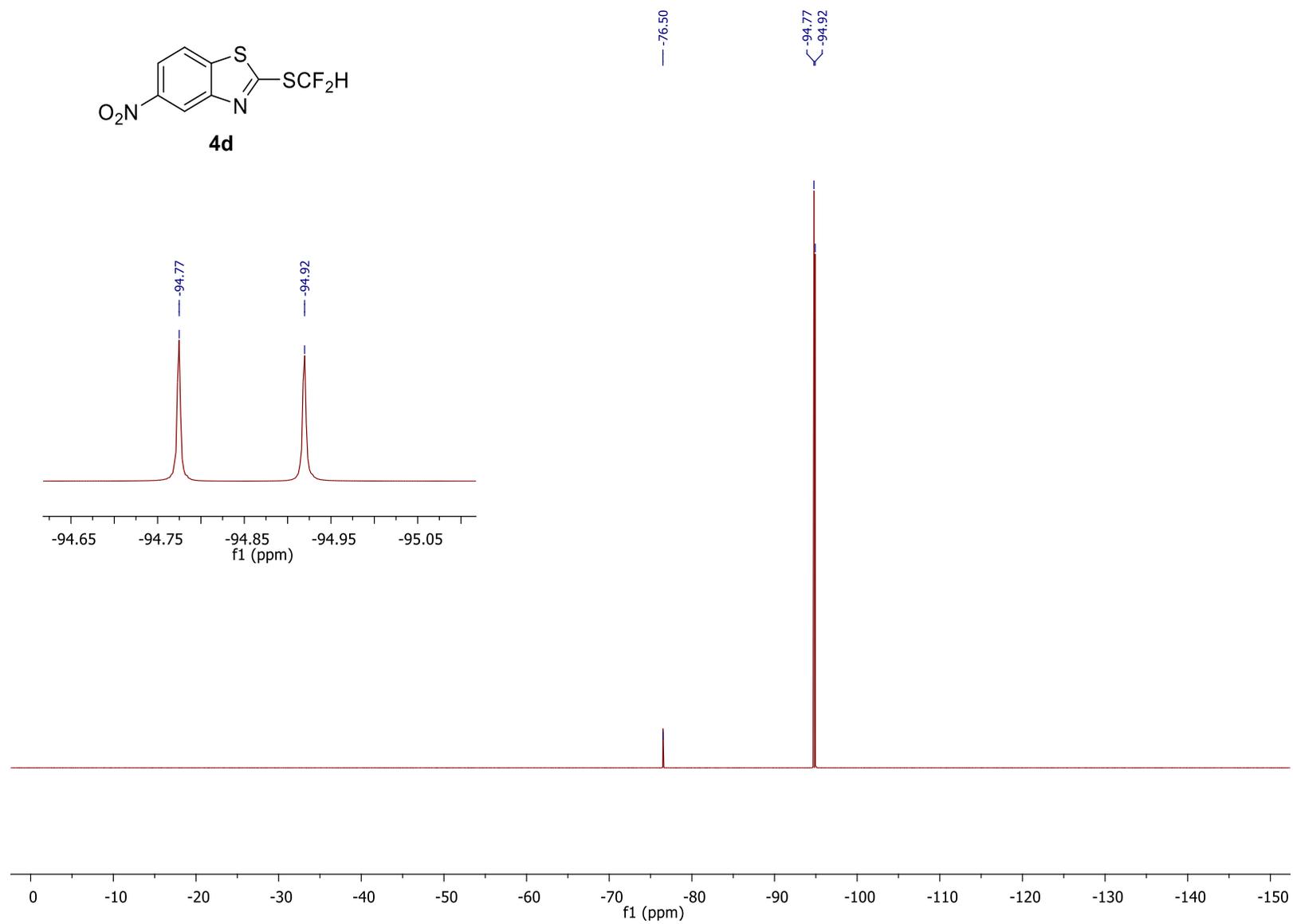
**Figure S9.**  $^{19}\text{F}$ -NMR spectrum of 2-((difluoromethyl)thio)-6-nitrobenzo[*d*]thiazole (**4c**).



**Figure S10.** <sup>1</sup>H-NMR spectrum of 2-((difluoromethyl)thio)-5-nitrobenzo[*d*]thiazole (**4d**).



**Figure S11.** <sup>13</sup>C-NMR spectrum of 2-((difluoromethyl)thio)-5-nitrobenzo[*d*]thiazole (**4d**).



**Figure S12.**  $^{19}\text{F}$ -NMR spectrum of 2-((difluoromethyl)thio)-5-nitrobenzo[*d*]thiazole (**4d**).

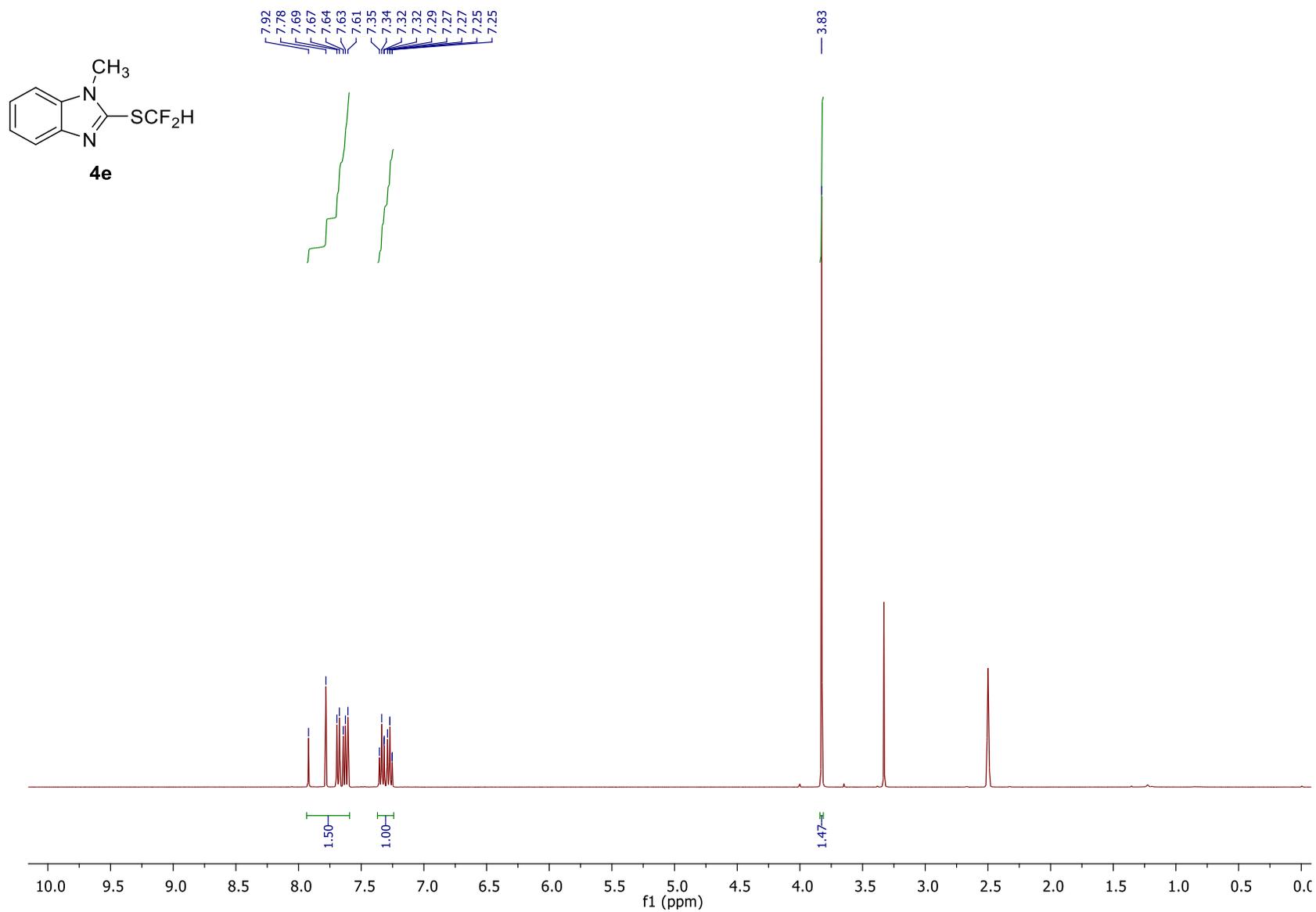


Figure S13. <sup>1</sup>H-NMR spectrum of 2-((difluoromethyl)thio)-1-methyl-1H-benzo[d]imidazole (4e).

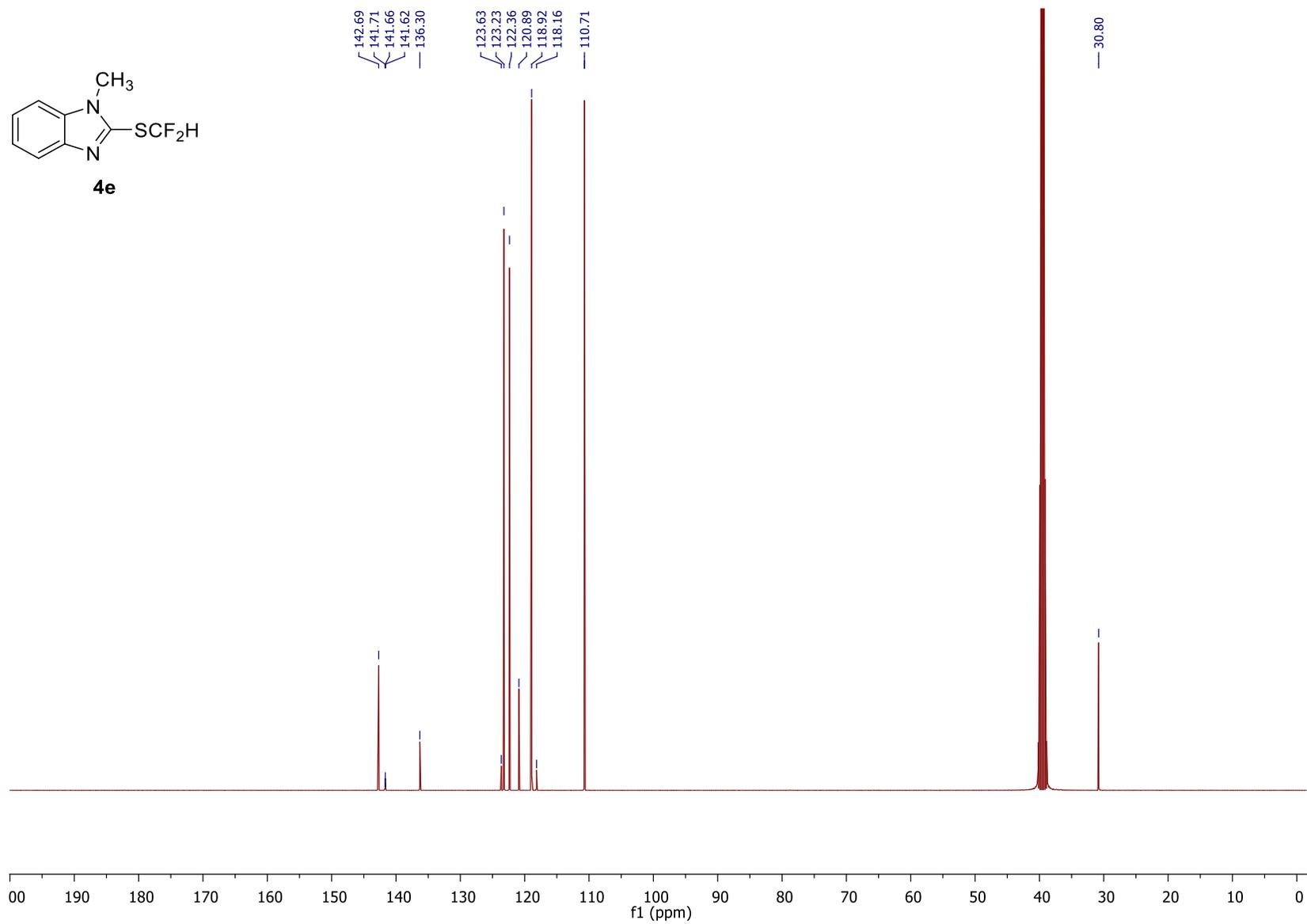
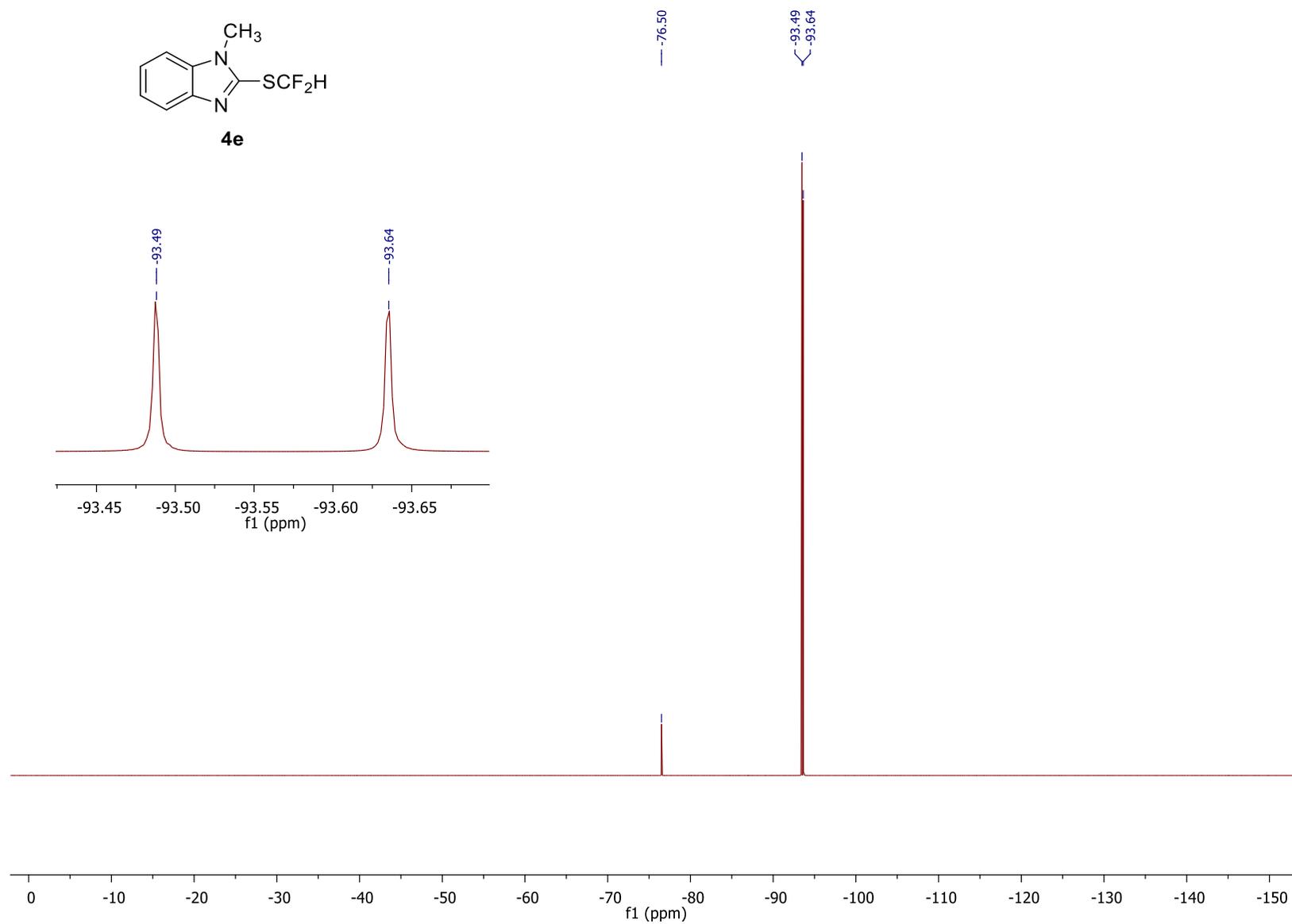
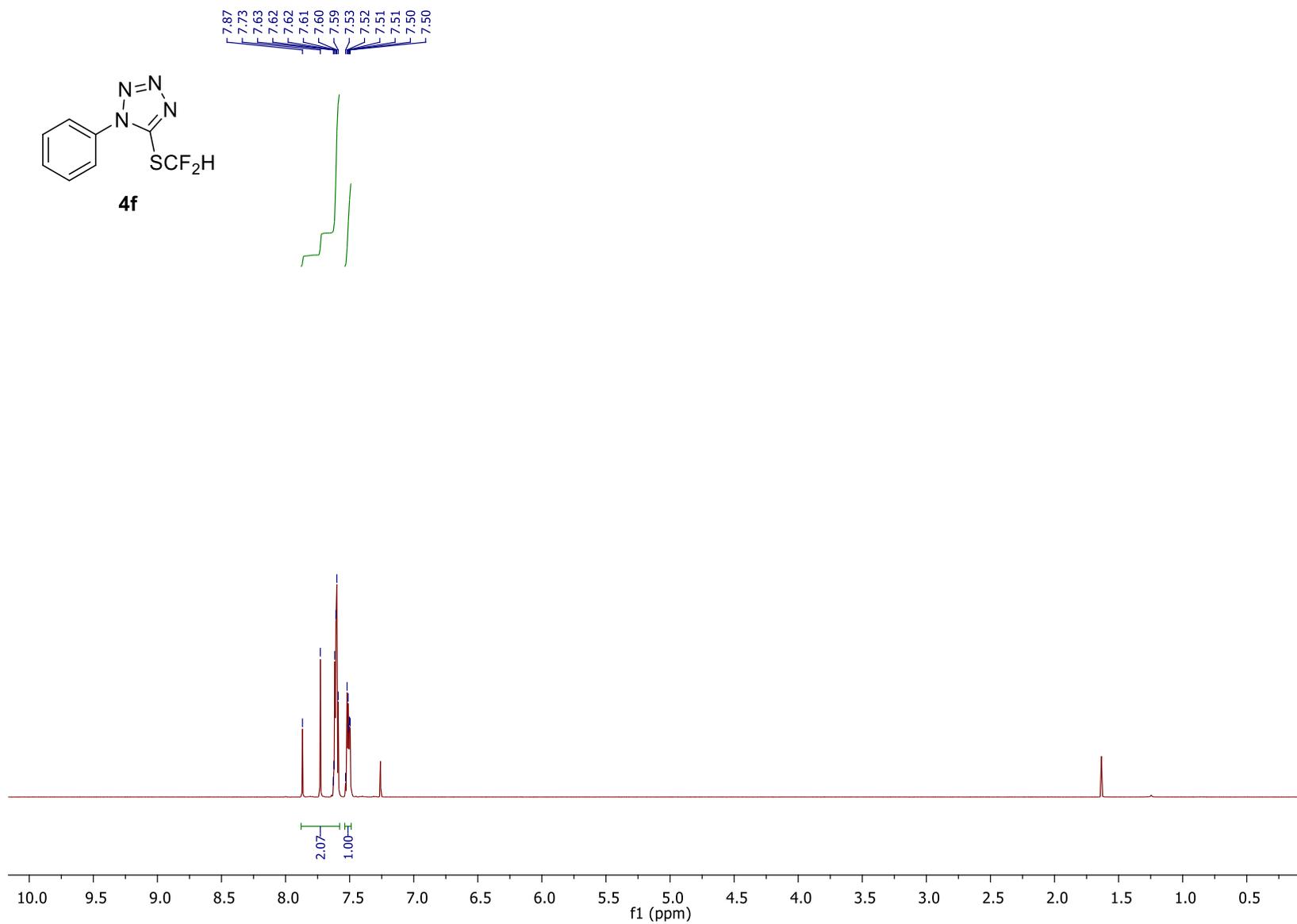


Figure S14. <sup>13</sup>C-NMR spectrum of 2-((difluoromethyl)thio)-1-methyl-1H-benzimidazole (4e).



**Figure S15.** <sup>19</sup>F-NMR spectrum of 2-((difluoromethyl)thio)-1-methyl-1H-benzo[d]imidazole (**4e**).



**Figure S16.** <sup>1</sup>H-NMR spectrum of 5-((difluoromethyl)thio)-1-phenyl-1H-tetrazole (**4f**).

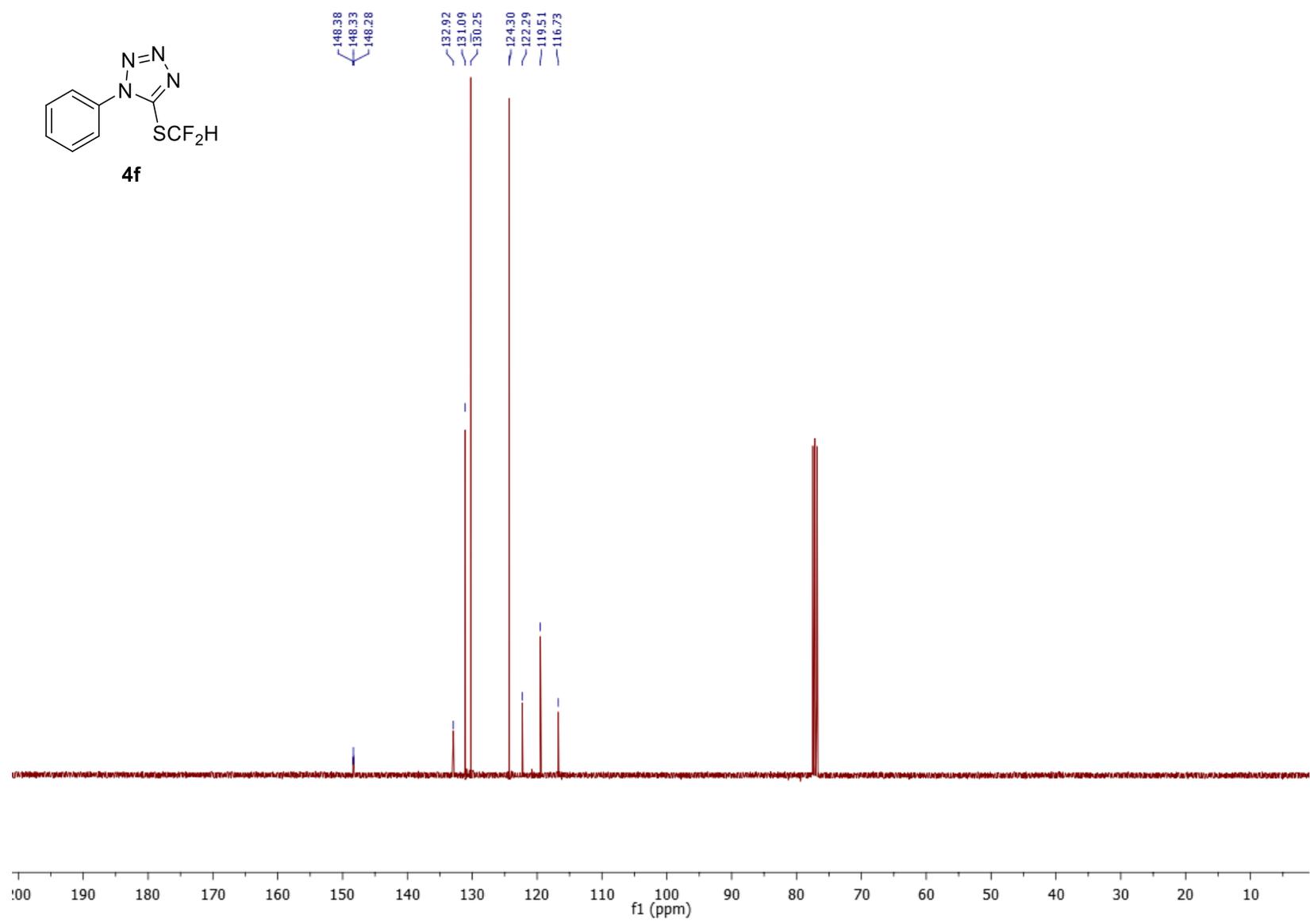
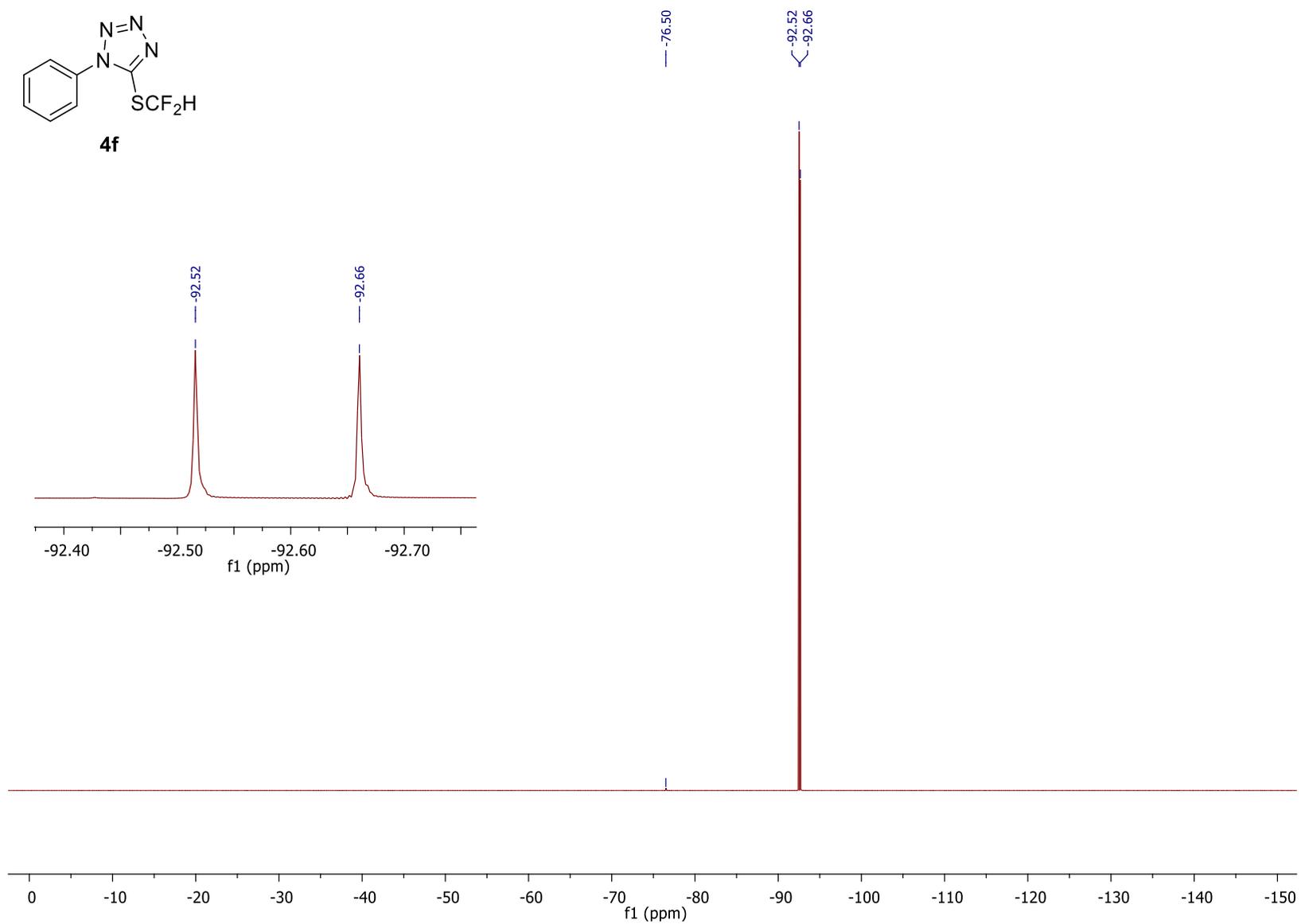
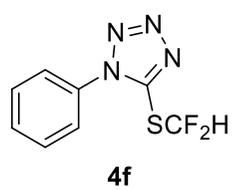


Figure S17. <sup>13</sup>C-NMR spectrum of 5-((difluoromethyl)thio)-1-phenyl-1H-tetrazole (**4f**).



**Figure S18.**  $^{19}\text{F}$ -NMR spectrum of 5-((difluoromethyl)thio)-1-phenyl-1*H*-tetrazole (**4f**).

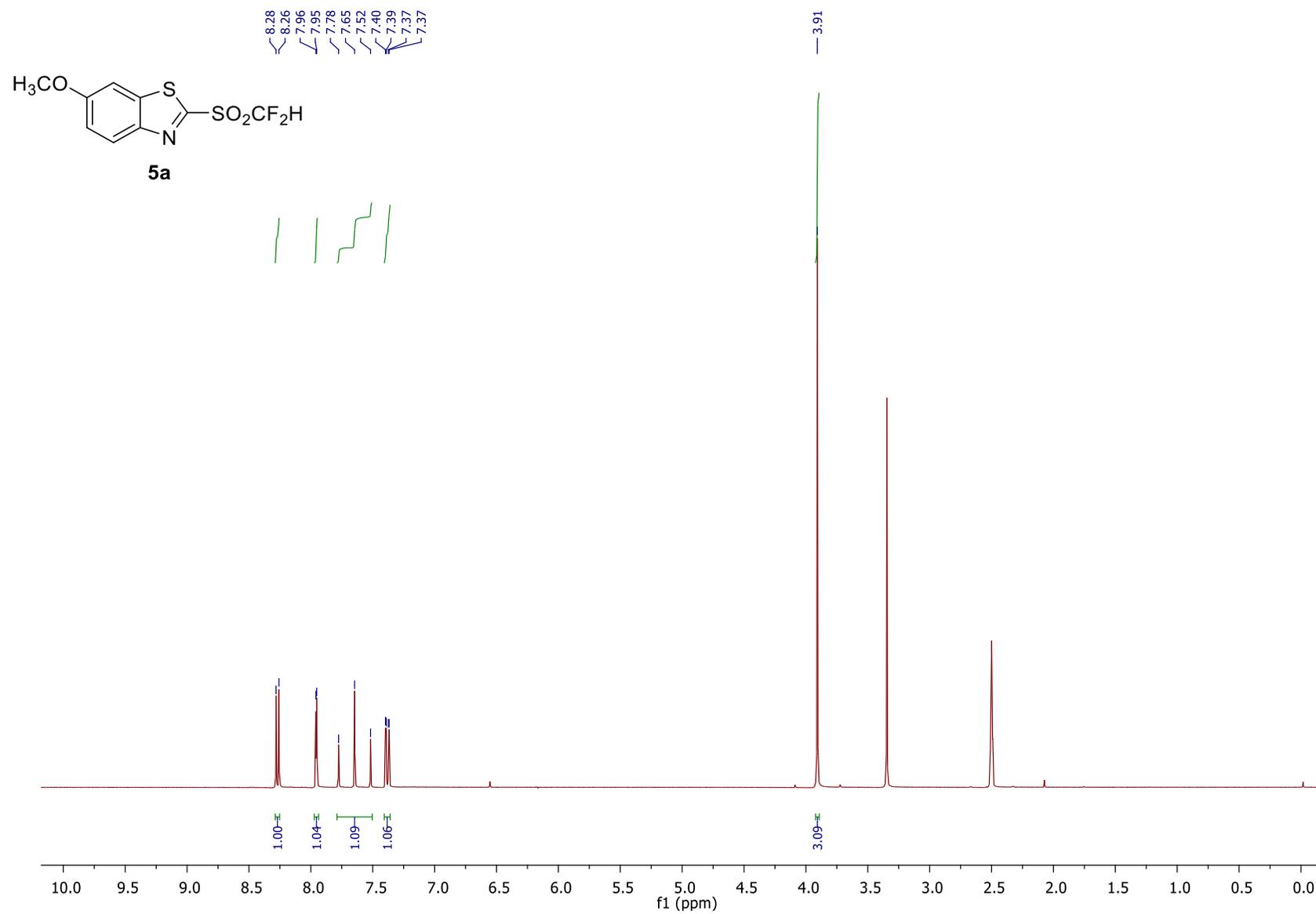
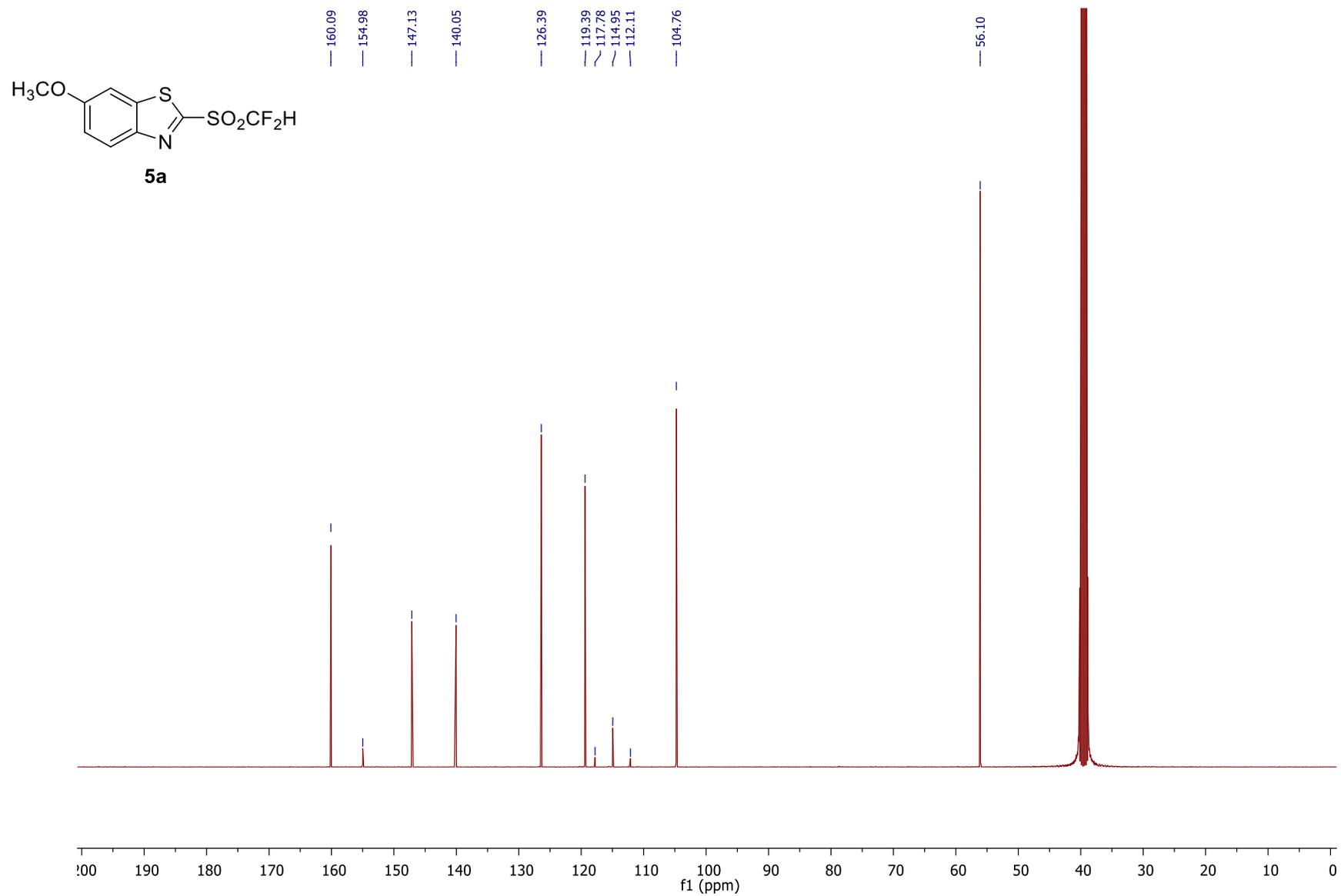
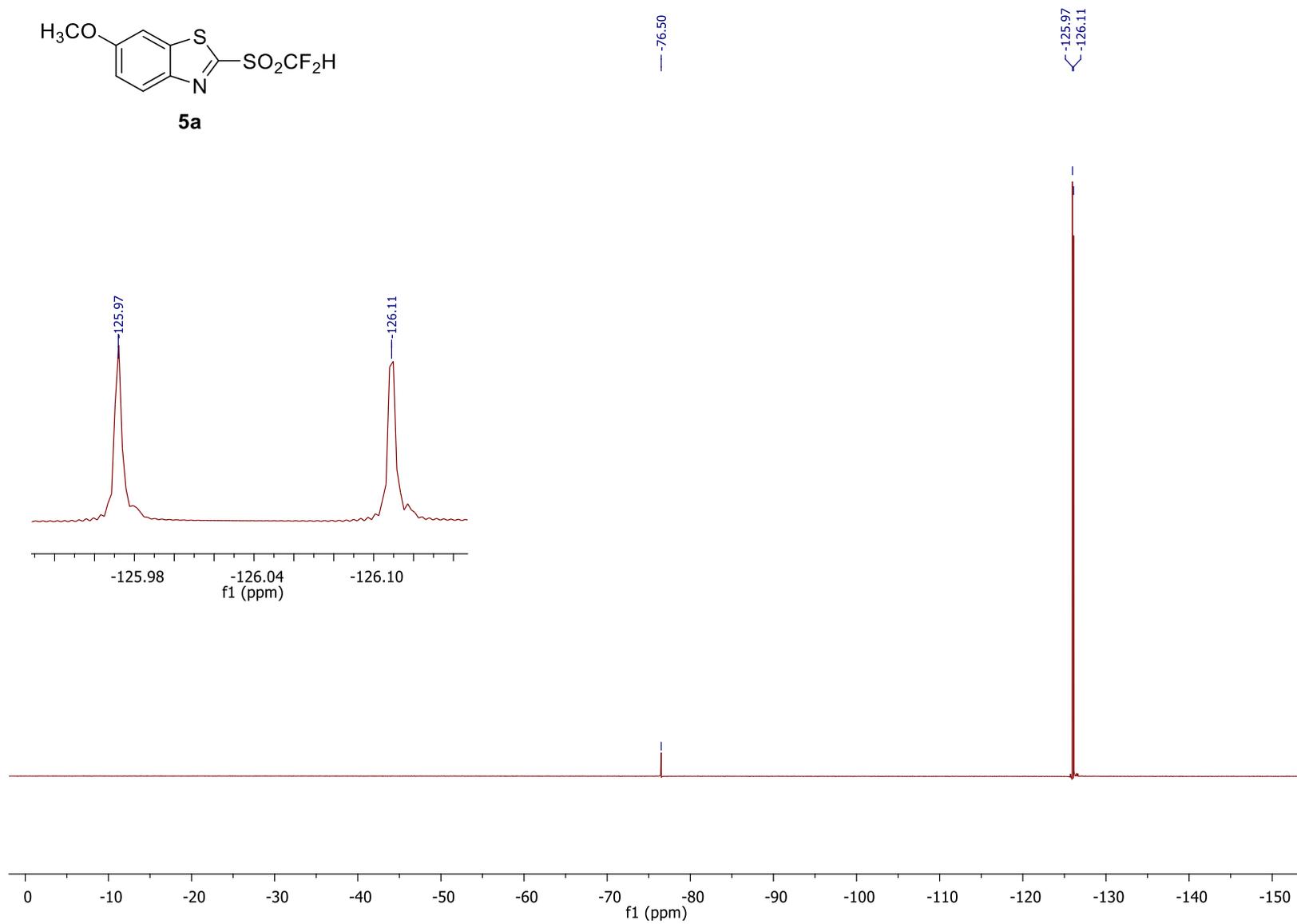


Figure S19. <sup>1</sup>H-NMR spectrum of 2-((difluoromethyl)sulfonyl)-6-methoxybenzo[d]thiazole (5a).



