

Article

Supporting Information: Hydroalkoxylation of Terminal and Internal Alkynes Catalyzed by Dinuclear Gold(I) Complexes with Bridging Di(N-Heterocyclic Carbene) Ligands

Elena Marcheggiani ¹, Cristina Tubaro ¹, Andrea Biffis ¹, Claudia Graiff ² and Marco Baron ^{1,*}

1. NMR spectra of compound AuIL⁷

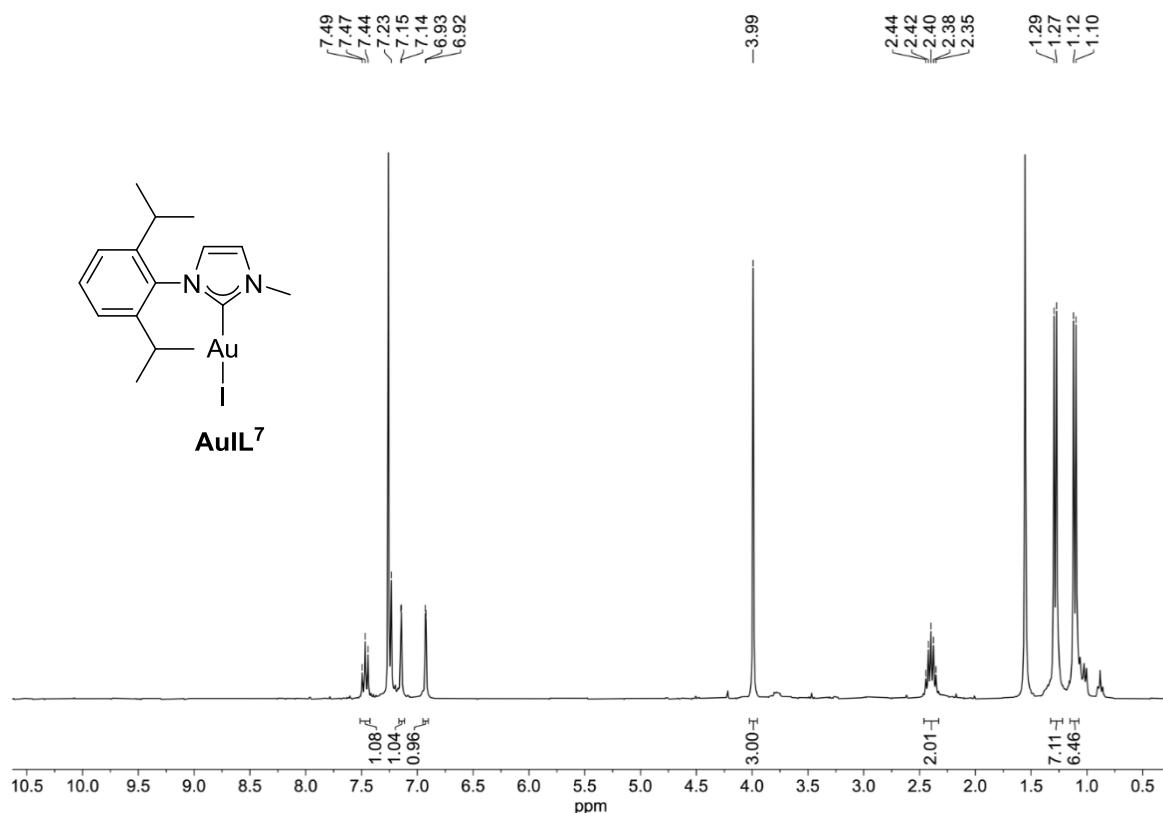


Figure S1. ¹H-NMR spectrum (CDCl₃, 300 MHz) of **AuIL⁷**.