**Figure S20.** <sup>13</sup>C-NMR spectrum of 2-((difluoromethyl)sulfonyl)-6-methoxybenzo[d]thiazole (**5a**).



**Figure S21.**  $^{19}\text{F}$ -NMR spectrum of 2-((difluoromethyl)sulfonyl)-6-methoxybenzo[d]thiazole (**5a**).

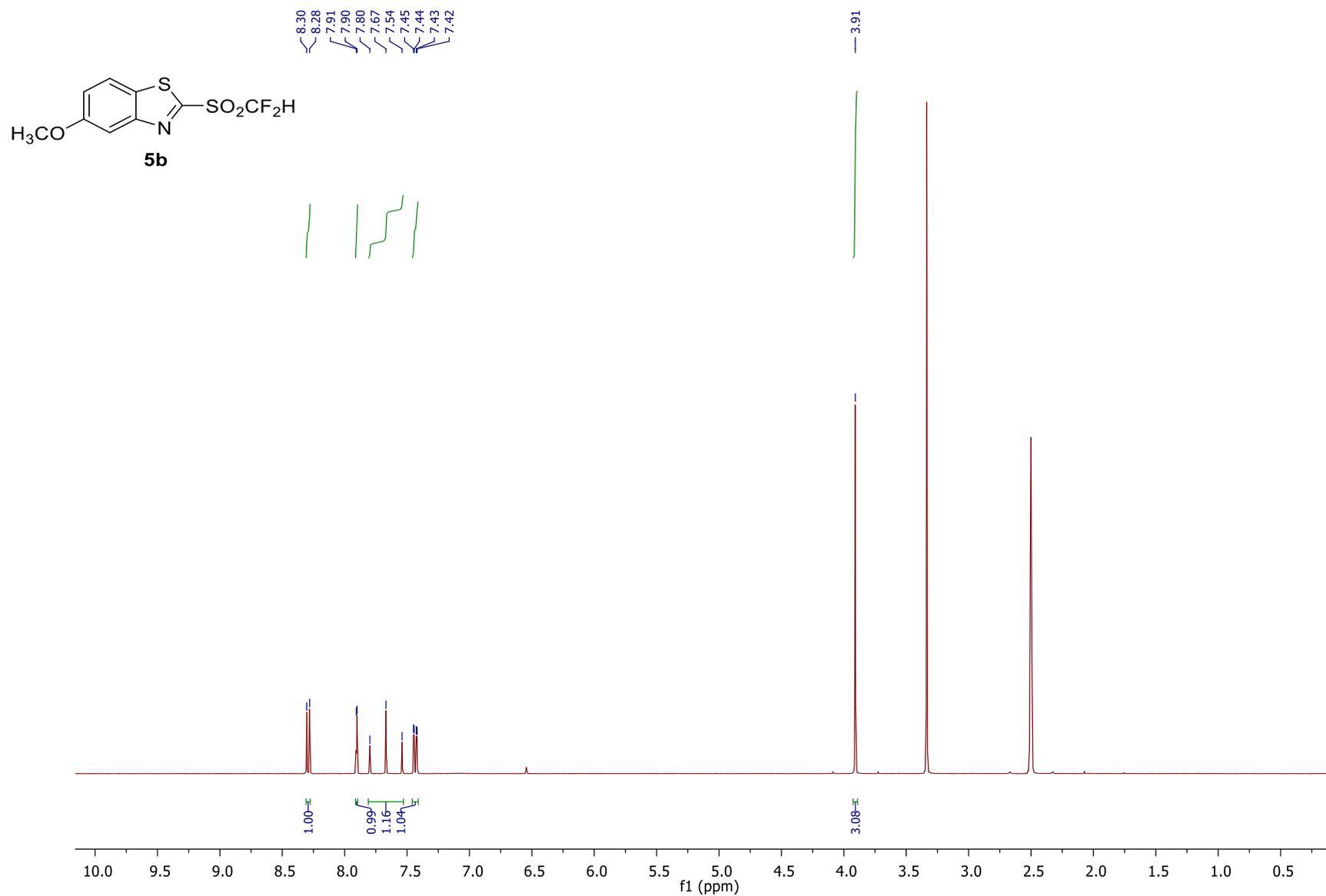
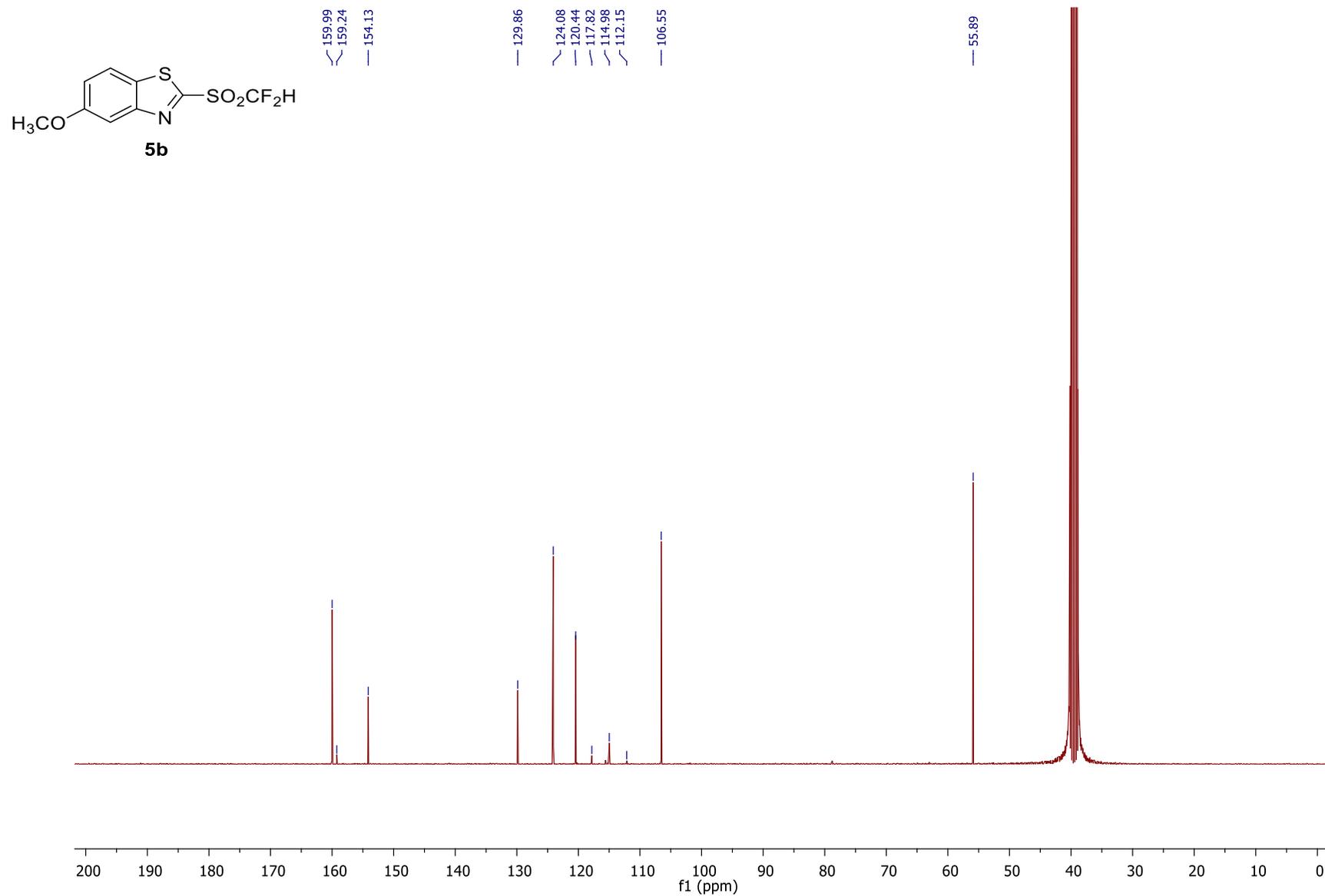
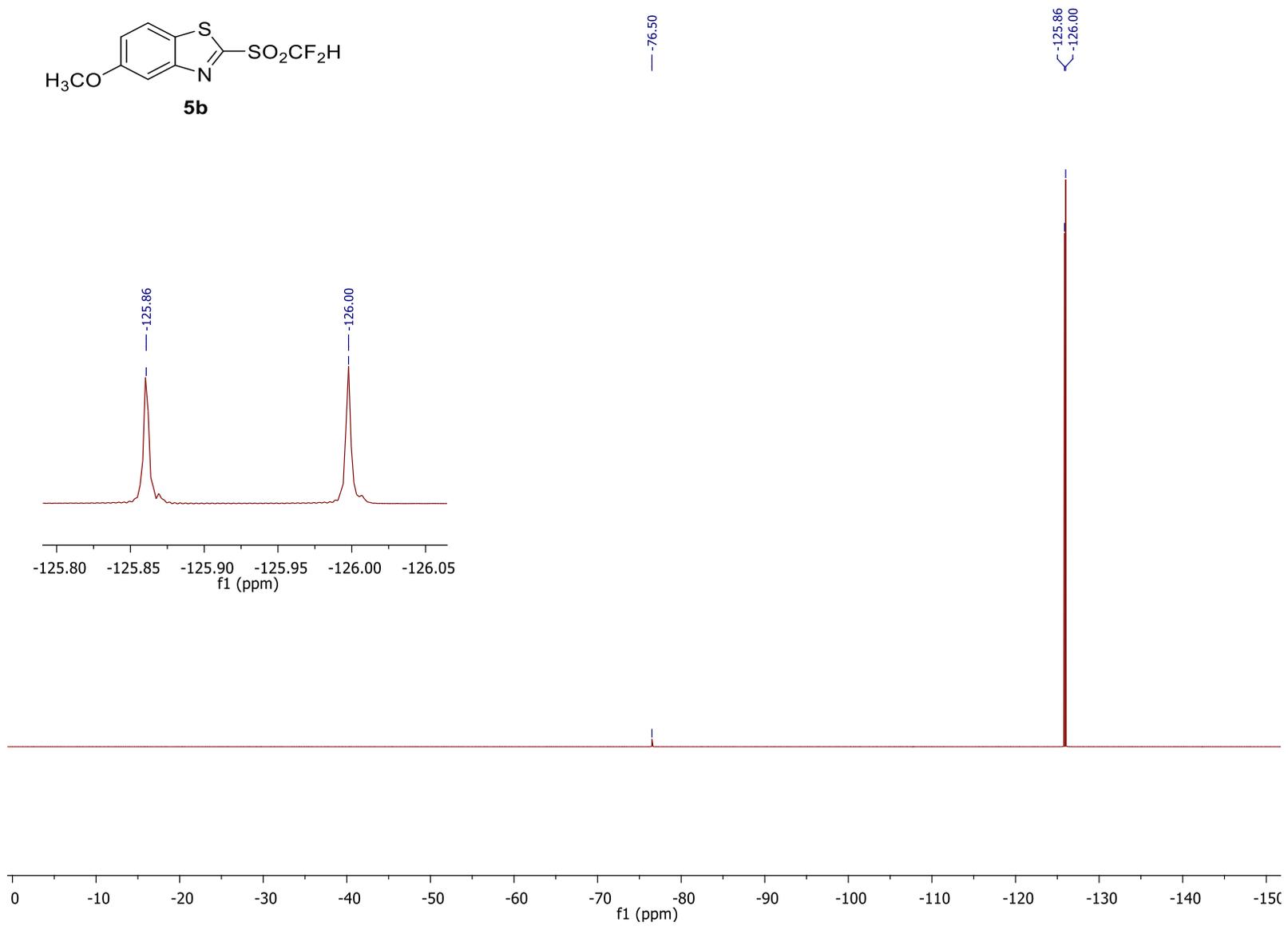


Figure S22. <sup>1</sup>H-NMR spectrum of 2-((difluoromethyl)sulfonyl)-5-methoxybenzo[d]thiazole (**5b**).



**Figure S23.** <sup>13</sup>C-NMR spectrum of 2-((difluoromethyl)sulfonyl)-5-methoxybenzo[d]thiazole (**5b**).



**Figure S24.**  $^{19}\text{F}$ -NMR spectrum of 2-((difluoromethyl)sulfonyl)-5-methoxybenzo[*d*]thiazole (**5b**).

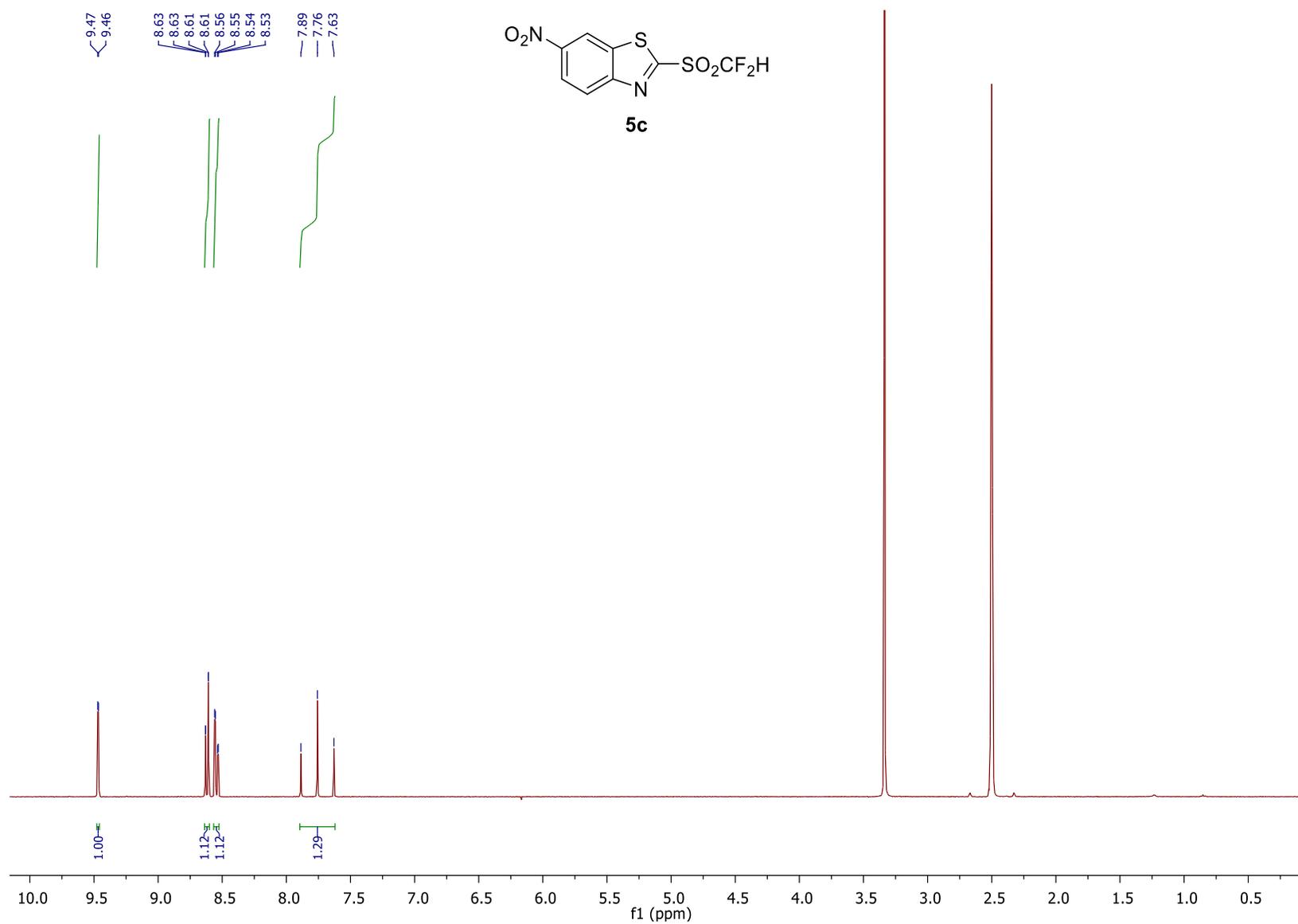
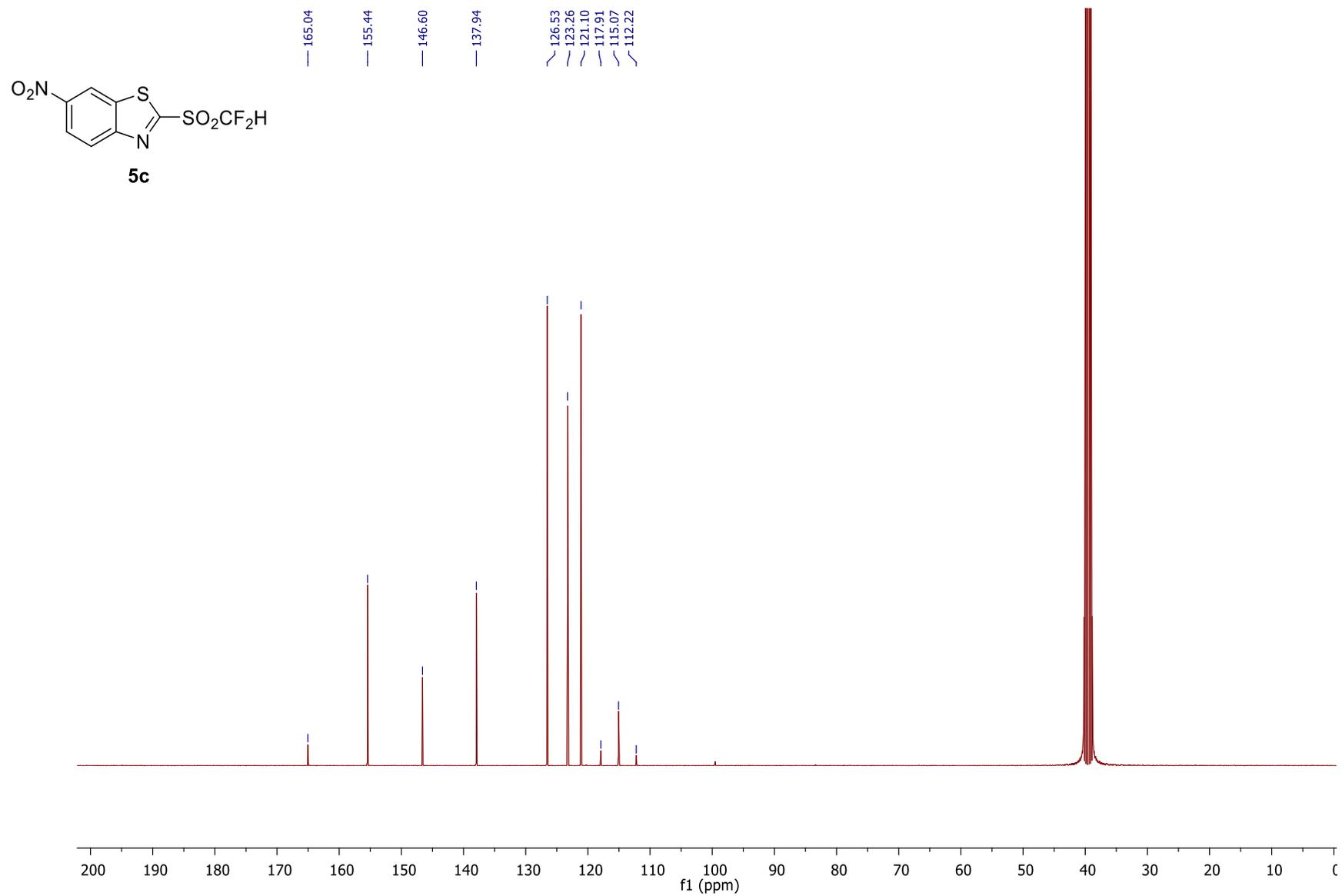
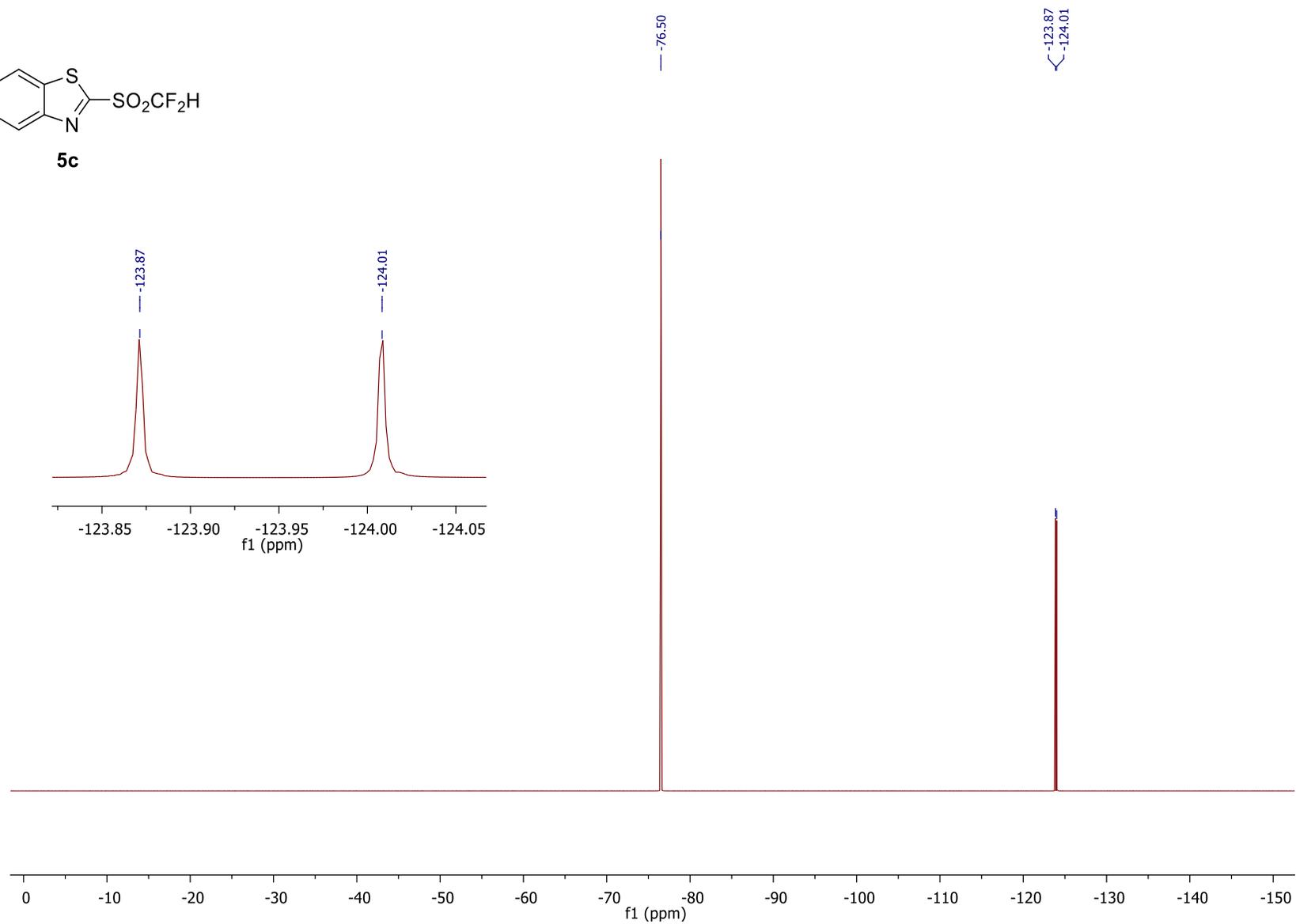
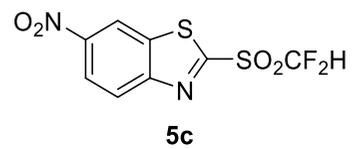


Figure S25. <sup>1</sup>H-NMR spectrum of 2-((difluoromethyl)sulfonyl)-6-nitrobenzo[d]thiazole (5c).



**Figure S26.** <sup>13</sup>C-NMR spectrum of 2-((difluoromethyl)sulfonyl)-6-nitrobenzo[d]thiazole (**5c**).



**Figure S27.**  $^{19}\text{F}$ -NMR spectrum of 2-((difluoromethyl)sulfonyl)-6-nitrobenzo[*d*]thiazole (**5c**).

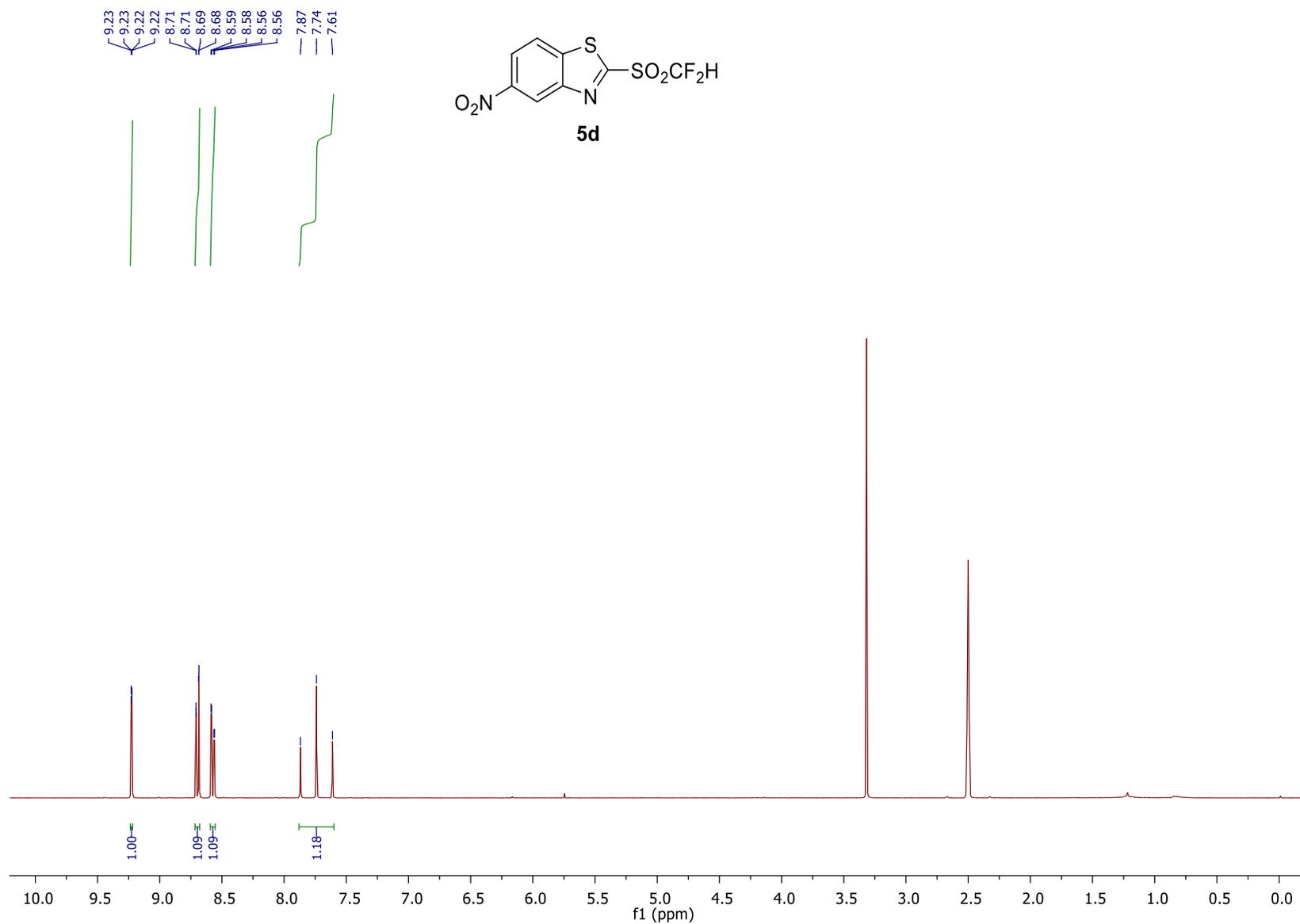
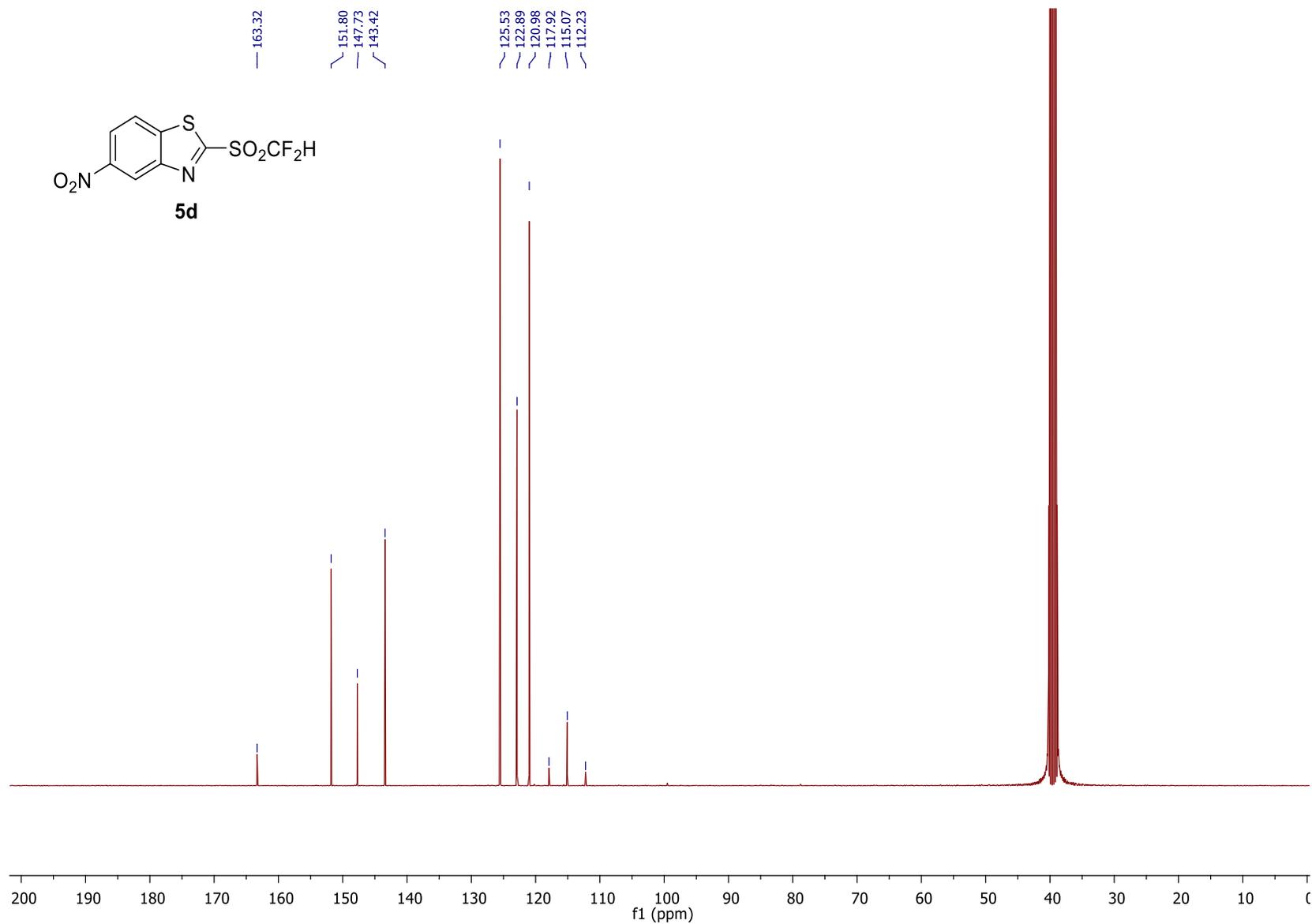
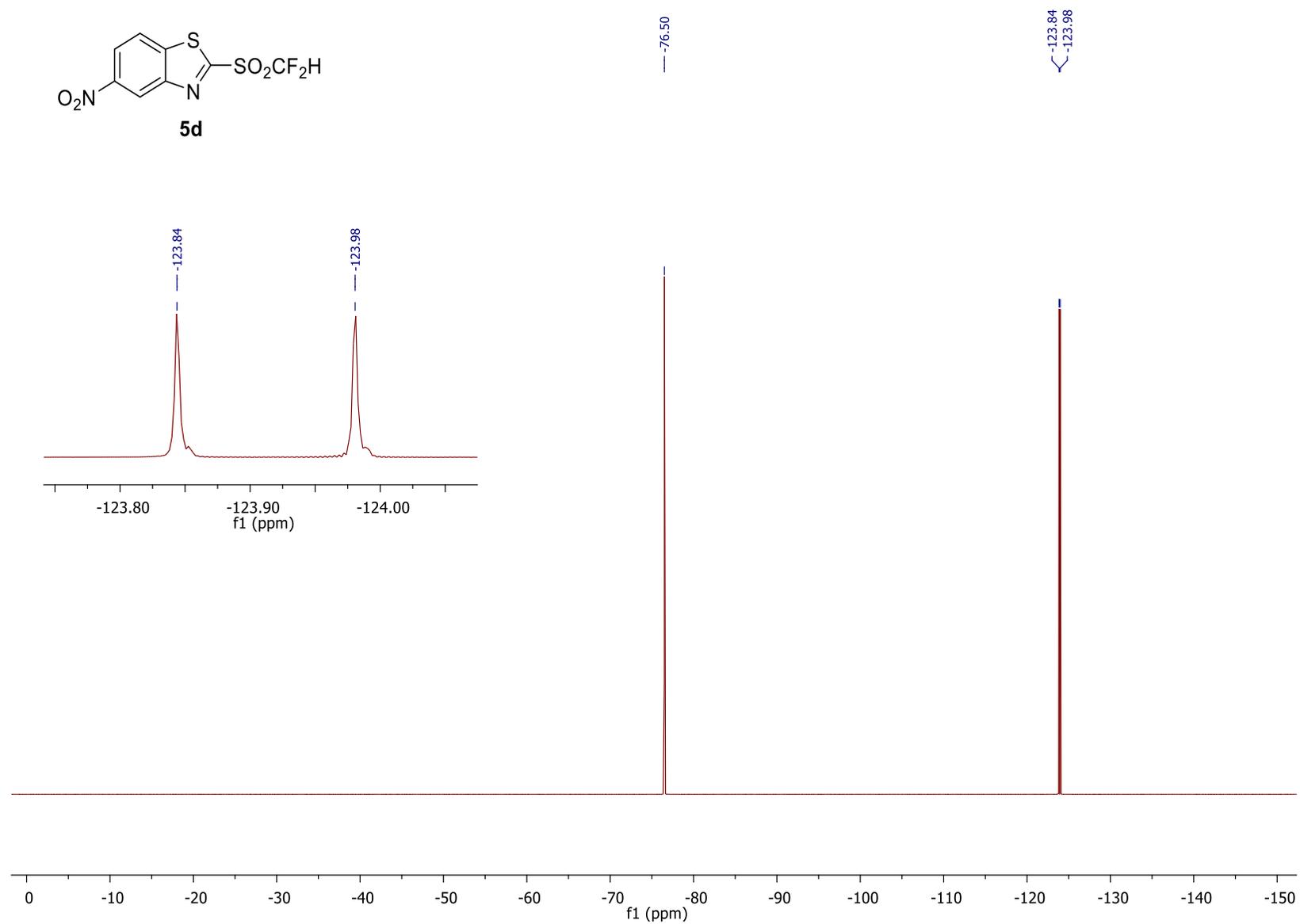


Figure S28. <sup>1</sup>H-NMR spectrum of 2-((difluoromethyl)sulfonyl)-5-nitrobenzo[d]thiazole (5d).



**Figure S29.** <sup>13</sup>C-NMR spectrum of 2-((difluoromethyl)sulfonyl)-5-nitrobenzo[d]thiazole (**5d**).



**Figure S30.**  $^{19}\text{F}$ -NMR spectrum of 2-((difluoromethyl)sulfonyl)-5-nitrobenzo[*d*]thiazole (**5d**).

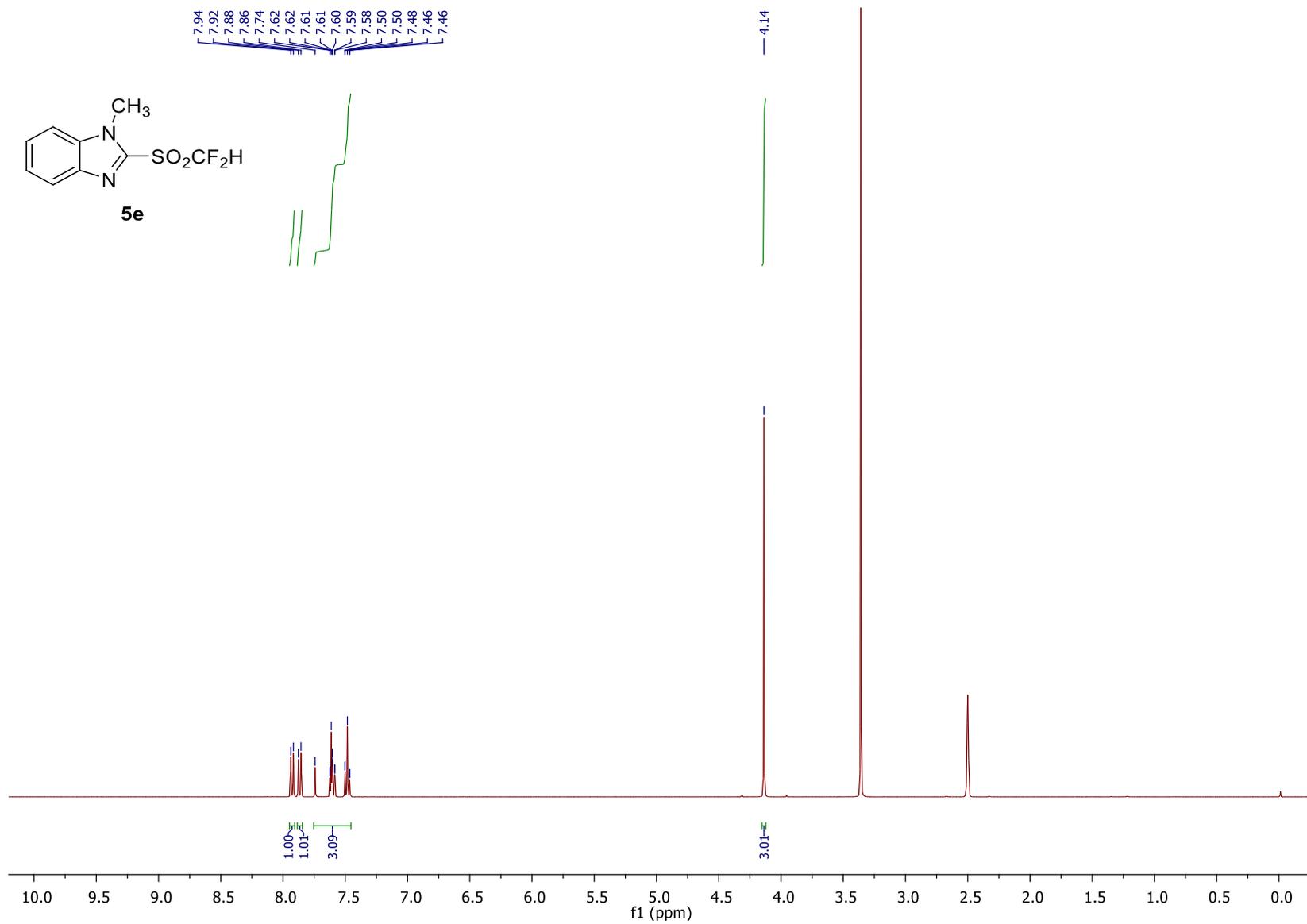


Figure S31. <sup>1</sup>H-NMR spectrum of 2-((difluoromethyl)sulfonyl)-1-methyl-1H-benzo[d]imidazole (5e).

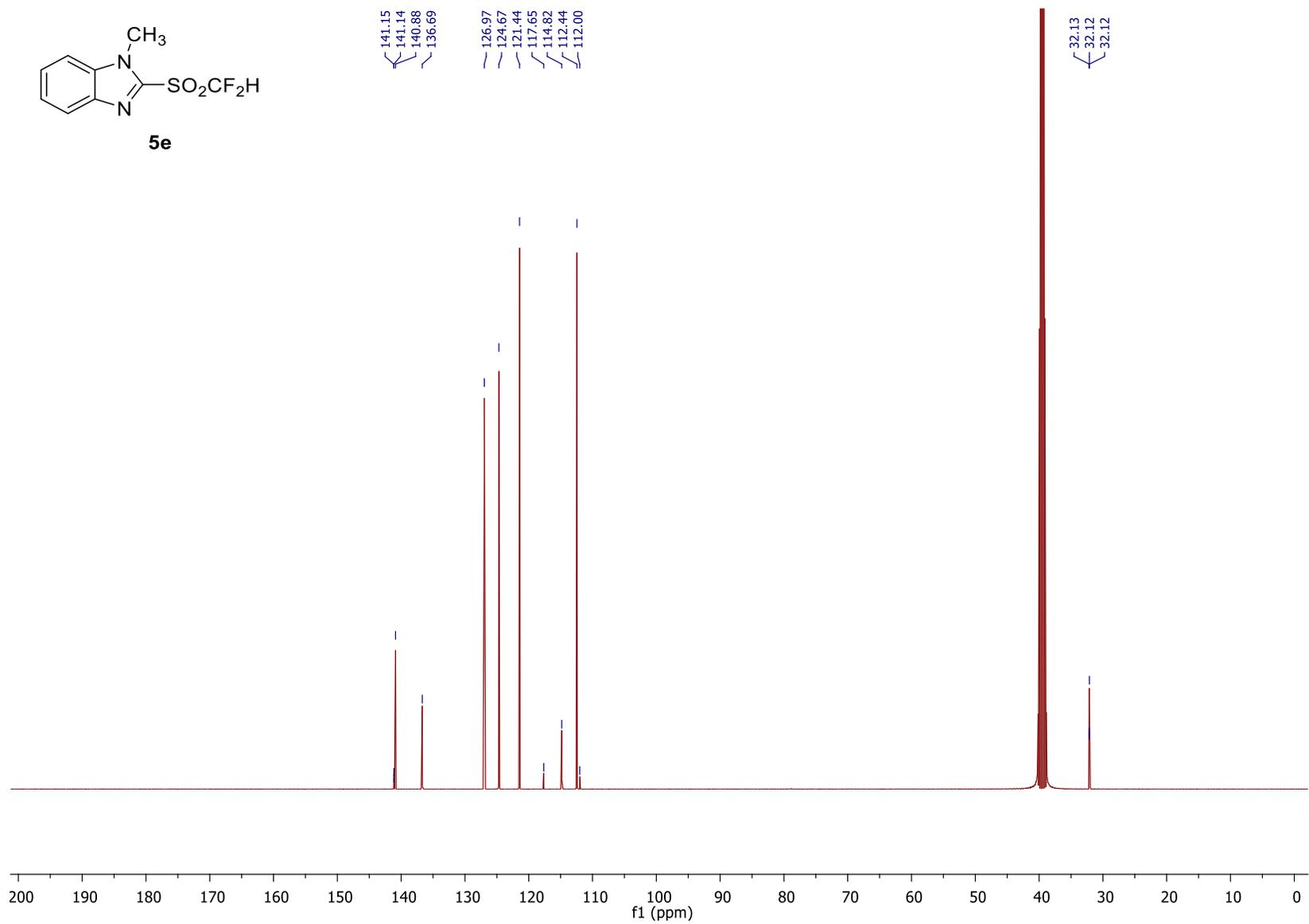
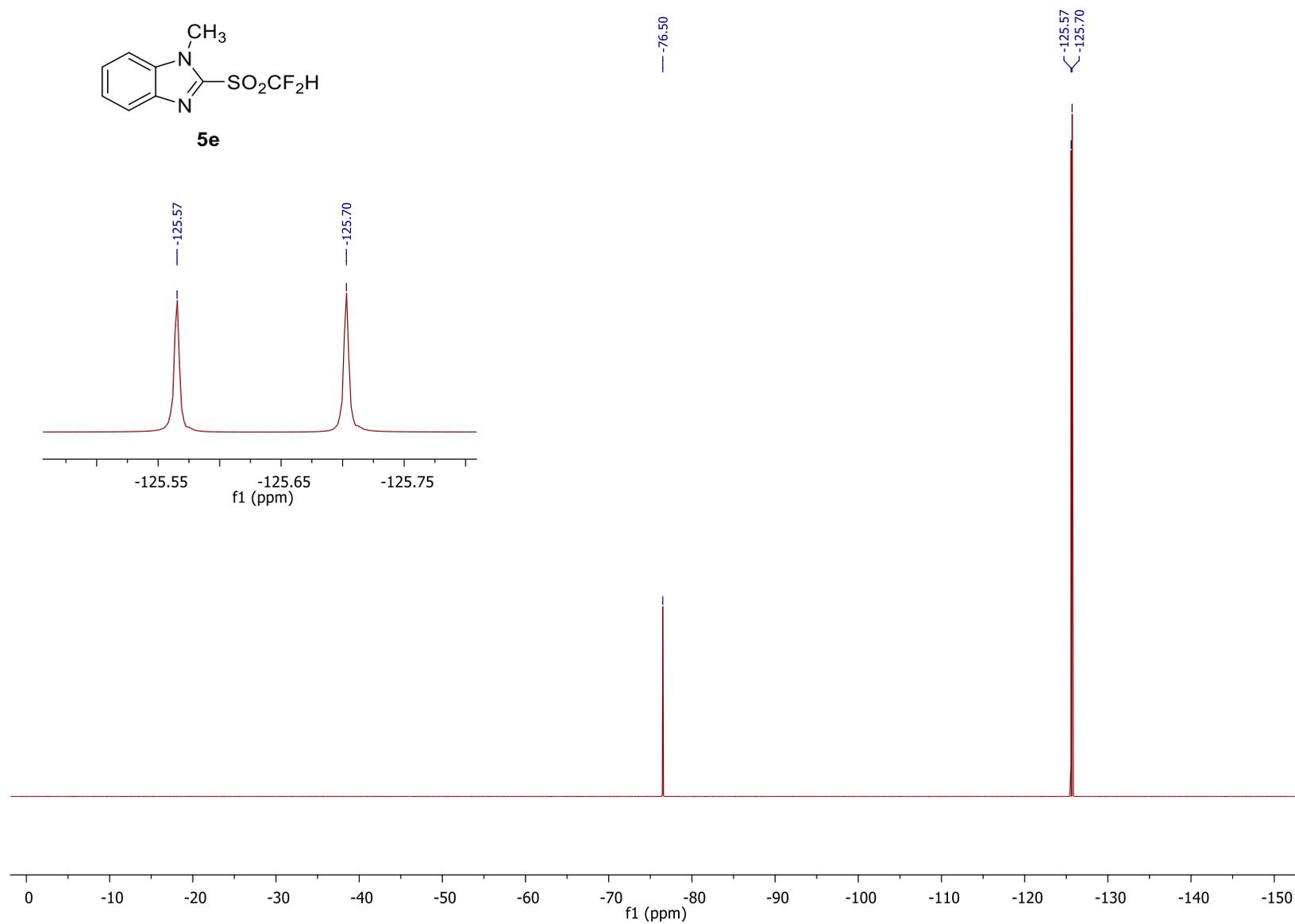


Figure S32. <sup>13</sup>C-NMR spectrum of 2-((difluoromethyl)sulfonyl)-1-methyl-1H-benzo[d]imidazole (5e).



**Figure S33.**  $^{19}\text{F}$ -NMR spectrum of 2-((difluoromethyl)sulfonyl)-1-methyl-1H-benzo[d]imidazole (**5e**).

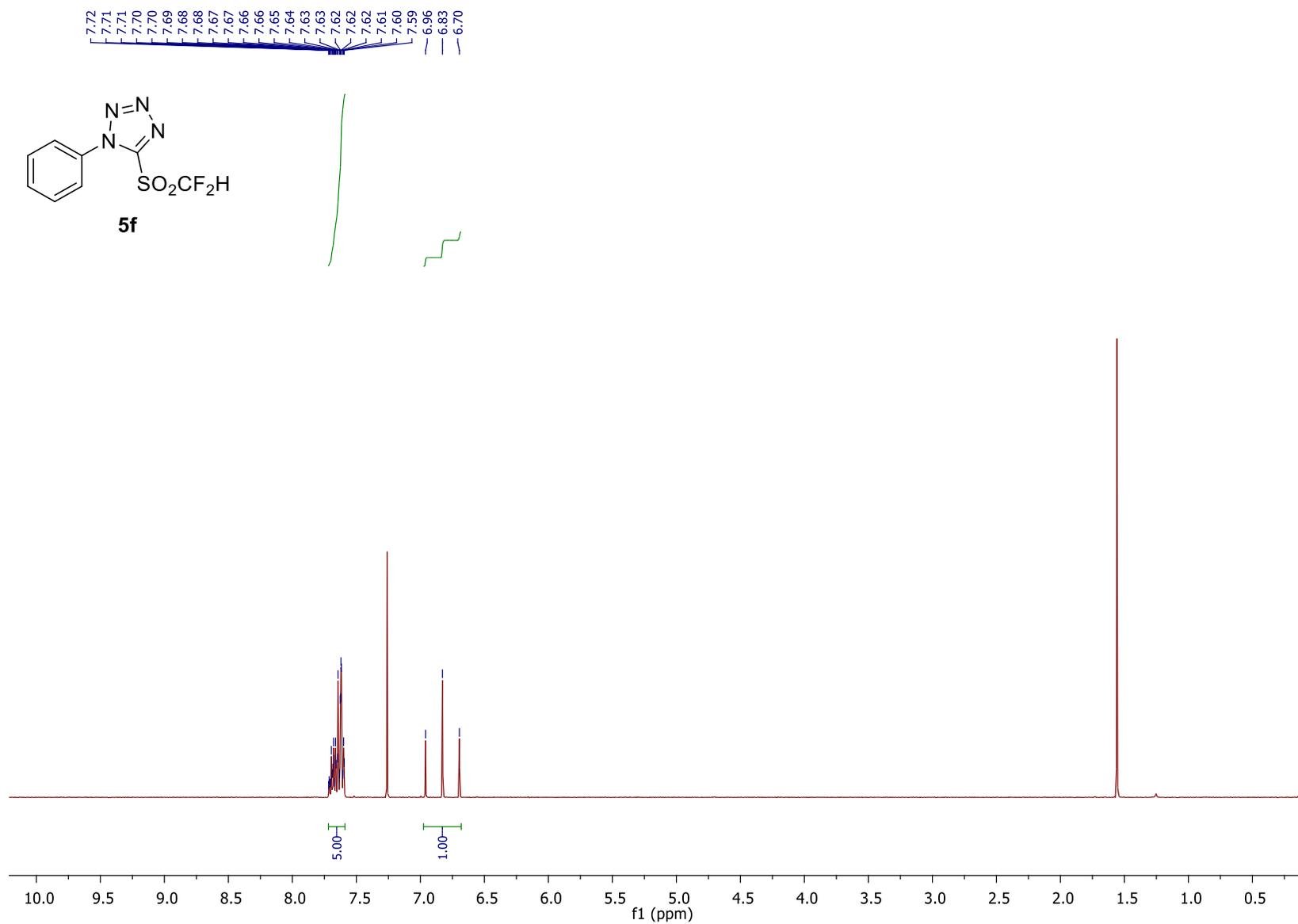


Figure S34.  $^1\text{H-NMR}$  spectrum of 5-((difluoromethyl)sulfonyl)-1-phenyl-1H-tetrazole (**5f**).

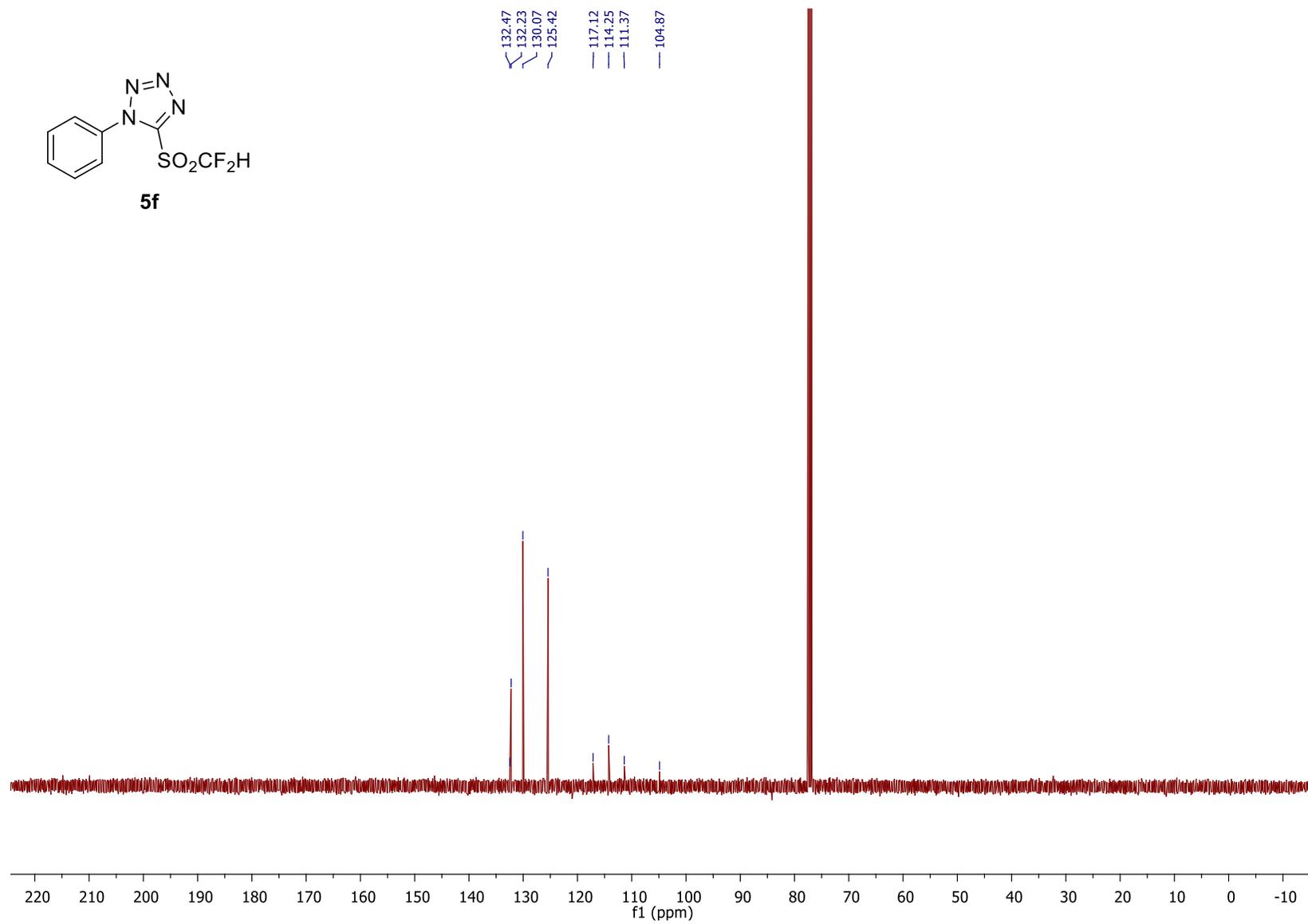


Figure S35. <sup>13</sup>C-NMR spectrum of 5-((difluoromethyl)sulfonyl)-1-phenyl-1H-tetrazole (5f).

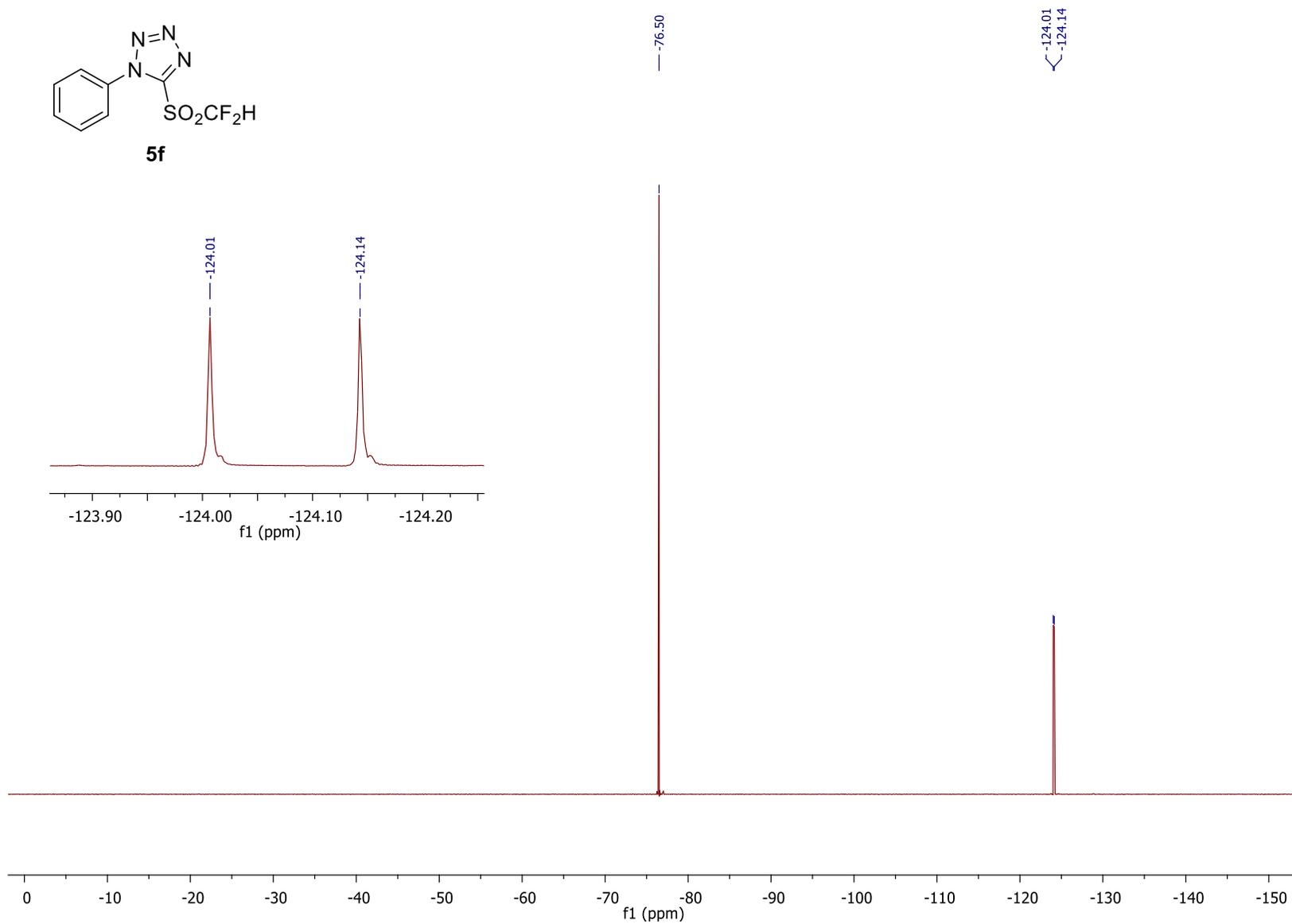


Figure S36.  $^{19}\text{F}$ -NMR spectrum of 5-((difluoromethyl)sulfonyl)-1-phenyl-1H-tetrazole (**5f**).

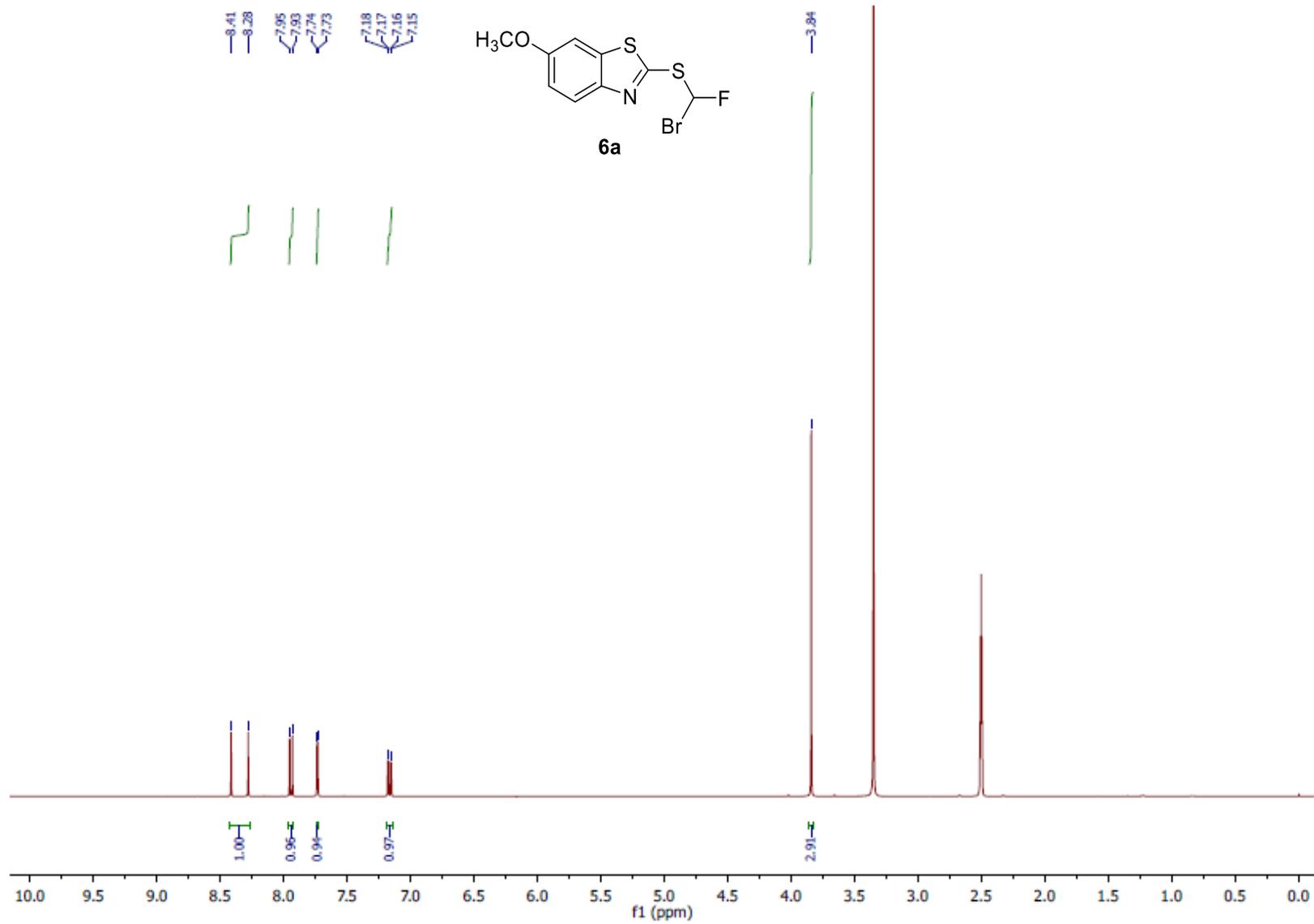


Figure S37. <sup>1</sup>H-NMR spectrum of 2-((bromofluoromethyl)thio)-6-methoxybenzo[d]thiazole (6a).

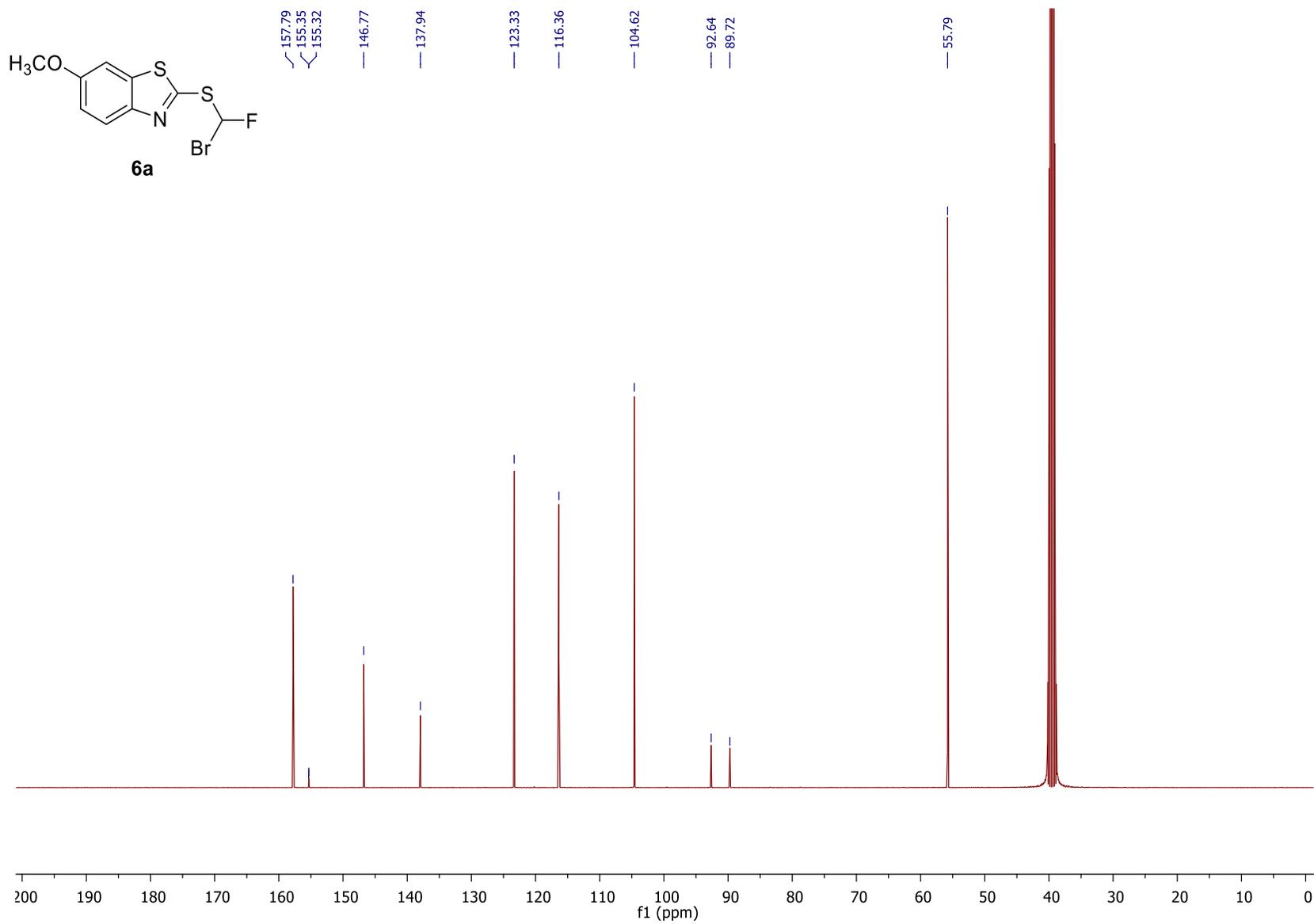
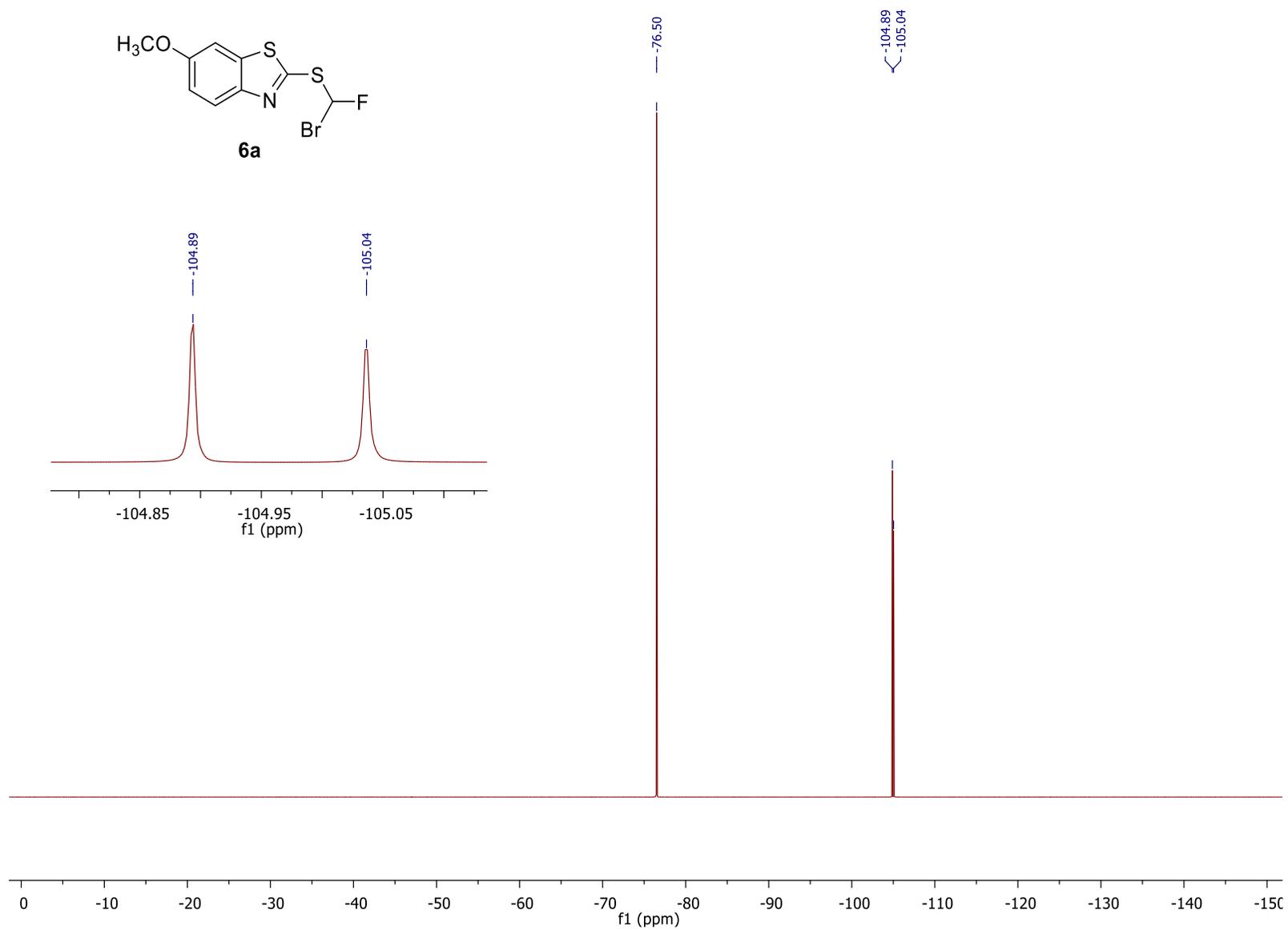


Figure S38. <sup>13</sup>C-NMR spectrum of 2-((bromofluoromethyl)thio)-6-methoxybenzo[d]thiazole (6a).



**Figure S39.**  $^{19}\text{F}$ -NMR spectrum of 2-((bromofluoromethyl)thio)-6-methoxybenzo[d]thiazole (**6a**).

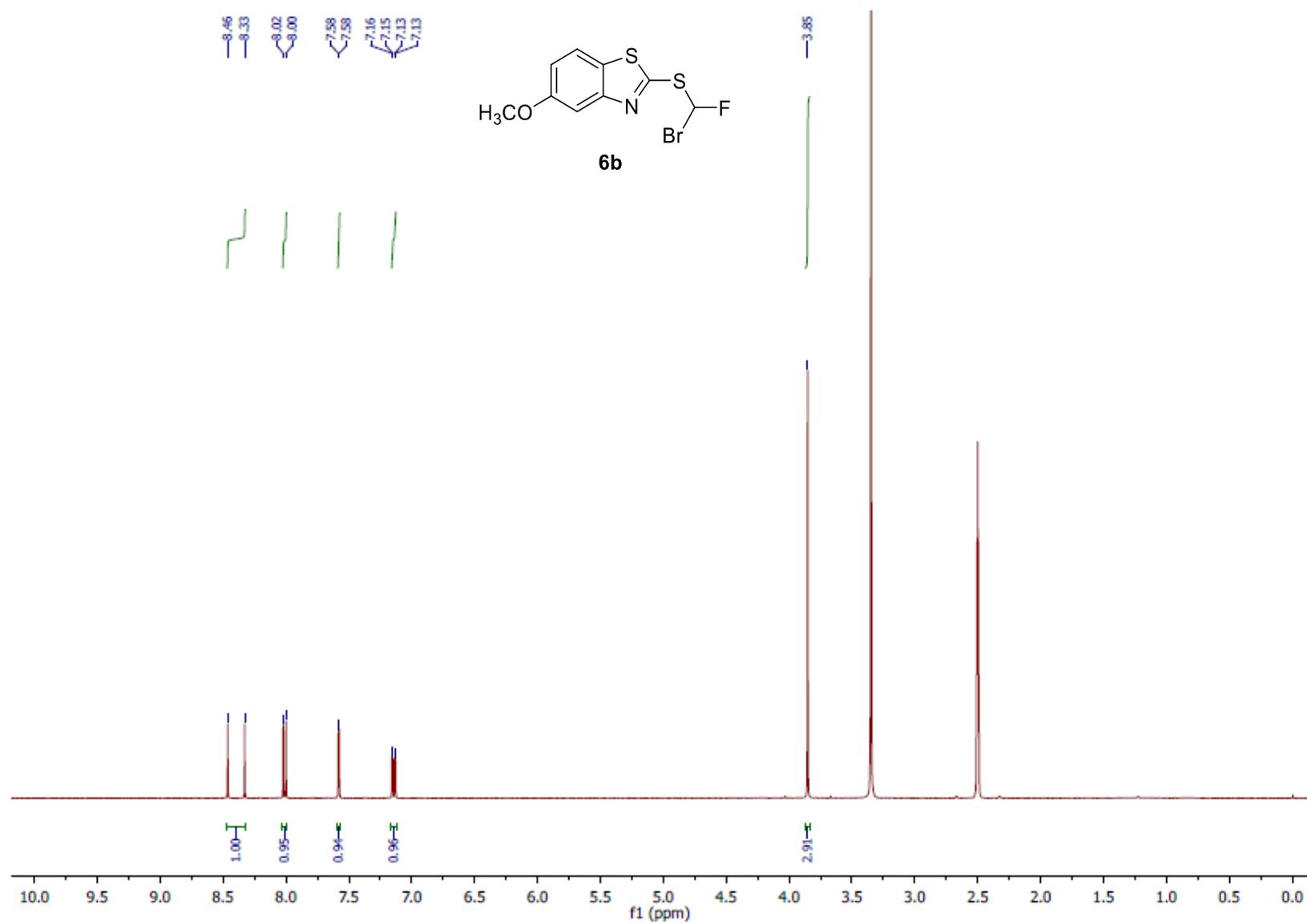
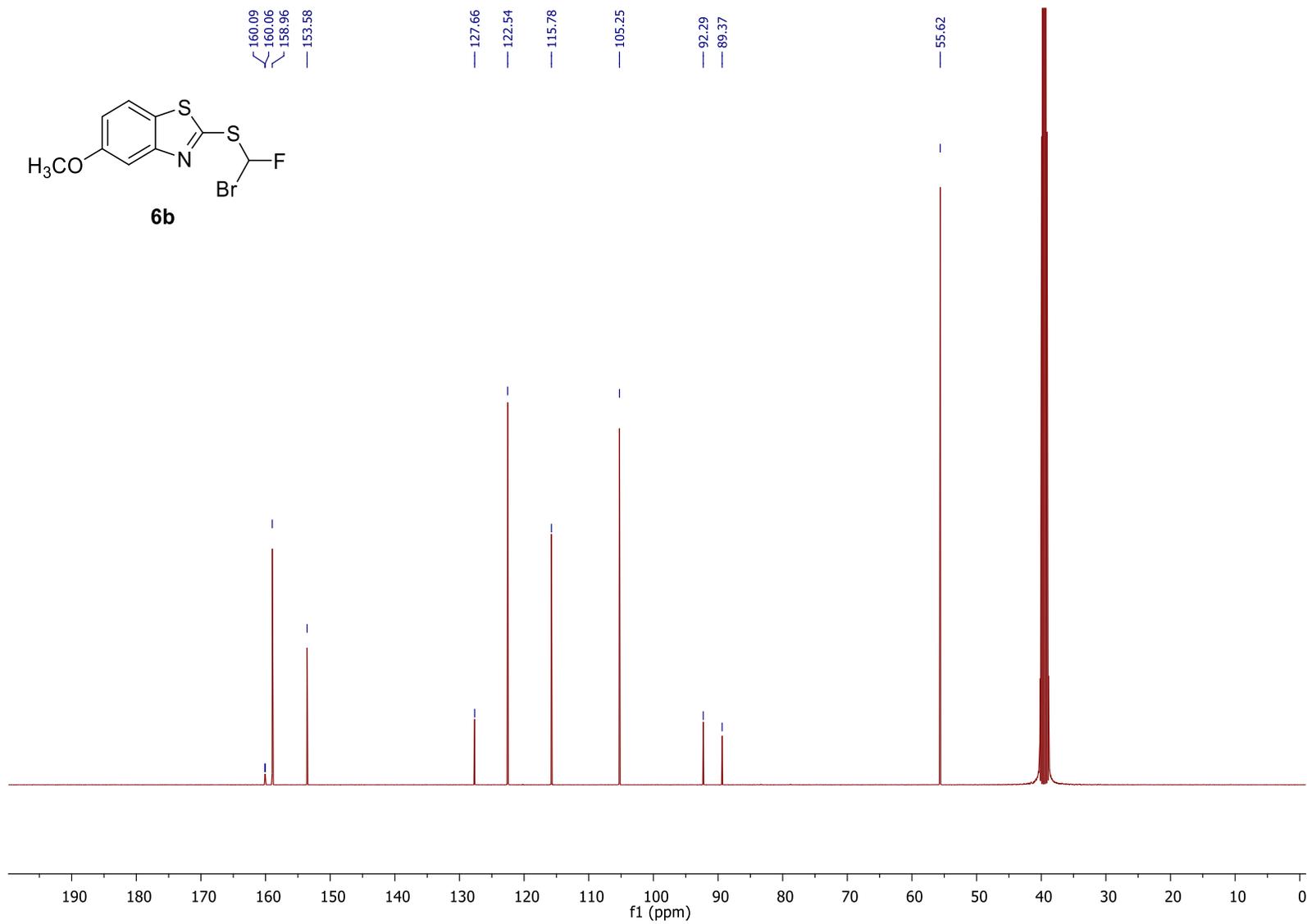
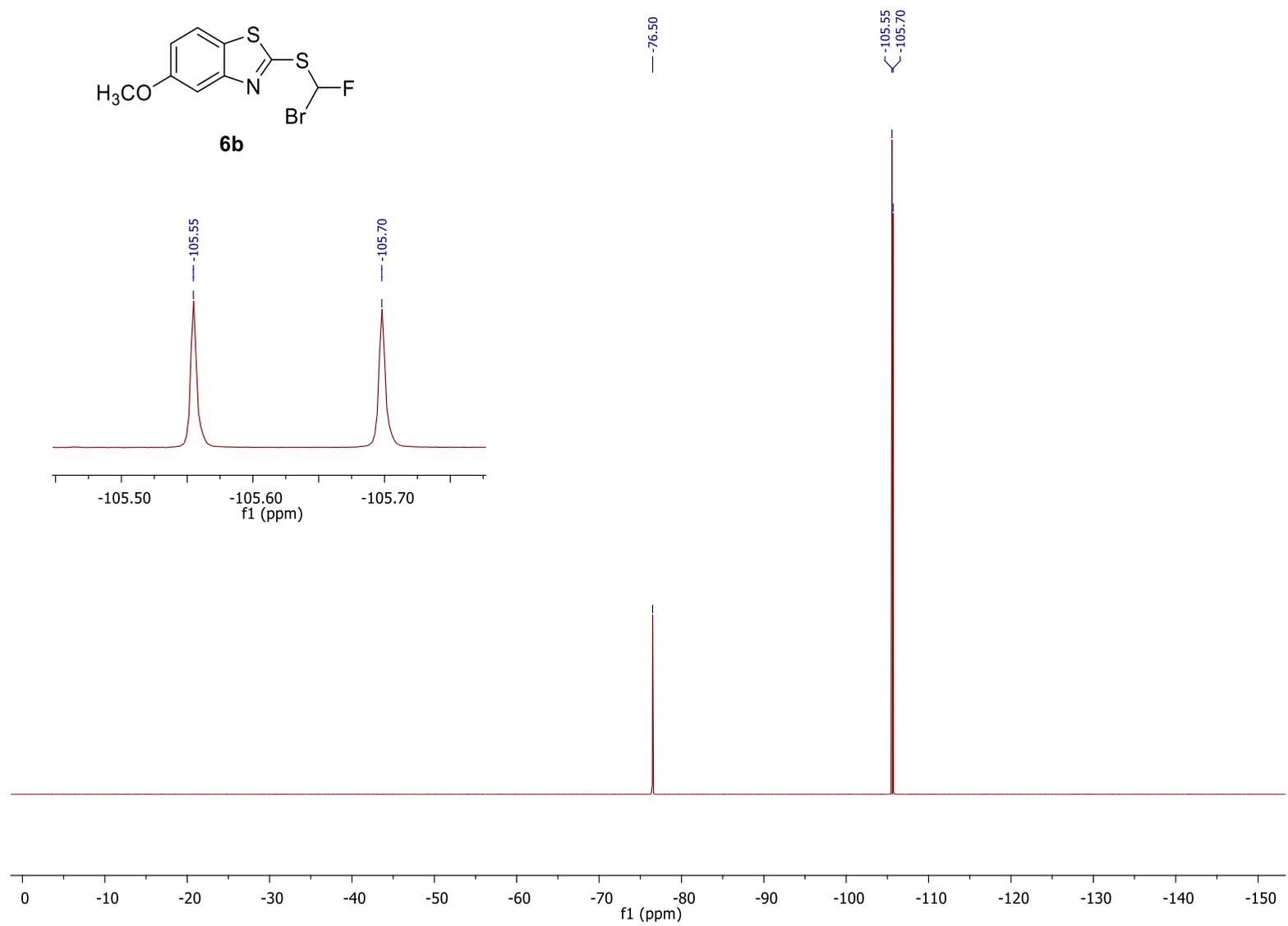


Figure S40. <sup>1</sup>H-NMR spectrum of 2-((bromofluoromethyl)thio)-5-methoxybenzo[d]thiazole (6b).



**Figure S41.** <sup>13</sup>C-NMR spectrum of 2-((bromofluoromethyl)thio)-5-methoxybenzo[*d*]thiazole (**6b**).



**Figure S42.**  $^{19}\text{F}$ -NMR spectrum of 2-((bromofluoromethyl)thio)-5-methoxybenzo[d]thiazole (**6b**).

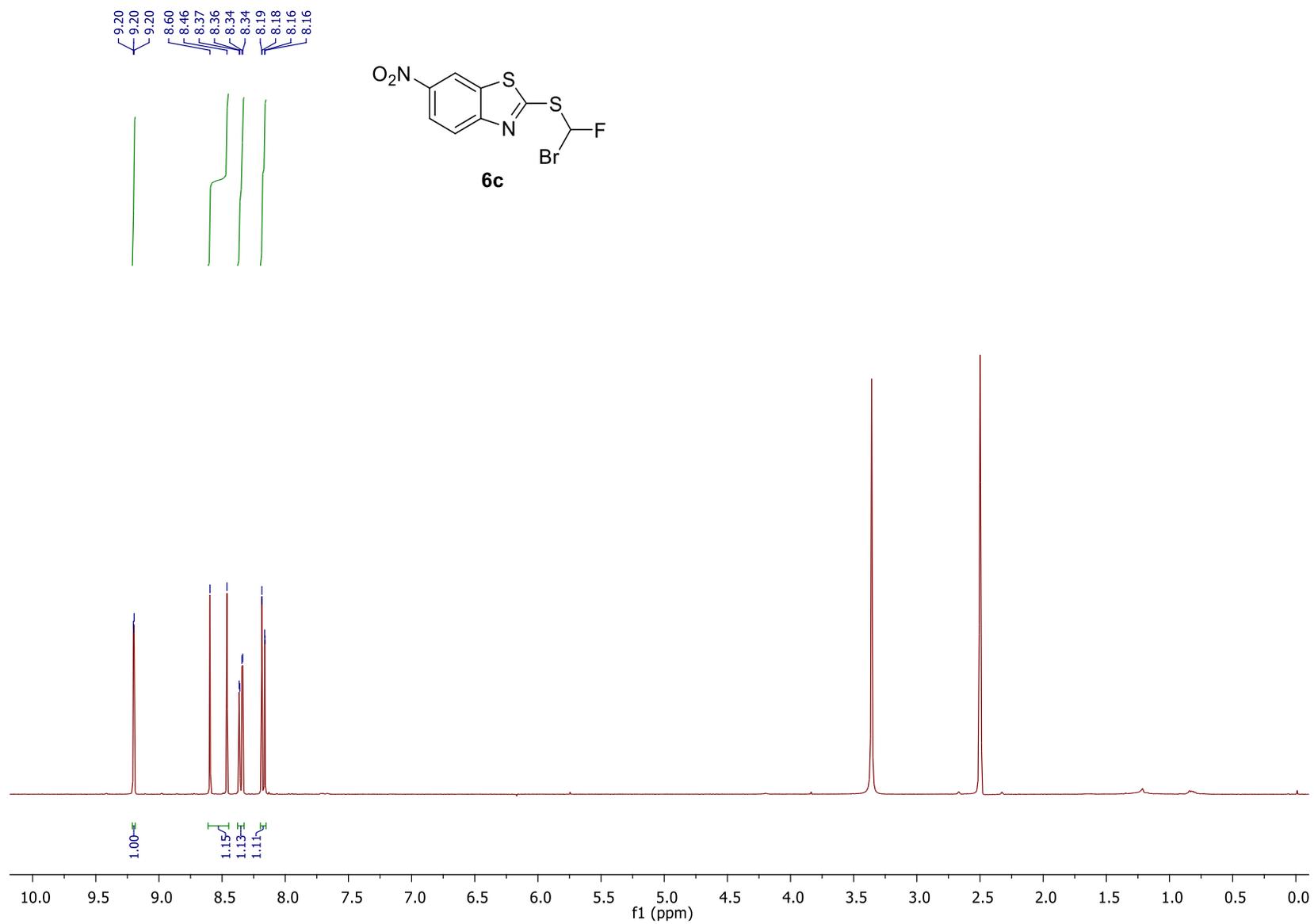
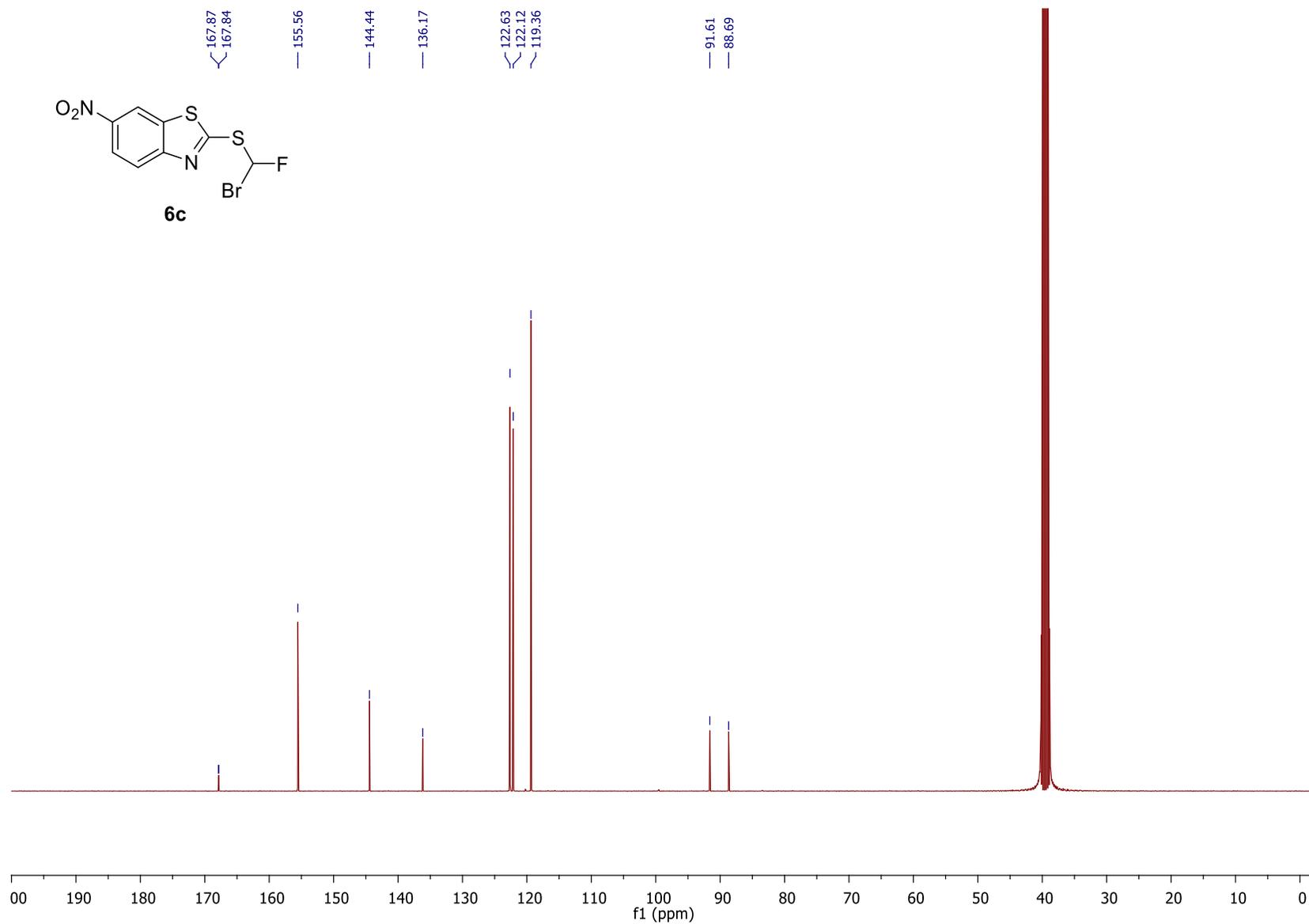
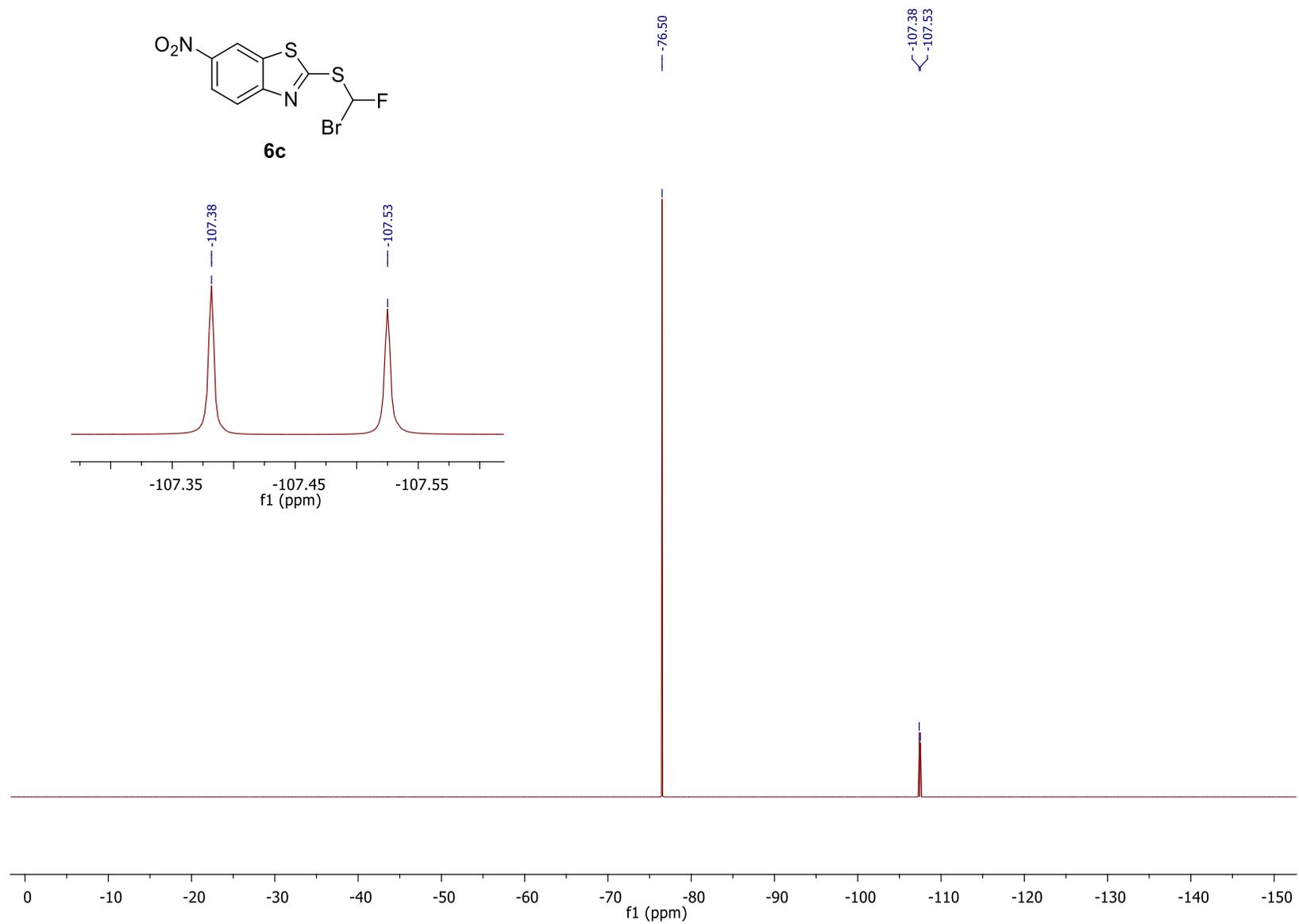