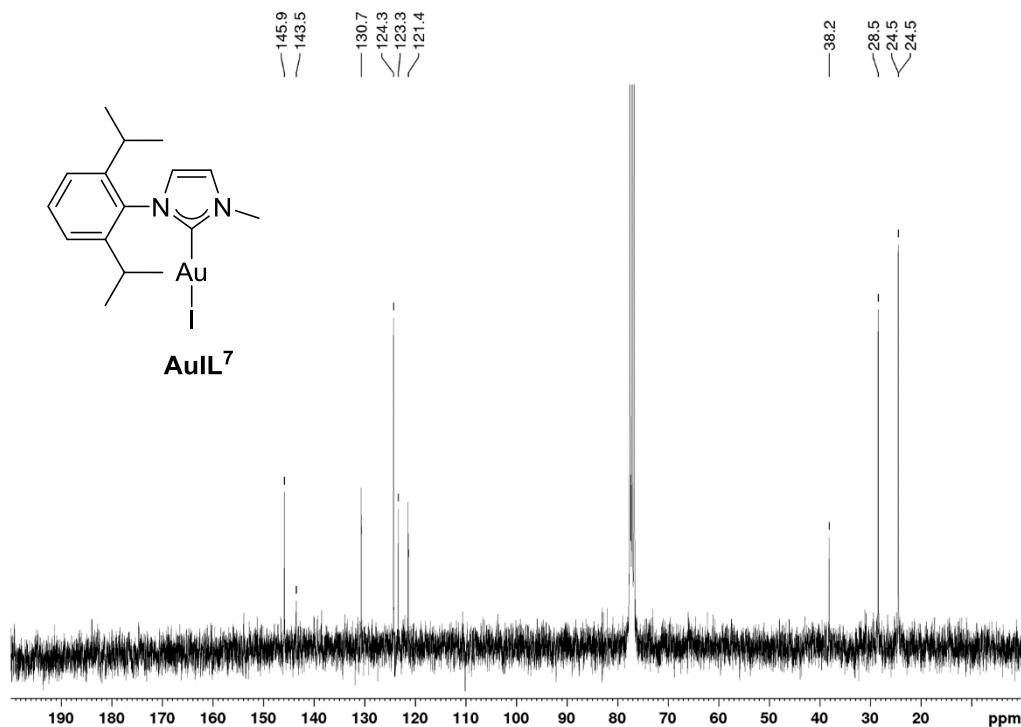


Figure S2. $^{13}\text{C}\{\text{H}\}$ NMR (CDCl_3 , 75.5 MHz) of AuIL^7 .

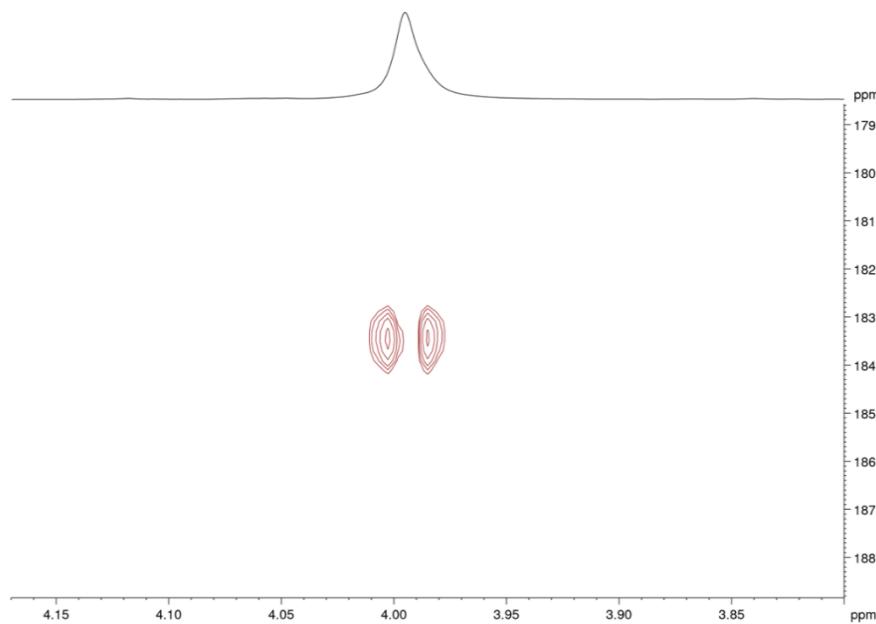


Figure S3. Portion of the $^1\text{H}, ^{13}\text{C}$ -HMBC spectrum of AuIL^7 in CDCl_3 showing the resonance of the carbene carbon.