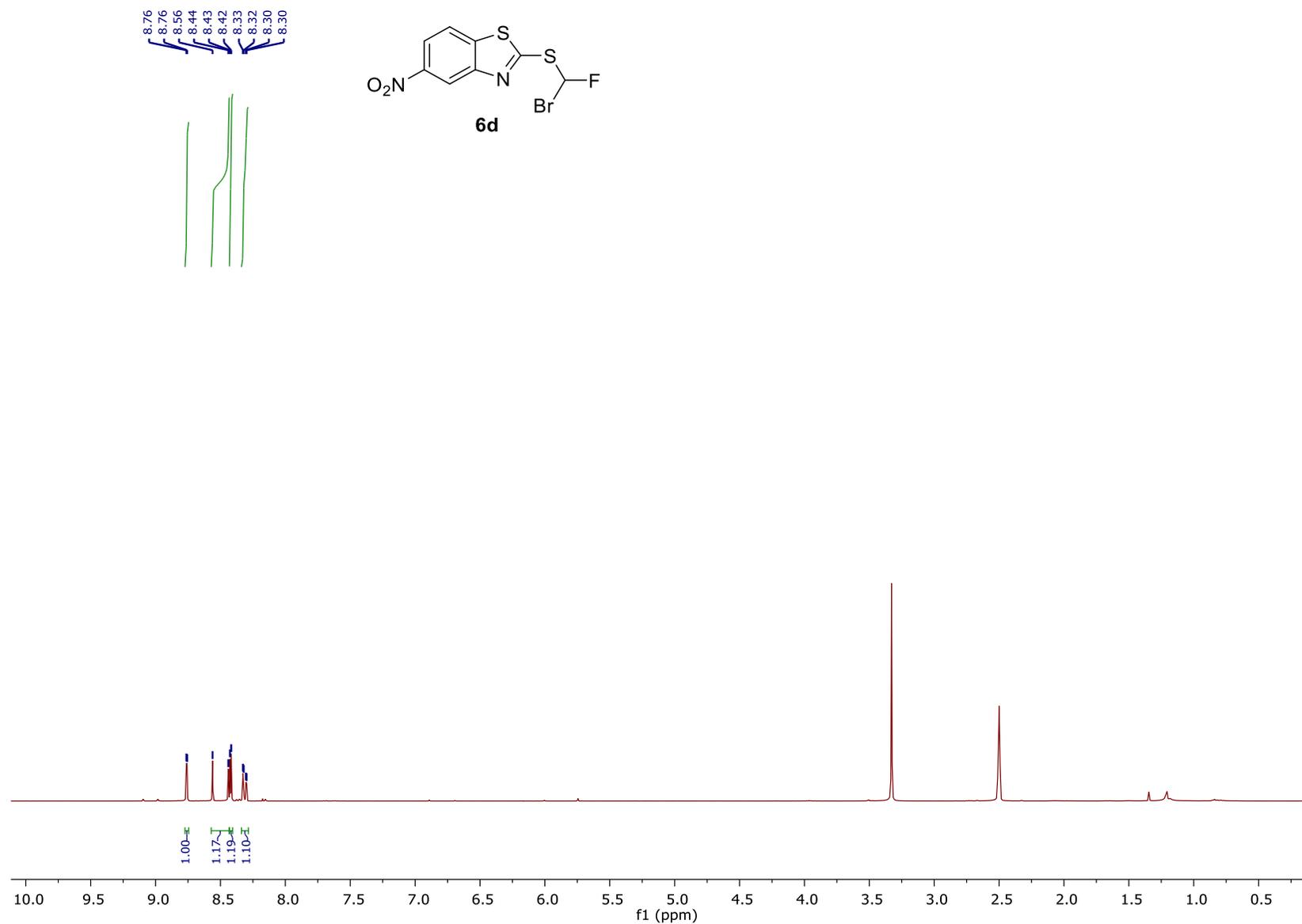
Figure S43. <sup>1</sup>H-NMR spectrum of 2-((bromofluoromethyl)thio)-6-nitrobenzo[*d*]thiazole (6c).



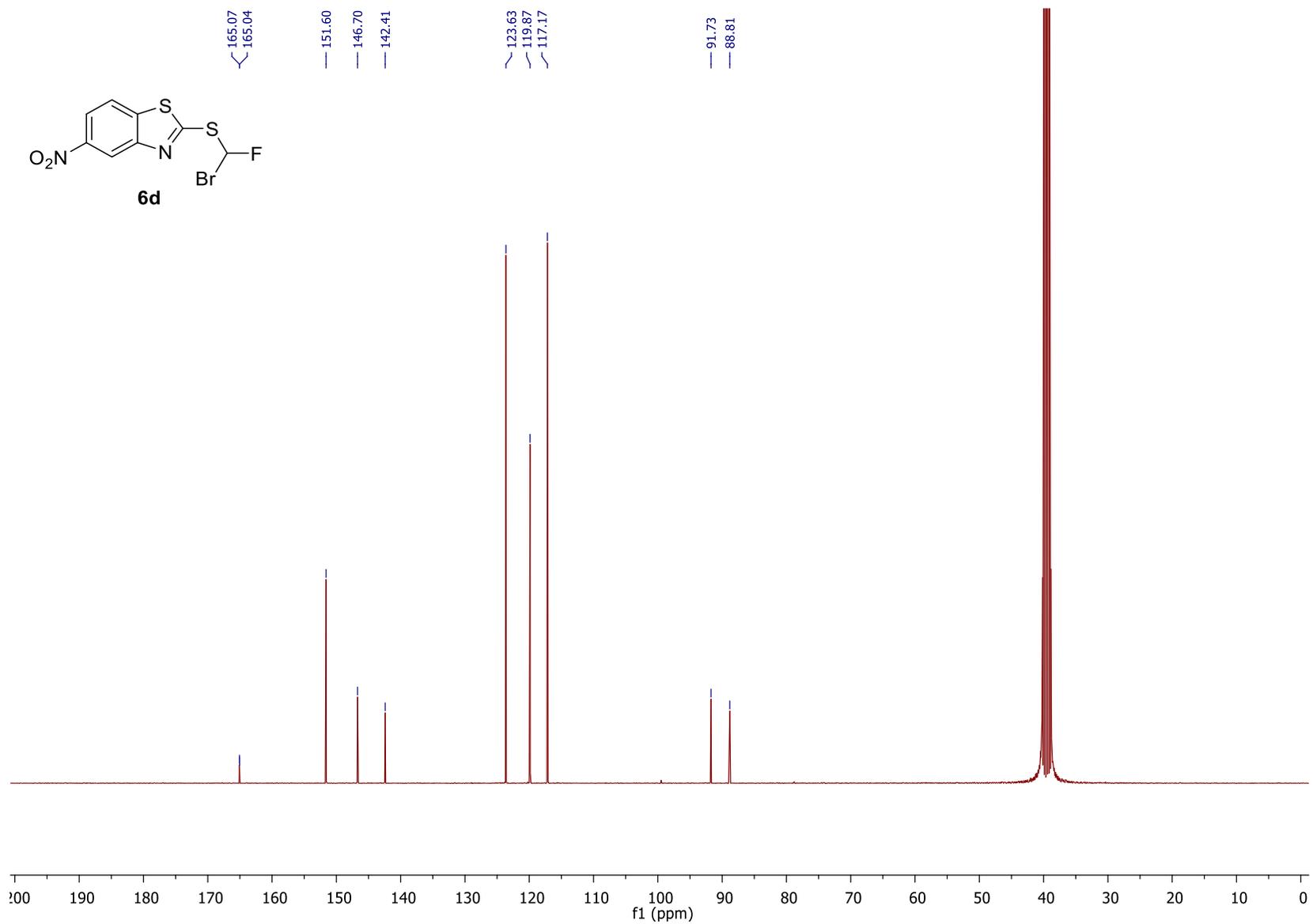
**Figure S44.** <sup>13</sup>C-NMR spectrum of 2-((bromofluoromethyl)thio)-6-nitrobenzo[*d*]thiazole (**6c**).



**Figure S45.**  $^{19}\text{F}$ -NMR spectrum of 2-((bromofluoromethyl)thio)-6-nitrobenzo[*d*]thiazole (**6c**).



**Figure S46.** <sup>1</sup>H-NMR spectrum of 2-((bromofluoromethyl)thio)-5-nitrobenzo[d]thiazole (**6d**).



**Figure S47.** <sup>13</sup>C-NMR spectrum of 2-((bromofluoromethyl)thio)-5-nitrobenzo[*d*]thiazole (**6d**).

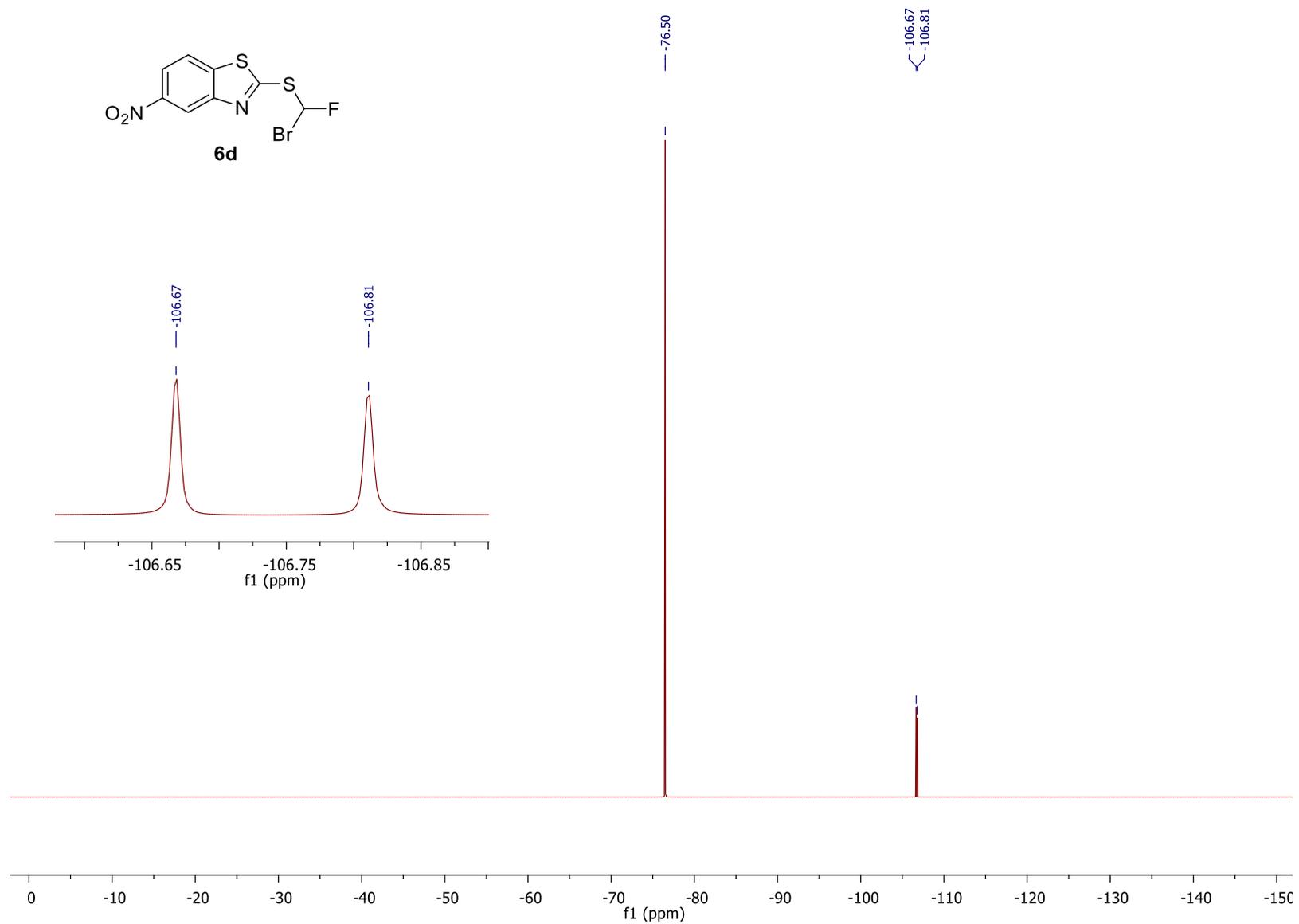


Figure S48.  $^{19}\text{F}$ -NMR spectrum of 2-((bromofluoromethyl)thio)-5-nitrobenzo[*d*]thiazole (6d).

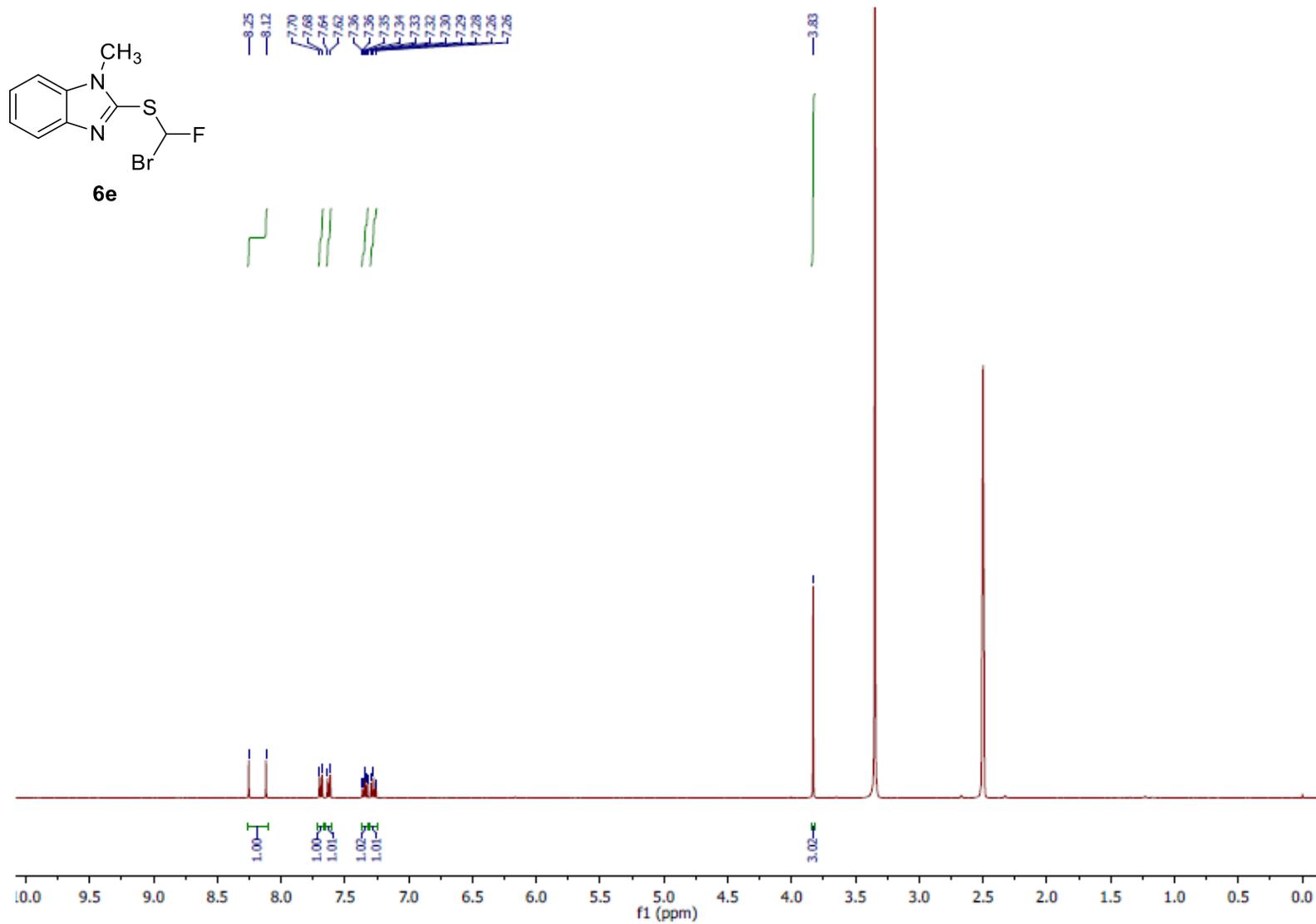


Figure S49. <sup>1</sup>H-NMR spectrum of 2-((bromofluoromethyl)thio)-1-methyl-1H-benzo[d]imidazole (6e).

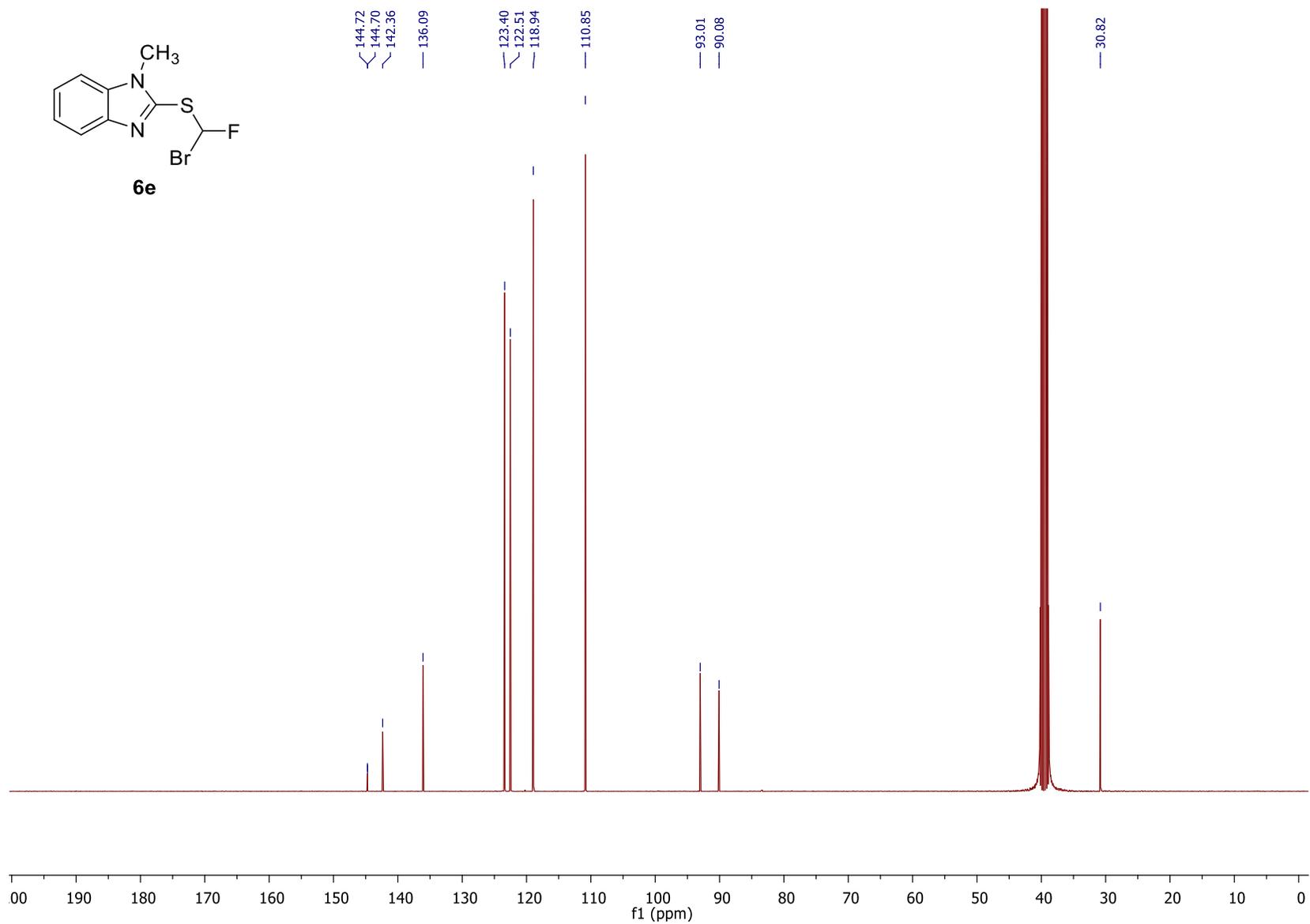
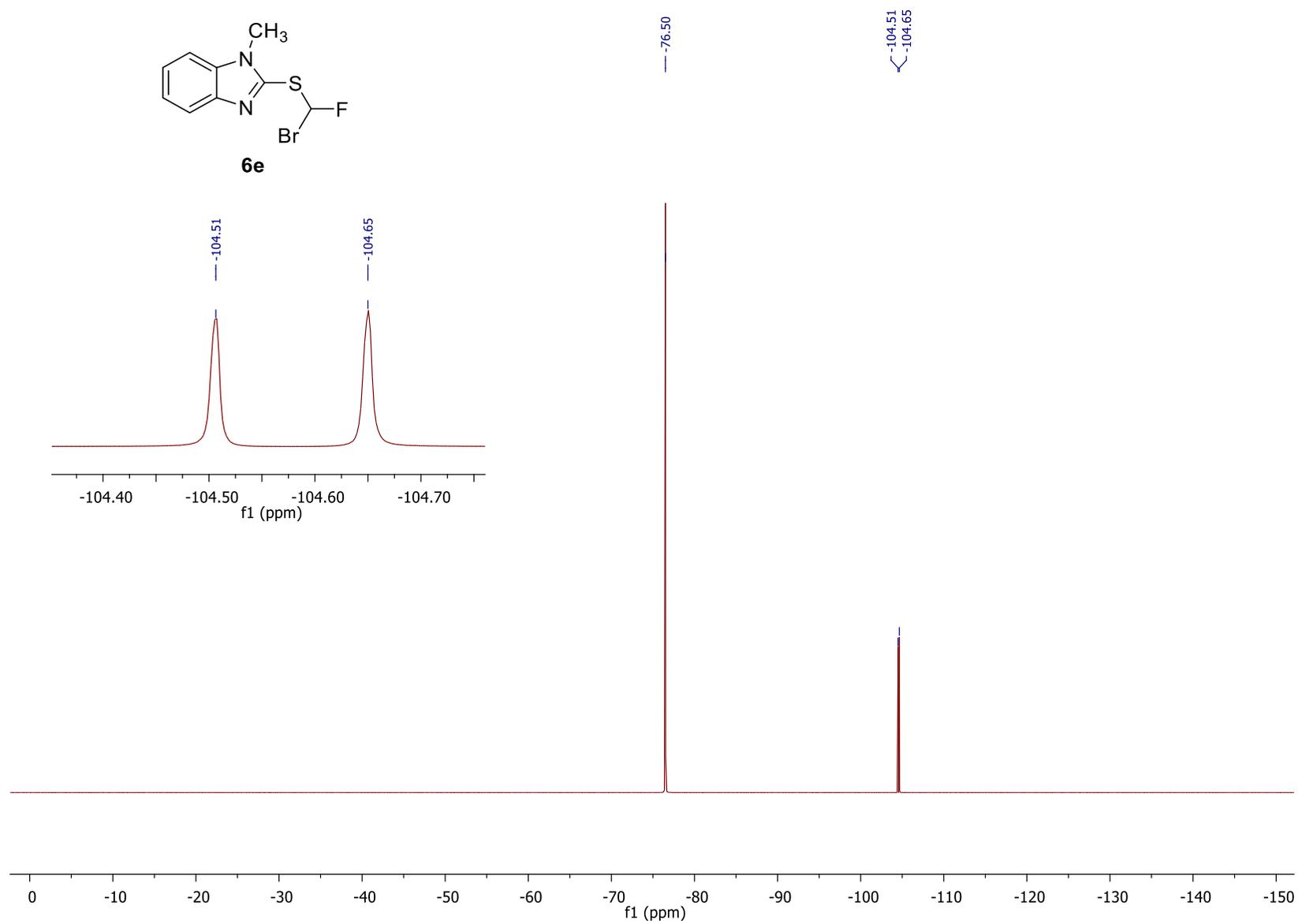


Figure S50. <sup>13</sup>C-NMR spectrum of 2-((bromofluoromethyl)thio)-1-methyl-1H-benzo[d]imidazole (6e).



**Figure S51.**  $^{19}\text{F}$ -NMR spectrum of 2-((bromofluoromethyl)thio)-1-methyl-1H-benzo[d]imidazole (**6e**).

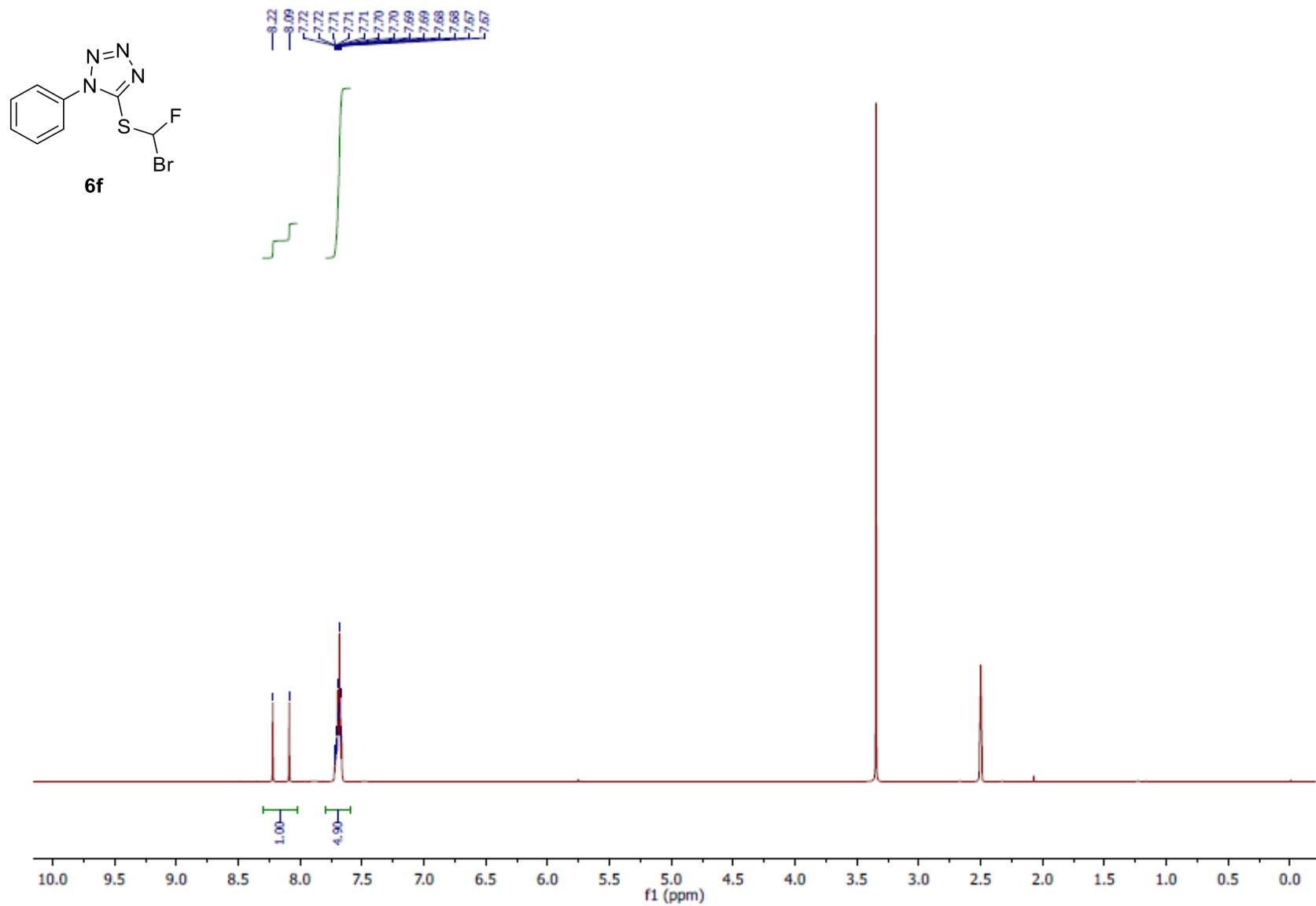
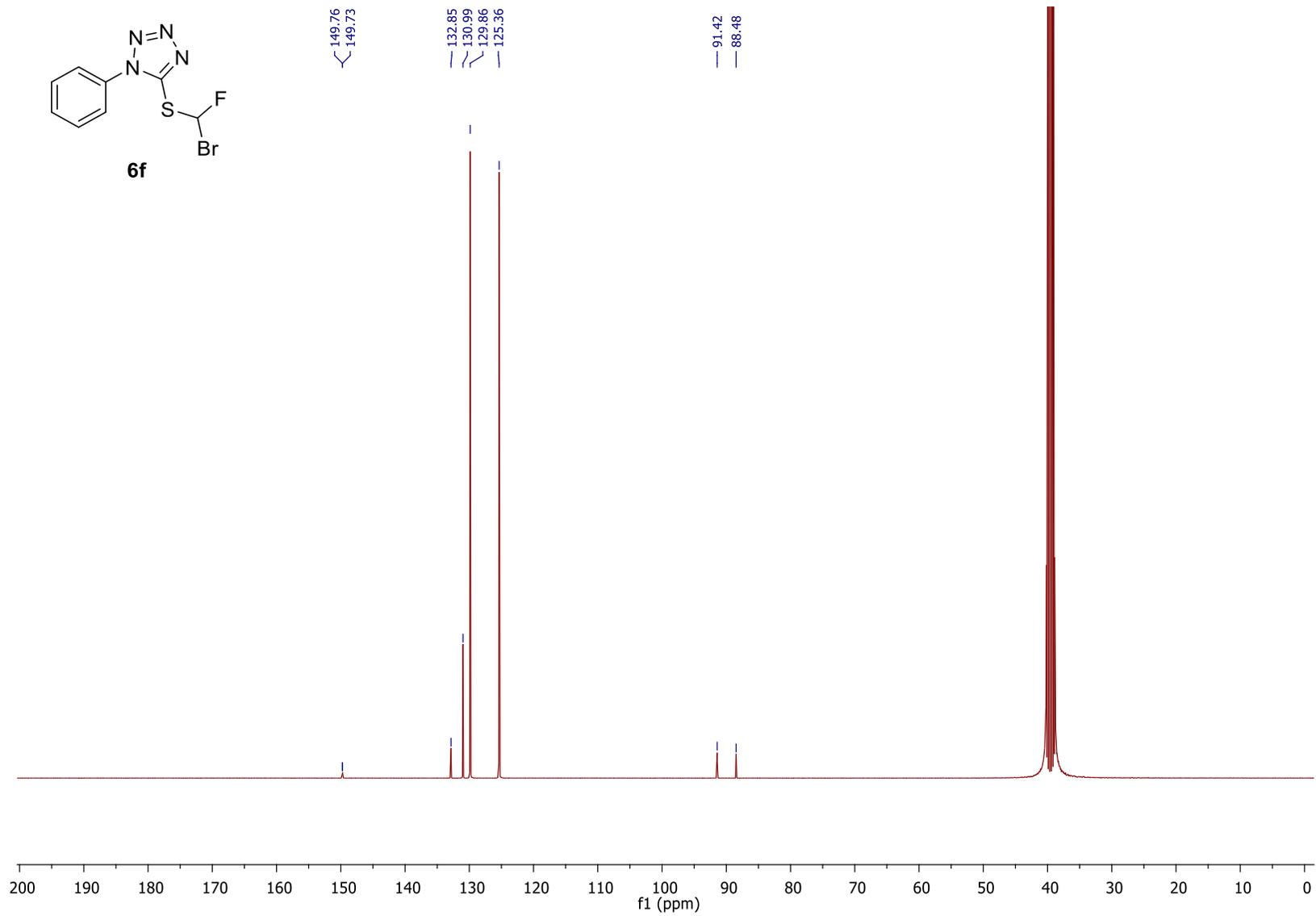
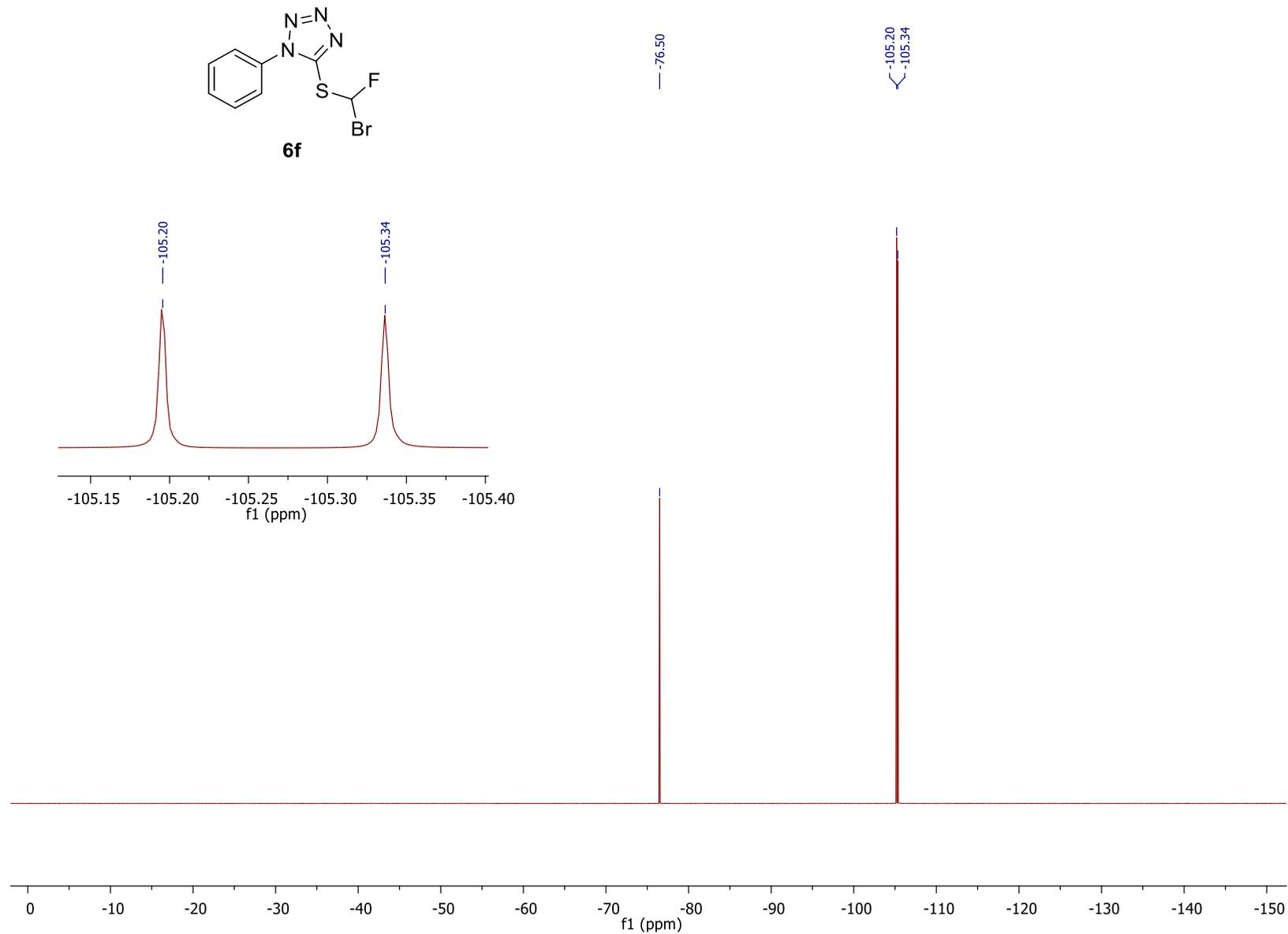


Figure S52. <sup>1</sup>H-NMR spectrum of 5-((bromofluoromethyl)thio)-1-phenyl-1H-tetrazole (6f).



**Figure S53.** <sup>13</sup>C-NMR spectrum of 5-((bromofluoromethyl)thio)-1-phenyl-1H-tetrazole (**6f**).



**Figure S54.** <sup>19</sup>F-NMR spectrum of 5-((bromofluoromethyl)thio)-1-phenyl-1H-tetrazole (**6f**).

## 2. Radiochemistry

**Table S1.** UPLC gradient for the analysis of the crude products [<sup>18</sup>F]4a–[<sup>18</sup>F]4f and [<sup>18</sup>F]5a–[<sup>18</sup>F]5f (gradient A)

Time (min)	HCO <sub>2</sub> H/H <sub>2</sub> O (0.05%, v/v)	MeCN	Flow rate (mL·min <sup>-1</sup> )
0	100	0	0.5
6	25	75	0.5
8	100	0	0.5

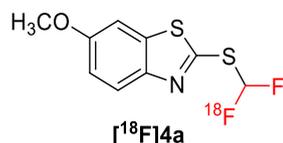
**Table S2.** UPLC gradient for the analysis of the crude products [<sup>18</sup>F]8a–[<sup>18</sup>F]8g (gradient B)

Time (min)	HCO <sub>2</sub> H/H <sub>2</sub> O (0.05%, v/v)	MeCN	Flow rate (mL·min <sup>-1</sup> )
0	100	0	0.5
6	0	100	0.5
8	100	0	0.5

### 2.1. General procedure for the <sup>18</sup>F-labeling of bromofluoromethyl heteroaryl-sulfides 6a-6f

Using the GE FASTlab™ synthesizer, an aliquot of [<sup>18</sup>F]fluoride (150-200 MBq) was trapped on a Sep-Pak® Accell™ Plus QMA Carbonate Plus Light cartridge and eluted with a solution of Kryptofix® 222 (K<sub>2.2.2</sub>; 7.5 mg in 600 μL of MeCN) and K<sub>2</sub>CO<sub>3</sub> (1.4 mg in 150 μL of H<sub>2</sub>O). Upon azeotropic drying, a solution of the precursors **6a** (12.3 mg, 0.04 mmol), **6b** (12.3 mg, 0.04 mmol), **6c** (6.5 mg, 0.02 mmol), **6d** (12.9 mg, 0.04 mmol), **6e** (11.0 mg, 0.04 mmol), or **6f** (11.6 mg, 0.04 mmol) in MeCN (1 mL) was transferred to the dry [<sup>18</sup>F]potassium fluoride/Kryptofix® 222 ([<sup>18</sup>F]KF/K<sub>2.2.2</sub>) complex and heated to 120 °C. After 5 min of <sup>18</sup>F-labeling and dilution of the reaction mixture with H<sub>2</sub>O, the labeled compounds [<sup>18</sup>F]4a–[<sup>18</sup>F]4f were trapped on Sep-Pak® C18 Plus Short cartridge. Subsequently, the C18 cartridge was removed and the trapped crude products [<sup>18</sup>F]4a–[<sup>18</sup>F]4f were recovered to a 4 mL-vial *via* manual elution with MeCN (1 mL).

#### 2.1.1. Synthesis of [<sup>18</sup>F]2-((difluoromethyl)thio)-6-methoxybenzo[d]thiazole ([<sup>18</sup>F]4a)



The implementation of the general procedure for the <sup>18</sup>F-labeling of 2-((bromofluoromethyl)thio)-6-methoxybenzo[d]thiazole (**6a**) (12.3 mg, 0.04 mmol) provided the labeled compound [<sup>18</sup>F]4a in 14.2 ± 0.7% RCY (d.c. at the SOS).

The radiochemical yield (RCY) of the <sup>18</sup>F-labeling step was determined based on the activity of the recovered crude products [<sup>18</sup>F]4a–[<sup>18</sup>F]4f, on their radio-TLC and radio-UPLC purities, and the starting radioactivity, according to the following formula:

$$RCY (\%, d. c.) = \frac{\text{radioTLC purity} (\%) \times \text{radioUPLC purity} (\%) \times \text{activity of the solution of } [^{18}\text{F}]\mathbf{4a} - [^{18}\text{F}]\mathbf{4f} (d. c.)}{\text{starting radioactivity} \times 100}$$

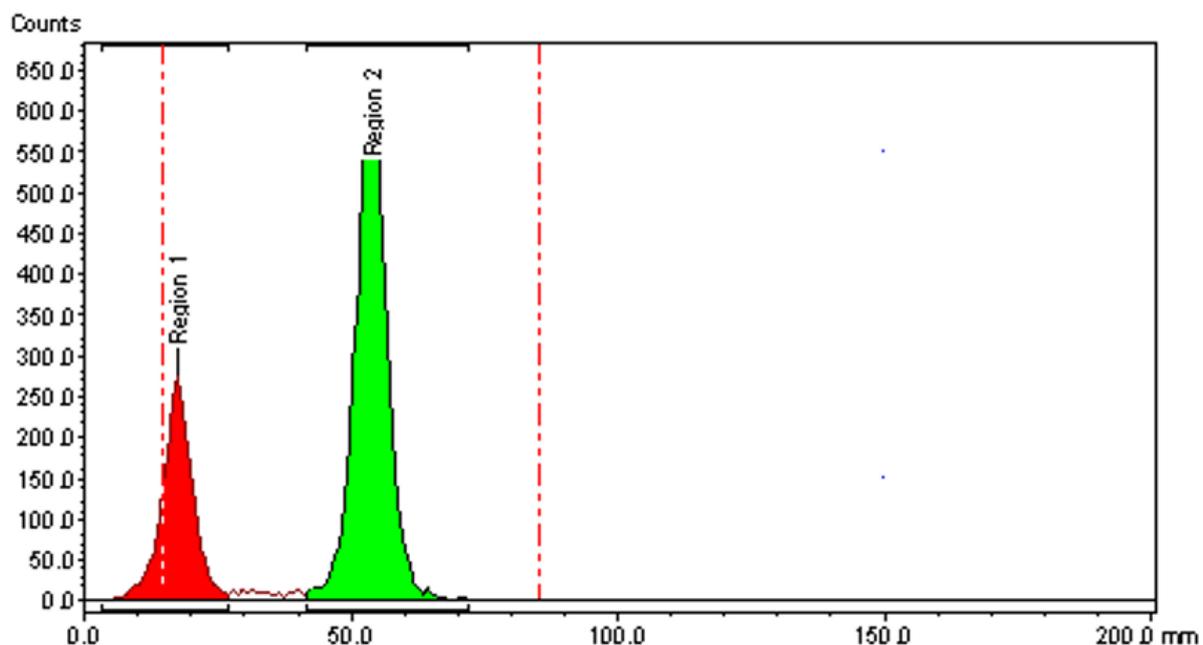


Figure S55. TLC radio-chromatogram of the crude product [ $^{18}\text{F}$ ]4a (eluent: methanol).

Table S3. Determination of the radio-TLC purity of the crude product [ $^{18}\text{F}$ ]4a

Retention factor ( $R_f$ , mm)	Ratio (%)
0.03	29 (impurity/by-product)
0.67	71 (desired crude product)

Table S4 furnishes more details of the RCY determination. The UPLC radio-chromatogram of the crude product [ $^{18}\text{F}$ ]4a is depicted in Figure S56. Figure S57 represents the UPLC UV-chromatogram of the non-radioactive reference 4a.

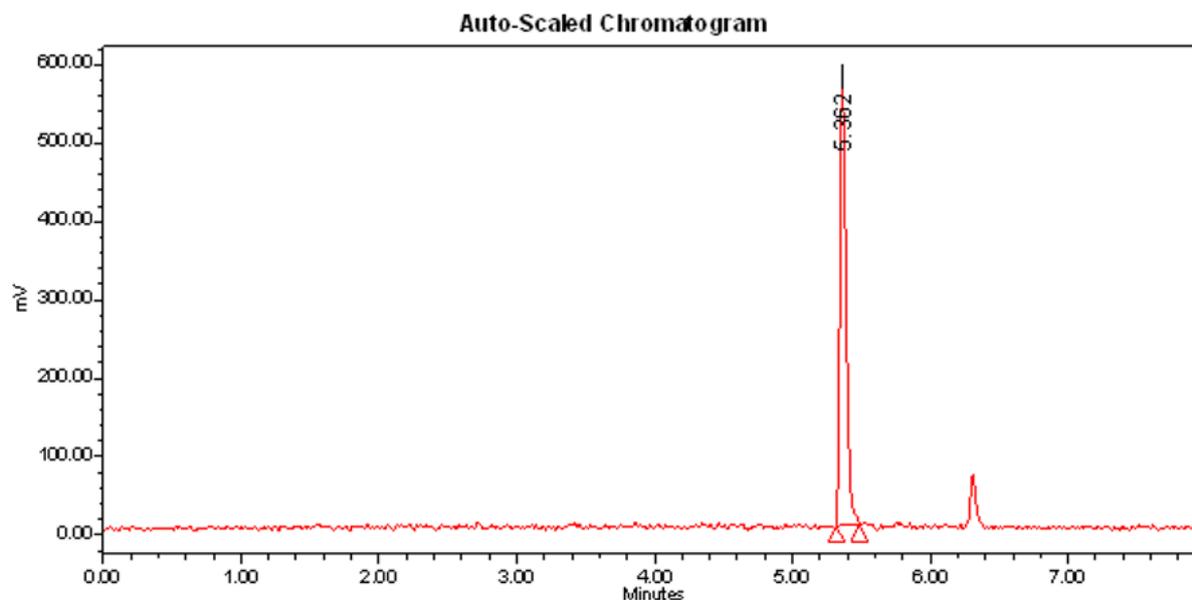


Figure S56. UPLC radio-chromatogram of the crude product [ $^{18}\text{F}$ ]4a (gradient A).

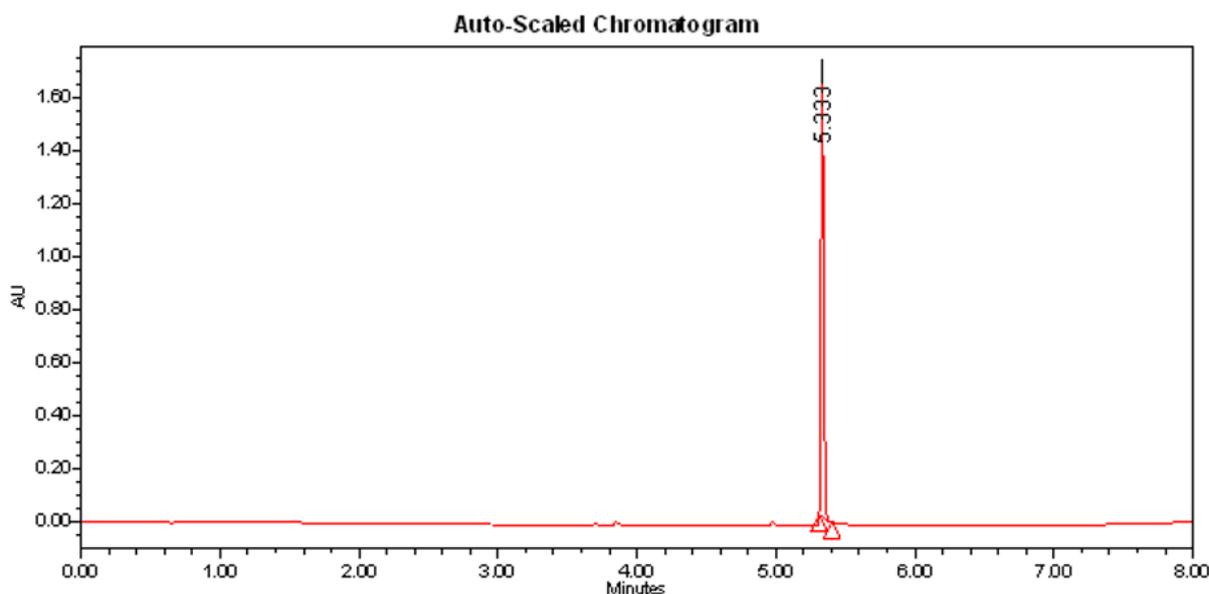


Figure S57. UPLC UV-chromatogram of the authentic reference **4a** (gradient A).

$$RCY (\%, d. c.) = \frac{\text{radioTLC purity (\%)} \times \text{radioUPLC purity (\%)} \times \text{activity of the solution of } [^{18}\text{F}]\mathbf{4a} (d. c.)}{\text{starting radioactivity} \times 100}$$

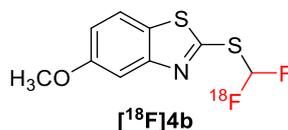
$$RCY (\%, d. c.) = \frac{71 \times 92 \times 37.0}{169.4 \times 100}$$

$$RCY (\%, d. c.) = 14.3 \%$$

Table S4. Determination of the radiochemical yield (%) of the synthesis of  $[^{18}\text{F}]\mathbf{4a}$

Reaction	Starting activity (MBq)	Activity of the crude product $[^{18}\text{F}]\mathbf{4a}$ (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	169.4	37.0	71	92	14.3
2	144.4	28.8	74	90	13.3
3	187.9	50.9	55	100	14.9
<b>Radiochemical Yield (%) ± Deviation</b>					14.2 ± 0.7

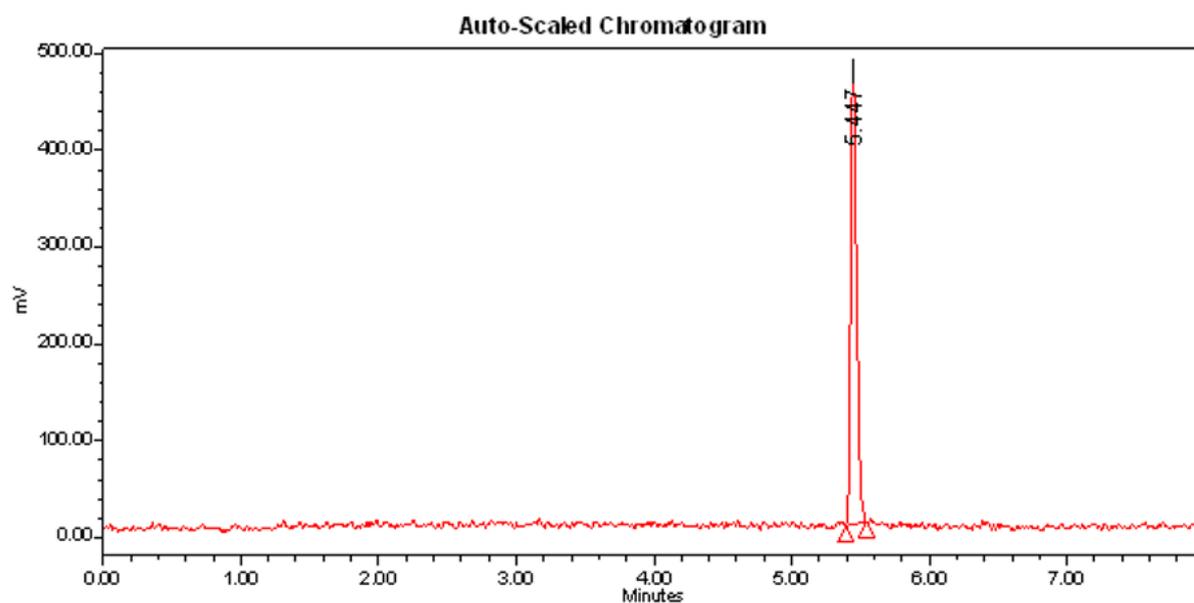
### 2.1.2. Synthesis of $[^{18}\text{F}]\mathbf{2}$ -((difluoromethyl)thio)-5-methoxybenzo[*d*]thiazole ( $[^{18}\text{F}]\mathbf{4b}$ )



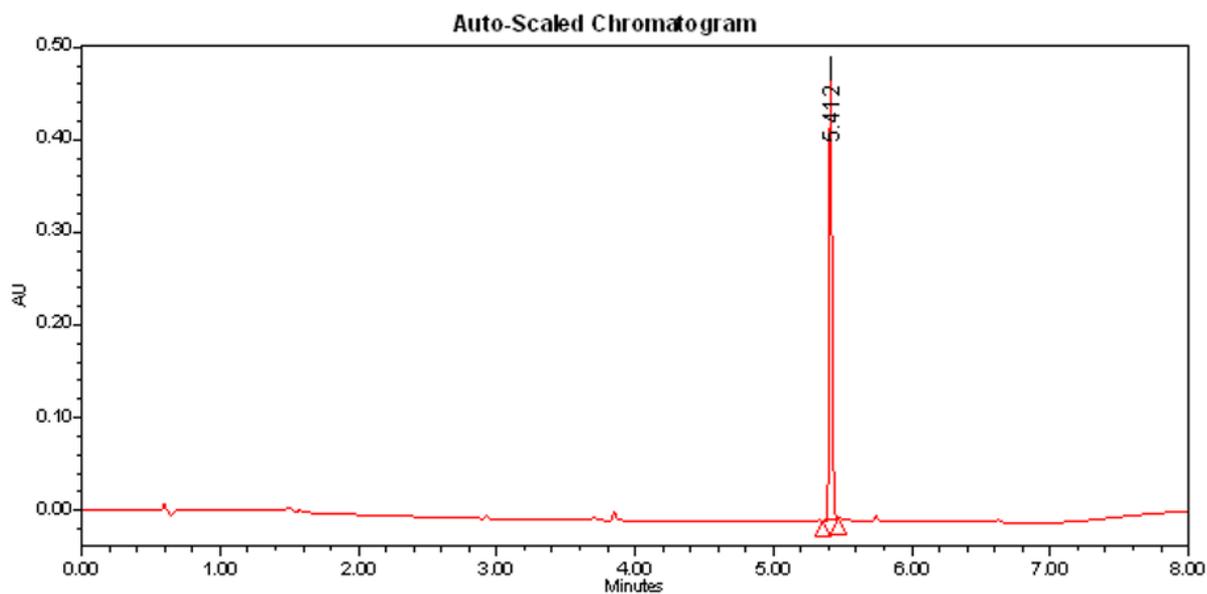
The implementation of the general procedure for the  $^{18}\text{F}$ -labeling of 2-((bromofluoromethyl)thio)-5-methoxybenzo[*d*]thiazole (**6b**) (12.3 mg, 0.04 mmol) provided the labeled compound  $[^{18}\text{F}]\mathbf{4b}$  in  $11.8 \pm 1.9\%$  RCY (d.c. at the SOS). Table S5 furnishes more details of the RCY determination. The UPLC radiochromatogram of the crude product  $[^{18}\text{F}]\mathbf{4b}$  is depicted in Figure S58. Figure S59 represents the UPLC UV-chromatogram of the non-radioactive reference **4b**.

**Table S5.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]4b

Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F]4b (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	187.5	41.5	42	100	9.3
2	142.2	24.2	68	100	11.6
3	173.4	30.5	67	100	11.8
4	167.4	34.2	73	97	14.5
Radiochemical Yield (%) ± Deviation					11.8 ± 1.9

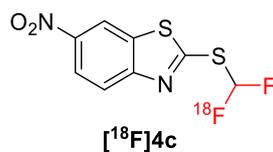


**Figure S58.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]4b (gradient A).



**Figure S59.** UPLC UV-chromatogram of the authentic reference 4b (gradient A).

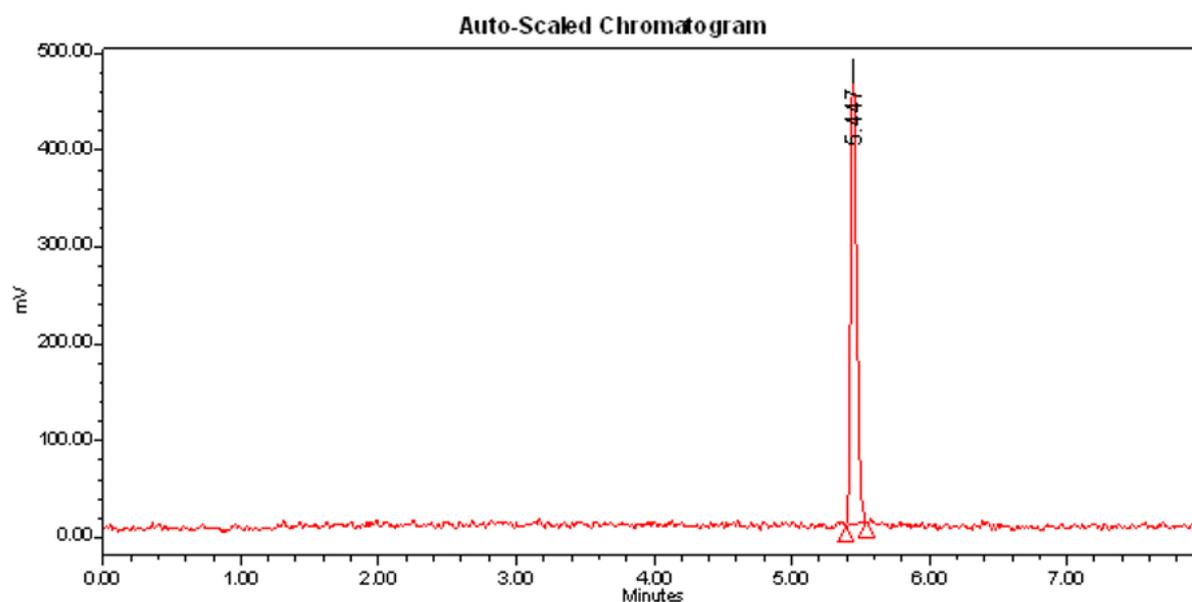
### 2.1.3. Synthesis of [<sup>18</sup>F]2-((difluoromethyl)thio)-6-nitrobenzo[*d*]thiazole ([<sup>18</sup>F]4c)



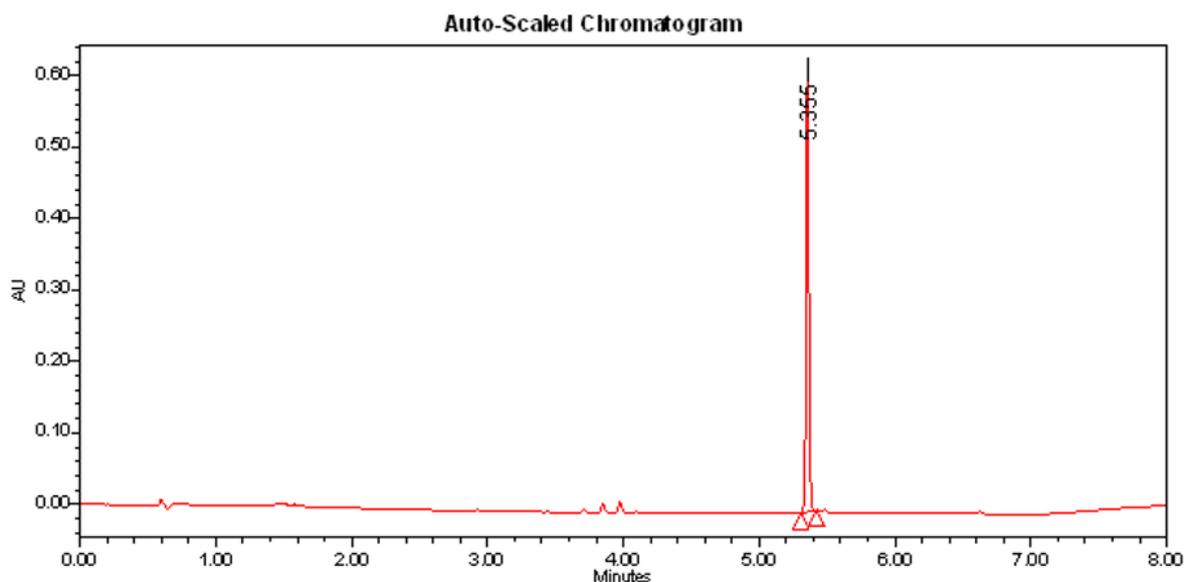
The implementation of the general procedure for the <sup>18</sup>F-labeling of 2-((bromofluoromethyl)thio)-6-nitrobenzo[*d*]thiazole (**6c**) (6.5 mg, 0.02 mmol) provided the labeled compound [<sup>18</sup>F]**4c** in 13.6 ± 0.6% RCY (d.c. at the SOS). Table S6 furnishes more details of the RCY determination. The UPLC radio-chromatogram of the crude product [<sup>18</sup>F]**4c** is depicted in Figure S60. Figure S61 represents the UPLC UV-chromatogram of the non-radioactive reference **4c**.

**Table S6.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**4c**

Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F] <b>4c</b> (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	150.9	32.6	75	82	13.3
2	173.1	36.1	81	86	14.5
3	196.1	37.8	70	97	13.1
<b>Radiochemical Yield (%) ± Deviation</b>					<b>13.6 ± 0.6</b>

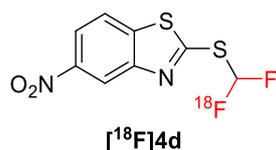


**Figure S60.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]**4c** (gradient A).



**Figure S61.** UPLC UV-chromatogram of the authentic reference **4c** (gradient A).

#### 2.1.4. Synthesis of [<sup>18</sup>F]2-((difluoromethyl)thio)-5-nitrobenzo[*d*]thiazole ([<sup>18</sup>F]**4d**)



The implementation of the general procedure for the <sup>18</sup>F-labeling of 2-((bromofluoromethyl)thio)-5-nitrobenzo[*d*]thiazole (**6d**) (12.9 mg, 0.04 mmol) provided the labeled compound [<sup>18</sup>F]**4d** in 12.7 ± 0.2% RCY (d.c. at the SOS). Table S7 furnishes more details of the RCY determination. The UPLC radiochromatogram of the crude product [<sup>18</sup>F]**4d** is depicted in Figure S62. Figure S63 represents the UPLC UV-chromatogram of the non-radioactive reference **4d**.

**Table S7.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**4d**

Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F] <b>4d</b> (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	196.8	62.1	46	86	12.5
2	161.4	38.7	67	79	12.7
3	180.6	34.9	75	89	12.9
<b>Radiochemical Yield (%) ± Deviation</b>					12.7 ± 0.2

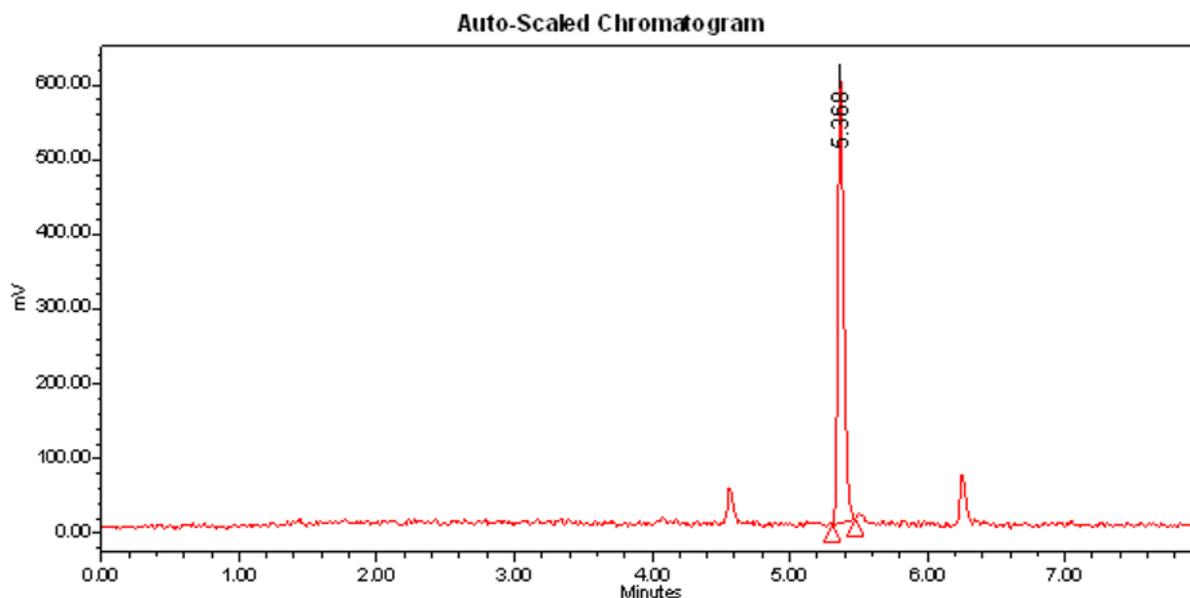


Figure S62. UPLC radio-chromatogram of the crude product  $[^{18}\text{F}]\mathbf{4d}$  (gradient A).

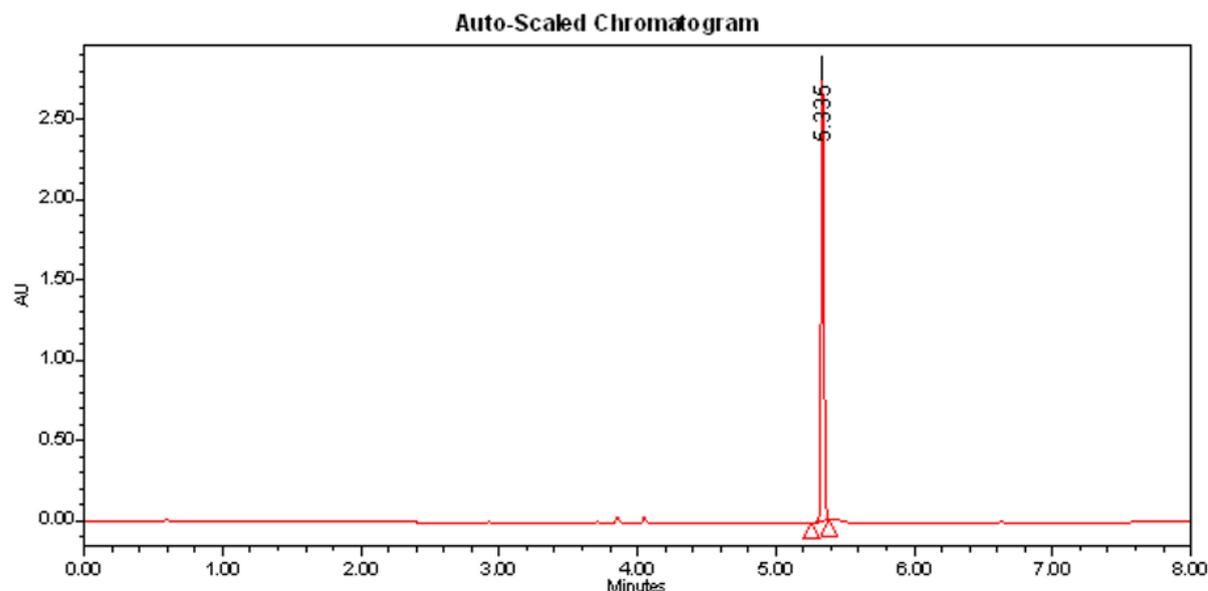
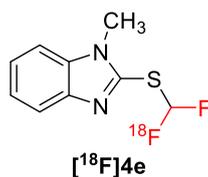


Figure S63. UPLC UV-chromatogram of the authentic reference  $\mathbf{4d}$  (gradient A).

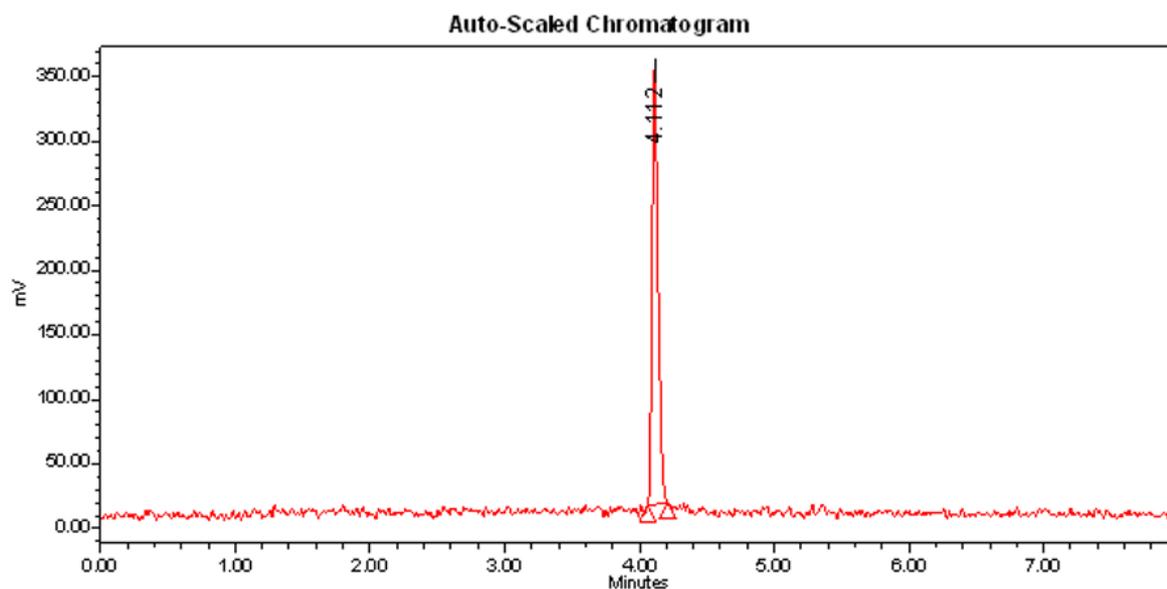
### 2.1.5. Synthesis of $[^{18}\text{F}]2\text{-}((\text{difluoromethyl})\text{thio})\text{-1-methyl-1H-benzo}[d]\text{imidazole}$ ( $[^{18}\text{F}]\mathbf{4e}$ )



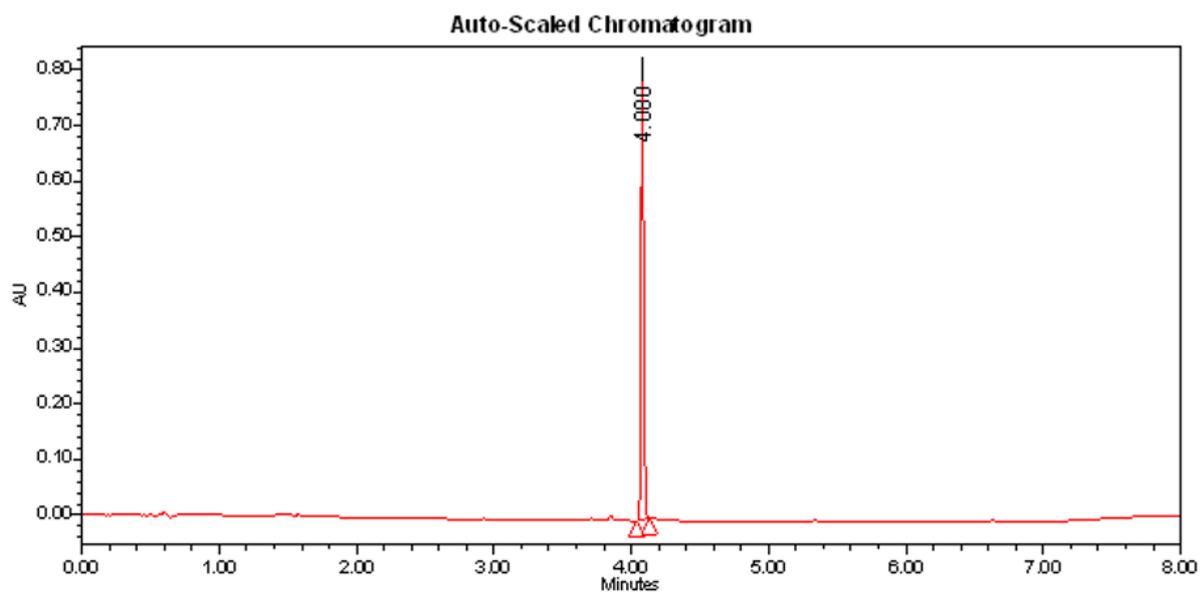
The implementation of the general procedure for the  $^{18}\text{F}$ -labeling of 2-((bromofluoromethyl)thio)-1-methyl-1H-benzo[*d*]imidazole ( $\mathbf{6e}$ ) (11.0 mg, 0.04 mmol) provided the labeled compound  $[^{18}\text{F}]\mathbf{4e}$  in  $8.3 \pm 1.9\%$  RCY (d.c. at the SOS). Table S8 furnishes more details of the RCY determination. The UPLC radio-chromatogram of the crude product  $[^{18}\text{F}]\mathbf{4e}$  is depicted in Figure S64. Figure S65 represents the UPLC UV-chromatogram of the non-radioactive reference  $\mathbf{4e}$ .

**Table S8.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]4e

Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F]4e (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	182.4	33	32	100	5.8
2	154.6	16.8	69	100	7.5
3	169.8	25.7	58	100	8.8
4	187.4	30.5	71	95	11.0
Radiochemical Yield (%) ± Deviation					8.3 ± 1.9

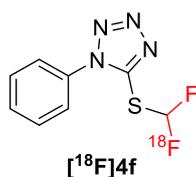


**Figure S64.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]4e (gradient A).



**Figure S65.** UPLC UV-chromatogram of the authentic reference 4e (gradient A).

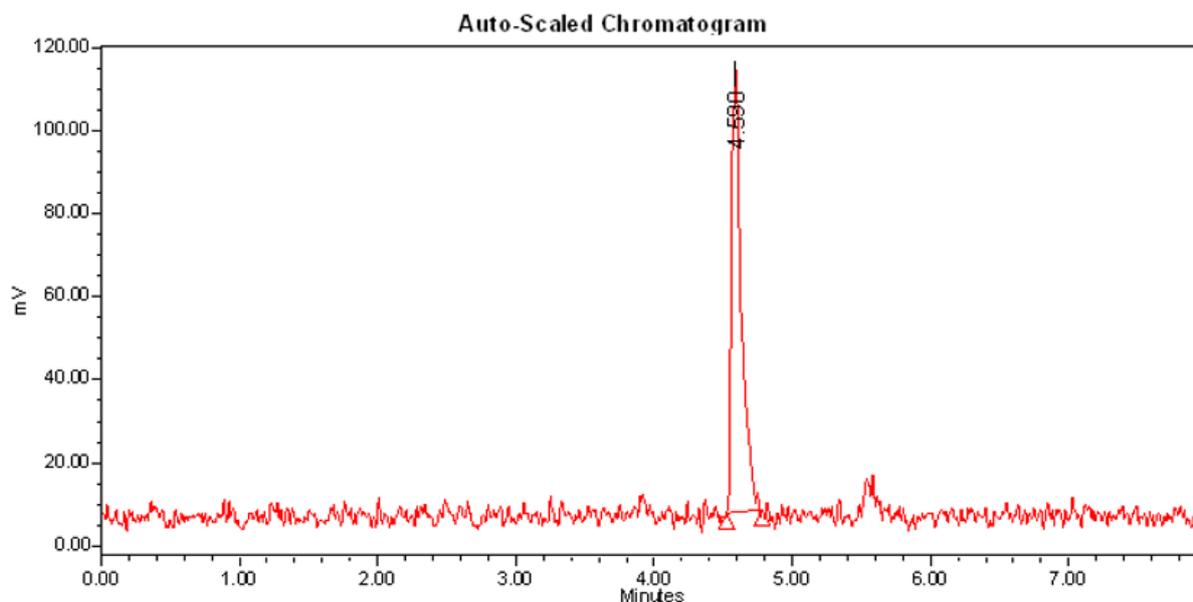
### 2.1.6. Synthesis of [<sup>18</sup>F]5-((difluoromethyl)thio)-1-phenyl-1H-tetrazole ([<sup>18</sup>F]4f)



The implementation of the general procedure for the <sup>18</sup>F-labeling of 5-((bromofluoromethyl)thio)-1-phenyl-1H-tetrazole (**6f**) (11.6 mg, 0.04 mmol) provided the labeled compound [<sup>18</sup>F]**4f** in 14.8 ± 0.8% RCY (d.c. at the SOS). Table S9 furnishes more details of the RCY determination. The UPLC radio-chromatogram of the crude product [<sup>18</sup>F]**4f** is depicted in Figure S66. Figure S67 represents the UPLC UV-chromatogram of the non-radioactive reference **4f**.

**Table S9.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**4f**

Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F] <b>4f</b> (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	130.2	26.8	81	96	16.0
2	190.4	38.4	74	95	14.2
3	195.8	39.3	75	95	14.3
<b>Radiochemical Yield (%) ± Deviation</b>					<b>14.8 ± 0.8</b>



**Figure S66.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]**4f** (gradient A).

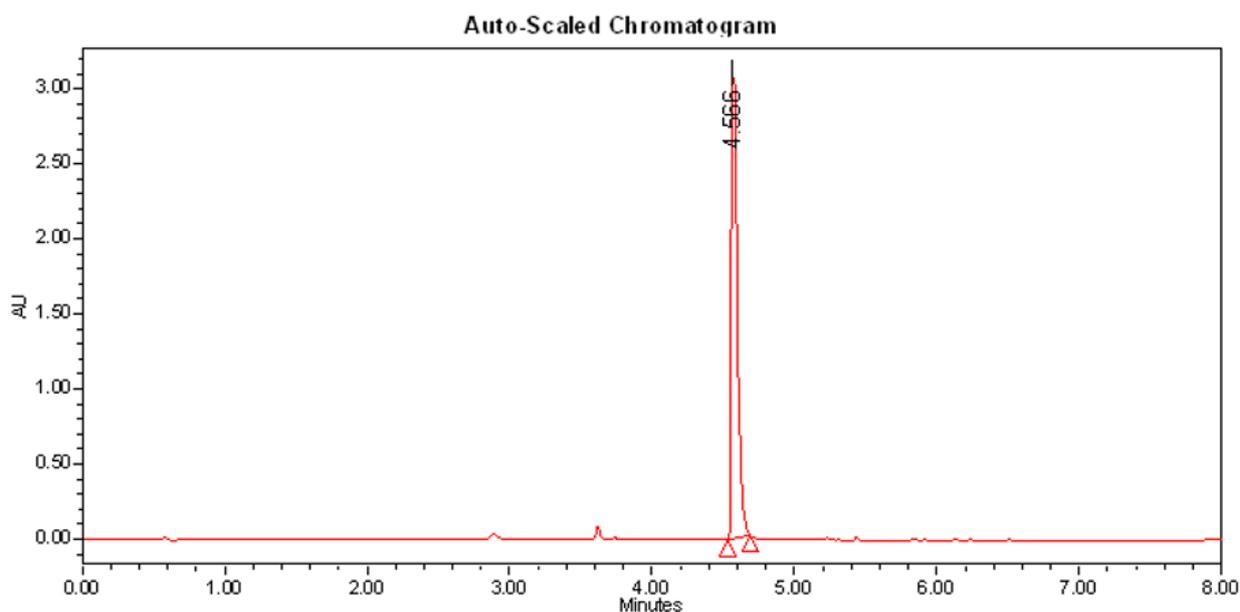
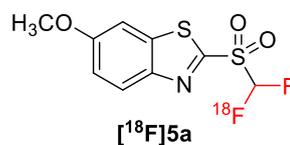


Figure S67. UPLC UV-chromatogram of the authentic reference **4f** (gradient A).

## 2.2. Optimization of the conditions for the oxidation of the [<sup>18</sup>F]difluoromethyl heteroaryl-sulfide [<sup>18</sup>F]**4a**

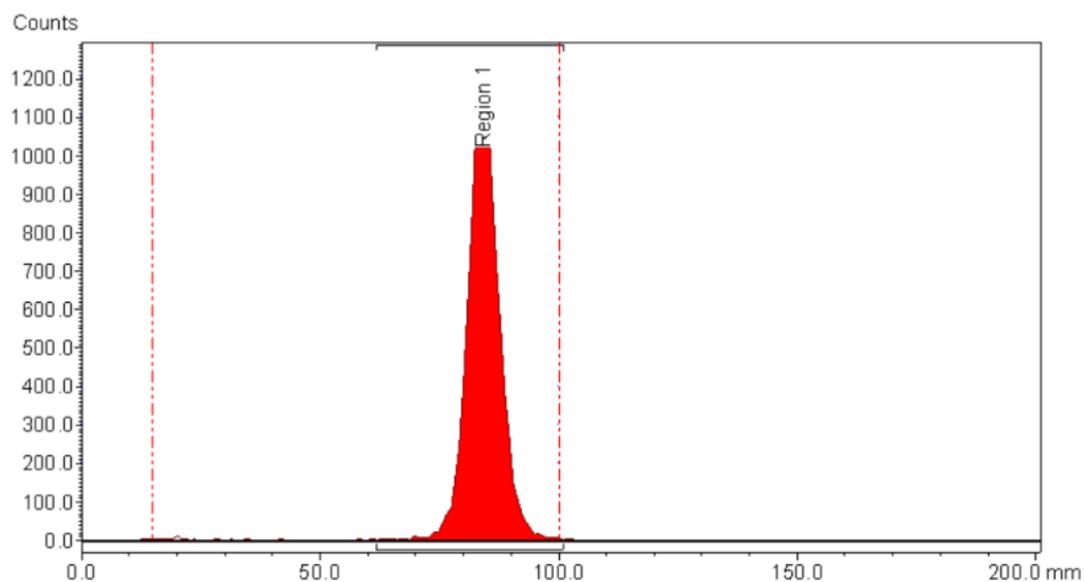
### 2.2.1. Synthesis of [<sup>18</sup>F]2-((difluoromethyl)sulfonyl)-6-methoxybenzo[*d*]thiazole ([<sup>18</sup>F]**5a**)



A solution containing NaIO<sub>4</sub> and RuCl<sub>3</sub>·*x*H<sub>2</sub>O in H<sub>2</sub>O (1 mL) was transferred to the 'C18 cartridge and the oxidation of the trapped crude products [<sup>18</sup>F]**4a** (10-20 MBq) was carried out in solid-phase for 5 min at room temperature. Afterwards, the corresponding [<sup>18</sup>F]difluoromethyl heteroaryl-sulfone [<sup>18</sup>F]**5a** was manually eluted from the 'C18 cartridge with MeCN (1 mL) to a 4 mL-vial.