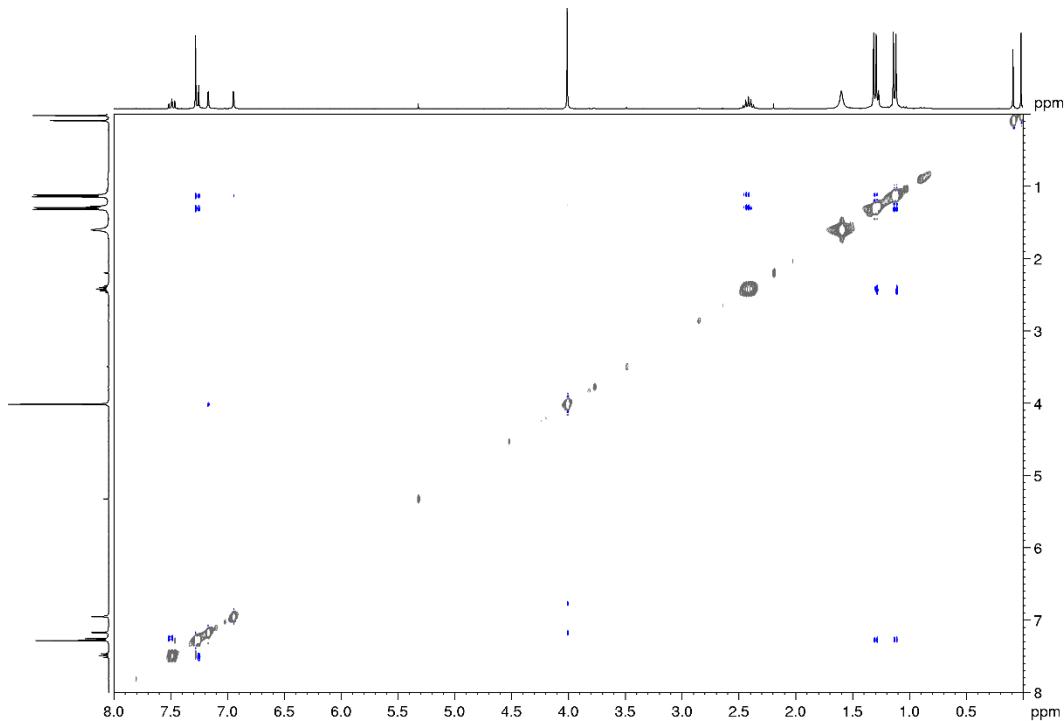


Figure S4. $^1\text{H},^1\text{H}$ -NOESY NMR spectrum of AuIL^7 in CDCl_3 .

2. Crystallographic data of compound $\text{Au}_2\text{Br}_2\text{L}^4$

Table S1. Crystallographic data of compound $\text{Au}_2\text{Br}_2\text{L}^4$

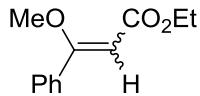
Compound	$\text{Au}_2\text{Br}_2\text{L}^4$
Formula	$\text{C}_{33.5}\text{H}_{44.5}\text{Au}_2\text{Br}_2\text{Cl}_3\text{N}_4$
Molecular Weight	1163.33
Crystal system	Monoclinic
Space group	$\text{P}2_1/\text{n}$
$a[\text{\AA}]$	13.5252(4)
$b[\text{\AA}]$	12.9755(4)
$c[\text{\AA}]$	23.9510(7)
$\alpha[^{\circ}]$	90
$\beta[^{\circ}]$	91.8390(10)
$\gamma[^{\circ}]$	90
$V[\text{\AA}^3]$	4201.1(2)
Temperature (K)	200(2)
Z	4
$D_{\text{calc}}[\text{g cm}^{-3}]$	1.839
$\mu[\text{cm}^{-1}]$	9.095