The RCY of the oxidation step was determined based on the activity of the crude products [<sup>18</sup>F]**4a** and [<sup>18</sup>F]**5a**, and on their radio-TLC and radio-UPLC purities, according to the following equation:

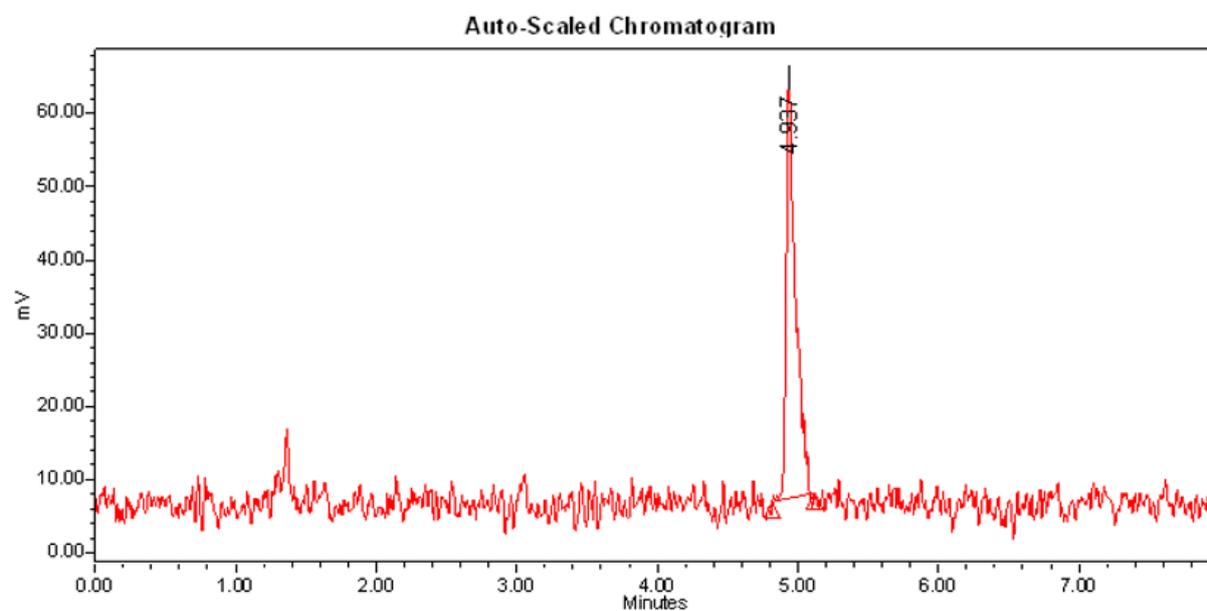
$$RCY (\%, d. c.) = \frac{\text{radioTLC purity (\%)} \times \text{radioUPLC purity (\%)} \times \text{activity of the solution of } [^{18}\text{F}]\mathbf{5a} (d. c.)}{\text{activity of the solution of } [^{18}\text{F}]\mathbf{4a} \times 100}$$



**Figure S68.** TLC radio-chromatogram of the crude product [ $^{18}\text{F}$ ]5a (eluent: methanol).

**Table S10.** Determination of the radio-TLC purity of the crude product [ $^{18}\text{F}$ ]5a

Retention factor ( $R_f$ , mm)	Ratio (%)
0	0 (impurity/by-product)
0.78	100 (desired crude product)



**Figure S69.** UPLC radio-chromatogram of the crude product [ $^{18}\text{F}$ ]5a (gradient A).

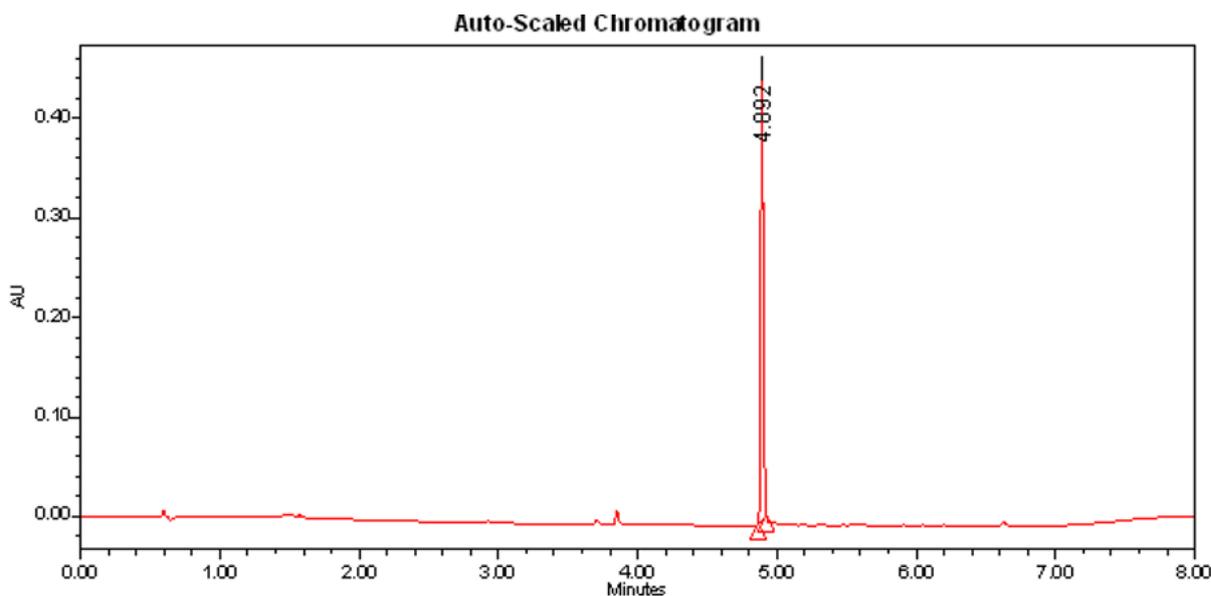


Figure S70. UPLC UV-chromatogram of the authentic reference **5a** (gradient A).

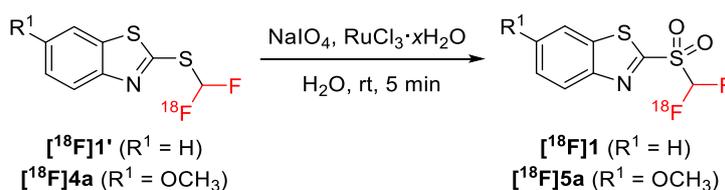
$$RCY (\%, d. c.) = \frac{\text{radioTLC purity (\%)} \times \text{radioUPLC purity (\%)} \times \text{activity of the solution of } [^{18}\text{F}]\mathbf{5a} (d. c.)}{\text{activity of the solution of } [^{18}\text{F}]\mathbf{4a} (d. c.) \times 100}$$

$$RCY (\%, d. c.) = \frac{100 \times 92 \times 12.0}{15.6 \times 100}$$

$$RCY (\%, d. c.) = 70.6 \%$$

According to the reported conditions for the oxidation of the [ $^{18}\text{F}$ ]difluoromethyl heteroaryl-sulfide [ $^{18}\text{F}$ ]**1'**, the following table summarizes the results of the different optimization tests conducted in the [ $^{18}\text{F}$ ]**4a**.

**Standard reaction conditions for the oxidation of [ $^{18}\text{F}$ ]**4a**:** [ $^{18}\text{F}$ ]difluoromethyl heteroaryl-sulfide [ $^{18}\text{F}$ ]**4a** (10-20 MBq),  $\text{NaIO}_4$  (mmol),  $\text{RuCl}_3 \cdot x\text{H}_2\text{O}$  (mmol),  $\text{H}_2\text{O}$  (1 mL), rt, 5 min.



Entry	Substrate	$\text{NaIO}_4$ (mmol)	$\text{RuCl}_3 \cdot x\text{H}_2\text{O}$ (mmol)	RCY (%) <sup>(a)</sup>
1	$[^{18}\text{F}]\mathbf{1}'$	0.24	0.008	$82.9 \pm 7.9$ ( $n=3$ ) <sup>(b)</sup>
2	$[^{18}\text{F}]\mathbf{4a}$	0.24	0.008	32.1
3	$[^{18}\text{F}]\mathbf{4a}$	0.24	0.016	59.6
4	$[^{18}\text{F}]\mathbf{4a}$	0.72	0.016	$70.9 \pm 6.1$ ( $n=3$ ) <sup>(b)</sup>

<sup>(a)</sup> All RCYs were decay-corrected at the SOS. <sup>(b)</sup> Full consumption of the substrates.

The best conditions for the oxidation of the substrate [ $^{18}\text{F}$ ]**4a** were: [ $^{18}\text{F}$ ]difluoromethyl heteroaryl-sulfide [ $^{18}\text{F}$ ]**4a** (10-20 MBq),  $\text{NaIO}_4$  (0.72 mmol),  $\text{RuCl}_3 \cdot x\text{H}_2\text{O}$  (0.016 mmol),  $\text{H}_2\text{O}$  (1 mL), rt, 5 min.

The implementation of the optimized procedure for the oxidation of [ $^{18}\text{F}$ ]2-((difluoromethyl)thio)-5-methoxybenzo[*d*]thiazole ( $[^{18}\text{F}]\mathbf{4a}$ ) (10-20 MBq) provided the labeled compound [ $^{18}\text{F}$ ]**5a** in  $70.9 \pm 6.1\%$  RCY (d.c. at the SOS). Table S11 furnishes more details of the RCY determination. The UPLC radio-

chromatogram of the crude product [<sup>18</sup>F]5a is depicted in Figure S69. Figure S70 represents the UPLC UV-chromatogram of the non-radioactive reference 5a.

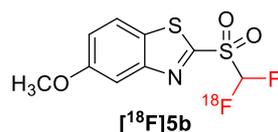
**Table S11.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5a

Reaction	Activity of the crude product [ <sup>18</sup> F]4a (MBq)	Activity of the crude product [ <sup>18</sup> F]5a (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	15.6	12.0	100	92	70.6
2	12.7	10.3	99	98	78.5
3	13.6	11.5	100	75	63.6
Radiochemical Yield (%) ± Deviation					70.9 ± 6.1

### 2.3. General procedure for oxidation of the [<sup>18</sup>F]difluoromethyl heteroaryl-sulfides [<sup>18</sup>F]4a-[<sup>18</sup>F]4f

A solution containing NaIO<sub>4</sub> (153.9 mg, 0.072 mmol) and RuCl<sub>3</sub>·xH<sub>2</sub>O (3.4 mg, 0.016 mmol) in H<sub>2</sub>O (1 mL) was transferred to the <sup>1</sup>C18 cartridge and the oxidation of the trapped crude products [<sup>18</sup>F]4a-[<sup>18</sup>F]4f (10-20 MBq) was carried out in solid-phase for 5 min at room temperature. Afterwards, the corresponding [<sup>18</sup>F]difluoromethyl heteroaryl-sulfones [<sup>18</sup>F]5a-[<sup>18</sup>F]5f were manually eluted from the <sup>1</sup>C18 cartridge with MeCN (1 mL) to a 4 mL-vial.

#### 2.3.1. Synthesis of [<sup>18</sup>F]2-((difluoromethyl)sulfonyl)-5-methoxybenzo[*d*]thiazole ([<sup>18</sup>F]5b)



The implementation of the general procedure for the oxidation of [<sup>18</sup>F]2-((difluoromethyl)thio)-5-methoxybenzo[*d*]thiazole ([<sup>18</sup>F]4b) (10-20 MBq) provided the labeled compound [<sup>18</sup>F]5b in 70.6 ± 5.1% RCY (d.c. at the SOS). Table S12 furnishes more details of the RCY determination. The UPLC radio-chromatogram of the crude product [<sup>18</sup>F]5b is depicted in Figure S71. Figure S72 represents the UPLC UV-chromatogram of the non-radioactive reference 5b.

**Table S12.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5b

Reaction	Activity of the crude product [ <sup>18</sup> F]4b (MBq)	Activity of the crude product [ <sup>18</sup> F]5b (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	17.4	13.1	99	85	63.5
2	16.2	13.7	99	87	72.9
3	14.9	11.2	100	100	75.4
Radiochemical Yield (%) ± Deviation					70.6 ± 5.1

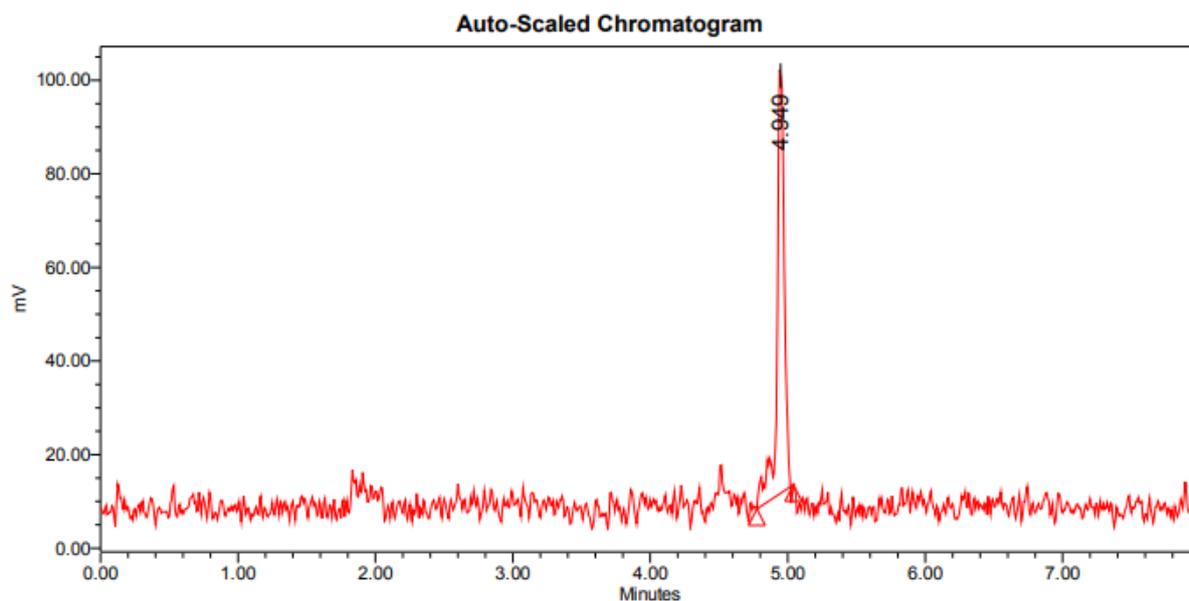


Figure S71. UPLC radio-chromatogram of the crude product  $[^{18}\text{F}]\mathbf{5b}$  (gradient A).

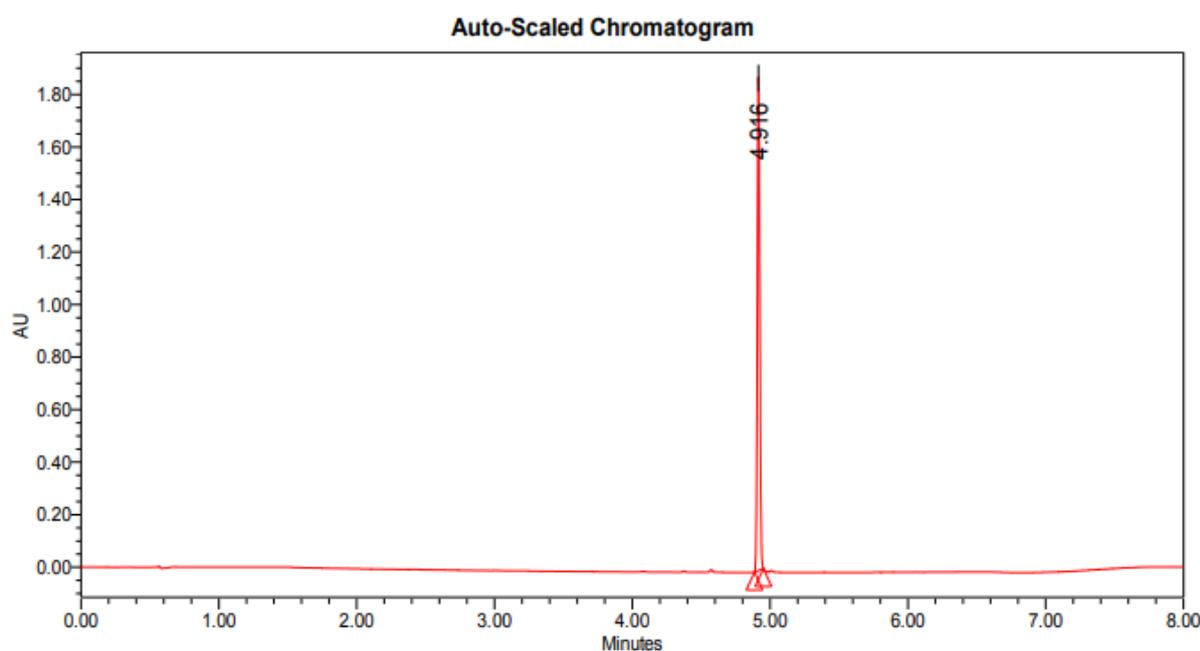
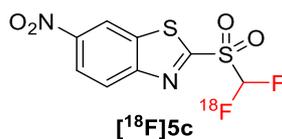


Figure S72. UPLC UV-chromatogram of the authentic reference  $\mathbf{5b}$  (gradient A).

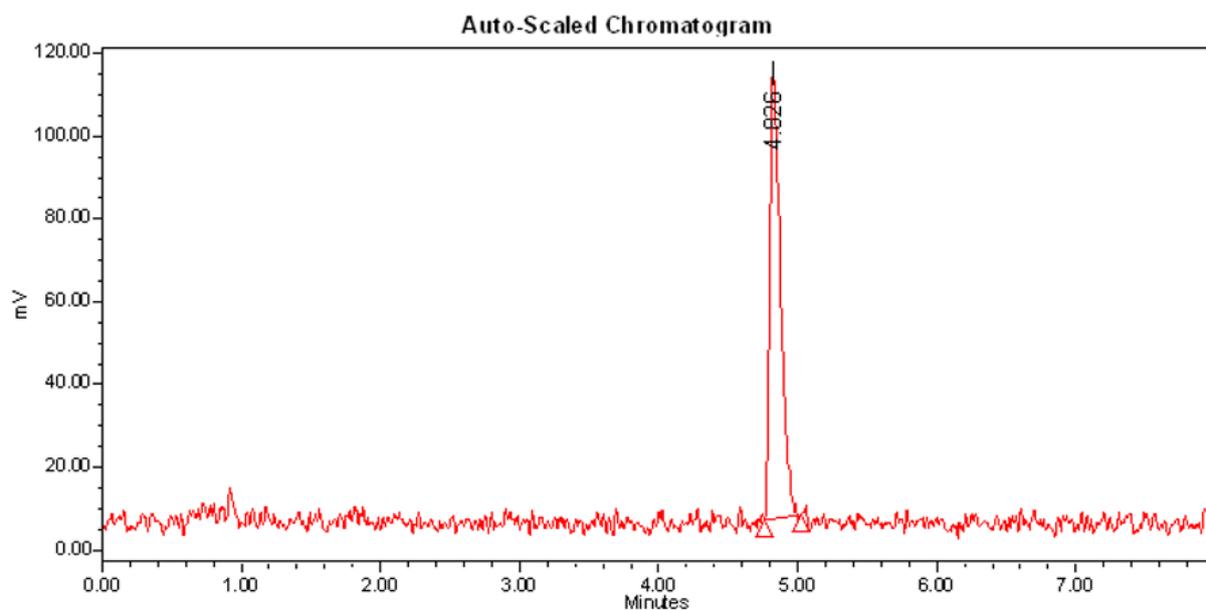
### 2.3.2. Synthesis of $[^{18}\text{F}]\mathbf{2}$ -((difluoromethyl)sulfonyl)-6-nitrobenzo[*d*]thiazole ( $[^{18}\text{F}]\mathbf{5c}$ )



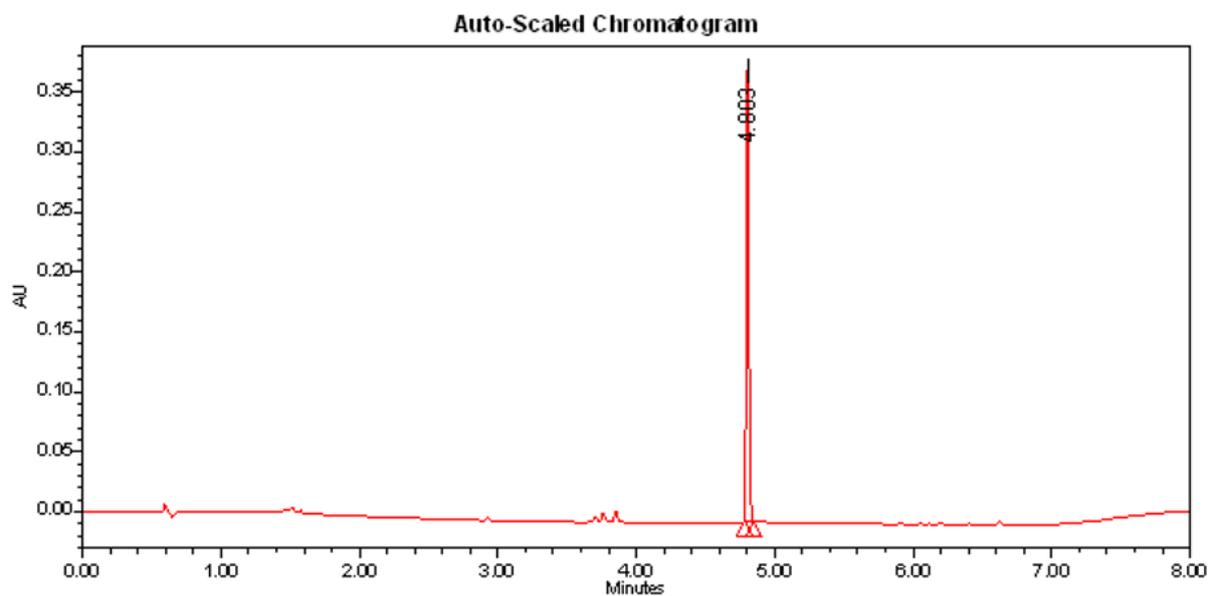
The implementation of the general procedure for the oxidation of  $[^{18}\text{F}]\mathbf{2}$ -((difluoromethyl)thio)-6-nitrobenzo[*d*]thiazole ( $[^{18}\text{F}]\mathbf{4c}$ ) (10-20 MBq) provided the labeled compound  $[^{18}\text{F}]\mathbf{5c}$  in  $88.2 \pm 0.2\%$  RCY (d.c. at the SOS). Table S13 furnishes more details of the RCY determination. The UPLC radio-chromatogram of the crude product  $[^{18}\text{F}]\mathbf{5c}$  is depicted in Figure S73. Figure S74 represents the UPLC UV-chromatogram of the non-radioactive reference  $\mathbf{5c}$ .

**Table S13.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5c

Reaction	Activity of the crude product [ <sup>18</sup> F]4c (MBq)	Activity of the crude product [ <sup>18</sup> F]5c (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	15.9	14.1	100	100	88.4
2	14.2	12.6	100	99	88
3	14.8	13.4	100	97	88.1
<b>Radiochemical Yield (%) ± Deviation</b>					<b>88.2 ± 0.2</b>

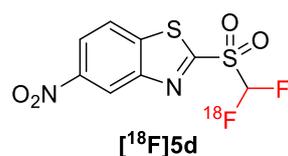


**Figure S73.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]5c (gradient A).



**Figure S74.** UPLC UV-chromatogram of the authentic reference 5c (gradient A).

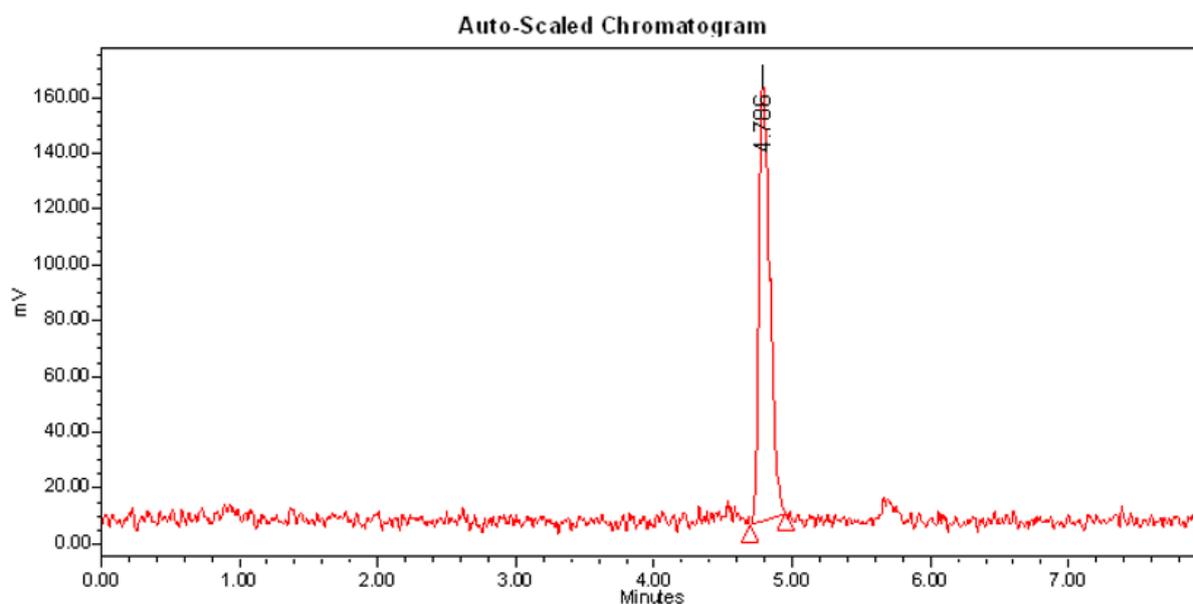
### 2.3.3. Synthesis of [<sup>18</sup>F]2-((difluoromethyl)sulfonyl)-5-nitrobenzo[*d*]thiazole ([<sup>18</sup>F]5d)



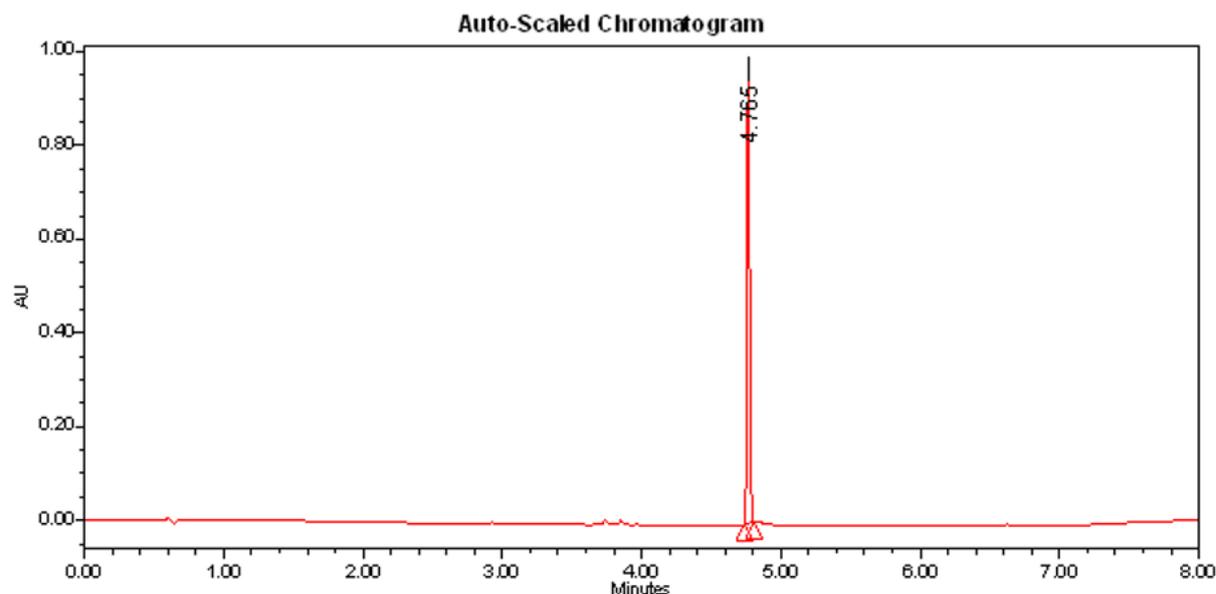
The implementation of the general procedure for the oxidation of [<sup>18</sup>F]2-((difluoromethyl)thio)-5-nitrobenzo[*d*]thiazole ([<sup>18</sup>F]4d) (10-20 MBq) provided the labeled compound [<sup>18</sup>F]5d in 88.4 ± 2.8% RCY (d.c. at the SOS). Table S14 furnishes more details of the RCY determination. The UPLC radio-chromatogram of the crude product [<sup>18</sup>F]5d is depicted in Figure S75. Figure S76 represents the UPLC UV-chromatogram of the non-radioactive reference 5d.

**Table S14.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5d

Reaction	Activity of the crude product [ <sup>18</sup> F]4d (MBq)	Activity of the crude product [ <sup>18</sup> F]5d (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	13.9	13.3	99	97	92.1
2	15.0	14.6	99	89	85.6
3	15.9	15.1	99	93	87.6
<b>Radiochemical Yield (%) ± Deviation</b>					<b>88.4 ± 2.8</b>

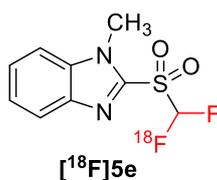


**Figure S75.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]5d (gradient A).



**Figure S76.** UPLC UV-chromatogram of the authentic reference **5d** (gradient A).

### 2.3.4. Synthesis of [<sup>18</sup>F]2-((difluoromethyl)sulfonyl)-1-methyl-1*H*-benzo[*d*]imidazole ([<sup>18</sup>F]**5e**)



The implementation of the general procedure for the oxidation of [<sup>18</sup>F]2-((difluoromethyl)thio)-1-methyl-1*H*-benzo[*d*]imidazole (**[<sup>18</sup>F]4e**) (10-20 MBq) provided the labeled compound [<sup>18</sup>F]**5e** in 86.1 ± 3.0% RCY (d.c. at the SOS). Table S15 furnishes more details of the RCY determination. The UPLC radio-chromatogram of the crude product [<sup>18</sup>F]**5e** is depicted in Figure S77. Figure S78 represents the UPLC UV-chromatogram of the non-radioactive reference **5e**.

**Table S15.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**5e**

Reaction	Activity of the crude product [ <sup>18</sup> F] <b>4e</b> (MBq)	Activity of the crude product [ <sup>18</sup> F] <b>5e</b> (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	13.7	12.4	100	100	90.2
2	17.4	16.0	93	97	83.1
3	14.7	13.2	100	95	85
<b>Radiochemical Yield (%) ± Deviation</b>					<b>86.1 ± 3.0</b>

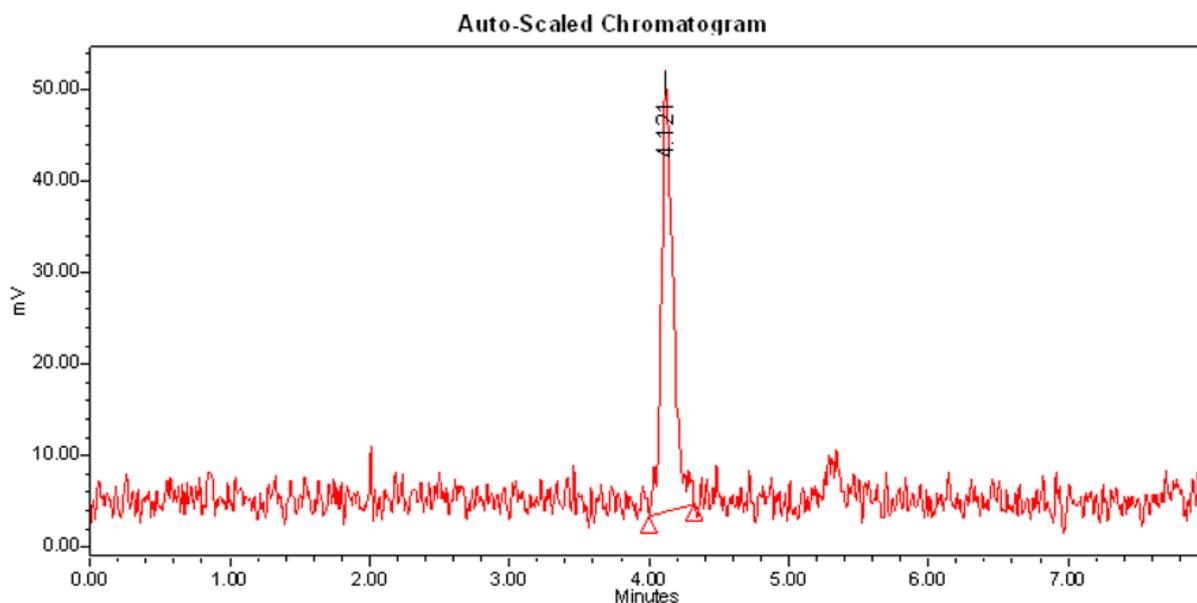


Figure S77. UPLC radio-chromatogram of the crude product  $[^{18}\text{F}]\mathbf{5e}$  (gradient A).

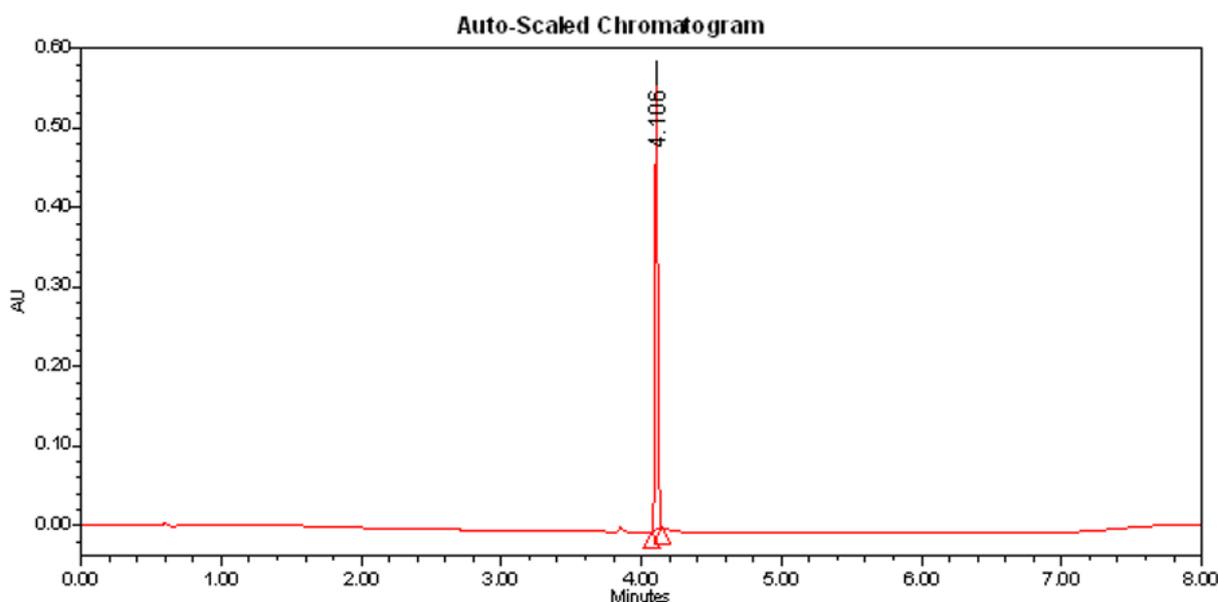
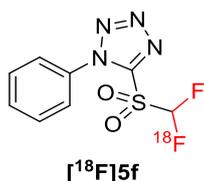


Figure S78. UPLC UV-chromatogram of the authentic reference  $\mathbf{5e}$  (gradient A).

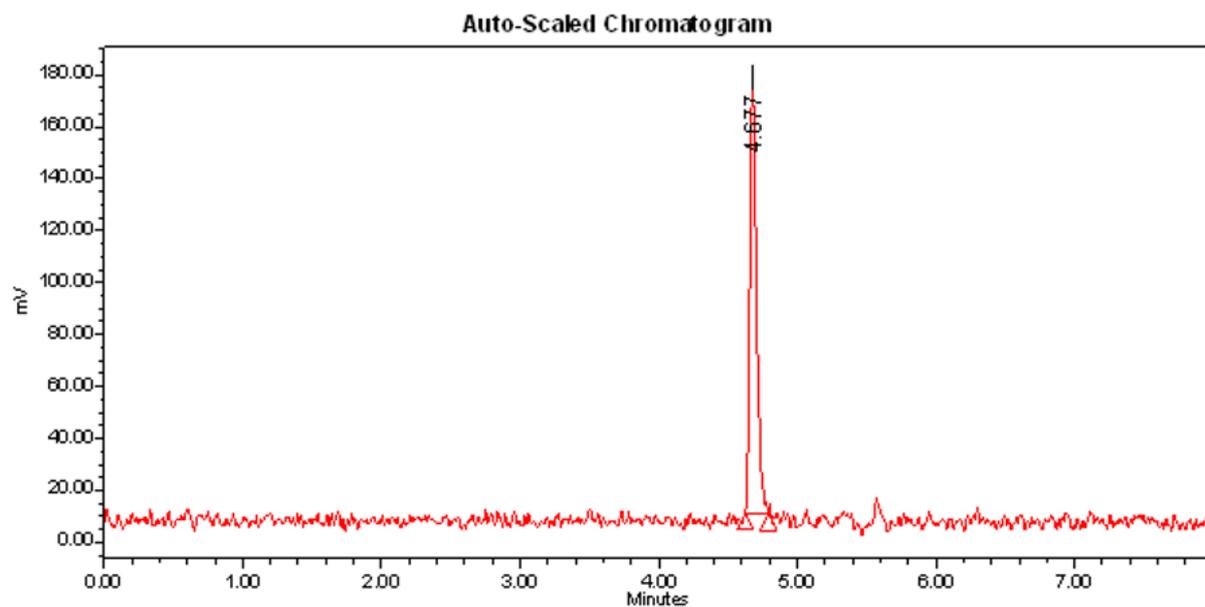
### 2.3.5. Synthesis of $[^{18}\text{F}]\mathbf{5}$ -((difluoromethyl)sulfonyl)-1-phenyl-1H-tetrazole ( $[^{18}\text{F}]\mathbf{5f}$ )



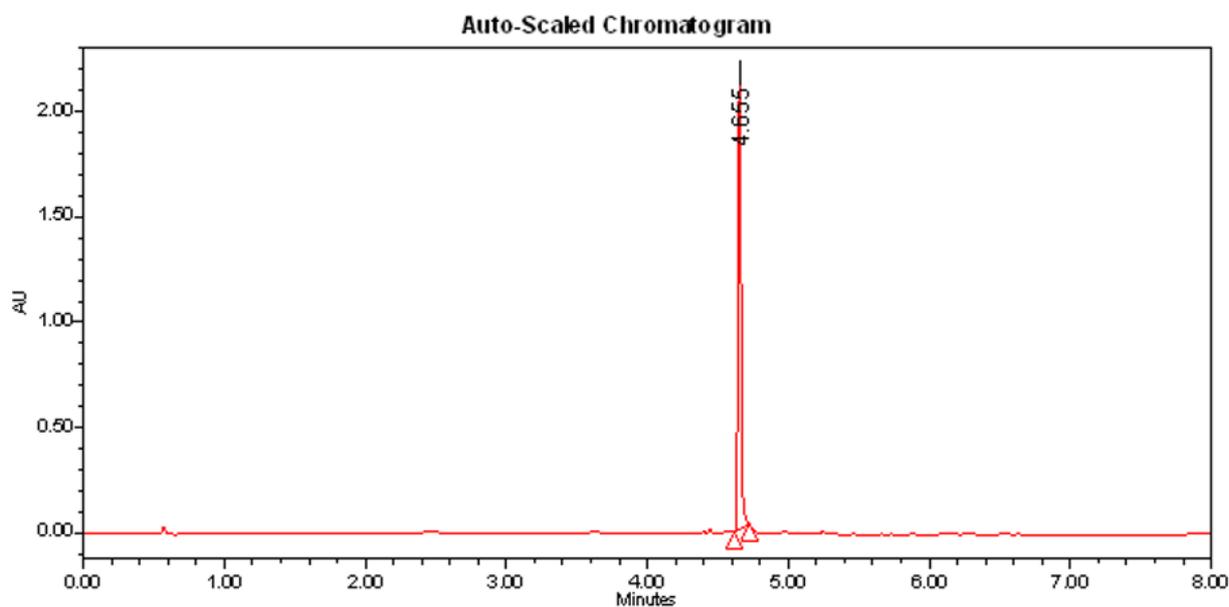
The implementation of the general procedure for the oxidation of  $[^{18}\text{F}]\mathbf{5}$ -((difluoromethyl)thio)-1-phenyl-1H-tetrazole ( $[^{18}\text{F}]\mathbf{4f}$ ) (10-20 MBq) provided the labeled compound  $[^{18}\text{F}]\mathbf{5f}$  in  $91.9 \pm 2.8\%$  RCY (d.c. at the SOS). Table S16 furnishes more details of the RCY determination. The UPLC radio-chromatogram of the crude product  $[^{18}\text{F}]\mathbf{5f}$  is depicted in Figure S79. Figure S80 represents the UPLC UV-chromatogram of the non-radioactive reference  $\mathbf{5f}$ .