F(000)	2210.0
Reflections collected	52909
Independent reflections	10401
Reflections in refinement	10401
R(int)	0.0400
Refined parameters	419
R ₁ [$I > 2\sigma(I)$]	R ₁ = 0.0509 wR ₂ = 0.1330
wR ₂ [all data]	R ₁ = 0.0735 wR ₂ = 0.1537
GOF	1.052
CCDC	1968216

$$R_1 = \sum |F_O - F_C| / \sum (F_O); wR_2 = [\sum [w(F_O^2 - F_C^2)^2] / \sum [w(F_O^2)^2]]^{1/2}.$$

3. Identification of the products of the alkyne hydroalkoxidation reactions

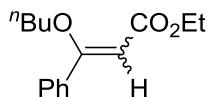
3aa. (E/Z) 3-methoxy-3-phenyl-ethylacrylate [1]



1H-NMR Z-Isomer (200 MHz, CDCl₃) δ ppm: 7.63-7.31 (m, 5H), 5.54 (s, 1H), 4.22 (q, J=7.1 Hz, 2H), 3.84 (s, 3H), 1.32 (t, J=7.1, 3H).

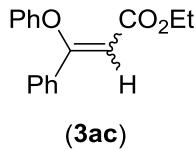
1H-NMR E-Isomer (200 MHz, CDCl₃) δ ppm: 7.63-7.31 (m, 5H), 5.27 (s, 1H), 4.05 (q, J=7.1 Hz, 2H), 3.81 (s, 3H), 1.15 (t, J=7.1Hz, 2H).

3ab. (E/Z)-Ethyl 3-butoxy-3-phenylacrylate



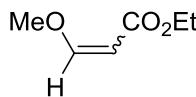
NMR signals of product **3ab** were identified by similarity with those of the analogous product **3aa**.

3ac. ethyl 3-phenoxy-3-phenylacrylate [2]



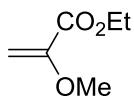
¹H NMR Z-Isomer (400 MHz, CDCl₃) δ 7.60 (d, J = 8.0 Hz, 2H), 7.37-7.31 (m, 3H), 7.25-7.20 (m, 2H), 6.97-6.94 (m, 3H), 6.15 (s, 1H), 4.12 (q, J = 8.0 Hz, 2H), 1.18 (t, J = 8.0 Hz, 3H) ppm.

3ca. (Z/E)-Ethyl 3-methoxyacrylate



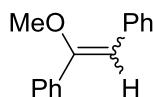
Product **3ca** is commercially available and the NMR signals can be found in on-line databases.

3ca'. Ethyl 2-methoxyacrylate



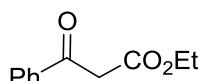
NMR signals of product **3ca'** were identified by similarity with those of the analogous species methyl 2-methoxyacrylate [3].

3ea. (Z)-1-methoxy-1,2-diphenylacetylene [4]



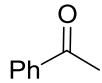
¹H-NMR Z-Isomer (250 MHz, CDCl₃) δ ppm: 3.55 (s, 3H), 6.02 (s, 1H), 7.13-7.16 (m, 1H), 7.24-7.32 (m, 5H), 7.48 (dd, J = 8.0 Hz, J = 1.5 Hz, 2H), 7.63 (d, J = 7.3 Hz, 2H).

5a. 3-oxo-3-phenyl Ethylpropionate [5]



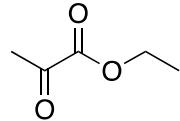
¹H-NMR of the keto form (400 MHz, CDCl₃) δ ppm: 1.22-1.26 (m, 3H), 3.96 (s, 2H), 4.17-4.29 (m, 2H), 7.37-7.48 (m, 2H), 7.55-7.59 (m, 1H), 7.92-7.95 (m, 2H).

5b. Acetophenone [6]



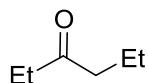
1H-NMR (250 MHz, CDCl₃) δ ppm: 2.54 (s, 3H), 7.36-7.53 (m, 3H), 7.88-7.93 (m, 2H).

5c. Ethyl pyruvate [7]



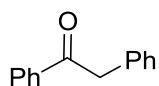
1H-NMR (400 MHz, CDCl₃) δ ppm: 1.38 (t, 3H), 2.47 (s, 3H), 4.32 (q, 2H).

5d. 3-hexanone [8]



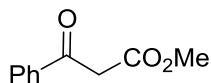
1H-NMR (200 MHz, CDCl₃) δ ppm: 0.89 (s, 3H), 1.02 (s, 3H), 1.60 (q, J = 7.4 Hz, 2H), 2.49-2.26 (m, 4H).

5e. 1,2-diphenylethanone [9]



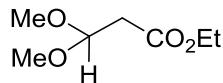
1H-NMR (400 MHz, CDCl₃) δ ppm: 4.32 (s, 2H), 7.31-7.29 (m, 3H), 7.36 (t, J = 7.3 Hz, 2H), 7.49 (t, J = 7.6 Hz, 2H), 7.59 (t, J = 7.4 Hz, 1H), 8.05 (d, J = 7.7 Hz, 2H).

5g. Methyl 3-oxo-3-phenylpropionate [10]



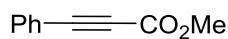
1H-NMR (400 MHz, CDCl₃) δ ppm: 3.75 (s, 3H), 4.00 (s, 2H), 7.50-7.46 (m, 2H), 7.61-7.57 (m, 1H), 7.95-7.93 (m, 2H).

4ca. Ethyl 3,3-dimethoxypropionate [11]



1H-NMR (400 MHz, CDCl₃) δ ppm: 1.25 (t, J=3.6 Hz, 3H), 2.63 (d, J=6.0 Hz, 2H), 3.35 (s, 6H), 4.15 (q, J=3.6 Hz, 2H), 4.82 (t, J=6.0 Hz, 1H).

1g. Methyl phenylpropiolate [12]



1H-NMR (300 MHz, CDCl₃) δ ppm: 3.85 (s, 3H), 7.32-7.42 (m, 2H), 7.42 - 7.49 (m, 1H), 7.54-7.64 (m, 2H).

4. Catalysts screening in the hydromethoxylation of ethyl phenylpropiolate

Table S2. Catalysts screening in the hydromethoxylation of ethyl phenylpropiolate.

Entry	Cat (mol %)	t (h)	Alkyne Conversion (%) ^a	Yield (%) ^a
1	Au₂Br₂L¹	1.5	67	(E)-3aa (9); (Z)-3aa (13); 4aa (5); 5a (40)
		6.5	91	4aa (30); 5a (61)
		22	92	4aa (5); 5a (87)
2	Au₂Br₂L²	1.5	78	(E)-3aa (14); 5a (64)
		6.5	90	(Z)-3aa (10); 5a (80)
3	Au₂Br₂L³	1.5	62	(E)-3aa (8); (Z)-3aa (43); 4aa (7); 5a (4)
		6.5	86	(E)-3aa (1); (Z)-3aa (2); 4aa (53); 5a (30)
		22	87	4aa (19); 5a (68)
4	Au₂Br₂L⁴	1	100	(E)-3aa (33); 5a (67)
		18.5	100	5a (100)
5	Au₂Br₂L⁵	1	86	(E)-3aa (27); (Z)-3aa (39); 4aa (9); 5a (11)
		18.5	89	4aa (17); 5a (72)
6	Au₂Br₂L⁶	1	72	(E)-3aa (32); (Z)-3aa (24); 4aa (6); 5a (10)
		23	100	5a (100)

Reaction conditions: Ethyl phenylpropiolate (0.6 mmol), MeOH (0.5 mL, 12.3 mmol), **Au₂Br₂L** (0.006 mmol), AgOTf co-cat. (0.012 mmol), 40 °C. ^a Alkyne conversion and product yields have been determined by ¹H NMR.

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