**Table S16.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5f

Reaction	Activity of the crude product [ <sup>18</sup> F]4f (MBq)	Activity of the crude product [ <sup>18</sup> F]5f (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	15.4	13.7	100	100	88.7
2	16.1	14.7	100	100	91.4
3	15.7	15.0	100	100	95.6
<b>Radiochemical Yield (%) ± Deviation</b>					<b>91.9 ± 2.8</b>

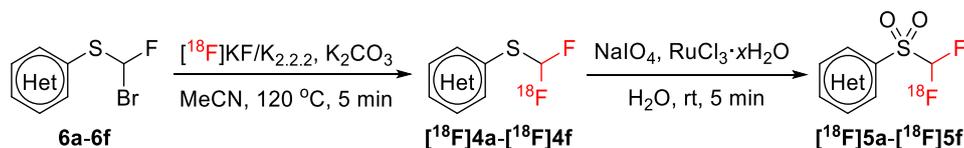


**Figure S79.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]5f (gradient A).



**Figure S80.** UPLC UV-chromatogram of the authentic reference 5f (gradient A).

## 2.4. Two-step radiosyntheses of the [<sup>18</sup>F]difluoromethyl heteroaryl-sulfones [<sup>18</sup>F]5a-<sup>18</sup>F]5f from the precursors 6a-6f



The overall RCY of the <sup>18</sup>F-labeling step of 6a-6f and the oxidation of [<sup>18</sup>F]4a-<sup>18</sup>F]4f was determined based on the activity of the recovered crude products [<sup>18</sup>F]5a-<sup>18</sup>F]5f, on their radio-TLC and radio-UPLC purities, and the starting radioactivity, according to the following formula:

$$RCY (\%, d. c.) = \frac{\text{radioTLC purity (\%)} \times \text{radioUPLC purity (\%)} \times \text{activity of the solution of } [^{18}\text{F}]\text{5a-} [^{18}\text{F}]\text{5f (d. c.)}}{\text{starting radioactivity} \times 100}$$

**Table S17.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5a from the precursor 6a

Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F]5a (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	100	11.0	98	93	10.0
2	98.6	11.8	98	95	11.1
3	131.8	13.6	99	89	9.1
<b>Radiochemical Yield (%) ± Deviation</b>					<b>10.1 ± 0.8</b>

**Table S18.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5b from the precursor 6b

Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F]5b (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	92.2	9.3	98	87	8.6
2	112.0	10.1	98	85	7.5
3	110.0	11.5	99	86	8.9
<b>Radiochemical Yield (%) ± Deviation</b>					<b>8.3 ± 0.6</b>

**Table S19.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5c from the precursor 6c

Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F]5c (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	112.8	14.6	100	99	12.8
2	101.3	12.1	100	98	11.7
3	122.9	14.5	100	98	11.6
<b>Radiochemical Yield (%) ± Deviation</b>					<b>12 ± 0.5</b>

**Table S20.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5d from the precursor 6d

Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F]5d (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	115.1	12.9	100	97	10.9
2	108.9	12.5	100	97	11.1
3	98.2	11.7	100	98	11.7
<b>Radiochemical Yield (%) ± Deviation</b>					<b>11.2 ± 0.3</b>

**Table S21.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5e from the precursor 6e

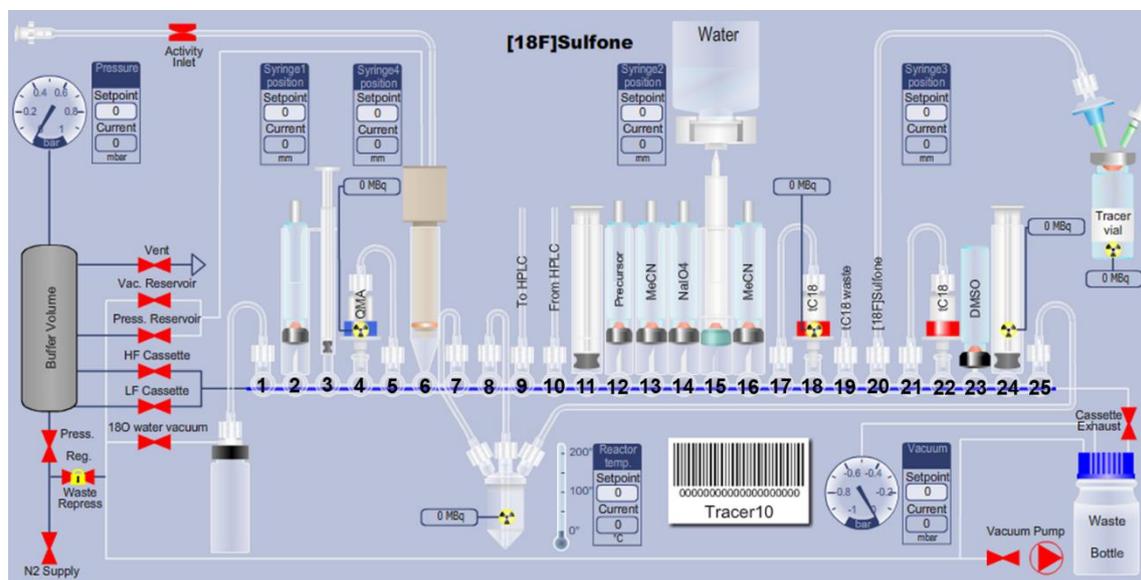
Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F]5e (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	106.8	8.2	100	98	7.5
2	99.1	7.4	99	96	7.1
3	105.3	7.6	99	96	6.9
<b>Radiochemical Yield (%) ± Deviation</b>					<b>7.2 ± 0.2</b>

**Table S22.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5f from the precursor 6f

Reaction	Starting activity (MBq)	Activity of the crude product [ <sup>18</sup> F]5f (MBq, d.c.)	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	116.8	15.8	100	100	13.5
2	111.3	14.6	100	100	13.1
3	110.6	15.6	100	100	14.1
<b>Radiochemical Yield (%) ± Deviation</b>					<b>13.6 ± 0.4</b>

## 2.5. Fully automated radiosyntheses of the labeled compounds [<sup>18</sup>F]5a, [<sup>18</sup>F]5c, and [<sup>18</sup>F]5f

### 2.5.1. Layout of the FASTlab™ cassette for the radiosyntheses of the labeled compounds [<sup>18</sup>F]5a, [<sup>18</sup>F]5c, and [<sup>18</sup>F]5f



**Figure S81.** Layout of the FASTlab™ cassette for the radiosyntheses of the labeled compounds [<sup>18</sup>F]5a, [<sup>18</sup>F]5c, and [<sup>18</sup>F]5f.

**Table S23.** Location of the reagents, solvents, and materials in the manifold of the FASTlab™ cassette

Manifold position	Reagents, solvents, and materials	Details
V1	Silicone tubing connected to [ <sup>18</sup> O]H <sub>2</sub> O recovery vial	14 cm
V2	K <sub>2.2</sub> <sup>®</sup> (7.5 mg) in MeCN (600 μL) and K <sub>2</sub> CO <sub>3</sub> (1.4 mg) in H <sub>2</sub> O (150 μL)	11 mm vial (volume = 750 μL)
V3	Syringe S1 (part of the manifold)	Maximum volume = 1 mL
V4	Sep-Pak® Accell™ Plus QMA Carbonate Plus Light Cartridge with silicone tubing at position V5	46 mg (40 μm) (Waters)
V5	Silicone tubing connected to the Sep-Pak® Accell™ Plus QMA Carbonate Plus Light Cartridge at position V4	14 cm
V6	[ <sup>18</sup> O]H <sub>2</sub> O/[ <sup>18</sup> F]F <sup>-</sup> inlet conical reservoir (part of the manifold)	Maximum volume = 5 mL
V7	Silicone tubing connected to the cyclic olefin copolymer (COC) reactor (left-hand side)	14 cm
V8	Silicone tubing connected to the COC reactor (central port)	14 cm
V9	Outlet "to HPLC loop" via silicone tubing connected to a Sterifix® Paed filter (B. Braun)	30 cm
V10	Inlet "from HPLC loop" enabling the recovery of the purified labeled compounds [ <sup>18</sup> F]5a, [ <sup>18</sup> F]5c, and [ <sup>18</sup> F]5f after semi-preparative HPLC purification	30 cm
V11	Syringe S2 (part of the manifold)	Maximum volume = 6 mL
V12	Precursors <b>6a</b> (12.3 mg, 40 μmol), <b>6c</b> (6.5 mg, 20 μmol), or <b>6f</b> (11.6 mg, 40 μmol) solubilized in MeCN	11 mm vial (volume = 1 mL)

V13	MeCN	13 mm vial (volume = 4 mL)
V14	NaIO <sub>4</sub> (153.9 mg) and RuCl <sub>3</sub> ·xH <sub>2</sub> O (3.4 mg) solubilized in H <sub>2</sub> O	13 mm vial (volume = 4 mL)
V15	Water bag spike	Volume = 100 mL
V16	MeCN	13 mm vial (volume = 4 mL)
V17	Silicone tubing connected to the Sep-Pak® C18 Plus Short Cartridge at position V18	14 cm
V18	Sep-Pak® C18 Plus Short Cartridge with silicone tubing at position V17	400 mg (37-55 μm)
V19	Outlet waste bottle	21 cm
V20	Final outlet vial for collection of the labeled compounds [ <sup>18</sup> F]5a, [ <sup>18</sup> F]5c, and [ <sup>18</sup> F]5f after semi-preparative HPLC purification and reformulation	50 cm
V21	Silicone tubing connected to the Sep-Pak® C18 Plus Short Cartridge at position V22	14 cm
V22	Sep-Pak® C18 Plus Short Cartridge with silicone tubing at position V21	400 mg (37-55 μm)
V23	Anhydrous DMSO	13 mm vial (volume = 4 mL)
V24	Syringe S3 (part of the manifold)	Maximum volume = 6 mL
V25	Silicone tubing connected to the COC reactor (right-hand side) and vent valve for the reactor	42 cm

The RCY of the fully automated radiosyntheses of the [<sup>18</sup>F]difluoromethyl heteroaryl-sulfones [<sup>18</sup>F]5a, [<sup>18</sup>F]5c, and [<sup>18</sup>F]5f was determined based on the radioactivity of the [<sup>18</sup>F]5a, [<sup>18</sup>F]5c, or [<sup>18</sup>F]5f present in DMSO solution and the radioactivity trapped on the QMA carbonate cartridge, according to the following formula:

$$RCY (\%, d. c.) = \frac{\text{activity of the solution of } [^{18}\text{F}]5a, [^{18}\text{F}]5c, \text{ or } [^{18}\text{F}]5f \text{ in DMSO (d. c.)}}{\text{activity trapped on the QMA carbonate cartridge} \times 100}$$

**Table S24.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5a from the precursor 6a

Reaction	Starting activity (GBq)	Activity of the isolated product [ <sup>18</sup> F]5a (GBq, d.c.)	Radiochemical Yield (%)
1	135.0	4.0	3.0
2	137.3	3.7	2.7
3	148.9	4.3	2.9
<b>Radiochemical Yield (%) ± Deviation</b>			<b>2.9 ± 0.1</b>

**Table S25.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5c from the precursor 6c

Reaction	Starting activity (GBq)	Activity of the isolated product [ <sup>18</sup> F]5c (GBq, d.c.)	Radiochemical Yield (%)
1	134.3	6.3	4.7
2	146.9	8.5	5.8
3	135.3	8.2	6.1
4	127.6	7.5	5.9
5	147.6	9.0	6.1
<b>Radiochemical Yield (%) ± Deviation</b>			<b>5.7 ± 0.5</b>

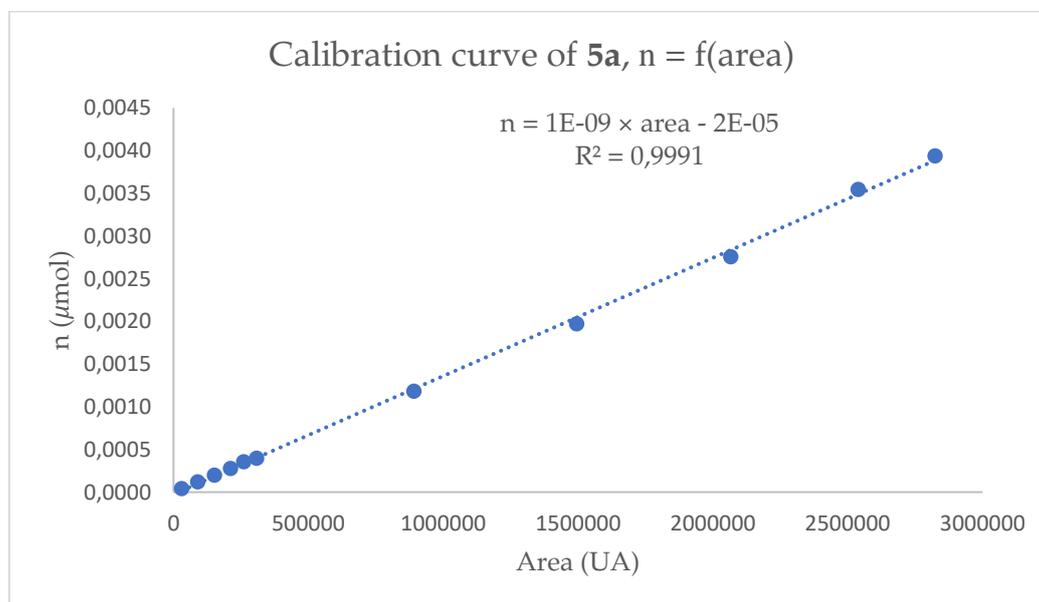
**Table S26.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]5f from the precursor 6f

Reaction	Starting activity (GBq)	Activity of the isolated product [ <sup>18</sup> F]5a (GBq, d.c.)	Radiochemical Yield (%)
1	140.1	10.2	7.3
2	130.9	9.7	7.4
3	124.9	11.6	9.3
Radiochemical Yield (%) ± Deviation			8.0 ± 0.9

### 2.5.2. Calibration curves of the difluoromethyl heteroaryl-sulfones 5a, 5c, and 5f for determination of the molar activity of [<sup>18</sup>F]5a, [<sup>18</sup>F]5c, and [<sup>18</sup>F]5f

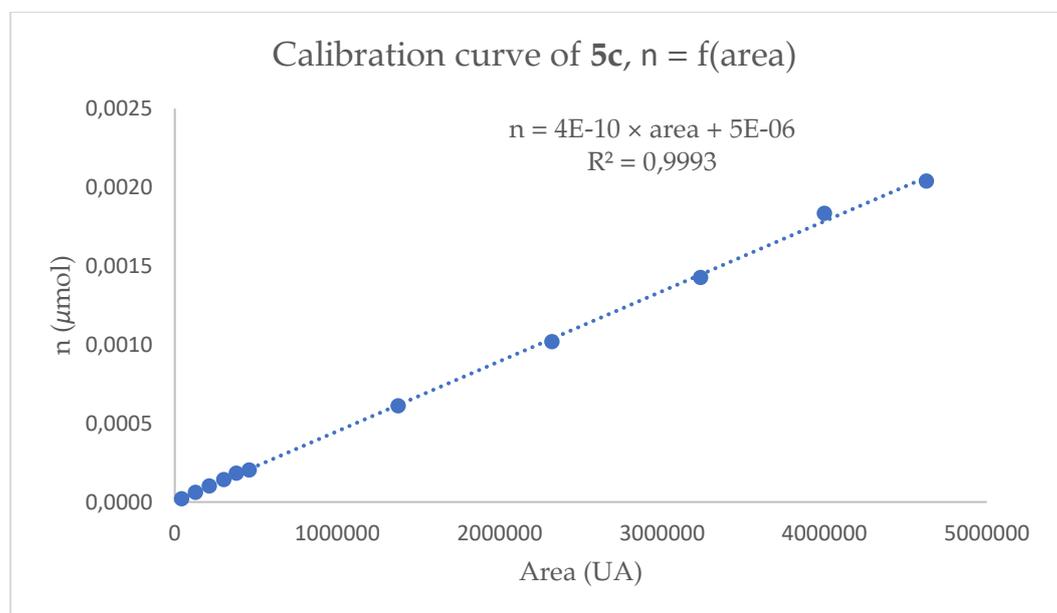
The fully automated radiosyntheses of the sulfones [<sup>18</sup>F]5a, [<sup>18</sup>F]5c, or [<sup>18</sup>F]5f were performed on a commercially available FASTlab™ synthesizer (GE Healthcare), using the optimized conditions for the labeling of precursors 6a (12.3 mg, 0.04 mmol), 6c (6.5 mg, 0.02 mmol), or 6f (11.6 mg, 0.04 mmol), and the oxidation of the labeled compounds [<sup>18</sup>F]4a, [<sup>18</sup>F]4c, and [<sup>18</sup>F]4f. The molar activity of the [<sup>18</sup>F]difluoromethyl heteroaryl-sulfones was determined using an aliquot of each reformulated solution (3 μL). After injection of an aliquot in UPLC, the radioactive peak of [<sup>18</sup>F]5a, [<sup>18</sup>F]5c, and [<sup>18</sup>F]5f associated to the non-radioactive sulfones 5a, 5c, and 5f, respectively, were collected and counted in an ionization chamber. The PDA UV area under the peak of the non-radioactive sulfones 5a, 5c, and 5f at 258 nm, 290 nm, and 244 nm, respectively, enabled the determination of the corresponding amount (in μmol) of the difluoromethyl heteroaryl-sulfones using the calibration curves described in Figures S82-S84. The molar activity was calculated by the ratio between the radioactivity of the [<sup>18</sup>F]5a, [<sup>18</sup>F]5c, and [<sup>18</sup>F]5f and the corresponding amount of non-radioactive compound, according to the following formula:

$$\text{Molar activity (GBq} \cdot \mu\text{mol}^{-1}) = \frac{\text{activity of the collected UPLC peak of } [^{18}\text{F}]5\text{a, } [^{18}\text{F}]5\text{c, or } [^{18}\text{F}]5\text{f}}{\text{amount of } 5\text{a, } 5\text{c, or } 5\text{f associated to the radioactive peak}}$$

**Figure S82.** Calibration curve of the difluoromethyl heteroaryl-sulfone 5a (wavelength: 258 nm).

**Table S27.** Determination of the molar activity of [<sup>18</sup>F]5a

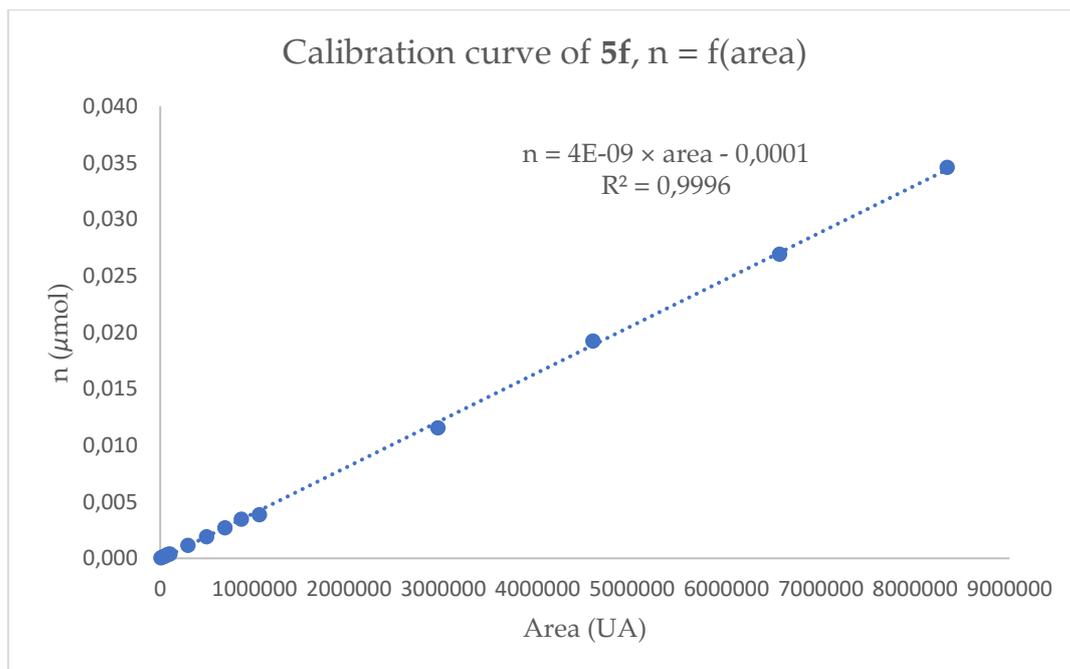
Reaction	Activity of the radioactive peak of [ <sup>18</sup> F]5a (GBq)	Area under the peak of 5a (UA) at 258 nm	Amount of 5a (μmol)	Molar activity (GBq·μmol <sup>-1</sup> )
1	$4.552 \times 10^{-3}$	40425	$3.146 \times 10^{-5}$	145
2	$4.454 \times 10^{-3}$	38473	$2.876 \times 10^{-5}$	155
3	$5.175 \times 10^{-3}$	50000	$4.471 \times 10^{-5}$	116
<b>Molar activity (GBq·μmol<sup>-1</sup>) ± Deviation</b>				<b>139 ± 17</b>



**Figure S83.** Calibration curve of the difluoromethyl heteroaryl-sulfone 5c (wavelength: 290 nm).

**Table S28.** Determination of the molar activity of [<sup>18</sup>F]5c

Reaction	Activity of the radioactive peak of [ <sup>18</sup> F]5c (GBq)	Area under the peak of 5c (UA) at 290 nm	Amount of 5c (μmol)	Molar activity (GBq·μmol <sup>-1</sup> )
1	$4.863 \times 10^{-3}$	247987	$1.150 \times 10^{-4}$	42
2	$2.662 \times 10^{-3}$	90325	$4.489 \times 10^{-5}$	59
3	$8.242 \times 10^{-3}$	244309	$1.134 \times 10^{-4}$	73
4	$2.038 \times 10^{-3}$	65547	$3.387 \times 10^{-5}$	60
5	$2.524 \times 10^{-3}$	66094	$3.412 \times 10^{-5}$	74
<b>Molar activity (GBq·μmol<sup>-1</sup>) ± Deviation</b>				<b>62 ± 12</b>



**Figure S84.** Calibration curve of the difluoromethyl heteroaryl-sulfone **5f** (wavelength: 244 nm).

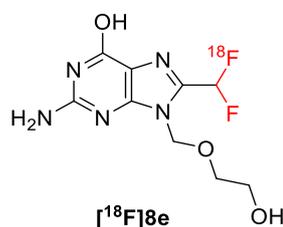
**Table S29.** Determination of the molar activity of [ $^{18}\text{F}$ ]**5f**

Reaction	Activity of the radioactive peak of [ $^{18}\text{F}$ ] <b>5f</b> (GBq)	Area under the peak of <b>5f</b> (UA) at 244 nm	Amount of <b>5f</b> ( $\mu\text{mol}$ )	Molar activity (GBq· $\mu\text{mol}^{-1}$ )
1	$1.702 \times 10^{-2}$	57459	$1.378 \times 10^{-4}$	124
2	$1.193 \times 10^{-2}$	57048	$1.361 \times 10^{-4}$	88
3	$1.435 \times 10^{-2}$	51785	$1.143 \times 10^{-4}$	126
<b>Molar activity (GBq·<math>\mu\text{mol}^{-1}</math>) <math>\pm</math> Deviation</b>				113 $\pm$ 17

## 2.6. Optimization of the conditions for the photocatalytic C-H $^{18}\text{F}$ -difluoromethylation with the reagents [ $^{18}\text{F}$ ]**5a**, [ $^{18}\text{F}$ ]**5c**, and [ $^{18}\text{F}$ ]**5f**

The propensity of the reagents [ $^{18}\text{F}$ ]**5a**, [ $^{18}\text{F}$ ]**5c**, and [ $^{18}\text{F}$ ]**5f** to undergo the desired photocatalytic C-H  $^{18}\text{F}$ -difluoromethylation reaction was carried out using 2-amino-9-((2-hydroxyethoxy)methyl)-9H-purin-6-ol (acyclovir, **7e**) as a model substrate.

### 2.6.1. Synthesis of [ $^{18}\text{F}$ ]2-amino-8-(difluoromethyl)-9-((2-hydroxyethoxy)methyl)-9H-purin-6-ol ([ $^{18}\text{F}$ ]**8e**)



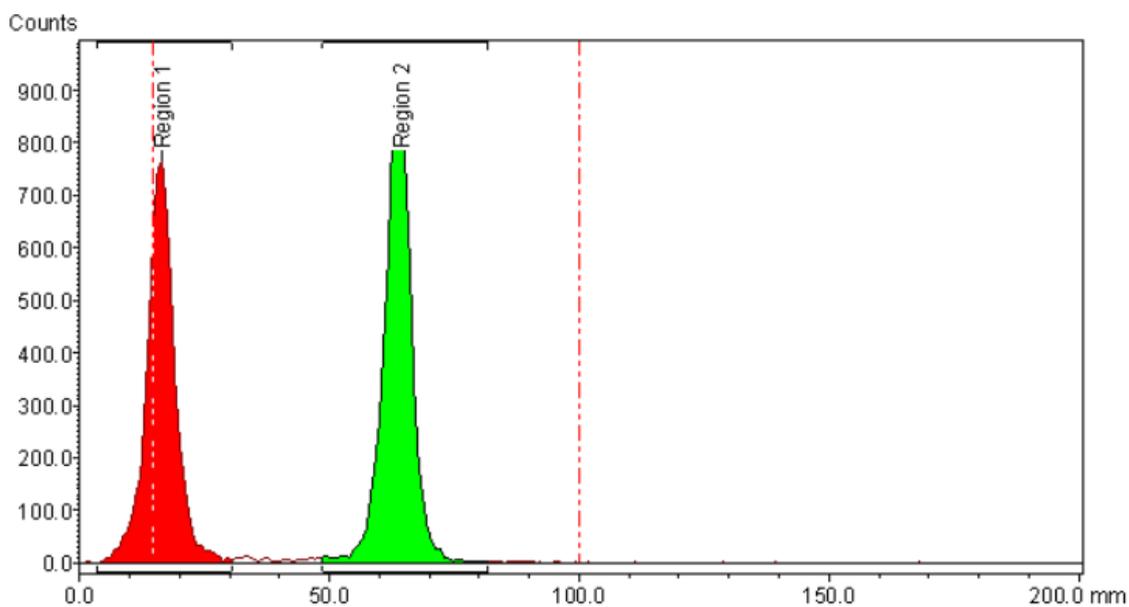
A solution of 2-amino-9-((2-hydroxyethoxy)methyl)-9H-purin-6-ol (acyclovir, **7e**) and *fac*- $\text{Ir}^{\text{III}}$ (ppy)<sub>3</sub> in DMSO (200  $\mu\text{L}$ ) was prepared. Next, a solution of [ $^{18}\text{F}$ ]**5a**, [ $^{18}\text{F}$ ]**5c**, or [ $^{18}\text{F}$ ]**5f** in DMSO (30-40 MBq, 50  $\mu\text{L}$ ) was added. The solution was injected in a 100  $\mu\text{L}$ -microchip pumped with DMSO at a flow rate and irradiated under blue LED (470 nm, 2 W), at a temperature of 35  $^{\circ}\text{C}$ . An aliquot of the crude product [ $^{18}\text{F}$ ]**8e** was then analyzed by radio-TLC and radio-UPLC for radiochemical yield (RCY) determination.



**Figure S85.** Instrument used for the C-H  $^{18}\text{F}$ -difluoromethylation reaction of the heteroarenes (FlowStart Evo, FutureChemistry).

The RCY of the C-H  $^{18}\text{F}$ -difluoromethylation reactions was determined according to the following formula:

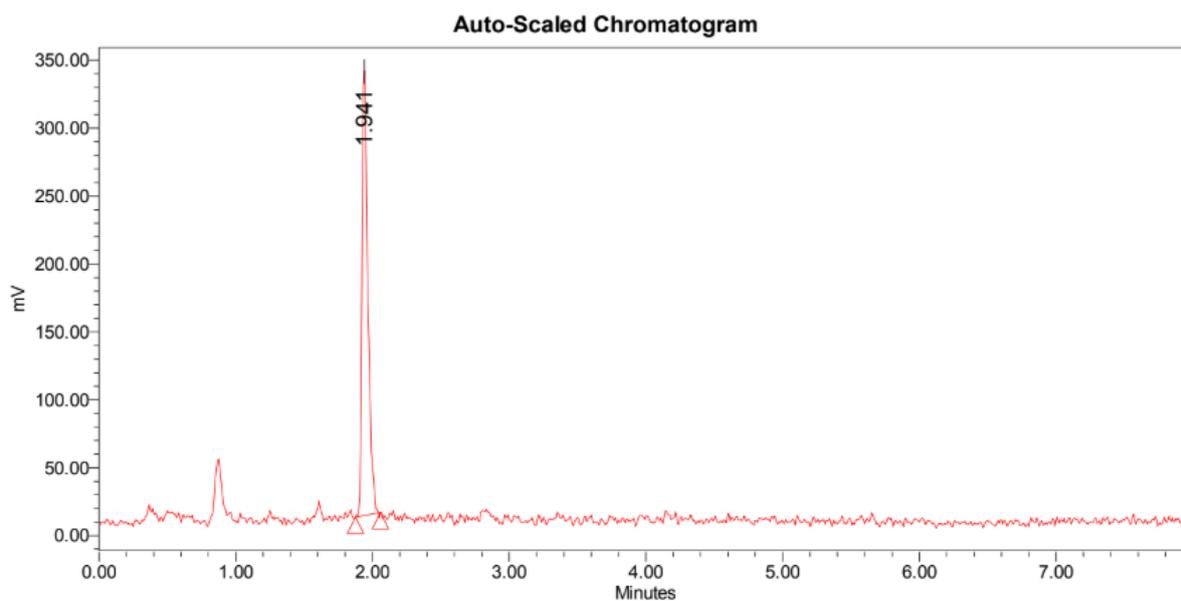
$$\text{RCY (\%)} = \frac{\text{radioTLC purity (\%)} \times \text{radioUPLC purity (\%)}}{100}$$



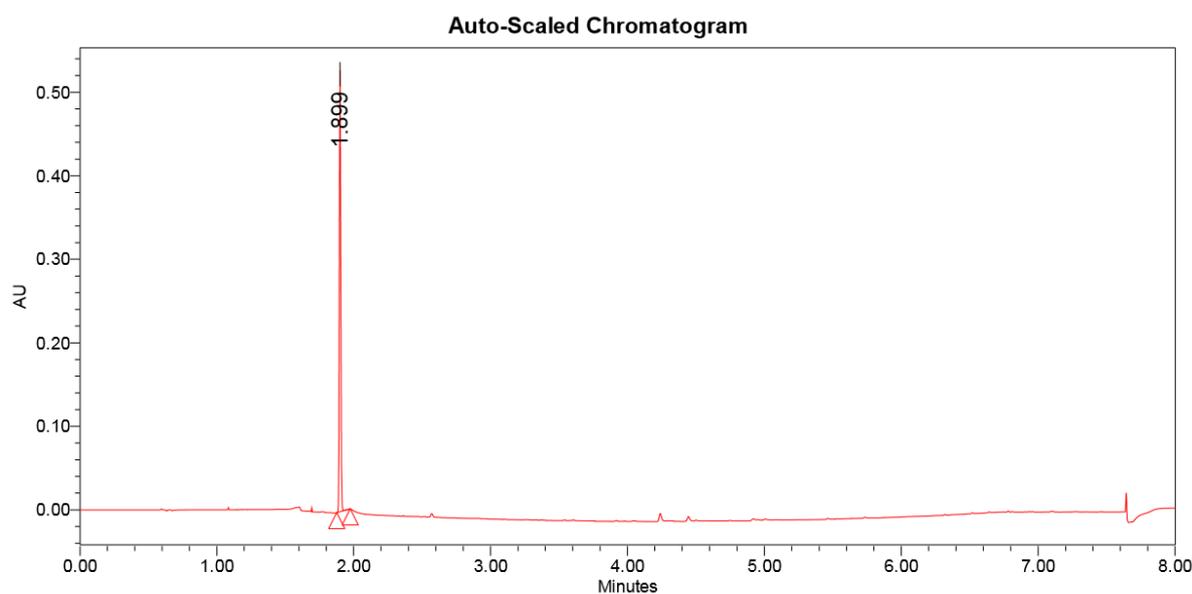
**Figure S86.** TLC radio-chromatogram of the crude product  $[^{18}\text{F}]\mathbf{8e}$  (eluent: methanol).

**Table S30.** Determination of the radio-TLC purity of the crude product  $[^{18}\text{F}]\mathbf{8e}$

Retention factor ( $R_f$ , mm)	Ratio (%)
0.02	46 (impurity/by-product)
0.77	54 (desired crude product)



**Figure S87.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]8e (gradient B).



**Figure S88.** UPLC UV-chromatogram of the authentic reference 8e (gradient B).

$$RCY (\%) = \frac{\text{radioTLC purity (\%)} \times \text{radioUPLC purity (\%)}}{100}$$

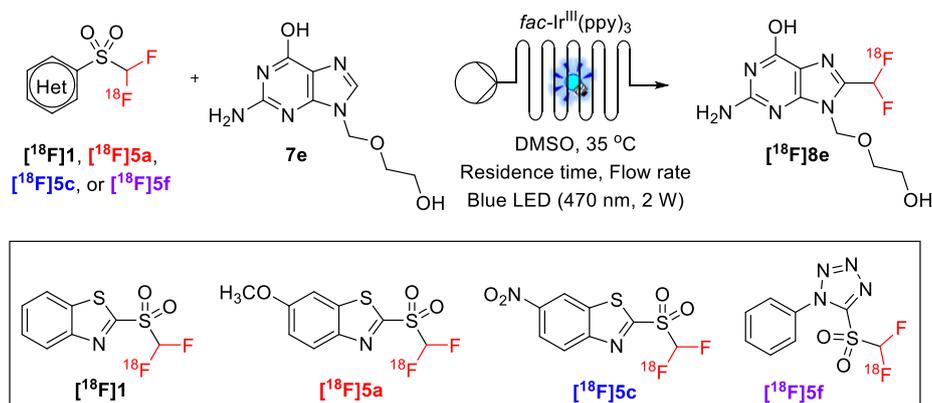
$$RCY (\%) = \frac{54 \times 100}{100}$$

$$RCY (\%) = 54 \%$$

Note: In some cases, some peaks at 0.6 and 0.9 min were observed on the radio-UPLC chromatograms. Those two peaks were collected and their radio-TLC purity was analyzed. Since the retention factor corresponding to these peaks is approximately 0, their contribution for the radio-UPLC purity was not taken into consideration. The contribution of these peaks was accounted for the determination of the radio-TLC purity.

On the basis of the reported conditions for the C-H  $^{18}\text{F}$ -difluoromethylation of the substrate **7e** with the reagent  $[^{18}\text{F}]\mathbf{1}$ , the following table summarizes the results of the different optimization tests performed in the presence of the reagents  $[^{18}\text{F}]\mathbf{5a}$ ,  $[^{18}\text{F}]\mathbf{5c}$ , and  $[^{18}\text{F}]\mathbf{5f}$ .

**Standard reaction conditions for the photoredox C-H  $^{18}\text{F}$ -difluoromethylation:** substrate **7e** (0.02 mmol),  $[^{18}\text{F}]$ difluoromethyl heteroaryl-sulfones  $[^{18}\text{F}]\mathbf{5a}$ ,  $[^{18}\text{F}]\mathbf{5c}$ , or  $[^{18}\text{F}]\mathbf{5f}$  (30-40 MBq), *fac*- $\text{Ir}^{\text{III}}(\text{ppy})_3$  (mol%), residence time (min), flow rate ( $\mu\text{L}\cdot\text{min}^{-1}$ ), DMSO (250  $\mu\text{L}$ ), 35  $^\circ\text{C}$ , blue LED (470 nm, 2 W).



Entry	Reagents	<i>fac</i> - $\text{Ir}^{\text{III}}(\text{ppy})_3$ (mol%)	Residence time (min)	Flow rate ( $\mu\text{L}\cdot\text{min}^{-1}$ )	Conversion (%) <sup>(a)</sup>	RCY (%) <sup>(b)</sup>
1	$[^{18}\text{F}]\mathbf{1}$	0.05	2	50	100	$70 \pm 7$ (n=4)
2	$[^{18}\text{F}]\mathbf{5a}$	0.05	2	50	100	$57 \pm 7$ (n=3)
3	$[^{18}\text{F}]\mathbf{5c}$	0.05	2	50	17	$14 \pm 1$ (n=3)
4	$[^{18}\text{F}]\mathbf{5c}$	0.5	2	50	36	$26 \pm 3$ (n=3)
5	$[^{18}\text{F}]\mathbf{5c}$	0.5	4	25	100	$51 \pm 7$ (n=4)
6	$[^{18}\text{F}]\mathbf{5f}$	0.05	2	50	73	$48 \pm 8$ (n=3)
7	$[^{18}\text{F}]\mathbf{5f}$	0.1	2	50	98	$55 \pm 1$ (n=3)
8	$[^{18}\text{F}]\mathbf{5f}$	0.1	2.5	40	100	$56 \pm 1$ (n=3)

<sup>(a)</sup> UPLC conversion of the reagents. <sup>(b)</sup> All RCYs were determined based on the radio-TLC and radio-UPLC purities of the crude product  $[^{18}\text{F}]\mathbf{8e}$ .

The best conditions for the C-H  $^{18}\text{F}$ -difluoromethylation of the substrate **7e** were:

- **Conditions A:**  $[^{18}\text{F}]$ difluoromethyl heteroaryl-sulfone  $[^{18}\text{F}]\mathbf{5a}$  (30-40 MBq), *fac*- $\text{Ir}^{\text{III}}(\text{ppy})_3$  (0.05 mol%), residence time (2 min), flow rate (50  $\mu\text{L}\cdot\text{min}^{-1}$ ), DMSO (250  $\mu\text{L}$ ), 35  $^\circ\text{C}$ , blue LED (470 nm, 2 W).
- **Conditions B:**  $[^{18}\text{F}]$ difluoromethyl heteroaryl-sulfone  $[^{18}\text{F}]\mathbf{5c}$  (30-40 MBq), *fac*- $\text{Ir}^{\text{III}}(\text{ppy})_3$  (0.5 mol%), residence time (4 min), flow rate (25  $\mu\text{L}\cdot\text{min}^{-1}$ ), DMSO (250  $\mu\text{L}$ ), 35  $^\circ\text{C}$ , blue LED (470 nm, 2 W).
- **Conditions C:**  $[^{18}\text{F}]$ difluoromethyl heteroaryl-sulfone  $[^{18}\text{F}]\mathbf{5f}$  (30-40 MBq), *fac*- $\text{Ir}^{\text{III}}(\text{ppy})_3$  (0.1 mol%), residence time (2.5 min), flow rate (40  $\mu\text{L}\cdot\text{min}^{-1}$ ), DMSO (250  $\mu\text{L}$ ), 35  $^\circ\text{C}$ , blue LED (470 nm, 2 W).

**Table S31.** Determination of the radiochemical yield (%) of the synthesis of  $[^{18}\text{F}]\mathbf{8e}$  using the reagent  $[^{18}\text{F}]\mathbf{5a}$

Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	54	100	54
2	67	100	67
3	51	100	51
Radiochemical Yield (%) $\pm$ Deviation			$57 \pm 7$

**Table S32.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**8e** using the reagent [<sup>18</sup>F]**5c**

Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	74	79	58
2	68	57	39
3	63	84	53
4	67	80	54
Radiochemical Yield (%) ± Deviation			51 ± 7

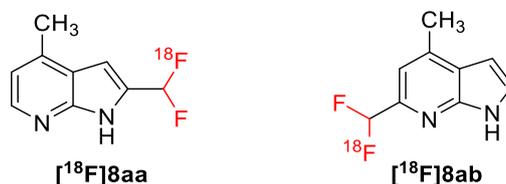
**Table S33.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**8e** using the reagent [<sup>18</sup>F]**5f**

Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	63	85	54
2	69	83	57
3	71	80	57
Radiochemical Yield (%) ± Deviation			56 ± 1

## 2.7. General procedure for the C-H <sup>18</sup>F-difluoromethylation reaction with the [<sup>18</sup>F]difluoromethyl heteroaryl-sulfones [<sup>18</sup>F]**5a**, [<sup>18</sup>F]**5c**, and [<sup>18</sup>F]**5f**

A solution of the heteroarenes (0.02 mmol) and *fac*-Ir<sup>III</sup>(ppy)<sub>3</sub> (0.05 mol% for [<sup>18</sup>F]**5a**; 0.5 mol% for [<sup>18</sup>F]**5c**; 0.1 mol% for [<sup>18</sup>F]**5f**) in DMSO (200 μL) was prepared. Next, a solution of [<sup>18</sup>F]**5a**, [<sup>18</sup>F]**5c**, or [<sup>18</sup>F]**5f** in DMSO (30-40 MBq, 50 μL) was added. The solution was injected in a 100 μL-microchip pumped with DMSO at a flow rate of 50 μL·min<sup>-1</sup> (residence time: 2 min for [<sup>18</sup>F]**5a**), 25 μL·min<sup>-1</sup> (residence time: 4 min for [<sup>18</sup>F]**5c**) or 40 μL·min<sup>-1</sup> (residence time: 2.5 min for [<sup>18</sup>F]**5f**) and irradiated under blue LED (470 nm, 2 W), at a temperature of 35 °C. An aliquot of the reaction mixture was then analyzed by radio-TLC and radio-UPLC for the RCY determination.

### 2.7.1. Synthesis of [<sup>18</sup>F]2-(difluoromethyl)-4-methyl-1*H*-pyrrolo[2,3-*b*]pyridine ([<sup>18</sup>F]**8aa**) and [<sup>18</sup>F]6-(difluoromethyl)-4-methyl-1*H*-pyrrolo[2,3-*b*]pyridine ([<sup>18</sup>F]**8ab**)



The implementation of the general procedure for the C-H <sup>18</sup>F-difluoromethylation of 4-methyl-1*H*-pyrrolo[2,3-*b*]pyridine (2.6 mg, 0.02 mmol) provided the labeled compound [<sup>18</sup>F]**8aa** in 32 ± 1%, 58 ± 3%, and 46 ± 3% RCY, using the reagents [<sup>18</sup>F]**5a**, [<sup>18</sup>F]**5c**, and [<sup>18</sup>F]**5f**, respectively. The labeled compound [<sup>18</sup>F]**8ab** was afforded in 5 ± 1%, 8%, and 7% RCY, using the reagents [<sup>18</sup>F]**5a**, [<sup>18</sup>F]**5c**, and [<sup>18</sup>F]**5f**, respectively (see the Tables S34-S36 for more details of the RCY determination). The UPLC radiochromatogram of the crude product [<sup>18</sup>F]**8a** is depicted in Figure S89. Figures S90 and S91 represent the UPLC UV-chromatograms of the non-radioactive references **8aa** and **8ab**, respectively.

**Table S34.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**8a** using the reagent [<sup>18</sup>F]**5a**

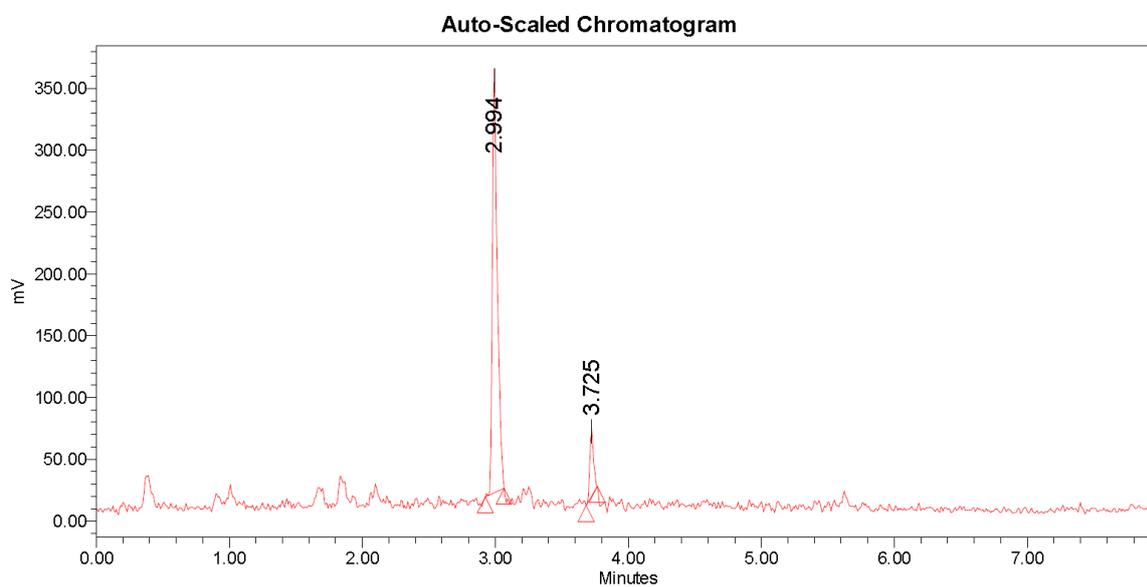
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)		Radiochemical Yield (%)	
	a + b	a	b	a	b
1	37	90	10	33	4
2	34	89	11	30	4
3	38	85	15	32	6
Radiochemical Yield (%) ± Deviation				32 ± 1	5 ± 1

**Table S35.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]8a using the reagent [<sup>18</sup>F]5c

Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)		Radiochemical Yield (%)	
	a + b	a	b	a	b
1	67	80	12	54	8
2	76	82	12	62	9
3	76	75	11	57	8
<b>Radiochemical Yield (%) ± Deviation</b>				<b>58 ± 3</b>	<b>8</b>

**Table S36.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]8a using the reagent [<sup>18</sup>F]5f

Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)		Radiochemical Yield (%)	
	a + b	a	b	a	b
1	64	74	11	47	7
2	58	73	10	42	6
3	67	72	10	48	7
<b>Radiochemical Yield (%) ± Deviation</b>				<b>46 ± 3</b>	<b>7</b>



**Figure S89.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]8a (gradient B).

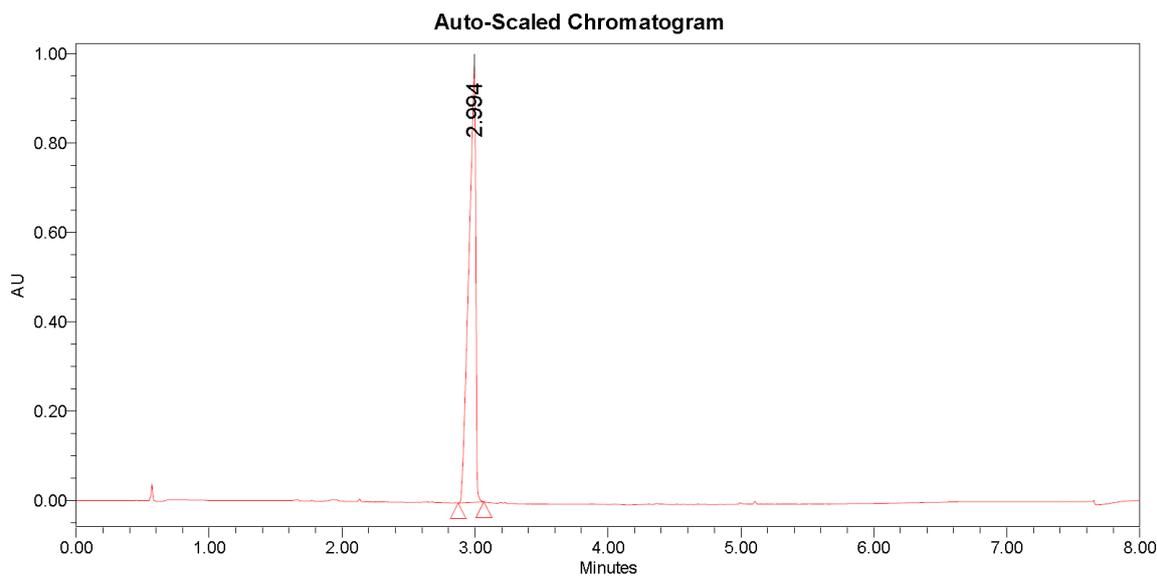


Figure S90. UPLC UV-chromatogram of the authentic reference **8aa** (gradient B).

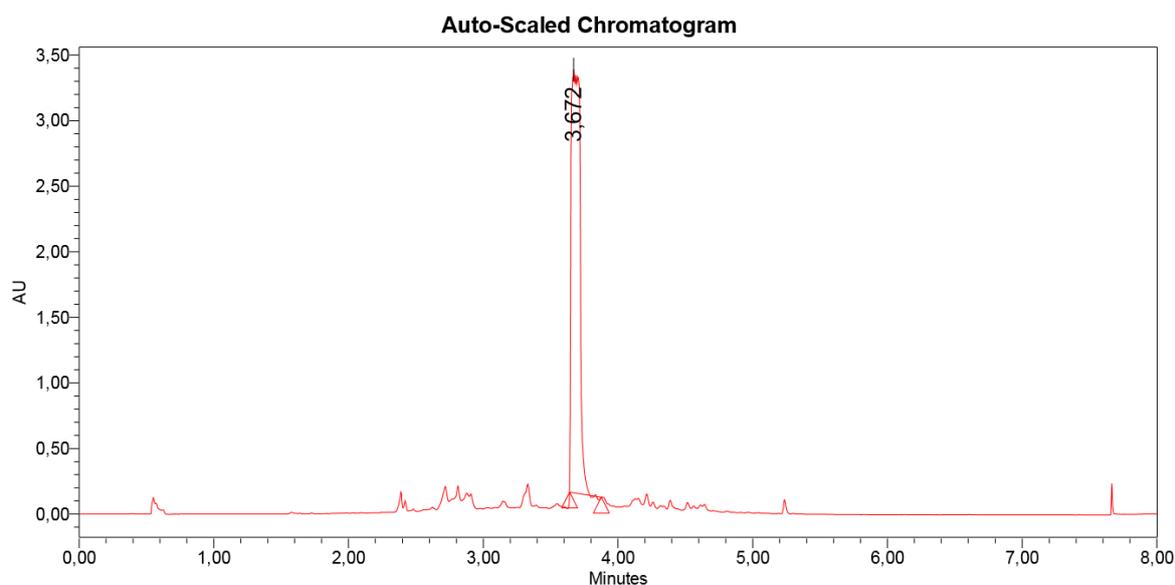
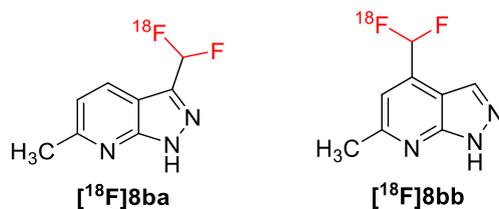


Figure S91. UPLC UV-chromatogram of the authentic reference **8ab** (gradient B).

### 2.7.2. Synthesis of [<sup>18</sup>F]3-(difluoromethyl)-6-methyl-1H-pyrazolo[3,4-*b*]pyridine ([<sup>18</sup>F]**8ba**) and [<sup>18</sup>F]4-(difluoromethyl)-6-methyl-1H-pyrazolo[3,4-*b*]pyridine ([<sup>18</sup>F]**8bb**)



The implementation of the general procedure for the C-H <sup>18</sup>F-difluoromethylation of 6-methyl-1H-pyrazolo[3,4-*b*]pyridine (2.7 mg, 0.02 mmol) provided the labeled compound [<sup>18</sup>F]**8aa** in 41 ± 6%, 43 ± 3%, and 41 ± 5% RCY, using the reagents [<sup>18</sup>F]**5a**, [<sup>18</sup>F]**5c**, and [<sup>18</sup>F]**5f**, respectively. The labeled compound

[<sup>18</sup>F]**8ab** was afforded in 16 ± 1%, 17 ± 3%, and 17 ± 1% RCY, using the reagents [<sup>18</sup>F]**5a**, [<sup>18</sup>F]**5c**, and [<sup>18</sup>F]**5f**, respectively (see the Tables S37-S39 for more details of the RCY determination). The UPLC radio-chromatogram of the crude product [<sup>18</sup>F]**8b** is depicted in Figure S92. Figures S93 and S94 represent the UPLC UV-chromatograms of the non-radioactive references **8ba** and **8bb**, respectively.

**Table S37.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**8b** using the reagent [<sup>18</sup>F]**5a**

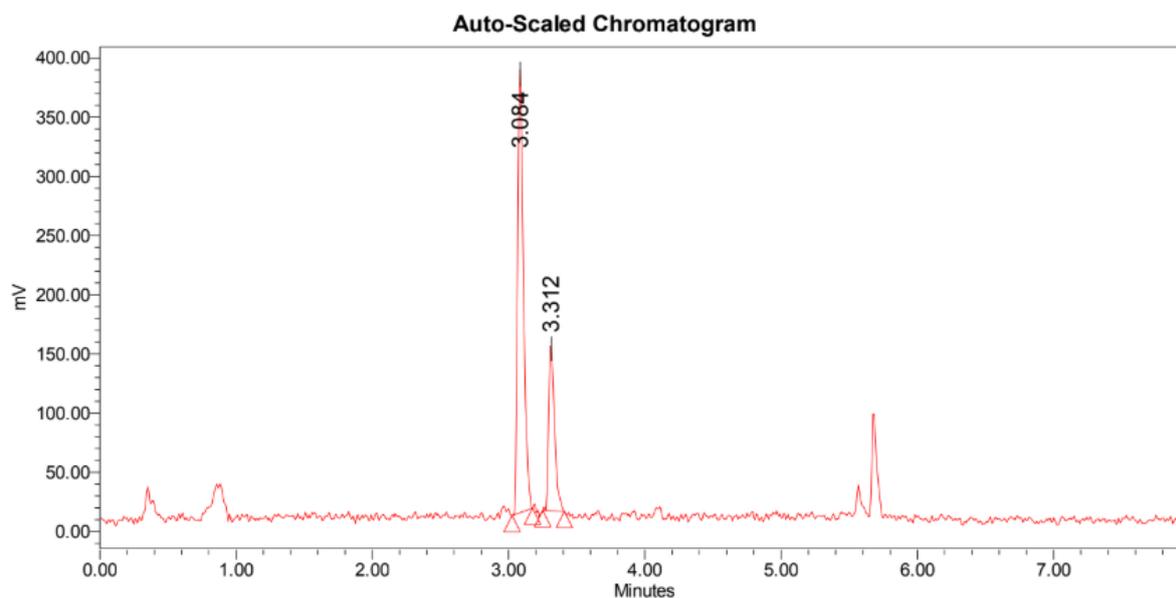
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)		Radiochemical Yield (%)	
	a + b	a	b	a	b
1	53	67	33	36	17
2	67	75	25	50	17
3	55	68	27	37	15
<b>Radiochemical Yield (%) ± Deviation</b>				41 ± 6	16 ± 1

**Table S38.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**8b** using the reagent [<sup>18</sup>F]**5c**

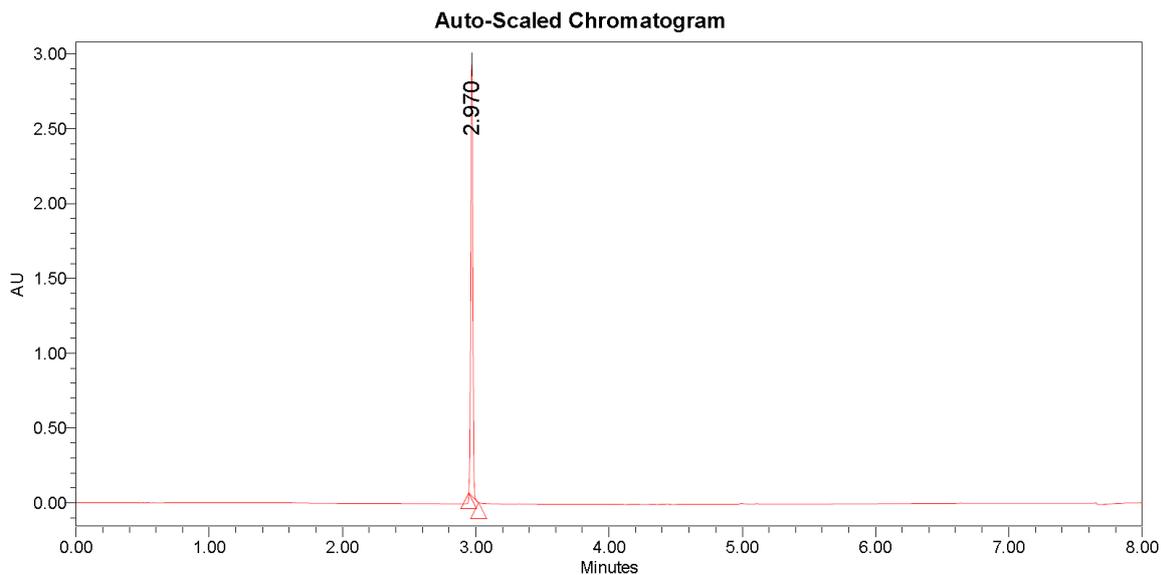
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)		Radiochemical Yield (%)	
	a + b	a	b	a	b
1	71	67	22	48	16
2	74	57	28	42	21
3	62	65	22	40	14
<b>Radiochemical Yield (%) ± Deviation</b>				43 ± 3	17 ± 3

**Table S39.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**8b** using the reagent [<sup>18</sup>F]**5f**

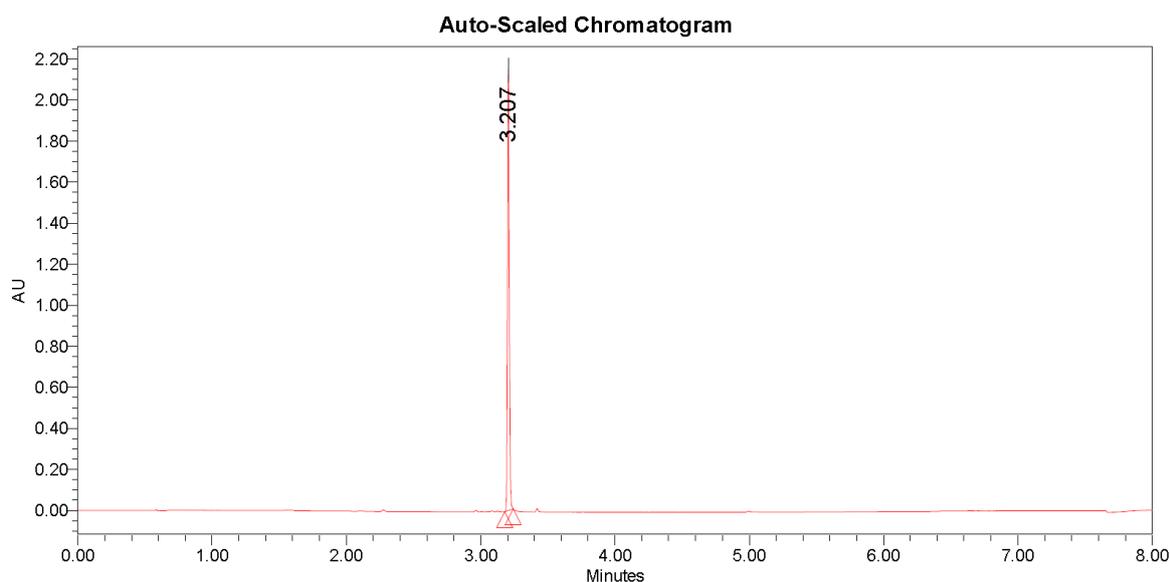
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)		Radiochemical Yield (%)	
	a + b	a	b	a	b
1	70	50	25	35	18
2	65	61	24	40	16
3	74	63	21	47	16
<b>Radiochemical Yield (%) ± Deviation</b>				41 ± 5	17 ± 1



**Figure S92.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]**8b** (gradient B).

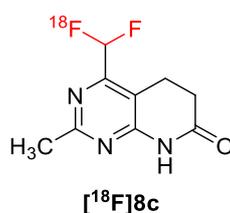


**Figure S93.** UPLC UV-chromatogram of the authentic reference **8ba** (gradient B).



**Figure S94.** UPLC UV-chromatogram of the authentic reference **8bb** (gradient B).

### 2.7.3. Synthesis of [<sup>18</sup>F]4-(difluoromethyl)-2-methyl-5,8-dihydropyrido[2,3-d]pyrimidin-7(6H)-one ([<sup>18</sup>F]**8c**)



The implementation of the general procedure for the C-H <sup>18</sup>F-difluoromethylation of 2-methyl-5,8-dihydropyrido[2,3-d]pyrimidin-7(6H)-one (3.3 mg, 0.02 mmol) provided the labeled compound [<sup>18</sup>F]**8c** in 17 ± 4%, 13 ± 3%, and 14% RCY, using the reagents [<sup>18</sup>F]**5a**, [<sup>18</sup>F]**5c**, and [<sup>18</sup>F]**5f**, respectively (see the Tables S40-S42 for more details of the RCY determination). The UPLC radio-chromatogram of the crude

product [<sup>18</sup>F]8c is depicted in Figure S95. Figure S96 represents the UPLC UV-chromatogram of the non-radioactive reference 8c.

**Table S40.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]8c using the reagent [<sup>18</sup>F]5a

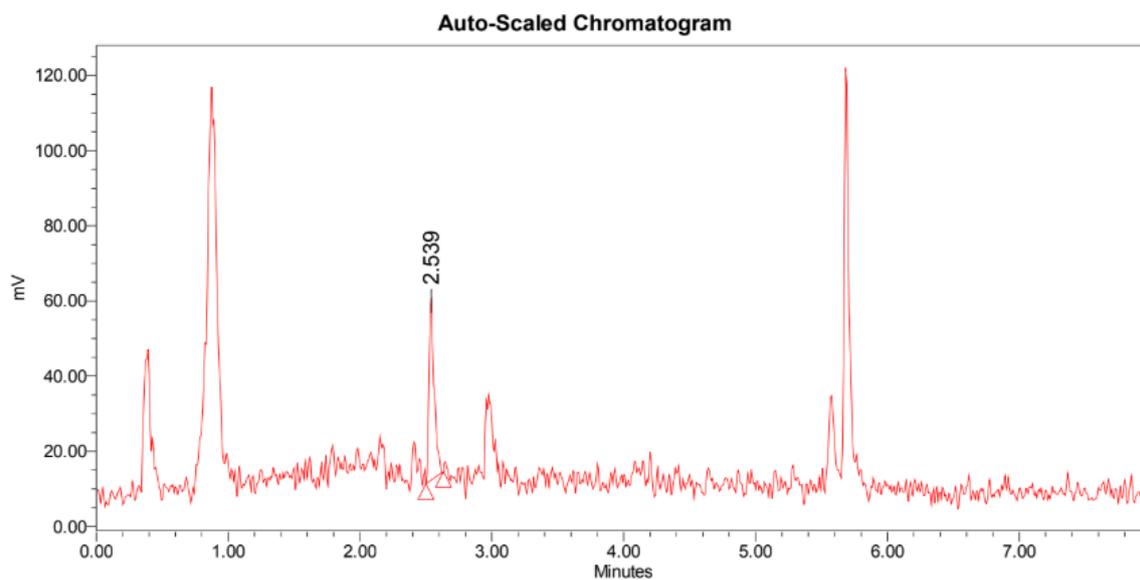
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	38	30	11
2	45	44	20
3	51	41	21
<b>Radiochemical Yield (%) ± Deviation</b>			<b>17 ± 4</b>

**Table S41.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]8c using the reagent [<sup>18</sup>F]5c

Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	57	30	17
2	55	20	11
3	19	52	10
<b>Radiochemical Yield (%) ± Deviation</b>			<b>13 ± 3</b>

**Table S42.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]8c using the reagent [<sup>18</sup>F]5f

Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	54	25	14
2	43	30	13
3	54	25	14
<b>Radiochemical Yield (%) ± Deviation</b>			<b>14</b>



**Figure S95.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]8c (gradient B).

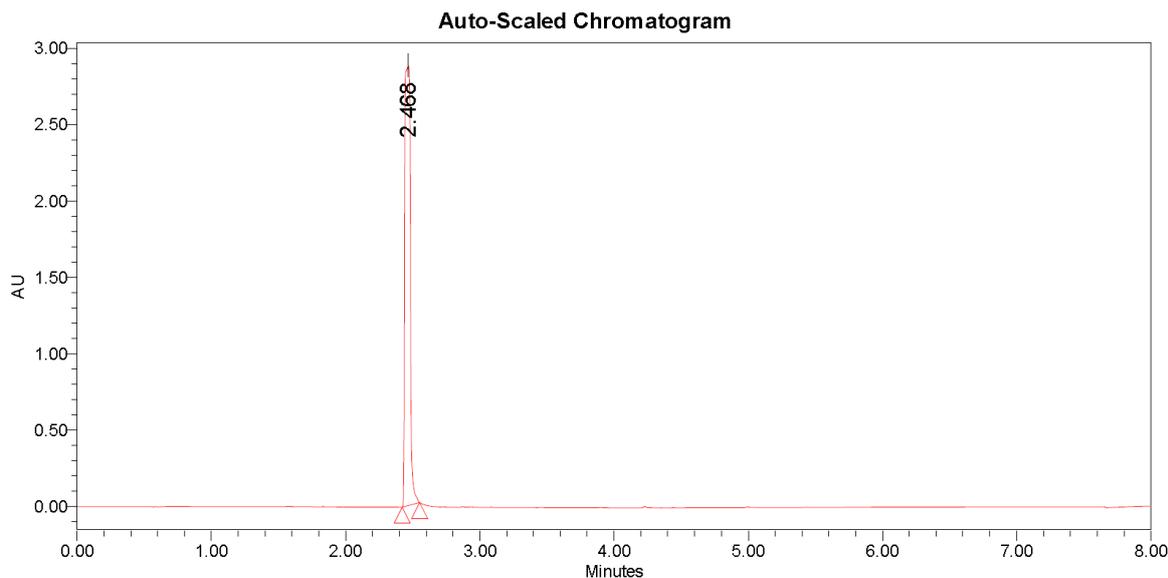
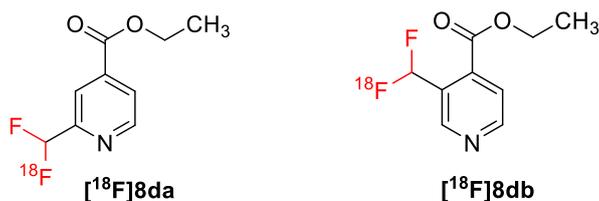


Figure S96. UPLC UV-chromatogram of the authentic reference **8c** (gradient B).

#### 2.7.4. Synthesis of [<sup>18</sup>F]ethyl 2-(difluoromethyl)isonicotinate ([<sup>18</sup>F]**8da**) and [<sup>18</sup>F]ethyl 3-(difluoromethyl)isonicotinate ([<sup>18</sup>F]**8db**)



The implementation of the general procedure for the C-H <sup>18</sup>F-difluoromethylation of ethyl isonicotinate (3.0 mg, 0.02 mmol) provided the labeled compound [<sup>18</sup>F]**8da** in 19 ± 7%, 24 ± 2%, and 36 ± 4% RCY, using the reagents [<sup>18</sup>F]**5a**, [<sup>18</sup>F]**5c**, and [<sup>18</sup>F]**5f**, respectively. The labeled compound [<sup>18</sup>F]**8db** was afforded in 10 ± 3%, 13 ± 1%, and 13 ± 2% RCY, using the reagents [<sup>18</sup>F]**5a**, [<sup>18</sup>F]**5c**, and [<sup>18</sup>F]**5f**, respectively (see the Tables S43-S45 for more details of the RCY determination). The UPLC radio-chromatogram of the crude product [<sup>18</sup>F]**8d** is depicted in Figure S97. Figures S98 and S99 represent the UPLC UV-chromatograms of the non-radioactive references **8da** and **8db**, respectively.

Table S43. Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**8d** using the reagent [<sup>18</sup>F]**5a**

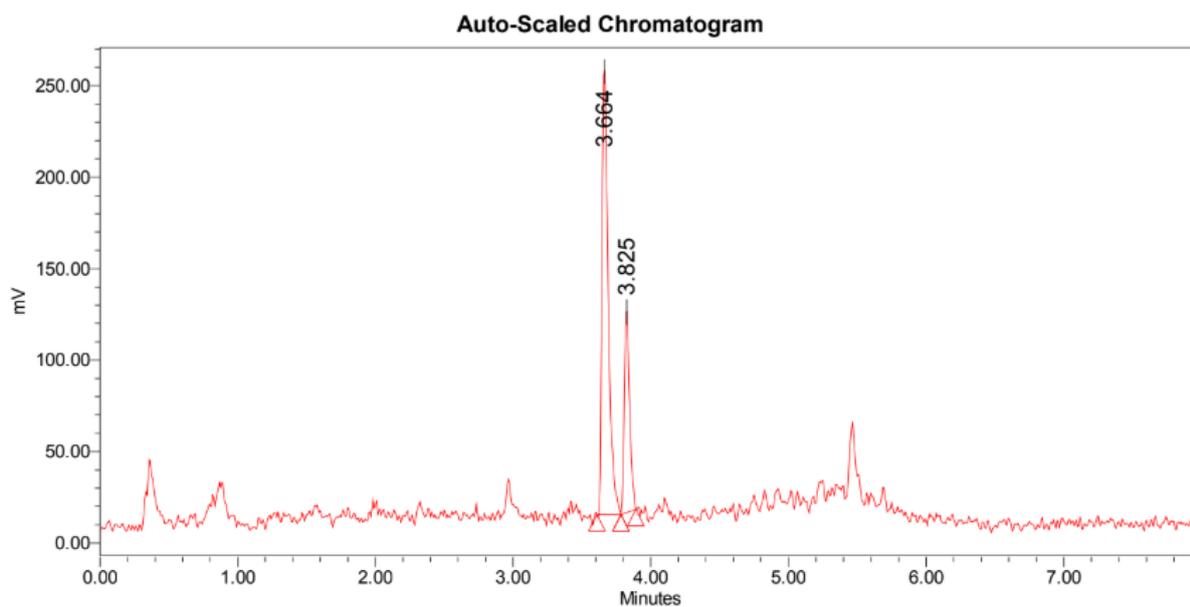
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)		Radiochemical Yield (%)	
	a + b	a	b	a	b
1	29	42	21	12	6
2	32	51	34	16	11
3	47	61	30	29	14
Radiochemical Yield (%) ± Deviation				19 ± 7	10 ± 3

Table S44. Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**8d** using the reagent [<sup>18</sup>F]**5c**

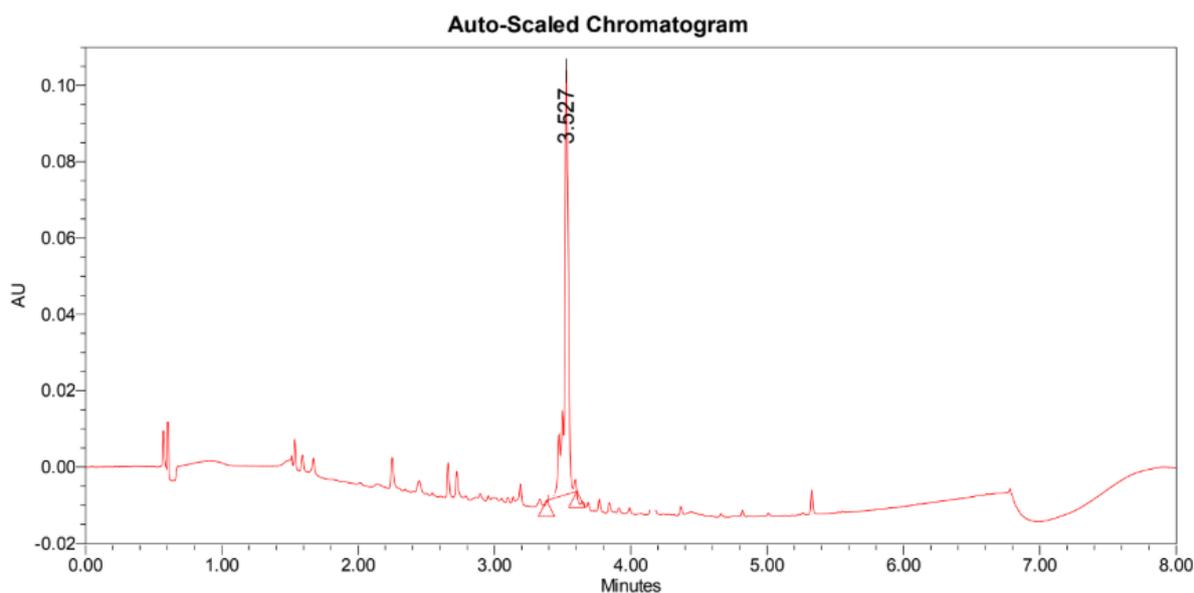
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)		Radiochemical Yield (%)	
	a + b	a	b	a	b
1	50	51	25	26	12
2	50	49	24	24	12
3	49	45	30	22	15
Radiochemical Yield (%) ± Deviation				24 ± 2	13 ± 1

**Table S45.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]8d using the reagent [<sup>18</sup>F]5f

Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)		Radiochemical Yield (%)	
	a + b	a	b	a	b
1	62	64	26	40	16
2	53	69	23	37	12
3	45	68	22	31	10
Radiochemical Yield (%) ± Deviation				36 ± 4	13 ± 2



**Figure S97.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]8d (gradient B).



**Figure S98.** UPLC UV-chromatogram of the authentic reference 8da (gradient B).

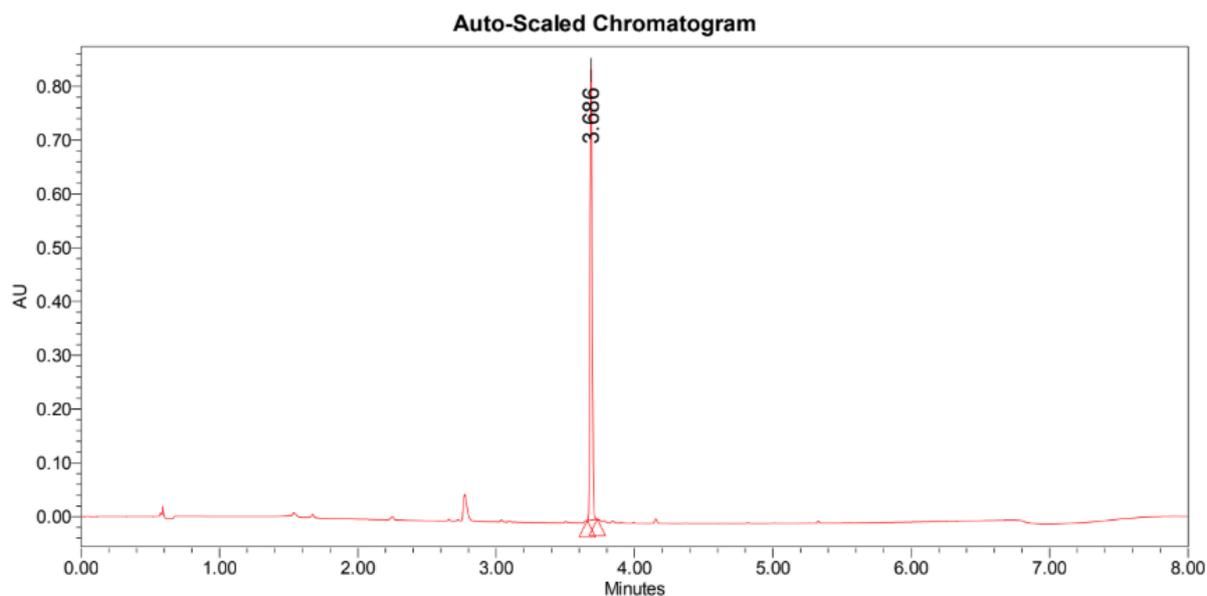
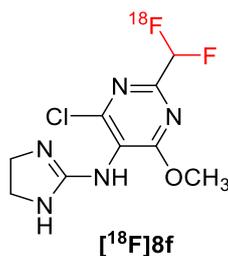


Figure S99. UPLC UV-chromatogram of the authentic reference **8db** (gradient B).

### 2.7.5. Synthesis of [<sup>18</sup>F]4-chloro-2-(difluoromethyl)-N-(4,5-dihydro-1H-imidazol-2-yl)-6-methoxypyrimidin-5-amine ([<sup>18</sup>F]**8f**)



The implementation of the general procedure for the C-H <sup>18</sup>F-difluoromethylation of 4-chloro-*N*-(4,5-dihydro-1*H*-imidazol-2-yl)-6-methoxypyrimidin-5-amine (4.6 mg, 0.02 mmol) provided the labeled compound [<sup>18</sup>F]**8f** in 52 ± 6%, 17 ± 3%, and 60 ± 3% RCY, using the reagents [<sup>18</sup>F]**5a**, [<sup>18</sup>F]**5c**, and [<sup>18</sup>F]**5f**, respectively (see the Tables S46-S48 for more details of the RCY determination). The UPLC radio-chromatogram of the crude product [<sup>18</sup>F]**8f** is depicted in Figure S100. Figure S101 represents the UPLC UV-chromatogram of the non-radioactive reference **8f**.

Table S46. Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**8f** using the reagents [<sup>18</sup>F]**5a**

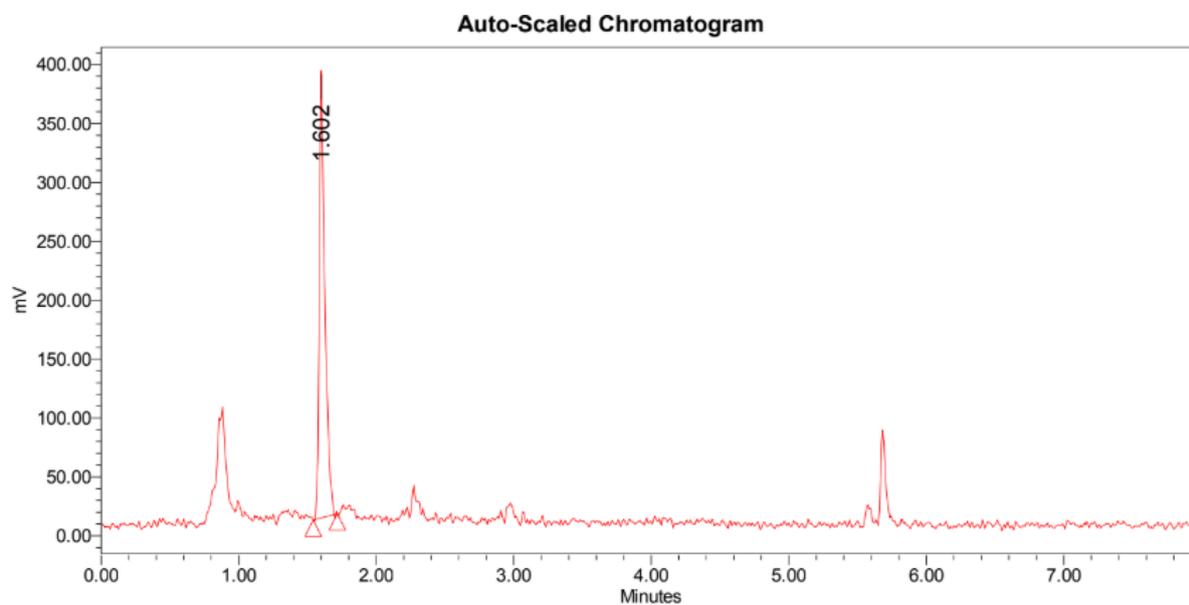
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	50	100	50
2	47	100	47
3	60	100	60
Radiochemical Yield (%) ± Deviation			52 ± 6

Table S47. Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]**8f** using the reagents [<sup>18</sup>F]**5c**

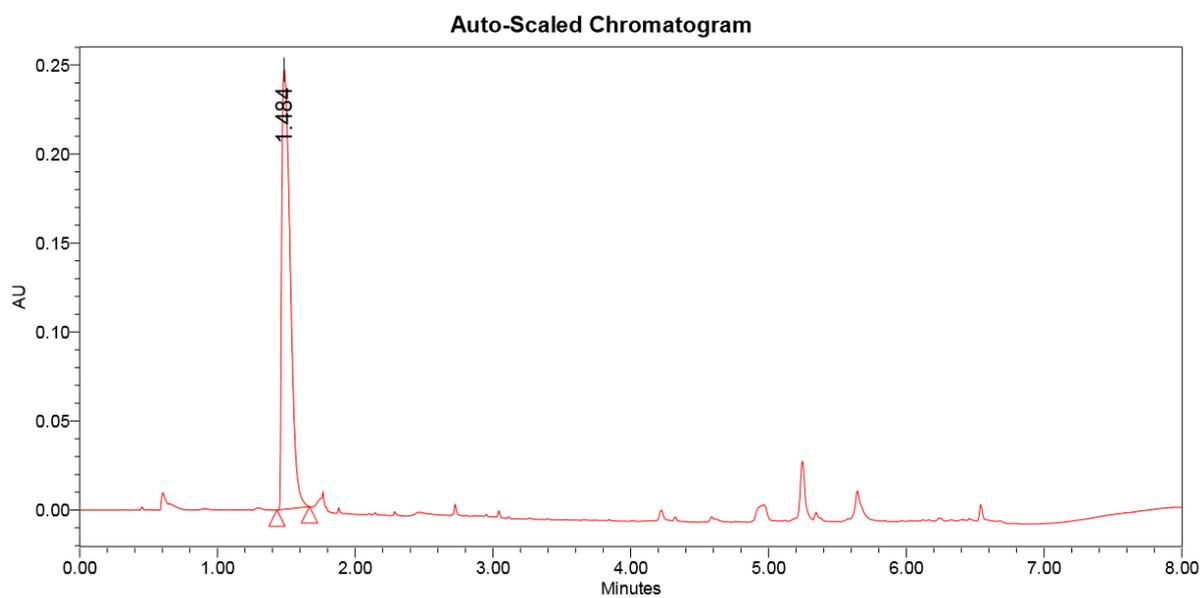
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	67	35	23
2	60	27	16
3	53	26	14
4	64	26	17
5	63	24	15
Radiochemical Yield (%) ± Deviation			17 ± 3

**Table S48.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]8f using the reagents [<sup>18</sup>F]5c

Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	73	84	61
2	67	83	56
3	78	80	62
Radiochemical Yield (%) ± Deviation			60 ± 3

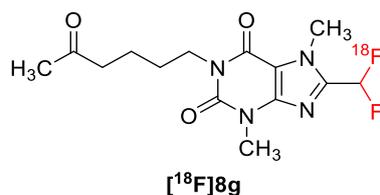


**Figure S100.** UPLC radio-chromatogram of the crude product [<sup>18</sup>F]8f (gradient B).



**Figure S101.** UPLC UV-chromatogram of the authentic reference 8f (gradient B).

### 2.7.6. Synthesis of 8-(difluoromethyl)-3,7-dimethyl-1-(5-oxohexyl)-3,7-dihydro-1H-purine-2,6-dione ([<sup>18</sup>F]8g)



The implementation of the general procedure for the C-H <sup>18</sup>F-difluoromethylation of 3,7-dimethyl-1-(5-oxohexyl)-3,7-dihydro-1H-purine-2,6-dione (5.6 mg, 0.02 mmol) provided the labeled compound [<sup>18</sup>F]8g in 52 ± 6%, 17 ± 3%, and 60 ± 3% RCY, using the reagents [<sup>18</sup>F]5a, [<sup>18</sup>F]5c, and [<sup>18</sup>F]5f, respectively (see the Tables S49-S51 for more details of the RCY determination). The UPLC radio-chromatogram of the crude product [<sup>18</sup>F]8g is depicted in Figure S102. Figure S103 represents the UPLC UV-chromatogram of the non-radioactive reference 8g.

**Table S49.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]8g using the reagents [<sup>18</sup>F]5a

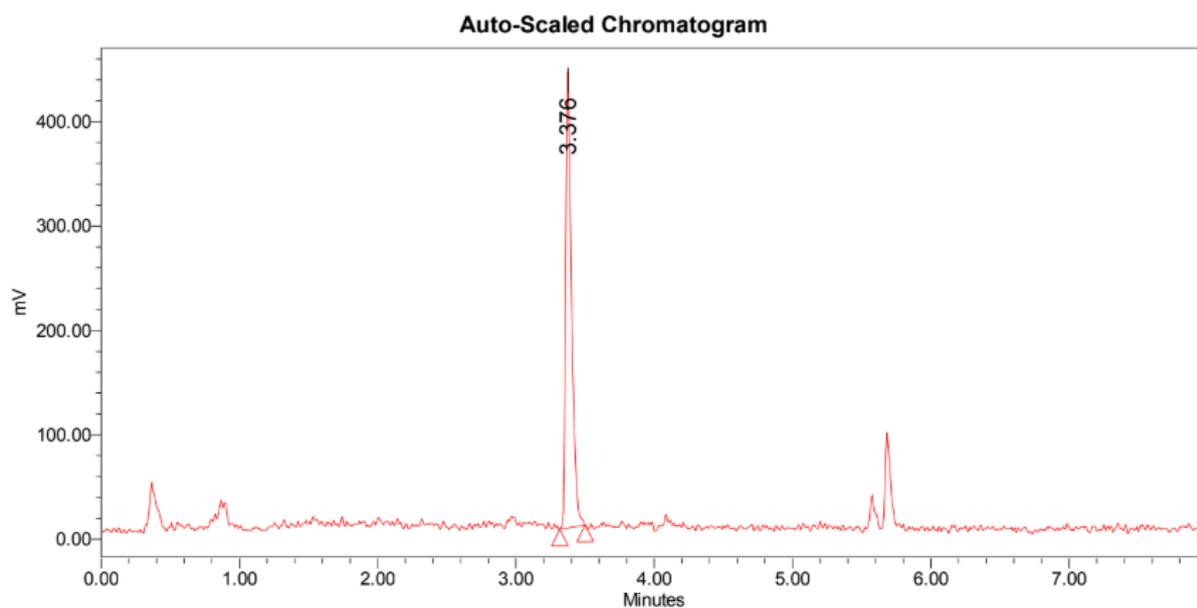
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	18	100	18
2	23	100	23
3	22	100	22
<b>Radiochemical Yield (%) ± Deviation</b>			<b>21 ± 2</b>

**Table S50.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]8g using the reagents [<sup>18</sup>F]5c

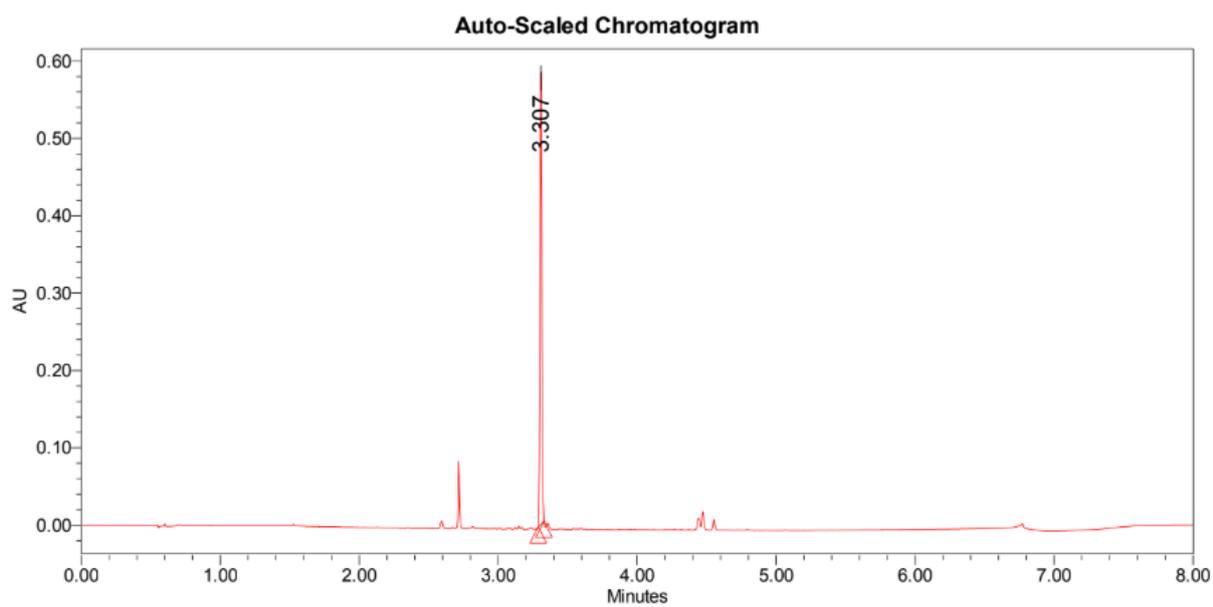
Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	42	58	24
2	56	62	35
3	61	61	37
4	48	53	25
<b>Radiochemical Yield (%) ± Deviation</b>			<b>30 ± 6</b>

**Table S51.** Determination of the radiochemical yield (%) of the synthesis of [<sup>18</sup>F]8g using the reagents [<sup>18</sup>F]5f

Reaction	Radio-TLC purity (%)	Radio-UPLC purity (%)	Radiochemical Yield (%)
1	52	60	31
2	47	82	39
3	42	81	34
<b>Radiochemical Yield (%) ± Deviation</b>			<b>35 ± 3</b>



**Figure S102.** UPLC radio-chromatogram of the crude product  $[^{18}\text{F}]\mathbf{8g}$  (gradient B).



**Figure S103.** UPLC UV-chromatogram of the authentic reference  $\mathbf{8g}$  (gradient B).